MiXed Layer CHemistry (MXLCH) model: Users Guide

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Chapter 1

Introduction

This document gives information of the input files for the current version of the MiXed Layer model with CHemistry (MXLCH). In short, the model solves simultaneously the temporal evolution of potential temperature, specific humidity, wind and reactant species in a diurnal convective boundary layer.

The chemical mechanisms in MXLCH are very flexible offering the possibility of adding and removing chemical species and reactants. Here we provide an example of a *reduced* chemical system and a *complex* one. The chemical input files for the one way can not be used for the other, but they can be modified and adjusted accordingly by the user. An overview of all namelist options is given in Chapter 2. In Chapter 3 the input files for chemistry are discussed. In Section 3.1 the input file for the simple chemistry is discussed. The chemistry mechanism is explained in Section 3.2.

More information on the dynamics and chemistry of the model and its performance can be found in the article:

The role of boundary layer dynamics on the diurnal evolution of isoprene and the hydroxyl radical over tropical forest (2011) Vilà-Guerau de Arellano J, Patton E.G, Karl T., van den Dries K., Barth M. C and Orlando J. J. *J. Geophys-Res-Atmospheres*.

Chapter 2

Namoptions

All options that can be set for experiments using the combined version of MXLCH will be discussed in the following paragraphs. For all name lists, the options are given with their default values, possible values, a description and the unit. The possible values are denoted by an x.

2.1 Namelist NAMRUN

Option	Default	Possible values	Description	Unit
outdir time dtime atime	'RUN00' 86400 1 60	$x =' \langle NAME \rangle'$ $x \in \mathbb{N}$ $x \in \mathbb{R}, x > 0$ $x \in \mathbb{R}, x > 0$	Name of output folder Simulated time Timestep Time interval for statistics	- S S
atime_vert	1800	$x \in \mathbb{R}, x > 0$	Time interval for vertical profile	S
h_max	3000	$x \in \mathbb{R}, x > 0$	Maximum height of simulated do-	m
latt	0	$x \in \mathbb{R}, -90 \le x \le 90$	Latitude of simulated location	0
long	0	$x \in \mathbb{R}, 0 \le x \le 360$	Longitude of simulated location	0
day	80	$x \in \mathbb{N}, 1 \le x \le 365$	Julian day of the year	-
hour	0	$x \in \mathbb{R}, 0 \le x \le 24$	Local time at which the simulation	hr
			starts	

2.2 Namelist NAMDYN

Option	Default	Possible values	Description	Unit
zi0	200	$x \in \mathbb{R}, x > 0$	Initial boundary layer height	m
beta	0.2	$x \in \mathbb{R}, x \ge 0$	Entrainment ratio	_
wsls	0	$x \in \mathbb{R}$	Flow divergence factor for subsidence	s^{-1}
wthetasmax	0	$x \in \mathbb{R}$	Maximum surface kinematic heat flux for standard flux	$\mathrm{K}\mathrm{m}\mathrm{s}^{-1}$
c_fluxes	.false.	$x \in \{.\text{false.},.\text{true.}\}$	profiles If true all fluxes are constant. It is better to use the options in Paragraph 2.7.	-
gamma	0.006	$x \in \mathbb{R}, x \ge 0$	Potential temperature lapse rate in the free troposphere	$ m K~m^{-1}$
thetam0	295	$x \in \mathbb{R}, x > 0$	Initial mixed layer potential temperature	K
dtheta0	4	$x \in \mathbb{R}, x \ge 0$	Initial potential temperature jump	K
advtheta	0	$x \in \mathbb{R}$	Advection of potential temperature	$ m Ks^{-1}$
ladvecFT	.false.	$x \in \{.\text{false.},.\text{true.}\}$	If true advection is also ap-	-
pressure	1013	$x \in \mathbb{R}, x > 0$	plied for free troposphere. Air pressure in the boundary layer	Pa
wqsmax	0	$x \in \mathbb{R}$	Maximum surface kinematic moisture flux for standard flux profiles	$g kg^{-1} m s^{-1}$
gammaq	0	$x \in \mathbb{R}$	Specific humidity lapse rate in the free troposphere	$g kg^{-1} m^{-1}$
qm0	0	$x \in \mathbb{R}, x \ge 0$	Initial mixed layer specific humidity	$g kg^{-1}$
dq0	0	$x \in \mathbb{R}, x \ge -qm0$	Initial specific humidity jump	$g kg^{-1}$
advq	0	$x \in \mathbb{R}$	Advection of specific humidity	
wcsmax	0	$x \in \mathbb{R}$	Maximum surface kinematic tracer flux for standard flux profiles	ppb m s ⁻¹

