## MM6010 - ASSIGNMENT 2

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## 1 Question 1

The Excess enthalpy of  $Co_{10}Cu_{30}Fe_{60-y}Ni_y$  system is calculated. The excess enthalpy of the binary systems are calculated as mentioned in the code snippet given below.

```
1 % functions to implement different binary systems
_2 function y = CoCu(x1,x2)
y = x1*x2*(39332 - 1356*(x1-x2) + 7953*(x1-x2)^2 - 1119*(x1-x2)^3);
function y = CoFe(x1, x3)
y = x1*x3*(-9312 - 1752*(x1-x3));
10 function y = CoNi(x1, x4)
y = x1*x4*(1331);
12 end
13
function y = CuFe(x2,x3)
y = x2*x3*(35626 - 1530*(x2-x3) + 12714*(x2-x3)^2 + 1177*(x2-x3)^3);
16 end
17
18 function y = CuNi(x2, x4)
19 y = x2*x4*(12049 - 1862*(x2-x4));
21
function y = FeNi(x3, x4)
y = x3*x4*(-18379 + 9228*(x3-x4));
```

The extrapolation is performed using different schemes as outlined in the code snippet given below.

```
1 % Muggianu Scheme
    _{2} beta = 1.0;
               for m = 1:no\_of\_comp-1
                                          for n = m+1: no of comp
                                                                   x_i = (1 + mole_frac(m) - mole_frac(n))/2;

x_j = (1 + mole_frac(n) - mole_frac(m))/2;
                                                                     f = mole_frac(m)*mole_frac(n)/(x_i*x_j);
                                                                     del\_H\_muggianu = del\_H\_muggianu + func(m,n,x\_i,x\_j)*f*\frac{beta}{};
   9
10 end
11 % Kohler Scheme
_{12} beta = 1.0;
for m = 1:no\_of\_comp-1
                                         for n = m+1: no of comp
14
                                                                     lamda = (mole frac(m) - mole frac(n))/(mole frac(m) + mole frac(n));
                                                                     x_i = ((1 + mole_frac(m) - mole_frac(n)) + lamda*(1 - mole_frac(m) - mole_frac(m)) + lamda*(1 - molefrac(m)) + lamda*(1 - mol
                                        mole_frac(n))/2;
                                                                   x_j = ((1 + mole_frac(n) - mole_frac(m)) + lamda*(1 - mole_frac(n) - mole_frac(n)) + lamda*(1 - mole
17
                                        mole_frac(m)))/2;
                                                                     f = mole_frac(m)*mole_frac(n)/(x_i*x_j);
18
                                                                     del_H_kohler = del_H_kohler + func(m,n,x_i,x_j)*f*beta;
19
                                         end
20
21 end
22 % Colinet Scheme
for m = 1:no of comp-1
```

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```
for n = m+1: no of comp
24
           lamda colinet = [1, -1];
25
           beta colinet = [0.5, 0.5];
26
27
           for o = 1:2
                x i = ((1 + mole frac(m) - mole frac(n)) + lamda colinet(o)*(1 - mole frac(n)))
      mole frac(m) - mole frac(n))/2;
               x_j = ((1 + mole_frac(n) - mole_frac(m)) + lamda_colinet(o)*(1 - mole_frac(m)))
29
      mole_frac(n) - mole_frac(m)))/2;
                f = mole_frac(m)*mole_frac(n)/(x_i*x_j);
30
               del_H_colinet = del_H_colinet + func(m,n,x_i,x_j)*f*beta_colinet(o);
31
           end
32
33
       end
34 end
```

The excess enthalpy is calculated for various atomic percentages of Ni. The plot of excess enthalpy is given in Fig. 1

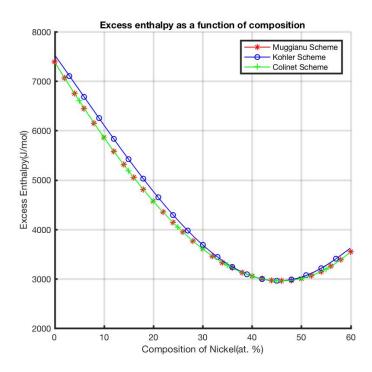


Figure 1: Excess enthalpy as function of composition of Ni

The calculation is repeated for equi-atomic composition and the output is given below.

```
Composition
Co=0.25
Cu=0.25
Fe=0.25
Ni=0.25
Excess Enthalpy in J/mol
through Muggianu scheme = 3790.4375
through Kohler scheme = 3790.4375
through Colinet scheme = 3790.4375
```

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