# Simon Fraser University

# Mechatronic Systems Engineering

Midterm Exam for ENSC 331: Introduction to MEMS

Instructor: Behraad Bahreyni

4 July 2012

Time: 90 minutes

Name:	SOLUTIONS	
Student i	number:	

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Question	Mark
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Deal-Groove model for oxidation rate of silicon:	$t_{ox}^2 + At_{ox} = B(t+\tau)$

Deal-Groove rate constants for oxidation of (100) silicon wafers

	A		$B(\mu m^2/hr)$		τ (	hr)
Temperature (°C)	Dry	Wet	Dry	Wet	Dry	Wet
900	0.423	1.136	0.004	0.172	2.79	0.169
1000	0.232	0.424	0.010	0.316	0.616	0.036
1100	0.139	0.182	0.024	0.530	0.174	0.010

Deal-Groove rate constants for oxidation of (111) silicon wafers

9	A (	$(\mu m)$	Β (μι	$m^2/hr)$	$\tau (hr)$	
Temperature (°C)	Dry	Wet	Dry	Wet	Dry	Wet
900	0.252	0.676	0.004	0.172	1.72	0.102
1000	0.138	0.252	0.010	0.316	0.391	0.022
1100	0.083	0.109	0.024	0.530	0.114	0.006

Doping profiles and surface density of dopants (x is the depth into the wafer):

Diffusion with constant concentration of dopants at the surface	$C(x,t) = C_s \operatorname{erfc}\left(\frac{x}{2\sqrt{Dt}}\right)$	$Q = \frac{2\sqrt{Dt}}{\sqrt{\pi}}C_s$
Diffusion with constant number of dopants at the surface	$C(x,t) = \frac{Q}{\sqrt{\pi Dt}} \exp\left(-\frac{x^2}{4Dt}\right)$	Q
Doping with implantation	$C(x) = C_p \exp\left(-\frac{(x - R_p)^2}{2\Delta R_p^2}\right)$	$Q = \sqrt{2\pi} C_P \Delta R_P$

	( ZDINP )
Diffusion constant	$D = D_0 \exp\left(-\frac{E_a}{k_B T}\right)$
Mean free path of gas molecules	$\lambda = \frac{k_B T}{\sqrt{2}\sigma P}$
Thickness of spun photoresist	$T = K \frac{C \eta}{\omega^{\alpha}}$
Resolution in proximity and contact printing	$R = \frac{3}{2} \sqrt{\lambda \left(s + \frac{T_{pr}}{2}\right)}$
Angle between $\langle h_1 k_1 l_1 \rangle$ and $\langle h_2 k_2 l_2 \rangle$ directions	$\cos \theta = \frac{h_1 h_2 + k_1 k_2 + l_1 l_2}{\sqrt{(h_1^2 + k_1^2 + l_1^2)(h_2^2 + k_2^2 + l_2^2)}}$

Boltzmann constant $k_B = 1.38 \times 10^{-23} J/K$	$1eV = 1.6 \times 10^{-19} J$	
1 atmosphere = 760 Torr = 101 325 Pa	$1\mu m = 1000nm = 10^{-4}cm$	

