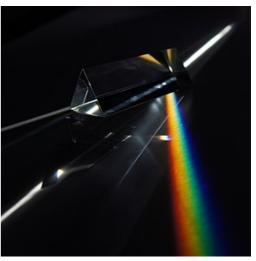


Light

Light or **visible light** is <u>electromagnetic radiation</u> that can be perceived by the <u>human eye</u>. Visible light is usually defined as having <u>wavelengths</u> in the range of 400–700 nanometres (nm), corresponding to <u>frequencies</u> of 750–420 <u>terahertz</u>, between the <u>infrared</u> (with longer wavelengths) and the ultraviolet (with shorter wavelengths).

In physics, the term "light" may refer more broadly to electromagnetic radiation of any wavelength, whether visible or not. [4][5] In this sense, gamma rays, X-rays, microwaves and radio waves are also light. The primary properties of light are intensity, propagation direction, frequency or wavelength spectrum and polarization. Its speed in vacuum, 299 792 458 m/s, is one of the fundamental constants of nature. [6] Like all types of electromagnetic radiation, visible light propagates by massless elementary particles called photons that represents the quanta of electromagnetic field, and can be analyzed as both waves and particles. The study of light, known as optics, is an important research area in modern physics.

The main source of natural light on Earth is the \underline{Sun} . Historically, another important source of light for humans has been \underline{fire} , from ancient campfires to modern $\underline{kerosene\ lamps}$. With the development of $\underline{electric\ lights}$ and power systems, electric lighting has effectively replaced firelight.



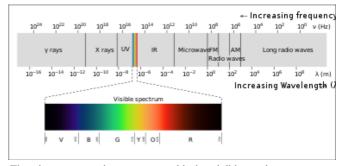
A triangular <u>prism</u> <u>dispersing</u> a beam of white light. The longer wavelengths (red) and the shorter wavelengths (green-blue) are separated.

Electromagnetic spectrum and visible light

Generally, <u>electromagnetic radiation</u> (EMR) is classified by wavelength into <u>radio</u> waves, <u>microwaves</u>, <u>infrared</u>, the <u>visible spectrum</u> that we perceive as light, <u>ultraviolet</u>, <u>X-rays</u> and <u>gamma rays</u>. The designation "<u>radiation</u>" excludes static electric, magnetic and near fields.

The behavior of EMR depends on its wavelength. Higher frequencies have shorter wavelengths and lower frequencies have longer wavelengths. When EMR interacts with single atoms and molecules, its behavior depends on the amount of energy per quantum it carries.

EMR in the visible light region consists of <u>quanta</u> (called <u>photons</u>) that are at the lower end of the energies that are



The $\underline{\text{electromagnetic spectrum}},$ with the $\underline{\text{visible portion}}$ highlighted

capable of causing electronic excitation within molecules, which leads to changes in the bonding or chemistry of the molecule. At the lower end of the visible light spectrum, EMR becomes invisible to humans (infrared) because its photons no longer have enough individual energy to cause a lasting molecular change (a change in conformation) in the visual molecule retinal in the human retina, which change triggers the sensation of vision.

There exist animals that are sensitive to various types of infrared, but not by means of quantum-absorption. <u>Infrared sensing in snakes</u> depends on a kind of natural <u>thermal imaging</u>, in which tiny packets of cellular water are raised in temperature by the infrared radiation. EMR in this range causes molecular vibration and heating effects, which is how these animals detect it.

Above the range of visible light, ultraviolet light becomes invisible to humans, mostly because it is absorbed by the comea below 360 <u>nm</u> and the internal lens below 400 nm. Furthermore, the <u>rods</u> and <u>cones</u> located in the <u>retina</u> of the human eye cannot detect the very short (below 360 nm) ultraviolet wavelengths and are in fact damaged by ultraviolet. Many animals

with eyes that do not require lenses (such as insects and shrimp) are able to detect ultraviolet, by quantum photon-absorption mechanisms, in much the same chemical way that humans detect visible light.

