Physically-Plausible Parameters

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1 Introduction

This note illustrates an approach to fitting the parameters of a bond graph model to experimental data. Insofar as the parameters are associated with a bond graph, they are *physically-plausible* Gawthrop et al. (2020).

The approach uses a bond-graph derived from a stoichiometric model of *e.coli* Orth et al. (2010) (using a method described elsewhere Gawthrop (2020)) combined with experimental values of *reaction potential, reaction flux* and *species concentration* from the literature Park et al. (2016).

1.1 Setup modules

```
[1]: ## Paths
NeedPath=True
if NeedPath:
    import sys
    sys.path += ['/usr/lib/python3/dist-packages']
```

```
[2]: ## Maths library
     import numpy as np
     import scipy
     ## BG tools
     import BondGraphTools as bgt
     ## SVG bond graph
     import svgBondGraph as sbg
     ## BG stoichiometric utilities
     import stoich as st
     ## Modular bond graphs
     import modularBondGraph as mbg
     ## Stoichiometric conversion
     import CobraExtract as Extract
     import stoichBondGraph as stbg
     ## Potentials
     import phiData
     ## Faraday constant
     import scipy.constants as con
     F = con.physical_constants['Faraday constant'][0]
     ## Display
     import IPython.display as disp
     ## Plotting
     import matplotlib.pyplot as plt
     import copy
     ## Allow output from within functions
     from IPython.core.interactiveshell import InteractiveShell
     InteractiveShell.ast_node_interactivity = "all"
```

```
import importlib as imp
quiet = True
showMu=True
```

1.2 Quadratic programming QP.

minimise
$$\frac{1}{2}x^TPx + q^Tx$$
 (1)

subject to
$$Gx \le h$$
 (2)

and
$$Ax = b$$
 (3)

In the case considered here, there is no equality constraint and

$$x = \hat{\phi} \tag{4}$$

$$P = NN^T + \mu I_{n_X \times n_X} \tag{5}$$

$$q = (N\Phi)^T \tag{6}$$

$$G = N^T (7)$$

$$h = -\Phi_{min} \tag{8}$$

 $\mu > 0$ is required to give a convex QP: in essence it turns a non-unique solution for ϕ into a minimum norm solution.

```
[3]: ## Quadratic programming stuff.
     import quadprog
     \textit{\#\# Function from https://scaron.info/blog/quadratic-programming-in-python.html}
     def quadprog_solve_qp(P, q, G=None, h=None, A=None, b=None):
         qp_G = .5 * (P + P.T) # make sure P is symmetric
         qp_a = -q
         if A is not None:
             qp_C = -np.vstack([A, G]).T
             qp_b = -np.hstack([b, h])
             meq = A.shape[0]
         else: # no equality constraint
             if G is None:
                 qp_C = None
                 qp_b = None
             else:
                 qp_C = -G.T
                 qp_b = -h
             meq = 0
         return quadprog.solve_qp(qp_G, qp_a, qp_C, qp_b, meq)[0]
     ## Function to compute phi from Phi subject to Phi>positive number
     ## NN Reduced N corresponding to known Phi
     def quadsolve_phi(N0,N1,Phi0,Phi_min=0.0,mu=1e-10):
```

```
(n_X,n_V) = N1.shape
P = 1.0*N0@(N0.T) + mu*np.eye(n_X)
q = (N0@Phi0).T
G = 1.0*N1.T
h = -Phi_min*np.ones((n_V))
phi = quadprog_solve_qp(P, q, G=G, h=h)
return phi
```

2 Conversion factor

```
[4]: Factor = st.F()/1e6
print(f'To convert from kJ/mol to mV, divide by {1/Factor:4.3}')
```

To convert from kJ/mol to mV, divide by 10.4

3 Extract Model

This example uses the Glycolysis and Pentose Phosphate pathways.

[5]: sm = Extract.extract(cobraname='textbook', Remove=['_C','__'],

Notes:

- Reactions RPI, PGK and PGM are reversed to correspond to positive flows.
- The resultant stoichiometric matrix N relates reaction flows (f) to species flows (\dot{x}):

$$\dot{x} = Nf \tag{9}$$

3.1 Extract stoichiometry

```
negReaction=['RPI', 'PGK', 'PGM'], quiet=quiet)

Extracting stoichiometric matrix from: textbook
Cobra Model name: e_coli_core BondGraphTools name: e_coli_core_abg
Extract.Integer only handles one non-integer per reaction
Multiplying reaction BIOMASS_ECOLIORE ( 12 ) by 0.6684491978609626 to avoid non-integer species 3PG ( 2 )
Multiplying reaction CYTBD ( 15 ) by 2.0 to avoid non-integer species 02 ( 55 )
Multiplying reaction PGK ( 54 ) by -1
Multiplying reaction PGM ( 56 ) by -1
Multiplying reaction RPI ( 65 ) by -1
```

```
[6]: name = 'GlyPPP_abg'
reaction = []

## Glycolysis
reaction += ['PGI','PFK','FBA','TPI']

## Pentose Phosphate
reaction += ['G6PDH2R','PGL','GND','RPI','TKT2','TALA','TKT1','RPE']
```

```
ss = Extract.choose(sm,reaction=reaction)

## Create BG
ss['name'] = name
stbg.model(ss)
import GlyPPP_abg
imp.reload(GlyPPP_abg)
s = st.stoich(GlyPPP_abg.model(),quiet=quiet)
```

```
[7]: ## Set up chemostats
    chemostats = ['ADP','ATP','H','H2O','NADP','NADPH','CO2']
    chemostats += ['G6P','G3P','R5P']
    #chemostats += ['G6P','R5P']
    chemostats.sort()
    print(chemostats)
    sc = st.statify(s,chemostats=chemostats)

sp = st.path(s,sc,pathname='PPP')
    print(st.sprintp(sc))
    disp.Latex(st.sprintrl(sp,chemformula=True))
```

['ADP', 'ATP', 'CO2', 'G3P', 'G6P', 'H', 'H2O', 'NADP', 'NADPH', 'R5P']
3 pathways
0: + PGI + PFK + FBA + TPI
1: + G6PDH2R + PGL + GND + RPI
2: - 2 PGI + 2 G6PDH2R + 2 PGL + 2 GND + TKT2 + TALA + TKT1 + 2 RPE

[7]:

$$ATP + G_6P \xrightarrow{PPP_1} ADP + 2G_3P + H \tag{10}$$

$$G_6P + H_2O + 2 \text{ NADP} \xrightarrow{PPP_2} CO_2 + 2 H + 2 \text{ NADPH} + R_5P$$
 (11)

$$2H_2O + 4NADP + R_5P \xrightarrow{PPP_3} 2CO_2 + G_3P + 4H + 4NADPH$$
 (12)

```
[8]: print(st.sprintl(sc,'K',transpose=True))
disp.Latex(st.sprintl(sc,'K',transpose=True))
```

[8]:

3.2 Extract reaction potentials Φ and deduce plausible species potentials ϕ .

Because of the energetic constaints implied by the bond graph, the reaction potentials Φ are related to the species potentials ϕ by

$$\Phi = -N^T \phi \tag{14}$$

Typically, there are more species than reactions and so N has more rows than columns. Given the reaction potentials Φ , the species potentials can be estimated using the *pseudo inverse* N^{\dagger} of $-N^{T}$:

$$\hat{\phi} = N^{\dagger} \Phi \tag{15}$$

Notes:

