NMRlipids IV: Headgroup & glycerol backbone structures, and cation binding in bilayers with PS lipids

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Primarily measured but also simulated NMR order parameters will be collected also for other than phophatidylcholine (these are discussed in NMRlipids I) headgroup. The information will be used to understand structural differences between different lipid molecules in bilayers.

INTRODUCTION

Phosphatidylserine (PS) is the most common negatively charged lipid in eykaryotic membranes. PS lipids compose 8.5% of total lipid weight of erythrocytes, but the abundance varies between different organelles up to 25-35% in plasma membrane [1–3]. Despite of the relatively low abundance, PS lipids are important signaling molecules. They interact with signaling proteins [2], regulate surface charge and protein localization [4], and induce protein aggregation [5, 6]. Some domains spesifically interact PS lipids, while others are attracted by general electrostatics and the binding can be regulated by calcium [2]. Therefore, the structural details of lipid headgroups and the details of cation binding are crucial for the PS mediated signaling processes.

Previous experimental studies have concluded that PS head-groups have more rigid conformation than phophocholines (PC) due to hydrogen bonding or electrostatic interactions [7, 8]. Multivalent cations and Li⁺ are able to form strong dehydrated molecular complexes with PS lipids, while monovalent ions interact more weakly with PS containing bilayers [9–16]. Dilution of bilayers with PC lipids makes PS head-groups less rigid and reduces propensity for the formation of strong complexes with multivalent ions [7, 8, 15, 16].

The structure of PS lipid headgroups and their interactions with ions have been studied with various experimental methods and theoretical techniques [15, 17?]. However, the consesus has not been reached due to the difficulties to interpret the experimental data [?] and the inaccuracies in simulation models at the headgroup region [18–20]. Some studies propose that the negatively charged lipids attract cations only due to the increase of local concentration in the vicinity of membranes and that the binding constant of cations is similar to zwitterionic and negatively charged lipids [21?, 22]. On the other hand, some studies propose specific binding of calcium directly to PS lipid headgroups [23?, 24], The NMR data proposes that the PS headgroup is more rigid than PC,PE or PG headgroups, but more detailed interperetation has not been done.

Headgroup and glycerol backbone C-H bond order parameters calculated from MD simulations have been recently used to interpret the lipid structures in NMR experiments and to validate lipid structure and ion binding in simulations of PC

lipid bilayers [18–20, 25]. In this work we apply this approach to PS lipid headgroup in order to elucidate the structural details and ion binding to negatively charged lipids. The results are expected to elucidate also PS mediated signalling events because glycerol backbone and headgroup structure and behaviour are similar in model membranes and in bacteria [21, 26, 27].

METHODS

Solid state NMR experiments

The magnitude and signs of the C-H bond order parameters in headgroup and glycerol backbone were measured using natural abundance ¹³C solid state NMR spectroscopy as described previously [25, 28]. Shortly, the absolute values of the order parameters were determined from the dipolar splittings given by the indirect dimension of 2D R-PDFL experiment [29] and the signs were measured using S-DROSS experiments [30].

2.Details of the used spectrometer and maybe some other details should be given.

3.Sample preparation should be described.

4. How is the peak assignment done?

Molecular dynamics simulations

Molecular dynamics simulation data was collected using the Open Collaboration method [18]. The NMR-lipids project blog (nmrlipids.blogspot.fi) and the GitHub repository (github.com/NMRLipids/NMRlipidsIVotherHGs) were used as the communication platforms. The simulated systems are listed in Table II and simulation details are given in the SI. The simulation data is also indexed in the searchable database (nmrlipids.fi), and in the NMRlipids/MATCH GitHub repository (https://github.com/NMRLipids/MATCH).

The C-H bond order parameters were calculated directly from the definition

$$S_{\rm CH} = \frac{1}{2} \langle 3\cos^2 \theta - 1 \rangle,\tag{1}$$

TABLE I: List of MD simulations without additional salt. CKPM refers to the version with Berger/Chiu NH_3 charges compatible with Berger (i.e. the NH_3 group having the same charges as in the $N(CH_3)_3$ group of the PC lipids; 'M' stands for Mukhopadhyay after the first published Berger-based PS simulation that used these charges) and CKP refers to the version with more Gromos compatible version (i.e. the charges for the NH_3 group taken from the lysine side-chain).

lipid/counter-ions	s force field for lipids / ions	NaCl (mM)	$CaCl_{2}\left(mM\right)$	$^{a}\mathrm{N}_{\mathrm{l}}$	$^b\mathrm{N_w}$ 5.	$^c\mathrm{N_c}$	$^{d}T\left(K\right)$	$e_{t_{sim}(ns)}$	t _{anal} (ns)	g files
DOPS/Na ⁺	CHARMM36 [31]	0	0	128	4480	0	303	500	100	[32]
DOPS/Na+	CHARMM36ua [?] 6.	0	0	128	4480	0	303	500	100	[33]
DOPS/Na ⁺	Slipids [34]	0	0	128	4480	0	303	500	100	[35]
DOPS/Na ⁺	Slipids [34]	0	0	288	11232	0	303	200	100	[36]
DOPS/Na ⁺	Berger [37]	0	0	128	4480	0	303	500	100	[38]
DOPS/Na ⁺	GROMOS-CKP1 [?] 7.	0	0	128	4480	0	303	500	100	[39]
DOPS/Na ⁺	GROMOS-CKP2 [?] 8.	0	0	128	4480	0	303	500	100	[40]
DOPS/Na ⁺	lipid17 [41] / JC [42]	0	0	128	4480	0	303	600	100	[43]
DOPS/Na ⁺	lipid17 [41] / ff99 [44]	0	0	128	4480	0	303	600	100	[45]
POPS/Na ⁺	CHARMM36 [31]	0	0	128	4480	0	298	500	100	[46]
POPS/K ⁺	CHARMM36 [31]	0	0	128	4480	0	298	500	100	[47]
POPS/Na ⁺	CHARMM36ua [?] 9.	0	0	128	4480	0	298	500	100	[48]
POPS/Na ⁺	Slipids [34]	0	0	128	4480	0	298	500	100	[49]
POPS/Na ⁺	Berger [?]	0	0	128	4480	0	298	500	100	[50]
POPS/Na ⁺	MacRog [51]	0	0	?	??	0	?	?	?	[?] 10.
OPPS/Na+	MacRog [51]	0	0	128	5120	0	298	200	100	[52]
POPS/Na ⁺	GROMOS-CKPM [?] 11.	0	0	128	4480	0	298	500	100	[53]
POPS/Na ⁺	GROMOS-CKP [?] 12.	0	0	128	4480	0	298	500	100	[54]
POPS/Na ⁺	lipid17 [41] / JC [42]	0	0	128	4480	0	298	600	100	[55]
POPS/Na ⁺	lipid17 [41] / ff99 [44]	0	0	128	4480	0	298	600	100	[56]

^aNumber of lipid molecules with largest mole fraction

where θ is the angle between the C-H bond and the membrane normal. Angular brackets point to the average over all sampled configurations. 29.Error estimation should be discussed.

