

Q: For a given microstate, what is $P(U_2 = E) = P(E)$.

What is the statistical weight?

$\Rightarrow W(U_2) = 1$ (one microstate)

Use $T = \text{const}$, integrate

$$\frac{1}{k_B T} = \frac{\partial \ln W_1(U_1)}{\partial U_1} \Rightarrow \ln W_1(U_1) = \frac{U_1}{k_B T} + B \quad \swarrow \text{const}$$

$$W_1 = e^B e^{U_1/k_B T}$$

Important

$$\begin{aligned} W(E) &= W_1(U_1) \times W_2(U_2) = 1 \\ &= W_1(U_0 - E) \\ &= e^B e^{U_0/k_B T} e^{-E/k_B T} \end{aligned}$$

$$U_0 = U_1 + U_2 = E$$

$$U_1 = U_0 - E$$

$$P(E) \propto W(E) \propto e^{-E/k_B T}$$

Boltzmann distribution =

$$P(E) = \frac{1}{Z} e^{-E/k_B T}$$

needs to be in a state

$$\sum_i P(E_i) = 1 = \frac{1}{Z} \sum_i e^{-E_i/k_B T}$$

all microstates

Boltzmann factor

$$Z = \sum_i e^{-E_i/k_B T}$$

$Z = \text{partit}^o \text{ function}$

$$\begin{aligned} \frac{\partial Z}{\partial T} &= \sum_i \frac{\partial}{\partial T} e^{-E_i/k_B T} \\ &= \sum_i \frac{E_i}{k_B T^2} e^{-E_i/k_B T} \\ &= \frac{1}{k_B T^2} \sum_i E_i e^{-E_i/k_B T} \\ &= \frac{1}{k_B T^2} \sum_i E_i P(E_i) \end{aligned}$$

$$= \frac{ZU}{k_B T^2}$$

$$U = k_B T^2 \frac{1}{Z} \left(\frac{\partial Z}{\partial T} \right)_{V, N} = k_B T^2 \left(\frac{\partial \ln(Z)}{\partial T} \right)_{V, N}$$

$$\frac{\partial \ln(Z)}{\partial T} = \frac{1}{Z} \frac{\partial Z}{\partial T}$$

... for E ...

Helium $h\nu = 2$ The energy (1 Hz)

$$F = U - TS \quad \text{thermodynamics}$$

$$U = F + TS$$

$$= F - T \frac{\partial F}{\partial T} \quad S = \left(\frac{\partial F}{\partial T} \right)_{V, \mu}$$

$$= -T^2 \frac{\partial (F/T)}{\partial T} = k_B T^2 \frac{\partial \ln Z}{\partial T}$$

$$F = -k_B T \ln(Z)$$

$$S = k_B \ln(W)$$

$$W \quad S \therefore Z \therefore -F/T$$

const V const V

$$S = - \left(\frac{\partial F}{\partial T} \right)_{V, \mu} = k_B \frac{\partial \ln(Z)}{\partial T}$$

$$= k_B T \left(\frac{\partial \ln(Z)}{\partial T} \right)_{V, \mu} + k_B \ln(Z)$$

Pressure $P = - \left(\frac{\partial F}{\partial V} \right)_{T, \mu} = k_B T \left(\frac{\partial \ln(Z)}{\partial V} \right)_{T, \mu}$

Chem potential $\mu = \left(\frac{\partial F}{\partial \mu} \right)_{T, V} = k_B T \left(\frac{\partial \ln(Z)}{\partial \mu} \right)_{T, V}$

Two level systems.

Energy $\overset{\text{up}}{E_0} \text{ } 2 - \overset{\text{down}}{E_0}$ at const T
levels

Q. What is Z, U, C ? *down*

A: $Z = \sum_i e^{-E_i/k_B T} = e^{-E_0/k_B T} + e^{-E_0/k_B T}$ *up*

$$= 2 \cosh(E_0/k_B T)$$

$$\cosh(x) = \frac{e^x + e^{-x}}{2}$$