• In general $\hat{\phi} \neq \phi$ but is physically plausible insofar as $-N^T \hat{\phi} = \Phi$.

```
[9]: def getPhi(s,Phi_hyd=0.5,phi_6PGL=None,quadprog=False):
         """Extract phi for given system using
         Reaction potentials from ParRubXu16"""
         ## Reaction potentials from ParRubXu16
         PHI = phiData.Phi_ParRubXu16_Measured()
           Phenotype = 'Mammalian'
          Phenotype = 'Yeast'
         Phenotype = 'Ecoli'
         Phi_reac = PHI[Phenotype]
         Phi = np.zeros((len(s['reaction']),1))
         N = copy.copy(s['N'])
         N_0 = None
         N_1 = None
         Phi_0 = []
         for j,reac in enumerate(s['reaction']):
             if (reac in Phi_reac.keys()) and not np.isnan(Phi_reac[reac]):
                 Phi_0.append(Phi_reac[reac])
                 if N_O is None:
                     N_0 = N[:,j]
                     N_0 = np.vstack((N_0,N[:,j]))
             else:
                 if N_1 is None:
                     N_1 = N[:,j]
                 else:
                     N_1 = np.vstack((N_1,N[:,j]))
         Phi_0 = np.array(Phi_0)
         #print(N_1)
```

```
N_O = N_O.T
          N_1 = N_1.T
          n_X, n_V = N_0.shape
          print(f'Extracting {n_X} values of phi from {n_V} values of Phi')
          if quadprog:
              phi = quadsolve_phi(N_0,N_1,Phi_0,Phi_min=1e-3,mu=1e-10)
              ## Compute Phi using pseudo inverse
              pinvNT = scipy.linalg.pinv(N_0.T)
              phi = -pinvNT@Phi_0
          if phi_6PGL is not None:
              ## Reset 6PGL
              i_6PGL = s['species'].index('6PGL')
              phi[i_6PGL] = phi_6PGL
              print (f'Resetting phi_6GPL to {int(1e3*phi[i_6PGL])} mV' )
          ## Sanity check
          Phi_new = -N_0.T_{ophi}
          err = np.linalg.norm(Phi_new-Phi_0)
          print(f'Phi error = {int(err*1000)}mV\n')
          Phi = -N.T_{ophi}
          return Phi,phi,Phi_0,Phi_reac
[10]: Phi_,phi_est_,Phi_0_,Phi_reac_ = getPhi(s,quadprog=False)
      print('Minimum Phi = ', int(round(np.min(1e3*Phi_))), 'mV')
     Extracting 19 values of phi from 10 values of Phi
     Phi error = OmV
     Minimum Phi = -3 mV
[11]: Phi,phi_est,Phi_0,Phi_reac = getPhi(s,quadprog=True)
      print('Minimum Phi = ', int(round(np.min(1e3*Phi))), 'mV')
      print('\nChange in phi')
      for i,spec in enumerate(s['species']):
          change = int(1e3*(phi_est[i]-phi_est_[i]))
          if not change==0:
              print(f'{i} {spec}\t {change}')
      print('\nChange in Phi')
      for i,reac in enumerate(s['reaction']):
          change = int(round(1e3*(Phi[i]-Phi_[i])))
```

Compute Phi

Extracting 19 values of phi from 10 values of Phi Phi error = 0mV

Minimum Phi = O mV

Change in phi 1 6PGL 1 12 H2O 1

Change in Phi 5 PGL 4 1 -3

3.3 Extracted reactions and reaction potentials

```
[12]: disp.Latex(st.sprintrl(s,chemformula=True,Phi=Phi,units=['mV','kJ']

→,showMu=showMu))
```

[12]:

$$G_{6}P \stackrel{PGI}{\longleftarrow} F_{6}P \qquad (16 \text{ mV}) [-1.59 \text{ kJ mol}^{-1}]$$

$$ATP + F_{6}P \stackrel{PFK}{\longleftarrow} ADP + FDP + H \qquad (68 \text{ mV}) [-6.64 \text{ kJ mol}^{-1}]$$

$$FDP \stackrel{FBA}{\longleftarrow} DHAP + G_{3}P \qquad (20 \text{ mV}) [-1.93 \text{ kJ mol}^{-1}]$$

$$DHAP \stackrel{TPI}{\longleftarrow} G_{3}P \qquad (7 \text{ mV}) [-0.77 \text{ kJ mol}^{-1}]$$

$$G_{6}P + NADP \stackrel{G_{6}PDH_{2}R}{\longleftarrow} {}_{6}PGL + H + NADPH \qquad (82 \text{ mV}) [-7.99 \text{ kJ mol}^{-1}]$$

$${}_{6}PGL + H_{2}O \stackrel{GND}{\longleftarrow} {}_{6}PGC + H \qquad (1 \text{ mV}) [-0.10 \text{ kJ mol}^{-1}]$$

$${}_{6}PGC + NADP \stackrel{GND}{\longleftarrow} CO_{2} + NADPH + RU_{5}PD \qquad (114 \text{ mV}) [-11.05 \text{ kJ mol}^{-1}]$$

$$RU_{5}PD \stackrel{RPI}{\longleftarrow} R_{5}P \qquad (0 \text{ mV}) [-0.00 \text{ kJ mol}^{-1}]$$

$$E_{4}P + XU_{5}PD \stackrel{TKT_{2}}{\longleftarrow} F_{6}P + G_{3}P \qquad (16 \text{ mV}) [-1.58 \text{ kJ mol}^{-1}]$$

$$G_{3}P + S_{7}P \stackrel{TALA}{\longleftarrow} E_{4}P + F_{6}P \qquad (54 \text{ mV}) [-5.25 \text{ kJ mol}^{-1}]$$

$$R_5P + XU_5PD \xrightarrow{TKT_1} G_3P + S_7P \qquad (4 \text{ mV}) [-0.39 \text{ kJ mol}^{-1}]$$

$$RU_5PD \xrightarrow{RPE} XU_5PD \qquad (0 \text{ mV}) [-0.08 \text{ kJ mol}^{-1}]$$

4 Deduce Pathway Flows

From basic stoichiometric analysis, steady-state flows can be written as:

$$f = K_p f_p \tag{16}$$

where
$$K_p N^{cd} = 0$$
 (17)

Note that the *pathway matrix* K_p is dependent on the choice of chemostats.

Given a set of experimental flows f, an estimate \hat{f}_p of f_p can be obtained from the *least-squares* formula:

$$(K_p^T K_p) \hat{f}_p = K_p^T f \tag{18}$$

Notes:

- v_p is a n_p vector containg the pathways flows
- $(K_p^T K_p)$ is a square $n_p \times n_p$ matrix where n_p is the number of pathways
- If some flows are not measured, the corresponding rows of K_p are deleted
- the reaction flows (including the missing ones) can be estimated from $\hat{f} = K_p \hat{f}_p$.
- the estimated chemostat flows are given by the non-zero elements of

$$\hat{\dot{x}} = N\hat{f} \tag{19}$$

```
[13]: def PathwayFlux(K, reaction, Reaction, flux):
          #KK = st.singleRemove(K)
          KK = K
          Kp = None
          Flux = {}
          reac_known = []
          #flux = phiData.ParRubXu16_flux()
          for i,reac in enumerate(reaction):
              if reac in flux.keys():
                  reac_known.append(reac)
                  fi = flux[reac]
                  \#Ki = np.abs(KK[i,:])
                  Ki = KK[i,:]
                  #print(reac,Ki)
                  if Kp is None:
                       Kp = Ki
                       f = fi
                  else:
                       Kp = np.vstack((Kp,Ki))
                       f = np.vstack((f,fi))
          #print(Kp)
          if Kp is not None:
              #print(f)
              f_p = np.linalg.solve(Kp.T@Kp,Kp.T@f)
              for i, Reac in enumerate (Reaction):
                  Flux[Reac] = f_p[i][0]
              #print(f_p)
              f_est = Kp@f_p
              #print(Kp@f_p-f)
          error = np.linalg.norm(f_est-f)/len(f)
          print(f'Flux error = {error:.2e}')
          return Flux,f_p,f_est,f,reac_known
```