Various sources define visible light as narrowly as $420-680 \text{ nm} \cdot \frac{[7][8]}{[13]}$ to as broadly as $380-800 \text{ nm} \cdot \frac{[9][10]}{[10]}$ Under ideal laboratory conditions, people can see infrared up to at least 1,050 nm; children and young adults may perceive ultraviolet wavelengths down to about $310-313 \text{ nm} \cdot \frac{[12][13][14]}{[12][13][14]}$

Plant growth is also affected by the colour spectrum of light, a process known as photomorphogenesis.

Speed of light

The speed of light in <u>vacuum</u> is defined to be exactly 299 792 458 <u>m/s</u> (approx. 186,282 miles per second). The fixed value of the speed of light in SI units results from the fact that the metre is now defined in terms of the speed of light. All forms of electromagnetic radiation move at exactly this same speed in vacuum.

Different <u>physicists</u> have attempted to measure the speed of light throughout history. <u>Galileo</u> attempted to measure the speed of light in the seventeenth century. An early experiment to measure the speed of light was conducted by <u>Ole Rømer</u>, a Danish physicist, in 1676. Using a <u>telescope</u>, Rømer observed the motions of <u>Jupiter</u> and one of its <u>moons</u>, <u>Io</u>. Noting discrepancies in the apparent period of Io's orbit, he calculated that light takes about 22 minutes to traverse the diameter of Earth's orbit. However, its size was not known at that time. If Rømer had known the diameter of the Earth's orbit, he would have calculated a speed of 227 000 000 m/s.



Beam of sun light inside the cavity of Rocca ill'Abissu at <u>Fondachelli-</u>Fantina, Sicily

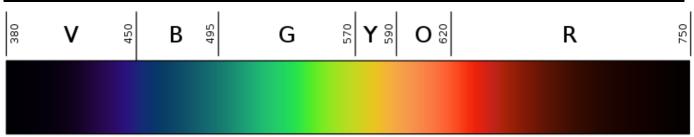
Another more accurate measurement of the speed of light was performed in Europe by <u>Hippolyte Fizeau</u> in 1849. Fizeau directed a beam of light at a mirror several kilometers away. A rotating <u>cog wheel</u> was placed in the path of the light beam as it traveled from the source, to the mirror and then returned to its origin. Fizeau found that at a certain rate of rotation, the beam would pass through one gap in the wheel on the way out and the next gap on the way back. Knowing the distance to the mirror, the number of teeth on the wheel and the rate of rotation, Fizeau was able to calculate the speed of light as 313 000 000 m/s.

<u>Léon Foucault</u> carried out an experiment which used rotating mirrors to obtain a value of 298 000 000 m/s $^{[16]}$ in 1862. <u>Albert A. Michelson</u> conducted experiments on the speed of light from 1877 until his death in 1931. He refined Foucault's methods in 1926 using improved rotating mirrors to measure the time it took light to make a round trip from <u>Mount Wilson</u> to <u>Mount San Antonio</u> in California. The precise measurements yielded a speed of 299 796 000 m/s $^{[17]}$

The effective velocity of light in various transparent substances containing ordinary <u>matter</u>, is less than in vacuum. For example, the speed of light in water is about 3/4 of that in vacuum.

Two independent teams of physicists were said to bring light to a "complete standstill" by passing it through a <u>Bose–Einstein condensate</u> of the element <u>rubidium</u>, one team at <u>Harvard University</u> and the <u>Rowland Institute for Science</u> in Cambridge, Massachusetts and the other at the <u>Harvard–Smithsonian Center for Astrophysics</u>, also in Cambridge. However, the popular description of light being "stopped" in these experiments refers only to light being stored in the excited states of atoms, then re-emitted at an arbitrary later time, as stimulated by a second laser pulse. During the time it had "stopped", it had ceased to be light.

Optics



The study of light and the interaction of light and $\underline{\text{matter}}$ is termed $\underline{\text{optics}}$. The observation and study of $\underline{\text{optical phenomena}}$ such as rainbows and the aurora borealis offer many clues as to the nature of light.

