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Charge source and the charging mechanism of the contact electrification of polymer powder



Masato Sakaguchi*, Masakazu Makino

Graduate School of Integrated Pharmaceutical and Nutritional Sciences, University of Shizuoka, 52 Yada, Suruga-ku, Shizuoka 422-8526, Japan

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ABSTRACT

The charge sources, as well as the charging mechanism of the contact electrification (CE) of polymers, are still debatable. Since CE is accompanied by destruction, it is considered that "hard contacting" via ball milling can induce covalent bond scission and produce naked-activated-charge sources. Regarding "soft contacting" via nano-scale sliding, which does not induce covalent bond scission, a frontier-electron, "f-electron," of the naked-activated-charge source is crucial to electron transfer among the naked-activated-charge sources. Here, we configure naked-activated-charge-source models, naked-activated-mechano-anion, and naked-activated-mechano-cation, which are produced by mechanical energy induced heterogeneous covalent bond scission, as well as naked-activated-mechano-radicals that are produced by homogeneous covalent bond scission. Regarding "soft contacting" among naked-activated-charge sources in a vacuum, f-electron can be transferred from a donor to an acceptor if the energy level of the donor is higher than that of the acceptor. The net amount of the normalized transferred-f-electrons is obtained by adopting settings in which the average energy level of the nakedactivated-charge sources (as the donors) is higher than that of the sources employed as acceptors. Thus, the surfaces comprising the donors and acceptors will exhibit positive and negative net surface charges, respectively. We conclude that net surface charges depend on the average energy level of naked-activated-charge sources. Further, we observe that the alignment of polyethylene (PE)-polyvinyl chloride (PVC)-polytetrafluoroethylene (PTFE) to the average energy level is identical to that of the triboelectric series.

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1. Introduction

Charging by contacting (contact electrification, CE) or rubbing (triboelectricity) is ubiquitous among polymeric materials; the phenomenon has existed since ancient times.

In conventional experiments, the amount of charge is measured by a Faraday cup (FC). However, the assignment of a charge source cannot be executed because the FC method cannot define a chemical species and its structure. Furthermore, the amount of charge is affected by many factors, including humidity, surface roughness, hot-spot temperature, oxygen, and polymer characteristics. Therefore, a charge source is still poorly defined, and the charging mechanism is still debatable.

Some studies recently reported that CE was accompanied by the fracture of polymer composites. Grzybowski and coworkers observed via Kelvin force microscopy (KFM) that the surface exhib-

* Corresponding author.

E-mail address: sakaguchi@u-shizuoka-ken.ac.jp (M. Sakaguchi).

ited a uniform charge potential map before contacting a polymer; however, the surface exhibited non-uniform charge potential maps, i.e., a charge-mosaic pattern, after contacting the polymer [1]. Although KFM revealed the charge potential in the nanoscale region, it could not identify the charge source. However, their result indicated that a charge source, which was trapped on the surface, might have been produced by the contact. Employing atomic force microscopy (AFM), they further reported that the contact between polymers produced a surface charge coarsened the surface [2]. Although AFM revealed the surface roughness, it could not identify the charge source, indicating that CE induced surface coarsening (the destruction of the polymer at the surface where the charge source might have been produced and trapped). Galembeck and coworkers reported that contacting polyethylene (PE) with polytetrafluoroethylene (PTFE) revealed the non-uniform and excess charges of PE and PTFE which were positive and negative, respectively, following the triboelectric series [3]. Furthermore, they reported a review in which triboelectricity was triggered in polymers via mechanochemical wear or mass transfer

Nomenclature

contact electrification hard contacting via ball milling can induce covalent bond scission and produce naked-activated-charge sources soft contacting via nano-scale sliding does not induce covalent bond scission, and can induce electron transfer between naked-activated-charge sources A frontier electron "f-electron" of naked-activated-charge source is crucial to electron transfer between the nakedactivated-charge sources PΕ polyethylene polyvinyl chloride PVC PTFE polytetrafluoroethylene POL = {PE, PVC, PTFE} POL-POL naked-activated-mechano-anion POL+ POL naked-activated-mechano-cation POI. POL naked-activated-mechano-radical POL_{ch.source.I} {POL⁻, POL⁺, POL[•]} HOMO and LUMO the highest occupied molecular orbital and lowest unoccupied MO of POL_{ch.source,i} POL_{ch.source,i}{POL⁻(HOMO), POL⁺(HOMO), POL⁻(LUMO), POL⁺(LUMO), POL*(HOMO), POL*(LUMO)} POL_{ch.source,i} at the HOMO $E_{POLch.source,i}$ $\{E_{POL^-(HOMO)}, E_{POL^+(HOMO)}, E_{POL^-(LUMO)}, E_{POL^+(LUMO)}, E_{POL^-(LUMO)}, E_{POL^-(LUMO)},$ •(HOMO), E_{POL}•(LUMO)} the energy level of POL_{ch.source,i} $I_{POL}^{+}(HOMO)$, $I_{POLch.source,i}\{I_{POL^-(HOMO)},$ $I_{POL^-(LUMO)}$, $I_{POL}^{\dagger}(LUMO)$, IPOL (HOMO), IPOL (LUMO), IPOL ized I_{POLch.source,i} $C^{norm}I_{POLch,source,i}\{C^{norm}I_{POL^{-}(HOMO)},C^{norm}I_{POL^{+}(HOMO)},C^{norm}I_{POL^{-}(LUMO)},C^{norm}I_{POL^{+}(LUMO)},C^{norm}I_{POL^{+}(LUMO)},C^{norm}I_{POL^{-}(LUMO)}\}$ the normalized concentration of $C^{norm}I_{POL,source,i}$ with a charge sign Total,normSC_{hard,POLch.source,i} the sum of C^{norm}I_{POLch.source,i}S POL_{don,i}{POL⁻(HOMO), POL[•](HOMO), POL[•](LUMO), POL⁺(HOMO)} f-

 $E_{POLdon,i}\{E_{POL^-(HOMO)}, E_{POL^+(HOMO)}, E_{POL^{\bullet}(HOMO)}, E_{POL^{\bullet}(LUMO)}\}$ energy level of $POL_{don,}$ $E_{POL^{-}(LUMO)}$, $E_{POL^{\bullet}(LUMO)}$, $E_{POL^{\bullet}(LUMO)}$, $E_{POL^{\bullet}(HOMO)}$ energy level of POL_{acc,i} $\Delta E_{POLdon,i/POLacc,i}$ energy difference between $E_{POLdon,i}$ and $E_{POLacc,i}$ FePOL_{don.i}{POL⁻(HOMO), POL•(HOMO), POL•(LUMO)} feasible felectron donor of POLdoni FePOLaccii{POL*(LUMO), POL*(HOMO), POL*(LUMO)} feasible felectron acceptor of POLacc,i $E_{FePOLdon,i}\{E_{FePOL^-(HOMO)}, E_{POL^{ullet}(HOMO)}, E_{POL^{ullet}(LUMO)}\}\$ energy level of $^{Fe}POL_{don,i}$ $E_{FePOLacc,i}\{E_{POL}\bullet_{(LUMO)}, E_{POL}\bullet_{(HOMO)}, E_{POL}\bullet_{(LUMO)}\}\$ energy level of $^{Fe}POL_{acc,i}$ $^{norm}TRfE_{FePOLdon,i/FePOLacc,i}$ Concentration transferred-f-electron from $^{\text{Fe}}\text{POL}_{\text{don},i}$ to $^{\text{Fe}}\text{POL}_{\text{acc},i}$ C^{norm}TRfE_{FePOLdon,i}/FePOLacc,i</sub> normTRfE_{FePOLdon,i}/FePOLacc,I</sub> with charge sign Total, FePOL_{don,i} sum of FePOL_{don,i} Total, FePOL_{don,i} sum of FePOL_{acc,i} sum of FePOL_{acc,i} sum of FePOL_{don,i} total number of FePOL_{don,i} Total N_{FePOLacc,i} total number of FePOL_{acc,i} Total N_{FePOLacc,i} total number of feasible f-electron transfer pathways from FePOLdon,i to FePOLacc,i Total E_{FePOLdon,i} total energy of E_{FePOLdon,i} Total EfePOLacc,i total energy of EfePOLacc,i
Total.normTRfEFePOLdon,i/FePOLacc,i
Sum of normTRfEFePOLdon,i/FePOLacc,i
Total.normTPfEFePOLdon,i/FePOLacc,i
Total.normTPfEFePOLdon,i/FePOLacc,i $C^{Total,norm}TRfE_{FePEdon,i/FePTEacc,i}$ the Total,normTRfE $_{FePOLdon,i/FePOLacc,i}$ with charge sign whole, FePOL_{don,i} sum of Total, FePOL_{don,i} and Total, FePOL_{acc,i} $TotalE_{whole,FePOLdon,i}$ sum of $TotalE_{FePOLdon,i}$ and $TotalE_{FePOLacc,i}$ $whole,FePOL_{acc,i}$ sum of $TotalE_{FePOLdon,i}$ and $TotalE_{FePOLacc,i}$ $TotalE_{FePOLacc,i}$ $TotalE_{FePOLacc,i}$ Total E_{whole,FePOLacc,i} sum of Total E_{FePOLacc,i} and Total E_{FePOLacc,i} Total N_{whole, FepEdon,i} sum of Total N_{FepOLdon,i} and Total N_{FepOLdon,i} $E_{Av.whole,FePOLacc,i}$ average energy of $^{\text{whole,FePOLacc},i}$ $\Delta C^{\text{Net.norm}}TRfE$ net amount of normalized transferred-f-electron

phenomena [4]. Moreover, Soh and coworkers reported that material transfer is key to CE [5]. These reports indicate that CE is accompanied by the destruction of polymers. Additionally, Coote et al. reported a perspective in which they outlined recent advances in the elucidation of the mechanism of CE involving insulators [6]. Whereas there are many studies on the molecular orbital calculations of "normal polymer," none has identified a charge source as the primary product of CE, thus hindering the nondisclosure of molecular orbital calculations on the primary product, and failing to clarify the charging mechanism. These limitations are due to the observation techniques, as well as the experimental conditions, which affect the charge sources. Moreover, their charging is affected by water and oxygen molecules, surface roughness, temperature, etc. Therefore, a paradigm shift in the experimental condition is necessary to reveal charge sources and their chemical structures.

electron donor of POL_{ch.source,i} POL_{acc,i}{POL⁻(LUMO),POL⁺(LUMO), POL⁺(LUMO)} f-

electron acceptor of POLch.source.i

Here, to reveal a charge source, mechanochemistry was applied in a vacuum via solid-state milling.

First, to reveal the molecular structure of a charge source, as well as its property, an experiment was performed in which a polymer sample was mechanically fractured in a vacuum for 21 h at 77 K via vibration glass-ball milling; therein, the glass ampoule containing the sample and balls was equipped with an electron

spin resonance (ESR) spectroscopy sample tube at the top. Employing this experiment, the polymer sample was mechanically fractured in a vacuum at 77 K, producing a fine powder (notably, the glass ampoule was opened at room temperature after ballmilling, revealing that the glass balls and wall of the glass ampoule were tightly covered with the polymer powder. After removing the cover, it was confirmed that the surfaces of the glass ball and glass wall were not fractured. Put differently, no free fine powder was produced on the surfaces of glass ball and wall, although it was produced on the covered surface). Further, the fractured free powder was dropped into the ESR sample tube after ball-milling by flipping the glass ampoule upside-down in the liquid nitrogenfilled Dewar, after which it was observed in a vacuum via ESR at 77 K. The chemical structure of the powder polymer was assigned by analyzing the ESR spectra employing self-built computer program (M.S.). We have reported that the mechanical fracture of a polymer in a vacuum (77 K, 21 h) via vibration glass-ball milling induced the homogeneous scissions of the existing covalent bonds comprising a polymer main chain, as well as produced mechanoradicals [7–13] (regarding low-molecular-weight compounds, although powdered samples were obtained, covalent bond scission due to mechanical energy did not occur, i.e., a mechano-radical was not produced [7]). The mechano-radical was a chain-end type,

Net.normSC_{soft,Av.POL} normalized net surface charge of POL

as well as a "naked" radical, which did not exhibit an interactive counter radical. Although the naked-mechano-radical was anchored to the fresh powder surface that was produced via mechanical destruction, its terminal, (the active end) protruding from the surface was highly reactive during ball milling [8-13]. This active end initiated radical polymerization in a vacuum at 77 K and produced a block copolymer, which was tethered on the surface [8-13]. These results indicated that the activation energy of the polymerization reaction, which was initiated by the naked-mechano-radical, might almost be zero. Molecular motion, e.g., that of the active end of the PE naked-mechano-radical, which was tethered to the PTFE surface in a vacuum, was extremely high, even at 77 K [8-10]. Further, the active end of the PE nakedmechano-radical at 77 K rotated freely around the carbon-carbon bond at the end of the PE main chain. These results indicated that the naked-mechano-radical, which was anchored to the surface in a vacuum, was in an isolated and activated state, also called a "naked-activated-mechano-radical."

Further, the $3.1 \times 10 \text{ nm}^2$ / (tethered point of the block copolymer) value was estimated [9]. Namely, one naked-activated-mechano-radical occupied the nano-order area on the powder surface.

Furthermore, we previously reported that the mechanical fracture of the polymer in vacuum (77 K, 21 h) via vibration glass-ball milling induced the homogeneous and heterogeneous scission of covalent bonds comprising a polymer main chain, as well as produced a *mechano-anion* [14–19] and a *mechano-cation* [18–20] as a pair of products. Although this pair was induced by the heterogeneous covalent bond scissions of the polymer main chain, it could not be directly detected by ESR spectroscopy because the constituent species (*mechano-anion* and *a mechano-cation*) did not possess an electron spin.

Regarding the mechano-anion, we first detected it via electron spin trapping employing tetracyanoethylene (TCNE) in vacuum at 77 K [14–19]. A polymer was ball-milled in a vacuum with TCNE for 21 h at 77 K, after which a mechano-anion that had been produced via the heterogeneous covalent bond scission of the polymer main chain, donated an electron to TCNE: TCNE accepted the electron, thus producing a TCNE-anion radical (TCNE-•). Further, TCNE^{-•} was confirmed via ESR spectroscopy. Therefore, the mechano-anion was produced via ball milling in a vacuum at 77 K, after which it donated an electron to TCNE. This mechanoanion was a naked-mechano-anion that did not exhibit an interactive cation as a counter ion because the mechano-anion was anchored to the fresh surface in a vacuum. The naked-mechanoanion on the surface in vacuum at 77 K was highly reactive to TCNE; the activation energy of the reaction might almost be zero. These results indicated that the naked-mechano-anion on the surface in vacuum was in the activated state; put differently, this is called a "naked-activated-mechano-anion."