5 Reaction constants (modified mass-action) and Michaelis-Menten

The modified mass-action formula is Gawthrop et al. (2020):

$$f = \kappa \left(\exp \frac{\Phi^f}{\alpha V_N} - \exp \frac{\Phi^r}{\alpha V_N} \right) \tag{20}$$

Thus an estimate for κ can be computed as:

$$\hat{\kappa} = \frac{\hat{f}}{f_0} \tag{21}$$

where
$$f_0 = \left(\exp\frac{\Phi^f}{\alpha V_N} - \exp\frac{\Phi^r}{\alpha V_N}\right)$$
 (22)

```
[14]: def reactionConstant(s,phi_est,f_est,alpha=1,K_E=100,K_C=1,rho=0.9):
          V_N = st.V_N()
          ## Extract stoichiometry
          N = s['N']
          Nf = s['Nf']
          Nr = s['Nr']
          reaction = s['reaction']
          ## Compute Phis from estimated phi
          Phi_ = -N.T@phi_est
          Phi_f = Nf.T@phi_est
          Phi_r = Nr.T@phi_est
           ## Compute normalised flow rates
          f_plus = np.exp(Phi_f/(alpha*V_N))
          f_minus = np.exp(Phi_r/(alpha*V_N))
          f0 = f_plus - f_minus
          parameter = {}
          MMparameter = {}
          for i,react in enumerate(reaction):
               MMpar = {}
               kap = f_est[i][0]/f0[i]
               parameter[f'kappa_{react}'] = kap
               \#print(f'\{react\}: \tPhi = \{int(Phi_[i]*1000)\}mV, \tf_est = \tulebox{$\sqcup$}
       \rightarrow {f_est[i][0]:.2e}, \tkappa = {kap:.2}')
               ## MM version
               X_{data} = np.array([1,f_{est}[i][0],-f_{est}[i][0]/f0[i]])
              Y_data = f_est[i][0]*f_plus[i]/f0[i]
                print(X_data)
      #
                 print(Y_data)
```

```
XTX = np.outer(X_data, X_data)
                                 XTy = X_data*Y_data
                                        print('XTX:', XTX)
                                         print('XTy:', XTy)
                                  ## Pseuso inverse eapproach
                                           theta = np.linalq.pinv(XTX)@XTy
                                          print(theta)
                                 ## QP approach
                                           f_{max_est} = 10
                                 rho_est = rho
                                k_v_{est} = K_C/K_E
                                 G = -np.eye(3)
                                h = -0*np.ones(3)
                                 A_{eq} = np.array([[0,1,0],[0,0,1]])
                                 b_eq = np.array([rho_est,k_v_est])
                                         A_{eq} = np.array([[1,0,0],[0,1,0]])
                                         b_eq = np.array([f_max_est, rho_est])
                                 theta = quadprog_solve_qp(XTX+1e-10*np.eye(3),-XTy,G,h,A_eq,b_eq)
                                 f_{max} = theta[0]
                                 rho = theta[1]
                                k_v = theta[2]
                                kappa = f_max/k_v
                                 MMpar['f_max'] = theta[0]
                                MMpar['rho'] = theta[1]
                                MMpar['k_v'] = theta[2]
                                         print(f'\{react\}: kappa=\{kappa:6.2f\} f_max=\{f_max:.2\} rho=\{rho:0.2f\}_{\sqcup} rho=\{rho:0.2f\}_
\hookrightarrow k_v = \{k_v : 6.2f\}'
                                 MMparameter[react] = MMpar
             return parameter, MMparameter
```

```
[15]: ## Convert to BG parameters
def MMtoBG(MMpar,K_E=100):
    kappa = np.zeros(2)
    K_CE = np.zeros(2)
    rho = MMpar['rho']
    K_C = K_E*MMpar['k_v']

# print(MMpar)
    kappa_bar = MMpar['f_max']/K_C
    kappa[0] = kappa_bar/rho
    kappa[1] = kappa_bar/(1-rho)
```

```
K_CE[0] = K_C
K_CE[1] = K_E
# print(kappa_1, kappa_2, K_E)
return kappa, K_CE
```

5.1 Normalise data

```
[16]: imp.reload(phiData)
    ## Extract experimetal data
    Concentration = phiData.ParRubXu16_conc() # M
    concentration = Concentration['Ecoli']

Flux = phiData.ParRubXu16_flux() # mM/min
    flux = Flux['Ecoli']

c_0 = concentration['G6P']
    f_0 = flux['PGI']/60
    t_0 = (1e3*c_0)/f_0

print(f'c_0 = {c_0*1000} mM, f_0 = {f_0} mM/sec, t_0 = {t_0} sec')
```

5.2 Show computed reaction flows

```
[17]: K = sc['K']
n_path = K.shape[1]
Reaction = []
for i in range(n_path):
    Reaction += [f'PPP{i+1}']
    print(Reaction)

for reac in flux.keys():
    flux[reac] *= 1/f_0

fluxp,f_p,f_est,f,reaction = PathwayFlux(sc['K'],s['reaction'],Reaction,flux)

## Assumed values:
K_E = 10
K_C = 1
rho = 0.2
```

```
## Reaction constants
f_est = sc['K']@f_p
parameter, MMparameter =
 →reactionConstant(s,phi_est,f_est,K_E=K_E,K_C=K_C,rho=rho)
K_C=K_C
#f_est = sc['K']@f_p
j=0
print('\n\n\% LaTeX table')
print('\\hline')
print('\\hline')
for i,reac in enumerate(s['reaction']):
    ## BG MM equivalent
    MMpar = MMparameter[reac]
    kappa_MM,K_CE = MMtoBG(MMpar,K_E=100.0)
    if reac in flux.keys():
       ff = f'\{f[j][0]:0.2f\}'
       j += 1
    else:
       ff = '--'
    if reac in Phi_reac.keys():
       PP = f'{1e3*Phi_reac[reac]:.2f}'
    else:
       PP = '--'
    kappa = 'kappa_'+reac'
    print(
       f'{reac} &\t {PP} &\t {1e3*Phi[i]:.2f} &\t {ff} & {f_est[i][0]:0.2f} \
       & {parameter[kappa]:.2f} & {kappa_MM[0]:.2f} & {kappa_MM[1]:.2f} \\\\
print('\\hline')
['PPP1']
['PPP1', 'PPP2']
['PPP1', 'PPP2', 'PPP3']
Flux error = 1.86e-01
%% LaTeX table
\hline
               $\Phi$~mV &
                             $\hat{\Phi}$~mV &
                                                   $f$ & $\hat{f}$ &
Reaction &
$\hat{\kappa}$ & $\hat{\kappa_1}$ & $\hat{\kappa_2}$\\
\hline
PGI &
        16.48 &
                      16.48 &
                                    60.00 & 59.52
                                                        & 154.39 & 66.44
& 16.61 \\
PFK &
       68.82 &
                      68.82 &
                                    62.62 & 63.12
                                                        & 54.85 & 30.59 &
7.65 \\
                                                        & 160.08 & 61.59
FBA &
      20.00 &
                      20.00 &
                                    63.43 & 63.12
```

```
& 15.40 \\
        7.98 & 7.98 & 62.82 & 63.12
                                          & 353.93 & 133.64 & 33.41 \\
G6PDH2R &
               -- &
                      82.84 &
                                     -- & 11.58
                                                      & 4.67 & 5.14 & 1.28
//
PGL &
        -- & 1.00 & -- & 11.58
                                        & 291.64 & 171.06 & 42.77 \\
GND &
        114.53 &
                      114.53 &
                                     11.70 & 11.58
                                                         & 1.27 & 4.78 &
1.19 \\
        0.04 & 0.04 & 7.87 & 7.98
                                         & 4206.98 & 2785.35 & 696.34 \\
RPI &
TKT2 & 16.38 &
                 16.38 &
                                     0.91 & 1.80
                                                  & 9.17 & 2.24 &
0.56 \\
TALA &
       54.41 &
                     54.41 &
                                     -- & 1.80
                                                & 1.66 & 0.94 & 0.23
//
TKT1 &
       4.04 & 4.04 & 2.92 & 1.80
                                       & 8.82 & 6.66 & 1.67 \\
RPE & 0.83 & 0.83 & 3.83 & 3.59
                                        & 96.07 & 63.27 & 15.82 \\
\hline
```

