# University Physics with Modern Physics Electromagnetism Notes

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### 21 Electric Charge and Electric Field

#### 21.1 Electric Charge

- Electrons have a much smaller mass than neutrons and protons
- Neutrons and protons have a very similar mass
- Electrons and protons have the same magnitude of charge
- The number of protons in an atom determins its atomic number
- If an electron is added to a neutral atom it becomes a **negative ion**, if one is removed it becomes a **positive ion** this is called **ionisation**
- The **principle of conservation of charge** states that the algebraic sum of all the electric charges in any closed system is constant
- The electron or proton's magnitude of charge is a natural unit of charge every observable amount of electric charge is an integer multiple of this

#### 21.2 Conductors, Insulators, and Incuded Charges

- Conductors pemit easy movement of charge, insulators do not
- Holding a charged object near an uncharged object causes free electrons in the latter to move away/towards the former, resulting in a net charge on either side this is called **induced charge**

#### 21.3 Coulomb's Law

- The SI unit of charge is called one **coulomb** (1 C) and is defined such that  $1.602176634 \times 10^{-19}$  C is equal to the charge of an electron or proton
- Coulomb's law describes the electric force between two point charges

$$F = \frac{1}{4\pi\epsilon_0} \frac{|q_1 q_2|}{r^2}$$

where the **electric constant**  $\epsilon_0 = 8.854 \times 10^{-12} \,\mathrm{C}^2/\mathrm{N} \cdot \mathrm{m}^2$ ,  $q_1$  and  $q_2$  are the magnitudes of the charges, and r is the distance between them

- The electric force is always directed along the line between the two charges, attracting opposite charges and repelling like charges
- $\frac{1}{4\pi\epsilon_0}$  can be approximated as  $9.0 \times 10^9 \,\mathrm{N}\cdot\mathrm{m}^2/\mathrm{C}^2$
- The principle of superposition of forces also applies to electric charges

#### 21.4 Electric Field and Electric Forces

- The electric force on a charged object is exerted by the electric field created by other charged objects
- We can determine if there is an electric field at a point by placing a test charge  $q_0$  there and seeing if it experiences an electric force the electric field at that point (the electric force per unit charge) is then given by

$$\mathbf{E}=rac{\mathbf{F}}{q_0}$$

• Rearranging, the force experienced by a charge  $q_0$  at a point is given by

$$\mathbf{F} = q_0 \mathbf{E}$$

- When considering an electric field produced by a point charge, the location of the point charge is called the **source point** and the location at which we're trying to determine the field is called the **field point**
- The electric field produced by a point charge is given by

$$\mathbf{E} = \frac{1}{4\pi\epsilon_0} \frac{q}{r^2} \hat{\mathbf{r}}$$

where q is the charge of the point charge, r is the distance between the source and field points, and  $\hat{\mathbf{r}}$  is the unit vector from the source to the field point

- Unlike Coulomb's law this equation doesn't use the absolute value of q meaning that the electric fields of positive charges point away from the charge, while those of negative charges point towards them
- $\bullet$  In electrostatics, the electric field inside the material of a conductor (but not holes within the material) is  ${\bf 0}$

#### 21.5 Electric-Field Calculations

• The **principle of superposition of electric fields** states that the total electric field at a point *P* is the vector sum of the fields at *P* due to each point charge in the charge distribution

$$\mathbf{E} = \mathbf{E}_1 + \mathbf{E}_2 + \cdots$$

- For a line charge distribution the **linear charge density** is represented by  $\lambda$  (the charge per unit length, measured in C/m)
- For a surface charge distribution the **surface charge density** is represented by  $\sigma$  (the charge per unit area, measured in C/m<sup>2</sup>)
- For a volume charge distribution the **volume charge density** is represented by  $\rho$  (the charge per unit volume, measured in C/m<sup>3</sup>)
- The electric field of an infinitely long line charge along the y-axis is

$$E = \frac{\lambda}{2\pi\epsilon_0 r}$$

#### 21.6 Electric Field Lines

- An **electric field line** is a line drawn through space such that its tangent at any point is in the direction of the electric field vector at that point
- Fewer lines are drawn in areas where the electric field is weak and more lines are drawn in areas where it's strong

