## Periodic table of elements

## Mendeleev's table

1 2.20 1s H Hydrogen 1.00784–1.00811																	${ {\rm He}\atop {\rm Helium}\atop {\rm 4.002602(2)}} {\rm 1s}$
	1.57 2s Be Beryllium 9.0121831(5)		$egin{array}{cccc} {\sf Z} & \chi & {\sf ss} \ & {f Sy} \ & { m Name} \ & { m saw} \ \end{array}$	ss: subshell Sy: symbol Name: elemen	gativity							5 2.04 2p B Boron 10.806-10.821	6 2.55 2p C Carbon 12.0096–12.0116	7 3.04 2p N Nitrogen 14.00643- 14.00728	8 3.44 2p O Oxygen 15.99903- 15.99977	9 3.98 2 <i>p</i> <b>F</b> Fluorine 18.998403163(6)	10 2p Ne Neon 20.1797(6)
Na Sodium	12 1.31 3s Mg Mg Magnesium 24.304-24.307											13 1.61 3p Al Aluminium 26.9815385(7)	14 1.90 3p Si Silicon 28.084-28.086	15 2.19 3 <i>p</i> P Phosphorus 30.973761998(5)	16 2.58 3 <i>p</i> S Sulfur 32.059–32.076	7 17 3.16 3p Cl Chlorine 35.446-35.457	18 3p Ar Argon 39.948(1)
19 0.82 4s 20 <b>K</b> Potassium 39.0983(1)	Calcium 40.078(4)	21 1.36 3 <i>d</i> Sc Scandium 44.955908(5)	22 1.54 3d <b>Ti</b> Titanium 47.867(1)	23 1.63 3 <i>d</i> <b>V</b> Vanadium 50.9415(1)	24 1.66 3 <i>d*</i> Cr  Chromium  51.9961(6)	25 1.55 3 <i>d</i> Mn Manganese 54.938044(3)	26 1.83 3 <i>d</i> Fe Iron 55.845(2)	27 1.88 3 <i>d</i> CO Cobalt 58.933194(4)	28 1.91 3 <i>d</i> Ni Nickel 58.6934(4)	29 1.90 3 <i>d*</i> Cu Copper 63.546(3)	30 1.65 3 <i>d</i> 2n Zinc 65.38(2)	31 1.81 4p Ga Gallium 69.723(1)	32 2.01 4p Ge Germanium 72.630(8)	33 2.18 4p <b>As</b> Arsenic 74.921595(6)	34 2.55 4p Se Selenium 78.971(8)	35 2.96 4p Br Bromine 79.901-79.907	36 3.00 4p <b>Kr</b> Krypton 83.798(2)
37 0.82 5s 38 Rb Rubidium 85.4678(3)	88 0.95 5s Sr Sr Strontium 87.62(1)	39 1.22 4 <i>d</i> <b>Y</b> Yttrium  88.90584(2)	40 1.33 4d <b>Zr</b> Zirconium 91.224(2)	41 1.6 4 <i>d*</i> Nb Niobium 92.90637(2)	42 2.16 4 <i>d*</i> Mo  Molybdenum  95.95(1)	Tc	44 2.2 4 <i>d*</i> Ru Ruthenium 101.07(2)	* 45 2.28 4 <i>d</i> * Rh Rhodium 102.90550(2)	46 2.20 4 <i>d*</i> Pd Palladium 106.42(1)	47 1.93 4 <i>d*</i> Ag Silver 107.8682(2)	48 1.69 4 <i>d</i> Cd Cadmium 112.414(4)	49 1.78 5 <i>p</i> In Indium 114.818(1)	50 1.96 5 <i>p</i> Sn Tin 118.710(7)	51 2.05 5 <i>p</i> Sb Antimony 121.760(1)	52 2.1 5 <i>p</i> <b>Te</b> Tellurium  127.60(3)	53 2.66 5p I Iodine 126.90447(3)	Xe Xenon 131.293(6)
