

Introduction to Quantum Mechanics by David J. Griffiths Notes

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Contents

I	Theory	1
1	The Wave Function	2
1.1	The Schrödinger Equation	2
1.2	The Statistical Interpretation	2
1.3	Probability	3
1.3.1	Discrete Variables	3
1.3.2	Continuous Variables	4
1.4	Normalization	4
1.5	Momentum	5
1.6	The Uncertainty Principle	6
2	Time-Independent Schrödinger Equation	6
2.1	Stationary States	6
2.2	The Infinite Square Well	9
2.3	The Harmonic Oscillator	11
2.3.1	Algebraic Method	12
2.3.2	Analytic Method	14
2.4	The Free Particle	15
2.5	The Delta-Function Potential	16
2.6	Bound States and Scattering States	16

Part I

Theory

1 The Wave Function

1.1 The Schrödinger Equation

- The **Schrödinger equation**

$$i\hbar \frac{\partial \Psi}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + V\Psi$$

is to quantum mechanics what Newton's second law is to classical mechanics. Given suitable initial conditions — typically $\Psi(x, 0)$ — the Schrödinger equation determines $\Psi(x, t)$ for all future time.

1.2 The Statistical Interpretation

- The **Born rule** states that $|\Psi(x, t)|^2$ gives the probability of finding the particle at point x at time t or

$$\int_a^b |\Psi(x, t)|^2 dx$$

gives the probability of finding the particle between a and b at time t .

- This statistical interpretation introduces indeterminacy to quantum mechanics — we can't predict with certainty the particle's position.
- Suppose we measure a particle's position to be C . Where was it before we took the measurement? In the past there were three main schools of thought:
 1. The **realist** position believes that the particle was at C but Ψ doesn't give us enough information to determine that — there's another **hidden variable** that would allow us to.
 2. The **orthodox** position (also known as the **Copenhagen interpretation**) believes that the particle didn't have a definite position but the act of measuring it forced it to do so.
 3. The **agnostic** position believes that it doesn't matter and is potentially unknowable.
- **Bell's theorem** confirms the orthodox interpretation.
- If we take two consecutive measurements of a particle, they will both yield the same result. The first measurement causes the wavefunction to **collapse** such that it is peaked only at the particle's measured location. If the system is allowed to evolve between the measurements the wavefunction will "spread out" but if done in quick succession the result won't change.

1.3 Probability

1.3.1 Discrete Variables

- The average value of a discrete variable j is

$$\langle j \rangle = \frac{\sum jN(j)}{N} = \sum_{j=0}^{\infty} jP(j)$$

where N is the size of the population, $N(j)$ is the number of j s in the population, and $P(j)$ is the probability of randomly selecting a j from the population.

- In quantum mechanics the average is usually the quantity of interest and it is called the **expectation value** (even though it may not be the most probable value).
- The average value of a function f of a discrete variable j is

$$\langle f(j) \rangle = \sum_{j=0}^{\infty} f(j)P(j).$$

- Two distributions could have the same median, mean, mode, and size but be spread out differently. One way to quantify this could be to calculate how far each element is from the average

$$\Delta j = j - \langle j \rangle$$

and calculate the average of Δj . However, it ends up being zero

$$\begin{aligned}\langle \Delta j \rangle &= \sum (j - \langle j \rangle)P(j) \\ &= \sum jP(j) - \langle j \rangle \sum P(j) \\ &= \langle j \rangle - \langle j \rangle \\ &= 0.\end{aligned}$$

To avoid this we calculate the average of the square of Δj

$$\sigma^2 = \langle (\Delta j)^2 \rangle$$

which is known as the **variance** of the distribution.

- The square root of the variance is called the **standard deviation**.