5.3 Show computed chemostat flows

```
[18]: dx_est = s['N']@f_est

print('\n\n% LaTeX table')
print('\hline')
print('Chemostat &\t flow \\\')
print('\hline')
for i,spec in enumerate(s['species']):
    if spec in chemostats:
        print(f'{spec} &\t {dx_est[i][0]:0.2f} \\\')
print('\hline')
```

```
%% LaTeX table
\hline
Chemostat & flow \\
\hline
ADP &
        63.12 \\
ATP &
        -63.12 \\
CO2 &
       11.58 \\
G3P &
       128.03 \\
G6P &
       -71.10 \\
H &
        86.27 \\
H20 &
       -11.58 \\
NADP &
        -23.16 \\
NADPH & 23.16 \\
R5P &
        6.19 \\
\hline
```

5.4 Show pathway flows

```
[19]: print('\n\n\% LaTeX table')
    print('\hline')
    print('Pathway &\t $\hat{f}_p$ \\\')
    print('\hline')
    for reac in fluxp.keys():
        print(f'{reac} &\t {fluxp[reac]:0.2f} \\\')
    print('\hline')

%% LaTeX table
\hline
```

```
\hline
Pathway & $\hat{f}_p$ \\
\hline
PPP1 & 63.12 \\
PPP2 & 7.98 \\
PPP3 & 1.80 \\
\hline
```

6 Species constants

$$K = \frac{\exp \phi}{x^{\circ}} = \frac{\exp \phi}{Vc^{\circ}} \tag{23}$$

```
[45]: #imp.reload(phiData)
     print('\n\n\% LaTeX table')
     print('\\hline')
     print('\\hline')
     #concentration['H'] = 1e-7
     ## Data in mM
     scale = 1e3
     K_spec = np.ones(s['n_X'])
     conc = np.ones(s['n_X'])
     c_G6P = concentration['G6P']
     #print('c_G6P',c_G6P)
     for i,spec in enumerate(s['species']):
         if spec in concentration.keys():
             conc[i] = concentration[spec]/c_G6P
             K_spec[i] = np.exp(phi_est[i]/st.V_N())/conc[i]
            print(f'{spec} & {int(round(1e3*phi_est[i]))} & \t{conc[i]:.4f} &_\(\subseteq\)
      \rightarrow{K_spec[i]:.4f} \\\')
         else:
             K_spec[i] = np.exp(phi_est[i]/st.V_N())
              print(f'{spec} &{phi_est[i]:.2} & -- & --\\\')
```

```
print('\\hline')

#print(conc)
print(s['species'])
```

```
%% LaTeX table
\hline
Species &
                \hat{K}^mV & \frac{c}{c_0} & \hat{K}
\hline
6PGC & 29 &
               0.4784 & 6.2335 \\
ADP & -27 &
               0.0704 & 5.1546 \\
ATP & 27 &
               1.2221 & 2.2539 \\
CO2 & -30 &
               0.0095 & 33.7942 \\
               0.3883 & 1.7790 \\
DHAP & -10 &
E4P & -27 &
               0.0062 & 57.9353 \\
F6P & -21 &
               0.3198 & 1.4140 \\
FDP & -8 &
               1.9289 & 0.3880 \\
               0.0344 & 14.9020 \\
G3P & -18 &
G6P & -5 &
               1.0000 & 0.8377 \\
NADP & 30 &
               0.0003 & 11747.0633 \\
NADPH & -30 &
               0.0154 & 21.0027 \\
R5P & 5 &
               0.0999 & 12.2419 \\
               0.0142 & 86.1551 \\
RU5PD & 5 &
S7P & 24 &
               0.1119 & 21.7513 \\
XU5PD & 5 &
               0.0230 & 51.6829 \\
\hline
['6PGC', '6PGL', 'ADP', 'ATP', 'CO2', 'DHAP', 'E4P', 'F6P', 'FDP', 'G3P', 'G6P',
'H', 'H2O', 'NADP', 'NADPH', 'R5P', 'RU5PD', 'S7P', 'XU5PD']
```

7 Simulation

7.1 Set up parameters

• Reaction constants already set

```
[21]: for i,spec in enumerate(s['species']):
    #K_spec = np.exp(phi_est[i]/st.V_N())
    parameter['K_'+spec] = K_spec[i]
[ ]:
```

7.2 Set up chemostats and flowstats

```
[22]: def setPath(s,path='R5P'):
    print('\n Path =', path)
```

```
if path == 'R5P':
       chemostats = ['ADP', 'ATP', 'CO2', 'G6P', 'H', 'H2O', 'NADP', __
→'NADPH', 'R5P']
       flowstats =['G6PDH2R']
       dX_G6P = 5
   elif path == 'NADPH':
       chemostats = ['ADP', 'ATP', 'CO2', 'G6P', 'H', 'H2O', 'NADP', 'NADPH']
       flowstats = []
       dX_G6P = 1
   elif path == 'both':
       chemostats = ['ADP', 'ATP', 'CO2', 'G6P', 'H', 'H2O', 'NADP', _
→'NADPH', 'R5P']
       flowstats = ['PGI', 'TKT2']
       dX_G6P = 1
   elif path == 'all':
       chemostats = ['ADP', 'ATP', 'CO2', 'G6P', 'H', 'H2O', 'NADP', _
→'NADPH', 'R5P','G3P']
       flowstats = []
       dX_G6P = 10
  sc = st.statify(s,chemostats=chemostats)
   sf = st.statify(s,flowstats=flowstats)
   return sc,sf,dX_G6P
```

