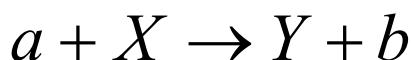


5. NUCLEAR REACTIONS

The aim of this section is to introduce:

- *the basic features of reactions;*
- *the concept of cross-section*
- *the centre of mass frame*
- *how to use this in fission and fusion reactions*

A typical reaction can be written as:

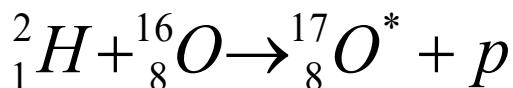


Examples:

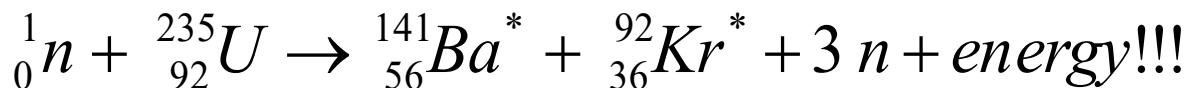


B.E. 64.75 28.3 39.25

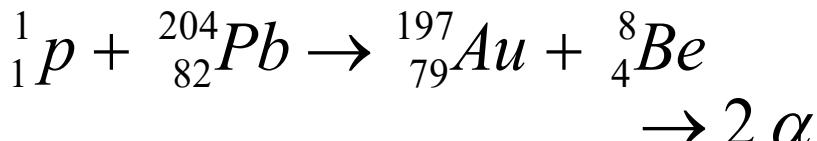
(used in reactor control, or in detecting neutrons ...)



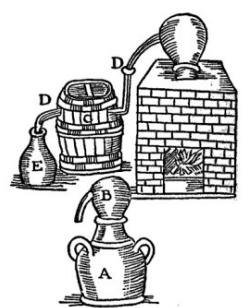
(deuteron stripping, creating excited states)



(an infamous reaction)



(another famous reaction ...)



Typical features:

(i) Usually, light nucleons (n or p) or light nucleus, incident on a heavier one

(ii) Energy released Q is:

$$Q = \text{total binding energy of products} - \text{total binding energy of reactants}$$

If $Q > 0$, the reaction is EXOTHERMIC

If $Q < 0$, the reaction is ENDOTHERMIC

(it requires energy to take place)
(the incident particle must have a threshold K.E. for the reaction to occur)

(iii) Strong interaction involved:

Nuclei must approach within ~ 1 fm to react ...

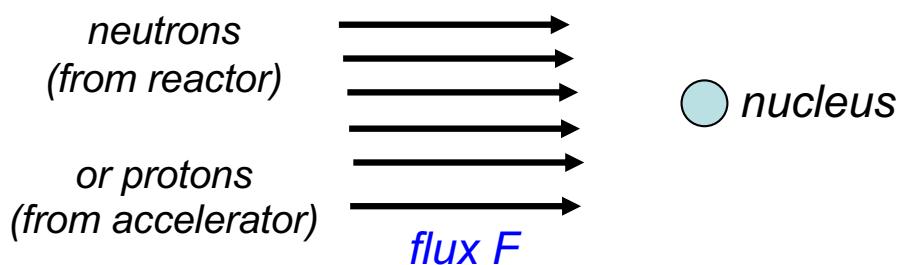
(iv) The strength of the reaction is measured by its cross-section σ

(v) We need to work in the “Centre of Mass” frame

5.1. CROSS-SECTION

The flux of incident particles is
the number of incident particles per unit area per unit time
(units: $m^{-2} s^{-1}$)

Let us consider a single nucleus



The number of particles interacting per unit time is proportional to the **flux F**.

We can write it as: $\sigma \times F$

σ is the **cross-section** (unit: m^2)

$\sigma \times F$ is the number of particles incident on an area σ per unit time: **σ is the effective cross-section of the nucleus.**

σ is not simply related to the cross-sectional area of the nucleus: it gives an idea of the magnitude ...

The cross-section area of an average nucleus is:

$$\approx \pi R_{nuc}^2$$

$$\approx \pi \times (1.1)^2 \times A^{\frac{2}{3}} \times 10^{-30} \text{ m}^2$$

For $^{238}_{92}U$: $\approx 1.45 \times 10^{-28} \text{ m}^2$

For $^{16}_8O$: $\approx 0.25 \times 10^{-28} \text{ m}^2$

Units: barns (1 barn = 10^{-28} m^2)

The cross-sections vary in range between

$$\sigma \sim 0 \text{ to } 10^4 \text{ barns}$$

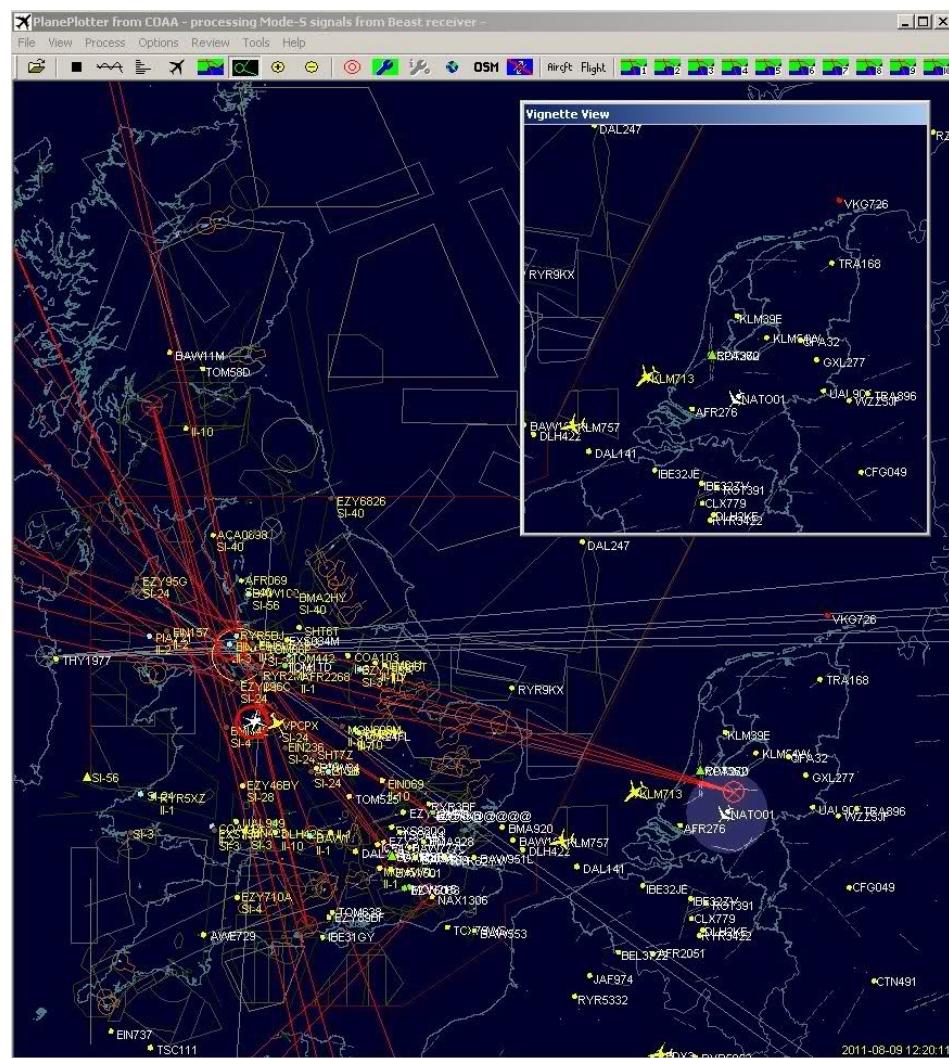
(this is much larger than the “physical” cross-section)

Analogy:

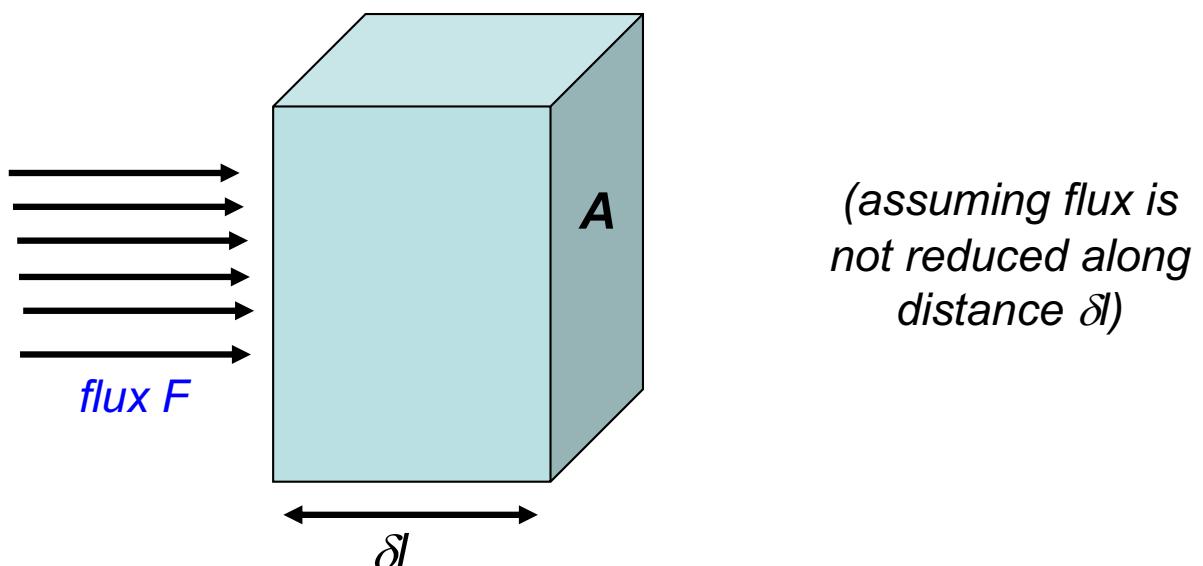
The radar cross-section of an object is related to how easily it interacts with electromagnetic waves:

typical bird	0.01 m^2
human	1 m^2
average car	100 m^2
B-2 stealth bomber	0.75 m^2

(all rather different from the physical cross-section)



For a target with ρ_{nuc} nuclei per unit volume:



The number of reactions per unit time is:

$$F \underbrace{\sigma}_{\text{nucleus}} \times \underbrace{\rho_{nuc} A \delta l}_{\text{no. of nuclei}}$$

(assuming single scattering)

The number of particles incident per unit time is:

$$F A$$

Therefore the probability of reacting in thickness δl is:

$$\rho_{nuc} \sigma \delta l = \frac{\delta l}{l_0}$$

l_0 is the mean free path of incident particles

For $\rho_{nuc} = 10^{29} \text{ m}^{-3}$

$$\sigma = 1 \text{ barn}$$

$$l_0 = 10 \text{ cm}$$

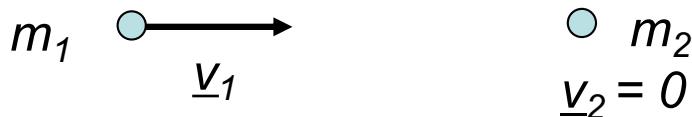
$$100 \text{ barns}$$

$$1 \text{ mm}$$

5.2. CENTRE OF MASS FRAME

We shall consider non-relativistic cases, with stationary targets.

Lab frame:



$$KE_{lab} = \frac{1}{2} m_1 v_1^2$$

Centre of Mass (CM) frame:

It moves with velocity v_{CM} so that:

$$m_1 (v_1 - v_{CM}) = m_2 v_{CM} \quad \text{i.e. the net momentum is 0}$$



Hence:

$$v_{CM} = \frac{m_1}{m_1 + m_2} v_1$$

$$KE_{CM} = \frac{1}{2} m_1 (v_1 - v_{CM})^2 + \frac{1}{2} m_2 v_{CM}^2$$

$$KE_{CM} = \frac{1}{2} m_1 (v_1 - v_{CM})^2 + \frac{1}{2} m_2 v_{CM}^2$$

Substituting for v_{CM} and rearranging, one gets:

$$KE_{CM} = \frac{1}{2} \frac{m_1 m_2}{m_1 + m_2} v_1^2$$

$$KE_{CM} = \frac{m_2}{m_1 + m_2} KE_{lab}$$

In the CM frame, there is no net momentum.
Thus, all KE can go into the reaction.

In the LAB frame, there is a net momentum, which must be conserved. All KE is not available.

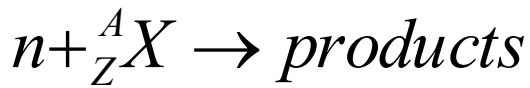
Only the KE in the CM frame is available for the reaction (e.g. to excite the target nucleus).

If Q is required to make an endothermic reaction occur, in the LAB frame, one must supply:

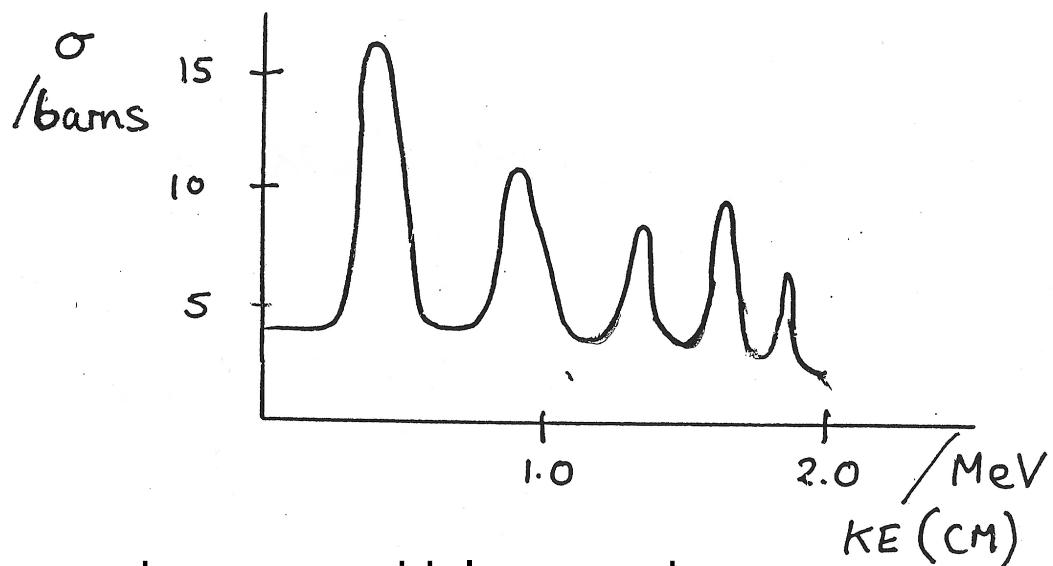
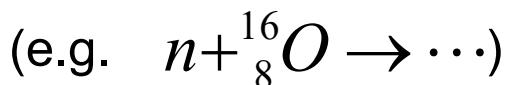
$$\frac{m_1 + m_2}{m_2} Q$$

This is why we fire light particles at heavy ones ...

