

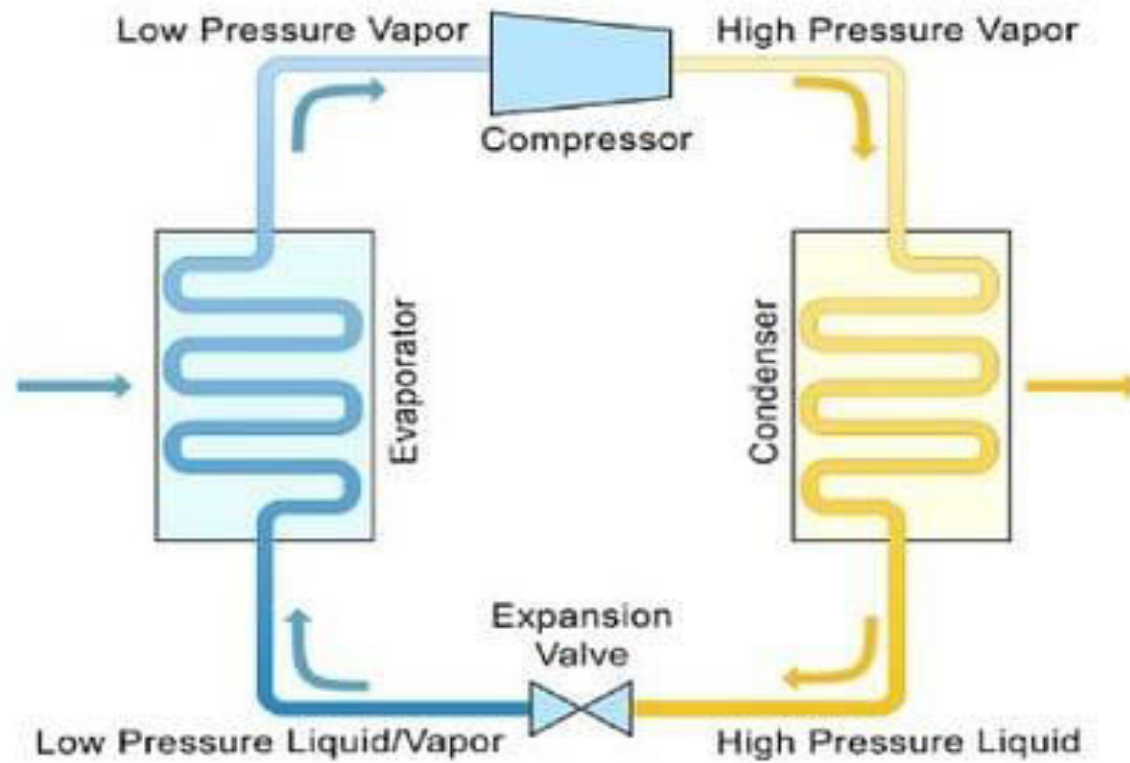
ME 161

Introduction to Mechanical Engineering

Refrigeration and air-conditioning

Please go through class notes and reference materials discussed in the class. This is just a guideline
for those who missed the classes

Refrigeration system components



Types of Air Conditioners

- Central air conditioning system
- Split type air conditioner
- Portable air conditioner
- Window air conditioner
- Cassette air conditioner
- Evaporative air conditioner

- **Air-conditioning** means control of temperature, humidity, purity and movement of air. The working conditions and comfort for human beings is improved by air-conditioning. The temperature is brought down during summer and increased during winter without causing any discomfort. There are 6 major parts to the system. The Compressor, The Condenser, The Receiver/Dryer or Accumulator, The Thermal Expansion Valve or Orifice Tube, The Evaporator, The Refrigerant
- **Refrigeration** may be defined as the process of removing heat from a substance under controlled conditions and reducing and maintaining the temperature of a body below the temperature of its surroundings by the aid of external work.
- **Coefficient of Performance of a Refrigerator : COP**
The performance of a refrigeration system is expressed by a factor known as Coefficient of Performance (COP). The COP of a refrigeration system is defined as the ratio of amount of heat absorbed or extracted in a system to the amount of work supplied.
$$\text{COP} = \frac{\text{Amount of heat absorbed}}{\text{Amount of work supplied}} \quad \text{COP}_r = Q_c / W$$
- **Refrigerants.** The refrigerant is a heat carrying medium which during their cycle in the refrigeration system absorb heat from a low temperature system and reject the heat to a high temperature system.

- **Unit of Refrigeration: Tons of refrigeration**

In a refrigeration system, the rate at which the heat is absorbed in a cycle from the interior space to be cooled is called refrigerating effect. The capacity of a refrigeration system is expressed in Tons of refrigeration which is the unit of refrigeration. A ton of refrigeration is defined as the quantity of heat absorbed to convert one ton of water at 0°C to one ton of ice at the same temperature in 24 hours.

- In S.I. System, 1 Ton of Refrigeration = 210 kJ/min = 3.5 kW,
- In FPS system 200 Btu/min
- In MKS system 50 kcal/min

Types of compressors

- Reciprocating compressors.
- Rotary vane compressors.
- Scroll compressors.
- Screw compressors.
- Centrifugal compressors.