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Option	Default	Possible values	Description	Unit
gammac	0	$x \in \mathbb{R}$	Tracer lapse rate in the free troposphere	ppb m ⁻¹
cm0	0	$x \in \mathbb{R}, x \ge 0$	Initial mixed layer tracer con- centration	ppb
dc0	0	$x \in \mathbb{R}, x \ge -\text{qm}0$	Initial tracer concentration jump	ppb
c_ustr	.true.	$x \in \{.\text{false.},.\text{true.}\}$	If true the momentum fluxes (and friction velocity) are constant.	-
z0	0.03	$x \in \mathbb{R}, x > 0$	Roughness length	m
uws0	0	$x \in \mathbb{R}$	Initial surface (<i>x</i> -)momentum	${ m m}^2{ m s}^{-2}$
			flux	
vws0	0	$x \in \mathbb{R}$	Initial surface (<i>y</i> -)momentum flux	$\mathrm{m}^2\mathrm{s}^{-2}$
gammau	0	$x \in \mathbb{R}$	Lapse rate of u in the free tro-	s^{-1}
gammav	0	$x \in \mathbb{R}$	posphere Lapse rate of v in the free troposphere	s^{-1}
um0	0	$x \in \mathbb{R}$	Initial u in the mixed layer	$\mathrm{m}\mathrm{s}^{-1}$
vm0	0	$x \in \mathbb{R}$	Initial v in the mixed layer	$\mathrm{m}\mathrm{s}^{-1}$
ug	0	$x \in \mathbb{R}$	Geostrophic wind in the x-	${ m m~s^{-1}}$
			direction	
vg	0	$x \in \mathbb{R}$	Geostrophic wind in the <i>y</i> -	$\mathrm{m}\mathrm{s}^{-1}$
			direction	
lencroachment	.false.	$x \in \{.\text{false.},.\text{true.}\}$	Enables encroachment	-

2.3 Namelist NAMSURFLAYER

Option	Default	Possible values	Description	Unit
lsurfacelayer	.false.	$x \in \{.\text{false.}, .\text{true.}\}$ $x \in \mathbb{R}, x > 0$ $x \in \mathbb{R}, x > 0$	Enable or disable surface layer	-
z0m	0.03		Roughness length momentum	m
z0h	0.03		Roughness length heat	m

2.4 Namelist NAMRAD

Option	Default	Possible values	Description	Unit
lradiation cc S0 albedo	.false. 0 1368 0.2	$x \in \{.\text{false.}, .\text{true.}\}$ $x \in \mathbb{R}, 0 \le x \le 1$ $x \in \mathbb{R}, x > 0$ $x \in \mathbb{R}, 0 \le x \le 1$	Enable or disable radiation Cloud cover Incoming shortwave solar radiation Albedo	- - W m ⁻²