7.3 Time unit

```
[43]: \#\#t_0 = ((1000*c_G6P)/flux_PGI)*100
print(f"Time unit: \{t_0:4.2f\} sec")
```

Time unit: 7.95 sec

7.4 Simulation

```
[24]: approximateFlowstats = True

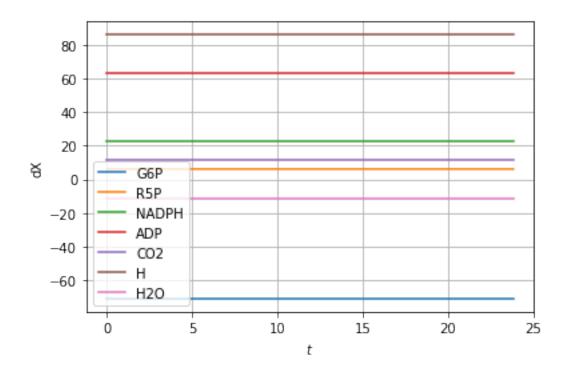
Spec = ['G6P','R5P','NADPH','ADP','C02','H','H20']
paths = ['all','both','R5P','NADPH']
#paths = ['R5P']
RATIO = {}
for path in paths:
    Ratio = {}
    normalisedRatio = {}

    ## Set up pathway]
    spec = sc['species']
    sc,sf,dX_G6P_0 = setPath(s,path=path)

## Set up parameters
par = copy.copy(parameter)
```

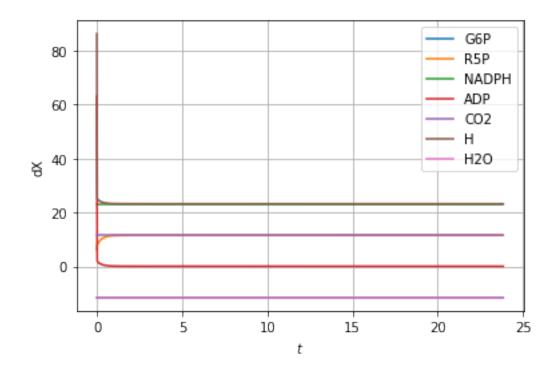
```
if approximateFlowstats:
      small = 1e-3
      par = copy.copy(parameter)
      for fs in sf['flowstats']:
           par['kappa_'+fs] = small
       sf = None
   ## Simulate
  t = np.linspace(0,3*t_0,1000)
    ## Find steady-state with no flowstats
    dat_ss = st.sim(s,sc=sc,sf=sf,t=t,parameter=parameter,X0=conc)
    X_ss = dat_ss['X'][-1,:]
  dat = st.sim(s,sc=sc,sf=sf,t=t,parameter=par,X0=conc)
   #st.plot(s,dat,species=[])
  st.plot(s,dat,reaction=[],species=Spec,dX=True)
  ## Extract some external flows
  DX = dat['dX']
  dX = \{\}
  for Sp in Spec:
       dX[Sp] = DX[:,spec.index(Sp)]
       Ratio[Sp] = -dX[Sp]/dX['G6P']
      normalisedRatio[Sp] = -dX_G6P_0*dX[Sp]/dX['G6P']
  RATIO[path] = normalisedRatio
   ## Print steady-state values
  for Sp in Spec:
       ratio = Ratio[Sp][-1]
       print(f'\{Sp\}: \t\{dX[Sp][0]:3.1f\} \t\{dX[Sp][-1]:3.
\rightarrow1f}\t{(dX_G6P_0*ratio):3.1f}\t{100*ratio:3.1f}\%')
```

Path = all



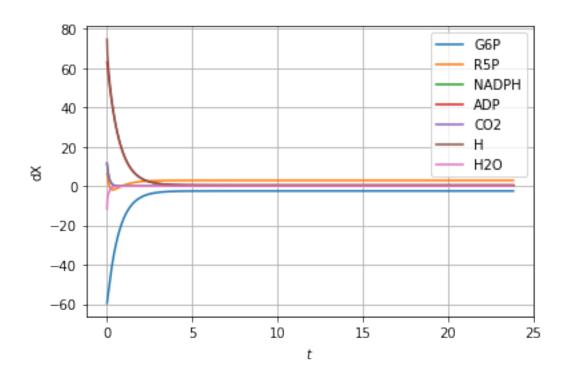
G6P:	-71.1	-71.1	-10.0	-100.0%
R5P:	6.2	6.2	0.9	8.7%
NADPH:	23.2	23.2	3.3	32.6%
ADP:	63.1	63.1	8.9	88.8%
CO2:	11.6	11.6	1.6	16.3%
H:	86.3	86.3	12.1	121.3%
H20:	-11.6	-11.6	-1.6	-16.3%

Path = both



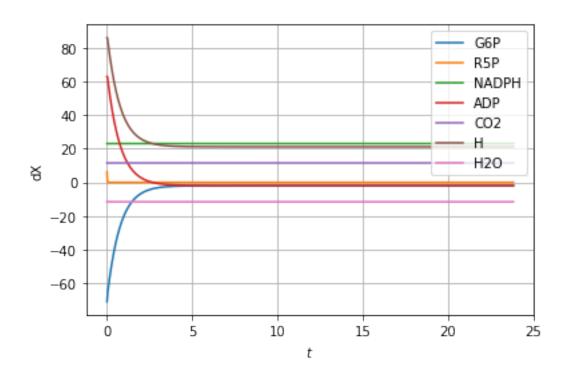
G6P:	-11.6	-11.6	-1.0	-100.0%
R5P:	6.2	11.5	1.0	99.4%
NADPH:	23.2	23.2	2.0	200.0%
ADP:	63.1	0.0	0.0	0.4%
CO2:	11.6	11.6	1.0	100.0%
H:	86.3	23.2	2.0	200.4%
H2O:	-11.6	-11.6	-1.0	-100.0%

Path = R5P



G6P:	-59.5	-2.4	-5.0	-100.0%
R5P:	6.2	2.9	6.0	120.0%
NADPH:	11.6	0.0	0.0	0.2%
ADP:	63.1	0.5	1.0	20.0%
CO2:	11.6	0.0	0.0	0.1%
H:	74.7	0.5	1.0	20.2%
H20:	-11.6	-0.0	-0.0	-0.1%

Path = NADPH



```
G6P:
                         -1.0
                                 -100.0%
        -71.1
                 -1.9
R5P:
        6.2
                 0.0
                         0.0
                                 0.0%
        23.2
                                 1200.0%
NADPH:
                 23.1
                         12.0
ADP:
        63.1
                 -1.9
                         -1.0
                                 -100.0%
CO2:
        11.6
                11.6
                         6.0
                                 600.0%
                 21.2
                                 1100.0%
Η:
        86.3
                         11.0
        -11.6
                 -11.6
                         -6.0
                                 -600.0%
H20:
```

```
[25]: ## Plot ratios
      name = ['i','ii','iii']
      for sp in ['R5P','NADPH']:
          BigFont = 24
          plt.rcParams.update({'font.size': BigFont})
          for i,path in enumerate(['both','R5P','NADPH']):
              Ratio = RATIO[path]
              label = f'Path {name[i]}'
              plt.plot(t/t_0,Ratio[sp],label=label,linewidth=5)
          if sp=='R5P':
              ylim = 8
          else:
              ylim=15
          plt.ylim((0,ylim))
          ylabel = r' \frac{r'}{rho_{i+sp+'}}
          plt.ylabel(ylabel)
          plt.xlabel('$t/t_0$')
          plt.legend()
          plt.grid()
          plt.savefig(f'Figs/{sp}.pdf',bbox_inches='tight')
```

plt.show()

[25]: [<matplotlib.lines.Line2D at 0x7fc620edcfd0>]

[25]: [<matplotlib.lines.Line2D at 0x7fc620f28250>]

