

CBSE Class 12 physics Important Questions Chapter 10 Haloalkanes and Haloarenes

1 Marks Questions

1. Give IUPAC names of following compounds

(i).

(ii).

(iii).

(iv).

(v). $C_6H_5CH_2CH_2CI$



(vi).

(vii). $C_6H_6Cl_6$

(viii).

(ix).

(x).

Ans.(i). 1, 3- Dibromobutane

(ii). 1- Cholopropan-2-ol

(iii). 2, 3 – Dibromo-1-chloro-3-methylpentane

(iv). 2-Choloro-3-ethyl-2-methyl pentane



- (v). 1-Chloro-2-phenylethane
- (vi). 1-Chloro-1-phenyl ethane
- (vii). 1, 2, 3, 4, 5, 6- hexachlorocyclohexane
- (viii). 2, 2- Dihexyl 1, 1, 1-Trichloro ethane
- (ix). 4, 4-dibromobiphenyl
- (x). 1, 3-Dibromo -3- methyl butane
- (2). Give the structures of following.:
- (i). 1,3-Dichloro -2-(bromomethyl) propane
- (ii). Isobutylchloride
- (iii). Ortho bromotoluene
- (iv). 1 Bromo 4 chlorobutane
- (v). 3 Bromo 5 chloro 3,5 dimethyloctane
- (vi). 2,3 Dibromo 1 chloro -3- methylpentane
- (vii). 2 Chloro 3 ethyl -1, 4- pentadieme
- (viii). 2,3 Dibromo 1 chloro -3- methylpentane
- (ix). 2 Chloro 1 phemylpropane
- (x). Tert butylchloride
- Ans.(i)



(ii).

(iii).

(iv).

(v).

(vi).

(vii).



$$H_2C = CH - CH - C = CH_2$$

$$\begin{vmatrix} I & I \\ C_2H_5 & CI \end{vmatrix}$$

(viii).

(ix).

(x).

3. Convert

(i). 1 - Butene to 1 — chlorobutane.



(ii). Ethene to ethanol.

(iii). Chlorobenzene to phenol.

(iv). Methyl bromide to acetic acid.

(v). 2- chlorobutane to sec- butyl ethyl ether.

Ans.
$$CI$$
 CH_2
 CH_3
 CH_2
 CH_3
 CH_3



(vi). Chlorobenzene to benzyl chloride.

Ans.

(vii). Chlorobenzene to Benzene.

Ans.

(viii). Methane to Ethane.

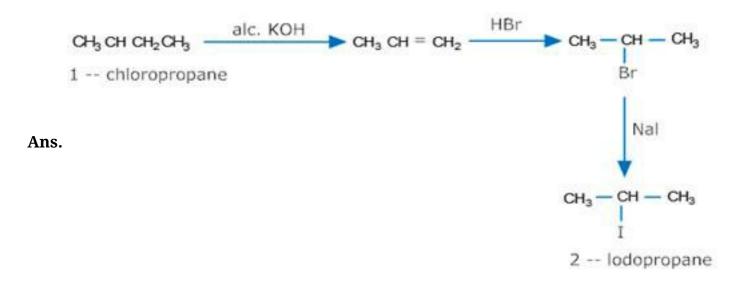
Ans. Methane

(ix). Benzene to o-chlorotoluene.

Ans.

(x). 1- chloropropane to 2- iodopropane.





4. What is lucas reagent?

Ans. A mixture of HCl and anhydrous $ZnCl_2$ is known as Lucas reagent.

5. Which of the following will show optical is omerism

1 – bromobutane or 2 – bromobutane?

$$CH_3CH_2CH_2Br$$
 CH_3CH CH_2CH_3 Ans. Br 1 -- bromobutane 2 -- bromobutane

2- bromobutane will be optically active as it has one chiral carbon $\left(C_{\scriptscriptstyle 2}\right)$.

6. Arrange $CH_3CH_2CH_2CH_2$ Br, $(CH_3)_3$ C Br, CH_3CH (CH_3) CH_2 Br in order of increasing boiling points.

Ans. The order of increasing boiling points is



Boiling point decreases with increase in branching as it reduces the surface area.

7. Give an example of

(a) Fittig reaction

(b) Finkelstein reaction.

Ans. (a) Fittig reaction

(b) Finkelstein reaction