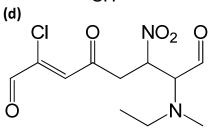
Wednesday April 25, 2018

1. Name the following compounds (3  $\times$  8 = 24 pts) (a)



2. Predict the major product(s) expected from the following reactions (3 x 16 = 48 pts) (a)

$$\begin{array}{c|c} & & & \\ \hline \\ & & \\ \hline \\ OMe \end{array} \begin{array}{c} & & \\ \hline \\ & &$$

(b) 
$$\frac{1. O_3}{2. CH_3SCH_3} \frac{H_2NNH_2}{H_3O^+}$$

(d)
$$\frac{1. \operatorname{Sia_2BH}}{2. \operatorname{H_2O_2}, \operatorname{NaOH}}$$

$$\frac{1. \operatorname{Sia_2BH}}{\operatorname{BuLi}}$$

(g)

$$\xrightarrow{\text{O} \\ \text{CI}} \xrightarrow{\text{NaHCO}_3}$$

(h)
$$\frac{\text{CN}}{2. \text{ H}_3\text{O}^+}$$

CN 1. LiAlH<sub>4</sub> 2. H<sub>3</sub>O<sup>+</sup>

(j)

$$\begin{array}{c|c}
 & HNO_3 \\
\hline
 & H_2SO_4
\end{array}$$

$$\frac{\text{H}_2\text{NOH}}{\text{H}_3\text{O}^{\dagger}}$$

(k)

## MCPBA → Heat

(m)

(n)

(p)

(q)

$$\begin{array}{ccc}
& & \text{Na}_2\text{Cr}_2\text{O}_7 \\
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3. Show how you would synthesize each of the following compounds from the given starting material(s). You **must** show all the intermediates to receive full credit  $(3 \times 6 = 18 \text{ pts})$ 

(a)

(c)

(d)

$$\begin{array}{c} \mathsf{H_2NH_2C} \\ & \mathsf{CH_2NH_2} \\ \\ & \mathsf{CH_2NH_2} \end{array}$$

(e)

(f)

OMe

4. Propose a mechanism consistent with the following reactions (you must show all the intermediates to receive full credit)  $(3.5 \times 3 = 10.5 \text{ pts})$ 

(a)

(b)