#### 21.7 Electric Dipoles

- ullet An **electric dipole** is a pair of point charges of equal magnitude q and opposite sign separated by a distance d
- The net force on an electric dipole in a uniform electric field is **0**
- The electric dipole moment  $\mathbf{p}$  of an electric dipole is a vector directed from the negative charge to the positive charge with magnitude qd
- The net torque on an electric dipole in a uniform electric field is  $\mathbf{p} \times \mathbf{E}$  or  $qEd\sin\phi$  where  $\phi$  is the angle between the electric dipole and the electric field
- The potential energy of an electric dipole in a uniform electric field is

$$U = -\mathbf{p} \cdot \mathbf{E}$$

#### 22 Gauss's Law

#### 22.1 Calculating Electric Flux

• The electric flux of a uniform electric field through a flat surface A is

$$\Phi_E = \mathbf{E} \cdot \mathbf{A}$$

where  $\mathbf{A}$  is normal to A and has a magnitude equal to its area

 $\bullet$  The electric flux of a nonuniform electric field through a curved surface A is

$$\Phi_E = \int \mathbf{E} \cdot \mathbf{dA}$$

#### 22.2 Gauss's Law

• Gauss's law states that the total electric flux through a closed surface is equal to the total electric charge enclosed by the surface divided by  $\epsilon_0$ 

$$\Phi_E = \oint \mathbf{E} \cdot \mathbf{dA} = \frac{Q_{\mathrm{enc}}}{\epsilon_0}$$

#### 22.3 Applications of Gauss's Law

- Gauss's law can be used in two ways:
  - If we know the charge distribution and it has enough symmetry to let us evaluate the integral in Gauss's law, we can find the field
  - If we know the field, we can use Gauss's law to find the charge distribution
- Under electrostatics, excess charge always lies of the surface of a conductor
- The electric field of an infinite line charge is

$$\mathbf{E} = \frac{1}{4\pi\epsilon_0} \frac{2\lambda}{r} \hat{\mathbf{r}}$$

#### 22.4 Charges on Conductors

- If there is excess charge at rest on a conductor, all of that charge must lie on the surface of the conductor and the electric field inside the conductor must be zero. If there is a cavity inside the conductor, the net charge on the cavity walls equals the amount of charge enclosed by the cavity
- Charges outside a conductor have no effect on the interior of the conductor, even if it has a cavity inside this is why Faraday cages work
- At the surface of a conductor, the component of the electric field that is perpendicular to the surface is

$$E_{\perp} = \frac{\sigma}{\epsilon_0}$$

#### 23 Electric Potential

#### 23.1 Electric Potential Energy

• The electric potential energy of two point charges is

$$U = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r}$$

• The electric potential energy of a point charge  $q_0$  and a collection of charges  $q_1, q_2$ , etc. is

$$U = \frac{q_0}{4\pi\epsilon_0} \left( \frac{q_1}{r_1} + \frac{q_2}{r_2} + \cdots \right) = \frac{q_0}{4\pi\epsilon_0} \sum_i \frac{q_i}{r_i}$$

- For every electric field due to a static charge distribution, the force exterted by that field is conservative
- The total electric potential energy of a collection of charges  $q_1, q_2$ , etc. is

$$U = \frac{1}{4\pi\epsilon_0} \sum_{i < j} \frac{q_i q_j}{r_{ij}}$$

where  $r_{ij}$  is the distance between  $q_i$  and  $q_j$ 

#### 23.2 Electric Potential

- Potential is potential energy per unit charge
- The unit of potential is the **volt**, equal to 1 joule per coulomb
- The potential difference between two points  $V_{ab} = V_a V_b$  is called the potential of a with respect to b and equals the amount of work done by the electric force when a unit (1 C) of charge moves from a to b
- The electric potential due to a point charge is

$$V = \frac{1}{4\pi\epsilon_0} \frac{q}{r}$$

• The electric potential due to a collection of point charges is

$$V = \frac{1}{4\pi\epsilon_0} \sum_i \frac{q_i}{r_i}$$

• The electric potential due to a continuous charge distribution is

$$V = \frac{1}{4\pi\epsilon_0} \int \frac{dq}{r}$$

• The electric potential difference between two points is given by

$$V_a - V_b = \int_a^b \mathbf{E} \cdot d\mathbf{l} = \int_a^b E \cos \phi \, dl$$

- Positive charges tend to "fall" from high- to low-potential regions while negative charges do the opposite
- When a particle with charge  $e = 1.602 \times 10^{-19} \,\mathrm{C}$  moves between two points with a potential difference of  $1 \,\mathrm{V} = 1 \,\mathrm{J/C}$  the change in energy is  $U_a U_b = q V_{ab} = (1.602 \times 10^{-19} \,\mathrm{C})(1 \,\mathrm{J/C}) = 1.602 \times 10^{-19} \,\mathrm{J}$  which is called 1 **electron volt**