A <u>transparent</u> object allows light to <u>transmit</u> or pass through. Conversely, an <u>opaque</u> object does not allow light to transmit through and instead <u>reflecting</u> or <u>absorbing</u> the light it receives. Most objects do not reflect or transmit light <u>specularly</u> and to some degree <u>scatters</u> the incoming light, which is called <u>glossiness</u>. Surface scatterance is caused by the <u>surface roughness</u> of the reflecting surfaces, and internal scatterance is caused by the difference of <u>refractive index</u> between the particles and <u>medium</u> inside the object. Like transparent objects, <u>translucent</u> objects allow light to transmit through, but translucent objects also scatter certain wavelength of light via internal scatterance.

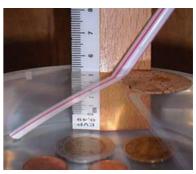
Refraction

Refraction is the bending of light rays when passing through a surface between one transparent material and another. It is described by Snell's Law:

$$n_1\sin\theta_1=n_2\sin\theta_2.$$

where θ_1 is the angle between the ray and the surface <u>normal</u> in the first medium, θ_2 is the angle between the ray and the surface normal in the second medium and n_1 and n_2 are the <u>indices of refraction</u>, n=1 in a <u>vacuum</u> and n>1 in a <u>transparent</u> substance.

When a beam of light crosses the boundary between a vacuum and another medium, or between two different media, the wavelength of the light changes, but the frequency remains constant. If the beam of light is not <u>orthogonal</u> (or rather normal) to the boundary, the change in wavelength results in a change in the direction of the beam. This change of direction is known as <u>refraction</u>.



Due to refraction, the straw dipped in water appears bent and the ruler scale compressed when viewed from a shallow angle.

The refractive quality of <u>lenses</u> is frequently used to manipulate light in order to change the apparent size of images. Magnifying glasses, spectacles, contact lenses, microscopes and refracting telescopes are all examples of this manipulation.

Light sources

There are many sources of light. A body at a given temperature emits a characteristic spectrum of <u>black-body radiation</u>. A simple thermal source is <u>sunlight</u>, the radiation emitted by the <u>chromosphere</u> of the <u>Sun</u> at around 6,000 <u>K</u> (5,730 °C; 10,340 °F). Solar radiation peaks in the visible region of the <u>electromagnetic spectrum</u> when plotted in wavelength units, $\overline{^{[20]}}$ and roughly 44% of the radiation that reaches the ground is visible. Another example is <u>incandescent light bulbs</u>, which emit only around 10% of their energy as visible light and the remainder as infrared. A common thermal light source in history is the glowing solid particles in <u>flames</u>, but these also emit most of their radiation in the infrared and only a fraction in the visible spectrum.

The peak of the black-body spectrum is in the deep infrared, at about 10 <u>micrometre</u> wavelength, for relatively cool objects like human beings. As the temperature increases, the peak shifts to shorter wavelengths, producing first a red glow, then a white one and finally a blue-white colour as the peak moves out of the visible part of the spectrum and into the ultraviolet. These colours can be seen when metal is heated to "red hot" or "white hot". Blue-white <u>thermal emission</u> is not often seen, except in stars (the commonly seen pure-blue colour in a <u>gas</u> flame or a <u>welder</u>'s torch is in fact due to molecular emission, notably by CH radicals emitting a wavelength band around 425 nm and is not seen in stars or pure thermal radiation).

Atoms emit and absorb light at characteristic energies. This produces "emission lines" in the spectrum of each atom. Emission can be spontaneous, as in light-emitting diodes, gas discharge lamps (such as neon lamps and neon signs, mercury-vapor lamps, etc.) and flames (light from the hot gas itself—so, for example, sodium in a gas flame emits characteristic yellow light). Emission can also be stimulated, as in a laser or a microwave maser.

Deceleration of a free charged particle, such as an <u>electron</u>, can produce visible radiation: <u>cyclotron radiation</u>, <u>synchrotron radiation</u> and <u>bremsstrahlung</u> radiation are all examples of this. Particles moving through a medium faster than the speed of light in that medium can produce visible <u>Cherenkov radiation</u>. Certain chemicals produce visible radiation by <u>chemoluminescence</u>. In living things, this process is called <u>bioluminescence</u>. For example, <u>fireflies</u> produce light by this means and boats moving through water can disturb plankton which produce a glowing wake.

