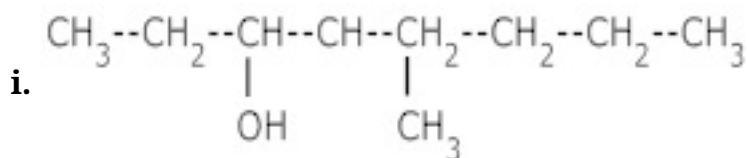


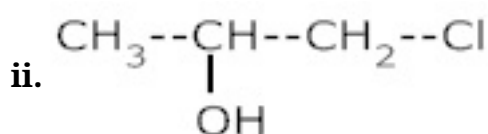
CBSE Class 12 physics
Important Questions
Chapter 11
Alcohols Phenols and Ethers

1 Mark Questions

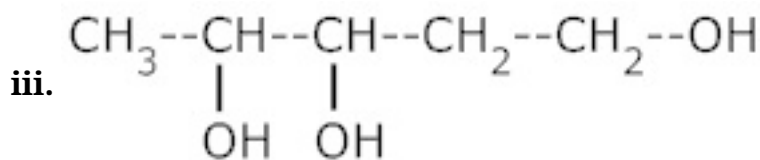
1. Write IUPAC names of :-



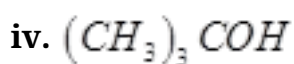
Ans. 5 – Methyloctan-3-ol



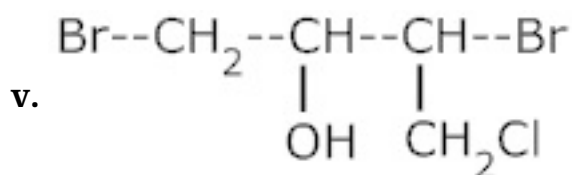
Ans. 1-Chloro propan-2-ol



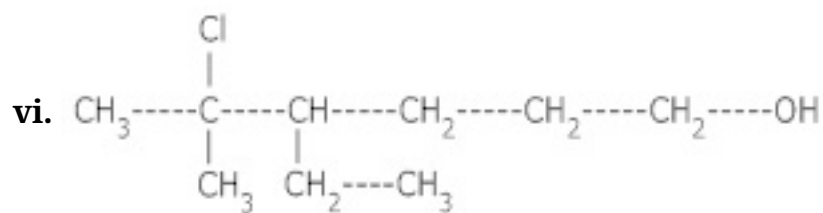
Ans. Pentan – 1,3,4 – triol



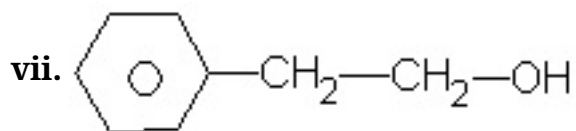
Ans. 2 – Methylpropan -2-ol



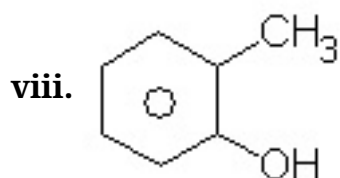
Ans. 1,3 – Dibromo – 4- chloro – 2- butanol



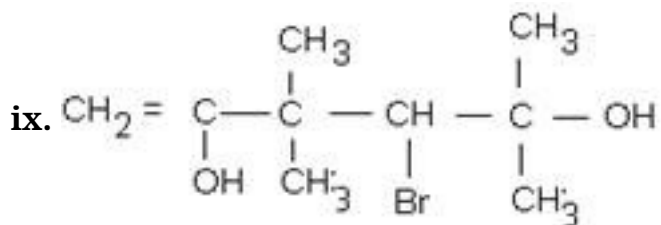
Ans. 5 - Chloro - 4 - ethyl - 5- methyl hexanol.



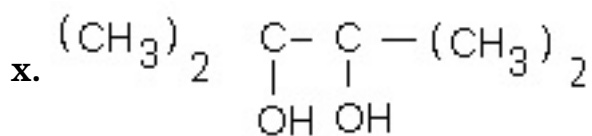
Ans. 2 - Phenyl ethanol



Ans. 2- Methyl phenol.



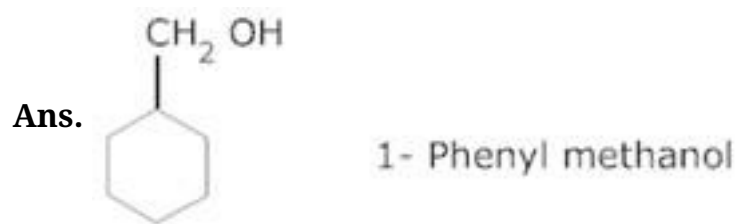
Ans. 4- Bromo -3, 3,5 - trimethyl - hex -1-ene- 2,5- diol