Next, a mechano-cation was produced as a paired product with the naked-activated-mechano-anion. However, the mechano-cation could also not be detected via ESR spectroscopy because it did not exhibit an electron spin. Thus, we first detected the mechanocation by a mechanochemical reaction with isobutyl vinyl ether (IBVE; characteristic monomer for cationic polymerization) in which a poly(vinylidene fluoride)(PVDF) naked-mechano-cation that was anchored to PVDF in a vacuum, attracted an electron from IBVE via physical contact due to mixing; IBVE was transformed into an IBVE cation radical (IBVE^{+•}), which initiated the cationic polymerization of IBVE in vacuum at 77 K to produce a poly(IBVE) homo-polymer with M_w and M_n of 6.2×10^4 and 2.3×10^4 g/mol, respectively [20]. Thus, the activation energies of the electron transfer from IBVE to the PVDF naked-mechano-cation, as well as polymerization reaction initiated by IBVE^{+•}, might almost be approximately zero. These results indicate that the naked-mechano-cation that was anchored to the surface in vacuum was in the activated state, also called a "naked-activated-mechano-cation".

From the foregoing, we concluded that *naked-activated-mechano-anion* and *naked-activated-mechano-cation*, which are produced via heterogeneous covalent bond scissions due to mechanical energy, as well as *naked-activated-mechano-radicals*, which are produced by homogeneous covalent bond scissions, account for polymer charge sources in vacuum; they are anchored to the fresh powder surface produced by the mechanical destruction and isolation of the polymer on the surface in vacuum.

Second, to reveal the charging mechanism, we calculated the highest occupied molecular orbital (*HOMO*) and lowest unoccupied MO (*LUMO*) of the charge sources via using the density functional theory (*DFT*). Although we intended to calculate their HOMO and LUMO via using DFT, it was challenging because the observed polymer charge sources exhibited high molecular weights. Next, we configured model molecules of charge sources based on the following three conditions: 1. naked-activated-mechano-anion, naked-activated-mechano-radical as the charge sources with low molecular weights; 2. they were anchored to the surface; 3. they were isolated on the powder surface in a vacuum, i.e., the effects of solvent, oxygen, water were eliminated, and their HOMO and LUMO were calculated via DFT.

Here, we configured model molecules of the charge sources and calculated their HOMOs and LUMOs. The signs and amounts of the net charges, which were induced via feasible electron transfer from the charge source (as the donor) to another charging source (as the acceptor), were estimated employing settings in which the energy level of the donor was higher than that of the acceptor (where the activation energy of the electron transfer is zero).

2. Methods

2.1. Calculations of HOMOs and LUMOs

All the calculations reported in this paper were performed via DFT, as implemented in the Gaussian R 09W ver.7.0 program suite [21]. Becke's 3-parameter hybrid functional combined with the Le-Yang-Parr correlation functional (the B3LYP level of DFT) was utilized [22]. The input geometry-data of the models, which were submitted to the Gaussian, were prepared by a GaussView ver.5.0; to adjust the geometry, the "Clean" option in the viewer program was utilized [23]. The geometries of all the molecules were optimized to ground states in vacuum by a 6-311G(d,p) basis set and the Berny algorithm with the geometry optimization employing direct inversion in the iterative subspace (GEDIIS) in redundant internal coordinates[24]. We observed the calculated ground-state geometries of (i) naked-activated-mechano-cation, (ii) naked-activated-mechano-anion, and (iii) naked-activatedmechano-radical, and calculated the HOMO and LUMO energy levels to estimate the ease of electron transfer among these species. Regarding the radicals, since we employed an unrestricted method, a single occupied MO was divided into occupied and unoccupied ones. In this paper, the former and latter will simply be expressed as HOMO and LUMO, respectively. The calculations were performed on an HP Compaq 8100 Elite SF/CT AY032AV-A with a Windows PC.

3. Result and discussion

- 3.1. Model molecules of the charge source and their properties
- 3.1.1. Model molecules of PE charge sources and their properties
 When mechanical energy, which was generated by "hard contacting" via ball milling, is loaded onto the PE model molecule

(*PE-PE*; *CH*₃*CH*₂*CH*₂*CH*₂*CH*₂*CH*₃) in a vacuum at 77 K, a covalent bond comprising the main chain is heterogeneously or homogeneously cleaved, as follows:

$$PE - PE \rightarrow PE^{-}(HOMO) + PE^{+}(HOMO)$$
...Heterogeneous bond scission

$$PE - PE \rightarrow PE^{-}(LUMO) + PE^{+}(LUMO)$$
...Heterogeneous bond scission (2)

$$PE - PE \rightarrow PE^{\bullet}(HOMO) + PE^{\bullet}(HOMO)...Homogenous bond scission$$
(3)

 $PE_{ch.source,i}\{PE^-(HOMO), PE^+(HOMO), PE^-(LUMO), PE^+(LUMO), PE^+(LUMO)\}$ is defined as a PE charge source,i. $E_{PEch.source,i}\{E_{PE^-(HOMO)}, E_{E^+(HOMO)}, E_{PE^-(LUMO)}, E_{PE^+(LUMO)}\}$ is defined as an energy level of $PE_{ch.source,i}$. Model structures of PE, $PE_{ch.source,i}$, and $PE_{PEch.source,i}$ are reported in Table 1.

Further, we reported the estimation of the ionic degree of the covalent bond comprising polymer main chain by calculating the "absolute \(\Delta \text{Mulliken} \) atomic charge," which was defined as the difference between the Mulliken atomic charges of the two adjacent atoms comprising the covalent bond of the polymer main chain. Ionic yield of polymer increased with the increasing absolute \(\Delta \text{Mulliken} \) atomic charge [18].

Next, we demonstrated the distribution of the orbital coefficient of PE_{ch.source,i}, which was also calculated via DFT (Fig. 1). For example, regarding PE $^-$ (HOMO), the distribution of the orbital coefficient at the C₃-atom (Fig. 1a) exhibited a large density lobe (Fig. 1b). Thus, the "f-electron," which is defined as a frontier electron on a large density lobe, for example, at the C₃-atom, can be key to the interaction as an f-electron donor. Moreover, regarding PE $^+$ (LUMO), the distribution of the orbital coefficient at the C₃-atom (Fig. 1g) exhibited a large density lobe (Fig. 1i). Thus, an "f-hole," which is defined as an f-electron hole on the large density lobe, for example, at the C₃-atom, can be key to the interaction as an f-electron acceptor.

Next, the concentration of PE_{ch.source,i} was estimated.

 $I_{PEch.source,i}\{I_{PE^-(HOMO)}, I_{PE^+(HOMO)}, I_{PE^-(LUMO)}, I_{PE^-(LUMO)}, I_{PE(HOMO)}\}$ is defined as a concentration of $PE_{ch.source,i}$. $I_{PE^-(HOMO)}$ is equal to $I_{PE^+(HOMO)}$ because they were produced as a pair (Eq. (1)). Further, $I_{PE^-(LUMO)}$ is equal to $I_{PE^+(LUMO)}$ because they were produced as a pair (Eq. (2)). Assuming that heterogeneous scission probability of Eq. (1) is identical to that of Eq. (2), $I_{PE^-(HOMO)}$ is equal to the $I_{PE^+(LUMO)}$, and the $I_{PE^+(HOMO)}$ is equal to the $I_{PE^+(LUMO)}$. Assuming that the homogeneous scission probability of Eq. (3) is equal to that of Eq. (4), $I_{PE^-(HOMO)}$ is equal to the $I_{PE^-(LUMO)}$.

Further, we assume that $PE^-(HOMO)$ exhibits one f-electron on the large density lobe, which is key to electron transfer and can release the f-electron. Conversely, $PE^+(LUMO)$ exhibits one f-hole on the large density lobe, and it is key to the acceptance the released f-electron. Although $PE^+(HOMO)$ exhibits one f-electron

or one *f-hole* on the large density lobe, $PE^{\bullet}(LUMO)$ also exhibits one *f-hole* or *f-electron*.

 $^{ion}I_{PEch.source,i}\{I_{PE^-(HOMO)}, I_{PE^+(HOMO)},$ Regarding I_{PEch.source,i}, $I_{PE^-(IUMO)}$, $I_{PE^+(IUMO)}$ was defined as an ionic concentration of $PE_{ch.source,i}$, $I_{PEch.source,i}$ $\{I_{PE(HOMO)}^{\bullet}, I_{PE(LUMO)}^{\bullet}\}$ was defined as a radical concentration of PE_{ch.source,i}. Total concentration of PE_{ch.source,i} (Total I_{PE.ch.source,i}) was obtained by the sum of ion I_{PEch.source,i} and $radI_{PEch.source,i}$. Further, $ionY_{PEch.source,i} = ionI_{PEch.source,i}$ / $TotalI_{PEch.source,i}$ and $radY_{PEch.source,i} = radI_{PEch.source,i}$ / $TotalI_{PEch.source,i}$ were defined as ionic yield and radical yield of PE_{ch.source,i}, respectively. We assume that all the PE_{ch.source,i} were produced by four-time scissions, and concentration of one f-electron was equal to that of one f-hole. Thus, ${}^{Total}I_{PE.ch.source,i}$ was given as (4×2) [f-electron]. Consequently, Thus, four $I_{PE.ch.source,i}$ was given as (4×2) [f-electron]. Consequently, a $normI_{PE.ch.source,i}$ from $I_{PE-(HOMO)}$, $normI_{PE-(HOMO)}$, $normI_{PE-(LUMO)}$, nversely, we previously reported that the $^{ion}Y_{PE}$ (0.11) and $^{rad}Y_{PE}$ (0.89) were obtained by ball milling PE in vacuum for 21 h at 77 K [17–19]. Using ionY_{PE} (0.11) and radY_{PE} (0.89), normI_{PEch.source,i} was obtained, as presented in Table 1. Further, C^{norm}I_{PEch.source,i}-{C_{norm}I_{PE}-(HOMO), C^{norm}I_{PE}+(HOMO), C^{norm}I_{PE}-(LUMO), C^{norm}I_{PE}+(LUMO), C^{norm}I $I_{PE(HOMO)}^{\bullet}$, $C^{norm}I_{PE(LUMO)}^{\bullet}$ is defined as the $I_{PEch.source,i}^{norm}$ with a charge sign. The $C^{\text{norm}}I_{\text{PEch.source,i}}$ value is presented in Table 1.

Contrarily, Total,normSC_{hard,PEch,source,i} is defined as the sum of the C^{norm}I_{PEch,source,i}s. Thus, Total,normSC_{hard,PEch,source,i} is 0.00000, i.e., neutral.

3.1.2. Model molecules of PTFE charge sources and their properties

When mechanical energy owing to hard contacting via ball milling is loaded onto a PTFE model molecule (PTFE – PTFE; CF₃CF₂-CF₂ – CF₂CF₂CF₃) in a vacuum at 77 K, a covalent bond comprising the main chain is cleaved heterogeneously or homogeneously. PTFE_{ch.source,i}[PTFE⁻(HOMO), PTFE⁺(HOMO), PTFE⁺(LUMO), PTFE⁺(LUMO), PTFE⁺(LUMO), PTFE⁺(LUMO), PTFE⁺(LUMO), EpTFE⁺(LUMO), EpTFE⁺(LUMO),

Next, we demonstrated the distribution of the orbital coefficient of PTFE_{ch.source,i}, which was also calculated by DFT (Fig. 2). For example, regarding PTFE $^{\bullet}$ (HOMO), the distribution of the orbital coefficient at the C₂-atom (Fig. 2d) revealed a large density lobe (Fig. 2e). Thus, the *f-electron* on the large density lobe at the C₂-atom can be key to the interaction as an *f-electron* donor. Conversely, regarding PTFE $^{+}$ (LUMO), the distribution of the orbital coefficient at the C₂-atom (Fig. 2g) revealed large density lobe (Fig. 2i). Thus, the f-hole on the large density lobe at the C₂-atom can be key to the interaction as an *f-electron* acceptor.

Next, we estimated the concentration of PTFE_{ch.source,i}. All the PTFE_{ch.source,i} were produced by four-time scissions. $I_{PTFE-h.source,i}$ { $I_{PTFE^-(HOMO)}$, $I_{PTFE^-(HOMO)}$, $I_{PTFE^-(LUMO)}$, $I_{PTFE^-(LUMO)}$, $I_{PTFE^-(LUMO)}$, $I_{PTFE^-(HOMO)}$, $I_{PTFE^-(HOMO)}$, $I_{PTFE-h.source,i}$ Similar to the $I_{PTFE-h.source,i}$ I_{PT

Table 1Model structures of PE, PE_{ch.source,i}, E_{PEch.source,i}, and C^{norm}l_{PEch.source,i}, and C^{norm}l_{PEch.source,i},

PE _{source,i}	Model structure	E _{PEch.source,i} [au]	norm _{IPEch.source,i}	C ^{norm} I _{PEch.source,i}
PE-(HOMO)	CH ₃ CH ₂ CH ₂ -	0.11694	0.01375	-0.01375
PE ⁻ (LUMO)	CH ₃ CH ₂ CH ₂ -	0.22878	0.01375	-0.01375
PE*(HOMO)	CH ₃ CH ₂ CH ₂	-0.20503	0.11125	0.00000
PE*(LUMO)	CH ₃ CH ₂ CH ₂	-0.05110	0.11125	0.00000
PE+(HOMO)	CH ₃ CH ₂ CH ₂ ⁺	-0.57603	0.01375	+0.01375
PE+(LUMO)	CH ₃ CH ₂ CH ₂ ⁺	-0.40480	0.01375	+0.01375

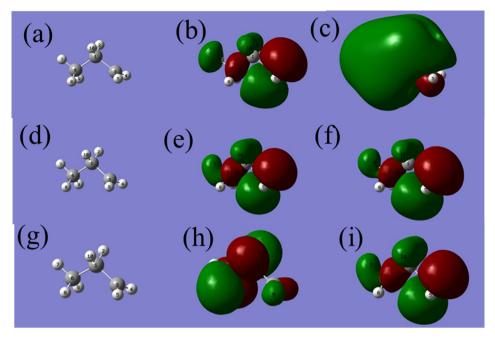


Fig. 1. Model molecule and (a) atom numbering of PE⁻: CH₃CH₂CH⁻₂. Distributions of the orbital coefficients of (b) PE⁻(HOMO) and (c) PE⁻(LUMO). Model molecule and (d) atom numbering of PE⁺: CH₃CH₂CH⁺₂. Distributions of the orbital coefficient of (e) PE⁺(HOMO) and (f) PE⁺(LUMO). Model molecule and (g) atom numbering of PE⁺: CH₃CH₂CH⁺₂. Distributions of the orbital coefficient of (h) PE⁺(HOMO) and (i) PE⁺(LUMO).