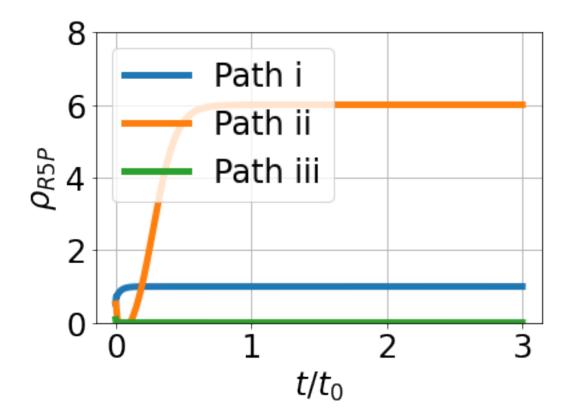
[25]: [<matplotlib.lines.Line2D at 0x7fc620ee74f0>]

[25]: (0.0, 8.0)

[25]: Text(0, 0.5, '\$\\rho_{R5P}\$')

[25]: Text(0.5, 0, '\$t/t_0\$')

[25]: <matplotlib.legend.Legend at 0x7fc6418b8a90>



[25]: [<matplotlib.lines.Line2D at 0x7fc620c53d90>]

[25]: [<matplotlib.lines.Line2D at 0x7fc620eadfa0>]

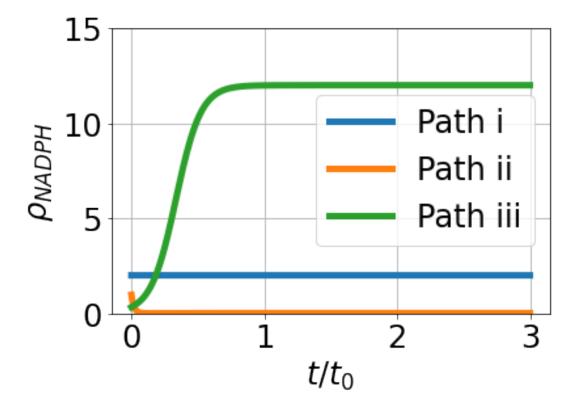
[25]: [<matplotlib.lines.Line2D at 0x7fc620c6b3d0>]

[25]: (0.0, 15.0)

[25]: Text(0, 0.5, '\$\\rho_{NADPH}\$')

```
[25]: Text(0.5, 0, '$t/t_0$')
```

[25]: <matplotlib.legend.Legend at 0x7fc620bd5400>



```
[26]: X = np.array([1,2,3])
      print(X)
      print(X.T)
      print(np.outer(X,X))
      print(np.linalg.pinv(np.outer(X,X)))
      print(np.linalg.pinv(np.outer(X,X))@X.T)
      print(X@X.T)
     [1 2 3]
     [1 2 3]
     [[1 2 3]
      [2 4 6]
      [3 6 9]]
     [[0.00510204 0.01020408 0.01530612]
      [0.01020408 0.02040816 0.03061224]
      [0.01530612 0.03061224 0.04591837]]
     [0.07142857 0.14285714 0.21428571]
     14
```

8 Michaelis-Menten formulation

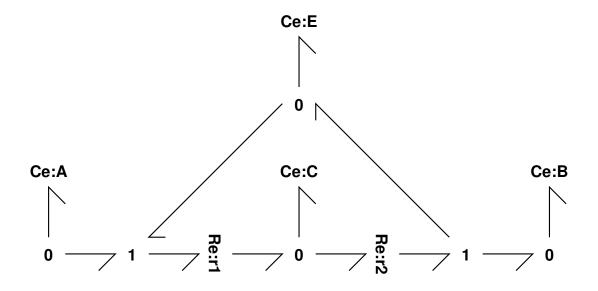
8.1 Show results

```
[27]: for reac in reaction:
          MMpar = MMparameter[reac]
          f_max = MMpar['f_max']
          rho = MMpar['rho']
          k_v = MMpar['k_v']
          print(f'Reaction {reac}:')
          print(f'f_max = \{f_max: 6.2f\}; rho = \{rho: 1.2f\}; k_v = \{k_v: 4.2f\}')
          kappa,K_CE = MMtoBG(MMpar,K_E=K_E)
          print(f'kappa_1 = {kappa[0]:3.1f}; kappa_2 = {kappa[1]:3.1f}; K_C =__
       \rightarrow \{K_CE[0]:3.1f\}; K_E = \{K_CE[1]:3.1f\}'\}
     Reaction PGI:
     f_{max} = 132.87; rho = 0.20; k_v = 0.10
     kappa_1 = 664.4; kappa_2 = 166.1; K_C = 1.0; K_E = 10.0
     Reaction PFK:
     f_{max} = 61.18; rho = 0.20; k_v = 0.10
     kappa_1 = 305.9; kappa_2 = 76.5; K_C = 1.0; K_E = 10.0
     Reaction FBA:
     f_{max} = 123.18; rho = 0.20; k_v = 0.10
     kappa_1 = 615.9; kappa_2 = 154.0; K_C = 1.0; K_E = 10.0
     Reaction TPI:
     f_{max} = 267.28; rho = 0.20; k_v = 0.10
     kappa_1 = 1336.4; kappa_2 = 334.1; K_C = 1.0; K_E = 10.0
     Reaction GND:
     f_{max} = 9.55; rho = 0.20; k_v = 0.10
     kappa_1 = 47.8; kappa_2 = 11.9; K_C = 1.0; K_E = 10.0
     Reaction RPI:
     f_{max} = 5570.71; rho = 0.20; k_v = 0.10
     kappa_1 = 27853.5; kappa_2 = 6963.4; K_C = 1.0; K_E = 10.0
     Reaction TKT2:
     f_{max} = 4.48; rho = 0.20; k_v = 0.10
     kappa_1 = 22.4; kappa_2 = 5.6; K_C = 1.0; K_E = 10.0
     Reaction TKT1:
     f_{max} = 13.32; rho = 0.20; k_v = 0.10
     kappa_1 = 66.6; kappa_2 = 16.7; K_C = 1.0; K_E = 10.0
     Reaction RPE:
     f_{max} = 126.53; rho = 0.20; k_{v} = 0.10
     kappa_1 = 632.7; kappa_2 = 158.2; K_C = 1.0; K_E = 10.0
```

8.2 Bond graph model of Enzyme Catalysed Reaction (RE)

```
[28]: sbg.model('RE_abg.svg')
import RE_abg as RE
disp.SVG('RE_abg.svg')
```

[28]:



[29]:

$$A + E \stackrel{r_1}{\Longleftrightarrow} C \tag{24}$$

$$C \stackrel{r_2}{\Longleftrightarrow} B + E$$
 (25)

[30]:

$$G_6P \stackrel{PGI}{\longleftrightarrow} F_6P$$
 (26)

$$ATP + F_6P \stackrel{PFK}{\longleftarrow} ADP + FDP + H$$
 (27)

$$FDP \stackrel{FBA}{\longleftarrow} DHAP + G_3P \tag{28}$$

$$DHAP \stackrel{TPI}{\longleftarrow} G_3P \tag{29}$$

$$G_6P + NADP \xrightarrow{G_6PDH_2R} {}_6PGL + H + NADPH$$
 (30)

$$_{6}PGL + H_{2}O \xrightarrow{PGL} _{6}PGC + H$$
 (31)

$$_{6}PGC + NADP \xrightarrow{GND} CO_{2} + NADPH + RU_{5}PD$$
 (32)

$$RU_5PD \stackrel{RPI}{\longleftarrow} R_5P$$
 (33)

$$E_4P + XU_5PD \xrightarrow{TKT_2} F_6P + G_3P$$
 (34)

$$G_3P + S_7P \xrightarrow{TALA} E_4P + F_6P \tag{35}$$

$$R_5P + XU_5PD \xrightarrow{TKT_1} G_3P + S_7P$$
 (36)

$$RU_5PD \stackrel{RPE}{\longleftarrow} XU_5PD$$
 (37)