- A useful theorem on variances is

$$\begin{aligned}
\sigma^2 &= \langle (\Delta j)^2 \rangle \\
&= \sum (\Delta j)^2 P(j) \\
&= \sum (j - \langle j \rangle)^2 P(j) \\
&= \sum (j^2 - 2j\langle j \rangle + \langle j \rangle^2) P(j) \\
&= \sum j^2 P(j) - 2\langle j \rangle \sum j P(j) + \langle j \rangle^2 \sum P(j) \\
&= \langle j^2 \rangle - 2\langle j \rangle^2 + \langle j \rangle^2 \\
&= \langle j^2 \rangle - \langle j \rangle^2
\end{aligned}$$

and thus the standard deviation can be calculated as

$$\sigma = \sqrt{\langle j^2 \rangle - \langle j \rangle^2}$$

which is usually easier than the full formula.

1.3.2 Continuous Variables

- The equations above translate as expected to continuous variables:

$$\begin{aligned}
P_{ab} &= \int_a^b \rho(x) dx \\
1 &= \int_{-\infty}^{+\infty} \rho(x) dx \\
\langle x \rangle &= \int_{-\infty}^{+\infty} x \rho(x) dx \\
\langle f(x) \rangle &= \int_{-\infty}^{+\infty} f(x) \rho(x) dx \\
\sigma^2 &= \langle x^2 \rangle - \langle x \rangle^2
\end{aligned}$$

1.4 Normalization

- In order for the statistical interpretation of the wavefunction to make sense, it must be the case that

$$\int_{-\infty}^{\infty} |\Psi(x, t)|^2 dx = 1,$$

i.e. the particle must be somewhere.

- The process of multiplying a candidate wavefunction by a complex constant A to make this hold is called **normalization**.

- For some solutions to the Schrödinger equation, the integral is infinite in which case there is no A that normalizes the wavefunction. The same goes for $\Psi = 0$. Such **non-normalizable** solutions can't represent particles so they must be rejected.
- But if we normalize the wavefunction at $t = 0$, how do we know it stays normalized? It turns out that the integral

$$\int_{-\infty}^{\infty} |\Psi(x, t)|^2 dx$$

is constant (independent of time) so if it's normalized at $t = 0$ it stays normalized.

1.5 Momentum

- What does

$$\langle x \rangle = \int_{-\infty}^{\infty} x |\Psi(x, t)|^2 dx$$

mean? It doesn't mean that if you measure the position of a particle over and over again and take the average of the results you'll get $\int x |\Psi|^2 dx$. The first measurement causes the wavefunction to collapse so you'll get the same measurement each time. Instead, it means if you have a large number of particles all in the same state Ψ , measure the position of each of them, and take the average of the results you'll get $\int x |\Psi|^2 dx$.

- The expectation value of the velocity is equal to the time derivative of the expectation value of the position

$$\langle v \rangle = \frac{d\langle x \rangle}{dt} = -\frac{i\hbar}{m} \int \Psi^* \frac{\partial \Psi}{\partial x} dx$$

however it is customary to work with momentum $p = mv$ rather than velocity

$$\langle p \rangle = m \frac{d\langle x \rangle}{dt} = -i\hbar \int \Psi^* \frac{\partial \Psi}{\partial x} dx.$$

- The above can be rewritten in the form

$$\begin{aligned} \langle x \rangle &= \int \Psi^* [x] \Psi dx \\ \langle p \rangle &= \int \Psi^* \left[-i\hbar \frac{\partial}{\partial x} \right] \Psi dx. \end{aligned}$$

The values in the square brackets are called **operators** and we say that the operator x “represents” position and the operator $-i\hbar(\partial/\partial x)$ “represents” momentum. To calculate expectation values we place the appropriate operator between Ψ^* and Ψ and integrate.

- All other values of interest can be expressed in terms of position and momentum, e.g. kinetic energy is

$$T = \frac{1}{2}mv^2 = \frac{p^2}{2m}.$$

Their operators can be determined by substituting $p = -i\hbar(\partial/\partial x)$, e.g. kinetic energy is

$$-\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2}$$

such that

$$\langle T \rangle = -\frac{\hbar^2}{2m} \int \Psi^* \frac{\partial^2 \Psi}{\partial x^2} dx.$$

In general

$$\langle Q(x, p) \rangle = \int \Psi^* [Q(x, -i\hbar\partial/\partial x)] \Psi dx.$$