PTFE _{ch.source,i}	Model structure	E _{PTFEch.source,i} [au]	norm _I _{PTFEch.source,i}	$C^{norm}I_{PTFEch.source,i}$
PTFE-(HOMO)	$CF_3CF_2CF_2^-$	0.09858	0.02125	-0.02125
PTFE ⁻ (LUMO)	CF ₃ CF ₂ CF ₂	0.26369	0.02125	-0.02125
PTFE*(HOMO)	CF ₃ CF ₂ CF ₂	-0.24455	0.10375	0.00000
PTFE*(LUMO)	CF ₃ CF ₂ CF ₂	-0.09374	0.10375	0.00000
PTFE+(HOMO)	CF ₃ CF ₂ CF ₂ ⁺	-0.61472	0.02125	+0.02125
PTFE+(LUMO)	CF ₃ CF ₂ CF ₂ ⁺	-0.45945	0.02125	+0.02125

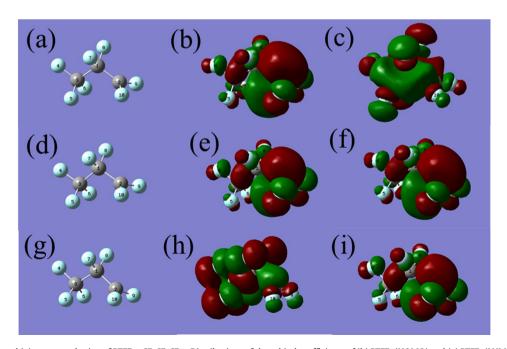


Fig. 2. Model molecule and (a) atom numbering of PTFE⁻: CF₃CF₂CF₂-. Distributions of the orbital coefficients of (b) PTFE⁻(HOMO) and (c) PTFE⁻(LUMO). Model molecule and (d) atom numbering of PTFE⁺: CF₃CF₂CF₂. Distributions of the orbital coefficients of (e) PTFE⁺(HOMO) and (f) PTFE⁺(LUMO). Model molecules and (g) atom numbering of PTFE⁺: CF₃CF₂CF₂. Distributions of the orbital coefficients of (h) PTFE⁺(HOMO) and (i) PTFE⁺(LUMO).

defined as $I_{PTFEch,source,i}\{^{ion}Y_{PTFE} \times 1/(4 \times 2), ^{ion}Y_{PTFE} \times 1/(4 \times 2), ^{ion}Y_{PTFE} \times 1/(4 \times 2), ^{ion}Y_{PTFE} \times 1/(4 \times 2), ^{rad}Y_{PTFE} \times 1/(4 \times 2$

Conversely, Total,normSChard,PTFEch.source,i is defined as the sum of the CnormlpTFEch.source,is. Thus, Total,normSChard,PTFEch.source,i is 0.00000, i.e., neutral.

Next, we estimated $total\ f$ -electrons ($^{Total}I_{f\text{-}e,PTFE}$) of PTFE $_{\text{ch.source,i}}$, which were produced by the mechanical fracture of PTFE in vacuum (77 K, 21 h). We previously reported that the concentration of PTFE mechano-radicals, which were produced by homogeneous covalent bond scissions via the vibratory ball-milling in vacuum for 21 h at 77 K was $6.8 \times 10^{16}\ [\text{spin/g}]$, and the specific surface of the fractured PTFE was $2.1\ [\text{m}^2/\text{g}]\ [9]$. Contrarily, the concentration of the f-electrons arising from the mechano-radicals ($^{rad}I_{f\text{-}e,PTFE}$) is given as $^{rad}I_{f\text{-}e,PTFE} = ^{Total}I_{f\text{-}e,PTFE} \times ^{rad}Y_{PTFE}$. Thus, $^{Total}I_{f\text{-}e,PTFE} = ^{rad}I_{f\text{-}e,PTFE}$ is given as $^{rad}Y_{PTFE}$, Assuming that one f-electron is equal to one electron spin. Therefore, estimated $^{Total}I_{f\text{-}e,PTFE}}$ in vacuum without the decay of the f-electrons is $(6.8 \times 10^{16}\ [\text{spin/g}]/\ 0.83)/2.1\ [\text{m}^2/\text{g}] = 3.901 \times 10^{16}\ [\text{f--electron/m}^2]$. In our discussion, we assumed that one f-electron is identical to one PTFE $_{\text{ch.source,i}}$, i.e., [one f-electron] = [one charge source]. Thus, $^{Total}I_{f\text{-}e,PTFE}(3.9013 \times 10^{16}\ [\text{f-electron/m}^2])$) is obtained as $^{Total}I_{f\text{-}e,PTFE}(3.9013 \times 10^{16}\ [\text{charge source/m}^2])$. Accordingly, each PTFE $_{\text{ch.source,i}}$ occupies the nanoscale region with 25.633 [nm²/charge source] on the surface. We can conclude that the non-uniform charge pattern of the PTFE surface comprise charge sources in the nanoscale region.

3.1.3. Model molecules of polyvinyl chloride (PVC) charge sources and their properties

When mechanical energy due to hard contacting via ball milling is loaded onto PVC model molecule (*PVC – PVC*; *CH*₂*ClCH*₂*-CHCl – CH*₂*CHClCH*₃) in vacuum at 77 K, a covalent bond comprising the main chain is heterogeneously or homogeneously cleaved. *PVC*_{ch.source,i}(*PVC1*⁻(*HOMO*), *PVC2*⁺(*HOMO*), *PVC2*⁻(*HOMO*), *PVC1*⁺(*HOMO*), *PVC1*⁺(*LUMO*), *PVC1*⁺(*LUMO*), *PVC1*⁺(*LUMO*), *PVC1*⁺(*LUMO*), *PVC1*⁺(*LUMO*), *PVC2*⁺(*LUMO*), *PVC2*⁺(*LUMO*), *PVC2*⁺(*LUMO*), *PVC2*⁺(*LUMO*), *EPVC2*⁺(*LUMO*), *EPVC2*⁺(*LUMO*),

Next, the concentration of $PVC_{ch.source,i}$ was estimated. All $PVC_{ch.source,i}$ were produced by eight-time scissions. I_{PVC2} - I_{PV

Further, C^{norm}I_{PVCch.source,i} is listed in Table 3.

Conversely, Total,normSC_{hard,PVCch.source,i} is defined as the sum of the C^{norm}I_{PVCch.source,i}s. Thus, Total,normSC_{hard,PVCch.source,i} is 0.00000 (neutral).

3.2. CE of polymers in vacuum

We had reported that the charging of polymer might be due to the transfer of an electron from a mechano-anion to a mechanoradical based on the "electron-release potential" of the mechanoanion [17,25]. Although our study was the first to report that mechano-anion and mechano-radical correlated with CE of a polymer, the proposed mechanism was inadequate because it eliminated a mechano-cation, which was produced by heterogeneous covalent bond scission, from the reaction system, Sequentially, we proposed a charging mechanism in which charging was due to the transfer of electrons among mechano-anion, mechanoradical, and mechano-cation [26]. However, our proposed electron-transfer pathways were insufficient. Accordingly, whole feasible electron-transfer pathways could not be elucidated and the net amount of transferred electrons could not be speculated. However, there was no evidence of the correlation between charging and the covalent bond scission of polymers during the study.

Recently, although a charge source was not clarified, some studies reported [1–5] that CE was accompanied by the fracture of polymers, (detail in Section 1).

Here, we assumed the sign and net amount of the normalized transferred *f-electrons* from the charge source (donor) to another charging source (acceptor) in the settings in which the average energy level of the donors is higher than that of the acceptors (activation energy of the *f-electron* = 0.0000). (Notably, regarding the transfer of the *f-electron* from the charge source (donor) to the charge source (acceptor), we assumed that the activation energy should be zero owing to the extremely high reactivity even at 77 K, the naked structure, in vacuum isolation, and the C–C single bond at the terminal of the chain). Here, we demonstrated that the sign and net normalized surface charge depend on the average energy level of the charge sources. Further, we demonstrated that the alignment of the polymers based on the average energy level of the charge sources is identical to the alignment of the polymers from the positive charge to the negative on the triboelectric series.

Table 3Model structures of PVC, PVC_{ch.source,i}, E_{PVCch.source,i}, normI_{PVCch.source,i}, and C^{norm}I_{PVCch.source,i},

PVC _{ch.source,i}	Model structure	E _{PVCch.source,i} [au]	norm _{IPVCch.source,i}	$C^{norm}I_{PVCch.source,i}$
PVC1 ⁻ (HOMO)	CH ₂ ClCH ₂ CHCl ⁻	0.06700	0.03125	-0.03125
PVC2 ⁻ (HOMO)	CH ₃ CHClCH ₂	0.09314	0.03125	-0.03125
PVC1 ⁻ (LUMO)	CH ₂ CICH ₂ CHCl ⁻	0.19426	0.03125	-0.03125
PVC2-(LUMO)	CH ₃ CHClCH ₂ -	0.21606	0.03125	-0.03125
PVC1°(HOMO)	CH2CICH2CHCI*	-0.22376	0.03125	0.00000
PVC2*(HOMO)	CH ₃ CHClCH ₂	-0.22663	0.03125	0.00000
PVC1°(LUMO)	CH2CICH2CHCI*	-0.08577	0.03125	0.00000
PVC2*(LUMO)	CH ₃ CHClCH ⁶	-0.07229	0.03125	0.00000
PVC1 ⁺ (HOMO)	CH ₂ CICH ₂ CHCI ⁺	-0.47085	0.03125	+0.03125
PVC2 ⁺ (HOMO)	CH ₃ CHClCH ₂ ⁺	-0.50460	0.03125	+0.03125
PVC1 ⁺ (LUMO)	CH ₂ ClCH ₂ CHCl ⁺	-0.39067	0.03125	+0.03125
PVC2 ⁺ (LUMO)	CH₃CHCICH ₂ ⁺	-0.41251	0.03125	+0.03125

3.2.1. Ce of PE and PTFE in vacuum

CE of PE and PTFE were discussed based on the transfer of the *f*-electron between PE_{ch.source,i} and PTFE_{ch.source,i} in a vacuum.

First, we considered the hard contact between PE and PTFE in a vacuum. PE_{ch.source,i} is produced by covalent bond scission, which is induced by mechanical energy due to the hard contacting and is anchored to the surface of PE. Concurrently, PTFE_{ch.source,i} is produced by covalent bond scission due to the hard contacting and is anchored to the surface of PTFE. Consequently, the PE surface exhibits a charge-mosaic pattern that is comprising a nano-scale regional charge from PE_{ch.source,i}, although this surface exhibits a charge-mosaic pattern, $^{Total,norm}SC_{hard,PEch.source,i}$ is 0.00000 (neutral). Although the PTFE surface also exhibits the charge-mosaic pattern comprising a nano-scale regional charge from PTFE_{ch.source,i}, $^{Total,norm}SC_{hard,PTFEch.source,i}$ is also 0.00000 (neutral).

Second, we considered the "soft contacting" between $PE_{ch.source,i}$ and $PTFE_{ch.source,i}$. Soft contacting refers to nano-scale sliding; it does not cause bond scission and can induce f-electron transfer between $PE_{ch.source,i}$ and $PTFE_{ch.source,i}$. The contacting combinations between $PE_{ch.source,i}$ and $PTFE_{ch.source,i}$ are considered in 32 combinations, which are divided into two categories: one category comprises 16 combinations of $PE_{don,i}$ and $PTFE_{acc,i}$, i.e., 16 pathways that might account for f-electron transfer from $PE_{don,i}$ to $PTFE_{don,i}$ and $PE_{acc,i}$, i.e., 16 pathways that might represent an f-electron transfer from $PTFE_{don,i}$ to $PE_{acc,i}$. Regarding soft contacting, it was assumed that the total f-electrons of $PE_{ch.source,i}$ (f-otal f-e,f-e) are identical to those of f-e,f-c,f-c,f-e,f-e,f-e,f-f, indicating that the concentrations of f-e,f-c,f-c,f-c,f-e,f-e,f-f, indicating that the concentrations of f-e,f-c,f-c,f-e,f-e,f-e,f-e,f-f, indicating that the concentrations of f-e,f-c,f-c,f-e,f-e,f-f, indicating that the concentrations of f-e,f-f, indicating that the concentrations of f-e,f-c,f-c,f-e,f-f, indicating that the concentrations of f-e,f-f, indicating that f-e,f-f, indicating tha

First, we considered f-electron transfer from $PE_{don,i}$ to $PTFE_{acc,i}$ in a vacuum.

For example, regarding the soft contacting of PE-(HOMO) with PTFE*(LUMO), the transfer an *f-electron* from PE-(HOMO) (donor) to PTFE*(LUMO) (acceptor) can be executed without consuming additional energy because $\Delta E_{PE^-(HOMO)/PTFE} \bullet_{(LUMO),2} = E_{PE^-(HOMO)}$ $(0.11694) - E_{PTFE (LUMO)}^{\bullet} (-0.09374) = 0.21068 > 0.00000$ a.u., and the density lobes at the C_3 - and C_2 -atoms of PE⁻(HOMO) (Fig. 1b) and PTFE (LUMO) (Fig. 2f), respectively, are large. This feasible f-electron transfer pathway is presented in Table 4 as PE⁻(HOMO) → PTFE[•](LUMO) in the line of the pathway, Arabic numeral 2. Regarding this feasible f-electron transfer pathway, $^{norm}TRfE_{PF^-(HOMO)/PTFE(IJJMO),2}^{\bullet}$ is defined as the concentration of normalized transferred f-electron from PE-(HOMO) (donor) to PTFE*(LUMO) (acceptor). The values of normTRfE_{PE}*(HOMO)/PTFE (LUMO).2 is 0.01375, and it is assigned to low $^{norm}I_{PE^-(HOMO)}$ (0.01375) compared with that of high norm I_{PTFE (LUMO)} (0.10375). $C^{norm}TRfE_{PE}^{\bullet}_{(HOMO)/PTFE}$ (LUMO),2 is defined $^{norm}TRfE_{PE}^{-\bullet}_{(HOMO)/PTFE}_{(LUMO),2}$ with a charge sign. $C^{norm}TRfE_{PE}^{-}_{(-)}$ $_{\text{HOMO})/\text{PTFE}}^{\bullet}_{\text{(LUMO)},2}$ is -0.01375. These terms are presented in Table 4 in the pathway line, Arabic numeral 2.