5.3. NEUTRON REACTIONS

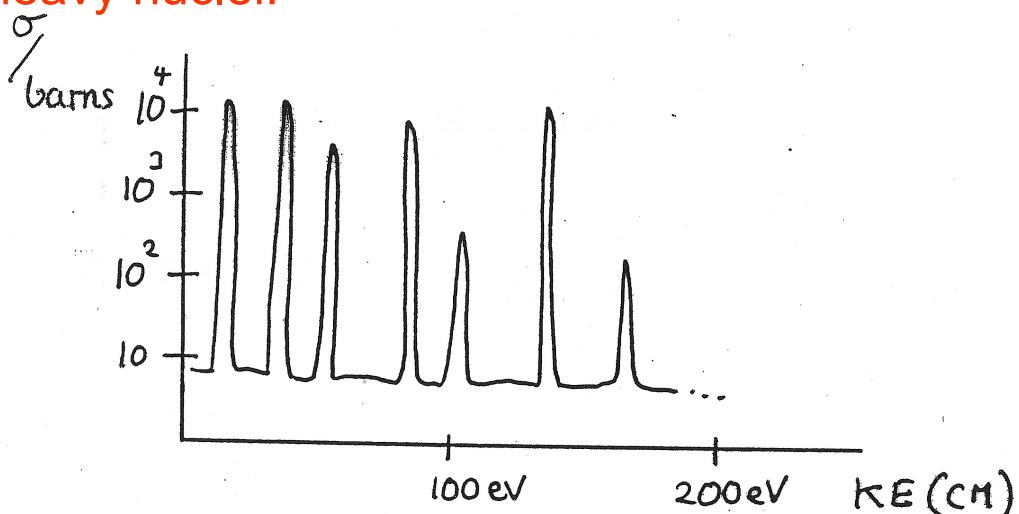


With light nuclei:



The peaks are:
widely spaced
broad
low (a few barns)

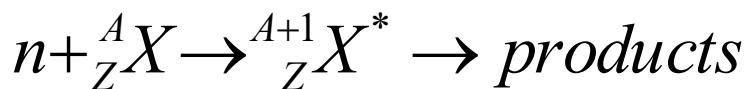
With heavy nuclei:



The peaks are:
closely spaced
narrow
high ($10^3 - 10^4$ barns)

Why do we have resonant peaks?

There is formation of a compound nucleus:



The neutron is absorbed into the nucleus.

Its energy is shared between nucleons.

∴ a while before any nucleon(s) have enough energy to escape

This is a two-stage process.

The decay mode carries little information about the formation process.

Resonance occurs when the initial KE (in the CM frame) matches the energy of the excited state of the compound nucleus ${}_{Z+1}^{A+1} X$

OFF RESONANCE

- energies do not match
- n doesn't enter nucleus (low probability)
- low σ

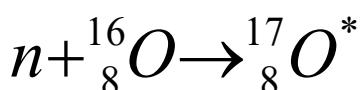
ON RESONANCE

- energies match (within ΔE)
- n enters target -> “long-lived” compound nucleus
- decay by breakup or γ -decay

Note: the binding energy of the neutron in the compound nucleus is available for the excitation of the target nucleus.

Typical values: 4 – 8 MeV ...

Let us look at a typical example:



We need the binding energies of each ...

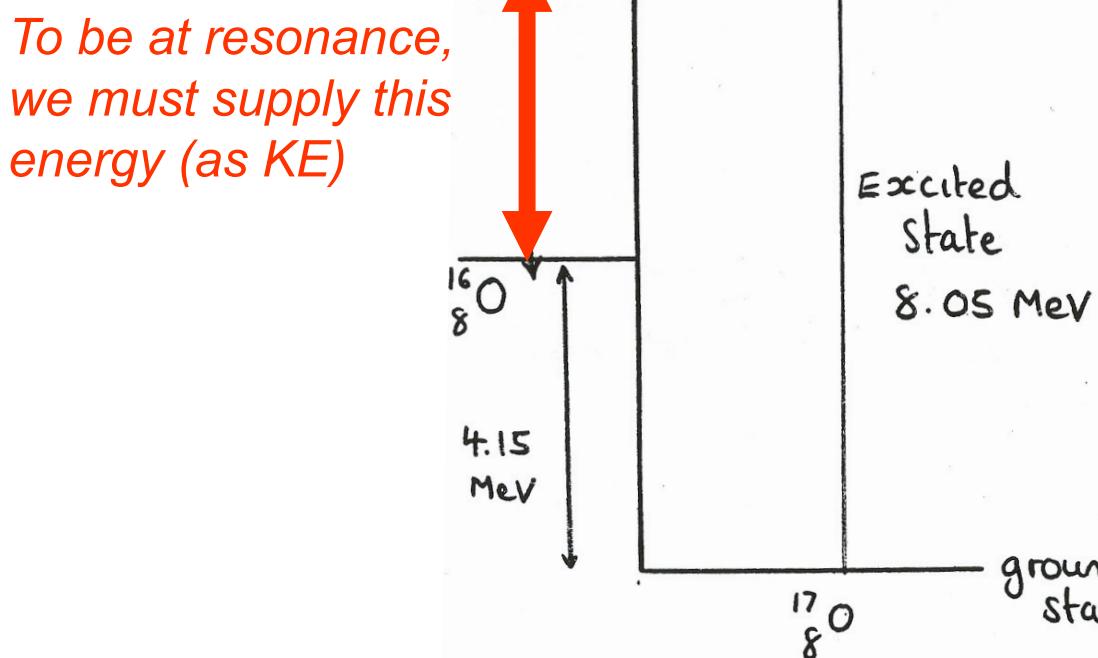
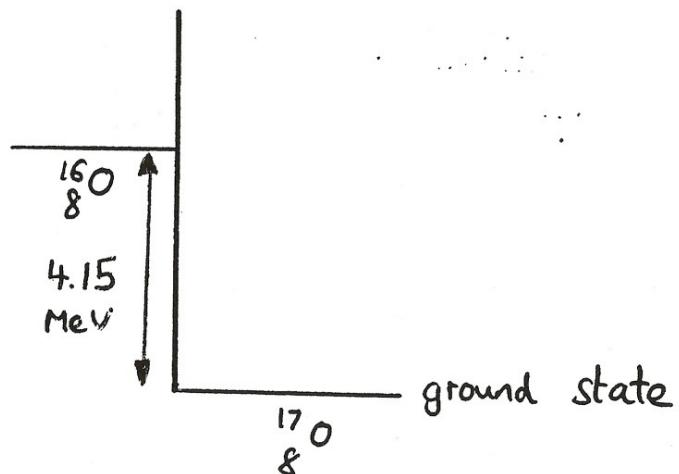
$^{16}_8O$ B.E. = 127.62 MeV

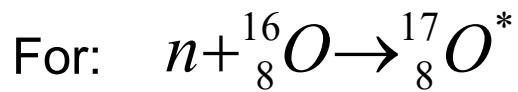
$^{17}_8O$ B.E. = 131.77 MeV (ground state)

$\therefore {}_{\frac{1}{8}}^{17}O$ is more stable than ${}_{\frac{1}{8}}^{16}O$

The difference is: $131.77 - 127.62 = 4.15$ MeV

We can use this value to draw the energy level diagram:





The resonance occurs when:

$4.15 \text{ MeV} + \text{KE(CM) of } n = \text{excited energy level of } {}_{8}^{17}O$



binding energy of last neutron in ${}_{8}^{17}O$

So, for 8.05 MeV excited state, we must supply:

$$\text{KE(CM)} = 8.05 - 4.15 = 3.9 \text{ MeV}$$

$$\begin{aligned} \text{In the LAB frame, } KE &= 3.9 \times \frac{M + m_n}{M} \\ &= 3.9 \times \frac{17}{16} = \underline{\underline{4.14 \text{ MeV}}} \end{aligned}$$

(here it is adequate to use A for mass ...)