The number density profiles were calculated using *gmx density* tool from Gromacs sofware package [63].

RESULTS AND DISCUSSION

Headgroup and glycerol backbone order parameters measured from POPS lipid bilayer

The INEPT and 2D R-PDLF experiments from POPS sample give well resolved spectras for all the carbons in head-group and glycerol backbone region, except for g_3 for which the resolution was not sufficient to determine the numerical value of the order parameter (Fig. 1). Slices of the R-PDFL spectra (Fig. 1 C)) show a single splitting for the β -carbon with the order parameter value of 0.12, and a superposition of a large and a very small splitting for the α -carbon. The

larger splitting gives a order parameter value of 0.09, while the numerical value from the small splitting cannot resolved with the available resolution. Since only the absolute values of the PS headgroup order parameters were measured previously [7, 16], we used the S-DROSS experiment [?] to determine the signs of the order parameters. The S-DROSS slice for the β -carbon (Fig. 1 D)) clearly shows that the order parameter is negative, which is confirmed by SIMPSON simulations. The beginning of the S-DROSS slice suggests that the higher order parameter of the α -carbon is positive and the deviation towards negative values with the longer T_1 times suggests that the smaller order parameter is negative. This is confirmed by a SIMPSON simulation where the value of -0.02 was taken from ²H NMR experiment [16] for the smaller order parameter. The literature value was used because the resolution of our experiment was not sufficient to determine the small value of the order parameter. The S-DROSS curve from SIMPSON simulation with a positive value for the smaller order parameter (dashed grey in Fig. 1 D)) did not agree with the experiment, confirming the interpretation that the smaller order

^bNumber of water molecules

^cNumber of additional cations

 $[^]d$ Simulation temperature

^eTotal simulation time

fTime used for analysis

gReference for simulation files

TABLE II: List of MD simulations. The salt concentrations calculated as [salt]= $N_c \times [water]/N_w$, where [water] = 55.5 M. CKPM refers to the version with Berger/Chiu NH₃ charges compatible with Berger (i.e. the NH₃ group having the same charges as in the N(CH₃)₃ group of the PC lipids; 'M' stands for Mukhopadhyay after the first published Berger-based PS simulation that used these charges [?]) and CKP refers to the version with more Gromos compatible version (i.e. the charges for the NH₃ group taken from the lysine side-chain).

lipid/counter-ions	force field for lipids / ions	NaCl (mM)	$CaCl_{2}\left(mM\right)$	$^a\mathrm{N}_\mathrm{l}$	$^b\mathrm{N_w}$ 13.	c N _c	$^{d}T(K)$	e t _{sim} (ns) f
POPC:POPS (5:1)/K ⁺	CHARMM36 [31, 57]	0	0	110:22	4935	0	298	100
POPC:POPS (5:1)/K ⁺	CHARMM36 [31, 57]	0	0	250:50	?	0	298	200
POPC:POPS (5:1)/K ⁺	CHARMM36 [31, 57]	0	0	110:22	4620	0	298	500
POPC:POPS (5:1)/Na ⁺	CHARMM36 [31, 57]	0	0	110:22	4620	0	298	500
POPC:POPS (1:1)/K ⁺	CHARMM36 [31, 57]	0	0	150:150	?	0	298	200
POPC:POPS (5:1)	CHARMM36 [31, 57, 61]	0	150 17 .	250:50	?	?	298	200
POPC:POPS (5:1)	CHARMM36 [31, 57, 61]	0	1000 19.	250:50	?	?	298	200
POPC:POPS21. (5:1)/K ⁺	MacRog [51]	0	0	120:24	5760	0	298	200
POPC:POPS (5:1)/K ⁺	MacRog [51]	0	100	120:24	5760	10	298	200
POPC:POPS (5:1)/K ⁺	MacRog [51]	0	300	120:24	5760	31	298	200
POPC:POPS (5:1)/K ⁺	MacRog [51]	0	1000	120:24	5760	104	298	200
POPC:POPS (5:1)/K ⁺	MacRog [51]	0	3000	120:24	5760	311	298	200

27.MacRog simulations with KCl to be added

28.Berger simulations with NaCl and CaCl to be added

parameter is negative.

The headgroup and glycerol backbone order parameters of POPS measured in this work are in good agreement with the previously reported values from 2 H NMR experiments of DOPS [7] (Fig. 2). The β -carbon order parameter is significantly more negative and α -carbon experiences a significant forking in PS headgroup when compared with the values previously measured for POPC [28] (Fig. 2). These features have been intepreted to arise from a rigid PS headgroup conformation, stabilized by hydrogen bonds or electrostatic interactions [7, 8], but detailed structrural interpretation is not available.

Headgroup and glycerol backbone in simulations of PS lipid bilayers without additional ions

The headgroup order parameters of DOPS and POPS bilayers from different simulation models are compared with the experimental data in Fig. 3. Subjective ranking of the quality of the models is shown in Fig. 4. The tested models perform generally less well than in the previous study for PC headgroup [18] and none of the models reproduce the experimental order parameters within experimental error bars. Therefore, the models cannot be straightforwardly used to interpret the structure of PS headgroup. However, the differences between PC and PS headgroups are partially reproduced by some of the models.

The two best performing models for the α and β -carbons of PS, Slipids and CHARMM36, reproduce the large forking in the α -carbon and the Slipids model gives also a good agreement with experiments for the β -carbon order parameter in both PC and PS headgroups (Fig. 3 and Ref. 18). Interestingly, the dihedral angle distributions of CHARMM36 and Slipids in Fig. 16 share significant similarities in the headgroup region. However, the experimental order parameters in the glycerol backbone region are not well reproduced by the Slipids model, which was also the case in for PC lipids [18]. This difference probably arises from the differences in the dihedral angle distributions of C1-C2-C3-O31 and C2-C3-O31-C31 in Figure 15, which are also illustrated in Figure 5.

