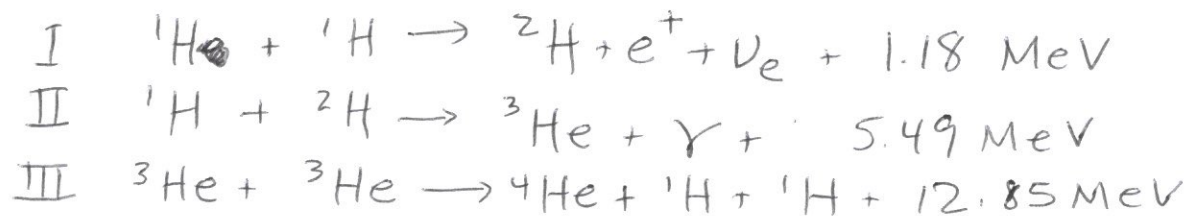


# Nuclear reactions

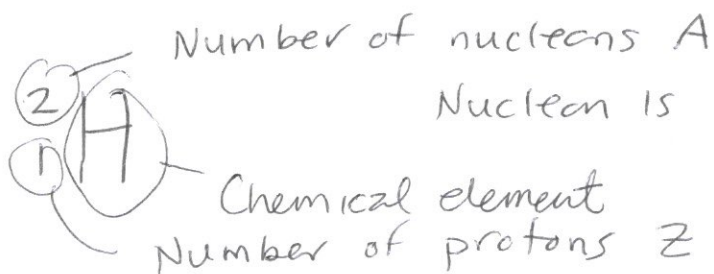
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## proton-proton chain (p-p chain)



Notation



Superfluous because the chemical element is in a 1:1 correspondence with Z.

Hydrogen means  $Z=1$

Helium means  $Z=2$

Lithium means  $Z=3$

etc.

Sometimes you will find  
p instead of  ${}^1_1\text{H}$  (proton) <sup>protium</sup>  
d instead of  ${}^2_1\text{H}$  (<sup>deuteron</sup>~~deuterium~~)  
deuterium

Isotope same number of protons but different number of neutrons, e.g.  $\begin{cases} {}^2_1\text{H} \text{ also written as } {}^2_1\text{H} \\ {}^1_1\text{H} \end{cases}$

~~$\begin{cases} {}^3_2\text{He} \text{ also written as } {}^3_2\text{He} \\ {}^4_2\text{He} \end{cases}$~~

${}^3_2\text{He}$

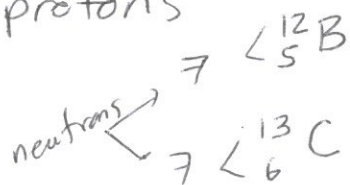
${}^3_2\text{He}$

${}^4_2\text{He}$

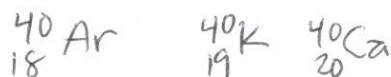
${}^4_2\text{He}$

Isotone Same number of neutrons but different number of protons

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Isobar Same number of nucleons (protons + neutrons)



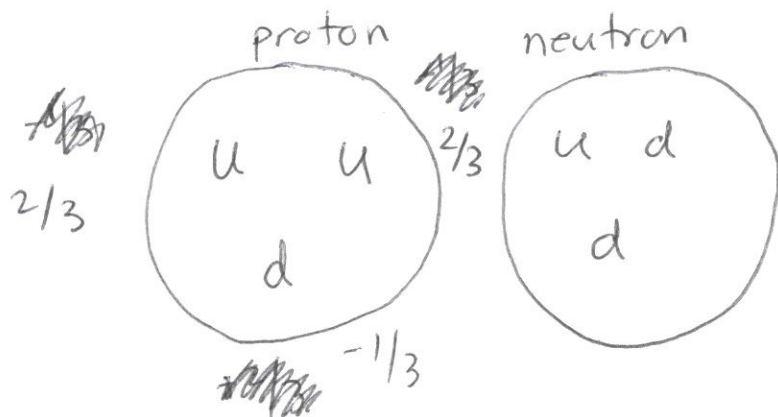
Weak nuclear force

Emission of neutrons is a telltale of weak interaction



Can you create an atom with ~~2 hydrogens~~ 2 protons and no neutrons?