Types of Condenser

- Air cooled
- Water cooled
- Evaporative

Too much humidity cause

- A stuffy nose, itchy and watery eyes, sneezing, difficulty breathing, skin rashes,
- A diminished immune system and resistance to respiratory allergies and infection

Benefits of dehumidification

- A better indoor 'smell'
- Protected clothes and fabrics
- Protected Property.
- Everything is fresher
- Less cleaning time
- Lower energy costs

Evaporator Purpose

- There are two purposes of evaporators:
 - Cooling
 - Dehumidification

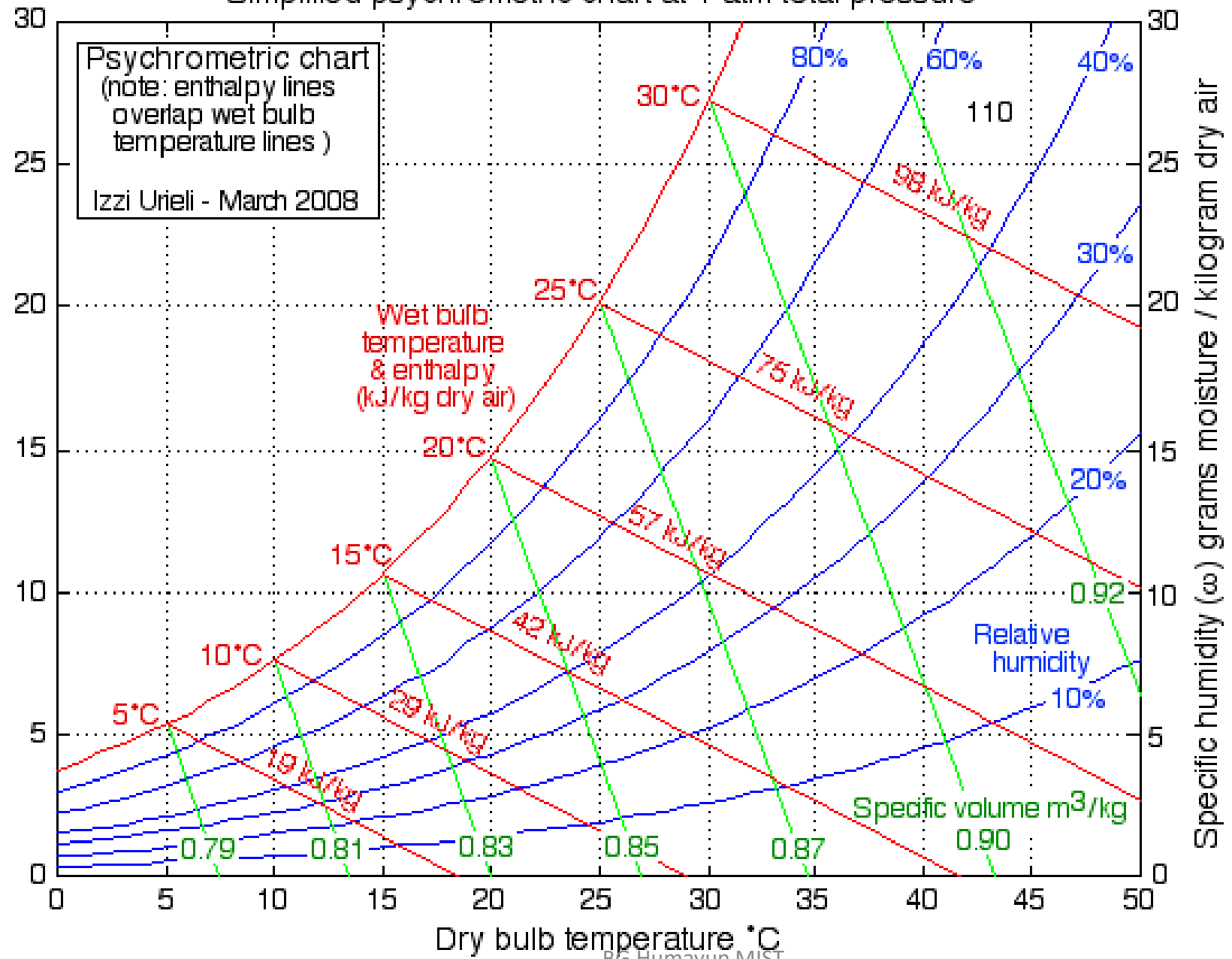
Evaporator Capacity

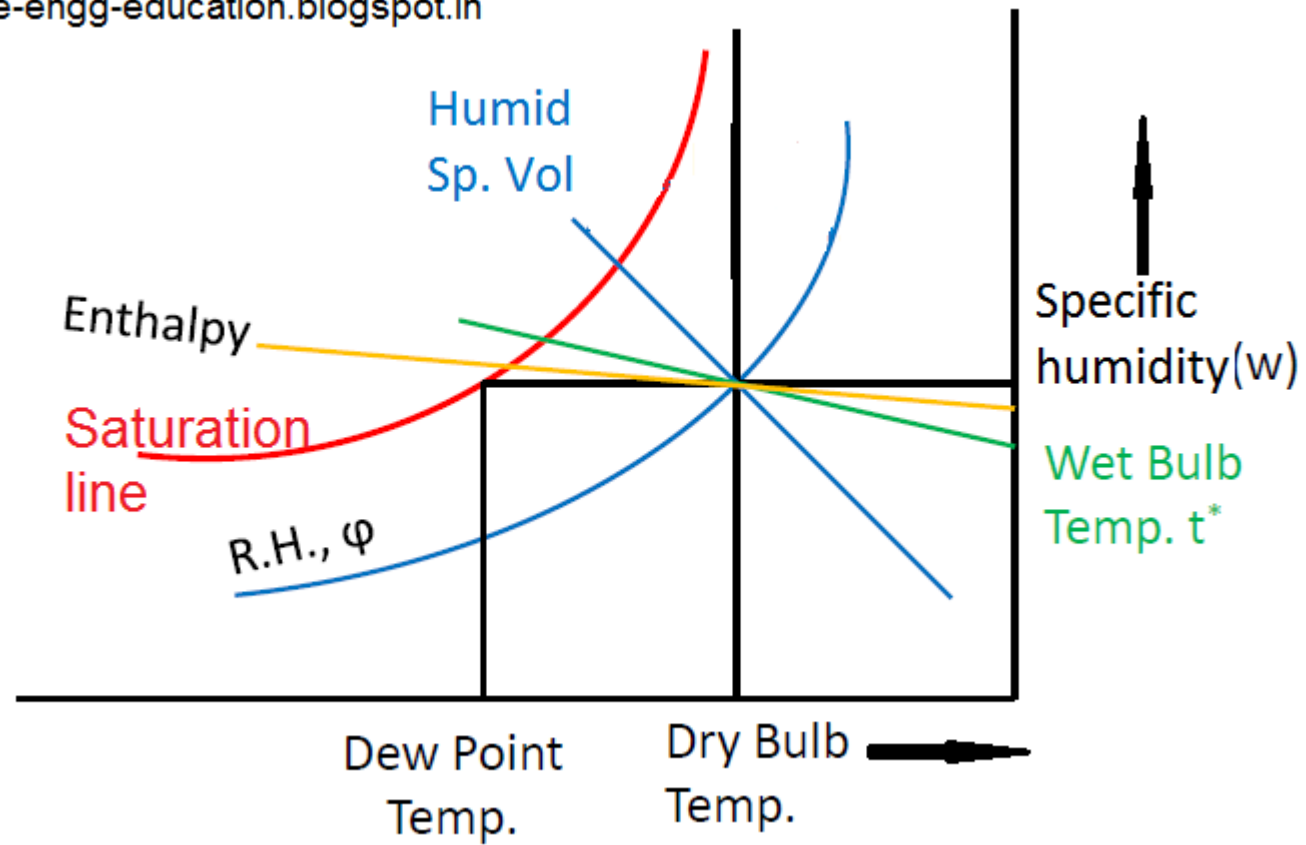
- Factors that effect evaporator capacity:
 - Surface Area
 - Temperature Difference
 - Refrigerant Velocity
 - Conductibility (How fast heat moves through metal)
 - Metal thickness
 - Air Volume

Laws of thermodynamics

- The [zeroth law of thermodynamics](#) defines [thermal equilibrium](#) and forms a basis for the definition of temperature: If two systems are each in thermal equilibrium with a third system, then they are in thermal equilibrium with each other.