2.5 Namelist NAMSURFACE

Option	Default	Possible values	Description	Unit
llandsurface	.false.	$x \in \{.\text{false.},.\text{true.}\}$	Enable or disable interactive land surface	-
Qtot	400	$x \in \mathbb{R}$	Net total radiation energy (recalculated if lradiation = .true.)	$ m W~m^{-2}$
Ts	thetam0	$x \in \mathbb{R}, x > 0$	Initial surface temperature	K
wwilt	0.314	$x \in \mathbb{R}, x \ge 0$	Wilting point	$\begin{array}{ c c c c c c c c c c c c c c c c c c c$
w2	0.420	$x \in \mathbb{R}, x \geq 0$	Volumetric water content deeper soil	${\sf m}^3 {\sf m}^{-3}$
			layer	
wg	0.400	$x \in \mathbb{R}, x \ge 0$	Volumetric water content top soil	$\mathrm{m}^3\mathrm{m}^{-3}$
		, _	layer	
wfc	0.491	$x \in \mathbb{R}, x \ge 0$	Volumetric water content field capac-	$\mathrm{m}^3\mathrm{m}^{-3}$
			ity	
wsat	0.600	$x \in \mathbb{R}, x \ge 0$	Saturated volumetric water content	$\mathrm{m}^3~\mathrm{m}^{-3}$
CLa	0.083	$x \in \mathbb{R}, x \ge 0$	Clapp and Hornberger retention	-
			curve parameter a	
CLb	11.4	$x \in \mathbb{R}, x \ge 0$	Clapp and Hornberger retention	-
			curve parameter b	
CLc	12.0	$x \in \mathbb{R}, x \ge 0$	Clapp and Hornberger retention	-
			curve parameter c	
C1sat	0.342	$x \in \mathbb{R}, x \ge 0$	Coefficient force term moisture	-
C2ref	0.3	$x \in \mathbb{R}, x \ge 0$	Coefficient restore term moisture	-
gD	0.0	$x \in \mathbb{R}, x \ge 0$	VPD correction factor for rs	-
rsmin	0.0	$x \in \mathbb{R}, x \ge 0$	Minimum resistance of transpiration	$\mathrm{s}\mathrm{m}^{-1}$
rssoilmin	0.0	$x \in \mathbb{R}, x \ge 0$	Minimum resistance of soiltranspira-	$ m s~m^{-1}$
			tion	
LAI	1.0	$x \in \mathbb{R}, x \ge 0$	Leave area index	\mid m 2 m $^{-2}\mid$
cveg	1.0	$x \in \mathbb{R}, 0 \le x \le 1$	Vegetation fraction	-

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Option	Default	Possible values	Description	Unit
Tooil	205	C ID > 0	Initial tomorphisms tomosil large	IV.
Tsoil T2	285 285		Initial temperature top soil layer	K K
Wl	0.0	$\begin{array}{ccc} x \in \mathbb{R}, & x > 0 \\ x \in \mathbb{R}, & x \ge 0 \end{array}$	Temperature deeper soil layer	
VVI	0.0	$x \in \mathbb{R}, x \geq 0$	Equivalent water layer depth for wet	m
Lambda	5.9	$x \in \mathbb{R}, x \ge 0$	vegetation Thermal conductivity skin layer di-	$ m W \ m^{-2} \ K^{-1}$
Lambaa	0.7	₩ C 112, ₩ <u>></u> 0	vided by depth	VVIII IX
CGsat	$3.6 \cdot 10^{-6}$	$x \in \mathbb{R}, x \ge 0$	Saturated soil conductivity for heat	$ m K~m^2~J^{-1}$
lrsAgs	.false.	$x \in \{.\text{false.},.\text{true.}\}$	Enable or disable A-gs model for sur-	_
			face resistances	
lCO2Ags	.false.	$x \in \{.\text{false.},.\text{true.}\}$	Enable or disable A-gs model for CO ₂	-
			surface fluxes	
CO2comp298	68.5	$x \in \mathbb{R}, x \ge 0$	CO ₂ compensation concentration	${ m mg~m^{-3}}$
Q10CO2	1.5	$x \in \mathbb{R}, x > 0$	Parameter to calculate CO ₂ compen-	-
			sation concentration	
gm298	7	$x \in \mathbb{R}, x > 0$	Mesophyll conductance at 298 K	mm s ⁻¹
Ammax298	2.2	$x \in \mathbb{R}, x > 0$	CO ₂ maximal primary productivity	${\rm mg}~{\rm m}^{-2}~{\rm s}^{-1}$
Q10gm	2	$x \in \mathbb{R}, x > 0$	Parameter to calculate mesophyll	-
		_	conductance	
T1gm	278	$x \in \mathbb{R}, x > 0$	Reference temperature to calculate mesophyll conductance	K
T2gm	301	$x \in \mathbb{R}, x > 0$	Reference temperature to calculate	K
6			mesophyll conductance	
Q10Am	2	$x \in \mathbb{R}, x > 0$	Parameter to calculate maximal pri-	_
			mary productivity	
T1Am	281	$x \in \mathbb{R}, x > 0$	Reference temperature to calculate	K
			maximal primary productivity	
T2Am	311	$x \in \mathbb{R}, x > 0$	Reference temperature to calculate	K
			maximal primary productivity	
f0	0.89	$x \in \mathbb{R}, x > 0$	Maximum value Cfrac	-
ad	0.07	$x \in \mathbb{R}, x > 0$	Regression coefficient to calculate	kPa ^{−1}
			Cfrac	_ 1
alpha0	0.017	$x \in \mathbb{R}, x \ge 0$	Initial low light conditions	$ m mg~J^{-1}$
Kx .	0.7	$x \in \mathbb{R}, x > 0$	Extinction coefficient PAR	1
gmin	$2.5 \cdot 10^{-4}$	$x \in \mathbb{R}, x \ge 0$	Cuticular (minimum) conductance	$m s^{-1}$
Cw	$1.6 \cdot 10^{-3}$	$x \in \mathbb{R}, x \ge 0$	Constant water stress correction	_
wsmax	0.55	$x \in \mathbb{R}, x > 0$	Upper reference value soil water	_
wsmin R10	0.005 0.23	$x \in \mathbb{R}, x \ge 0$	Lower reference value soil water	$ \frac{1}{100} \text{ mg m}^{-2} \text{ s}^{-1} $
KIU	0.23	$x \in \mathbb{R}, x > 0$	Respiration at 10 °C	Ing III - S
1	I.	n e e e e e e e e e e e e e e e e e e e	T. Control of the Con	1

