

Chemistry 221
Fall, 2000

Instructor: Monica Ali, Pierce Hall, Room 210, Telephone 4-8340

Text and Supplies: Ege, Seyhan, "Organic Chemistry," 4th edition, Houghton Mifflin, Boston, MA, 1999

Ege, Seyhan, Study Guide

Houghton-Mifflin, CD-ROM

Model kit

Memo Pad

Attendance: Attendance in class is required and will be taken for every class. Anyone, who is present for every class during the semester, will have one point added onto the final numerical grade, before the determination of the letter grade. Anyone, who is absent from more than three classes, will have one point deducted from the final numerical grade, before the determination of the letter grade.

Tests: Tests will be held on the following Friday afternoons at 2:15 PM in rooms 201 and 223 of Pierce Hall. A maximum of two hours in which to complete the test will be allowed. **There are no makeup tests.** Ordinarily, the material on each test will be the material covered in class from the time of the previous test up to and including that covered on the Wednesday before the test.

September 8
September 22
October 6
October 20
November 3
November 17
December 8

If a student must be absent from a test because of serious illness or a death in the family (the only acceptable excuses), the instructor **MUST** be notified before the time of the test.

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Final Exam: The final examination for the class is comprehensive and is scheduled for three hours at the following times. Final exams must be taken at the scheduled time.

9:00 AM class	Tuesday, December 19 th , 2:00 PM
10:00 AM class	Friday, December 15 th , 9:00 AM

Any student having an A or A- total test average at the end of the semester may (if he or she wishes) be exempted from taking the final exam. In such a case, the grade for the course will be the average of the semester tests.

Calculation of Course Grade The letter grade for the course will be determined in the following manner.

The test average will be multiplied by 80%
The final exam will be multiplied by 20%
The final numerical grade will be the sum of the two above numbers
The final numerical grade will be converted to the letter grade for the course

93-100	A
90-92	A-
87-89	B+
83-86	B
80-82	B-
77-79	C+
73-76	C
70-72	C-
67-69	D+
60-66	D

Honor Code: The Honor Code is observed at Oxford College. Cheating of any sort will not be tolerated in the chemistry class. Each student is expected to submit to the instructor only his or her own work and to report to the instructor any cheating observed.

Responsibilities: For every chapter of the text covered in class, unless the instructor indicates otherwise, the student is responsible for everything in the chapter and for all problems at the end of the chapter.

Memo Pads: During the last few minutes of each class, the student will write in a memo pad, according to instructions given in class, either about material learned or about material not understood in the class. One copy of the writings will be retained by the student and one copy will be submitted to the instructor.

Office Hours: I am in the office most days from 9 AM to 5 PM. However, to be certain that I will have the time to speak with you when you come by, please try to schedule an appointment when possible.

Help Sessions: I will hold help sessions (non-mandatory) for about one hour each Thursday afternoon at 4:30 or 5:00 PM.

Advice:

1. Save this syllabus for future reference.
2. To prepare before attending each class,
 - a. Review class notes as soon as possible after the class, to recall as much as possible of what happened in class.
 - b. Read the text material covered in class that day, to deepen understanding of the material.
 - c. Work any problems which pertain to the material covered that day in class, to apply the material to both similar and new situations.
 - d. Read ahead of the instructor's lecture, to have some familiarity with the material, which will be covered in the next class.

At **least** two or three hours of preparation are necessary per class.

3. After working individually, work together in small groups, to help each other.
4. When working on a problem in the text, do not look at the answer in the Study Guide until you have made a serious effort to solve the problem.
5. Work hard at the beginning and ease up later if you can. Don't do the reverse.
6. Please take the course seriously. I do.