Certain substances produce light when they are illuminated by more energetic radiation, a process known as <u>fluorescence</u>. Some substances emit light slowly after excitation by more energetic radiation. This is known as <u>phosphorescence</u>. Phosphorescent materials can also be excited by bombarding them with subatomic particles. <u>Cathodoluminescence</u> is one

example. This mechanism is used in cathode-ray tube television sets and computer monitors.

Certain other mechanisms can produce light:

- Bioluminescence
- Cherenkov radiation
- Electroluminescence
- Scintillation
- Sonoluminescence
- Triboluminescence

When the concept of light is intended to include very-high-energy photons (gamma rays), additional generation mechanisms include:



Hong Kong illuminated by colourful artificial lighting

- Particle—antiparticle annihilation
- Radioactive decay

Measurement

Light is measured with two main alternative sets of units: <u>radiometry</u> consists of measurements of light power at all wavelengths, while <u>photometry</u> measures light with wavelength weighted with respect to a standardized model of human brightness perception. Photometry is useful, for example, to quantify Illumination (lighting) intended for human use.

The photometry units are different from most systems of physical units in that they take into account how the human eye responds to light. The <u>cone cells</u> in the human eye are of three types which respond differently across the visible spectrum and the cumulative response peaks at a wavelength of around 555 nm. Therefore, two sources of light which produce the same intensity (W/m²) of visible light do not necessarily appear equally bright. The photometry units are designed to take this into account and therefore are a better representation of how "bright" a light appears to be than raw intensity. They relate to raw <u>power</u> by a quantity called <u>luminous efficacy</u> and are used for purposes like determining how to best achieve sufficient illumination for various tasks in indoor and outdoor settings. The illumination measured by a <u>photocell</u> sensor does not necessarily correspond to what is perceived by the human eye and without filters which may be costly, photocells and <u>charge-coupled</u> devices (CCD) tend to respond to some infrared, ultraviolet or both.

Light pressure

Light exerts physical pressure on objects in its path, a phenomenon which can be deduced by Maxwell's equations, but can be more easily explained by the particle nature of light: photons strike and transfer their momentum. Light pressure is equal to the power of the light beam divided by c, the speed of light. Due to the magnitude of c, the effect of light pressure is negligible for everyday objects. For example, a one-milliwatt <u>laser pointer</u> exerts a force of about 3.3 <u>piconewtons</u> on the object being illuminated; thus, one could lift a <u>U.S. penny</u> with laser pointers, but doing so would require about 30 billion 1-mW laser pointers. However, in <u>nanometre</u>-scale applications such as <u>nanoelectromechanical systems</u> (NEMS), the effect of light pressure is more significant and exploiting light pressure to drive NEMS mechanisms and to flip nanometre-scale physical switches in integrated circuits is an active area of research. At larger scales, light pressure can cause <u>asteroids</u> to spin faster, acting on their irregular shapes as on the vanes of a <u>windmill</u>. The possibility of making <u>solar sails</u> that would accelerate spaceships in space is also under investigation.

Although the motion of the <u>Crookes radiometer</u> was originally attributed to light pressure, this interpretation is incorrect; the characteristic Crookes rotation is the result of a partial vacuum. This should not be confused with the <u>Nichols radiometer</u>, in which the (slight) motion caused by torque (though not enough for full rotation against friction) is directly caused by light pressure. As a consequence of light pressure, <u>Einstein</u> in 1909 predicted the existence of "radiation friction" which would oppose the movement of matter. He wrote, "radiation will exert pressure on both sides of the plate. The forces of pressure exerted on the two sides are equal if the plate is at rest. However, if it is in motion, more radiation will be reflected on the surface that is ahead during the motion (front surface) than on the back surface. The backwardacting force of pressure exerted on the front surface is thus larger than the force of pressure acting on the back. Hence, as the resultant of the two forces, there remains a force that counteracts the motion of the plate and that increases with the velocity of the plate. We will call this resultant 'radiation friction' in brief."

