Lab 2: Miscellaneous Calculations

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i Date

- Lab Date: 23 September 2024 (Monday)
- **Due Date:** 7:00, 30 September 2024 (Monday)

1 Background

Students will work as individuals; assignments are per individual. This lab is due on Monday 30 September 2024 at 7:00 on iKamva.

1.1 Pre-Lab

Read this lab and contextualise within the pertinent material in your text.

1.2 Post-Lab

Upon completion of this lab:

• transcribe all tables and questions (Exercises A-E) to an electronic document and submit on iKamva. To submit online on Monday 30 September 2024 at 7:00.

2 Questions

2.1 Question 1: Dilutions (10 marks)

A 1.5% (mass:volume) carrageenan gel consists of 0.75g of carrageenan dissolved in 50 ml of 1% KCl. You accidentally added 0.87g to the 50 ml.

- i. What percentage gel have you inadvertently prepared?
- ii. How much additional water must be added to the 50 ml to achieve the desired 1.5% gel?
- iii. Carrageenan is a polysaccharide. Identify its biological origin: which photoautotrophs synthesise it?
- iv. Explain its ecological role in those organisms.
- v. Describe how humans exploit carrageenan's properties in food or industry.

2.2 Question 2: Quantum Light Measurements (4 marks)

monochromatic blue light source (420 nm) provides a photon flux density of 120 μ mol photons.m $^{-2}$.s $^{-1}$.

How many photons will fall on an area of 25 cm² over a 2-hour period?

(Be sure to convert the area into square metres, and assume flux density remains constant.)

2.3 Question 3: Plant Growth Rates (9 marks)

For each scenario below:

- Identify the biological process the measurements represent;
- Write down a suitable equation for calculating the process;
- Calculate the rate;
- State the resulting units.

Scenario i (4 marks)	Scenario ii (5 marks)
• Day 1: Plant biomass of 99 g	• Time, 0 minutes: 7.95 mg/L O ₂
• Day 100: Plant biomass of 149 g	• Time, 20 minutes: 11.39 mg/L O_2
	 Algal biomass: 2.3 g fresh mass

(In Scenario ii, calculate the oxygen production rate per gram of algal fresh mass.)

2.4 Question 4: Light Attenuation (15 marks)

You are a marine scientist assessing light penetration in the water column off Richards Bay, KZN. Measurements are taken in 5 m increments down to 50 m. Two stations are sampled:

- 1 km from shore (incident radiation at surface, 8:00: 1,213 μmol photons.m⁻².s⁻¹)
- 20 km from shore (incident radiation at surface, 9:35: 2,166 µmol photons.m⁻².s⁻¹)

Unfortunately, your submersible light meter was left in the lab; only surface values are available.

1. Using the Beer–Lambert law $I_z = I_0 e^{-kz}$, construct vertical light intensity curves for each site (surface to 50 m). State any assumptions made about the attenuation coefficient k.

- 2. Justify why these theoretical curves are a reasonable approximation of reality.
- 3. Identify physical and methodological factors that would cause deviation from the actual in situ profiles.

2.5 Question 5: Photosynthetic Rate Calculation (10 marks)

A leaf in full sunlight absorbs 10 mol of photons per square meter per second (mol m^{-2} s⁻¹). Its quantum yield is 0.05 mol of CO₂ fixed per mol of photons absorbed.

Calculate the photosynthetic rate (in µmol CO₂ m⁻² s⁻¹) of the leaf under these conditions.

2.6 Question 6: Relative Growth Rate (RGR) (5 marks)

The biomass of a plant at time t_0 is 50 g. After 10 days (time t_1), the biomass is 80 g.

Calculate the relative growth rate (RGR) in g g⁻¹ day⁻¹ using the equation:

$$RGR = \frac{\ln(W_1) - \ln(W_0)}{t_1 - t_0}$$

Where:

- (W_1) is the biomass at time (t_1)
- (W_0) is the biomass at time (t_0)

Explain why logarithmic growth is used instead of simple arithmetic increase.

2.7 Question 7: Respiration Rate and Plant Carbon Balance (5 marks)

A plant in darkness consumes 5 mg CO_2 per hour for respiration. During the day, its photosynthetic rate is 15 mg CO_2 per hour. Assume 12 hours light and 12 hours dark.

Calculate the net carbon balance of the plant over a 24-hour period. Is it positive or negative?

(Clarify whether the rates refer to whole-plant values or per unit biomass in your calculation.)

2.8 Question 8: Additive Light Intensity at Different Depths in Water (10 marks)

In an aquatic experiment, photon flux density at 2 m depth arises from multiple sources:

- Direct sunlight = $400 \mu mol photons m^{-2} s^{-1}$
- Diffuse underwater reflections = 120 μ mol photons m⁻² s⁻¹
- Scattered light from particles = $50 \mu mol photons m^{-2} s^{-1}$
- 1. Calculate the total photon flux density at a depth of 2 meters.
- 2. If a plant requires $\geq 500 \, \mu \text{mol photons m}^{-2} \, \text{s}^{-1}$ for photosynthesis, is the requirement met?
- 3. Briefly discuss how turbidity or scattering would affect these additive contributions in natural waters.

2.9 Question 9: Spectrally resolved attenuation and the euphotic boundary (12 marks)

At noon the surface PAR is $I_0=1,800~\mu{\rm mol}$ photons ${\rm m^{-2}s^{-1}}$. Treat the incident spectrum as two quasi-monochromatic bands:

- "Blue–green" centred at 490 nm, comprising 45% of I_0 , with diffuse attenuation $k_{BG}=0.040~\rm m^{-1}$.
- "Red–yellow" centred at 620 nm, comprising 55% of I_0 , with diffuse attenuation $k_{RY}=0.25~{\rm m}^{-1}.$

Assume independent exponential loss with depth and no internal sources.

- 1. Write an explicit expression for total PAR at depth z as the sum of the two bands, $I(z) = I_{BG}(z) + I_{RY}(z)$. State the units at each step.
- 2. Compute the 1% light level $z_{1\%}$ defined by $I(z_{1\%})=0.01\,I_0$. Because the sum of two exponentials lacks a closed-form inverse, determine $z_{1\%}$ to the nearest metre by a transparent numerical bracketing (show at least three evaluated depths and your final interpolation).
- 3. At $z=30\,\mathrm{m}$ and $z=60\,\mathrm{m}$, calculate the fractional contribution of the blue–green band to total PAR, i.e., $f_{BG}(z)=I_{BG}(z)/I(z)$. Interpret the change in f_{BG} with depth in one or two sentences.

(Note. Answers that hide the unit conversions or skip the bracketing will be penalised.)

Bibliography