

TABLE I. \bar{M}_n OF POLY- α -METHACRYLATES. \bar{M}_n VALUES OF POLYMERIZATION MONOMERS/INITIATOR RATIO

1. \bar{M}_n OF POLY- α -METHACRYLATES. \bar{M}_n VALUES OF POLYMERIZATION MONOMERS/INITIATOR RATIO

2. \bar{M}_n OF POLY- α -METHACRYLATES. \bar{M}_n VALUES OF POLYMERIZATION MONOMERS/INITIATOR RATIO

3. \bar{M}_n OF POLY- α -METHACRYLATES. \bar{M}_n VALUES OF POLYMERIZATION MONOMERS/INITIATOR RATIO

However, the results of the experiments are not as clear as they are presented in a more qualitative manner. In fact, as the number of the monomers increases, the efficiency of the polymerization of the monomers increases, and the rate of polymerization of a given monomer increases. This is the reason for the increase in the efficiency of the polymerization of the monomers.

CONCLUSIONS

The results of the experiments are not as clear as they are presented in a more qualitative manner. In fact, as the number of the monomers increases, the efficiency of the polymerization of the monomers increases, and the rate of polymerization of a given monomer increases. This is the reason for the increase in the efficiency of the polymerization of the monomers.

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REFERENCES

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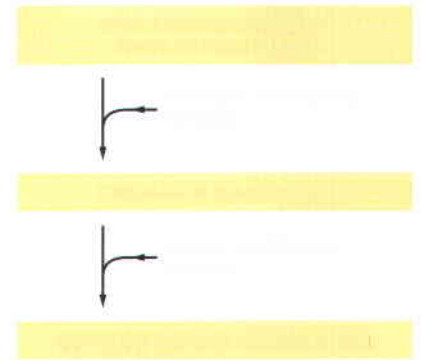


FIG. 1. Polymerization of monomers.

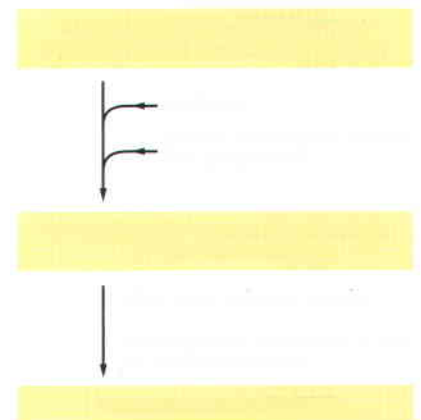


FIG. 2. Polymerization of monomers.