Usually light momentum is aligned with its direction of motion. However, for example in <u>evanescent waves</u> momentum is transverse to direction of propagation. [30]

Historical theories about light, in chronological order

Classical Greece and Hellenism

In the fifth century BC, <u>Empedocles</u> postulated that everything was composed of <u>four elements</u>; fire, air, earth and water. He believed that <u>Aphrodite</u> made the human eye out of the four elements and that she lit the fire in the eye which shone out from the eye making sight possible. If this were true, then one could see during the night just as well as during the day, so Empedocles postulated an interaction between rays from the eyes and rays from a source such as the sun. [31]

In about 300 BC, <u>Euclid</u> wrote *Optica*, in which he studied the properties of light. Euclid postulated that light travelled in straight lines and he described the laws of reflection and studied them mathematically. He questioned that sight is the result of a beam from the eye, for he asks how one sees the stars immediately, if one closes one's eyes, then opens them at night. If the beam from the eye travels infinitely fast this is not a problem. [32]

In 55 BC, <u>Lucretius</u>, a Roman who carried on the ideas of earlier Greek <u>atomists</u>, wrote that "The light & heat of the sun; these are composed of minute atoms which, when they are shoved off, lose no time in shooting right across the interspace of air in the direction imparted by the shove." (from *On the nature of the Universe*). Despite being similar to later particle theories, Lucretius's views were not generally accepted. <u>Ptolemy</u> (c. second century) wrote about the <u>refraction</u> of light in his book *Optics*. [33]

Classical India

In ancient India, the <u>Hindu</u> schools of <u>Samkhya</u> and <u>Vaisheshika</u>, from around the early centuries AD developed theories on light. According to the Samkhya school, light is one of the five fundamental "subtle" elements (tanmatra) out of which emerge the gross elements. The <u>atomicity</u> of these elements is not specifically mentioned and it appears that they were actually taken to be continuous. [34] On the other hand, the Vaisheshika school gives an <u>atomic theory</u> of the physical world on the non-atomic ground of <u>ether</u>, space and time. (See <u>Indian atomism</u>.) The basic atoms are those of earth (prthivi), water (pani), fire (agni) and air (vayu) Light rays are taken to be a stream of high velocity of tejas (fire) atoms. The particles of light can exhibit different characteristics depending on the speed and the arrangements of the tejas atoms. The <u>Vishnu Purana</u> refers to sunlight as "the seven rays of the sun". [34]

The Indian $\underline{\text{Buddhists}}$, such as $\underline{\text{Dign\bar{a}ga}}$ in the fifth century and $\underline{\text{Dharmakirti}}$ in the seventh century, developed a type of atomism that is a philosophy about reality being composed of atomic entities that are momentary flashes of light or energy. They viewed light as being an atomic entity equivalent to energy.

Descartes

René Descartes (1596–1650) held that light was a mechanical property of the luminous body, rejecting the "forms" of Ibn al-Haytham and Witelo as well as the "species" of Bacon, Grosseteste and Kepler. In 1637 he published a theory of the refraction of light that assumed, incorrectly, that light travelled faster in a denser medium than in a less dense medium. Descartes arrived at this conclusion by analogy with the behaviour of sound waves. Although Descartes was incorrect about the relative speeds, he was correct in assuming that light behaved like a wave and in concluding that refraction could be explained by the speed of light in different media.

Descartes is not the first to use the mechanical analogies but because he clearly asserts that light is only a mechanical property of the luminous body and the transmitting medium, Descartes's theory of light is regarded as the start of modern physical ontics. [35]

Particle theory

<u>Pierre Gassendi</u> (1592–1655), an atomist, proposed a particle theory of light which was published posthumously in the 1660s. <u>Isaac Newton</u> studied Gassendi's work at an early age and preferred his view to Descartes's theory of the *plenum*. He stated in his *Hypothesis of Light* of 1675 that light was composed of corpuscles (particles of matter) which were emitted in all

directions from a source. One of Newton's arguments against the wave nature of light was that waves were known to bend around obstacles, while light travelled only in straight lines. He did, however, explain the phenomenon of the <u>diffraction</u> of light (which had been observed by <u>Francesco Grimaldi</u>) by allowing that a light particle could create a localised wave in the aether.