35.Also the discussion about POPS/OPPS issue with MacRog model should be added.

Counterion binding to lipid bilayers containing PS lipids

Membranes containing PS lipids are always accomppanied with counterions, which modulate electrostatic interactions between lipids and other biomolecules. Counterions are also suggested screen the repulsion between charged lipid headgroups in MD simulations and reduce the area per lipid of PS bilayers to be smaller than in PC bilayers [37, 64, 65]. The counterion density profiles along membrane normal show significant differences between simulation models (Fig. 6).

^aNumber of lipid molecules with largest mole fraction

 $[^]b$ Number of water molecules

^cNumber of additional cations

^dSimulation temperature

^eTotal simulation time

fTime used for analysis

^gReference for simulation files

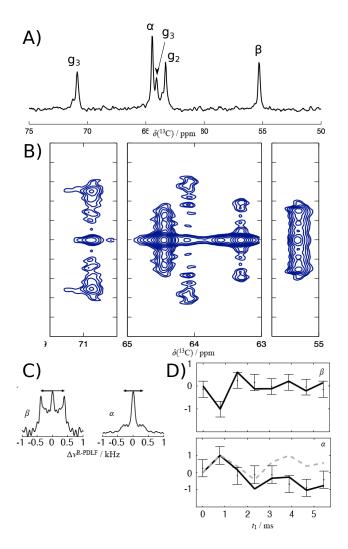


FIG. 1: (a) The headgroup region of the INEPT spectrum with headgroup and glycerol backbone carbons assigned. (b) 2D R-PDLF spectra for headgroup and glycerol backbone regions. (c) Slices for α and β barbons. (d) Experimental SDROSS data (points) and SIMP-SON simulations (lines). Order parameter values of -0.12 for the β -carbon, and 0.09 and -0.02 for the larger and smaller α -carbon slittings were used in the SIMPSON calculations. The S-DROSS curve from SIMPSON simulation with positive value for the smaller order parameter (dashed grey).

30.This is preliminary figure, should be polished. 31.Should we show slices for all the analyzed carbons in (c)?

The strongest counterion binding, i.e., the lowest concentrations in bulk water, are observed in MacRog, Berger and Lipid17/JC simulations. CHARMM36, CHARMM36ua and Gromos-CKP models exhibit two local maxima in counterion density, while a single maxima is observed in the other models. 36.More detailed discussion may be possible after comparing monovalent ion binding to bilayers between CHARMM simulations and experiments. Area per lipid is in agreement with experiments [66] only in the Gromos-CKP models, while other models give significantly lower values (Fig. 6). The difference cannot be explained by the electrostatic screening of the headgroup repulsion due

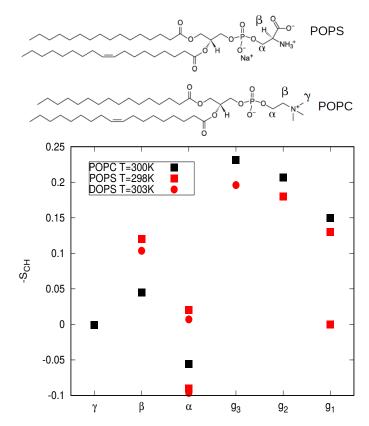


FIG. 2: Headgroup and glycerol backbone order parameters of POPS measured in this work compared with values for DOPS (²H NMR, 0.1M of NaCl) [7] and POPC (¹³C NMR) [28] from literature. Signs for PS order parameters as measured in this work and signs for PC as measured in Refs [25?].

32. There should be values in [15] which should be added.

to counterion binding because CHARMM36, CHARMM36ua and Slipid models give smaller area per lipid than Gromos-CKP models with similar counterion binding affinity.

To evaluate counterion binding in different simulation models against experimental data [15], we plot the headgroup order parameters measured from POPC:POPS 5:1 mixture as a function of different monovalent ions added to the buffer (Fig. 7). Experimental order parameter data for POPC headgroup in the mixture is available as a function of LiCl and KCl concentrations, while POPS headgroup order parameters are measured also as a function of NaCl. Lithium interacts more strongly with PS headgroups than other monovalent ions [11, 12, 14, 15, 67], as also observed for PC headgroups [68]. This is evident also in the changes of PS headgroup order parameters, which decrease with the addition of lithium but increase with the addition of sodium or potassium (Fig. 7). POPC headgroup order parameters exhibit a clear decrease as a function of LiCl concentration but only modest changes as a function of KCl concentration, indicating singificant Li⁺ binding but only weak Na⁺ binding to the mixture when interpreted using the electrometer concept [69-71]. In simulations with the Berger model, the headgroup order parameter response of POPC to the added NaCl is similar to the

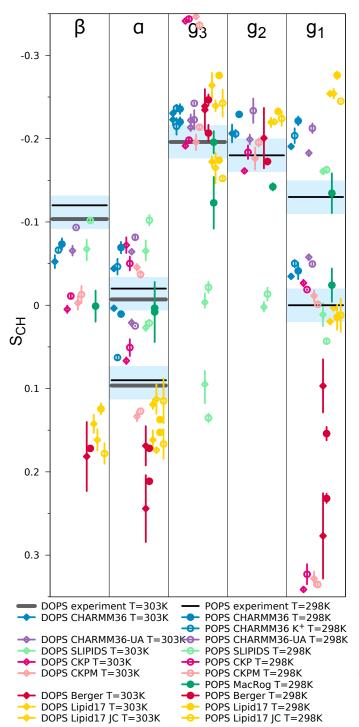


FIG. 3: Order parameters for PS headgroup and glycerol backbone from simulations with different models and experiments without CaCl₂. All DOPS data at 303 K, POPS at 298 K. Experimental data from [7] contain 0.1 M of NaCl. Signs are taken from experiments for POPS described in Supplementary Information. The vertical bars shown are not error bars, but demonstrate that we had at least two data sets; the ends of the bars mark the extreme values from the sets, and the dot marks their measurement-time-weighted average.