2.6 Namelist NAMCHEM

Option	Default	Possible values	Description	Unit
lchem lcomplex	.false.	$x \in \{.\text{false.}, .\text{true.}\}$ $x \in \{.\text{false.}, .\text{true.}\}$	Enable or disable chemistry Choice between complex chemical	- -
lwritepl	.true.	$x \in \{.\text{false.},.\text{true.}\}$	scheme and simplified scheme Enable the output of production and loss terms per chemical	-
ldiuvar	.true.	$x \in \{.\text{false.},.\text{true.}\}$	If false the UV radiation during day is calculating at time h_ref	-
h_ref	12	$x \in \mathbb{R}, 0 \le x \le 24$	Reference time for calculated UV radiation if ldiuvar is set to .false.	hr
lflux	.false.	$x \in \{.\text{false.},.\text{true.}\}$	If set to .true. the times of sunrise and sunset are input. The options in Paragraph 2.7 are preferred.	-
fluxstart	0	$x \in \mathbb{R}, 0 \le x \le 24$	Time of sunrise if Iflux is set to .true.	hr
fluxend	0	$x \in \mathbb{R}, 0 \le x \le 24$	Time of sunset if Iflux is set to .true.	hr
pressure_ft	pressure	$x \in \mathbb{R}, x > 0$	Air pressure in the free troposphere	Pa
lchconst	.false.	$x \in \{\text{.false., .true.}\}$	Switch to calculate reaction rates	_
			using reference temperatures, hu-	
			midities and pressures instead of actual values	
t_ref_cbl	298	$x \in \mathbb{R}, x > 0$	Reference temperature in the boundary layer	K
p_ref_cbl	1013.5	$x \in \mathbb{R}, x > 0$	Reference pressure in the boundary layer	Pa
q_ref_cbl	10	$x \in \mathbb{R}, x \ge 0$	Reference specific humidity in the boundary layer	$g kg^{-1}$
t_ref_ft	298	$x \in \mathbb{R}, x > 0$	Reference temperature in the free troposphere	K
p_ref_ft	1013.5	$x \in \mathbb{R}, x > 0$	Reference pressure in the free tro- posphere	Pa
q_ref_ft	10	$x \in \mathbb{R}, x \ge 0$	Reference specific humidity in the free troposphere	g kg ⁻¹

