University Physics with Modern Physics Electromagnetism Notes

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21.1 Electric Charge
\bullet Electrons have a much smaller mass than neutrons and protons
• Neutrons and protons have a very similar mass
• Electrons and protons have the same magnitude of charge
• The number of protons in an atom determins its atomic number
• If an electron is added to a neutral atom it becomes a negative ion , if

one is removed it becomes a **positive ion** — this is called **ionisation**

• The principle of conservation of charge states that the algebraic sum

• The electron or proton's magnitude of charge is a natural unit of charge — every observable amount of electric charge is an integer multiple of this

of all the electric charges in any closed system is constant

21.2 Conductors, Insulators, and Incuded Charges

- Conductors pemit easy movement of charge, insulators do not
- Holding a charged object near an uncharged object causes free electrons in the latter to move away/towards the former, resulting in a net charge on either side this is called **induced charge**

21.3 Coulomb's Law

- The SI unit of charge is called one **coulomb** (1 C) and is defined such that $1.602176634 \times 10^{-19}$ C is equal to the charge of an electron or proton
- Coulomb's law describes the electric force between two point charges

$$F = \frac{1}{4\pi\epsilon_0} \frac{|q_1 q_2|}{r^2}$$

where the **electric constant** $\epsilon_0 = 8.854 \times 10^{-12} \,\mathrm{C}^2/\mathrm{N} \cdot \mathrm{m}^2$, q_1 and q_2 are the magnitudes of the charges, and r is the distance between them

- The electric force is always directed along the line between the two charges, attracting opposite charges and repelling like charges
- $\frac{1}{4\pi\epsilon_0}$ can be approximated as $9.0\times10^9\,\mathrm{N\cdot m^2/C^2}$
- The principle of superposition of forces also applies to electric charges

21.4 Electric Field and Electric Forces

- The electric force on a charged object is exerted by the electric field created by other charged objects
- We can determine if there is an electric field at a point by placing a test charge q_0 there and seeing if it experiences an electric force the electric field at that point (the electric force per unit charge) is then given by

$$\mathbf{E} = \frac{\mathbf{F}}{q_0}$$

 \bullet Rearranging, the force experienced by a charge q_0 at a point is given by

$$\mathbf{F} = q_0 \mathbf{E}$$

• When considering an electric field produced by a point charge, the location of the point charge is called the **source point** and the location at which we're trying to determine the field is called the **field point**

• The electric field produced by a point charge is given by

$$\mathbf{E} = \frac{1}{4\pi\epsilon_0} \frac{q}{r^2} \hat{\mathbf{r}}$$

where q is the charge of the point charge, r is the distance between the source and field points, and $\hat{\mathbf{r}}$ is the unit vector from the source to the field point

- Unlike Coulomb's law this equation doesn't use the absolute value of q meaning that the electric fields of positive charges point away from the charge, while those of negative charges point towards them
- ullet In electrostatics, the electric field inside the material of a conductor (but not holes within the material) is ullet

21.5 Electric-Field Calculations

• The **principle of superposition of electric fields** states that the total electric field at a point *P* is the vector sum of the fields at *P* due to each point charge in the charge distribution

$$\mathbf{E} = \mathbf{E}_1 + \mathbf{E}_2 + \cdots$$

- For a line charge distribution the **linear charge density** is represented by λ (the charge per unit length, measured in C/m)
- For a surface charge distribution the surface charge density is represented by σ (the charge per unit area, measured in C/m²)
- For a volume charge distribution the **volume charge density** is represented by ρ (the charge per unit volume, measured in C/m³)
- \bullet The electric field of an infinitely long line charge along the y-axis is

$$E = \frac{\lambda}{2\pi\epsilon_0 r}$$