	β	α	g 3	g ₂	g 1	Σ
CHARMM 36	M	M F	М	М	M F	8
CHARMM 36-UA	M	М	М	М	M F	8
GROMOS- CKP1	M	M F	M		M	14
GROMOS- CKP2	M	M F	M F		M F	14
Slipid	M	М	M F	M	M F	14
Berger	M	M F	M F	М	M F	15

FIG. 4: Rough subjective ranking of force fields based on Figure 3. Here M indicates a magnitude problem, F a forking problem; letter size increases with problem severity. Color scheme: within experimental error (dark green), almost within experimental error (light green), clear deviation from experiments (light red), and major deviation from experiments (dark red). The Σ -column shows the total deviation of the force field, when individual carbons are given weights of 0 (matches experiment), 1, 2, and 4 (major deviation). For full details of the assessment, see Supplementary Information.

33.Issue about possible updates to this plot: https://github.com/NMRLipids/NMRlipidsIVotherHGs/issues/4 34.Lipid17 and MacRog results should be added into this plot.

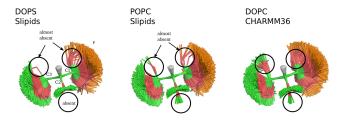


FIG. 5: Snapshots overlayed from different simulations for glycerol backbone region by Pavel Buslaev.

experiments of LiCl, indicating overestimated binding affinity of sodium, in line with the results for PC bilayers [19]. Indeed, the sodium density profile shows a significant binding peak in the Berger model (Fig. 8). Potassium binding in the MacRog simulation is significantly weaker (Fig. 8) and the headgroup order parameter changes are also in better agreement with simulations (Fig. 7). 37.Discussion about Lipid17 to be written when we have the density profiles. All the tested models overestimate the changes of POPS headgroup order parameters as a function of monovalent ions (Fig. 7), suggesting that model development is necessary to interpret the PS headgroup-ion interactions from MD simulations.

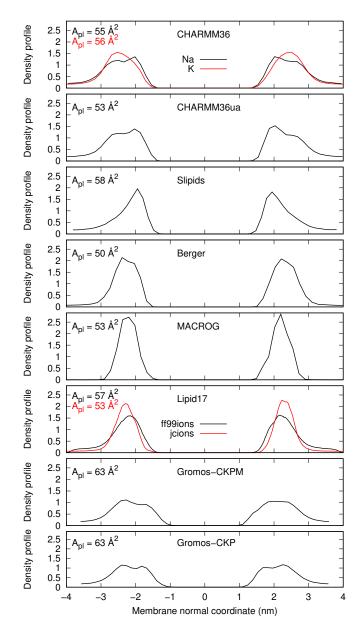


FIG. 6: Counterion densities of POPS lipid bilayer along the membrane normal from simulations with different force fields.

Headgroup structure in PS and PC mixtures

Dilution of PS lipid bilayers with PC lipids reduces propensity for the formation of strong complexes with multivalent ions and is proposed to make PS headgroups less rigid [7, 8, 15, 16]. Therefore, the intermolecular interactions at the headgroup region seems to determine important physical properties of mixed lipid bilayers. The headgroup order parameters of POPC increase in experiments with increasing amount of POPS in the mixed lipid bilayer (Fig. 9) [27]. This behaviour is generally observed when negatively charged lipids or sur-

factants are mixed with PC lipids [27, 72] and can be understood by the tilting of lipid headgroup more parallel to the membrane plane according to the electrometer concept [71]. The headgroup order parameters of PS lipids shift closer to zero when bilayer is diluted with PC lipids in experiments (Fig. 9) [7, 15, 27], which is intepreted to indicate reduced rigidity [7, 8].

The increase of POPC headgroup order parameters with the increasing amount of negatively charged POPS lipid is reproduced in MacRog simulations with potassium counterions, but not in Berger simulations with sodium or in CHARMM36 simulations with potassium or sodium conterions (Fig. 9). The observations can be explained using the electrometer concept. The Berger simulation exhibits very strong sodium binding (Fig. 8), which surpasses the effect of negatively charged lipids as the amount of counterions increase with increasing amount of PS. In CHARMM36 simulations, the counterion binding neutralizes the effect of PS and the headgroup order parameters are not changed with increasing amount of PS. Finally, the weak binding of potassium in the MacRog simulations enables the increase of order parameters with the added amount of PS (Figs. 9 and 8).

Also the headgroup order parameters of POPS mixed with varying amounts of POPC from simulations and experiments [15, 73] are shown in Fig. 9. The β -carbon order parameter of POPS slightly decreases and positive order parameter of α -carbon slightly increases with increasing amount of PS lipids. This may indicate increasing order of the headgroup. 43. The existing literature about this should be mentioned/discussed. It should be, however, noted that the experimental data for pure POPS and POPC/POPS mixture come from different experimental sets, 13C NMR in this work and 2H NMR from Ref. 15, respectively. Therefore the accuracy of the order parameter change is not as high as typically in the measurements of order parameter changes, see discussion about qualitative and quantitative accuracy in Ref. 20. The changes of PS headgroup order parameters are not reproduced by the tested simulation models. The β -carbon order parameter increase with increasing amount of PS in CHARMM36 and MacRog simulations in contrast to the experimental data. The smaller α carbon order parameter increase in both simulation models with increasing amount of PS, while it is almost unchanged in experiments. The larger α -carbon order parameter increase in MacRog and decrease in CHARMM36 with increasing amount of PS, both model exhibiting a poor agreement with experiments. Signifincant improvement in the MD simulation models are needed to interpret the PS headgroup structures and mutual interactions with PC lipids. 44.This discussion should be updated.