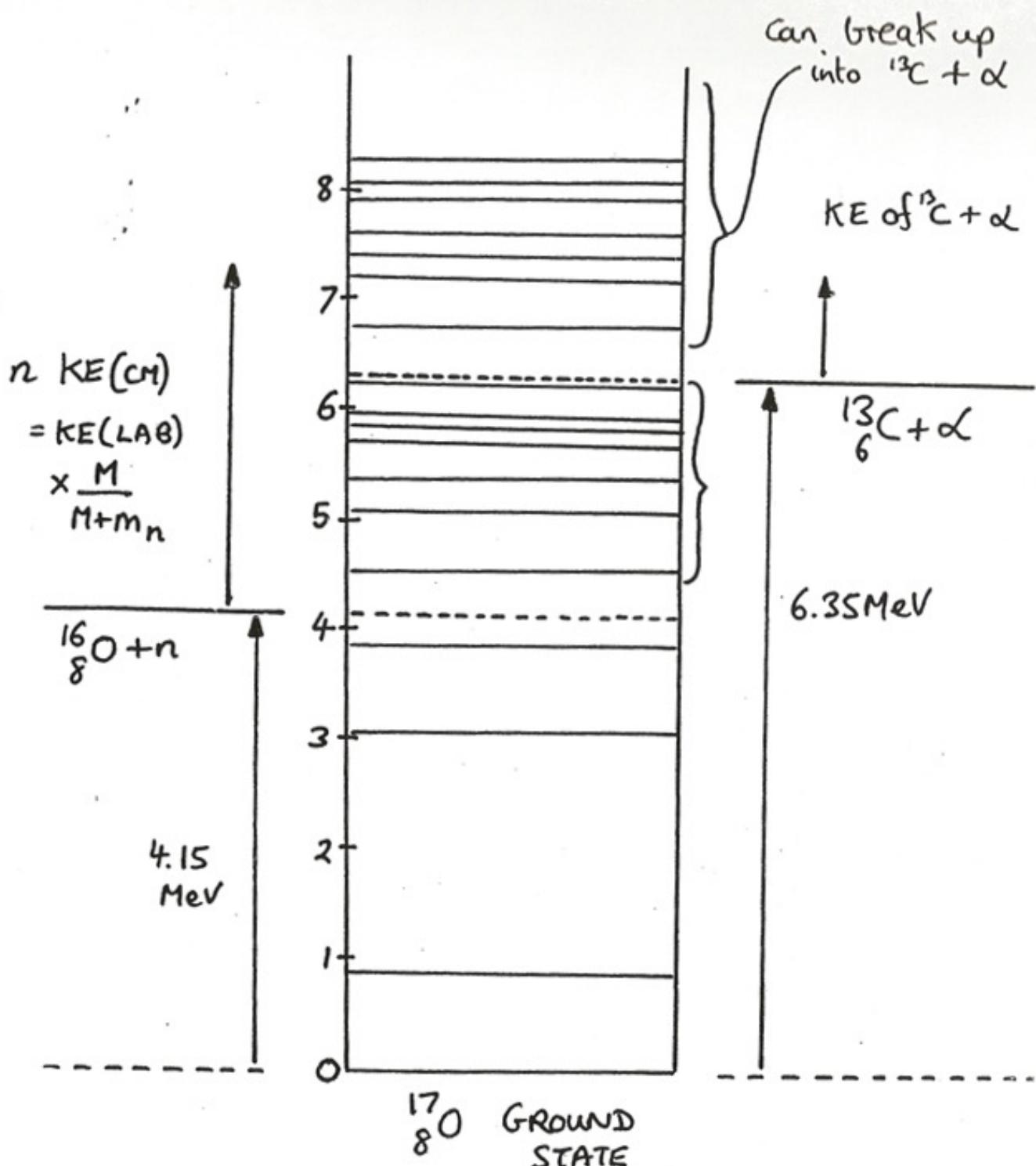


$Q = -2.2 \text{ MeV}$ (endothermic reaction)

Hence, if n has 4.14 MeV of KE(LAB), products will have:

$$4.14 - 2.2 = 1.94 \text{ MeV of KE(LAB)}$$

$n + {}^{16}_8 O$ Reaction

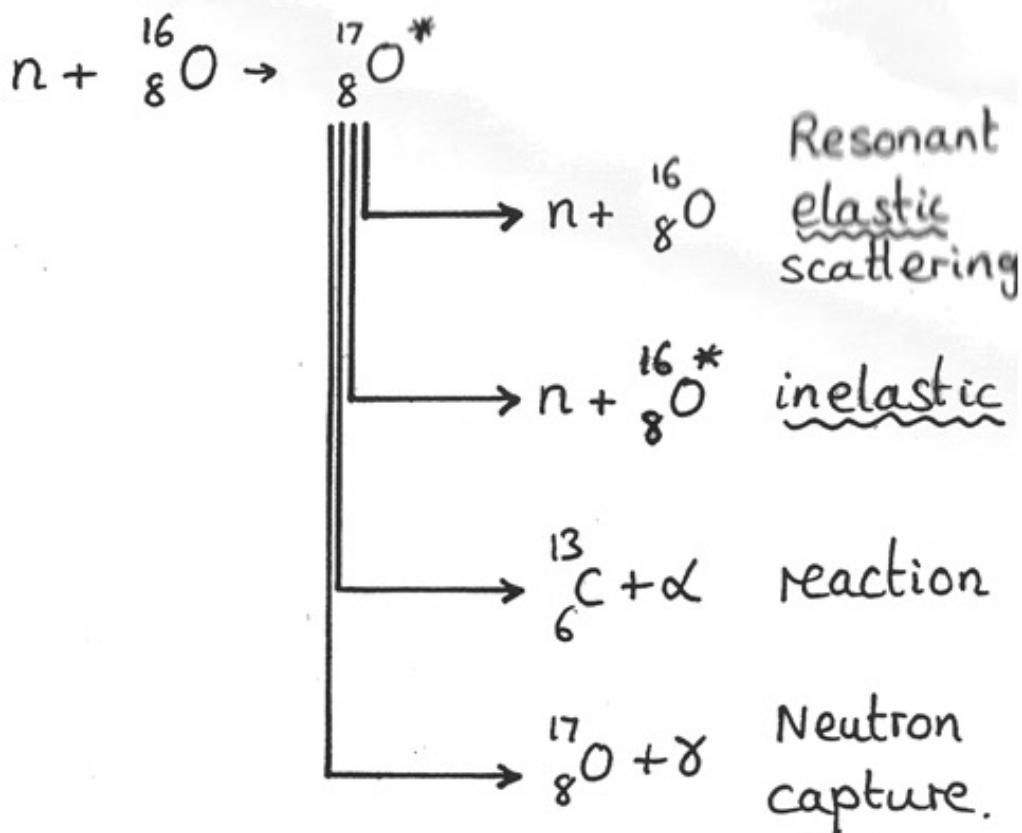


Stationary (no KE)
in CM frame.

Reaction products:

They are determined by the decay of intermediate ${}^{A+1}_Z X^*$

e.g.



Neutron capture is weak for low values of A (they break up faster) but it becomes important for high values of A .

5.4. CHARGED PARTICLE REACTIONS

Very similar to neutron reactions, except for the
Coulomb barrier (cf. α -decay)

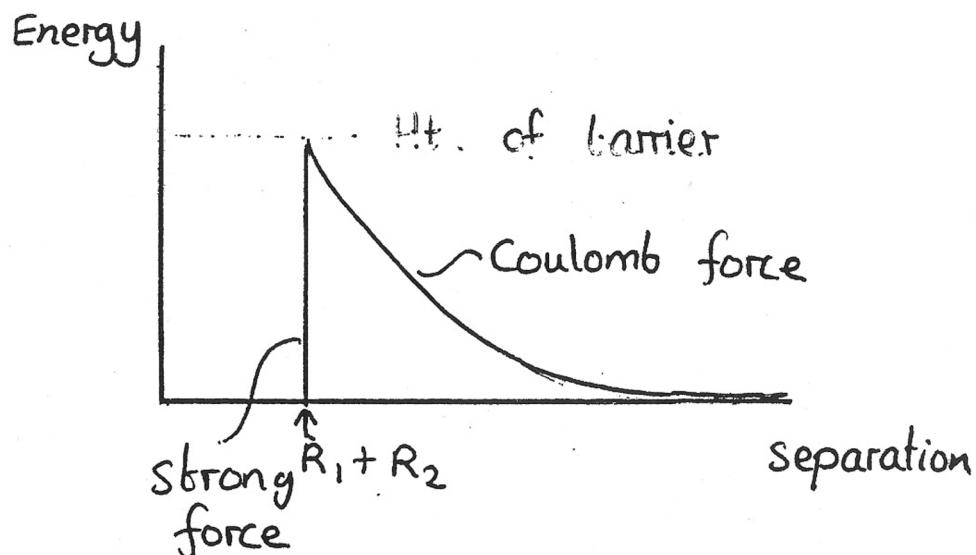
To experience the strong force, classically, nuclei must touch.