Newton's theory could be used to predict the <u>reflection</u> of light, but could only explain <u>refraction</u> by incorrectly assuming that light accelerated upon entering a denser <u>medium</u> because the <u>gravitational</u> pull was greater. Newton published the final version of his theory in his <u>Opticks</u> of 1704. His reputation helped the <u>particle theory of light</u> to hold sway during the eighteenth century. The particle theory of light led <u>Laplace</u> to argue that a body could be so massive that light could not escape from it. In other words, it would become what is now called a <u>black hole</u>. Laplace withdrew his suggestion later, after a wave theory of light became firmly established as the model for light (as has been explained, neither a particle or wave theory is fully correct). A translation of Newton's essay on light appears in *The large scale structure of space-time*, by <u>Stephen Hawking</u> and George F. R. Ellis.



Pierre Gassendi

The fact that light could be <u>polarized</u> was for the first time qualitatively explained by Newton using the particle theory. <u>Étienne-Louis Malus</u> in 1810 created a mathematical particle theory of polarization. <u>Jean-Baptiste Biot</u> in 1812 showed that this theory explained all known phenomena of light polarization. At that time the polarization was considered as the proof of the particle theory.

Wave theory

To explain the origin of <u>colours</u>, <u>Robert Hooke</u> (1635–1703) developed a "pulse theory" and compared the spreading of light to that of waves in water in his 1665 work <u>Micrographia</u> ("Observation IX"). In 1672 Hooke suggested that light's vibrations could be perpendicular to the direction of propagation. <u>Christiaan Huygens</u> (1629–1695) worked out a mathematical wave theory of light in 1678 and published it in his <u>Treatise on Light</u> in 1690. He proposed that light was emitted in all directions as a series of waves in a medium called the <u>luminiferous aether</u>. As waves are not affected by gravity, it was assumed that they slowed down upon entering a denser medium. [36]

The wave theory predicted that light waves could interfere with each other like sound waves (as noted around 1800 by Thomas Young). Young showed by means of a diffraction experiment that light behaved as waves. He also proposed that different colours were caused by different wavelengths of light and explained colour vision in terms of three-coloured receptors in the eye. Another supporter of the wave theory was Leonhard Euler. He argued in *Nova theoria lucis et colorum* (1746) that diffraction could more easily be explained by a wave theory. In 1816 André-Marie Ampère gave Augustin-Jean Fresnel an idea that the polarization of light can be explained by the wave theory if light were a transverse wave. [37]

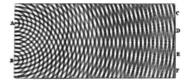
Later, Fresnel independently worked out his own wave theory of light and presented it to the <u>Académie des Sciences</u> in 1817. <u>Siméon Denis Poisson</u> added to Fresnel's mathematical work to produce a convincing argument in favor of the wave theory, helping to overturn Newton's corpuscular theory. By the year 1821, Fresnel was able to show via mathematical methods that polarization could be explained by the wave theory of light if and only if light was entirely transverse, with no longitudinal vibration whatsoever.

The weakness of the wave theory was that light waves, like sound waves, would need a medium for transmission. The existence of the hypothetical substance luminiferous aether proposed by Huygens in 1678 was cast into strong doubt in the late nineteenth century by the Michelson–Morley experiment.

Newton's corpuscular theory implied that light would travel faster in a denser medium, while the wave theory of Huygens and others implied the opposite. At that time, the speed of light could not be measured accurately enough to decide which theory was



Christiaan Huygens

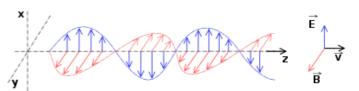


Thomas Young's sketch of a double-slit experiment showing diffraction. Young's experiments supported the theory that light consists of waves.

correct. The first to make a sufficiently accurate measurement was <u>Léon Foucault</u>, in 1850. His result supported the wave theory and the classical particle theory was finally abandoned, only to partly re-emerge in the twentieth century.

Electromagnetic theory

In 1845, Michael Faraday discovered that the plane of polarization of linearly polarized light is rotated when the light rays travel along the magnetic field direction in the presence of a transparent dielectric, an effect now known as Faraday rotation. This was the first evidence that light was related to electromagnetism. In 1846 he speculated that light might be some form of disturbance propagating along magnetic field lines. Faraday proposed in 1847 that light was a high-frequency electromagnetic vibration, which could propagate even in the absence of a medium such as the ether.