Ca²⁺ binding affinity in bilayers with negatively charged PS lipids

The headgroup order parameters of PC lipids decrease proportionally to the bound positive charge in to a bilayer [19, 71]

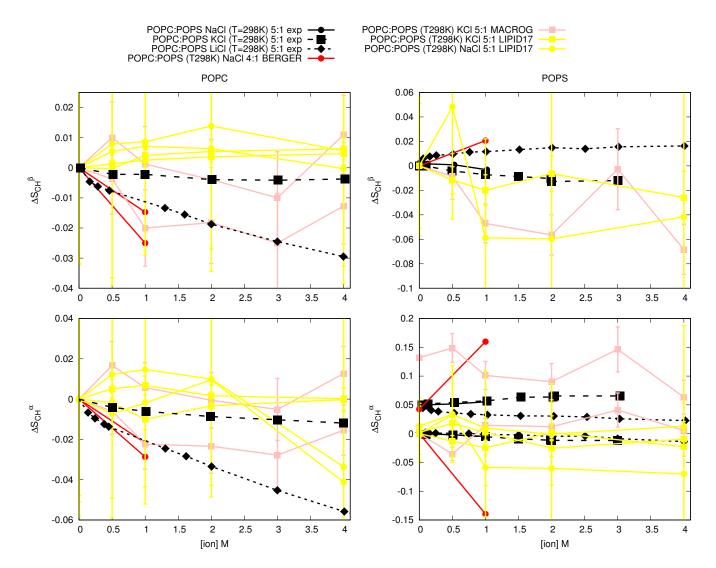


FIG. 7: Changes of the PC (left) and PS (right) headgroup order parameters as a function of added NaCl, KCl and LiCl from POPC:POPS (5:1) mixture. The experimental data is from Ref. 15. The values from counterion-only systems are set as a zero point of y-axis. To correctly illustrate the significant forking of the α -carbon order parameter in PS headgroup (bottom, right), the y-axis is transferred with the same value for both order parameters such that the lower order parameter value is at zero.

38.CHARMM36 results for this plot would be highly useful.

and can be therefore used to measure the ion binding affinity. This molecular electrometer concept can be also applied to lipid bilayers with mixtures of PC and negatively charged lipids [15, 73, 74] (see Fig. 12).

The headgroup order parameter changes of POPC and POPS from POPC:POPS (5:1) mixtures are shown in Fig. 10 as a function of Ca^{2+} concentration from different simulations and experiments [15]. The ion density distributions from the simulations are shown in Fig. 11. The results suggest that Ca^{2+} ions clearly overbind in simulations with MacRog and Berger models, as expected from previous results for PC lipid bilayers [19]. It should be noted, however, that the lowest concentration (100mM) gives a good agreemet with experiments 45.Should be analyze/discuss this further? Binding with \sim 100 mM is saturated in both Berger and MacRog simulations. Maybe this is realistic? It should be noted

that Berger simulation do not have counterions.. Surprisingly, the calcium binding seems to be too weak CHARMM36 simulations. This is due to the NBfix interaction parameters from Ref. 61, incorporated in the parameters from CHARMM-GUI at the time of running the simulations (January 2018). The binding of calcium to pure POPC bilayer is also too weak with these parameters as shown in Figs. 18 and 19. 46.The discussion is to be finished when we have all the data in the plot.

Also the order parameters of PS headgroup from POPC:POPS (5:1) mixture are shown in Fig. 10 as a function of CaCl₂ concentration. In experiments the order parameters exhibit a strong dependence of CaCl₂ with small concentrations with a rapid saturation around 50 mM. The changes of PS headgroup order parameters with added CaCl₂ are overestimated in all tested simulation models. Furthermore, the

SUPPLEMENTARY INFORMATION

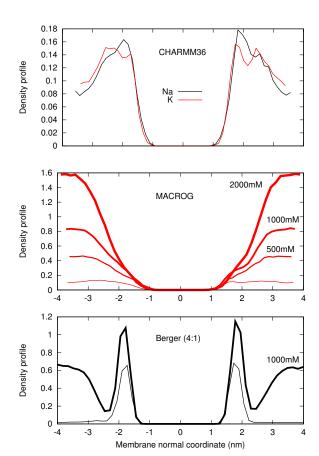


FIG. 8: Counterion density distributions from PC:PS mixtures.

39.Lipid 17 is to be added.

changes of the headgroup order parameters do not qualitatively agree with experiments. This is in contrast to previous results for PC headgroup [19], where qualitatively correct reponse to bound ions was observed despite of significant discrepancies in the headgroup structure without additional ions.

CONCLUSIONS

Simulated systems

CHARMM36

53.To be written by Piggot, Madsen and Ollila

CHARMM36ua

54.To be written by Piggot

Slipids

55.To be written by Piggot and Favela

Berger

56.To be wiritten by Piggot and Ollila Simulations with sodium were taken directly from Ref. ? and simulations with calcium directly from 17. Simulation of POPC at 310 K was taken directly from Ref. 75.

GROMOS-CKP

57.To be written by Piggot

Lipid17

58.To be written by Kav and Miettinen

MacRog

59.To be written by Javanainen and Piggot

Cation binding affinity to lipid bilayers with different amount of charge

Before using the headgroup order parameters to compare ion binding affinity between simulations and experiments, it is important to quantify the response of the order parameters to the bound charge in simulations. The response of headgroup order parameters to the fixed amount of cationic surfactants in POPC bilayer is compared between simulations and experiments [72] In Fig. 14. The figure shows that the order parameters are too sensitive to bound charge in Lipid14 model, while CHARMM36 is in better agreement with experiments. This has to be taken into account when analysin the binding affinities.

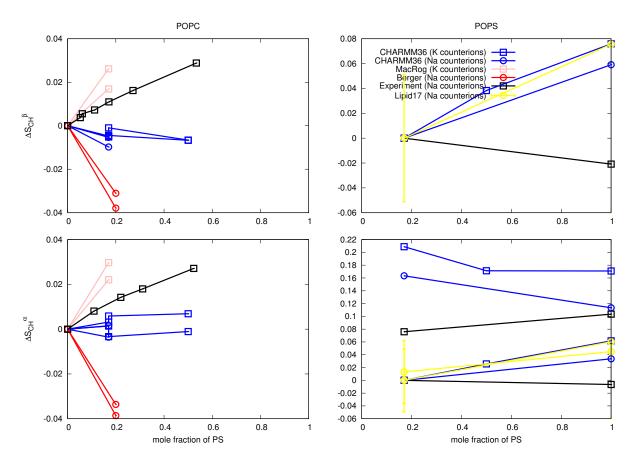


FIG. 9: Changes of PC (left panel) and PS (right panel) headgroup order parameters from POPC:POPS mixtures with increasing amount of POPS. Experimental results of POPC are taken from Ref. 27 (signs are determined as discussed in [18, 20]). Experimental values for POPS in pure bilayer and in mixture are measured in this work and in Ref. 15 at 298K, respectively. For POPC (left panel) the zero point of y-axis is set to the value of pure bilayer. For β -carbon of POPS (right panel, top) the zero point of y-axis is set to the value from POPC:POPS (5:1) mixture. For α -carbon of POPS (right panel, bottom) the y-axis is transferred with the same value for both order parameters such that the lower order parameter value from POPC:POPS (5:1) mixture is at zero to correctly illustrate the significant forking.

