Tuesday November 28, 2017

1. Name the following compounds (3 \times 8 = 24 pts)

(a)

(c)

(d)

(e)

2. Predict the major product(s) expected from the following reactions (3 \times 16 = 48 pts) (a)

(b)

$$1. \frac{\text{MgBr}}{2. \text{H}_3\text{O}^+}$$

(c)

$$\frac{1. \text{LiAIH}_4}{2. \text{H}_3\text{O}^+}$$

(e)
$$CI \longrightarrow NH_3$$

(f)

$$\begin{array}{c}
O \\
CI \\
\hline
NaHCO_3
\end{array}$$

(g)

(h) OONH2 MeO OMe

$$\frac{CH_2N_2}{1}$$

(i)

$$\begin{array}{c|c}
& & \text{HNO}_3 \\
& & \xrightarrow{} \\
& & \text{H}_2\text{SO}_4
\end{array}$$

(j)

$$\begin{array}{c|c} \text{(k)} & & \\ & & \\ \hline & & \\ & & \\ \hline & & \\ & &$$

(I)

(m)

(n)
$$H_2NOH$$
 H_3O^+

(p)

$$= \underbrace{\frac{1. \operatorname{Sia_2BH}}{2. \operatorname{H_2O_2, NaOH}}}$$

(q)

$$\frac{\mathsf{H_2NNH_2}}{\mathsf{H_3O}^+}$$

3. Show how you would synthesize each of the following compounds from the given starting material(s). You must show all the intermediates to receive full credit ($3 \times 6 = 18 \text{ pts}$)

(a)

(b)

(d) N

(e)

(f)

$$\bigcirc$$

, N

4. Propose a mechanism consistent with the following reactions (you must show all the intermediates and arrows indicating the electron flow to receive full credit) $(3.5 \times 3 = 10.5 \text{ pts})$

(a)

(b)

(c)