Wednesday April 25, 2018

1. Name the following compounds (3  $\times$  8 = 24 pts)

(a)

3-amino-9,N-diethyl-7-hydroxy-N-isopropyl -4-(methylthio) dec-9-ene-2,6,8-trione

(b) O O H Br OH

4-bromo-5-hydroxy-6-oxocyclohex-3-ene-1,3-dicarbaldehyde

(c) Br MeO SH

4-bromo-3-methoxy-1-(2-methylbutyl)pyrrolidine-2-thiol

(d) CI O NO<sub>2</sub> CO NO<sub></sub>

7-amino-2-chloro-N-ethyl-N-methyl-6-nitro-4-oxooct-2-enedial

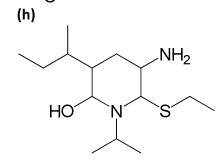
 $H_2N$  N O O

6-sec-butoxy-3-isopropyl-5-nitropyridin-2-amine

O NH<sub>2</sub>

3-benzyl-5-methoxy-2,6-dimethylnon-4-en-7-yne-1,4-diamine

6-amino-2-bromo-N-ethyl-4-formyl-3-isopropoxy-N-methylbenzoic acid



5-amino-3-sec-butyl-6-(ethylthio)-1-isopropylpiperidin-2-ol

2. Predict the major product(s) expected from the following reactions (3  $\times$  16 = 48 pts)

(a)

$$H_2NNH_2$$
 $H_3O^+$ 

(d)

$$= \underbrace{\frac{1. \operatorname{Sia_2BH}}{2. \operatorname{H_2O_2, NaOH}}}$$

3. H<sub>3</sub>O<sup>+</sup>, Hg<sub>2</sub><sup>+</sup>

(I)

H

$$\begin{array}{c}
1. & CI \\
2. \text{ LiAlH}_4 \\
3. \text{ H}_3\text{O}^+
\end{array}$$

Heat

$$\begin{array}{c}
\text{MCPBA} \\
\text{Heat}
\end{array}$$

(II)

$$\begin{array}{c}
\text{CI} & \text{Li} \\
\text{Cu} \\
\text{O} & \text{HCI}
\end{array}$$

$$\begin{array}{c}
\text{HCN} & \text{HO} & \text{CN} \\
\text{O} & \text{I. LiAlH}_4 \\
2. \text{H}_3\text{O}^+
\end{array}$$

(II)

$$\begin{array}{c}
\text{HCN} & \text{HO} & \text{CH}_2\text{NH}_2 \\
\text{O} & \text{I. NH}_2 \\
2. \text{ LiAlH}_4 \\
3. \text{ H}_3\text{O}^+
\end{array}$$
(II)

$$\begin{array}{c}
\text{HO} & \text{CH}_2\text{NH}_2 \\
\text{O} & \text{I. NH}_2 \\
\text{O} & \text{I. LiAlH}_4 \\
3. \text{ H}_3\text{O}^+
\end{array}$$
(II)

$$\begin{array}{c}
\text{HO} & \text{CH}_2\text{NH}_2 \\
\text{O} & \text{OH} & \text{I. NH}_2 \\
\text{OH} & \text{OH} & \text{OH} & \text{OH} \\
\text{O$$

3. Show how you would synthesize each of the following compounds from the given starting material(s). You **must** show all the intermediates to receive full credit  $(3 \times 6 = 18 \text{ pts})$ 

(a)

$$OH \xrightarrow{PCC} O \xrightarrow{1. NH_3} O \xrightarrow{1. NH_4} NH_2$$

$$3. H_3O^+$$

## (d)

$$\begin{array}{c|c} O \\ \hline \\ NH_2 \end{array} \begin{array}{c} Br_2 \\ \hline \\ NaOH \end{array} \begin{array}{c} NH_2 \\ \hline \\ OMe \end{array}$$

OMe 
$$\frac{1. \operatorname{Sia_2BH}}{2. \operatorname{H_2O_2}, \operatorname{NaOH}}$$
OMe 
$$\frac{1. \operatorname{LiAlH_4}}{2. \operatorname{H_3O^+}}$$
OMe 
$$\frac{1. \operatorname{LiAlH_4}}{2. \operatorname{H_3O^+}}$$

4. Propose a mechanism consistent with the following reactions (you must show all the intermediates to receive full credit)  $(3.5 \times 3 = 10.5 \text{ pts})$