Periodic Table of the Elements

\mathbf{He}^{1s}	$\overset{2p}{ ext{Neon}}_{\overset{ ext{Neon}}{ ext{20.1797}}}$	$\mathbf{Ar}_{^{\mathrm{Argon}}}^{3p}$	\mathbf{K}^{5}	$\overset{2.60}{\mathrm{Xe}}$ 5p $\overset{5}{\mathrm{Xe}}$ Senon 131.293	$\mathop{\mathbf{Rn}}_{(222)}^{2.2}$ 6 p	$\bigcup_{\substack{\text{Oganesson}\\(294)}}^{7p}$
0	$\frac{3.98}{\textbf{F}} \begin{array}{c} 2p \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ $	$ \begin{array}{c cccc} & 3.16 & 3p \\ & C1 \\ & Chlorine \\ & 35.451 \end{array} $	$\left egin{array}{c} 2.96 & 4p \ \mathbf{Br} \end{array} ight _{79.904}$	Lodine 126.65 5p 54 Lodine 126.90447	$\overset{2.2}{\mathbf{At}}$ 6p 86 Astatine (210)	$\left. \prod_{\substack{7p \ \mathrm{mnessine} \ (294)}}^{7p} \right _{1}$
	3.44 2p 99 Oxygen 15.999	6 2.58 3p 17 Sulphur (2.55 4p 35 Se elemium 78.971	$egin{array}{c cccc} \mathbf{Z} & \underline{2.1} & 5p & 53 \\ \mathbf{Te} & \mathbf{Pe} \\ \mathrm{Tellurium} & 127.60 & 1 \end{array}$	\mathbf{Po}^{4} Polonium (209)	$ \begin{bmatrix} 7p \\ LV \\ \text{Imporium} \\ (293) \end{bmatrix} $
	$\begin{bmatrix} 7 & 3.04 & 2p \\ \hline N \\ Nitrogen \\ 14.007 \end{bmatrix}$	15 2.19 3p 16 Phosphorus 30.973761998	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	S1 2.05 5p S2 SD Antimony 121.760	33 2.02 6p 84 Bismuth F	$\sum_{\substack{\text{Moscovium} \\ (289)}}^{7p} \sum_{\text{Live}}^{7p} \frac{116}{\text{Live}} $
	6 2.55 2p 7 Carbon 12.011	Silicon PP 28.085 30.0	$\begin{array}{c c} 32 & \underline{2.01} & 4p \\ \mathbf{Ge} \\ \mathbf{Germanium} \\ 72.630 \end{array}$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\Pr_{\mathrm{Lead}}^{82} \stackrel{\mathbf{6p}}{=} \mathbb{B}^{33}$	$\stackrel{7p}{\text{F1}}$
	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c c} \textbf{13} & \underline{1.61} & 3p \\ \hline & \textbf{A1} \\ \text{Aluminium} \\ 26.9815385 \end{array}$	$\left(egin{array}{c} 1.81 & 4p & 32 \ \textbf{Callium} & \text{Gallium} \ 69.723 & \end{array} ight)$	$\overset{1.78}{\text{Indium}}$	5d 81 1.62 6p 82 Thallium 204.384	$\sum_{\substack{\text{Nihonium} \\ (286)}}^{7p}$
			$\sum_{\substack{\mathrm{Zinc} \\ 65.38}}^{30} \frac{3d}{1.65} \frac{31}{3}$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\mathop{H_{\mathrm{Brcury}}}\limits_{^{2.00-592}}$	$ \begin{matrix} 6d & 112 & 6d \\ & \mathbf{Cn} \\ & \mathbf{Copernicium} \\ & (285) \end{matrix} $
		$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\mathop{A_{\mathrm{Silver}}}_{107.8682}^{47}$	${f A}_{{ m Cold}}^{2.54}$ 5 d^*	$\begin{array}{c c} 110 & 6d & 111 & 6d \\ \hline \textbf{DS} & \textbf{Rg} \\ \text{Darmstadtium} & \text{Roentgenium} \\ (281) & (282) \end{array}$	
Z= atomic number; eneg = electronegativity; ss = subshell; Sy = Symbol, Name = element name, saw = standard atomic weight			3d 28 1.91 3d Nickel 58.6934	$\overset{46}{\overset{2.20}{\overset{4}d^*}}$	54 78 2.28 54* 79 Pt Platinum 195.084	
			3d 27 1.88 3d Co Cobalt 58.933194	${ m Rhodium \atop 102.90550}$	54 77 2.20 54 Ir Iridium 192.217	$\begin{array}{ c c }\hline & 109 & \mathbf{6d} \\ \hline & \mathbf{Mt} \\ \mathbf{Meitnerium} \\ & (278) \\ \hline \end{array}$
			3d 26 1.83 3d Fe Iron 55.845		54 76 2.2 54 Osmium 190.23	6d 108 6d HS Hassium (269)
			$\sqrt{\mathbf{h}}$	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	5d 75 1.9 5d Renium 186.207	$\begin{bmatrix} 6d \\ \mathbf{Bh} \end{bmatrix}$
omic number; en 7mbol, Name =		$\begin{array}{c c} 24 & \underline{1.66} & 3d^* & 25 \\ & \mathbf{Cr} & & \mathbf{I} \\ \text{Chromium} & & Max \\ 51.9961 & & 54 \end{array}$	$\overline{\mathbf{Mo}}^{2.16 \ 4d^*}_{\mathrm{Molybdenum}}$	54 74 2.36 54 W Tungsten 183.84	$\begin{array}{c c} 6d & 106 & 6d \\ & \mathbf{Sg} \\ & \text{Seaborgium} \\ & (269) \end{array}$	