- The [first law of thermodynamics](#) states that, when energy passes into or out of a system (as [work](#), [heat](#), or [matter](#)), the system's [internal energy](#) changes in accordance with the law of [conservation of energy](#).
- The [second law of thermodynamics](#) states that in a natural [thermodynamic process](#), the sum of the [entropies](#) of the interacting [thermodynamic systems](#) never decreases.
- The [third law of thermodynamics](#) states that a system's entropy approaches a constant value as the temperature approaches [absolute zero](#). The entropy of a system at absolute zero is typically close to zero.

Simplified psychrometric chart at 1 atm total pressure







Psychrometric Chart



Ex. 2 : In winter air-conditioning system, 100 m^3 of air per minute at 15°C DBT and 80% relative humidity is heated until its DBT is 22°C with constant specific humidity. Find heat added to the air per minute

Given -

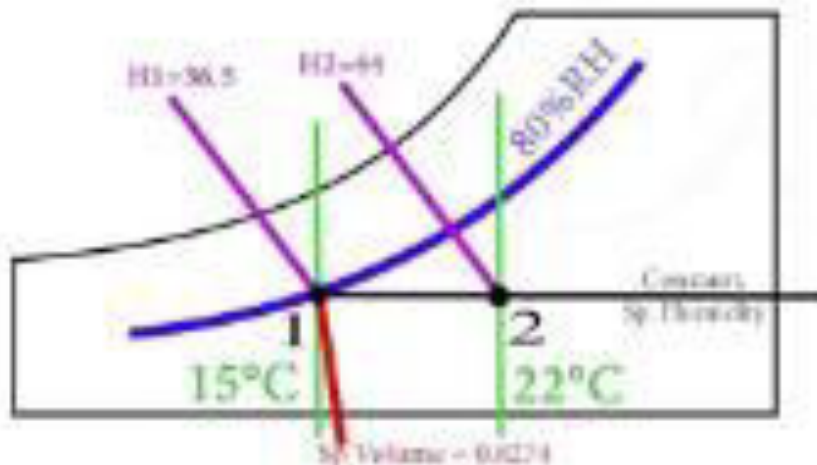
1. Initial Condition: 15°C DBT & 80% RH
2. Final Condition: 22°C DBT at Const. Sp Humidity