#### 23.4 Equipotential Surfaces

- An **equipotential surface** is a three-dimensional surface on which the electric potential is the same at every point
- Because electric potential energy doesn't change as a test charge moves over an equipotential surface, the electric field can do no work and thus field lines and equipotential surfaces are always perpendicular
- When all charges are at rest, the surface of a conductor is an equipotential surface
- When all charges are at rest, the entire solid volume of a conductor is at the same potential

#### 23.5 Potential Gradient

• The relationship between  $\mathbf{E}$  and V is given by

$$\mathbf{E} = -\nabla V = -\left(\frac{\partial V}{\partial x}\hat{\mathbf{i}} + \frac{\partial V}{\partial y}\hat{\mathbf{j}} + \frac{\partial V}{\partial z}\hat{\mathbf{k}}\right)$$

• If E has a radial component  $E_r$  with respect to an axis or a point and r is the distance from that axis or point, then

$$E_r = -\frac{\partial V}{\partial r}$$

# 24 Capacitance and Dielectrics

#### 24.1 Capacitors and Capacitance

- Any two conductors separated by an insulator (or a vacuum) form a capacitor
- The **capacitance** of a capacitor measures its ability to store charge

$$C = \frac{Q}{V_{AB}}$$

 $\bullet$  Capacitance is measured in  ${\bf farads}$  where

$$1 \, \mathrm{F} = 1 \, \mathrm{C/V}$$

• The capacitance of a parallel plate capacitor in a vacuum is

$$C = \epsilon_0 \frac{A}{d}$$

#### 24.2 Capacitors in Series and Parallel

- In a series connection, the magnitude of charge on all plates is the same
- The **equivalent capacitance** of a combination of capacitors is the capacitance of a single capacitor that would have equivalent behaviour
- In a series connection, the reciprocal of the equivalent capacitance equals the sum of the reciprocals of the individual capacitances

$$\frac{1}{C_{\text{eq}}} = \frac{1}{C_1} + \frac{1}{C_2} + \cdots$$

meaning the equivalent capacitance is always less than any individual capacitance

- In a parallel connection, the potential difference is the same for all capacitors
- In a parallel connection, the equivalent capacitance equals the sum of the individual capacitances

$$C_{\text{eq}} = C_1 + C_2 + \cdots$$

meaning the equivalent capacitance is always greater than any individual capacitance

# 24.3 Energy Storage in Capacitors and Electric-Field Energy

• The potential energy stored in a capacitor is

$$U = \frac{Q^2}{2C} = \frac{1}{2}CV^2 = \frac{1}{2}QV$$

• The **energy density** of a parallel plate capacitor is its energy per unit volume

$$u = \frac{\frac{1}{2}CV^2}{Ad} = \frac{1}{2}\epsilon_0 E^2$$

#### 24.4 Dielectrics

- **Dielectrics** are nonconducting materials
- Most capacitors have a dielectric material between their plates because
  - 1. It preserves the distance between the plates
  - 2. It increases the maximum potential difference between the plates by avoiding **dielectric breakdown** when the material between the plates becomes ionized and becomes conductive this happens more easily for air

- 3. It increases the capacitance by decreasing the potential difference for a given charge
- The dielectric constant of a material is defined as

$$K = \frac{C}{C_0}$$

where  $C_0$  is the capacitance of a capacitor with vacuum between the plates and C is the capacitance of the same capacitor with the material between the plates

• If  $E_0$  is the magnitude of the electric field between the plates of a parallel plate capacitor when separated by a vacuum and E is the magnitude when separated by a dielectric then

$$E = \frac{E_0}{K}$$

• The electric field (and electric potential) are reduced because the dielectric becomes **polarized** and an induced surface charge appears of magnitude

$$\sigma_i = \sigma \left( 1 - \frac{1}{K} \right)$$

• The **permittivity** of a dielectric is defined as

$$\epsilon = K\epsilon_0$$

• The capacitance of a parallel plate capacitor with dielectric between the plates is thus

$$C = KC_0 = K\epsilon_0 \frac{A}{d} = \epsilon \frac{A}{d}$$

and the electric energy density is

$$u = \frac{1}{2}K\epsilon_0 E^2 = \frac{1}{2}\epsilon E^2$$

• The maximum electric-field magnitude that a material can withstand without the occurrence of breakdown is called its **dielectric strength** and is denoted  $E_m$ 

#### 24.5 Molecular Model of Induced Charge

• If a material is comprised of polar molecules where the net charge of the molecule is 0 but the charge isn't distributed equally, electric fields cause the molecules to rotate which induces a charge

- Even if a material isn't comprised of polar molecules, electric fields cause molecules' positive and negative charges to separate slightly resulting in a dipole which again experiences a torque
- The charges in conductors are free to move so they're known as **free charges** while the charges in dielectrics aren't so they're known as **bound charges**