1.6 The Uncertainty Principle

- If a wave is spread out (e.g. if you shake a rope up and down repeatedly), its position isn't well defined but its wavelength is. On the other hand, if a wave is localised (e.g. if you shake a rope up and down once to make a spike), its position is well defined but its wavelength isn't.
- The momentum of a particle is related to its wavelength by the **de Broglie formula**

$$p = \frac{h}{\lambda} = \frac{2\pi\hbar}{\lambda}.$$

- If a wavefunction is periodic its position isn't well defined but its wavelength is and thus so is its momentum. If a wavefunction is localised its position is well defined but its wavelength isn't and thus neither is its momentum. This is formalised by the **Heisenberg uncertainty principle**

$$\sigma_x \sigma_p \geq \frac{\hbar}{2}$$

where σ_x and σ_p are the standard deviations in x and p , respectively.

2 Time-Independent Schrödinger Equation

2.1 Stationary States

- The Schrödinger equation

$$i\hbar \frac{\partial \Psi}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + V\Psi$$

can be solved via separation of variables if V is independent of t . In that case we assume $\Psi(x, t) = \psi(x)\varphi(t)$ and separation gives the two ODEs

$$i\hbar \frac{1}{\varphi} \frac{d\varphi}{dt} = E$$

which has no name and

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \psi}{\partial x^2} + V\psi = E\psi$$

which is called the **time-independent Schrödinger equation**. The solution to the former is

$$\varphi(t) = Ce^{-iEt/\hbar}$$

but the constant C can be absorbed into ψ . The latter can't be solved until V is specified.

- While there are many solutions to the Schrödinger equation that don't have the form $\psi(x)\varphi(t)$, those that do have three interesting properties:

1. They are stationary states. The wave function has the form

$$\Psi(x, t) = \psi(x)e^{-iEt/\hbar}$$

so the probability density is

$$|\Psi|^2 = \Psi^* \Psi = \psi^* e^{iEt/\hbar} \psi e^{-iEt/\hbar} = |\psi|^2,$$

i.e. it's constant in time. Similarly, time dependence drops out of the expectation value formula so they're also constant in time.

2. They are states of definite total energy. In classical mechanics the total energy is called the Hamiltonian

$$H(x, p) = \frac{p^2}{2m} + V(x).$$

The equivalent quantum operator is

$$\hat{H} = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} + V(x)$$

thus the time-independent Schrödinger equation can be written

$$\hat{H}\psi = E\psi$$

and the expectation value of the total energy can be written

$$\langle H \rangle = \int \psi^* \hat{H} \psi dx = E \int |\psi|^2 dx = E \int |\Psi|^2 dx = E.$$

Moreover,

$$\hat{H}^2\psi = \hat{H}(\hat{H}\psi) = \hat{H}(E\psi) = E(\hat{H}\psi) = E^2\psi$$

so

$$\langle H^2 \rangle = \int \psi^* \hat{H}^2 \psi dx = E^2 \int |\psi|^2 dx = E^2 \int |\Psi|^2 dx = E^2.$$

Thus the variance of H is

$$\sigma_H^2 = \langle H^2 \rangle - \langle H \rangle^2 = 0$$

meaning every measurement of the total energy of the system will return the same value E .

3. The general solution to the time-dependent Schrödinger equation is a linear combination of separable solutions. The time-independent Schrödinger equation yields an infinite number of solutions $\psi_1(x)$, $\psi_2(x)$, ..., $\psi_n(x)$ each with its associated separation constant E_1 , E_2 , ..., E_n — one for each **allowed energy**. The general solution is then given by

$$\Psi(x, t) = \sum_{n=1}^{\infty} c_n \psi_n(x) e^{-iE_n t/\hbar}$$

with appropriate choice of c_n to fit the initial conditions.

- The general approach to solving the time-dependent Schrödinger equation with given initial conditions and potential energy function is:
 1. Solve the time-independent Schrödinger equation to get an infinite set of solutions $\psi_n(x)$.
 2. Fit the linear combination of those solutions

$$\Psi(x, 0) = \sum_{n=1}^{\infty} c_n \psi_n(x)$$

to the initial conditions.