All the cases involving the soft contacting of PE_{don,i} with PTFE_{acc,i} are listed in Table 4 (pathway lines, Arabic numerals 1–16). Although the total number of pathways were 16, we emphasized that FePE_{don,i} could donate an *f-electron* to FePTFE_{acc,i} in the settings where ΔΕ_{FePEdon,i/FePTFEacc,i} > 0, i.e., Ε_{FePEdon,i} > E_{FePTFEacc,i}. Accordingly, each normTRfE_{FePEdon,i/FePTFEacc,i} was obtained in the settings where Ε_{FePEdon,i} > E_{FePTFEacc,i}, i.e., those in which normTRfE_{FePEdon,i/FePTFEacc,i} was assigned to low normI_{ch.source,i} compared with high normI_{ch.source,i}. The feasible *f-electron* transfer pathways are presented in Table 4 (Pathways 2, 3, 4, 7, 8, 10, 11 and 12). PE_{don,i,} PTFE_{acc,i}, E_{PEdon,i,} E_{PTFEacc,i}, ΔΕ_{PEdon,i/FePTFEacc,i} are presented in Table 4 (Pathway numerals). Furthermore, Total, normTRfE_{FePEdon,i/FePTFEacc,i} (0.39500) was obtained from the sum of

We emphasized that $C^{Total,norm}TRfE_{FePEdon,i/FePTFEacc,i}$ (-0.39500) indicates the amount of the normalized transferred f-electron from $^{Total,Fe}PE_{don,i}$ to $^{Total,Fe}PTFE_{acc,i}$ via the $^{Total}N_{FePEdon,i/FePTFEacc,i}$ (8) in the settings where $^{Total}E_{FePEdon,i}$ (-0.21254 a.u.) > $^{Total}E_{FePTFEacc,i}$ (-2.29948 a.u.).

Second, we considered f-electron transfer from PTFE $_{don,i}$ to PE $_{acc,i}$ in a vacuum.

For example, regarding the soft contacting of PTFE⁻(HOMO) with PE[•](LUMO), the transfer of an f-electron from PTFE⁻(HOMO) as the donor to PE[•](LUMO) as the acceptor can be achieved without additional extra-energy because $\Delta E_{PTFE^{-}(HOMO)/PE(LUMO),18}$ =- $E_{\text{PTFE}}^{-}(\text{HOMO})$ (0.09858) - $E_{\text{PE}}^{\bullet}(\text{LUMO})$ (-0.05110) = 0.14968 > 0.00000 a.u., and the density lobes at the C_2 - and C_3 -atoms of the PTFE-(HOMO) (Fig. 2b) and PE*(LUMO) (Fig. 1f), respectively, are large. This feasible *f-electron* transfer pathway is reported in Table 5 as PTFE⁻(HOMO) → PE[•](LUMO) (Pathway 18). Regarding this feasible f-electron transfer pathway, normTRfE_{PTFE}-(HOMO)/PE (LUMO).18</sub> is defined as the concentration of normalized transferred f-electron from PTFE-(HOMO) as the donor to PE*(LUMO) as the acceptor. normTRfE_{PTFE}-(HOMO)/PE(LUMO),18</sub> is 0.02125, and normTRfE_{PTFE}-(HOMO)/ PE (LUMO),18 is assigned to low norm I_{PTFE} (HOMO) (0.02125) compared with high $^{\text{norm}}I^{\bullet}_{\text{PE}(\text{LUMO})}$ (0.11125). $C^{\text{norm}}TRfE_{PTFE^{-}(\text{HOMO})/PE(\text{LUMO}),18}$ is defined as $^{norm}TRfE_{PTFE}\ ^{\bullet}_{(HOMO)/PE\ (LUMO),18}$ with a charge sign. $C^{norm}TRfE_{PTFE^-(HOMO)/PE\bullet(LUMO),18}$ is -0.02125. These two terms are listed in Table 5 (Pathway 18).

All the cases regarding the soft contacting of PTFE_{don,i} with PE_{acc,i} are reported in Table 5 (Pathways 17-32). Although the total number of Pathways were 16, we emphasized that FePTFEdon,i could donate an f-electron to ${}^{\mathrm{Fe}}\mathrm{PE}_{\mathrm{acc},i}$ in the settings where $\Delta E_{\text{FePTFEdon,i/FePEacc,i}} > 0$, i.e., $E_{\text{FePTFEdon,i}} > E_{\text{FePEacc,i}}$. Accordingly, each normTRfE_{FePTFEdon,i/FePEacc,i} was obtained in the settings where $E_{FePTFEdon,i} > E_{FePEacc,i}$, i.e., those in which $^{norm}TRfE_{FePTFEdon,i/FePEacc,i}$ was assigned to low normI_{ch.source,i} compared with high norm I_{ch.source,i}. The feasible *f-electron* transfer pathways are presented in Table 5 (Pathways 18, 19, 20, 24, 27, and 28). PTFEdon,i, PEacc,i, Eptfedon,i, Epeacc,i, \(\Delta Eptfedon,i/PEacc,i, \) \(\text{FePTFE}_{don,i}, \) \(\text{FePE}_{acc,i,} \) normTRfE_{FePTFEdon,i}/FePEacc,i, and C^{norm}TRfE_{FePTFEdon,i}/FePEacc,i</sub> are presented in Table 5 (Pathway numerals). Further, Total,normTRfE_{FePTFEdon,i/FePEacc,i} (0.187500) was obtained from the sum of $^{norm}TRfE_{FePTFEdon,i/FePEacc,i}S$. $C^{Total,norm}TRfE_{FePTFEdon,i/FePEacc,i}$ (-0.187500) was the $^{Total,norm}TRfE_{FePTFEdon,i/FePEacc,i}$ with charge sign.

Regarding these feasible *f-electron* transfer pathways, C^{Total,norm}-TRfE^{FePTFEdon,i/FePEacc,i} (-0.187500) is obtained via the *f-electron* transfer pathways of ^{Total}N_{FePTFEdon,i/FePEacc,i} (6) is in the settings where $\Delta^{\text{Total}}E_{\text{FePTFEdon,i/FePEacc,i}}$ (1.53927 a.u.) = { $^{\text{Total}}E_{\text{FePTFEdon,i}}$ (-0.13629 a.u.) - $^{\text{Total}}E_{\text{FePEacc,i}}$ (-1.67556 a.u.)} > 0.

We emphasized that the $C^{Total,norm}TRfE_{FePTFEdon,i/FePEacc,i}$ (-0.187500) indicated the amount of the normalized transferred-f-electron from $^{Total,Fe}PTFE_{don,i}$ to $^{Total,Fe}PE_{acc,i}$ via the $^{Total}N_{FePTFEdon,i/FePEacc,i}$ (6) in the settings where $^{Total}E_{FePTFEdon,i}$ (-0.13629~a.u.) > $^{Total}E_{FePEacc,i}$ (-1.67556~a.u.).

Next, after soft contacting, we speculated a normalized net surface charge of PE ($^{Net.norm}SC_{soft,Av.PE}$) and that of PTFE ($^{Net.norm}SC_{soft,Av.PE}$) based on the $C^{Total,norm}TRfE_{FePEdon,i/FePTFEacc,i}$ (-0.39500) and $C^{Total,norm}TRfE_{FePTFEdon,i/FePEacc,i}$ (-0.187500).

Regarding the soft contacting, the $\Delta C^{\text{Net.norm}}$ TRfE is given as $\Delta C^{\text{Net.norm}}$ TRfE (-0.20750) = $C^{\text{Total,norm}}$ TRfE_{FePEdon,i/FePTFEacc,i} (-0.39

Table 4Soft contacting between PE_{don,i} and PTFE_{acci}, E_{PEdon,i}, Δ E_{PEdon,i}/PTFEacci</sub>, FePE_{don,i}, FePTFE_{acci}, normTRfE_{FePEdon,i}/FePTFEacci</sub>, and C^{norm}TRfE_{FePEdon,i}/FePTFEacci</sub>, and C^{norm}TRfE_{FePEdon,i}/FePTFEacci</sub>.

Path. No.	Soft contacting between PE _{don,i} and PTFE _{acc,i}	E _{PEdon,i} [a.u.]	E _{PTFEacc,i} [a.u.]	ΔΕ _{PEdon,i} /PTFEacc,i [a.u.]	Feasible f-electron transfer pathway from FePE _{don,i} to FePTFE _{acc,i}	normTRfE _{FePEdon,i} /FePTFEacc,i	$C^{norm}TRfE_{FePEdon,i/FePTFEacc,i}$
1	PE-(HOMO) / PTFE-(LUMO)	0.11694	0.26369	-0.14675		0.00000	0.00000
2	PE-(HOMO) / PTFE*(LUMO)	0.11694	-0.09374	0.21068	$PE^{-}(HOMO) \rightarrow PTFE^{\bullet}(LUMO)$	0.01375	-0.01375
3	PE-(HOMO) / PTFE*(HOMO)	0.11694	-0.24455	0.36149	$PE^{-}(HOMO) \rightarrow PTFE^{\bullet}(HOMO)$	0.01375	-0.01375
4	PE-(HOMO) / PTFE+(LUMO)	0.11694	-0.45945	0.57639	$PE^{-}(HOMO) \rightarrow PTFE^{+}(LUMO)$	0.01375	-0.01375
5	PE*(HOMO) / PTFE-(LUMO)	-0.20503	0.26369	-0.46872		0.00000	0.00000
6	PE*(HOMO) / PTFE*(LUMO)	-0.20503	-0.09374	-0.11129		0.00000	0.00000
7	PE*(HOMO) / PTFE*(HOMO)	-0.20503	-0.24455	0.03952	$PE^{\bullet}(HOMO) \rightarrow PTFE^{\bullet}(HOMO)$	0.10375	-0.10375
8	PE*(HOMO) / PTFE*(LUMO)	-0.20503	-0.45945	0.25442	$PE^{\bullet}(HOMO) \rightarrow PTFE^{+}(LUMO)$	0.02125	-0.02125
9	PE*(LUMO) / PTFE-(LUMO)	-0.05110	0.26369	-0.31479		0.00000	0.00000
10	PE*(LUMO) / PTFE*(LUMO)	-0.05110	-0.09374	0.04264	$PE^{\bullet}(LUMO) \rightarrow PTFE^{\bullet}(LUMO)$	0.10375	-0.10375
11	PE*(LUMO) / PTFE*(HOMO)	-0.05110	-0.24455	0.19345	$PE^{\bullet}(LUMO) \rightarrow PTFE^{\bullet}(HOMO)$	0.10375	-0.10375
12	PE*(LUMO) / PTFE*(LUMO)	-0.05110	-0.45945	0.40835	$PE^{\bullet}(LUMO) \rightarrow PTFE^{+}(LUMO)$	0.02125	-0.02125
13	PE ⁺ (HOMO) / PTFE ⁻ (LUMO)	-0.57603	0.26369	-0.83972		0.00000	0.00000
14	PE ⁺ (HOMO) / PTFE [•] (LUMO)	-0.57603	-0.09374	-0.48229		0.00000	0.00000
15	PE ⁺ (HOMO) / PTFE [•] (HOMO)	-0.57603	-0.24455	-0.33148		0.00000	0.00000
16	PE ⁺ (HOMO) / PTFE ⁺ (LUMO)	-0.57603	-0.45945	-0.11658		0.00000	0.00000

Soft contacting between PTFE_{don,i} and PE_{acc,i}, E_{PTFEdon,i}, E_{PEacc,i}, Δ E_{PTFEdon,i}/PEacc,i</sub>, Γ PTFE_{don,i}, Γ PE_{acc,i}, Γ PTFE_{don,i}/FePE_{acc,i}, and Γ CormTRfE_{FePTFEdon,i}/FePE_{acc,i}, and Γ CormTRfE_{FePTFEdon,i}/FePE_{acc,i}, Γ FePE_{acc,i}, Γ FePE_a

Path. No.	Soft contacting between PTFE _{don,i} and PE _{acc,i}	E _{PTFEdon,i} [a.u.]	E _{PEacc,i} [a.u.]	$\Delta E_{PTFEdon,i/PEacc,i}$ [a.u.]	Feasible f-electron transfer pathway from ^{Fe} PTFE _{don,i} to ^{Fe} PE _{acc,i}	$^{ m norm} TRf E_{ m FePTFEdon,i/FePEacc,i}$	$C^{norm}TRfE_{FePTFEdon,i/FePEacc,i}$
17	PTFE-(HOMO) / PE-(LUMO)	0.09858	0.22878	-0.13020		0.00000	0.00000
18	PTFE-(HOMO) / PE*(LUMO)	0.09858	-0.05110	0.14968	$PTFE^{-}(HOMO) \rightarrow PE^{\bullet}(LUMO)$	0.02125	-0.02125
19	PTFE-(HOMO) / PE*(HOMO)	0.09858	-0.20503	0.30361	$PTFE^{-}(HOMO) \rightarrow PE^{\bullet}(HOMO)$	0.02125	-0.02125
20	PTFE ⁻ (HOMO) / PE ⁺ (LUMO)	0.09858	-0.40480	0.50338	$PTFE^{-}(HOMO) \rightarrow PE^{+}(LUMO)$	0.01375	-0.01375
21	PTFE*(HOMO) / PE-(LUMO)	-0.24455	0.22878	-0.47333		0.00000	0.00000
22	PTFE*(HOMO) / PE*(LUMO)	-0.24455	-0.05110	-0.19345		0.00000	0.00000
23	PTFE*(HOMO) / PE*(HOMO)	-0.24455	-0.20503	-0.03952		0.0000	0.0000
24	PTFE*(HOMO) / PE*(LUMO)	-0.24455	-0.40480	0.16025	$PTFE^{\bullet}(HOMO) \rightarrow PE^{+}(LUMO)$	0.01375	-0.01375
25	PTFE*(LUMO) / PE-(LUMO)	-0.09374	0.22878	-0.32252		0.00000	0.00000
26	PTFE*(LUMO) / PE*(LUMO)	-0.09374	-0.05110	-0.04264		0.00000	0.00000
27	PTFE*(LUMO) / PE*(HOMO)	-0.09374	-0.20503	0.11129	$PTFE^{\bullet}(LUMO) \rightarrow PE^{\bullet}(HOMO)$	0.10375	-0.10375
28	PTFE*(LUMO) / PE*(LUMO)	-0.09374	-0.40480	0.31106	$PTFE^{\bullet}(LUMO) \rightarrow PE^{\dagger}(LUMO)$	0.01375	-0.01375
29	PTFE ⁺ (HOMO) / PE ⁻ (LUMO)	-0.61472	0.22878	-0.84350		0.00000	0.00000
30	PTFE ⁺ (HOMO) / PE [•] (LUMO)	-0.61472	-0.05110	-0.56362		0.00000	0.00000
31	PTFE ⁺ (HOMO) / PE [•] (HOMO)	-0.61472	-0.20503	-0.40969		0.00000	0.00000
32	PTFE+(HOMO) / PE+(LUMO)	-0.61472	-0.40480	-0.20992		0.00000	0.00000