#### 24.6 Gauss's Law in Dielectrics

• Gauss's Law in a dielectric material relates the flux of KE through the surface to the amount of free (not bound) charge enclosed by the surface

$$\oint K\mathbf{E} \cdot d\mathbf{A} = \frac{Q_{\text{encl-free}}}{\epsilon_0}$$

• This shows that filling a volume with a dielectric with relative permittivity K reduces the magnitude of the electric field by a factor of 1/K

# 25 Current, Resistance, and Electromotive Force

#### 25.1 Current

- A current is any motion of charge from one region to another
- $\bullet$  The **drift velocity**  $\mathbf{v_d}$  of a current is the velocity of its particles
- While a current may come about through the movement of negative and/or
  positive charges, conventional current dictates that by convention we
  describe currents as if they were carried by positive charges
- The unit of current is the ampere which is defined to be one coulomb per second

$$1 A = 1 C/s$$

- ullet The **charge concentration** n is the number of moving charged particles per unit volume
- The current through an area is given by

$$I = \frac{dQ}{dt} = n|q|v_dA$$

• The current density is the current per unit cross-sectional area

$$J = nqv_d$$

#### 25.2 Resistivity

• The **resistivity**  $\rho$  of a material is defined by **Ohm's law** 

$$\rho = \frac{E}{I}$$

- The unit of resistivity is ohm-meters  $(\Omega m)$
- The reciprocal of resistivity is **conductivity**
- Materials that obey Ohm's law are called **ohmic** or **linear** conductors
- Materials that don't obey Ohm's law are called **nonohmic** or **nonlinear** conductors
- The resistivity of a metallic conductor nearly always increases with increasing temperature

$$\rho(T) = \rho_0[1 + \alpha(T - T_0)]$$

where  $\rho_0$  is the resistivity at reference temperature  $T_0$  and  $\alpha$  is the **temperature coefficient of resistivity** 

- The resistivity of semiconductors decreases with increasing temperature
- Some materials exhibit **superconductivity** where their resistivity drops to 0 below a critical temperature

#### 25.3 Resistance

• The ratio of the voltage and current in a conductor is called its **resistance** 

$$R = \frac{V}{I} = \frac{\rho L}{A}$$

where  $\rho$  is the resistivity of the conductor, L is its length, and A is its cross-sectional area

- If  $\rho$  is constant (as in ohmic materials), then R is also constant
- The unit of resistance is the ohm

$$1\Omega = 1V/A$$

• Because the resistivity of a material varies with temperature, so too does the resistance of a specific conductor

$$R(T) = R_0[1 + \alpha(T - T_0)]$$

• A device made to have a specific resistance is called a **resistor** 

#### 25.4 Electromotive Force and Circuits

- When a charge goes around a complete circuit and returns to its starting position its electric potential energy must be the same, but it experienced losses due to resistance along the way
- Electromotive force or emf  $\mathcal{E}$  is the influence that makes current flow from lower to higher potential in a circuit and restores its original potential energy
- A device that provides emf is called a source of emf
- The SI unit of emf is the volt
- In an ideal source of emf
  - The potential difference between its terminals is constant regardless of the current passing through it
  - $-\mathcal{E} = V = IR$
- Real sources of emf have internal resistance r that reduce the terminal voltage

$$V_{ab} = \mathcal{E} - Ir$$

• Real sources of emf can be modelled as an ideal source of emf  $\mathcal E$  in series with a resistor r

#### 25.5 Energy and Power in Electric Circuits

• Power is the time rate change of energy transfer

$$P = VI$$

where V is the voltage across a circuit element and I is the current in it

• The SI unit of power is the watt

$$1 \, \text{W} = 1 \, \text{J/s}$$

• If the circuit element is a resistor then V = IR and

$$P=VI=I^2R=\frac{V^2}{R}$$

• If the circuit element is a source of emf outputting power then

$$P = VI = (\mathcal{E} - Ir)I = \mathcal{E}I - I^2r$$

where the  $\mathcal{E}I$  term is the power generated by the element and the  $I^2r$  term is the power dissipated by its internal resistance

• If the circuit element is a source of emf consuming power (charging) then

$$P = VI = (\mathcal{E} + Ir)I = \mathcal{E}I + I^2r$$

where the terms are the same as above

## 25.6 Theory of Metallic Conduction

- $\bullet$  The average time between collisions of an electron and positive ions is called the **mean free time**  $\tau$
- The resistivity of a metal can be approximated as

$$\rho = \frac{m}{ne^2\tau}$$

where m is the mass of an electron, n is the number of free electrons per unit volume, e is the charge of an electron, and  $\tau$  is the mean free time