

Understanding pore limiting diameter in MOFs:

Features Identified by XAI Analysis

- Volume per Atom
- Symmetry Function G1
- Unoccupied Energy Levels Near the Conduction Band Minimum

Volume per Atom:

Explanation: The volume per atom in a material can influence the pore limiting diameter (PLD) in metal-organic frameworks (MOFs) by affecting the overall density and packing of the framework. A larger volume per atom might suggest a more open structure, potentially leading to larger pore sizes, while a smaller volume per atom could indicate tighter packing and smaller pores.

Scientific Evidence: The documents discuss the importance of crystal density as a descriptor for MOF structures, which is related to the volume per atom. Variations in atomic packing, which are influenced by the volume per atom, can lead to different structures and pore sizes (Chung et al., 2019).

Hypothesis: A higher volume per atom in MOFs may correlate with larger pore limiting diameters due to less dense atomic packing, allowing for larger channels within the framework.

Symmetry Function G1:

Explanation: Symmetry functions, such as G1, are often used in computational chemistry to describe the local environment around atoms. While the documents do not explicitly mention Symmetry Function G1, symmetry in general can influence the arrangement of atoms and thus the pore structure in MOFs.

Scientific Evidence: An explicit relationship between Symmetry Function G1 and the pore limiting diameter was not found in the given documents. However, the documents do discuss the geometric properties of MOFs, which are inherently related to symmetry (Haldoupis et al., 2010).

Hypothesis: Symmetry Function G1 may impact the pore limiting diameter by influencing the spatial arrangement of atoms, potentially affecting the uniformity and size of the pores.

Unoccupied Energy Levels Near the Conduction Band Minimum:

Explanation: Unoccupied energy levels near the conduction band minimum can affect the electronic properties of MOFs, which in turn might influence the interaction between the framework and adsorbates. This could indirectly affect the pore limiting diameter by altering the stability and structure of the MOF.

Scientific Evidence: The documents do not explicitly link unoccupied energy levels to the pore limiting diameter. However, they discuss the impact of energy barriers and adsorption properties on MOF performance, which could be related to electronic structure (Haldoupis et al., 2010).

Hypothesis: Unoccupied energy levels near the conduction band minimum may influence the pore limiting diameter by affecting the electronic interactions within the MOF, potentially leading to structural changes that alter pore sizes.

Summary

The relationship between the identified features and the pore limiting diameter in MOFs is complex and not explicitly detailed in the provided documents. The volume per atom is likely to influence the pore limiting diameter through its effect on atomic packing and density, potentially leading to larger or smaller pores depending on the volume. Symmetry Function G1, while not explicitly discussed, could impact the pore structure

by affecting the spatial arrangement of atoms. Unoccupied energy levels near the conduction band minimum might indirectly influence the pore limiting diameter by altering the electronic properties and stability of the MOF. These hypotheses are based on the general principles discussed in the documents, such as the importance of geometric and electronic properties in determining MOF structure and function.

References

1. Chung, Y. G., Haldoupis, E., Bucior, B. J., Haranczyk, M., Lee, S., Zhang, H., Vogiatzis, K. D., Milisavljevic, M., Ling, S., Camp, J. S., Slater, B., Siepmann, J. I., Sholl, D. S., & Snurr, R. Q. (2019). Advances, Updates, and Analytics for the Computation-Ready, Experimental Metal–Organic Framework Database: CoRE MOF 2019.
2. Haldoupis, E., Nair, S., & Sholl, D. S. (2010). Efficient Calculation of Diffusion Limitations in Metal Organic Framework Materials: A Tool for Identifying Materials for Kinetic Separations.

Explanation generated with XpertAI (2024)