Answer - From Psychrometric chart,

1) $H_1 = 36.5 \text{ kJ/kg}$

2) $H_2 = 44 \text{ kJ/kg}$

3) Sp. Volume = $0.8274 \text{ m}^3/\text{kg}$



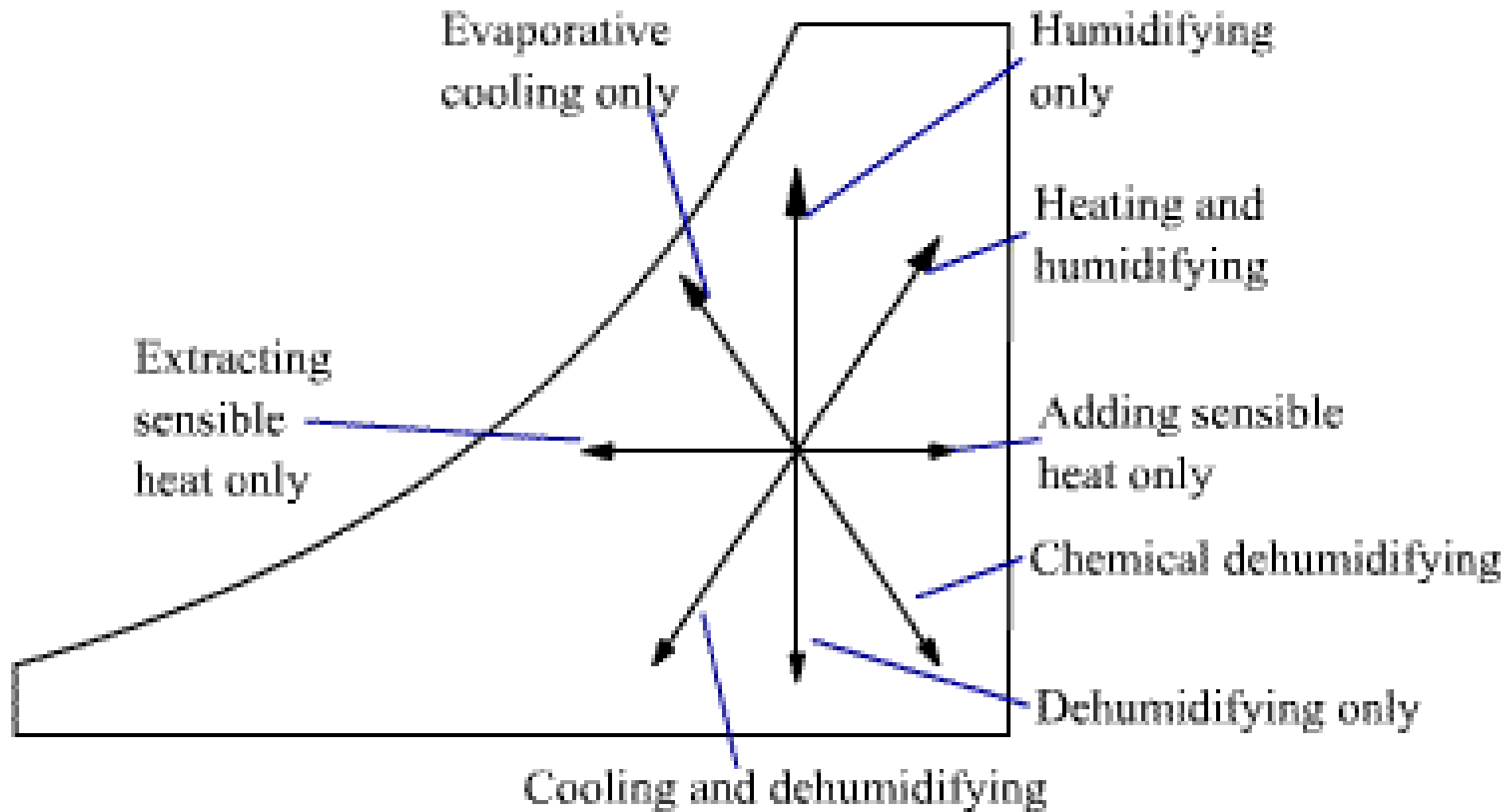
$$m_a = \frac{\text{Volume}}{\text{Sp. Volume}} = \frac{100}{0.8274} = 120.860 \text{ kg/min}$$