2.7 Namelist NAMFLUX

Option	Default	Possible values	Description	Unit
offset_wt	0	$x \in \mathbb{R}$ $x \in \mathbb{R}$	Offset for the kinematic heat flux Offset for the kinematic moisture flux	$ m K \ m \ s^{-1}$ $ m g \ kg^{-1} \ m \ s^{-1}$
function_wt	2	$x \in \{0, 1, 2, 3\}$	Shape of the kinematic heat flux 0 = No flux 1 = Constant flux 2 = Sinusoid flux evolution with a start and an end time 3 = Constant flux with a start and an end time 4 = Cosine shaped flux with a start and an end time. Equal to 0 at start and end and to wthetasmax in the middle. (Standard cosine is multiplied by -wthetasmax/2 and shifted by wthetasmas/2.)	-
function_wq	2	$x \in \{0, 1, 2, 3\}$	Shape of the kinematic moisture flux (see function_wt)	-
starttime_wt	sunrise	$x \in \mathbb{R}, 0 \le x \le 86400$	Time after which the heat flux starts in case of functions 2 and 3	s
endtime_wt	sunset	$x \in \mathbb{R}, 0 \le x \le 86400$	Time after which the heat flux ends in case of functions 2 and 3	s
starttime_wq	sunrise	$x \in \mathbb{R}, 0 \le x \le 86400$	Time after which the moisture flux starts in case of functions 2 and 3	S
endtime_wq	sunset	$x \in \mathbb{R}, 0 \le x \le 86400$	Time after which the moisture flux ends in case of functions 2 and 3	S

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Option	Default	Possibl	e values	Description	Unit
starttime_adv	sunrise	$x \in \mathbb{R}, 0 \le$	$\leq x \leq 86400$	Time after which the advection of potential temperature	s
endtime_adv	sunset	$x \in \mathbb{R}, 0$	$\leq x \leq 86400$	and moisture starts Time after which the advection of potential temperature	s
starttime_chem	sunrise	$x \in \mathbb{R}, 0$	$\leq x \leq 86400$	and moisture ends Time after which the chemical emissions start in case of	s
endtime_chem	sunset	$x \in \mathbb{R}, 0$	$\leq x \leq 86400$	functions 2 and 3 Time after which the chemical emissions end in case of functions 2 and 3	S

Chapter 3

Chemical mechanisms: input files

In this chapter the other files with input than the file with options in namelists, namoptions, are discussed. In case of *reduced* chemistry, we describe only one files **chem.inp**. In case of *complex* chemistry two files are required: **chem.inp** and **chemicals.txt**

Please note that the chem.inp files are different for *reduced* and *complex*. In Section 3.1 the *reduced* chemistry is explained and the input files for *complex* chemistry are described in Section 3.2.

3.1 Input files for reduced chemistry

For chemistry in this setup 1 extra file is needed, chem.inp. In this file, lines starting with an # are considered comments. The first uncommented line should start with an % followed by two integers. The first integer denotes the total number of chemicals (including passive tracers) and the second integer represents the number of reactions specified. These numbers are upper limits, so the exact numbers are not important but they should be large enough. The second uncommented line should start with an @. This is followed by a line with the names of all chemicals. The row below that the corresponding initial boundary layer concentrations are specified in ppb, followed by a row with the initial free tropospheric concentrations in ppb. The line below that specifies in order the emissions of the chemicals at the surface in ppb m s $^{-1}$. The line after that consists of a row of integers, specifying the shape of the emission evolution for the corresponding concentrations. These integers correspond to the same functions as specified in Section 2.7. The functions are:

- 0: No emission at all
- 1: A constant emission
- 2: An emission that evolves in time as a sinusoid. It is 0 until starttime_chem (Paragraph 2.7) and after endtime_chem. In between the emission is a sinusoid whos phase ranges from 0 to π . The maximum value of the sinusoid is given by the emission value listed in the line above.
- 3: An emission that is constant during daytime. It is 0 until starttime_chem (Paragraph 2.7) and after endtime_chem. In between the emission is constantly equal to the emission value listed in the line above.
- 4: An emission that evolves in time as a cosine. It is 0 until starttime_chem (Paragraph 2.7) and after endtime_chem. In between the emission behaves like a flipped cosine (a cosine with a phase ranging from $-\pi$ to π) with a mean and an amplitude which are both equal to half the emission value listed in the line above.