The complimentary error function

	(		p	The compl	imentary e	rror function	on			
λ	0.05	0.10	0.15	0.20	0.25	0.30	0.35	0.40	0.45	0.50
$erfc(\lambda)$	9.44E-1	8.88E-1	8.32E-1	7.77E-1	7.24E-1	6.71E-1	6.21E-1	5.72E-1	5.25E-1	4.80E-1
λ	0.55	0.60	0.65	0.70	0.75	0.80	0.85	0.90	0.95	1.00
erfc(λ)	4.37E-1	3.96E-1	3.58E-1	3.22E-1	2.89E-1	2.58E-1	2.29E-1	2.03E-1	1.79E-1	1.57E-1
λ	1.05	1.10	1.15	1.20	1.25	1.30	1.35	1.40	1.45	1.50
erfc(λ)	1.38E-1	1.20E-1	1.04E-1	8.97E-2	7.71E-2	6.60E-2	5.62E-2	4.77E-2	4.03E-2	3.39E-2
λ	1.55	1.60	1.65	1.70	1.75	1.80	1.85	1.90	1.95	2.00
erfc(λ)	2.84E-2	2.37E-2	1.96E-2	1.62E-2	1.33E-2	1.09E-2	8.89E-3	7.21E-3	5.82E-3	4.68E-3
λ	2.05	2.10	2.15	2.20	2.25	2.30	2.35	2.40	2.45	2.50
erfc(λ)	3.74E-3	2.98E-3	2.36E-3	1.86E-3	1.46E-3	1.14E-3	8.89E-4	6.89E-4	5.31E-4	4.07E-4
λ	2.50	2.55	2.60	2.65	2.70	2.75	2.80	2.85	2.90	2.95
erfc(λ)	4.07E-4	3.11E-4	2.36E-4	1.78E-4	1.34E-4	1.01E-4	7.50E-5	5.57E-5	4.11E-5	3.02E-5

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Ouestion 1 (6+2+4+2+6 marks)  1-I. Name three applications of silicon dioxide  a) Chemical Protection  b) Sacrificial lay	>n	ion?	
c) Electrical isola	•		
1-II. E-beam lithography can be used to create not replaced UV lithography in microelectronic		Explain why	it has
Slow speed			<del></del>
a) Wille-in the crystal			iter ion
1-IV. What is the major advantage of lift-o deposition of a metal layer on a substrate? Men	tion one reason only.		
No metal etchin	g /etchant is	ne	eded
1-V. List three main applications of plasma in those applications.			
a) Hicking (RIE):	Activation o	f ch	emicals
b) Sputtering: Tox	bombardment	011	he target
o) De Position (PEC)	D): Lowering 1	he re	eaction

temperature.

#### Question 2 (10+10 marks)

We have an *n*-doped silicon wafer with a background concentration of  $2 \times 10^{16}$  dopants/cm<sup>3</sup>. It is required to have a junction depth of  $3\mu m$  at a Boron diffusion temperature of 1075°C. What is the required time under the following circumstances? For boron,  $E_a = 3.46eV$  and  $D_0 = 0.76$  cm<sup>2</sup>/s.

**2.I.** The wafer is doped using constant surface concentration method with  $10^{18}$  dopants/cm<sup>3</sup>.

Time = 
$$\frac{25.4}{0.00}$$
 Hours

 $C(9l) = C_5 \, erfc\left(\frac{9l}{2\sqrt{D}t}\right)$ 
 $2 \times 10^{16} = 10^{18} \, erfc\left(\frac{3x10^{-4}}{2\sqrt{D}t}\right) \Rightarrow erfc\left(\frac{2x10^{16}}{10^{18}}\right) = 1.65$ 
 $\rightarrow \sqrt{D}t = 9.09 \times 10^{-5} \rightarrow Dt = 8.26 \times 10^{-9} \, cm^2$ 
 $0 = 0.76 \, e^{-\frac{3.46 \times 1.6 \times 10^{-19}}{1.38 \times 10^{-23}(273 + 1075)} = 9.04 \times 10^{-14} \, cm^2$ 
 $t = \frac{8.26 \times 10^{-9}}{9.04 \times 10^{-14}} = 91371 \, sec = 25.44 \, krs$ 

**2.II.** The final doping profile is Gaussian with a surface concentration of  $10^{18}$  dopants/cm<sup>3</sup>.

Time = 
$$\frac{14 \cdot 4}{\sqrt{70}t}$$
 Hours

$$C(9) = \frac{Q}{\sqrt{70}t} = \frac{-912}{40t}$$

$$C(0) = \frac{Q}{\sqrt{70}t} = \frac{18}{-3 \times 10^{-4}} = \frac{-9 \times 10^{-8}}{40t}$$