8.3 Replace Re components by RE

[31]: imp.reload(mbg)
mbg.ReRE(PPP,quiet=quiet)

[31]: <module 'modularBondGraph' from '/home/peterg/WORK/Research/SystemsBiology/lib/python/modularBondGraph.py'>

[32]: sPPPRE = st.stoich(PPP,quiet=quiet)

[]:

[33]: disp.Latex(st.sprintrl(sPPPRE,chemformula=True))

[33]:

$$G_6P + PGIase \xrightarrow{PGI_1} PGIcmp$$
 (38)

$$PGIcmp \stackrel{PGI_2}{\longleftarrow} F_6P + PGIase$$
 (39)

$$ATP + F_6P + PFKase \xrightarrow{PFK_1} PFKcmp$$
 (40)

$$PFKcmp \stackrel{PFK_2}{\longleftarrow} ADP + FDP + H + PFKase$$
 (41)

$$FDP + FBAase \xrightarrow{FBA_1} FBAcmp$$
 (42)

$$FBAcmp \xrightarrow{FBA_2} DHAP + G_3P + FBAase$$
 (43)

$$DHAP + TPIase \xrightarrow{TPI_1} TPIcmp$$
 (44)

TPIcmp
$$\stackrel{\text{TPI}_2}{\longleftarrow}$$
 G₃P + TPIase (45)

$$G_6P + NADP + G_6PDH_2Rase \xrightarrow{G_6PDH_2R_1} G_6PDH_2Rcmp$$
 (46)

$$G_6PDH_2Rcmp \xrightarrow{G_6PDH_2R_2} {}_6PGL + H + NADPH + G_6PDH_2Rase$$
 (47)

$$_{6}PGL + H_{2}O + PGLase \xrightarrow{PGL_{1}} PGLcmp$$
 (48)

$$PGLcmp \stackrel{PGL_2}{\longleftarrow} {}_{6}PGC + H + PGLase$$
 (49)

$$_{6}$$
PGC + NADP + GNDase $\stackrel{\text{GND}_{1}}{\Longleftrightarrow}$ GNDcmp (50)

GNDcmp
$$\stackrel{\text{GND}_2}{\longleftarrow}$$
 CO₂ + NADPH + RU₅PD + GNDase (51)

$$RU_5PD + RPIase \xrightarrow{RPI_1} RPIcmp$$
 (52)

$$RPIcmp \stackrel{RPI_2}{\longleftarrow} R_5P + RPIase$$
 (53)

$$E_4P + XU_5PD + TKT_2ase \xrightarrow{TKT_{21}} TKT_2cmp$$
 (54)

$$TKT_{2}cmp \stackrel{TKT_{22}}{\longleftarrow} F_{6}P + G_{3}P + TKT_{2}ase$$
 (55)

$$G_3P + S_7P + TALAase \xrightarrow{TALA_1} TALAcmp$$
 (56)

$$TALAcmp \xrightarrow{TALA_2} E_4P + F_6P + TALAase$$
 (57)

$$R_5P + XU_5PD + TKT_1ase \xrightarrow{TKT_{11}} TKT_1cmp$$
 (58)

$$TKT_1 cmp \xrightarrow{TKT_{12}} G_3P + S_7P + TKT_1 ase$$
 (59)

$$RU_5PD + RPEase \xrightarrow{RPE_1} RPEcmp$$
 (60)

$$RPEcmp \stackrel{RPE_2}{\longleftarrow} XU_5PD + RPEase$$
 (61)

[34]: print(chemostats)

['ADP', 'ATP', 'CO2', 'G3P', 'G6P', 'H', 'H2O', 'NADP', 'NADPH', 'R5P']

8.4 Set up parameters

```
[36]: parameter = {}
for i,spec in enumerate(s['species']):
    parameter['K_'+spec] = K_spec[i]

for reac in reaction:
    MMpar = MMparameter[reac]
    kappa,K_CE = MMtoBG(MMpar,K_E=100.0)
    for i in ['1','2']:
        Kappa = f'kappa_{reac}{i}'
        parameter[Kappa] = kappa[int(i)-1]
    for i,spec in enumerate(['cmp','ase']):
        K = f'K_{reac}{spec}'
        parameter[K] = K_CE[i]

# print(parameter)
```

```
[37]: ## Initial conds
n_X = sPPPRE['n_X']
X0 = 0.5*np.ones(n_X)
for i,spec in enumerate(sPPPRE['species']):
    if spec in s['species']:
        X0[i] = conc[s['species'].index(spec)]
```

8.5 Simulate

```
[38]: approximateFlowstats = True
      Spec = ['G6P','R5P','NADPH','ADP','C02','H','H20']
      paths = ['all','both','R5P','NADPH']
      #paths = ['R5P']
      RATIO = \{\}
      for path in paths:
          Ratio = {}
          normalisedRatio = {}
          ## Set up pathway]
          spec = sPPPRE['species']
          sc,sf,dX_G6P_0 = setPath(sPPPRE,path=path)
          ## Set up parameters
          par = copy.copy(parameter)
          if approximateFlowstats:
              small = 1e-3
              par = copy.copy(parameter)
              for fs in sf['flowstats']:
                  par['kappa_'+fs+'1'] = small
                  par['kappa_'+fs+'2'] = small
              sf = None
          ## Find the initial condion X_s after the initial transient due to E/C
          t_ss = np.linspace(0,t_0/100,100)
          dat_ss = st.sim(sPPPRE,sc=sc,t=t_ss,parameter=parameter,X0=X0)
          X_ss = dat_ss['X'][-1,:]
          ## Simulate from after transient
          dat = st.sim(sPPPRE,sc=sc,t=t,parameter=par,X0=X_ss)
            st.plot(s, dat, reaction=[], species=Spec, dX=True)
          ## Extract some external flows
          DX = dat['dX']
          dX = \{\}
          for Sp in Spec:
              dX[Sp] = DX[:,spec.index(Sp)]
              Ratio[Sp] = -dX[Sp]/dX['G6P']
              normalisedRatio[Sp] = -dX_G6P_0*dX[Sp]/dX['G6P']
```

```
RATIO[path] = normalisedRatio
     ## Print steady-state values
     for Sp in Spec:
         ratio = Ratio[Sp][-1]
         print(f'{Sp}:\t{dX[Sp][0]:3.1f} \t{dX[Sp][-1]:3.}
  \rightarrow1f}\t{(dX_G6P_0*ratio):3.1f}\t{100*ratio:3.1f}\%')
 Path = all
G6P:
        -63.8
                -62.5
                         -10.0
                                 -100.0%
R5P:
        0.8
                -0.7
                         -0.1
                                 -1.0%
NADPH: 2.3
                1.0
                         0.2
                                 1.6%
ADP:
                62.8
                         10.0
                                 100.4%
        62.8
CO2:
        1.8
                0.5
                         0.1
                                 0.8%
Η:
        63.7
                63.7
                         10.2
                                 102.0%
H20:
        -0.1
                -0.5
                         -0.1
                                 -0.8%
Path = both
Flowstat PGI is not a model reaction
Flowstat TKT2 is not a model reaction
G6P:
        -0.7
                -0.5
                         -1.0
                                 -100.0%
R5P:
        1.5
                0.5
                         0.9
                                 92.9%
NADPH: 2.3
                1.0
                         2.0
                                 197.1%
ADP:
        53.3
                0.0
                         0.0
                                 4.7%
CO2:
        1.8
                0.5
                         1.0
                                 98.2%
Η:
        54.2
                1.0
                         2.0
                                 201.8%
H20:
        -0.1
                -0.5
                         -1.0
                                 -98.3%
Path = R5P
Flowstat G6PDH2R is not a model reaction
G6P:
        -55.7
                -0.4
                         -5.0
                                 -100.0%
R5P:
        1.5
                0.4
                         6.0
                                 119.9%
NADPH: 1.8
                0.0
                         0.0
                                 0.3%
                         1.0
ADP:
        53.3
                0.1
                                 20.0%
CO2:
        1.8
                0.0
                         0.0
                                 0.2%
                0.1
                                 20.3%
H:
        53.8
                         1.0
H20:
        -0.1
                -0.0
                         -0.0
                                 -0.2%
 Path = NADPH
G6P:
        -56.9
                -0.1
                         -1.0
                                 -100.0%
R5P:
        0.1
                0.0
                         0.0
                                 0.0%
NADPH:
        3.7
                1.0
                         11.6
                                 1155.1%
ADP:
        53.2
                -0.1
                         -1.0
                                 -95.7%
CO2:
        3.3
                0.5
                         5.8
                                 575.7%
H:
        54.2
                0.9
                         10.6
                                 1059.4%
```