55 0.79 6s 56 CS Cesium 132.90545196(6)	Barium 137.327(7)	* Lanthanides	72 1.3 5 <i>d</i> Hf Hafnium 178.49(2)	73 1.5 5 <i>d</i> <b>Ta</b> Tantalum  180.94788(2)	74 2.36 5 <i>d</i> W  Tungsten  183.84(1)	75 1.9 5 <i>d</i> Re Rhenium 186.207(1)	76 2.2 5 <i>d</i> OS OSmium 190.23(3)	77 2.20 5 <i>d</i> Ir  Iridium  192.217(3)	78 2.28 5 <i>d*</i> Pt  Platinum  195.084(9)	79 2.54 5 <i>d*</i> Au Gold 196.966569(5)	80 2.00 5 <i>d</i> Hg Mercury 200.592(3)	81 1.62 6p Tl Thallium 204.382-204.385	82 1.87 6 <i>p</i> Pb  Lead 207.2(1)	83 2.02 6p Bi Bismuth 208.98040(1)	Po Polonium (209)	$\begin{array}{c} \mathbf{p} & 85 & 2.2 & 6p \\ \mathbf{At} & \mathbf{Astatine} \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ \end{array}$	86 2.2 6p Rn Radon (222)
87 0.7 7s 88 <b>Fr</b> Francium (223)	88 0.9 7s Ra Radium (226)	** Actinides	$rac{ ext{Rf}}{ ext{Rf}}$ Rutherfordium $rac{ ext{(261)}}{ ext{}}$	Db	$\begin{array}{ccc} 106 & 6d \\ \mathbf{Sg} \\ & & \\ \mathbf{Seaborgium} \\ & & $	${ m Bh} \atop { m Bohrium} \atop { m (270)}$	108 6 <i>d</i> HS Hassium (269)	Mt	${f Ds} \ {f Ds} \ {f Darmstadtium}$	Rg	Cn	$\begin{array}{c} 113 & 7  ho \\ \mathbf{Nh} \\ \mathrm{Nihonium} \\ \mathrm{(286)} \end{array}$	$\begin{array}{c} 114 & 7p \\ \hline Fl \\ Flerovium \\ \tiny (289) \end{array}$	Mc $Mc$ $Moscovium$ $(289)$	${ m Lv}_{ m Livermorium}$	Ts	O(118) $O(118)$ $O$
		*	57 1.1 5 <i>d*</i> La  Lanthanum 138.90547(7)	Ce	F 59 1.13 4 <i>f</i> <b>Pr</b> Praseodymium 140.90766(2)	Nd	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	62 1.17 4f Sm  Samarium  150.36(2)	63 1.2 4f Eu Europium 151.964(1)	64 1.2 4 <i>f</i> *  Gd  Gadolinium  157.25(3)	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	06 1.22 4f Dy Dysprosium 162.500(1)	67 1.23 4f Ho Holmium 164.93033(2)	68 1.24 4f Er Erbium 167.259(3)	69 1.25 4f Tm Thulium 168.93422(2)	$egin{array}{cccccccccccccccccccccccccccccccccccc$	71 1.27 4f  Lu  Lutetium  174.9668(1)
		**	89 1.1 6 <i>d*</i> Ac Actinium (227)	90 1.3 5f* <b>Th</b> Thorium  232.0377(4)	Protactinium 231.03588(2)	U	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	94 1.28 5f Pu Plutonium (244)	95 1.13 5 <i>f</i> Am Americium (243)	96 1.28 5 <i>f</i> *  Cm  Curium  (247)	97 1.3 5 <i>f</i> <b>Bk</b> Berkelium (247)	$\begin{array}{ccc} 98 & 1.3 & 5f \\ \textbf{Cf} & \\ \text{Californium} & \\ & & \\ & & \\ & & \\ & & \\ & & \\ \end{array}$	99 1.3 5 <i>f</i> <b>Es</b> Einsteinium (252)	100 1.3 5 <i>f</i> <b>Fm</b> Fermium  (257)	101 1.3 5 <i>f</i> Md  Mendelevium (258)	No	$\begin{array}{ccc} 103 & 1.3 & 5f \\ \mathbf{Lr} \\ \mathrm{Lawrencium} \\ ^{(266)} \end{array}$

Standard atomic weights taken from the Commission on Isotopic Abundances and Atomic Weights (ciaaw.org/atomic-weights.htm).

An asterisk (\*) next to a subshell indicates an anomalous (Aufbau rule-breaking) ground state electron configuration.