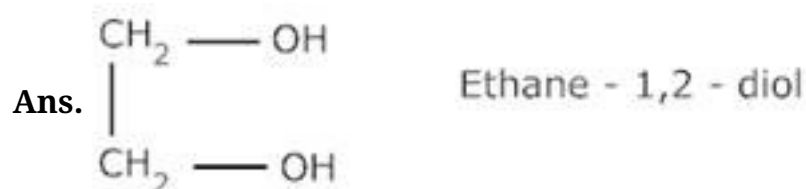
Ans. 2,3 - Dimethylbutan - 2,3 -diol

2. Write structural formula and give IUPAC names :-

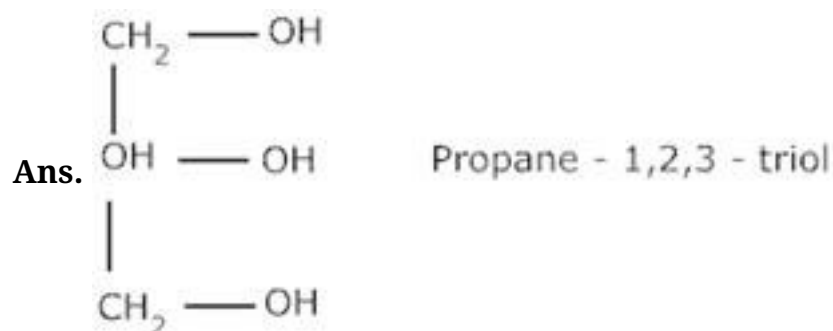
i. Benzyl Alcohol



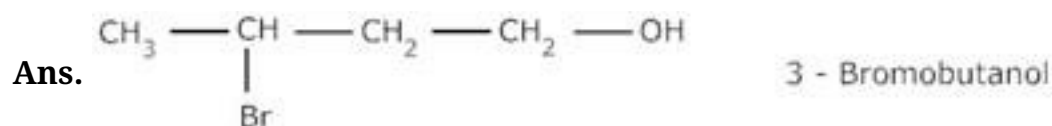
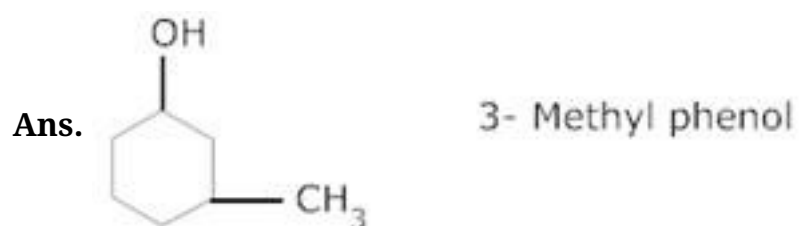
ii. Ethylene Glycol



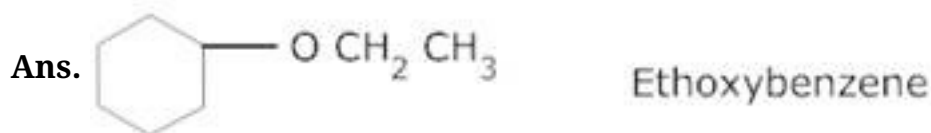
iii. Glycerol



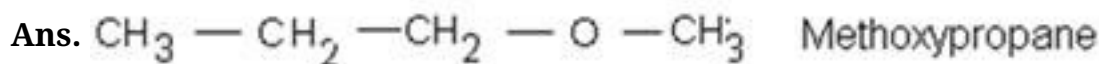
iv. m- cresol



vi. Ethylphenylether



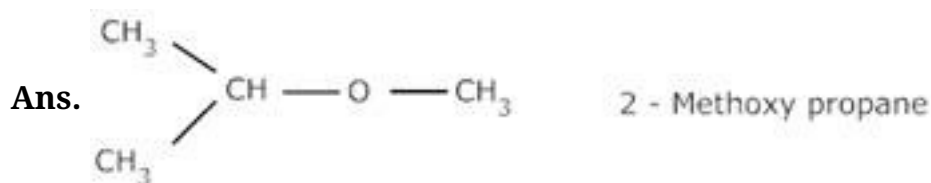
vii. Methylpropylether



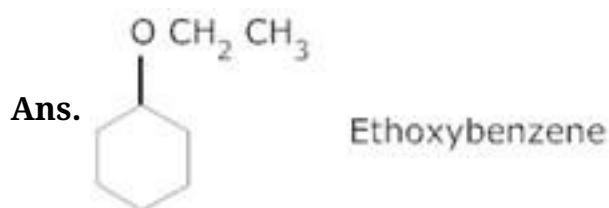
viii. Anisole



ix. Isopropyl methylether



x. Phenetole



3. What is denatured alcohol?

Ans. Industrial alcohol (ethyl alcohol) is made unfit for drinking purpose by addition of a small quantity of poisonous substance, methanol. This is called denatured alcohol. For denaturation, copper sulphate or pyridine can also be added.

4. Mention two important uses of methanol.

Ans. Uses of methanol:-

(i) It is used for denaturing alcohol.

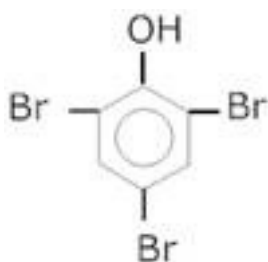
(ii) It is used as a solvent for paints & varnishes.

5. Arrange C_4H_5OH , C_2H_5OH & H_2O in the increasing order of acid strength.

Ans. The order is $C_2H_5OH < H_2O < C_4H_5OH$.

6. Give the structure of main product of action of excess of Br_2 on phenol.

Ans. The main product will be 2,4,6-tribromophenol.



7. What is wood spirit? Why is it so called?

Ans. Wood spirit is methanol. It is so called because earlier it was obtained from destructive distillation of wood only.

8. Identify allylic alcohols in the above examples.

Ans. The alcohols given in (ii) and (vi) are allylic alcohols.