40.Simulation of CHARMM36 at 298K should be maybe rerun with Gromacs 5.

41.Simulation of pure POPC at 298K with Lipid14 would be useful for this plot (only at 303 K is available from NMRlipids I) 42.MacRog simulations of pure POPS with potassium counterions only would be useful for this and other plots.

Difference between POPC and OPPS in MacRog model

Dihedrals

Dihedrals

The experimental results show essentially no changes in the order parameters as a function of added NaCl, while significant changes are observed in simulations. However, the minimum buffer concentration of NaCl in the experimental was 100mM [67]. Therefore, we cannot exclude the possibility that the NaCl induced changes were already saturated with 100mM NaCl concentration, which was the case for CaCl₂ in Fig. 10.

Details of the rough subjective force field ranking (Fig. 4)

The assessment was based fully on the Fig. 3. First, for each carbon (the columns in Fig. 3) in each force field (the rows), we looked separately at deviations in magnitude and forking.

Magnitude deviations, i.e., how close to the experimentally obtained C–H order parameters (OPs) the force-field-produced OPs were. For each carbon, the following 5-step scale was used:

0 (): More than half of all the calculated OPs (that is, of all different hydrogens in all different lipids) were within the *subjective sweet spots* (SSP, blue-shaded areas in Fig. 3).

1 (M): All the calculated OPs were < 0.03 units away from the SSP.

2 (M): All the calculated OPs were < 0.05 units away

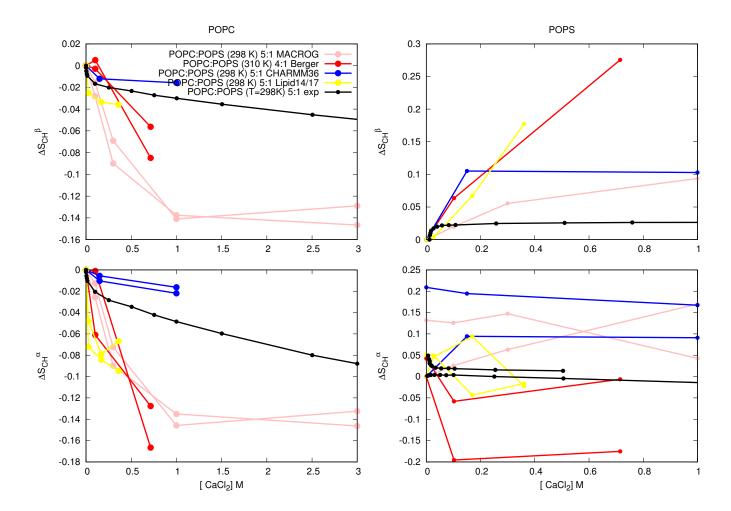


FIG. 10: Changes of POPC (left) and POPS (right) headgroup order parameters in POPC:POPS (5:1) mixture as a function CaCl₂ concentration. Experimental data is taken from 15. The values from counterion-only systems are set as a zero point of y-axis. To correctly illustrate the significant forking of the α -carbon order parameter in PS headgroup (bottom, right), the y-axis is transferred with the same value for both order parameters such that the lower order parameter value is at zero.

47.Information about the cuonterions in different simulations should be added

 $48. Upcoming \ simulations \ with \ original \ CHARMM36 \ have \ been \ mentioned \ in \ the \ blog:$

49.Upcoming Lipid17 simulations have been mentioned in the blog

http://nmrlipids.blogspot.com/2017/12/nmrlipids-iv-current-status-and.html? show Comment = 1515177306419 # c99482561231623546712 # c9948256123162354 # c9948256123162354 # c9948256123162 # c994825612 # c994825612 # c994825612 # c994825612 # c99482612 # c99482612

from the SSP.

3 (M): All the calculated OPs were < 0.10 units away from the SSP.

4 (M): Some of the calculated OPs were > 0.10 units away from the SSP.

Forking deviations, i.e., how well the difference in order parameters of two hydrogens attached to a given carbon matched that obtained experimentally. Note that this is not relevant for β and g_2 , which have only one hydrogen. For the α carbon, for which a considerable forking of 0.105 is experimentally seen, the following 5-step scale was used:

 ${\bf 0}$ (): The distance D between the dots (that mark the

measurement-time-weighted averages in Fig. 3) was 0.08 < D < 0.13 units for all the calculated OPs (that is, for all different lipids).

1 (F):
$$(0.06 < D < 0.08)$$
 OR $(0.13 < D < 0.15)$.

2 (F):
$$(0.04 < D < 0.06)$$
 OR $(0.15 < D < 0.17)$.

3 (F):
$$(0.02 < D < 0.04)$$
 OR $(0.17 < D < 0.19)$.

4 (F):
$$(D < 0.02)$$
 OR $(0.19 < D)$.

For the g_3 carbon, for which no forking is indicated by experiments, the following 5-step scale was used:

0 ():
$$D < 0.02$$
.

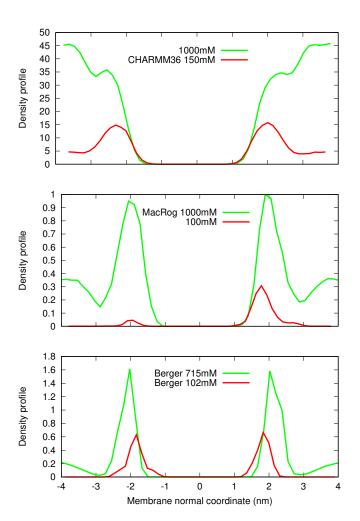


FIG. 11: Ca2+ density profiles from simulations.

50.The CHARMM results are mass densities, numbers should be used.

51.Should we include also counterions into the plot?

52.Not all the data from MacRog is included.

1 (F): 0.02 < D < 0.04.

2 (F): 0.04 < D < 0.06.

3 (F): 0.06 < *D* < 0.08.

4 (F): 0.08 < *D*.