Heat added to the air per minute,

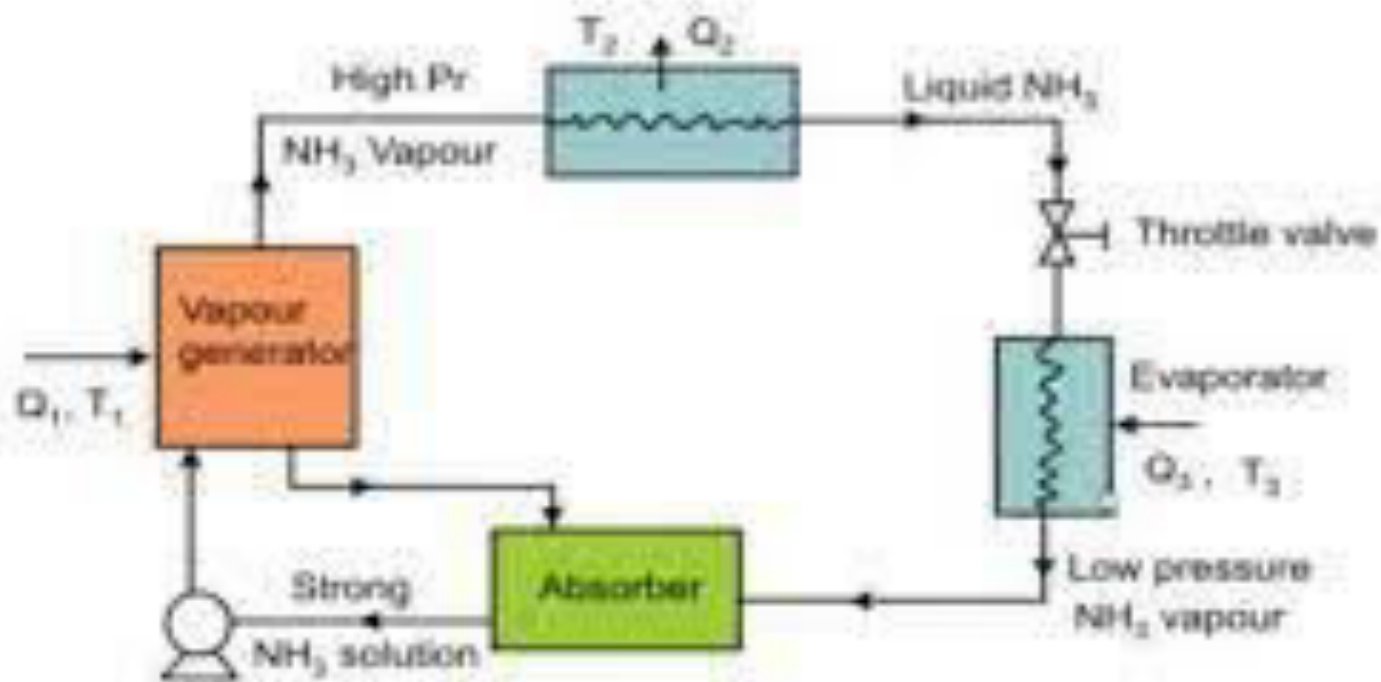
$$= m_a (H_2 - H_1) \text{ kJ/kg}$$
$$= 120.860 (44 - 36.5)$$
$$= 906.4539 \text{ kJ}$$

1:50

YouTube/Winfotech



Simple vapour absorption system



Simple Vapour Absorption System

- The VARS refrigeration system uses two working fluids for refrigeration i.e. refrigerant and absorbent. In the $\text{NH}_3\text{-H}_2\text{O}$ refrigeration system, ammonia (NH_3) is used as a refrigerant while the water (H_2O) is used as an absorbent.
- The purpose of the **absorber** is to absorb the low-pressure refrigerant vapours in the solution of the refrigerant and absorbent. Here the refrigerant vapours are absorbed to form a stronger solution.
- The weaker solution from the generator and the low-pressure refrigerant vapours from the evaporator enters the absorber.
- **Generator** is used to heat the strong solution by use of heating coils, solar energy or waste heat. As the refrigerant has a lower boiling point than the absorbent, the refrigerant inside the solution gets vapourised leaving the solution weaker.
- **Expansion valve** is located between the condenser & evaporator. Here the high-pressure liquid refrigerant is converted into a mixture of low-pressure refrigerant (liquid + vapour).

Advantages:

- It can utilize solar energy or any waste heat from the furnace or turbine for the heating the generator.
- The refrigerant won't produce the greenhouse effect.
- The moving component is a pump which consumes less power than the compressor.
- It has a quieter operation.
- It can handle huge refrigeration loads.

