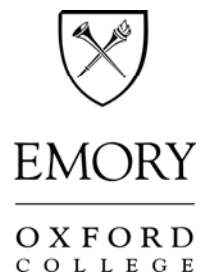


Chemistry 142

General Chemistry

Spring 2009



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COURSE DESCRIPTION: Chem 142 is the continuation of Chem 141. This course covers transition metal chemistry, solution chemistry, chemical kinetics, chemical equilibrium, thermodynamics, acid-base chemistry, electrochemistry, and nuclear chemistry. The class consists of three hours of lecture and one three-hour laboratory per week.

RESOURCES:

- Ebbing, General Chemistry, 9th edition, Houghton Mifflin
- Optional study guide and solutions manual
- Laboratory Manual, sold by the Chemistry Department
- A scientific calculator (not programmable)
- Carbon-copy lab notebook
- Goggles or safety glasses must be purchased and must be worn in the lab at all times. If you come to the lab without safety glasses, you will not be admitted.
- Lab coat or apron optional
- Learnlink Class Conference

MISSION: Oxford College is dedicated to the goal of excellence in undergraduate liberal arts education. Science, including chemistry is an integral part of the liberal arts. In this course you will have an opportunity to master these liberal arts skills:

Reasoning:

- Problem solving
- Critical Thinking
- Logic
- Calculation/Computation
- Investigation
- Analysis of data

Language:

- Listening and interpreting
- Reading
- Writing

Aesthetics:

- Observing
- Seeing relationships among form, pattern, harmony and shape

Imagination:

- Prediction
- Developing scientific insight (hypotheses)

LEARNING GOALS: The primary learning goals for this class are for you to:

- Utilize critical thought and reasoning to understand chemical behavior at the microscopic and macroscopic levels.
- From your knowledge of chemistry and chemical systems, be able to develop solutions to problems which you have not encountered before.

CLASS POLICIES:

Lecture: Lectures will cover material from the text as well as supplemental material not in the text. We will be covering chapters 22 (sections 3 & 7), 11-16, and 18-20. You should read the assigned material before each lecture and be prepared to ask any questions you may have during the lectures. Please remember to turn off your cell phones before class.

Attendance: Attendance is required for each lecture and laboratory class. It is **YOUR** responsibility to sign the attendance sheet each class period. Forgetting to sign the attendance sheet is treated the same as an absence. You are allowed three absences in lecture (regardless of the reason) and NO absences in lab. There are no “excused” absences. For each absence beyond three, three points will be deducted from your final average. If you arrive late to class, you must see me after class to sign the attendance sheet. If you are more than fifteen minutes late to class it will count as an absence. Makeup exams are not given. If you miss an exam and present me with acceptable written documentation (i.e. note from a doctor or a jury summons, etc.), the grade on the final exam will count in place of the missed grade. If the excuse is not acceptable, the exam grade will be a zero. In general, illness or an emergency situation are the only acceptable excuses for missing an exam. Missing an exam also counts as an absence in the course.

Grades: The grade breakdown for lecture is as follows:

Exam Average	75%
<u>Final Exam</u>	<u>25%</u>
Total	100%

Exams: Exams will be closed book, and generally consist of 10-15 problems which must be worked out with all work shown and 10-20 true false or multiple choice. There will be four short papers that will also count as one exam. Instructions regarding the papers will be posted on Learnlink. Tentative exam dates are as follows (These are subject to change):

Exam 1	Friday, Feb.6
Exam 2	Friday, Mar. 7
Exam 3	Friday, April 3
Exam 4	Friday, April 24

Final Exam: The final exam will be cumulative. The final exam will take place during the regularly scheduled final exam period.

Homework: Homework will not be graded. However, students are expected to work problems from the book in preparation for the exams. A student assistant will be available outside of class to assist in problem solving and reviewing for the exams. Times and locations will be posted on Learnlink.

Lab Grades: Your lab grade will count in one of two ways, whichever results in a higher grade in the course for you:

- A) Your course grade will be computed by adjusting your grade on the lecture portion using your lab average as shown below. This method normally benefits students whose exam average is a high B or an A:

<u>Lab average</u>	<u>Lecture grade adjustment</u>
90 to 92.9	+1
93.0 and up	+2

- B) Your course grade will be computed by taking 80% of your lecture grade and 20% of your lab grade. This method usually benefits students whose exam average is a B or lower. If your lab average is below 90, your grade will be automatically computed using this method.

Grading Scale:

The grading scale is as follows:

A = 90 - 100
B = 80 - 89
C = 70 - 79
D = 60 - 69
F = 59 OR BELOW

There will be no “curve” in this class. A grade of “incomplete” will only be assigned in extreme circumstances and only after presentation of written documentation of hardship is presented to the dean’s office. Your exam average AND your lab average must both be passing or you will receive a grade of F in the course regardless of your final numerical average. Please do not ask me to calculate what you need to get on the final to make an A (or B, C, etc.) in the class. This is a simple algebra problem and if you do not know how to do it you can see the math center for help.

Honor Code: The responsibility for maintaining standards of unimpeachable honesty in all academic work and in campus judicial proceedings falls upon every individual in Oxford College of Emory University. The Honor Code is based on the fundamental expectation that every person in Oxford College will conduct his or her life according to the dictates of the Honor Code and will refuse to tolerate actions in others that violate the Honor Code. On exams, only the use of a pen/pencil and a non-programmable calculator is allowed. Lab reports and papers should be the student’s own work; collaboration is not allowed. Any outside references should be properly cited. Any student suspected of cheating or plagiarism will be referred to the Dean’s office for disciplinary action.