5: Dry deposition of atmospheric compounds according to $-v_c C$. The deposition velocity v_c in m s⁻¹ needs to be specified.

After these specific lines all other lines denote chemical reactions that are evaluated. MXLCH stops reading the input file after it encounters a \$ at the beginning of a line. A line starting with \$ is needed after all chemical reactions are specified.

The lines with the chemical reactions consist out of 12 columns, listing in order the variables kr2nd, name, raddep, func1, A, B, C, D, E, F, G and reaction. The meaning of these columns is:

kr2nd: Deprecated: put a random number here

name: Name of the reaction: used for output

raddep: Flag for UV-radiation dependent reaction constants: 1 = true, 0 = false

func1: Flag to choose the definition of the reaction rate constant, *k*:

```
\begin{array}{llll} & \text{If raddep} = 0: \\ & \text{func1} = 1 & : & k = A \ [\text{cm}^3 \ \text{mol}^{-1} \ \text{s}^{-1}] \\ & \text{func1} = 2 & : & k = A \cdot \exp^{B/T} \ [\text{cm}^3 \ \text{mol}^{-1} \ \text{s}^{-1}] \\ & \text{func1} = 3 & : & k = A \cdot (T/B)^C \cdot \exp^{D/T} \ [\text{cm}^3 \ \text{mol}^{-1} \ \text{s}^{-1}] \\ & \text{func1} = 4 & : & k = \frac{k' \cdot k''}{k' + k''} \cdot G \ [\text{cm}^3 \ \text{mol}^{-1} \ \text{s}^{-1}] \\ & & k' = A \cdot (T/300)^B \cdot \exp^{C/T} \cdot [M] \\ & & k'' = D \cdot (T/300)^E \cdot \exp^{F/T} \\ & \text{func1} = 5 & : & \text{like func1} = 4, \text{but in } [\text{s}^{-1}] \\ & \text{func1} = 6 & : & k = (k' + k'') \cdot (1 + k''') \ [\text{cm}^3 \ \text{mol}^{-1} \ \text{s}^{-1}] \\ & : & k' = A \cdot \exp^{B/T} \\ & : & k'' = A \cdot \exp^{B/T} \cdot [M] \\ & : & k''' = 1 + E \cdot \exp^{F/T} \cdot [H20] \\ & \text{func1} = 7 & : & k = A \cdot (T/B)^C \cdot \exp^{D/T} \ [\text{cm}^6 \ \text{mol}^{-2} \ \text{s}^{-1}] ) \\ & \text{If raddep} = 1: \\ & \text{func1} = 1 & : & k = A \ [\text{s}^{-1}] \\ & \text{func1} = 2 & : & k = A \cdot \exp^{B/\cos(\chi)} \ [\text{s}^{-1}] \\ & \text{func1} = 3 & : & k = A \cdot \cos(\chi)^B \ [\text{s}^{-1}] \\ & \text{func1} = 4 & : & k = A \cdot (\cos(\chi)^B) \cdot \frac{D \cdot [H2O]}{D \cdot [H2O] + E \cdot [M]} \ [\text{s}^{-1}] \end{array}
```

In these equations, [H2O] and [M] are the concentrations of water and air molecules in the ambient air and χ is the solar zenith angle.

A - G: Constants used for the calculations

reaction: The chemical equation. On the left hand side input chemicals are written with their stoichiometric coefficients. The chemicals are separated by ' + '. On the right hand side the output chemicals are written in a similar way. Input and output chemicals are separated by ' -> '. If chemicals are enclosed in brackets they are just shown for information and are not used in the calculations. The stoichiometric coefficients for the input chemicals have to be integers.