 $500) - C^{Total,norm}TRfE_{FePTFEdon,i/FePEacc,i}$ (-0.18750). Accordingly, $\Delta C^{Net.norm}TRfE$ (-0.20750) is obtained via feasible *f-electron* transfer pathways in the settings where

$$\begin{array}{l} \varDelta^{Total}E\left(0.54767\text{ a.u.}\right)\\ =\Delta^{Total}E_{\text{FePEdon,i/FePTFEacc,i}}\left(2.08694\text{ a.u.}\right)-\Delta^{Total}E_{\text{FePTFEdon,i/FePEacc,i}}\\ _{\text{i}}\left(1.53927\text{ a.u.}\right) \end{array}$$

$$= \left\{ \begin{array}{l} {}^{Total}E_{FePEdon,i}(-0.21254a.u.) + {}^{Total}E_{FePEacc,i}(-1.67556a.u.) \right\} \\ - \left\{ \begin{array}{l} {}^{Total}E_{FePTFEdon,i}(-0.13629a.u.) + {}^{Total}E_{FePTFEacc,i}(-2.29948a.u.) \right\} \\ = {}^{Total}E_{whole,FePEdon,i}(-1.88810a.u.) - {}^{Total}E_{whole,FePTFEacc,i}(-2.43577a.u.) > 0 \end{array}$$

By introducing Total N_{whole,FePEdon,i} (14), and Total N_{whole,FePTFEacc,i} (14) into Eq. (5), the next equation would be obtained, as follows:

 $= \{^{Total}E_{whole,FePEdon,i} \; (-1.88810 \; a.u.) / \; ^{Total}N_{whole,FePEdon,i} \; (14)\} \times \\ ^{Total}N_{whole,FePEdon,i} \; (14) \; - \; \{^{Total}E_{whole,FePTFEacc,i} \; (-2.43577 \; a.u.) / \\ ^{Total}N_{whole,FePTFEacc,i} \; (14)\} \times \\ ^{Total}N_{whole,FePTFEacc,i} \; (14).$

$$=E_{Av.whole,FePEdon,i} \ (-0.13486 \ a.u.) \ \times ^{Total} \ N_{whole,FePEdon,i} \ (14)$$

$$-E_{Av.whole,FePTFEacc,i} (-0.17398 \ a.u.) \ \times ^{Total} \ N_{whole,FePTFEacc,i} (14) > 0$$

$$(6)$$

Therefore, $\Delta C^{\text{Net.norm}}$ TRfE (-0.20750) is obtained from the settings, where $E_{\text{Av.whole,FePEdon,i}}$ (-0.13486 a.u.) > $E_{\text{Av.whole,FePTFEacc,i}}$ (-0.17398 a.u.).

We concluded that the PE surface comprising whole, Fep Edon, i donated the *f*-electrons of $\Delta C^{\text{Net.norm}}$ TRfE(-0.20750) to whole, Fep PTFEacci via the feasible *f*-electron transfer pathways in the settings, where $E_{\text{Av.whole,FepPEdon,i}}$ (-0.13486 a.u.) > $E_{\text{Av.whole,FepTFEacc,i}}$ (-0.17398 a.u.). Consequently, the PE surface resulted in the positive net surface charge of PE (PE(+)) with Net.normSC soft, Av. PE(+0.20750). Although the PTFE surface comprising whole, Fep TFEacci accepts *f*-electrons of $\Delta C^{\text{Net.norm}}$ TRfE(-0.20750), and results in negative net surface charge of PTFE (PTFE(-)) with Net.normSC soft, Av. PTFE(-0.20750).

Here, we proposed the following charging mechanism of CE between PE and PTFE in a vacuum without the decay of the charge sources (illustrated in Fig. 3). Before "Hard Contacting," the PE and PTFE surfaces did not exhibit any electric charge. Stage 1.H.C.: The "Hard Contacting," between PE and PTFE produces PE_{ch.source.i}-{PE-(HOMO), PE*(HOMO), PE*(LUMO), PE*(HOMO), PE*(HOMO), PTFE*(HOMO), PTFE*(HOMO), PTFE*(HOMO), PTFE*(HOMO), PTFE*(HOMO), PTFE*(HOMO), PTFE*(HOMO), PTFE*(LUMO)} on the PTFE surface. PE (+, -, •) and PTFE(+, -, •)

refer to PE_{ch.source.i} and PTFE_{ch.source.i}, respectively. Although each surface-charge profile of PE or PTFE is exhibited as a nonuniform charge pattern arising from $PE_{ch.souce,i}$ or $PTFE_{ch.souece,i}$, respectively, $Total,norm SC_{hard,PEch.source,i}$ and $Total,norm SC_{hard,PTFEch.source,i}$ source,i are 0.00000, i.e., each surface charge is neutral. Stage 2.S. C.: "Soft Contacting," which refers to contacting without covalent bond scissions, between PE_{ch.source,i} and PTFE_{ch.souece,i} can induce f-electron transfer between PE_{ch.source,i} and PTFE_{ch.souece,i}. Regarding soft contacting, all the contact combinations between PE_{ch.source,i} and PTFE_{ch.source,i} were considered in 32 combinations. One category comprises 16 combinations of PEdoni and PTFEacci, i.e., 16 pathways that might account for f-electron transfer from PEdoni to PTFE_{acc.i}. We assumed that the transfer of f-electron from PE_{don.} i to PTFE_{acc,i} will be executed if E_{PEdon,i} is higher than E_{PTFEacc,i}, where the activation energy of f-electron transfer reaction is zero. Although the total number of pathways are 16, FepE_{don,i} can donate f-electron to FePTFE $_{acc,i}$ because $E_{FePEdon,i} > E_{FePTFEacc,i}$. $C^{Total,norm}$ -TRfE FePEdon,i /FePTFEacc,i. (-0.39500) indicates the amount of the normalized transferred f-electron from Total,FePEdon, to Total,FePTFEacc, via the $N_{\text{FePEdon,i}/\text{FePTFEacc,i}}$ (8) in the settings where $T_{\text{Total}}E_{\text{FePEdon,i}}$ (-0.21254 a.u.) > $T_{\text{Total}}E_{\text{FePTFEacc,i}}$ (-2.29948 a.u.). Another category comprises 16 combinations of PTFE_{don,i} and PE_{acc,i}. C^{Total,norm}- $TRfE^{\text{FePTFEdon,i/FePEacc,i}}$ (-0.187500) indicated the amount of the normalized transferred-f-electron from Total, FePTFEdon, to Total, FePEacc, via the Total N_{FePTFEdon,i}/FePEacc,i (6) in the settings where Total E_{FePTFEdon,i} $(-0.13629 \text{ a.u.}) > \text{Total}_{\text{EPPEacc,i}} (-1.67556 \text{ a.u.}). \text{ Thus, } \Delta C^{\text{Net.norm}} \text{TRfE}(-0.20750) = \{C^{\text{Total,norm}}\text{TRfE}_{\text{EPEdon,i}}, \text{feptfeach,i}_{\text{Feptfeach,i}}, \text{fept$ norm TRfE $_{FeFTFEdon,i/FePEacc,i}$ (-0.18750)] is obtained from the f-electron transfer from whole,Fe PE $_{don,i}$ to whole,Fe PTFE $_{acc,i}$ in the setting, where $E_{\text{Av.whole,FePEdon,i}}$ (-0.13486 a.u.) > $E_{\text{Av.whole,FePTFEacc,i}}$ (-0.17398 a.u.). Consequently, the PE surface comprising whole,FePEdon,i donated the f-electrons of $\Delta C^{\text{Net.norm}}$ TRfE (-0.20750) to obtain the positive net surface charge of PE (PE(+)) with Net.normSC_{soft,Av.whole}, FePEdon,i (+0.20750). However, the PTFE surface comprising whole, Fe PTFE_{acc,i} accepts *f*-electrons of $\Delta C^{\text{Net.norm}}$ TRfE(-0.20750) to obtain the negative net surface charge of PTFE (PTFE(-)) with Net.norm-SC_{soft,Av,PTFE}(-0.20750). The non-uniform surface charge patterns of PE(+) and PTFE(-) were altered compared with that before soft contacting. Stage 3.Sep: After soft contacting, they might be separated infinitely on a molecular scale. They might not be encountered each other. We consider that this process for Stage 1 to 3 is one set of contacting process. For conventional CE, many sets of contacting might be executed on the surface at a molecular

Next, after soft contact, we estimated the net amounts of the surface charges of PE and PTFE per square meter.

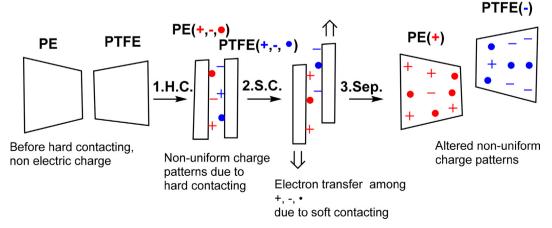


Fig. 3. Illustration of CE between PE and PTFE in a vacuum.

 $\textbf{Table 6} \\ \textbf{Soft contacting between $PE_{don,i}$ and $PVC_{acc,i}$, $E_{PEdon,i}$, $E_{PVCacc,i}$, $\Delta E_{PEdon,i/PVCacc,i}$, $F^{e}PE_{don,i}$, $F^{e}PVC_{acc,i}$, $normTRfE_{PEdon,i/PVCacc,i}$, and $C^{norm}TRfE_{PEdon,i/PVCacc,i}$, $archive a contacting between $PE_{don,i}$, $archive a contacting $PVC_{acc,i}$, $archive a contacting $PVC_$

Path. No.	Soft contacting between	$E_{PEdon,i}$	E _{PVCacc,i}	$\Delta E_{PEdon,i/PVCacc,i}$	Feasible f-electron transfer pathway	normTRfE _{PEdon,i/PVCacc,i}	C ^{norm} TRfE _{PEdon,i/PVCacc,i}
-	PE _{don,i} and PVC _{acc,i}	[a.u.]	[a.u.]	[a.u.]	from ^{Fe} PE _{don,i} to ^{Fe} PVC _{acc,i}		
33	PE-(HOMO) / PVC1-(LUMO)	0.11694	0.19426	-0.07732		0.00000	0.00000
34	PE-(HOMO) / PVC2-(LUMO)	0.11694	0.21606	-0.09912		0.000000	0.000000
35	PE-(HOMO) / PVC1*(LUMO)	0.11694	-0.08577	0.20271	$PE^{-}(HOMO) \rightarrow PVC1^{\bullet}(LUMO)$	0.006875	-0.006875
36	PE-(HOMO) / PVC1*(HOMO)	0.11694	-0.22376	0.34070	$PE^{-}(HOMO) \rightarrow PVC1^{\bullet}(HOMO)$	0.006875	-0.006875
37	PE-(HOMO) / PVC2*(LUMO)	0.11694	-0.07229	0.18923	$PE^{-}(HOMO) \rightarrow PVC2^{\bullet}(LUMO)$	0.006875	-0.006875
38	PE ⁻ (HOMO) / PVC2*(HOMO)	0.11694	-0.22663	0.34357	$PE^{-}(HOMO) \rightarrow PVC2^{\bullet}(HOMO)$	0.006875	-0.006875
39	PE-(HOMO) / PVC1+(LUMO)	0.11694	-0.39067	0.50761	$PE^{-}(HOMO) \rightarrow PVC1^{+}(LUMO)$	0.006875	-0.006875
40	PE-(HOMO)) /PVC2+(LUMO)	0.11694	-0.41251	0.52945	$PE^{-}(HOMO) \rightarrow PVC2^{+}(LUMO)$	0.006875	-0.006875
41	PE·(HOMO) / PVC1-(LUMO)	-0.20503	0.19426	-0.39929		0.000000	0.000000
42	PE*(HOMO) / PVC2-(LUMO)	-0.20503	0.21606	-0.42109		0.000000	0.000000
43	PE*(HOMO) / PVC1*(LUMO)	-0.20503	-0.08577	-0.11926		0.000000	0.000000
44	PE*(HOMO) / PVC1*(HOMO)	-0.20503	-0.22376	0.01873	PE [•] (HOMO) → PVC1 [•] (HOMO)	0.031250	-0.031250
45	PE*(HOMO) / PVC2*(LUMO)	-0.20503	-0.07229	-0.13274		0.000000	0.000000
46	PE*(HOMO) / PVC2*(HOMO)	-0.20503	-0.22663	0.02160	$PE^{\bullet}(HOMO) \rightarrow PVC2^{\bullet}(HOMO)$	0.031250	-0.031250
47	PE*(HOMO) / PVC1*(LUMO)	-0.20503	-0.39067	0.18564	$PE^{\bullet}(HOMO) \rightarrow PVC1^{+}(LUMO)$	0.031250	-0.031250
48	PE*(HOMO) / PVC2*(LUMO)	-0.20503	-0.41251	0.20748	$PE^{\bullet}(HOMO) \rightarrow PVC2^{+}(LUMO)$	0.031250	-0.031250
49	PE•(LUMO) / PVC1-(LUMO)	-0.05110	0.19426	-0.24536		0.000000	0.000000
50	PE*(LUMO) / PVC2*(LUMO)	-0.05110	0.21606	-0.26716		0.000000	0.000000
51	PE*(LUMO) / PVC1*(LUMO)	-0.05110	-0.08577	0.03467	PE [•] (LUMO) → PVC1 [•] (LUMO)	0.031250	-0.031250
52	PE*(LUMO) / PVC1*(HOMO)	-0.05110	-0.22376	0.17266	$PE^{\bullet}(LUMO) \rightarrow PVC1^{\bullet}(HOMO)$	0.031250	-0.031250
53	PE*(LUMO) / PVC2*(LUMO)	-0.05110	-0.07229	0.02119	$PE^{\bullet}(LUMO) \rightarrow PVC2^{\bullet}(LUMO)$	0.031250	-0.031250
54	PE*(LUMO) / PVC2*(HOMO)	-0.05110	-0.22663	0.17553	$PE^{\bullet}(LUMO) \rightarrow PVC2^{\bullet}(HOMO)$	0.031250	-0.031250
55	PE*(LUMO) / PVC1*(LUMO)	-0.05110	-0.39067	0.33957	$PE^{\bullet}(LUMO) \rightarrow PVC1^{+}(LUMO)$	0.031250	-0.031250
56	PE*(LUMO) / PVC2*(LUMO)	-0.05110	-0.41251	0.36141	$PE^{\bullet}(LUMO) \rightarrow PVC2^{+}(LUMO)$	0.031250	-0.031250
57	PE ⁺ (HOMO) / PVC1 ⁻ (LUMO)	-0.57603	0.19426	-0.77029		0.000000	0.000000
58	PE ⁺ (HOMO) / PVC2 ⁻ (LUMO)	-0.57603	0.21606	-0.79209		0.000000	0.000000
59	PE ⁺ (HOMO) / PVC1 [•] (LUMO)	-0.57603	-0.08577	-0.49026		0.000000	0.000000
60	PE ⁺ (HOMO) / PVC1•(HOMO)	-0.57603	-0.22376	-0.35227		0.000000	0.000000
61	PE ⁺ (HOMO) / PVC2•(LUMO)	-0.57603	-0.07229	-0.50374		0.000000	0.000000
62	PE ⁺ (HOMO) / PVC2 [•] (HOMO)	-0.57603	-0.22663	-0.34940		0.000000	0.000000
63	PE+(HOMO) / PVC1+(LUMO)	-0.57603	-0.39067	-0.18536		0.000000	0.000000
64	PE ⁺ (HOMO) / PVC2 ⁺ (LUMO)	-0.57603	-0.41251	-0.16352		0.000000	0.000000

In previous section, we obtained $^{Total}I_{f\text{-e,PTFE}}(3.9013\times10^{16}~[f\text{-electron/m}^2]),$ and we assumed that $^{Total}I_{f\text{-e,PTE}}=^{Total}I_{f\text{-e,PTFE}}.$ Thus, $\Delta\text{C}^{\text{Net.norm}}\text{TRfE}(-0.20750)$ was estimated as $-8.09524\times10^{15}~[f\text{-electron/m}^2].$ Assuming that one f-electron was equal to the elementary charge of one electron (1.602177 \times 10 $^{-19}~[C]).$ Thus, $\Delta\text{C}^{\text{Net.norm}}\text{TRfE}$ ($-8.09524\times10^{15}~[f\text{-electron/m}^2])$ was estimated as $-1.2970\times10^{-3}~[C/m^2].$ Namely, PE donated $-1.2970\times10^{-3}~[C/m^2]$ to PTFE, and results in PE(+) with a positive charge (+1.2970 \times 10 $^{-3}~[C/m^2]).$ However, PTFE accepted $-1.2970\times10^{-3}~[C/m^2]$ to obtain PTFE(-) with a negative charge ($-1.2970\times10^{-3}~[C/m^2]$).

