"Historically" informed reconstructions of 19th century chrome yellow pigments

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Abstract

This report deals with the fabrication of 19th century historically informed chrome yellow pigments by Vanessa Otero and Gauthier Patin in April 2019. In the time frame of five days, about 44 g of pigments have been made. The majority of the pigments will be used to paint a reconstructed version of the Sunflowers painting from Vincent Van Gogh while the rest of the pigments will be faded in the framework of model paint-outs colour changes experiments.

Pigment Lemon chrome yellow L2b : PbCr_{1-x}S_xO₄ with x = 0.5

Description Mixed crystals of lead chromate and lead sulfate. Pigment found in the Sunflowers painting of Van Gogh

Recipe Winsor & Newton, recipe originally called *Super Lemon Chrome*Recipe transcribed and tested during the PhD Thesis of Vanessa Otero (2018)

Date From 08/04/2019 to 12/04/2019

Location Department of Conservation and Restoration, University NOVA of Lisbon, Campus da Caparica, Portugal

Participants Pigment making: Vanessa Otero¹ assisted by Gauthier Patin² Pigment selection: Vanessa Otero, Muriel Geldof³, Ella Hendriks⁴

Purpose The pigment will be implemented in three different projects:

- 1. The majority of the pigment will be used by the artist Charlotte Caspers in order to paint a reconstruction of the Sunflowers painting by Van Gogh. This reconstruction has been ordered by the Van Gogh Museum in the frame of the 2019 summer exhibition on the sunflowers.
- 2. A small part of the pigment will used by Gauthier Patin for his PhD experiments on colour changes in Van Gogh artworks.
- 3. Another small amount will be given to Laurens van Giersbergen for his Master thesis experiments on the comparison of different light aging devices.

Results Three batches of pigment have been made (cf. Illustration 6):

1. On the 09/04/2019: 4.4563 g

2. On the 11/04/2019: 21.0083 g

3. On the 12/04/2019: 18.7489 g

The total amount of pigment is 44.2135 g. A tiny quantity of each batch has been stored in small plastic container and given to Vanessa for further analysis if needed. More information about each batch is given below.

Safety Since chromium (VI) oxide is extremely toxic, each step of the recipe has been performed under a fume-hood with blouse, gloves, and glasses. We recommend to read the International Chemical Safety Card on chromium (VI) oxide before making such pigment.

Ingredients

- Water (H₂0)
- Potassium dichromate (K₂Cr₂O₇)
- Sodium carbonate (Na₂CO₃)
- Barium sulfate (BaSO₄)
- Nitric acid (HNO₃)
- Lead (II) oxide (PbO)
- Sulfuric acid (H₂SO₄)
- Sodium sulfate (Na₂SO₄)

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Lemon Chrome L2b - Batch N°1.

Synthesis

56 ml of H_2O + 2.120 g of Na_2SO_4 + 1.325 g of $K_2Cr_2O_7$ + 1.113 g of Na_2CO_3 (boil and stir for 30 min ; add water to fill up to 13ml) + 13 ml of $Pb(NO_3)_2^\#$ + 0.308 ml of H_2SO_4 in 1 ml of H_2O + stir for 30 min (cf. Illustration 3)

washing the pigment with milliQ water and then with ethanol and finally with ethylic ether (cf. Illustration 4)

Quantity of pigment obtained: 4.4563 g

pH measurements

Ingredients	pH value
$H_2O + Na_2SO_4$	5.33
$H_2O + Na_2SO_4 + K_2Cr_2O_7 + Na_2CO_3$	10.13
Final solution	2.30

 $^{^{\#}}$ 150 ml of boiling H₂O + 32.3 ml of HNO₃ + 53.5 g of PbO (cf. Illustration 1)

Lemon Chrome L2b - Batch N°2.

Synthesis

 $\begin{array}{l} 250 \text{ ml of } H_2O \\ + \\ 9.41 \text{ g of } Na_2SO_4 \\ + \\ 60 \text{ ml of boiling } H_2O + 5.88 \text{ g of } K_2Cr_2O_7 + 4.94 \text{ g of } Na_2CO_3 \text{ (boil and stir for } 30 \text{ min ; add water to fill up to } 60\text{ml)} \\ + \\ 57 \text{ ml of } Pb(NO_3)_2^\# \\ + \\ 1.7 \text{ ml of } H_2SO_4 \text{ in } 2 \text{ ml of } H_2O \\ + \\ \text{stir for } 30 \text{ min (cf. Illustration } 3) \\ + \end{array}$

washing the pigment with milliQ water and then with ethanol and finally with ethylic ether (cf. Illustration 4)

Quantity of pigment obtained: 21.0083 g

pH measurements

Ingredients	pH value
$H_2O + Na_2SO_4$	6.01
$H_2O + Na_2SO_4 + K_2Cr_2O_7 + Na_2CO_3$	10.30
$H_2O + H_2SO_4$	2.02
Final solution	2.52

 $^{^{\#}}$ 150 ml of boiling H₂O + 32.3 ml of HNO₃ + 53.5 g of PbO (cf. Illustration 1)

Lemon Chrome L2b - Batch N°3.

Synthesis

 $\begin{array}{l} 250 \text{ ml of } H_2O \\ + \\ 9.41 \text{ g of } Na_2SO_4 \\ + \\ 60 \text{ ml of boiling } H_2O + 5.88 \text{ g of } K_2Cr_2O_7 + 4.94 \text{ g of } Na_2CO_3 \text{ (boil and stir for 30 min ; add water to fill up to 60ml)} \\ + \\ 57 \text{ ml of } Pb(NO_3)_2^\# \\ + \\ 1.7 \text{ ml of } H_2SO_4 \text{ in 2 ml of } H_2O \\ + \\ \text{stir for 30 min (cf. Illustration 3)} \end{array}$

Quantity of pigment obtained: 18.7489 g

pH measurements

Ingredients	pH value
$H_2O + Na_2SO_4$	5.90
$H_2O + Na_2SO_4 + K_2Cr_2O_7 + Na_2CO_3$	10.16
Final solution	2.29

washing the pigment with milliQ water and then with ethanol and finally with ethylic ether (cf. Illustration 4) $\frac{1}{2}$

 $[\]overline{}^{*}$ 150 ml of boiling H₂O + 32.3 ml of HNO₃ + 53.5 g of PbO (cf. Illustration 1)

Photos



Illustration 1: Lead nitrate solution



Illustration 3: Final solution (before filtering and washing)



Illustration 5: Pigment drying



Illustration 2: Chromate solution



Illustration 4: Washing of the pigment



Illustration 6: The three batches