Energy supplied must be:

$$\frac{Z_1 Z_2 e^2}{4\pi \epsilon_0 (R_1 + R_2)} \approx \left(A_1^{1/3} + A_2^{1/3} \right) \times 1.3 \text{ MeV}$$

So, for example, for: ${}_{6}^{13}C + \alpha \rightarrow {}_{8}^{16}O + n$

the barrier is around 4 MeV

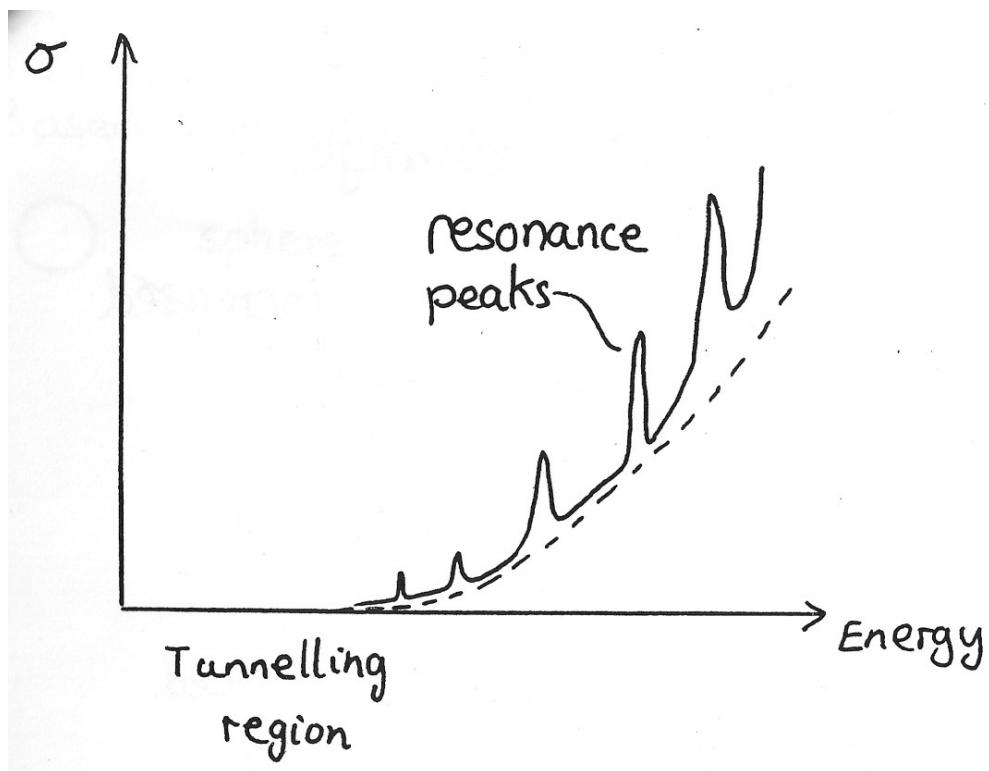


- If $KE \ll$ Barrier Height

- must tunnel to interact
- very small σ
- only light particles react

- If $KE \gg$ Barrier Height

- same as neutron reactions



↑ missed
lecture

5.5. NUCLEAR FISSION



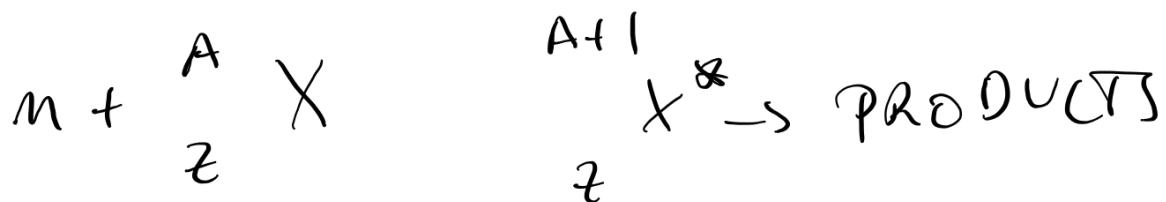
Two cases:

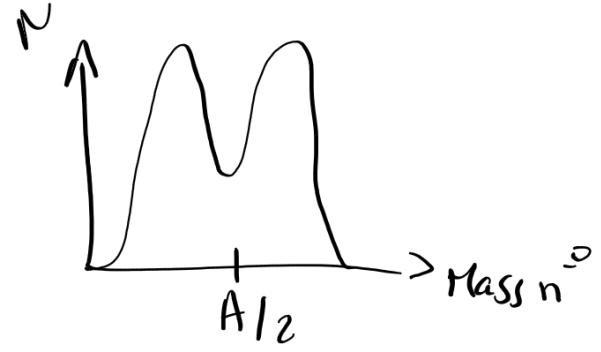
1. spontaneous fission

some heavy nuclei are unstable
(form of radioactive decay)

2. induced fission

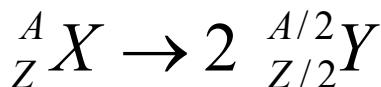
follows a nuclear reaction
(usually neutron + heavy nucleus)





Spontaneous Fission

Let us consider the symmetric case (rare, but easier):



The energy released is:

$$Q = 2B({}_{Z/2}^{A/2}) - B(A, Z)$$

The Liquid Drop Model gives the binding energy of a nucleus of mass number A , containing Z protons and N neutrons as:

$$B(Z, N) = a A - b A^{\frac{2}{3}} - s \frac{(N - Z)^2}{A} - d \frac{Z^2}{A^{\frac{1}{3}}} + \delta$$

where $a = 15.84$ MeV, $b = 18.33$ MeV, $s = 23.20$ MeV, $d = 0.714$ MeV.

Using the liquid drop model (and ignoring the pairing term δ)

bulk term

$$\left. \begin{aligned} & a A \\ & s \frac{(N - Z)^2}{A} \end{aligned} \right\}$$

$$\Delta A = 2a \left(\frac{A}{2} \right)$$

no influence

symmetry term

do not promote or inhibit

Only the surface and Coulomb terms change.

$$\frac{(N-Z)^2}{A} s = 2 \frac{s \left(\frac{N}{2} - \frac{Z}{2} \right)^2}{A/2}$$

- the surface energy is increased:

$$b A^{2/3} \Rightarrow 2 b \left(\frac{A}{2} \right)^{\frac{2}{3}} = 1.26 b A^{\frac{2}{3}}$$

~~This decreases B
i.e makes fission less likely~~

- the Coulomb energy is reduced:

$$\frac{d Z^2}{A^{1/3}} \Rightarrow \frac{2 d \left(\frac{Z^2}{4} \right)}{\left(\frac{A}{2} \right)^{\frac{1}{3}}} \Rightarrow 0.62 \frac{d Z^2}{A^{\frac{1}{3}}}$$

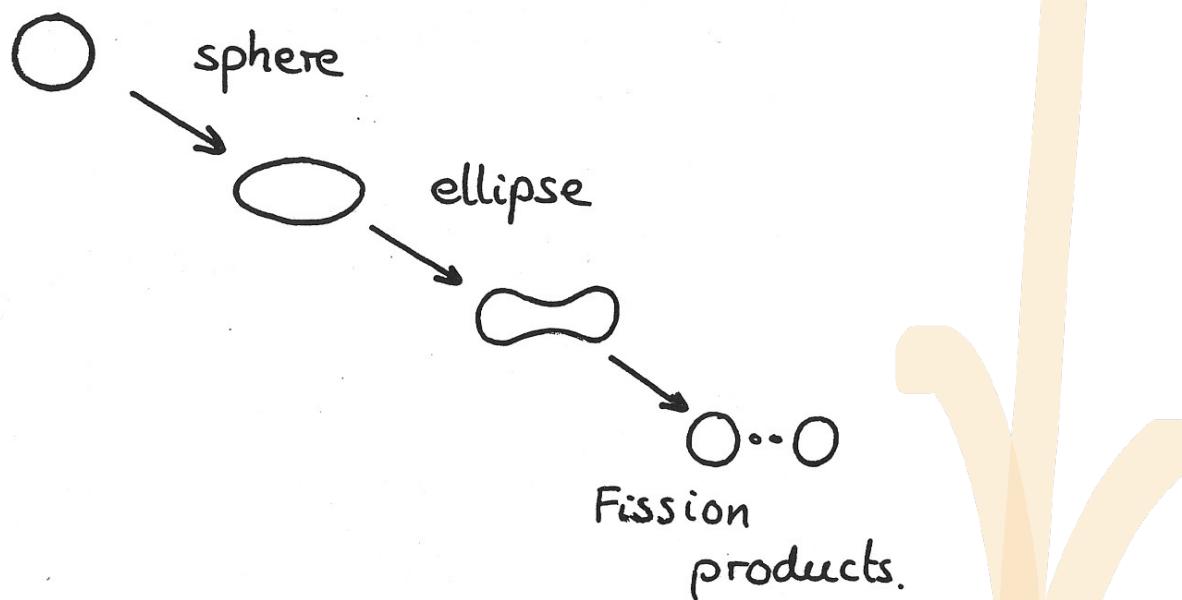
~~Up to P $\Rightarrow P^*$ + likely~~

Works out when $Q \geq 0$ (i.e. for $Z \geq 45$)