$$C(4) = \frac{3 \times 10^{-4}}{\sqrt{70}t} = \frac{Q}{\sqrt{70}t} = \frac{-3 \times 10^{-4}}{40t} = \frac{2 \times 10^{-8}}{10^{-18}}$$

$$A = \frac{14 \cdot 4}{\sqrt{70}t} = \frac{Q}{\sqrt{70}t} = \frac{-9 \times 10^{-8}}{10^{-18}}$$

$$A = \frac{14 \cdot 4}{\sqrt{70}t} = \frac{Q}{\sqrt{70}t} = \frac{-9 \times 10^{-8}}{10^{-18}}$$

$$A = \frac{14 \cdot 4}{\sqrt{70}t} = \frac{-912}{40t}$$

$$A = \frac{18}{10^{-18}} = \frac{14 \cdot 4}{10^{-18}} = \frac{$$

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## Question 3 (5+8+7 marks)

We want to grow a layer of oxide on a bare <100> silicon wafer with only native oxide on it. Determine the oxide thickness under each of the following scenarios.

3-I. Wet oxidation at 1100°C for 2 hours.

tox: 94.5 nm

A = 0.182, B=0.330, T=0.010 tox + 0.182 tox = 0.530 (2+0.01) -) ton = 56.945 pm V

3-II. Wet oxidation for 2 hours followed by dry oxidation for 22 hours, both at 1000°C.

ton + Anton = Bu(++Tw)

(0.618) + 0.232 (0.618) = 0.010 (T+0.616)  $t_{\text{on}}^2 + 0.232 t_{\text{ox}} = 0.010 (22 + 0.616 + 51.91)$ 

- ton= {0.755 mm

/A=6.424 B=0.316 T=0.036

3-III. Dry oxidation for 22 hours followed by wet oxidation for 2 hours, both at 1000°C.

$$t_{00}: FF6 \text{ nm}$$

$$t_{00}: FF6 \text{ nm}$$

$$t_{00}: FF6 \text{ nm}$$

$$t_{00}: FF6 \text{ nm}$$

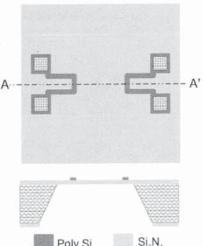
$$f_{00}: FF6 \text{ nm}$$

## Question 4 (20 marks)

Top view and cross section of a microstructure consisting of a released nitride membrane with poly silicon piezoresistors is shown to the right. Design the process flow using ONLY the following processes. Specify the orientation of the wafer that you need (i.e., <100> or <111>). If needed, use a process more than once.

- Front side lithography
- Back side lithography
- Photoresist stripping
- Thermal oxidation
- Thermal evaporation
- Sputtering
- Nitride CVD
- Nitride RIE
- Wet etching of Al

- Doping
- KOH etching
- XeF<sub>2</sub> etching
- HF etching
- · CVD of poly silicon
- Silicon DRIE
- Silicon RIE
- · Front side protection
- · Back side protection



Poly Si

Write down the processing steps for your fabrication flow in the order they need to be performed. Do NOT provide details on chemistry, etch rate, etc. Use as few steps as

needed for a reliable process. The lithography steps include all the required steps to transfer the pattern to the photoresist on the substrate (i.e., spinning, exposure, and development).

Number of masks:	Orientation of substrate:
Witche CUA	
Step 1: /////// CVV	
Step 2: (V) of Poly Si	
Step 3: Front side Lithe	(Mask 1)
Step 4: Silicon RIF	
Step 5: Photo yesist Strip	Ping
Step 6: Sputtering of Al	
Step 7: Front side litho(	Mask 2)
Step 8: Wet etching of A	4
Step 9: Photo resist strip	ping
Step 10: Front side Prote	ection
Step 11: Back Side litho	(Mask 3)
Step 12: Nitride RIE	
Step 13: Photo resist sti	Pping
Step 14: KoH lfching	
Step 15: Removal of Plon	tective layer (release)
Step 16:	
Step 17:	
Step 18:	

You can use this space to answer any question with proper reference or for your calculations.