3. Multiply each term in the sum by its associated time dependence

$$\Psi(x, t) = \sum_{n=1}^{\infty} c_n \psi_n(x) e^{-iE_n t/\hbar} = \sum_{n=1}^{\infty} c_n \Psi_n(x, t).$$

- The physical meaning of the constants c_n is thus: $|c_n|^2$ is the probability that a measurement of the system's energy would return the value E_n . This means that

$$\sum_{n=1}^{\infty} |c_n|^2 = 1$$

and

$$\langle H \rangle = \sum_{n=1}^{\infty} |c_n|^2 E_n.$$

The probability of measuring any particular energy is constant in time and thus the expectation value of H — this is part of energy conservation.

- For a solution to the Schrödinger equation to be normalisable, E must be real and must exceed the minimum value of V .

2.2 The Infinite Square Well

- The potential of an infinite square well is

$$V(x) = \begin{cases} 0 & 0 \leq x \leq a \\ \infty & \text{otherwise} \end{cases}$$

and thus $\Psi(x, t) = 0$ for x outside $[0, a]$.

- Inside such a well the time-independent Schrödinger equation reads

$$-\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} = E\psi$$

where E must be real and greater than 0. This can be rearranged to

$$\frac{d^2\psi}{dx^2} = -k^2\psi$$

where

$$k = \frac{\sqrt{2mE}}{\hbar}.$$

This has the form of a simple harmonic oscillator and has the solution

$$\psi = A \sin(kx) + B \cos(kx).$$

Applying the boundary condition $\psi(0) = 0$ gives $B = 0$ and thus

$$\psi = A \sin(kx).$$

Applying the boundary condition $\psi(a) = 0$ gives $A \sin(ka) = 0$. We must reject $A = 0$ as that would result in the non-normalisable solution $\psi = 0$, so $\sin(ka) = 0$ and thus $ka = 0, \pm\pi, \pm2\pi, \dots$. Again, $k = 0$ results in $\psi = 0$ so it must be rejected and we're left with

$$\psi = A \sin(k_n x)$$

where

$$k_n = \frac{n\pi}{a}, \quad n \in \mathbb{Z}^+.$$

- From the above, the possible values of E are

$$E_n = \frac{\hbar^2 k_n^2}{2m} = \frac{n^2 \pi^2 \hbar^2}{2ma^2}, n \in \mathbb{Z}^+.$$

In other words, unlike in classical mechanics, a quantum particle in an infinite square well can't have any energy — it must be one of these values.

- To determine the value of the constant A above we normalise ψ

$$\int_0^a |A|^2 \sin^2(kx) dx = |A|^2 \frac{a}{2} = 1.$$

This only determines the magnitude of A but the phase doesn't affect anything so we might as well pick

$$A = \sqrt{\frac{2}{a}}$$

and the solutions to the time-independent Schrödinger equation inside the infinite square well are

$$\psi_n(x) = \sqrt{\frac{2}{a}} \sin\left(\frac{n\pi}{a}x\right).$$

- The solutions ψ_n have some interesting properties:

1. They alternate even and odd with respect to the centre of the well.
2. Each subsequent solution has one more node (crossing of the axis).
3. They are mutually orthogonal, i.e.

$$\int \psi_m^* \psi_n dx = \begin{cases} 0 & m \neq n \\ 1 & m = n \end{cases}.$$

4. They are complete, in that any other function $f(x)$ can be expressed as a linear combination of them

$$f(x) = \sum_{n=1}^{\infty} c_n \psi_n(x) = \sqrt{\frac{2}{a}} \sum_{n=1}^{\infty} c_n \sin\left(\frac{n\pi}{a}x\right)$$

where

$$c_n = \int \psi_n(x)^* f(x) dx.$$

- The stationary states of the infinite square well are thus

$$\Psi_n(x, t) = \sqrt{\frac{2}{a}} \sin\left(\frac{n\pi}{a}x\right) e^{-i(n^2 \pi^2 \hbar / 2ma^2)t}$$

and the general solution to the time-dependent Schrödinger equation is

$$\Psi(x, t) = \sum_{n=1}^{\infty} c_n \Psi_n(x, t)$$

where c_n are given by

$$c_n = \sqrt{\frac{2}{a}} \int_0^a \sin\left(\frac{n\pi}{a}x\right) \Psi(x, 0) dx.$$