For larger values of Z , Q can be large ($\sim 100\text{-}200$ MeV)

Fission Model

Based on the idea of a liquid drop

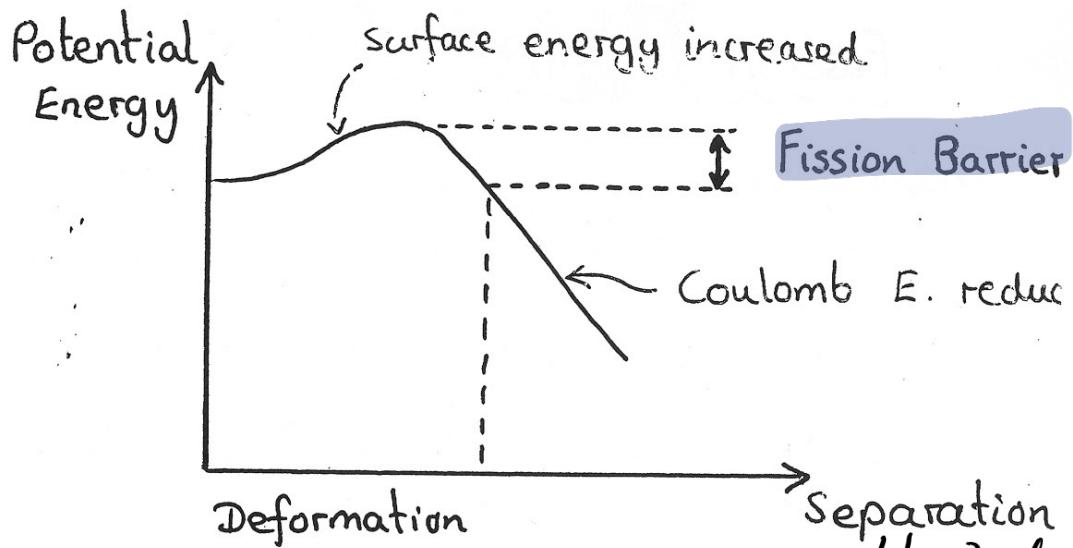


Fission yields:

an increase in surface energy

a decrease in Coulomb energy

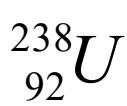
But only gain Coulomb energy when the products are separated from each other ...



The fission barrier is $\sim 5\text{-}6 \text{ MeV}$ for $A \sim 240$

$$\log +^{1/2}$$

Fission is a tunnelling process and lifetimes are long



α -decay
 $\sim 10^9$ years

fission
 $\sim 10^{16}$ years

For $Z \geq 144$, the barrier tends to 0.

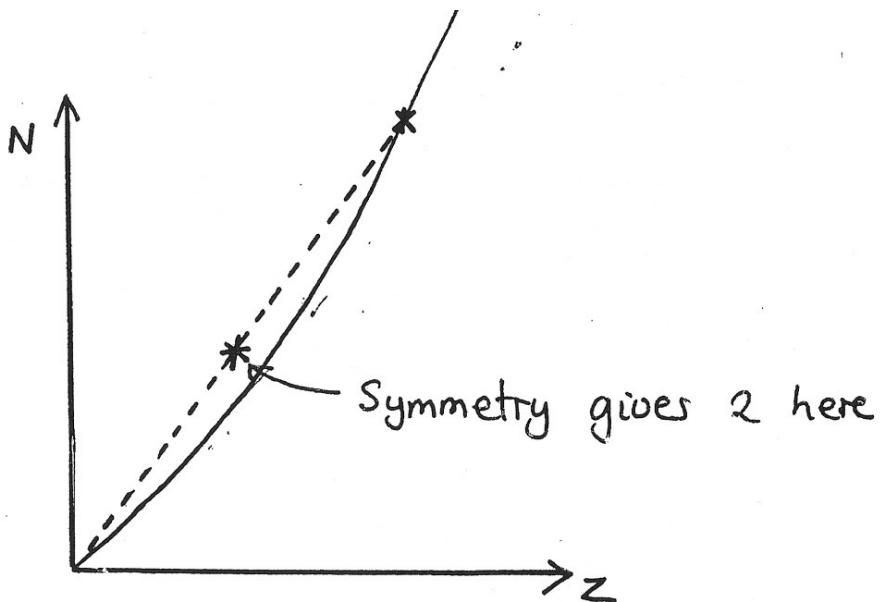
For $A \geq 260$, fission becomes the dominant decay mode.

Fission products

- fission jets

neutron rich

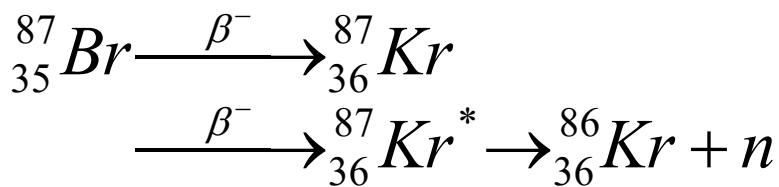
highly excited



Symmetric fission releases maximum energy.
Fragments usually asymmetric.

Following fission:

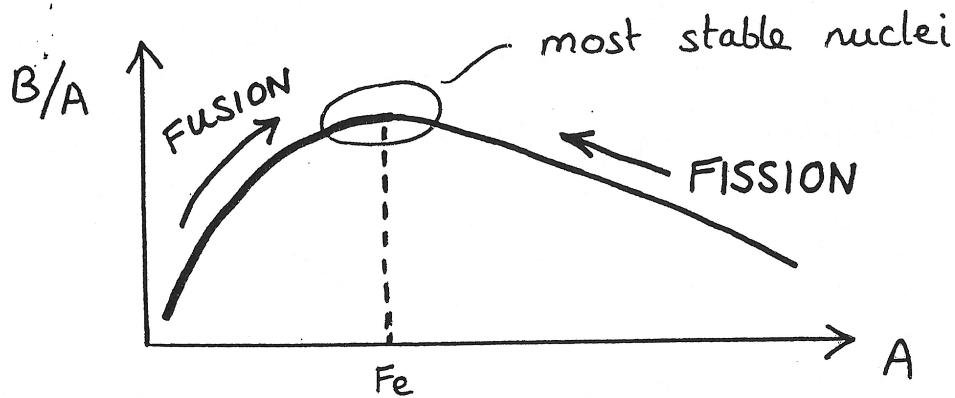
- “prompt” neutrons “boil off”
(fast: $\sim 10^{-20}$ s)
- γ -decay to ground states
(fast: $\sim 10^{-16}$ s)
- products are still n-rich: β -decay back to β -stability line
(slow: minutes/years)
- more ‘delayed neutrons’ sometimes emitted after β -decay
(e.g. important for control of nuclear power plants)



5.6. NUCLEAR FUSION

BIG EXPERT BY

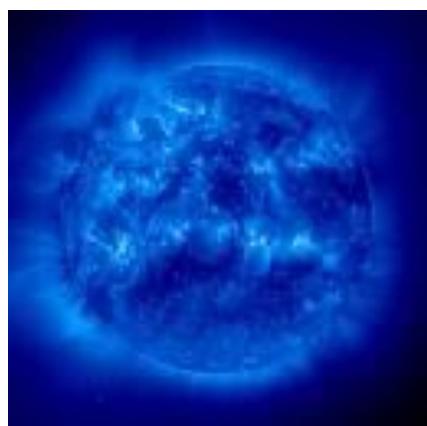
Let us consider the binding energy/nucleon curve:



Two light nuclei fusing can increase the total B.E.
∴ releases energy

This is important in:

stars (like the Sun)
fusion reactors

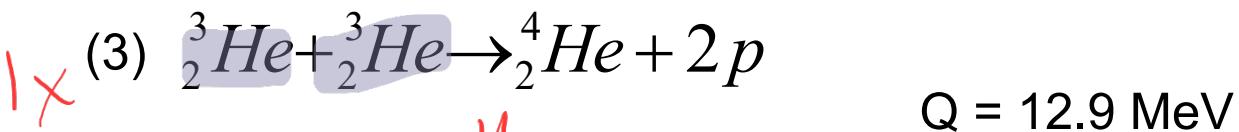
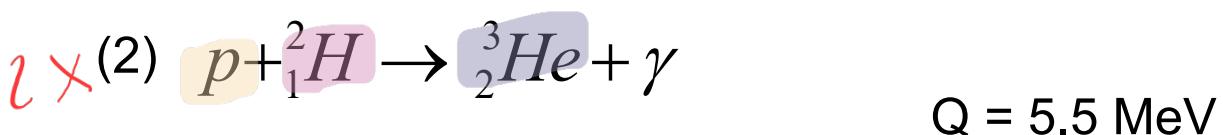
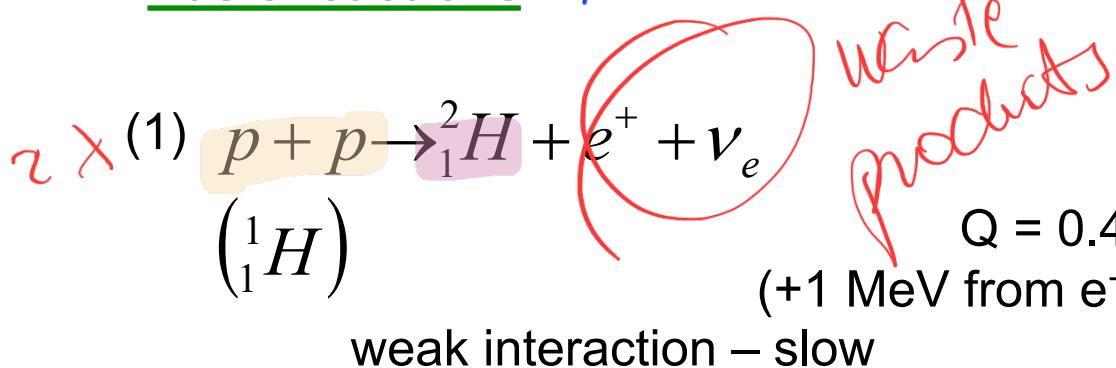


The solar energy flux at the surface of the Earth is 1.4 kW/m^2
Total output = $4 \times 10^{26} \text{ W}$

Where does this energy come from?

Basic reactions

2 protons → 1 goes through β decay
inputs



Q = 24.7 MeV
(+2 MeV)

This is the main hydrogen burning cycle in the Sun.
It is called the PPI chain
(there is also a PP-II, PP-III etc.)

But the protons/nuclei are charged: there is a Coulomb barrier to fusion. How much?

The barrier height is:

$$\frac{Z_1 Z_2 e^2}{4\pi \epsilon_0 (R_1 + R_2)} \approx \frac{1.44 Z_1 Z_2}{(R_1 + R_2)^2} \text{ MeV} \geq 1 \text{ MeV}$$

What energies do the particles have?

High temperature \Rightarrow nuclei are ionised

Gas \Rightarrow plasma (mix of nuclei and electrons)

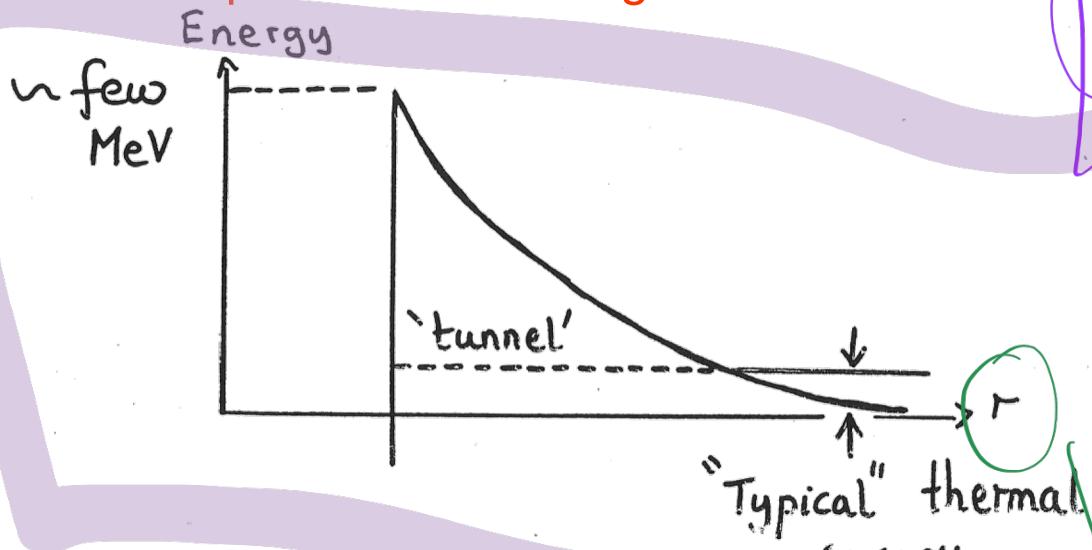
Distribution of energies = Maxwell-Boltzmann distribution

The average kinetic energy is: $\frac{3}{2} k T$

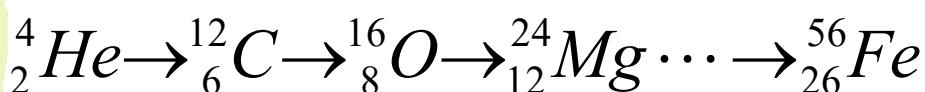
For the temperature of the Sun ($> 10^7$ K), this means that

typical KE \ll barrier height

➤ reactions require QM tunnelling



Fusion can continue (nucleosynthesis):



But barriers are higher, requiring significantly higher T
(that's why heaviest elements are only created in supernovae)

distance apart

$T \propto \sqrt[3]{E}$

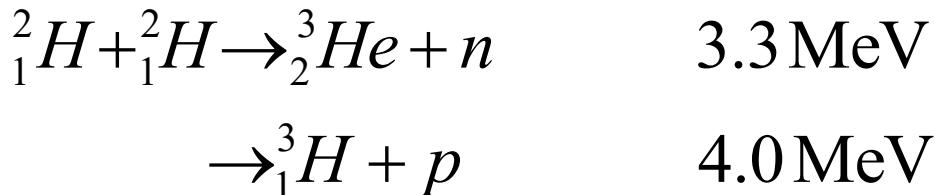
$$D = \frac{1}{(1 + e^{-x})}$$

FUSION REACTORS

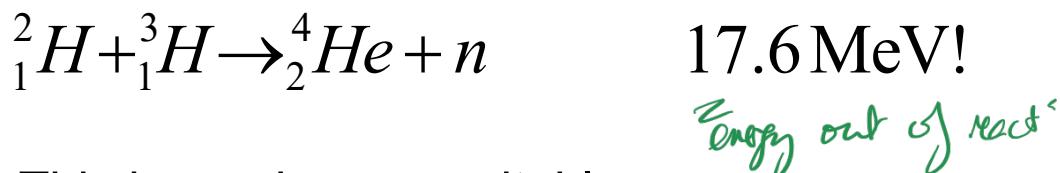
Can we generate fusion energy on Earth?