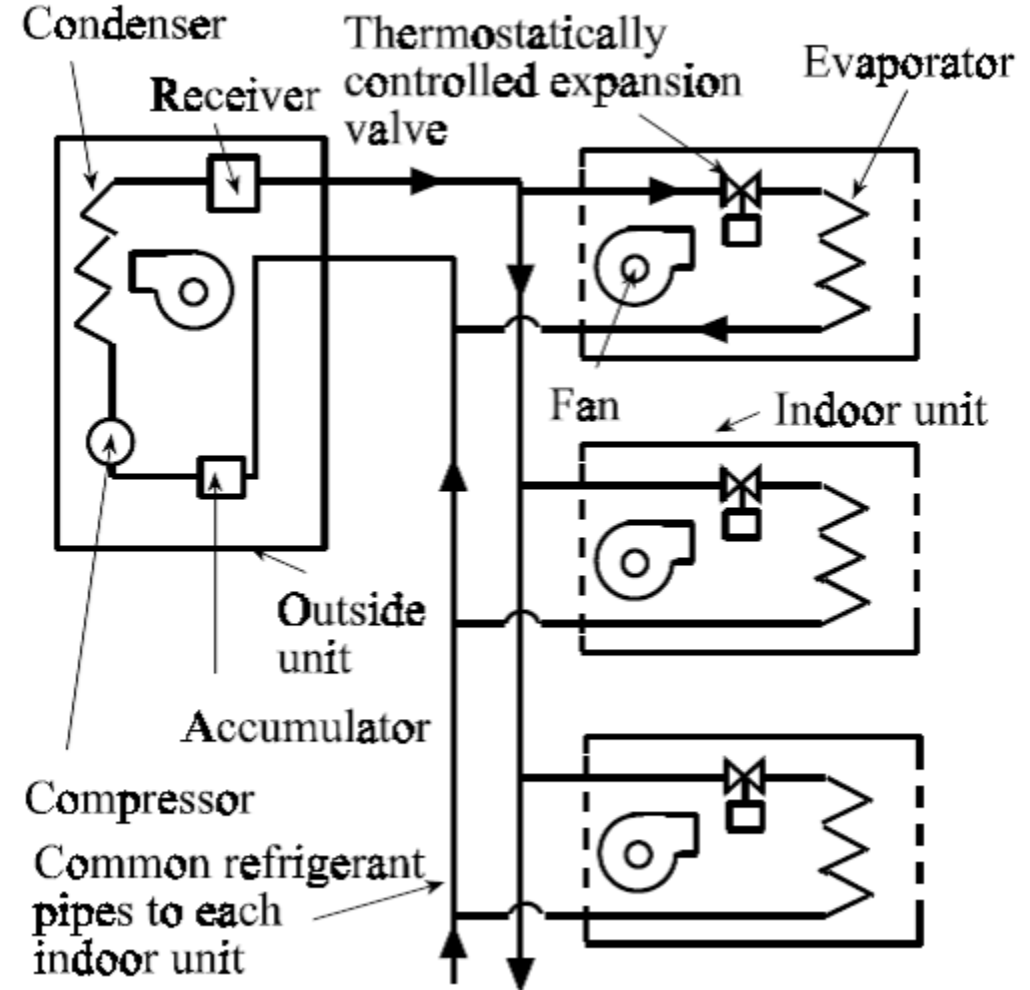
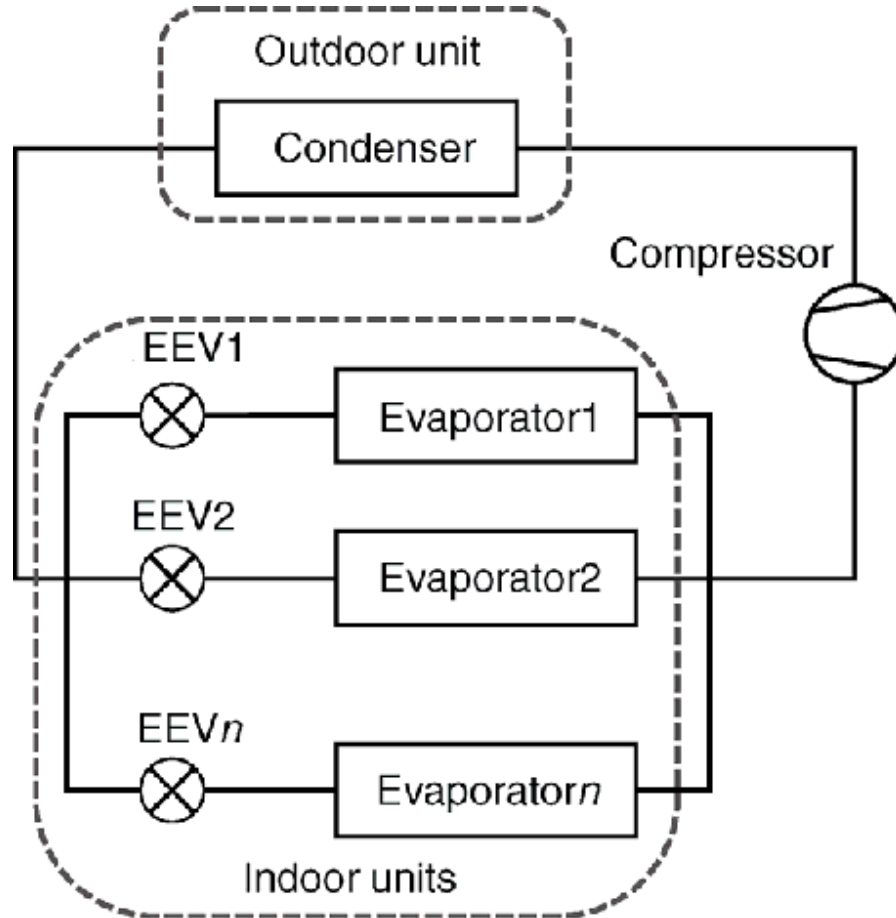
Disadvantages:

- It uses more components than the VCR system.
- It has a bulky setup.
- Lower coefficient of performance (COP).
- The Li-Br absorbent is corrosive in nature.
- Higher initial cost.

Applications:

- Large-capacity refrigeration in bigger plants.
- Suitable for larger air conditioning .
- Used in places having electricity problems for refrigeration and air conditioning.