For the g_1 carbon, for which a considerable forking of 0.13 is experimentally seen, the following 5-step scale was used:

0 (): 0.11 < D < 0.15.

1 (F): (0.09 < D < 0.11) OR (0.15 < D < 0.17).

2 (F): (0.07 < D < 0.09) OR (0.17 < D < 0.19).

3 (F): (0.05 < D < 0.07) OR (0.19 < D < 0.21).

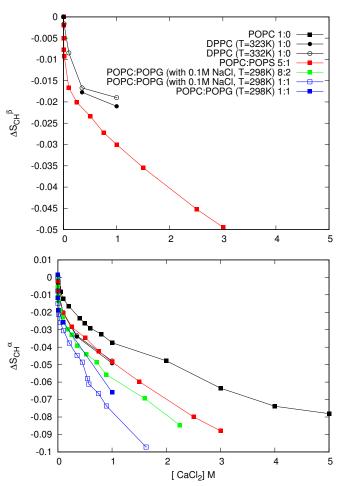


FIG. 12: The change of PC headgroup order parameters as a function of CaCl₂ measured from bilayers containing different amount of negatively charged lipids. The values are taken from 2H NMR experiments reported in the literature (DPPC [69], POPC [70], POPC:POPS (5:1) [15], POPC:POPG mixtures with 0.1M NaCl [74] and POPC:POPG (1:1) without NaCl [73]). As expected, the decrease of order parameters with the added CaCl₂ is more pronounced for systems with larger fraction of negatively charged lipids, indicating larger amount amount of bound cations.

4 (**F**):
$$(D < 0.05)$$
 OR $(0.21 < D)$.

Based on these assessments of magnitude and forking deviations, each carbon was then assigned to one of the following groups: "within experimental error" (magnitude and forking deviations both on step 0 of the scales described above), "almost within experimental error" (sum of the magnitude and forking deviation steps 1 or 2), "clear deviation from experiments" (sum of magnitude and forking deviation steps from 3 to 5), and "major deviation from experiments" (sum of magnitude and forking deviation steps from 6 to 8). These groups are indicated by colors in Fig. 4. (Note that for β and g_2 , for which there can be no forking, the corresponding group assigment limits were: 0, 1, 2, and 3.)

Finally, the total ability of the force field to describe the

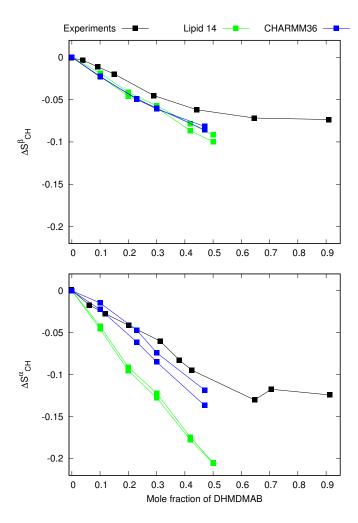


FIG. 13: The response of headgroup order parameters to the fixed amount of cationic surfactants in POPC bilayer is compared between simulations and experiments [72].

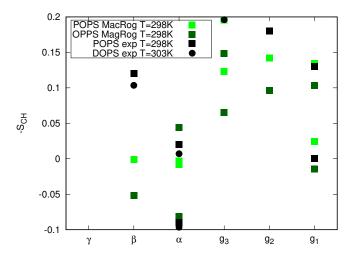


FIG. 14: Headgroup order parameters from POPS and OPPS simulations with MacRog model.

headgroup and glycerol structure was estimated. To this end, the groups were given the following weights: 0 (within experimental error), 1 (almost within experimental error), 2 (clear deviation from experiments), 4 (major deviation from experiments), and the weights of the five carbons were summed up. The sum, given in the Σ -column of Fig. 3, was then used to (roughly and subjectively, as should be clear from the above description) rank the force fields.

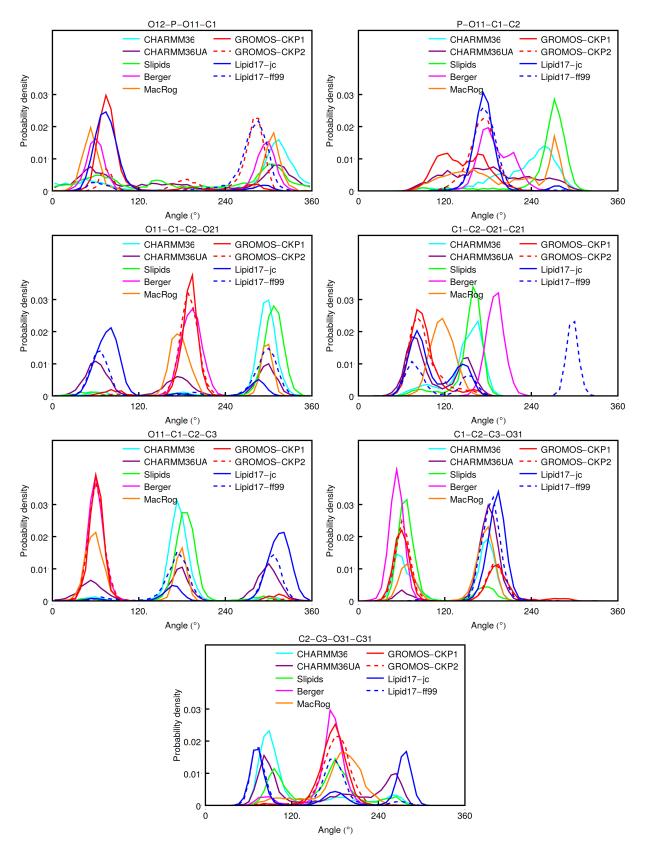


FIG. 15: Dihedral angle distributions of bonds from phosphate to acyl chain carbonyls from different simulation models.

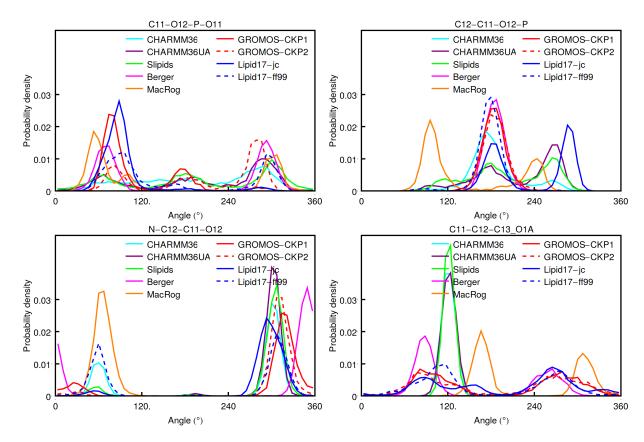


FIG. 16: Dihedral angle distributions of bonds from phosphate to headgroup from different simulation models.