Contrarily, Soh et al. reported that the charge value of PTFE induced by CE was PTFE(–) = $-6.7~[\mu\text{C/m}^2]$; there in, the materials were contacted 60 times [27]. Conversely, our result for PTFE (–1. $2970\times10^{-3}~[\text{C/m}^2]$) was over two orders of magnitude higher than that by Soh et al. (–6.7 $[\mu\text{C/m}^2]$). This large value might be due to the large concentration of $^{\text{Total}}I_{\text{Fe,PTFE}}(3.9013\times10^{16}~[\text{f-electron/m}^2]$, which was produced during vibratory ball-milling in a vacuum for 21 h at 77 K with 5 Hz. Additionally, in our experiment, charge leaking, the decay of the charge sources, as well as the effects of water and temperature, were abrogated.

3.2.2. CE of PE and PVC in vacuum

We discussed CE of PE and PVC based on the *f-electron* transfer between PE_{ch source,i} and PVC_{ch source,i} in a vacuum.

First, we considered the hard contact between PE and PVC in a vacuum. $PE_{ch.source,i}$ and $PVC_{ch.source,i}$ were produced and anchored to the PE and PVC surfaces via hard contacting. Consequently, the PE and PVC surfaces exhibited charge-mosaic patterns comprising nano-scale regional charges from $PE_{ch.source,i}$ and $PVC_{ch.source,i}$. Although both surfaces exhibited charge-mosaic patterns, $P_{ch.source,i}$ and $P_{ch.source,i}$ and $P_{ch.source,i}$ and $P_{ch.source,i}$ and $P_{ch.source,i}$ and $P_{ch.source,i}$ were 0.00000 (neutral).

Second, we considered the soft contact between $PE_{ch.source,i}$ and $PVC_{ch.source,i}$ in a vacuum. A total of 64 combinations were considered for all the possible contact between $PE_{ch.source,i}$ and $PVC_{ch.source,i}$, and they were divided into two categories. One category composed 32 combinations of $PE_{don,i}$ and $PVC_{acc,i}$, i.e., 32 pathways might account for f-electron transfer from $PE_{don,i}$ to $PVC_{acc,i}$. Another category comprised 32 combinations of $PVC_{don,i}$ and $PE_{acc,i}$, i.e., 32 pathways might account for f-electron transfer from $PVC_{don,i}$ to $PE_{acc,i}$. Regarding the soft contacting, we assumed that the total f-electrons of $PVC_{ch.source,i}$ ($f^{Total}I_{f^{-e},PVC}$) were identical to the $f^{Total}I_{f^{-e},PE}$, indicating that $f^{Total}I_{f^{-e},PE}$, indica

First, we considered f-electron transfer from $PE_{don,i}$ to $PVC_{acc,i}$ in a vacuum.

All the cases involving the soft contact of PE_{don,i} with PVC_{acc,i} are presented in Table 6 in (Pathways 33 –64). Although the total number of pathways were 32, we emphasized that FePE_{don,i} could donate *f-electron* to FePVC_{acc,i} in settings where ΔΕ_{FePEdon,i}/FePVCacc, i > 0, i.e., Ε_{FePEdon,i} > Ε_{FePVCacc,i}. Accordingly, each normTRfE_{FePEdon,i}/FePVCacc,i was obtained in settings where Ε_{FePEdon,i} > Ε_{FePVCacc,i}, i.e., those in which normTRfE_{FePEdon,i}/FePVCacc,i was assigned to low norm I_{ch.source,i} compared with high normI_{ch.source,i}. The feasible *f-electron* transfer pathways are presented in Table 6 (Pathways 35, 36, 37, 38, 39, 40, 44, 46, 47, 48, 51, 52, 53, 54, 55, and 56). PE_{don,i}, PVC_{acc,i}, E_{PEdon,i}, E_{PVCacc,i}, ΔΕ_{PEdon,i}/PVCacc,i</sub>, FePE_{don,i}, FePVC_{acc,i}, norm_TRfE_{FePEdon,i}/FePVCacc,i</sub>, and CnormTRfE_{FePEdon,i}/FePVCacc,i</sup> are presented in Table 6 (Pathways numerals). Further, Total,normTRfE_{FePEdon,i}/FePVCacc,i</sub> (0.353750) was obtained from the sum of normTRfE_{FePEdon,i}/FePVCacc,i or Total, normTRfE_{FePEdon,i}/FePVCacc,i</sub> (-0.353750) was the Total, normTRfE_{FePEdon,i}/FePVCacc,i</sup> (-0.353750) was

Regarding these feasible *f-electron* transfer pathways, $C^{Total,norm}$ -TRfE^{FePEdon,i/FePVCacc,i} (-0.353750) was obtained via the *f-electron* transfer pathways of $C^{Total}N_{FePEdon,i/FePVCacc,i}$ (16) in the settings, where

 $\begin{array}{l} \Delta^{Total} E_{FePEdon,i/FePVCacc,i} \ (3.65175 \ a.u.) = \{^{Total} E_{FePEdon,i} \ (-0.42508 \ a. \\ u.) - {}^{Total} E_{FePVCacc,i} \ (-4.07683 \ a.u.)\} > 0. \end{array}$

We emphasized that C^{Total,norm}TRfE_{FePEdon,i/FePVCacc,i} (-0.35375) indicated the amount of the normalized transferred-f-electron from Total,FePE_{don,i} to Total,FePVC_{acc,i} via the Total N_{FePEdon,i/FePVCacc,i} (16) in the settings where Total E_{FePEdon,i} (-0.42508 a.u.) > Total E_{FePVCacc,i} (-4.07683 a.u.).

Second, we considered f-electron transfer from $PVC_{don,i}$ to $PE_{acc,i}$ in a vacuum.

All the cases regarding the soft contacting of PVC $_{don,i}$ with PE $_{acc,i}$ are reported in Table 7 (Pathways 65 –96). Although the total number of Pathways are 32, we emphasized that $^{\rm Fe}$ PVC $_{don,i}$ could donate an f-electron to a $^{\rm Fe}$ PE $_{acc,i}$ in the settings where $\Delta E_{\rm FePVCdon,i}$ /FePEacc, $_i$ > 0, i.e., $E_{\rm FePVCdon,i}$ > $E_{\rm FePEacc,i}$. Accordingly, each $^{\rm norm}$ TRfE $_{\rm FePVCdon,i}$ /FePEacc, $_i$ is obtained in the settings where $E_{\rm FePVCdon,i}$ > $E_{\rm FePEacc,i}$, where $^{\rm norm}$ TRfE $_{\rm FePVCdon,i}$ /FePEacc, $_i$ was assigned to low $^{\rm norm}$ I $_{\rm ch,source,i}$ as compared with high $^{\rm norm}$ I $_{\rm ch,source,i}$. The feasible f-electron transfer pathways are presented in Table 7 (Pathways 66, 67, 68, 70, 71, 72, 76, 79, 80, 84, 87, and 88). PVC $_{\rm don,i}$, PE $_{\rm acc,i}$, $E_{\rm PVCdon,i}$, $E_{\rm PEacc,i}$, $^{\rm Te}$ PE $_{\rm acc,i}$, $^{\rm norm}$ TRfE $_{\rm FePVCdon,i}$ /FePEacc, $_{\rm i}$, and the C $_{\rm norm}$ TRfE $_{\rm FePVCdon,i}$ /FePEacc, $_{\rm i}$ are presented in Table 7 (Pathways numerals). Further, $^{\rm Total,norm}$ TRfE $_{\rm FePVCdon,i}$ /FePEacc, $_{\rm i}$ (0.228750) was obtained from the sum of $^{\rm norm}$ TRfE $_{\rm FePVCdon,i}$ /FePEacc, $_{\rm i}$ (0.228750) was obtained from the sum of $^{\rm norm}$ TRfE $_{\rm FePVCdon,i}$ /FePEacc, $_{\rm i}$ (0.228750) was the $^{\rm Total,norm}$ TRfE $_{\rm FePVCdon,i}$ /FePEacc, $_{\rm i}$ (0.228750) was obtained from the sum of $^{\rm norm}$ TRfE $_{\rm FePVCdon,i}$ /FePEacc, $_{\rm i}$ (0.228750) was obtained from the sum of $^{\rm norm}$ TRfE $_{\rm FePVCdon,i}$ /FePEacc, $_{\rm i}$ (0.228750) was obtained from the sum of $^{\rm norm}$ TRfE $_{\rm FePVCdon,i}$ /FePEacc, $_{\rm i}$ (0.228750) was obtained from the sum of $^{\rm norm}$ TRfE $_{\rm FePVCdon,i}$ /FePEacc, $_{\rm i}$ (0.228750) was obtained from the sum of $^{\rm norm}$ TRfE $_{\rm FePVCdon,i}$ /FePEacc, $_{\rm i}$ (0.228750) was obtained from the sum of $^{\rm norm}$ TRfE $_{\rm FePVCdon,i}$ /FePEacc, $_{\rm i}$ (0.228750) was obtained from the sum of $^{\rm norm}$ TRfE $_{\rm FePVCdon,i}$ /FePEacc, $_{\rm i}$ (0.228750) was obtained from the sum of $^{\rm n$

Regarding these feasible *f-electron* transfer pathways, $C^{Total,norm}$ -TRfE^{FePVCdon,i/FePEacc,i} (-0.228750) is obtained via the *f-electron* transfer pathways of $^{Total}N_{FePVCdon,i/FePEacc,i}$ (12) in the settings where Δ^{Total} -E_{FePVCdon,i/FePEacc,i} (3.06503 a.u.) = $^{Total}E_{FePVCdon,i}$ (-0.28609 a.u.) - $^{Total}E_{FePEacc,i}$ (-3.35112 a.u.)} > 0.

We emphasized that the C^{Total,norm}TRfE_{FePVCdon,i}/FePEacc,i</sub> (-0.228750) indicated the amount of the normalized transferred-f-electron from Total,FePVCdon,i to Total,FePEacc,i via the Total N_{FePVCdon,i}/FePEacc,i (12) in the settings where Total E_{FePVCdon,i} (-0.28609 a.u.) > Total E_{FePEacc,i} (-3.35112 a.u.).

Next, after soft contacting, we speculated Net.normSC_{soft,Av,PE} and a normalized net surface charge of PVC ($^{Net.norm}SC_{soft,Av,PVC}$) based on C^{Total,norm}TRfE_{FePEdon,i}/FePEdon,i</sub>/FePEdon,i (-0.353750) and C^{Total,norm}TRfE_{FePVCdon,i}/FePEdoc,i</sub> (-0.228750)

Regarding the soft contacting, $\Delta C^{\text{Net.norm}}TRfE$ is given: $\Delta C^{\text{norm}}-TRfE (-0.12500) = C^{\text{Total,norm}}TRfE_{\text{FePEdon,i/FePVCacc,i}} (-0.353750) - C^{\text{Total,norm}}TRfE_{\text{FePVCdon,i/FePEacc,i}} (-0.228750)$. Accordingly, $\Delta C^{\text{Net.norm}}TRfE(-0.12500)$ is obtained via feasible *f-electron* transfer pathways in the settings where

$$\Delta^{Total}E(0.58672 \text{ a.u.}) = \Delta^{Total}E_{FePEdon,i/FePVCacc,i} (3.65175 \text{ a.u.}) - \Delta^{Total}E_{FePVCdon,i/FePEacc,i} (3.06503 \text{ a.u.}).$$

$$=^{Total} E_{whole,FePEdon,i}(-3.77620 \ a.u.) - ^{Total} E_{whole,FePVCacc,i}(-4.36292 \ a.u.)$$
> 0.

