

Roll No. ....

Total Pages : 3

BT-I/D-19

**31037**

CHEMISTRY

Paper : BS-101-A

Time : Three Hours]

[Maximum Marks : 75]

**Note :** Attempt *five* questions in all selecting at least *one* question from each unit. All questions carry equal marks.

**UNIT-I**

1. (a) Write salient features of Molecular orbital theory. Draw the molecular orbital energy level diagram of  $O_2$  molecule, and find its bond order. 8

(b) Explain the bond structure of solids on the basis of molecular orbital theory and also explain the concept of doping. 7

2. (a) What do you mean by Crystal field splitting ? Explain the splitting in octahedral field. On the basis of crystal field splitting explain the structure of  $[Co(NH_3)_6]^{3+}$ . 10

(b) Why conjugated butadiene is more stable than non-conjugated butadiene. 5

**UNIT-II**

3. (a) What is Molecular spectroscopy ? How does it differ from Atomic spectroscopy ? 3

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- ~~(b)~~ Explain the concept of Fluorescence and how it differs from Phosphorescence. Give important applications of Fluorescence in Medicine. 6
- ~~(c)~~ Write a short note on Diffraction. 6
4. (a) What do you mean by Electromagnetic spectrum ? Give basic principle of Electronic spectroscopy and also explain Frank-Condon principle. 10
- (b) "IR spectra is often characterised as molecular fingerprints." Justify the statement. 5

### UNIT-III

- ~~5.~~ (a) Write down the important applications of Nernst equation and also derive Nernst equation. 6
- (b) Differentiate between Gibb's free energy and Helmholtz free energy. which out of these two is called Work function, and why ? 4
- ~~(c)~~ Draw molecular geometries of  $\text{H}_2\text{O}$ ,  $\text{PCl}_5$  on the basis of Valence shell electron pair repulsion theory. 5
- ~~6.~~ (a) What is Concept of Entropy ? Find out the expression for entropy change when volume and temperature are the two variables. 6
- (b) What do you mean by Electronic configuration of element ? Explain the three principles used while writing electronic configuration. 6
- (c) Write about Fajan rules. 3

## **UNIT-IV**

7. (a) What are Structural isomers ? Write about position and chain isomers in detail giving examples. 6
- (b) Write a note on  $S_N1$  and  $S_N2$  reactions with examples. 6
- (c) Write Markownikoff's rule with example. 3
8. (a) What are ring opening type of organic reactions ? Explain. 5
- (b) Write a short note on Diastereomers. 5
- (c) Write a method for the synthesis of Paracetamol. Where this drug is used ? 5