- * samuli.ollila@helsinki.fi
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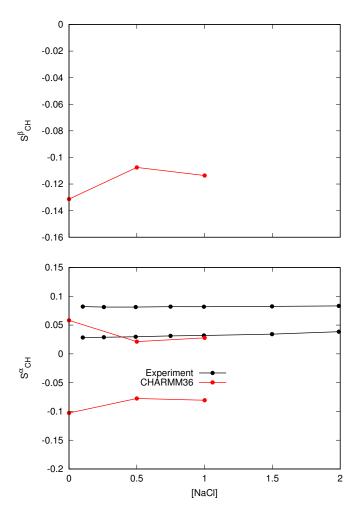


FIG. 17: Order parameters of PS headgroup as a function of added NaCl measured from DMPC:DMPS (3:1) mixture [67].

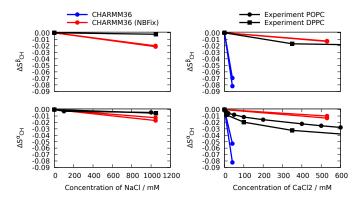


FIG. 18: The response of headgroup order parameters to the fixed amount of cationic surfactants in POPC bilayer is compared between simulations and experiments [72].

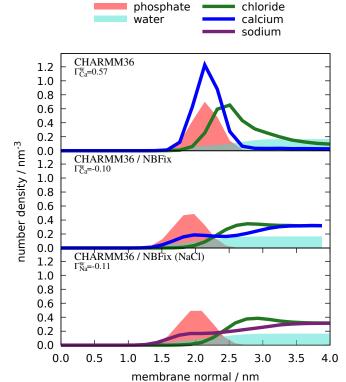


FIG. 19: The response of headgroup order parameters to the fixed amount of cationic surfactants in POPC bilayer is compared between simulations and experiments [72].

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ToDo

2. Details of the used spectrometer and maybe some other details should be given	
other details should be given	1. Authorship query to be sent soon
3. Sample preparation should be described	2. Details of the used spectrometer and maybe some
4. How is the peak assignment done?	other details should be given
5. Should confirm that the amounts of water in experiments matched those in simulations	3. Sample preparation should be described
ments matched those in simulations	4. How is the peak assignment done?
6. Correct citation for CHARMMua DOPS	5. Should confirm that the amounts of water in experi-
7. Correct citation(s) for CKP	ments matched those in simulations
	6. Correct citation for CHARMMua DOPS
8. Correct citation(s) for CKP	7. Correct citation(s) for CKP
	8. Correct citation(s) for CKP

9. Correct citation for CHARMMua DOPS	2	43. The existing literature about this should be men-	
10. Data to be added by Piggot	2	tioned/discussed	6
11. Correct citation(s) for CKP	2	44. This discussion should be updated	6
12. Correct citation(s) for CKP	2	38. CHARMM36 results for this plot would be highly	
29. Error estimation should be discussed	2	useful	7
13. Should confirm that the amounts of water in exper-		45. Should be analyze/discuss this further? Binding	
iments matched those in simulations	3	with \sim 100 mM is saturated in both Berger and MacRog	
14. Equilibration?	3	simulations. Maybe this is realistic? It should be noted	
15. Trajectories and further details to be added by J.		that Berger simulation do not have counterions	7
Madsen	3	46. The discussion is to be finished when we have all	•
16. Trajectories and further details to be added by J.		the data in the plot.	7
Madsen	3		
17. Concentration to be checked	3	39. Lipid 17 is to be added	8
18. Trajectories and further details to be added by J.		53. To be written by Piggot, Madsen and Ollila	8
Madsen	3	54. To be written by Piggot	8
19. Concentration to be checked	3	55. To be written by Piggot and Favela	8
20. Trajectories and further details to be added by J.		56. To be wiritten by Piggot and Ollila	8
Madsen	3	57. To be written by Piggot	8
21. This is also probably OPPS? These should be cor-		58. To be written by Kav and Miettinen	8
rected in this table as well	3	59. To be written by Javanainen and Piggot	8
22. Equilibration?	3	40. Simulation of CHARMM36 at 298K should be	
23. Equilibration?	3	maybe rerun with Gromacs 5	9
24. Equilibration?	3	41. Simulation of pure POPC at 298K with Lipid14	
25. Equilibration?	3	would be useful for this plot (only at 303 K is available	
26. Equilibration?	3	from NMRlipids I)	9
27. MacRog simulations with KCl to be added	3	42. MacRog simulations of pure POPS with potassium	
28. Berger simulations with NaCl and CaCl to be added	3	counterions only would be useful for this and other plots.	9
35. Also the discussion about POPS/OPPS issue with		47. Information about the cuonterions in different sim-	
MacRog model should be added	3	ulations should be added	10
30. This is preliminary figure, should be polished	4		10
31. Should we show slices for all the analyzed carbons		48. Upcoming simulations with original	
in (c)?	4	CHARMM36 have been mentioned in the blog:	
36. More detailed discussion may be possible after		http://nmrlipids.blogspot.com/2017/12/nmrlipids-iv-	. 55(02(0
comparing monovalent ion binding to bilayers between		current-status-and.html?showComment=1520090718976#	C3369269
CHARMM simulations and experiments	4	49. Upcoming Lipid17 simulations	
32. There should be values in [15] which should be added.	4	have been mentioned in the blog	
33. Issue about possible updates to this plot:		http://nmrlipids.blogspot.com/2017/12/nmrlipids-iv-	0040056
https://github.com/NMRLipids/NMRlipidsIVotherHGs/iss	ues/4	gurrent-status-and.html?showComment=1515177306419#	c9948256
34. Lipid17 and MacRog results should be added into		50. The CHARMM results are mass densities, numbers	
this plot	5	should be used	11
37. Discussion about Lipid17 to be written when we		51. Should we include also counterions into the plot? .	11
have the density profiles	5	52. Not all the data from MacRog is included	11