A $\underline{\text{linearly polarized}}$ electromagnetic wave traveling along the z-axis, with E denoting the $\underline{\text{electric field}}$ and perpendicular B denoting magnetic field.

Faraday's work inspired <u>James Clerk Maxwell</u> to study electromagnetic radiation and light. Maxwell discovered that self-propagating electromagnetic waves would travel through space at a constant speed, which happened to be equal to the previously measured speed of light. From this, Maxwell concluded that light was a form of electromagnetic radiation: he first stated this result in 1862 in *On Physical Lines of Force*. In 1873, he published <u>A Treatise on Electricity and Magnetism</u>, which contained a full mathematical description of the behavior of electric and magnetic fields, still known as <u>Maxwell's equations</u>. Soon after, <u>Heinrich Hertz</u> confirmed Maxwell's theory experimentally by generating and detecting radio waves in the laboratory and demonstrating that these waves behaved exactly like visible light, exhibiting properties such as reflection, refraction, diffraction and <u>interference</u>. Maxwell's theory and Hertz's experiments led directly to the development of modern radio, radar, television, electromagnetic imaging and wireless communications.

In the quantum theory, photons are seen as <u>wave packets</u> of the waves described in the classical theory of Maxwell. The quantum theory was needed to explain effects even with visual light that Maxwell's classical theory could not (such as spectral lines).

Quantum theory

In 1900 Max Planck, attempting to explain black-body radiation, suggested that although light was a wave, these waves could gain or lose energy only in finite amounts related to their frequency. Planck called these "lumps" of light energy "quanta" (from a Latin word for "how much"). In 1905, Albert Einstein used the idea of light quanta to explain the photoelectric effect and suggested that these light quanta had a "real" existence. In 1923 Arthur Holly Compton showed that the wavelength shift seen when low intensity X-rays scattered from electrons (so called Compton scattering) could be explained by a particle-theory of X-rays, but not a wave theory. In 1926 Gilbert N. Lewis named these light quanta particles photons. [41]

Eventually the modern theory of <u>quantum mechanics</u> came to picture light as (in some sense) *both* a particle and a wave and (in another sense), as a phenomenon which is *neither* a particle nor a wave (which actually are macroscopic phenomena, such as baseballs or ocean waves). Instead, modern physics sees light as something that can be described sometimes with mathematics appropriate to one type of macroscopic metaphor (particles) and sometimes another macroscopic metaphor (water waves), but is actually something that cannot be fully imagined. As in the case for radio waves and the X-rays involved in Compton scattering, physicists have noted that electromagnetic radiation tends to behave more like a classical wave at lower frequencies, but more like a classical particle at higher frequencies, but never completely loses all qualities of one or the other. Visible light, which occupies a middle ground in frequency, can easily be shown in experiments to be describable using either a wave or particle model, or sometimes both.

In February 2018, scientists reported, for the first time, the discovery of a new form of light, which may involve <u>polaritons</u>, that could be useful in the development of quantum computers. [42][43]

Use for light on Earth

<u>Sunlight</u> provides the <u>energy</u> that <u>green plants</u> use to create <u>sugars</u> mostly in the form of <u>starches</u>, which release energy into the living things that digest them. This process of <u>photosynthesis</u> provides virtually all the energy used by living things. Some species of animals generate their own light, a process called <u>bioluminescence</u>. For example, <u>fireflies</u> use light to locate mates and vampire squid use it to hide themselves from prey.

See also



- Automotive lighting
- Ballistic photon
- Colour temperature
- Fermat's principle
- Huygens' principle
- Journal of Luminescence
- Light art
- Light beam in particular about light beams visible from the side
- Light Fantastic (TV series)
- Light mill
- Light painting
- Light pollution
- Light therapy
- Lighting
- List of light sources
- Luminescence: The Journal of Biological and Chemical Luminescence
- Photic sneeze reflex
- Right to light
- Risks and benefits of sun exposure
- Spectroscopy

Notes

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