2.3 The Harmonic Oscillator

- The equation of motion for a harmonic oscillator is

$$\frac{d^2x}{dt^2} = -kx$$

where k is the force constant and x is the displacement from equilibrium. The solution to this differential equation is

$$x = A \sin(\omega t) + B \cos(\omega t)$$

where $\omega = \sqrt{k/m}$ is the angular frequency of oscillation.

- The potential energy of a harmonic oscillator is

$$V(x) = \frac{1}{2}kx^2$$

which is a parabola.

- Any potential function can be approximated by a parabola in the neighborhood of a local minimum x_0 providing the amplitude is small by expanding it as a Taylor series, dropping the first term as it's constant and doesn't affect the force, dropping the second term because $V'(x_0) = 0$, keeping the third term, and dropping higher order terms.
- The quantum problem is to solve the Schrödinger equation for potential

$$V(x) = \frac{1}{2}m\omega^2x^2$$

where k has been replaced with $m\omega^2$. The first step is to solve the time-independent Schrödinger equation:

$$-\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} + \frac{1}{2}m\omega^2x^2\psi = E\psi.$$

This is typically done in one of two ways: a clever algebraic technique using ladder-operators, and a more straightforward “brute-force” technique using power series.

2.3.1 Algebraic Method

- We can factor the time-independent Schrödinger equation above as

$$\frac{1}{2m} \left[-\hbar^2 \frac{d^2}{dx^2} + (m\omega x)^2 \right] \psi = \frac{1}{2m} [\hat{p}^2 + (m\omega x)^2] \psi = E\psi.$$

As the Hamiltonian is equal to the total energy of the system, this means

$$\hat{H} = \frac{1}{2m} [\hat{p}^2 + (m\omega x)^2].$$

- This form invites factorisation of the form $u^2 + v^2 = (iu + v)(-iu + v)$ however \hat{p} and x are operators and they do not, in general, commute (i.e. $\hat{p}x \neq x\hat{p}$). We can try anyway by introducing the quantities

$$\hat{a}_{\pm} = \frac{1}{\sqrt{2\hbar m\omega}} (\mp i\hat{p} + m\omega x).$$

Their product is

$$\begin{aligned} \hat{a}_- \hat{a}_+ &= \frac{1}{2\hbar m\omega} (i\hat{p} + m\omega x)(-i\hat{p} + m\omega x) \\ &= \frac{1}{2\hbar m\omega} [\hat{p}^2 + (m\omega x)^2 - im\omega(x\hat{p} - \hat{p}x)]. \end{aligned}$$

The extra term involving $x\hat{p} - \hat{p}x$ is called the **commutator** of x and \hat{p} — it's a measure of how badly they fail to commute.

- In general, the commutator of two operators \hat{A} and \hat{B} is

$$[\hat{A}, \hat{B}] = \hat{A}\hat{B} - \hat{B}\hat{A}.$$

It's often easier to calculate the commutator of two operators if you introduce a test function f , expand, and remove f at the end.

- Using this notation, the product above can be rewritten

$$\hat{a}_- \hat{a}_+ = \frac{1}{2\hbar m\omega} [\hat{p}^2 + (m\omega x)^2] - \frac{i}{2\hbar} [x, \hat{p}].$$