The PPI chain is not suitable (too slow)

One possibility: D-D reaction (D = Deuterium)



Even better: D-T reaction (T = Tritium)



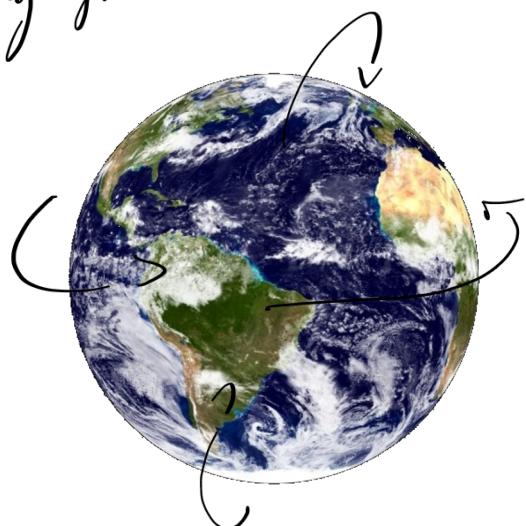
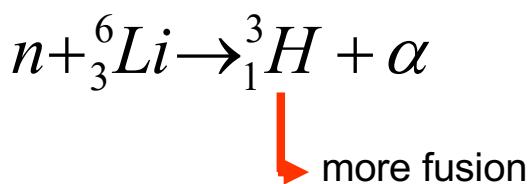
This is much more suitable:

- high Q
 - lower barrier
- continuously*
- } ↑ likely fusi'

Where do we get D from?

0.015% D in water

Where do we get T from?



Requirements

“Breakeven”

“Ignition”

energy out \cong energy in
reaction self-sustaining

↳ needed for energy product

(1) Plasma temperature must be high ($>10^8$ K)

(2) Plasma density n (ions/m³) must be high

(3) Must hold plasma for sufficient time τ ([confinement time](#))

For $T > 10^8$ K, i.e. $kT \sim 10$ keV, we need for breakeven

$$n \tau > 10^{20} \text{ s m}^{-3} \quad (\text{"Lawson criterion"})$$

This is higher still for ignition.

Confinement methods

- *Magnetic Confinement Fusion (MCF)*

Plasma confined in magnetic field (e.g. Tokamak).

It is held in a torus (e.g. Joint European Torus, 2.8 m Ø)

Maximum output: 16 MW

power out/power in = 65%

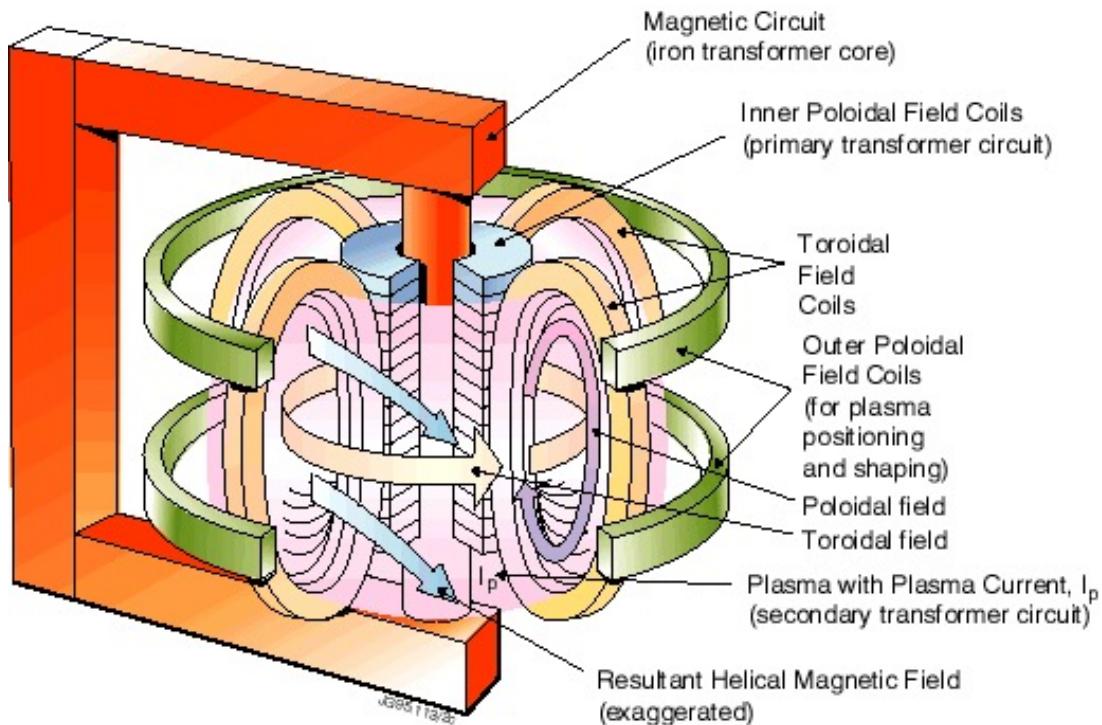
- *Inertial Confinement Fusion (ICF)*

DT mixture in plastic spheres (1 mm Ø)

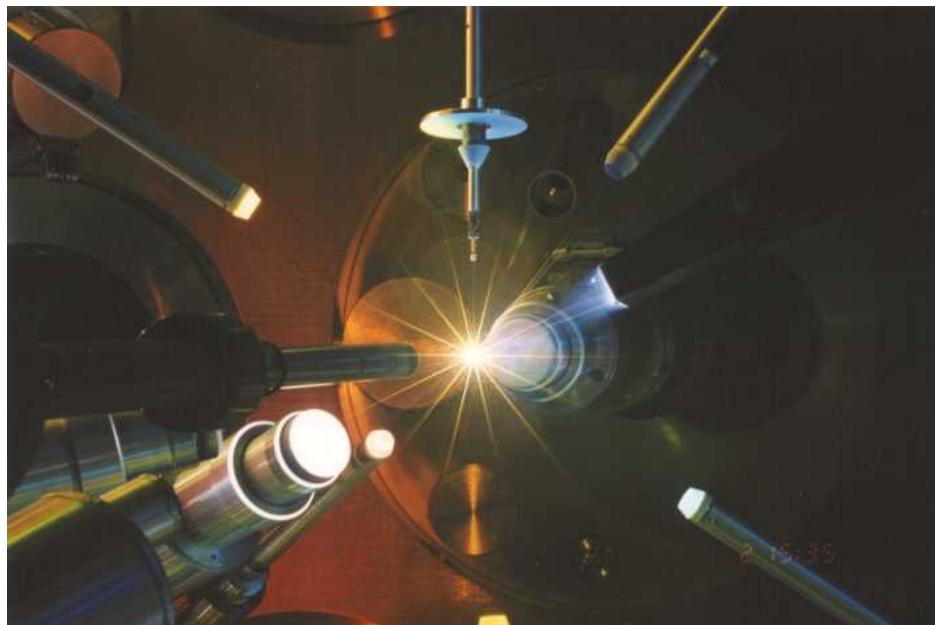
Vaporised by intense energy burst, applied symmetrically (with lasers or ion beams)

Pellet implosion generates $T \sim 10^8$ K and very high n

Confinement time $\sim 10^{-9}$ s



Typical Tokamak design



Typical ICF chamber

REACTION RATE

$$\text{Rate for 1 nucleus} = F \sigma$$

flux of incident nuclei *cross-section*

$$\text{Reaction rate per unit volume} = F \sigma \rho_1$$

number density of target nuclei

$$\text{For colliding beams of particles: } F = \rho_2 v$$

number density of incident nuclei *velocity of incident nuclei*

$$\therefore \text{reaction rate per unit volume} = \rho_1 \rho_2 \sigma v$$

We need to average over all possible relative velocities.

$$\text{Hence reaction rate per unit volume} = \rho_1 \rho_2 \overline{\sigma v}$$

$$\overline{\sigma v} = \int_0^{+\infty} P(v) \sigma(v) v \ dv$$

Where $P(v)dv$ is the probability of having a relative velocity between v and $v + dv$ (Maxwell-Boltzmann distribution)

$$\rho_1 \rho_2 \overline{\sigma v} \quad \text{for different nuclei}$$

$$\frac{1}{2} \rho_1^2 \overline{\sigma v} \quad \text{for identical nuclei}$$

(division by 2 in order not to count pairs twice)

$$\overline{\sigma v} = \int_0^{+\infty} P(v) \sigma(v) v \, dv$$

$\frac{mv^2}{2kT}$

tunnelling:
 exponentially decreases
 at low v

exponentially decreases
 at large v