H20:

-0.1

-0.5

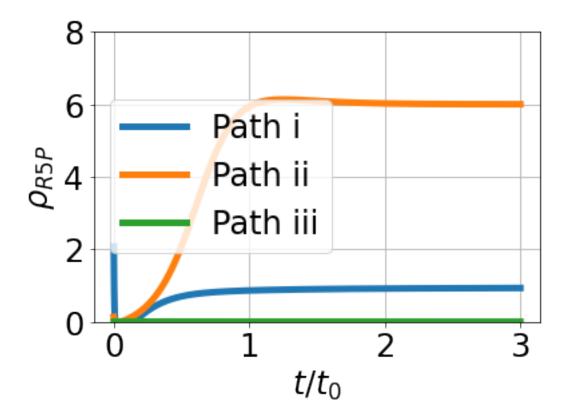
-5.8

-576.2%

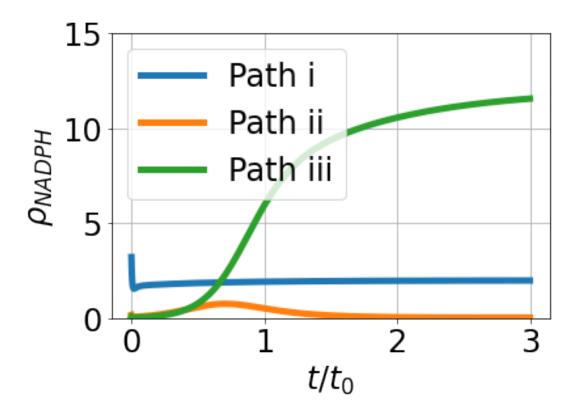
8.6 Plot ratios

```
[39]: | ## Plot ratios
      name = ['i','ii','iii']
      for sp in ['R5P','NADPH']:
          BigFont = 24
          plt.rcParams.update({'font.size': BigFont})
          for i,path in enumerate(['both','R5P','NADPH']):
              Ratio = RATIO[path]
              label = f'Path {name[i]}'
              plt.plot(t/t_0,Ratio[sp],label=label,linewidth=5)
          ylabel = r' \frac{r'}{rho_{i+sp+'}}
          plt.ylabel(ylabel)
          plt.xlabel('$t/t_0$')
          plt.legend()
          plt.grid()
          if sp=='R5P':
              ylim = 8
          else:
              ylim=15
          plt.ylim((0,ylim))
          plt.savefig(f'Figs/{sp}_MM.pdf',bbox_inches='tight')
          plt.show()
[39]: [<matplotlib.lines.Line2D at 0x7fc62100d460>]
```

```
[39]: [<matplotlib.lines.Line2D at 0x7fc62100d460>]
[39]: [<matplotlib.lines.Line2D at 0x7fc6418d8fd0>]
[39]: [<matplotlib.lines.Line2D at 0x7fc6418d8490>]
[39]: Text(0, 0.5, '$\\rho_{R5P}$')
[39]: Text(0.5, 0, '$t/t_0$')
[39]: <matplotlib.legend.Legend at 0x7fc620ee7280>
[39]: (0.0, 8.0)
```



- [39]: [<matplotlib.lines.Line2D at 0x7fc641da0b80>]
- [39]: [<matplotlib.lines.Line2D at 0x7fc641da0f70>]
- [39]: [<matplotlib.lines.Line2D at 0x7fc641da0700>]
- [39]: Text(0, 0.5, '\$\\rho_{NADPH}\$')
- [39]: Text(0.5, 0, '\$t/t_0\$')
- [39]: <matplotlib.legend.Legend at 0x7fc620ee7490>
- [39]: (0.0, 15.0)



```
[40]: ## Compare concentrations
      conc_mam = Concentration['Mammalian']
      conc_eco = Concentration['Ecoli']
      for spec in conc_eco.keys():
          ratio = conc_eco[spec]/conc_mam[spec]
          print(f'{spec}: {ratio:2.2e}')
     6PGC: 2.28e+02
     ADP: 9.75e-01
     ATP: 2.06e+00
     CO2: 9.86e-03
     DHAP: 1.88e+00
     E4P: 4.76e+00
     F6P: 2.60e+01
     FDP: 1.00e+01
     G3P: 1.92e+00
     G6P: 1.17e+01
     NADP: 7.32e-02
     NADPH: 1.85e+00
     R5P: 2.77e+01
     RU5PD: 2.13e+01
     S7P: 4.87e+01
     XU5PD: 6.05e+00
```

[]:

[]:

References

- Peter J Gawthrop. Energy-based Feedback Control of Biomolecular Systems with Cyclic Flow Modulation. Available at arXiv:2007.14762, July 2020.
- Peter J. Gawthrop, Peter Cudmore, and Edmund J. Crampin. Physically-plausible modelling of biomolecular systems: A simplified, energy-based model of the mitochondrial electron transport chain. *Journal of Theoretical Biology*, 493:110223, 2020. ISSN 0022-5193. doi: 10. 1016/j.jtbi.2020.110223.
- J. Orth, R. Fleming, and B. Palsson. Reconstruction and use of microbial metabolic networks: the core escherichia coli metabolic model as an educational guide. *EcoSal Plus*, 2010. doi: 10.1128/ecosalplus.10.2.1.
- Junyoung O. Park, Sara A. Rubin, Yi-Fan Xu, Daniel Amador-Noguez, Jing Fan, Tomer Shlomi, and Joshua D. Rabinowitz. Metabolite concentrations, fluxes and free energies imply efficient enzyme usage. *Nat Chem Biol*, 12(7):482–489, Jul 2016. ISSN 1552-4450. doi: 10.1038/nchembio.2077.