- It turns out that

$$[x, \hat{p}] = i\hbar.$$

This equation is called the **canonical commutation relation** and using it we find

$$\hat{a}_- \hat{a}_+ = \frac{1}{\hbar\omega} \hat{H} + \frac{1}{2}$$

or

$$\hat{H} = \hbar\omega \left(\hat{a}_- \hat{a}_+ - \frac{1}{2} \right).$$

- Note that the order of the operators is important. Going through the same process for $\hat{a}_+\hat{a}_-$ gives

$$\hat{a}_+\hat{a}_- = \frac{1}{\hbar\omega}\hat{H} - \frac{1}{2}$$

and

$$\hat{H} = \hbar\omega \left(\hat{a}_+\hat{a}_- + \frac{1}{2} \right).$$

- If ψ satisfies the Schrödinger equation with energy E , i.e. $\hat{H}\psi = E\psi$, then $\hat{a}_+\psi$ satisfies it with energy $E + \hbar\omega$

$$\hat{H}(\hat{a}_+\psi) = (E + \hbar\omega)\psi$$

and $\hat{a}_-\psi$ satisfies it with energy $E - \hbar\omega$

$$\hat{H}(\hat{a}_-\psi) = (E - \hbar\omega)\psi.$$

- \hat{a}_\pm are called **ladder operators** because, if we know one solution to the Schrödinger equation, they can be used to climb up and down the energy ladder.
- However, we can't apply the lowering operator \hat{a}_- an infinite number of times — eventually we'll reach a state of zero energy. The “lowest rung” ψ_0 occurs when $\hat{a}_-\psi_0 = 0$. Solving this equation we find

$$\psi_0(x) = \left(\frac{m\omega}{\pi\hbar} \right)^{1/4} e^{-\frac{m\omega}{2\hbar}x^2}$$

and

$$E_0 = \frac{1}{2}\hbar\omega.$$

- From here we can apply the raising operator \hat{a}_+ to find higher energy states

$$\psi_n(x) = \frac{1}{\sqrt{n!}}(\hat{a}_+)^n\psi_0 \text{ with } E_n = \left(n + \frac{1}{2} \right) \hbar\omega.$$

- In the case of the infinite square well, the stationary states of the harmonic oscillator are orthogonal

$$\int_{-\infty}^{\infty} \psi_m^* \psi_n dx = \delta_{mn}.$$

2.3.2 Analytic Method

- By introducing the term

$$\xi = \sqrt{\frac{m\omega}{\hbar}}x$$

the Schrödinger equation for the harmonic oscillator

$$-\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} + \frac{1}{2}m\omega^2 x^2 \psi = E\psi$$

can be rewritten

$$\frac{d^2\psi}{d\xi^2} = (\xi^2 - K)\psi$$

where

$$K = \frac{2E}{\hbar\omega}$$

is the energy of the system in units of $\hbar\omega/2$.

- For large ξ this has the approximate solution

$$\psi(\xi) = h(\xi)e^{-\xi^2/2}.$$

Substituting this into the Schrödinger equation gives

$$\frac{d^2h}{d\xi^2} - 2\xi \frac{dh}{d\xi} + (K - 1)h = 0.$$

If we assume the solution $h(\xi)$ is a power series in ξ we get

$$\sum_{j=0}^{\infty} [(j+1)(j+2)a_{j+2} - 2ja_j + (K-1)a_j] \xi^j = 0.$$

By the uniqueness of power series expansions, the coefficient of each power of ξ must be 0 in order for the series overall to equal 0

$$(j+1)(j+2)a_{j+2} - 2ja_j + (K-1)a_j = 0.$$

This gives a recurrence relation

$$a_{j+1} = \frac{2j+1-K}{(j+1)(j+2)}a_j$$

that can be used to generate the even- and odd-numbered coefficients by starting with a_0 and a_1 , respectively.

- However, not all choices of a_j result in normalisable wavefunctions. In order for them to be valid, the power series must terminate — there must be some “highest” j (call it n) such that $a_{n+2} = 0$. This will truncate **either** the even series or the odd series. The other one must be zero from the start — $a_1 = 0$ if n is even, $a_0 = 0$ if n is odd.