★ Why or why not?



$$q_p = \frac{2}{3} + \frac{2}{3} - \frac{1}{3} = \frac{3}{3} = 1$$

$$q_n = \frac{2}{3} - \frac{1}{3} - \frac{1}{3} = \frac{0}{3} = 0$$

just like electrons can absorb energy in an atom and move to a higher energy orbital so can the proton absorb energy

$$m_p = 1.672 \times 10^{-27} \text{ kg} = 938.27 \frac{\text{MeV}}{c^2}$$

(38)

$$m_n = 1.674 \times 10^{-27} \text{ kg} = 939.56 \frac{\text{MeV}}{c^2}$$

★ If the neutron is more massive, which one is more stable? The mean life time of The proton! a neutron is 881 s (free neutron)

In addition to quantum tunneling of the Coulomb interaction (which occurs with low probability to begin with) a weak interaction is necessary to "flip" one of the (up) quarks into a (down) quark.

$$\Delta q = -1/3 - 2/3 = -1$$

$\beta^+$  decay only occurs within a nucleus, the two protons form a "nucleus" for a short time with quantum tunnel

In order to conserve charge, a positron is emitted along with a neutrino

Although a nucleus with two protons is not stable, a nucleus with 1 proton and 1 neutron is stable.

The antielectron annihilates with an electron right away producing 2  $\gamma$  rays. The neutrino interacts very weakly, so it essentially escapes.

Reaction I occurs very rarely, about once every  $10^{10}$  years. (in the sun)

The mass of an up quark is  $2.2^{+0.5}_{-0.4} \frac{\text{MeV}}{c^2}$ , the mass of a down quark is  $4.7^{+0.5}_{-0.3} \frac{\text{MeV}}{c^2}$ , so

Proton

$$2 \left( 2.2 \frac{\text{MeV}}{c^2} \right) + ~~4.7~~ \left( 4.7 \frac{\text{MeV}}{c^2} \right) = 9.1 \frac{\text{MeV}}{c^2}$$

★

But  $m_p = 938 \frac{\text{MeV}}{c^2}$  where is all the 'extra' mass coming from??

For neutron

$$\left( 2.2 \frac{\text{MeV}}{c^2} \right) + 2 \left( 4.7 \frac{\text{MeV}}{c^2} \right) = 11.6 \frac{\text{MeV}}{c^2}$$

From the strong force binding energy!

But  $m_n = 939 \frac{\text{MeV}}{c^2}$



does not require weak interaction, Coulomb barrier not that high, so it happens once every 1s (in the sun)

A consequence of this is that deuterium does not exist in the core of stars. The ratio of deuterium to hydrogen in the universe is 26 in 1 million. In the earth's oceans it is 156 per 1 million (1 in 6,410). All remain from Big Bang.



Occurs once every 300,000 years or so.

Although another path exists



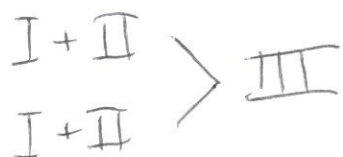
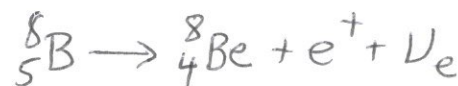
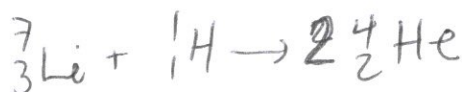
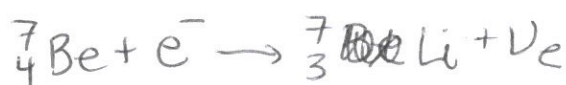
69%  
↙

31%  
↘



99.7%  
↙

0.3%  
↘



each time this occurs, 4 protons are converted into a  $^4\text{He}$  nucleus, 2  $\nu_e$ , photons, and kinetic energy. Adding all the energies,

Get the 0.7% from here.

you get 26.73 MeV, the mass of  $^4\text{He}$  is 25.71 MeV, 2(0.511 MeV) annihilation of existing  $e^-$ , 25.71 MeV.