An example chem.inp file describing a system of 2 reactions and 3 chemicals is shown on the next page.

```
# Input file for chemistry
# Reactants Reactions
%32
# Chemicals
@1
      2
           3
      O3
           O1D
 O2
 0.2e0930.
           0.0
 0.2e0930.
           0.0
      1.e-3 0.0
 0
 0
      2
           0
# Kr2nd Name Rad func1 A
                               C D E F G Reaction
                          В
      R01
          1
               2
                   3.83e-5 -.575 1.0 1.0 1.0 1.0 03 + (hv) -> O1D + O2
  2
      R02
          0
               2
                   $ end of chemical reactions
```

3.2 Input files for complex chemistry

For complex chemistry two input files are used next to the file 'namoptions'. These are chem.inp and chemicals.txt and are treated in Paragraphs 3.2.1 and 3.2.2 respectively.

3.2.1 chem.inp

This file differs from the chem.inp file for simplified chemistry as described in Paragraph 3.1. Again lines starting with an # are considered comments. The first uncommented line should start with an % followed by one integer which denotes the total number of chemical reactions specified. This number is an upper limits, so the exact number is not important if large enough. The lines after that are input for chemical reactions. MXLCH stops reading the input file after it encounters a \$ at the beginning of a line. A line starting with \$ is needed after all chemical reactions are specified.

This input is provided in blocks of 2 lines. The first line contains the chemical equation. On the left hand side input chemicals are written with their stoichiometric coefficients. The chemicals are separated by '+'. On the right hand side the output chemicals are written in a similar way. Input and output chemicals are separated by '->'. If chemicals are enclosed in brackets they are just shown for information and are not used in the calculations. The stoichiometric coefficients for the input chemicals have to be integers. The second line contains, in order, the variables name, raddep, func1, A, B, C, D, E, F and G. A to G are constants that can be used for the calculation of the reaction rate constant. Other than for the simplified chemistry, for complex chemistry the constants have to be specified only if they are used. So if the reaction rate constant is calculated using 3 constants, D - G don't have to be specified. The name is used for the output and raddep is an integer set to either 1 or 0, representing whether the reaction is or is not a photolysis reaction using UV light. The variable func1 is a flag to choose the definition of the reaction rate constant,k:

```
If \operatorname{raddep} = 0: \operatorname{func1} = 0 : k = A \left[ \operatorname{cm}^{3N} \operatorname{mol}^{-N} \operatorname{s}^{-1} \right] \operatorname{func1} = 1 : k = A \cdot \exp^{B/T} \left[ \operatorname{cm}^{3N} \operatorname{mol}^{-N} \operatorname{s}^{-1} \right] \operatorname{func1} = 2 : k = A \cdot (B/T)^C \left[ \operatorname{cm}^{3N} \operatorname{mol}^{-N} \operatorname{s}^{-1} \right] \operatorname{func1} = 3 : k = A \cdot (T/B)^C \cdot \exp^{D/T} \left[ \operatorname{cm}^{3N} \operatorname{mol}^{-N} \operatorname{s}^{-1} \right] \operatorname{func1} = 4 : k = \frac{k' \cdot k''}{k' + k''} \cdot G \left[ \operatorname{cm}^3 \operatorname{mol}^{-1} \operatorname{s}^{-1} \right] k' = A \cdot (T/300)^B \cdot \exp^{C/T} \cdot [M] k'' = D \cdot (T/300)^E \cdot \exp^{F/T} \operatorname{func1} = 5 : \operatorname{like} \operatorname{func1} = 4, \operatorname{but} \operatorname{in} \left[ \operatorname{s}^{-1} \right] \operatorname{func1} = 6 : k = (k' + k'') \cdot (1 + k''') \left[ \operatorname{cm}^3 \operatorname{mol}^{-1} \operatorname{s}^{-1} \right]) : k' = A \cdot \exp^{B/T} : k'' = 1 + E \cdot \exp^{D/T} \cdot [M] : k''' = 1 + E \cdot \exp^{F/T} \cdot [H20] \operatorname{func1} = 7 : k = A \cdot (T/B)^C \cdot \exp^{D/T} \left[ \operatorname{cm}^6 \operatorname{mol}^{-2} \operatorname{s}^{-1} \right] If \operatorname{raddep} = 1:
```

```
\begin{array}{lll} \mathrm{func1} = 1 & : & k = A \, [\mathrm{s}^{-1}] \\ \mathrm{func1} = 2 & : & k = A \cdot \exp^{B/\cos(\chi)} \, [\mathrm{s}^{-1}] \\ \mathrm{func1} = 3 & : & k = A \cdot \cos(\chi)^B \, [\mathrm{s}^{-1}] \\ \mathrm{func1} = 4 & : & k = A \cdot (\cos(\chi)^B) \cdot \frac{D \cdot [H2O]}{D \cdot [H2O] + E \cdot [M]} \, [\mathrm{s}^{-1}] \\ \mathrm{func1} = 5 & : & k = A \cdot \cos(\chi)^B \cdot \exp^{-C/\cos(\chi)} \, [\mathrm{s}^{-1}] \\ \mathrm{func1} = 6 & : & k = A \cdot B^{\cos(\chi)} \cdot \cos(\chi)^C \, [\mathrm{s}^{-1}] \end{array}
```