- In general $h_n(\xi)$ will be a polynomial of degree n in ξ , involving even powers only if n is even, and odd powers only if it's odd. Apart from the overall factor a_0 or a_1 these are the **Hermite polynomials**, $H_n(\xi)$.
- By convention the arbitrary multiplicative factor is chosen such that the coefficient of the highest power of ξ is 2^n . With this convention, the normalised stationary states for the harmonic oscillator are

$$\psi_n = \left(\frac{m\omega}{\pi\hbar}\right)^{1/4} \frac{1}{\sqrt{2^n n!}} H_n(\xi) e^{-\xi^2/2}.$$

2.4 The Free Particle

- For a free particle, $V = 0$ everywhere. This gives a time-independent Schrödinger equation of

$$-\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} = E\psi$$

or

$$\frac{d^2\psi}{dx^2} = -k^2\psi \text{ where } k = \frac{\sqrt{2mE}}{\hbar}$$

the solution to which is

$$\psi(x) = Ae^{ikx} + Be^{-ikx}.$$

- Introducing the standard time dependence gives

$$\Psi(x, t) = Ae^{ik(x - \frac{\hbar k}{2m}t)} + Be^{-ik(x + \frac{\hbar k}{2m}t)}.$$

The first term represents a wave travelling to the left and the second term represents a wave travelling to the right. We can combine them

$$\Psi_k(x, t) = Ae^{i(kx - \frac{\hbar k^2}{2m}t)}$$

if we let k go negative

$$k = \pm \frac{\sqrt{2mE}}{\hbar} \text{ with } \begin{cases} k > 0 & \text{travelling to the right,} \\ k < 0 & \text{travelling to the left.} \end{cases}$$

- Unfortunately this wavefunction can't be normalised

$$\int_{-\infty}^{\infty} \Psi_k^* \Psi_k = |A|^2 \int_{-\infty}^{\infty} dx = |A|^2(\infty).$$

This means that it doesn't represent a physically realisable state. In other words, there's no such thing as a free particle with a definite energy.

- The solution is still a linear combination of separable solutions, but it's an integral over the continuous variable k instead of a sum over the discrete variable n

$$\Psi(x, t) = \frac{1}{\sqrt{2\pi}} \int_{-\infty}^{\infty} \phi(k) e^{i(kx - \frac{\hbar k^2}{2m}t)} dk$$

where $(1/\sqrt{2\pi})\phi(k) dk$ plays the role of the coefficient c_n in previous wavefunctions. This wavefunction can be normalised but it carries a range of k s and thus a range of energies and speeds. This is called a **wave packet**.

- In a typically quantum problem you'll be given $\Psi(x, 0)$ and be asked to find $\Psi(x, t)$. How do you determine $\phi(k)$ to match the initial wave function

$$\Psi(x, 0) = \frac{1}{\sqrt{2\pi}} \int_{-\infty}^{\infty} \phi(k) e^{ikx} dk.$$

The answer is given by **Plancherel's theorem**

$$f(x) = \frac{1}{\sqrt{2\pi}} \int_{-\infty}^{\infty} F(k) e^{ikx} dk \iff F(k) = \frac{1}{\sqrt{2\pi}} \int_{-\infty}^{\infty} f(x) e^{-ikx} dx$$

where $F(k)$ is the **Fourier transform** of $f(x)$. Thus

$$\phi(k) = \frac{1}{\sqrt{2\pi}} \int_{-\infty}^{\infty} \Psi(x, 0) e^{-ikx} dx.$$

- A wave packet is comprised of multiple constituent waves that have different velocities and wavelengths. The velocity of a constituent wave is called the **phase velocity** while the velocity of the packet as a whole is called the **group velocity**.
- The relationship between ω and k (in the equation above it's $\omega = \hbar k^2/2m$) is called the **dispersion relation** and it determines the group and phase velocities

$$v_{\text{group}} = \frac{d\omega}{dk}$$

$$v_{\text{phase}} = \frac{\omega}{k}.$$

2.5 The Delta-Function Potential

2.6 Bound States and Scattering States

- In classical mechanics, if the potential energy function $V(x)$ rises higher than the system's total energy E then the particle is "stuck" in a potential well. It moves back and forth between the **turning points** but it can't escape. This is called a **bound state**.

- On the other hand, if E exceeds $V(x)$ on one or both sides then the particle comes in from “infinity”, slows down or speeds up under the influence of the potential, and returns to infinity. This is called a **scattering state**.
- Some potentials admit only bound states (e.g. the harmonic oscillator), some admit only scattering states, and some admit both kinds depending on the energy of the particle.