VRF Air-conditioning System

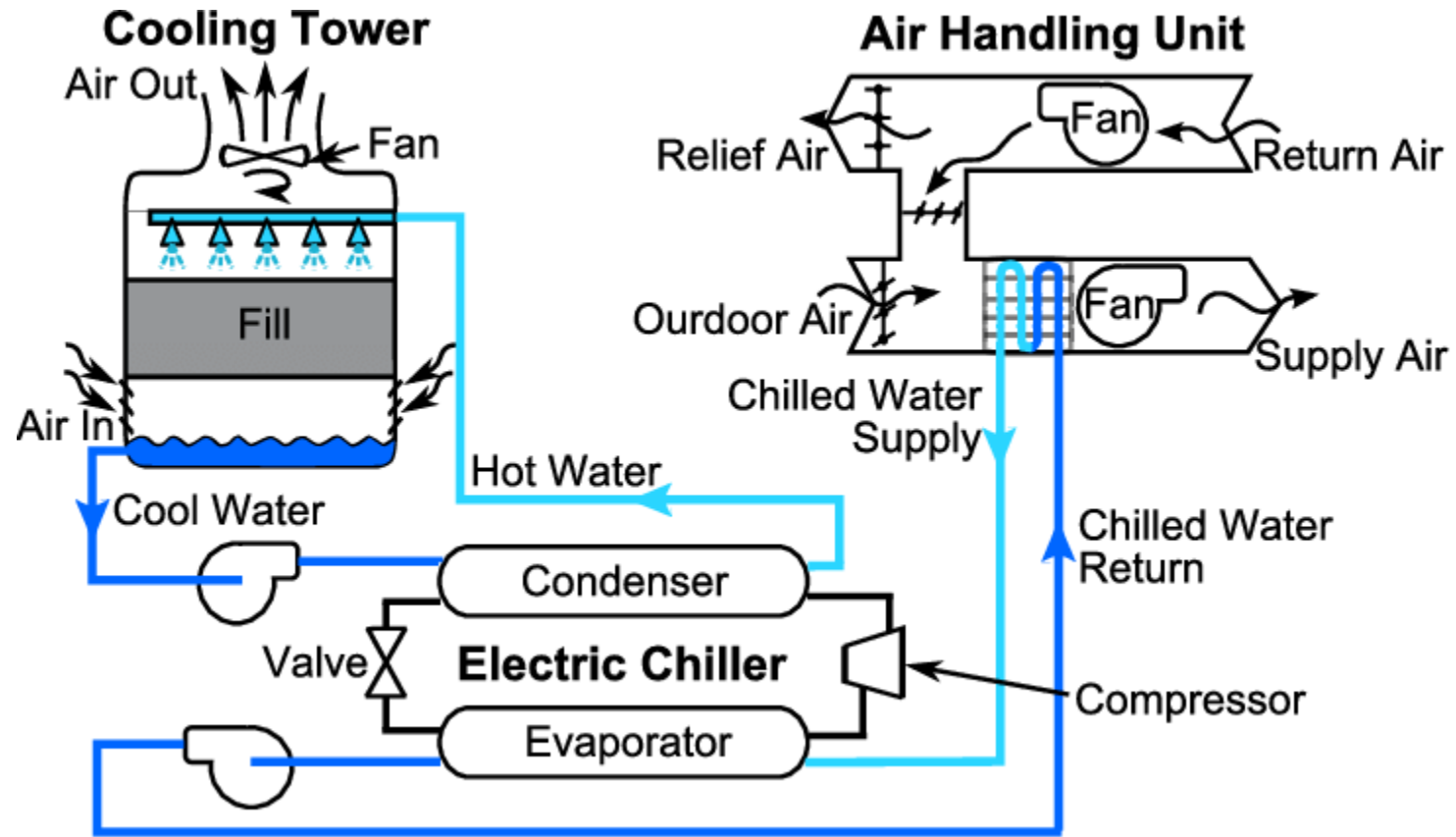


Key benefits of VRF air conditioning systems:

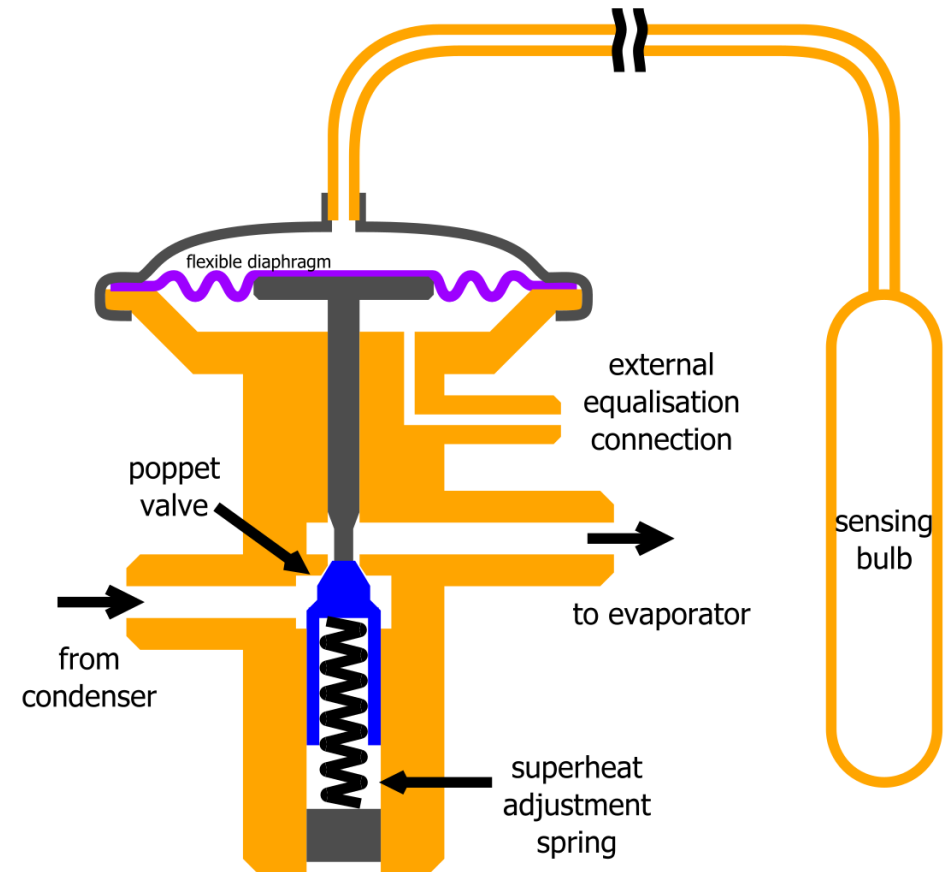
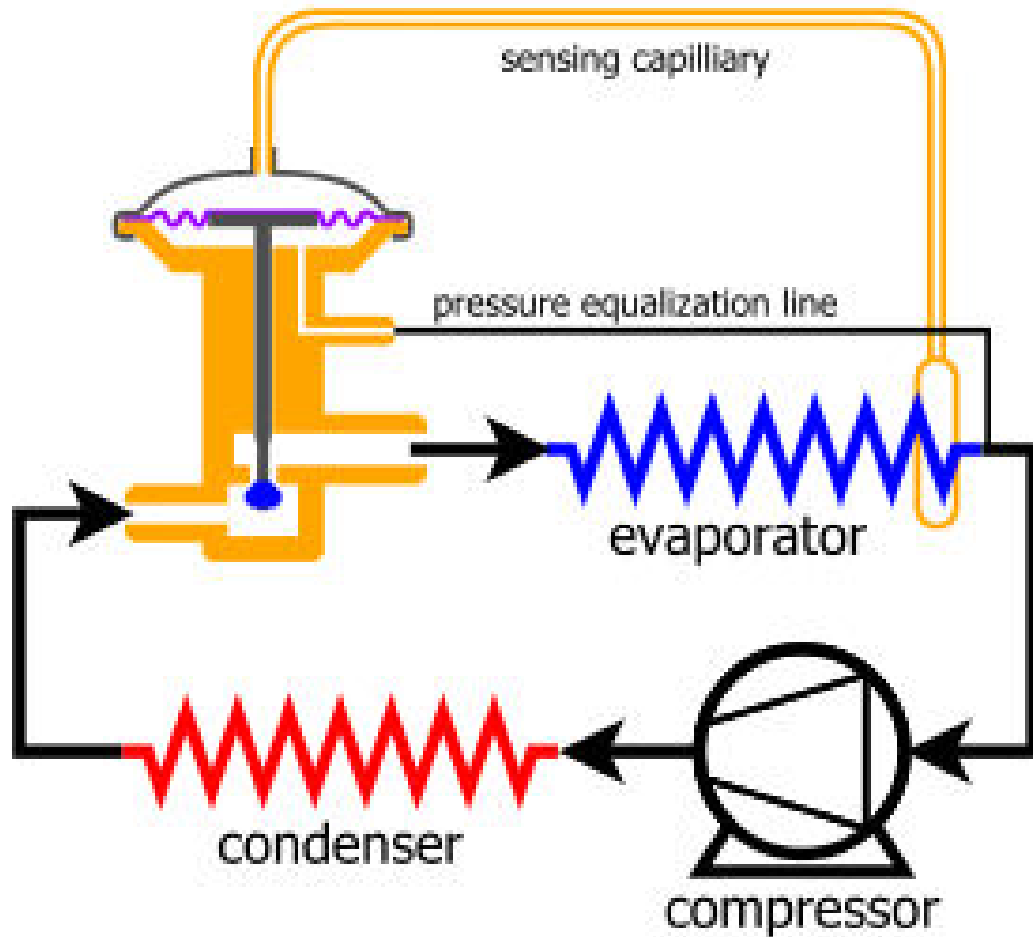
- Highly energy efficient
- Precise temperature control
- Simultaneous heating & cooling
- Heat recovery
- Zoned comfort
- Flexible & modular designs
- Simpler to install
- Sophisticated, easy to use and controls

VRF models are used in many commercial and residential applications as follows:

Hospitals , Hotels, Airports, Offices in high-rise office buildings, Commercial buildings such as shopping malls.



How does expansion valve work in refrigeration system



Properties of a good refrigerant

Desirable Properties of a Good Refrigerant

1) Thermodynamic Properties :-

- a) Boiling Point :- It should have Low Boiling Point.
- b) Freezing Point :- It should be below the Evaporator temperature.
- c) Evaporative Pressure :- It should be above Atmospheric pressure.
- d) Condensing Pressure :- It should have LOW Condensing pressure.
- e) Latent Heat of Vapourisation :- It should have HIGH Latent heat of Vaporisation.
- f) Critical Temperature & Pressure :- It should be above the condensing Temperature & Pressure.

2) Chemical Properties :-

- a) Toxicity :- It should not be Poisonous or injurious . It should not be non-irritating to eyes.
- b) Corrosiveness :- It should not be corrosive & should not have any effect on materials used in equipment.
- c) Leak Detection :- It should have less tendency to leak & if it is leaking it should be easily detectable.
- d) Flammability :- It should not be Inflammable.
- e) Miscibility with Oil :- It should be immiscible with oil & should not have any effect on the properties of Oil used for Lubrication.
- f) Effect on Foodstuff :- It should not affect on food articles Or make them poisonous or unportable.

3) Physical Properties :-

- a) Specific Volume :- It should be LOW in Vapour state.
- b) Viscosity :- It should have LOW viscosity.
- c) Thermal Conductivity :- It should have HIGH Thermal Conductivity.
- d) Di-Electric Strength :- It should have High strength.

Terminology

Refrigerator- machine used to extract heat from body at low temperature to high temperature.

Refrigerant- It is liquid capable of absorbing heat from the other substance.

Capacity of Refrigerator- It is the rate at which heat can be removed from the cold body. Unit- ton of refrigeration

One ton of refrigeration is defined as the quantity of heat required to freeze one ton of water at 0°C in 24 hours. Value is 3.5 kJ/sec.

Coefficient of Performance- Ratio of heat absorbed to the workdone.

COP- $\text{Heat absorbed} / \text{Work done}$

Refrigerant effect- Ratio between heat removed and the time taken. $\text{Heat removed} / \text{time taken}$.