In these equations, [H2O] and [M] are the concentrations of water and air molecules in the ambient air and χ is the solar zenith angle. N is equal to the total reaction order minus 1. An example chem.inp file describing a system of 2 reactions is shown below.

```
# Input file for chemical reactions
# Amount of reactions
% 2
# Reaction
                           func1
                                              В
                                                       C
                                                                 D
                                                                          E
                                                                                    F
                                                                                             G
        Name
                  Rad
                                     Α
  O_3
           + (hv)
                   -> O1D
                                + O2
                  1
                                     3.83e-05 -0.575
        R01
  O<sub>1</sub>D
           + O2
                    -> O3
        R02
                  0
                                     3.30e-11 55
$ end of chemical reactions
```

3.2.2 chemicals.txt

When using the complex chemical scheme, the chemicals are specified in the file chemicals.txt. Again lines starting with an # are considered comments. The first uncommented line should start with an % followed by one integer which denotes the total number of chemical species. This number is an upper limits, so the exact number is not important if large enough. The lines after that are input for the chemical species and their initial concentrations and their emissions. MXLCH stops reading the input file after it encounters a \$ at the beginning of a line. A line starting with \$ is needed after all chemical species are specified.

The lines contain, in order, the species name, the initial concentration in the boundary layer in ppb, the initial concentration in the free troposphere in ppb, the emission at the surface in ppb m $\rm s^{-1}$, an integer specifying the shape of the emission evolution and an

integer that specifies whether production and loss statistics should be generated for that species (0 = write no output, 1 = write output). The integer specifying the shape of the emission evolution is exactly equal to the one described in Paragraph 3.1, so

- 0: No emission at all
- 1: A constant emission
- 2: An emission that evolves in time as a sinusoid. It is 0 until starttime_chem (Paragraph 2.7) and after endtime_chem. In between the emission is a sinusoid whos phase ranges from 0 to π . The maximum value of the sinusoid is given by the emission value listed in the line above.
- 3: An emission that is constant during daytime. It is 0 until starttime_chem (Paragraph 2.7) and after endtime_chem. In between the emission is constantly equal to the emission value listed in the line above.
- 4: An emission that evolves in time as a cosine. It is 0 until starttime_chem (Paragraph 2.7) and after endtime_chem. In between the emission behaves like a flipped cosine (a cosine whos phase ranges from $-\pi$ to π) with a mean and an amplitude which are both equal to half the emission value listed in the line above.
- 5: Dry deposition of atmospheric compounds according to $-v_c C$. The deposition velocity v_c in m s⁻¹ needs to be specified.

An example chemicals.txt file describing a system with 3 chemical species is shown below.

Input file for chemical species

Amount of compounds

%3

# Chemical	C_CBL	C_FT	Emission	Function	Statistics
O2	0.2e09	0.2e09	0.0	0	1
O3	30.0	30.0	1.e-3	2	1
O1D	0.0	0.0	0.0	0	1

\$ end of chemical species