(7)

By introducing Total N_{whole,FePEdon,i} (28), and Total N_{whole,FePVCacc,i} (28) into Eq. (7), the next equation would be obtained, as follows:

$$= \{^{Total}E_{whole,FePEdon,i} \; (-3.77620 \; a.u.) / ^{Total}N_{whole,FePEdon,i} \; (28)\} \times \\ ^{Total}N_{whole,FePEdon,i} \; (28) \; - \; \{^{Total}E_{whole,FePVCacc,i} \; (-4.36292 \; a.u.) / \\ ^{Total}N_{whole,FePVCacc,i} \; (28)\} \times \\ ^{Total}N_{whole,FePVCacc,i} \; (28).$$

$$= E_{A\nu,FePEdon,i}(-0.13486a.u.) \times^{Total} N_{whole,FePEdon,i}(28)$$

$$- E_{A\nu,FePVCacc,i}(-0.15582a.u.) \times^{Total} N_{whole,FePVCacc,i}(28)$$

$$> 0$$
(8)

Table 7Soft contacting between PVCdon,i and PEacc,i, $E_{PVCdon,i}$, $E_{PEacc,i}$, $\Delta E_{PVCdon,i/PEacc,i}$, $F_{epVCdon,i/PEacc,i}$,

Path. No.	Soft contacting between PVC _{don,i} and PE _{acc,i}	E _{PVCdon,i} [a.u.]	E _{PEacc,i} [a.u.]	$\Delta E_{PVCdon,i/PEacc,i}$ [a.u.]	Feasible f-electron transfer pathway from ^{Fe} PVC _{don,i} to ^{Fe} PE _{acc,i}	$^{ m norm} TRfE_{ m PVCdon,i/PEacc,i}$	$C^{norm}TRfE_{PVCdon,i/PEacc,i}$
65	PVC1 ⁻ (HOMO) / PE ⁻ (LUMO)	0.06700	0.22878	-0.16178		0.000000	0.000000
66	PVC1-(HOMO) / PE*(LUMO)	0.06700	-0.05110	0.11810	$PVC1^{-}(HOMO) \rightarrow PE^{\bullet}(LUMO)$	0.031250	-0.031250
67	PVC1 ⁻ (HOMO) / PE [•] (HOMO)	0.06700	-0.20503	0.27203	$PVC1^{-}(HOMO) \rightarrow PE^{\bullet}(HOMO)$	0.031250	-0.031250
68	PVC1 ⁻ (HOMO) / PE ⁺ (LUMO)	0.06700	-0.40480	0.47180	$PVC1^{-}(HOMO) \rightarrow PE^{+}(LUMO)$	0.006875	-0.006875
69	PVC2 ⁻ (HOMO) / PE ⁻ (LUMO)	0.09314	0.22878	-0.13564		0.000000	0.000000
70	PVC2 ⁻ (HOMO) / PE [•] (LUMO)	0.09314	-0.05110	0.14424	$PVC2^{-}(HOMO) \rightarrow PE^{\bullet}(LUMO)$	0.031250	-0.031250
71	PVC2-(HOMO) / PE*(HOMO)	0.09314	-0.20503	0.29817	$PVC2^{-}(HOMO) \rightarrow PE^{\bullet}(HOMO)$	0.031250	-0.031250
72	PVC2-(HOMO) / PE+(LUMO)	0.09314	-0.40480	0.49794	$PVC2^{-}(HOMO) \rightarrow PE^{+}(LUMO)$	0.006875	-0.006875
73	PVC1*(HOMO) / PE-(LUMO)	-0.22376	0.22878	-0.45254		0.000000	0.000000
74	PVC1*(HOMO) / PE*(LUMO)	-0.22376	-0.05110	-0.17266		0.000000	0.000000
75	PVC1*(HOMO) / PE*(HOMO)	-0.22376	-0.20503	-0.01873		0.000000	0.000000
76	PVC1*(HOMO) / PE*(LUMO)	-0.22376	-0.40480	0.18104	$PVC1^{\bullet}(HOMO) \rightarrow PE^{+}(LUMO)$	0.006875	-0.006875
77	PVC1°(LUMO) / PE-(LUMO)	-0.08577	0.22878	-0.31455		0.000000	0.000000
78	PVC1°(LUMO) / PE°(LUMO)	-0.08577	-0.05110	-0.03467		0.000000	0.000000
79	PVC1°(LUMO) / PE°(HOMO)	-0.08577	-0.20503	0.11926	PVC1*(LUMO) → PE*(HOMO)	0.031250	-0.031250
80	PVC1*(LUMO) / PE*(LUMO)	-0.08577	-0.40480	0.31903	$PVC1^{\bullet}(LUMO) \rightarrow PE^{+}(LUMO)$	0.006875	-0.006875
81	PVC2*(HOMO) / PE-(LUMO)	-0.22663	0.22878	-0.45541		0.000000	0.000000
82	PVC2*(HOMO) / PE*(LUMO)	-0.22663	-0.05110	-0.17553		0.000000	0.000000
83	PVC2*(HOMO) / PE*(HOMO)	-0.22663	-0.20503	-0.02160		0.000000	0.000000
84	PVC2*(HOMO) / PE*(LUMO)	-0.22663	-0.40480	0.17817	$PVC2^{\bullet}(HOMO) \rightarrow PE^{+}(LUMO)$	0.006875	-0.006875
85	PVC2*(LUMO) / PE-(LUMO)	-0.07229	0.22878	-0.30107		0.000000	0.000000
86	PVC2*(LUMO) / PE*(LUMO)	-0.07229	-0.05110	-0.02119		0.000000	0.000000
87	PVC2*(LUMO) / PE*(HOMO)	-0.07229	-0.20503	0.13274	PVC2*(LUMO) → PE*(HOMO)	0.031250	-0.031250
88	PVC2*(LUMO) / PE*(LUMO)	-0.07229	-0.40480	0.33251	$PVC2^{\bullet}(LUMO) \rightarrow PE^{+}(LUMO)$	0.006875	-0.006875
89	PVC1 ⁺ (HOMO) / PE ⁻ (LUMO)	-0.47085	0.22878	-0.69963		0.000000	0.000000
90	PVC1 ⁺ (HOMO) / PE [•] (LUMO)	-0.47085	-0.05110	-0.41975		0.000000	0.000000
91	PVC1 ⁺ (HOMO) / PE [•] (HOMO)	-0.47085	-0.20503	-0.26582		0.000000	0.000000
92	PVC1 ⁺ (HOMO) / PE ⁺ (LUMO)	-0.47085	-0.40480	-0.06605		0.000000	0.000000
93	PVC2+(HOMO) / PE-(LUMO)	-0.50460	0.22878	-0.73338		0.000000	0.000000
94	PVC2+(HOMO) / PE+(LUMO)	-0.50460	-0.05110	-0.45350		0.000000	0.000000
95	PVC2 ⁺ (HOMO) / PE [•] (HOMO)	-0.50460	-0.20503	-0.29957		0.000000	0.000000
96	PVC2 ⁺ (HOMO) / PE ⁺ (LUMO)	-0.50460	-0.40480	-0.09980		0.000000	0.000000

Path No.	Soft contacting between $PVC_{don,i}$ and $PTFE_{acc,i}$	E _{PVCdon,i} [a.u.]	E _{PTFEacc,i} [a.u.]	$\Delta E_{PVCdon,i/PTFEacc,i}$ [a.u.]	Feasible f-electron transfer pathway from $^{\rm Fe}$ PVC $_{ m don,i}$ to $^{\rm Fe}$ PTFE $_{ m acc,i}$	normTRfE _{PVCdon,i} /PTFEacc,i	$C^{norm}TRfE_{PVCdon,i/PTFEacc,i}$
97	PVC1 ⁻ (HOMO) / PTFE ⁻ (LUMO)	0.06700	0.26369	-0.19669		0.00000	0.00000
98	PVC1 ⁻ (HOMO) / PTFE*(LUMO)	0.06700	-0.09374	0.16074	PVC1 ⁻ (HOMO) → PTFE*(LUMO)	0.031250	-0.031250
99	PVC1 ⁻ (HOMO) / PTFE*(HOMO)	0.06700	-0.24455	0.31155	$PVC1^{-}(HOMO) \rightarrow PTFE^{\bullet}(HOMO)$	0.031250	-0.031250
100	PVC1 ⁻ (HOMO) / PTFE ⁺ (LUMO)	0.06700	-0.45945	0.52645	$PVC1^-(HOMO) \rightarrow PTFE^+(LUMO)$	0.010625	-0.010625
101	PVC2 ⁻ (HOMO) / PTFE ⁻ (LUMO)	0.09314	0.26369	-0.17065		0.00000	0.00000
102	PVC2-(HOMO) / PTFE*(LUMO)	0.09314	-0.09374	0.18688	$PVC2^{-}(HOMO) \rightarrow PTFE^{\bullet}(LUMO)$	0.031250	-0.031250
103	PVC2-(HOMO) / PTFE*(HOMO)	0.09314	-0.24455	0.33769	$PVC2^{-}(HOMO) \rightarrow PTFE^{\bullet}(HOMO)$	0.031250	-0.031250
104	PVC2-(HOMO) / PTFE+(LUMO)	0.09314	-0.45945	0.55259	$PVC2^{-}(HOMO) \rightarrow PTFE^{+}(HOMO)$	0.010625	0.010625
105	PVC1*(HOMO) / PTFE-(LUMO)	-0.22376	0.26369	-0.48745		0.000000	0.000000
106	PVC1*(HOMO) / PTFE*(LUMO)	-0.22376	-0.09374	-0.13002		0.000000	0.000000
107	PVC1*(HOMO) / PTFE*(HOMO)	-0.22376	-0.24455	0.33769	PVC1*(HOMO) → PTFE*(HOMO)	0.031250	-0.031250
108	PVC1*(HOMO) / PTFE*(LUMO)	-0.22376	-0.45945	0.23569	PVC1*(HOMO) → PTFE*(LUMO)	0.010625	-0.010625
109	PVC1*(LUMO) / PTFE-(LUMO)	-0.08577	0.26369	-0.34946		0.000000	0.000000
110	PVC1*(LUMO) / (PTFE*(LUMO)	-0.08577	-0.09374	0.00797	PVC1°(LUMO) → PTFE°(LUMO)	0.031250	-0.031250
111	PVC1*(LUMO) / PTFE*(HOMO)	-0.08577	-0.24455	0.15878	$PVC1^{\bullet}(LUMO) \rightarrow PTFE^{\bullet}(HOMO)$	0.03125	-0.03125
112	PVC1*(LUMO) / PTFE*(LUMO)	-0.08577	-0.45945	0.37368	$PVC1^{\bullet}(LUMO) \rightarrow PTFE^{+}(LUMO)$	0.010625	-0.010625
113	PVC2*(HOMO) / PTFE-(LUMO)	-0.22663	0.26369	-0.49032		0.000000	0.000000
114	PVC2*(HOMO) / PTFE*(LUMO)	-0.22663	-0.09374	-0.13289		0.000000	0.000000
115	PVC2*(HOMO) / PTFE*(HOMO)	-0.22663	-0.24455	0.01792	PVC2*(HOMO) → PTFE*(HOMO)	0.031250	-0.031250
116	PVC2*(HOMO) / PTFE*(LUMO)	-0.22663	-0.45945	0.23282	PVC2*(HOMO) → PTFE*(LUMO)	0.010625	-0.010625
117	PVC2*(LUMO) / PTFE-(LUMO)	-0.07229	0.26369	-0.33598		0.000000	0.000000
118	PVC2*(LUMO) / PTFE*(LUMO)	-0.07229	-0.09374	0.02145	PVC2°(LUMO) → PTFE°(LUMO)	0.031250	-0.031250
119	PVC2*(LUMO) / PTFE*(HOMO)	-0.07229	-0.24455	0.17226	$PVC2^{\bullet}(LUMO) \rightarrow PTFE^{\bullet}(HOMO)$	0.031250	-0.031250
120	PVC2*(LUMO) / PTFE*(LUMO)	-0.07229	-0.45945	0.38716	$PVC2^{\bullet}(LUMO) \rightarrow PTFE^{+}(LUMO)$	0.010625	-0.010625
121	PVC1 ⁺ (HOMO) / PTFE ⁻ (LUMO)	-0.47085	0.26369	-0.73454		0.000000	0.000000
122	PVC1 ⁺ (HOMO) / PTFE [•] (LUMO)	-0.47085	-0.09374	-0.37711		0.000000	0.000000
123	PVC1 ⁺ (HOMO) / PTFE*(HOMO)	-0.47085	-0.24455	-0.22630		0.000000	0.000000
124	PVC1 ⁺ (HOMO) / PTFE ⁺ (LUMO)	-0.47085	-0.45945	-0.01140		0.000000	0.000000
125	PVC2 ⁺ (HOMO) / PTFE ⁻ (LUMO)	-0.50460	0.26369	-0.76829		0.000000	0.000000
126	PVC2+(HOMO) / PTFE+(LUMO)	-0.50460	-0.09374	-0.41086		0.000000	0.000000
127	PVC2*(HOMO) / PTFE*(HOMO)	-0.50460	-0.24455	-0.26005		0.000000	0.000000
128	PVC2 ⁺ (HOMO) / PTFE ⁺ (LUMO)	-0.50460	-0.45945	-0.04515		0.000000	0.000000

 $\textbf{Table 9} \\ \textbf{Soft contacting between PTFE}_{don,i} \text{ and PVC}_{acc,i}, \text{ } E_{PTFEdon,i}, E_{PVCacc,i}, \text{ } \Delta E_{PTFEdon,i/PVCacc,i}, \text{ } ^{Fe}PTFE_{don,i}, \text{ } ^{Fe}PVC_{acc,i}, \text{ } \text{ } ^{norm}TRfE_{PTFEdon,i/PVCacc,i}, \text{ } and \text{ } C^{norm}TRfE_{PTFEdon,i/PVCacc,i}, \text{ }$