		٦	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	41 1.6 4d* Nobium 92.90637	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\mathop{\mathrm{Db}}_{\text{Dubnium}}^{105}$
	$\sum_{\substack{\text{c eneg} \\ \text{Name} \\ \text{saw}}} S$		$\prod_{47.867}^{1.54}$	$\sum_{\substack{1.33 \text{ Zirconium} \\ 91.224}} 4d$	$\overset{72}{\text{Halfnium}}\overset{5d}{\text{Halfnium}}$	$\Pr^{104}_{\mathbf{R}\mathbf{f}}$
			48 21 1.36 3d 22 Scandium 44.955908	$\begin{array}{c c} 39 & \underline{1.22} & 4d \\ \hline & \mathbf{Y} \\ \text{Yttrium} \\ 88.90584 \end{array}$	57-71 *	* ** * * * * * * * * * * * * * * * * *
8	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	5s 38 0.95 5s Sr Strontium 87.62	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{ccc} 3 & \underline{0.98} & 2s \\ & \mathbf{Lithium} \\ \text{6.990} \end{array}$	$\overset{11}{\overset{0.93}{\overset{3s}}{\overset{3s}}{\overset{3s}}{\overset{3s}}{\overset{3s}}{\overset{3s}}{\overset{3s}}{\overset{3s}}}}{\overset{3s}}{\overset{3s}}{\overset{3s}}}{\overset{3s}}{\overset{3s}{\overset{3s}{\overset{3s}{\overset{3s}{\overset{3s}}{\overset{3s}}{\overset{3s}}{\overset{3s}}}}{\overset{3s}{\overset{3s}{\overset{3s}{\overset{3s}{\overset{3s}}{\overset{3s}}{\overset{3s}}{\overset{3s}}}}}{\overset{3s}}{\overset{3s}}{\overset{3s}}}}{\overset{3s}{\overset{3s}{\overset{3s}{\overset{3s}}{\overset{3s}}{\overset{3s}}}}}}{\overset{3s}}{\overset{3s}}}}}{\overset{3s}}{\overset{3s}}{\overset{3s}}}}}{\overset{3s}}{\overset{3s}}{\overset{3s}}}}}}{3s$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\frac{37}{\mathbf{Rb}}$	$ \begin{array}{cccc} 55 & \underline{0.79} & 6s \\ \mathbf{CS} \\ \mathbf{Cesium} \\ 132.90545196 \end{array} $	$\mathbf{Francium}^{77}$ Francium (223)

4.	6.
4f 71 1.27 4f Lu Lutetium 174.9668	103 1.3 5 <i>f</i> Lawrencium (266)
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	102 1.3 5f Nobelium (259)
tf 69 1.25 4f Tm Thulium 168.93422	$\underset{(258)}{\overset{101}{\text{L}.3}} \overset{5f}{\text{5}}$
$\frac{68}{\mathbf{Er}}_{\text{brium}}^{1.24}$ Erbium $_{167.259}$	$ \begin{array}{c c} 100 & \underline{1.3} & 5f \\ \hline F & \mathbf{n} \\ \text{Fermium} \\ (257) \end{array} $
44 67 1.23 44 HO Holmium 164.93033	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$
$\mathop{D_{\rm ysprosium}}_{\rm 162.500}$	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$
$\prod_{\substack{\text{Terbium}\\158.92535}}^{4f}$	L3 X celium 247)
4f 64 1.2 $4f^*$ 65 C	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$
$\overset{63}{\overset{1.2}{\overset{1.2}{\mathbf{bu}}}}\overset{4f}{\mathbf{bu}}$	5f 95 1.13 5f 96 Am Americium (243)
$\mathop{\mathrm{Smarium}}_{150.36}$	$\overset{94}{\overset{1.28}{\text{Pu}}}$
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$
$\overset{60}{\overset{1.14}{\overset{1.4}{\text{Nodymium}}}}\overset{4f}{\overset{Neodymium}{\overset{144.242}{\overset{1}{\text{A}}}}}$	92 $1.38 5f^*$ U_{ramium} Uranium 238.02891
$\Pr_{ ext{Praseodymium}} egin{array}{c} 1.13 & 4f \\ Pr & Pr \end{array}$	$\Pr_{231.03588}^{91 \ \underline{1.5} \ 5f^*}$
	$egin{array}{c c c c c c c c c c c c c c c c c c c $
$\overset{57}{\overset{1.1}{\overset{5d}{\mathbf{a}}}}\overset{5d^*}{\overset{Lanthanum}{\overset{138.90547}{{{{{{{}{}{$	89 $\frac{1.1}{\mathbf{A}}$ 6 d^* Actinium (227)
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Standard atomic weights taken from the Commission on Isotopic Abundances and Atomic Weights (ciaaw.org/atomic-weights.htm). Adapted from Ivan Griffin's LAEX Periodic Table. © 2016 Paul Danese