Path No.	Soft contacting between $PTFE_{don,i}$ and $PVC_{acc,i}$	E _{PTFEdon,i} [a.u.]	E _{PVCacc,i} [a.u.]	$\Delta E_{PTFEdon,i/PVCacc,i}$ [a.u.]	Feasible f-electron transfer pathway from FePTFE _{don,I} to FePVC _{acc,i}	$^{ m norm}$ TRfE $_{ m PTFEdon,i/PVCacc,i}$	C ^{norm} TRfE _{PTFEdon,i} /PVCacc,i
129	PTFE ⁻ (HOMO) / PVC1 ⁻ (LUMO)	0.09858	0.19426	-0.09568		0.000000	0.000000
130	PTFE-(HOMO) / PVC2-(LUMO)	0.09858	0.21606	-0.11748		0.000000	0.000000
131	PTFE-(HOMO) / PVC1*(LUMO)	0.09858	-0.08577	0.18435	PTFE ⁻ (HOMO) → PVC1*(LUMO)	0.010625	-0.010625
132	PTFE-(HOMO) / PVC1*(HOMO)	0.09858	-0.22376	0.32234	PTFE ⁻ (HOMO) → PVC1*(HOMO)	0.010625	-0.010625
133	PTFE-(HOMO) / PVC2*(LUMO)	0.09858	-0.07229	0.17087	PTFE ⁻ (HOMO) → PVC2*(LUMO)	0.010625	-0.010625
134	PTFE-(HOMO) / PVC2*(HOMO)	0.09858	-0.22663	0.32521	$PTFE^{-}(HOMO) \rightarrow PVC2^{\bullet}(HOMO)$	0.010625	-0.010625
135	PTFE ⁻ (HOMO) / PVC1 ⁺ (LUMO)	0.09858	-0.39067	0.48925	$PTFE^{-}(HOMO) \rightarrow PVC1^{+}(LUMO)$	0.010625	-0.010625
136	PTFE-(HOMO) / PVC2+(LUMO)	0.09858	-0.41251	0.51109	$PTFE^{-}(HOMO) \rightarrow PVC2^{+}(LUMO)$	0.010625	-0.010625
137	PTFE*(HOMO) / PVC1-(LUMO)	-0.24455	0.19426	-0.43881		0.000000	0.000000
138	PTFE*(HOMO) / PVC2-(LUMO)	-0.24455	0.21606	-0.46061		0.000000	0.000000
139	PTFE*(HOMO) / PVC1*(LUMO)	-0.24455	-0.08577	-0.15878		0.000000	0.000000
140	PTFE*(HOMO) / PVC1*(HOMO)	-0.24455	-0.22376	-0.02079		0.000000	0.000000
141	PTFE*(HOMO) / PVC2*(LUMO)	-0.24455	-0.07229	-0.17226		0.000000	0.000000
142	PTFE*(HOMO) / PVC2*(HOMO)	-0.24455	-0.22663	-0.01792		0.000000	0.000000
143	PTFE*(HOMO) / PVC1*(LUMO)	-0.24455	-0.39067	0.14612	$PTFE^{\bullet}(HOMO) \rightarrow PVC1^{+}(LUMO)$	0.031250	-0.031250
144	PTFE*(HOMO) / PVC2*(LUMO)	-0.24455	-0.41251	0.16796	$PTFE^{\bullet}(HOMO) \rightarrow PVC2^{+}(LUMO)$	0.031250	-0.031250
145	PTFE*(LUMO) / PVC1-(LUMO)	-0.09374	0.19426	-0.28800		0.000000	0.000000
146	PTFE*(LUMO) / PVC2*(LUMO)	-0.09374	0.21606	-0.30980		0.000000	0.000000
147	PTFE*(LUMO) / PVC1*(LUMO)	-0.09374	-0.08577	-0.00797		0.000000	0.000000
148	PTFE*(LUMO) / PVC1*(HOMO)	-0.09374	-0.22376	0.13002	PTFE*(LUMO) → PVC1*(HOMO)	0.031250	-0.031250
149	PTFE*(LUMO) / PVC2*(LUMO)	-0.09374	-0.07229	-0.02145		0.000000	0.000000
150	PTFE*(LUMO) / PVC2*(HOMO)	-0.09374	-0.22663	0.13289	PTFE*(LUMO) → PVC2*(HOMO)	0.031250	-0.031250
151	PTFE*(LUMO) / PVC1*(LUMO)	-0.09374	-0.39067	0.29693	$PTFE^{\bullet}(LUMO) \rightarrow PVC1^{+}(LUMO)$	0.031250	-0.031250
152	PTFE*(LUMO) / PVC2*(LUMO)	-0.09374	-0.41251	0.31877	$PTFE^{\bullet}(LUMO) \rightarrow PVC2^{+}(LUMO)$	0.031250	-0.031250
153	PTFE ⁺ (HOMO) / PVC1 ⁻ (LUMO)	-0.61472	0.19426	-0.80898		0.000000	0.000000
154	PTFE ⁺ (HOMO) / PVC2 ⁻ (LUMO)	-0.61472	0.21606	-0.83078		0.000000	0.000000
155	PTFE ⁺ (HOMO) / PVC1*(LUMO)	-0.61472	-0.08577	-0.52895		0.000000	0.000000
156	PTFE ⁺ (HOMO) / PVC1*(HOMO)	-0.61472	-0.22376	-0.39096		0.000000	0.000000
157	PTFE*(HOMO) / PVC2*(LUMO)	-0.61472	-0.07229	-0.54243		0.000000	0.000000
158	PTFE+(HOMO) / PVC2+(HOMO)	-0.61472	-0.22663	-0.38809		0.000000	0.000000
159	PTFE+(HOMO) / PVC1+(LUMO)	-0.61472	-0.39067	-0.22405		0.000000	0.000000
160	PTFE*(HOMO) / PVC2*(LUMO)	-0.61472	-0.41251	-0.20221		0.000000	0.000000

Therefore, $\Delta C^{\text{Net.norm}}$ TRfE (-0.12500) is obtained from the settings, where $E_{\text{Av.whole,FePEdon,i}}$ (-0.13486 a.u.) > $E_{\text{Av.whole,FePVCacc,i}}$ (-0.15582 a.u.).

We concluded that the PE surface comprising whole, FepEdon, i donated the *f*-electrons of $\Delta C^{\text{Net.norm}}$ TRfE(-0.12500) to whole, FePVCacci via the feasible *f*-electron transfer path ways in the settings, where EAV. whole, FepEdon, i (-0.13486 a.u.) > EAV. whole, FepVCacci (-0.15582 a.u.). Consequently, the PE surface resulted in the positive net surface charge of PE (PE(+)) with Net. norm SCsoft, AV. PE(+0.12500). Although the PVC surface comprising whole, FepVCacci accepts *f*-electrons of $\Delta C^{\text{Net.norm}}$ TRfE(-0.12500), and results in negative net surface charge of PVC (PVC(-)) with Net. norm SCsoft, AV. PVC(-0.12500).

3.2.3. Ce of PVC and PTFE in a vacuum

CE of PVC and PTFE were discussed based on the transfer of *felectron* between PVC_{ch.source.i} and PTFE_{ch.source.i} in a vacuum.

First, we considered the hard contact between PVC and PTFE in a vacuum. $PVC_{ch.source,i}$ and $PTFE_{ch.source,i}$ were produced and anchored to the PVC and PTFE surfaces via hard contacting. Consequently, both surfaces exhibited charge-mosaic patterns comprising nano-regional charges that arose from $PVC_{ch.source,i}$ and $PTFE_{ch.source,i}$. Although the PVC and PTFE surfaces exhibited charge-mosaic patterns, $Total,normSC_{hard,PVCch.source,i}$ and $Total,normSC_{hard,PVCch.source,i}$ were 0.00000 (neutral).

Second, we considered the soft contact between PVC_{ch.source,i} and PTFE_{ch.source,i} in a vacuum. A total of 64 possible contact combinations between PVC_{ch.source,i} and PTFE_{ch.source,i}, which are divided into two categories, were considered. One category comprised 32 combinations of the *f-electron* transfer from PVC_{don,i} and PTFE_{acc,i}, i.e., 32 pathways that might account for the *f-electron* transfer from PVC_{don,i} to PTFE_{acc,i}, and the other category comprised 32 combinations of the *f-electron* transfer from PTFE_{don,i} to PVC_{acc,i}, i.e., 32 pathways that might account for the *f-electron* transfer from PTFE_{don,i} to PVC_{acc,i}. We assumed that Total I_{f-e,PVC} was identical to Total I_{f-e,PTFE}, indicating that normI_{PVCch.source,i} and normI_{PTFEch.source,i} were normalized to the products by eight-time scissions.

First, we considered f-electron transfer from $PVC_{don,i}$ to $PTFE_{acc,i}$ in a vacuum.

All the combinations of the soft contacting of PVCdon,i with PTFE_{acc.i} are listed in Table 8 (Pathways 97–128). Although the total number of pathways were 32, we emphasized that FePVCdoni could donate an f-electron to FePTFE_{acc,i} in settings where $\Delta E_{FePVCdon,i/FePT}$ FEACC,i > 0, i.e., $E_{FePVCdon,i}$ > $E_{FePTFEACC,i}$. Accordingly, each normTRfE_{FePVdon,i/FePTFacc,i} was obtained in settings where $E_{FePVCdon,i} > E_{FePTFEacc,i}$, i.e., where ${}^{norm}TRfE_{FePVdon,i/FePTFacc,i}$ was assigned to low normI_{ch.source,i} compared with high normI_{ch.source,i}. The feasible f-electron transfer pathways are listed in Table 8 (Pathways 98, 99, 100, 102, 103, 104, 107, 108, 110, 111, 112, 115, 116, 118, 119, and 120). PVC_{don,i}, PTFE_{acc,i}, E_{PVCdon,i}, E_{PTFEacc,i}, Δ E_{FePVCdon,i}/ FePTFEacc,i, FePVCdon,i, FePTFEacc,i, normTRfEFePVCdon,i/FePTFEacc,i, and the CnormTRfE_{FePVCdon,i/FePTFEacc,i} are presented in Table 8 (Pathway numerals). Further, ${}^{Total,norm}TRfE_{FePVCdon,i/FePTFEacc,i}$ (0.37625) was obtained from the sum of $^{norm}TRfE_{FePVCdon,i/FePTFEacc,i}s$. $C^{Total,norm}$ - $TRfE^{FepVCdon,i/FePTFEacc,i}$ (-0.37625) was the ${}^{Total,norm}TRfE_{FepVCdon,i/FePTFEacc,i}$ with charge sign.

Regarding these feasible *f-electron* transfer pathways, C^{Total,norm}-TRfE^{FePVCdon,i/FePTFEacc,i} (-0.376250) is obtained via the *f-electron* transfer pathways of ^{Total}N_{FePVCdon,i/FePTFEacc,i} (16) in the settings where $\Delta^{\text{Total}}E_{\text{FePVCdon,i/FePTFEacc,i}}$ (3.70442 a.u.) = { $^{\text{Total}}E_{\text{FePVCdon,i}}$ (-0.89454 a.u.) - $^{\text{Total}}E_{\text{FePTFEacc,i}}$ (-4.59896 a.u.)} > 0.

We emphasized that the CTotal,normTRfEFePVCdon,i/FePTFEacc,i (-0.37625) indicated the amount of the normalized transferred-f-electron from Total,FePVCdon,i to Total,FePTFEacc,i via the TotalNFePVCdon,i/FePTFEacc,i (16) in settings where TotalEFePVCdon,i (-0.89454 a.u.) > TotalEFePTFEacc,i (-4.59896 a.u.).

Second, we considered f-electron transfer from PTFE $_{don,i}$ to PVC $_{acc,i}$ in a vacuum.

All the cases regarding the soft contacting of PTFE $_{don,i}$ with PVC $_{acc,i}$ are reported in Table 9 (Pathways 129–160). Although the total number of Pathways were 32, we emphasized that Fe-PTFE $_{don,i}$ could donate an f-electron to FePVC $_{acc,i}$ in the settings where $\Delta E_{FePTFEdon,i/PVCacc,i} > 0$, i.e., $E_{FePTFEdon,i} > E_{FePVCacc,i}$. Accordingly, each normTRfE $_{FePTFEdon,i/FePVCacc,i}$ was obtained in the settings where $E_{FePTFEdon,i} > E_{FePVCacc,i}$, i.e., those in which normTRfE $_{FePTFEdon,i/FePVCacc,i}$ was assigned to low normI $_{ch,source,i}$ as compared with high normI $_{ch,source,i}$. The feasible f-electron transfer pathways are presented in Table 9 (Pathways 131, 132, 133, 134, 135, 136, 143, 144, 148, 150, 151 and 152). PTFE $_{don,i}$, PVC $_{acc,i}$, $_{eptFEdon,i/FePVCacc,i}$, and $_{eptFEdon,i/FePVCacc,i}$, $_{eptFEdon,i/FePVCacc,i}$, $_{eptFEdon,i/FePVCacc,i}$, are presented in Table 9 (Pathways numerals). Further, the $_{eptFEdon,i/FePVCacc,i}$ are presented in Table 9 (Pathways numerals). Further, the $_{eptFEdon,i/FePVCacc,i}$ are presented in Table 9 (Pathways numerals). Further, the $_{eptFEdon,i/FePVCacc,i}$ (0.251250) was obtained from the sum of $_{eptFEdon,i/FePVCacc,i}$ (0.251250) was obtained from the sum of $_{eptFEdon,i/FePVCacc,i}$ (0.251250) rotal, normTRfE $_{eptFEdon,i/FePVCacc,i}$ (0.251250) was the $_{eptFEdon,i/FePVCacc,i}$ (0.251250)

Regarding these feasible *f-electron* transfer pathways, C^{Total,norm}-TRfE^{FePTFEdon,i/FePVCacc.i} (-0.251250) is obtained via the *f-electron* transfer pathways of Total,FeN_{FePTFEdon,i/FePVCacc.i} (12) in the settings where $\Delta^{\text{Total}}E_{\text{FePTFEdon,i/FePVCacc.i}}$ (3.19580 a.u.) = { $^{\text{Total}}E_{\text{FePTFEdon,i}}$ (-0.27258 a.u.) - Total EFePVCacc.i (-3.46838 a.u.)} > 0.

We emphasized that the $C^{Total,norm}TRfE_{FePTFEdon,i/FePVCacc,i}$ (-0.25125) indicated the amount of the normalized transferred-f-electron from $^{Total,Fe}PTFE_{don,i}$ to $^{Total,Fe}PVC_{acc,i}$ via the $^{Total}N_{FePTFEdon,i/FePVCacc,i}$ (12) in settings where $^{Total}E_{FePTFEdon,i}$ (-0.27258~a.u.) > $^{Total}E_{FePVCacc,i}$ (-3.46838~a.u.).

Next, after soft contacting, we speculated Net.normSC_{soft,Av,PVC} and Net.normSC_{soft,Av,PTFE} based on C^{Total,norm}TRfE_{FePVCdon,i}/FePTFEacc,i</sub> (-0.37625) and C^{Total,norm}TRfE_{FePTFEdon,i}/FePVCacc,i</sub> (-0.25125). Regarding soft contacting, Δ CNet.normTRfE is given as Δ CNet.norm-

Regarding soft contacting, $\Delta C^{\text{Net.norm}}$ TRfE is given as $\Delta C^{\text{Net.norm}}$ TRfE(-0.12500) = $C^{\text{Total.norm}}$ TRfE_{FePVCdon,i/FePTFEacc,i} (-0.37625) - $C^{\text{Total.norm}}$ TRfE_{FePTFEdon,i/FePVCacc,i} (-0.25125). Accordingly, $\Delta C^{\text{Net.norm}}$ TRfE (-0.12500) is obtained via feasible *f-electron* transfer pathways in the settings where.

$$\Delta^{Total}E(0.50862 \text{ a.u.})$$
= $\Delta^{Total}E_{FePVCdon,i/FePTFEacc,i}$ (3.70442 a.u.) - $\Delta^{Total}E_{FePTFEdon,i/FePV-Cacc,i}$ (3.19580 a.u.)

$$=^{\textit{Total}} \textit{E}_{\textit{whole},\textit{FePVCdon},i}(-4.36292 a.u.) - ^{\textit{Total}} \textit{E}_{\textit{whole},\textit{FePTFEacc},i}(-4.87154 a.u.) \\ > 0$$

By introducing Total N_{whole,FePVCdon,i} (28), and Total N_{whole,FePTFEacc,i} (28) into Eq. (9), the next equation would be obtained, as follows:

 $\begin{array}{l} = \{^{Total}E_{whole,FePVCdon,i}\;(-4.36292\;\text{a.u.})/\;^{Total}N_{whole,FePVCdon,i}\;(28)\}\\ \times\;^{Total}N_{whole,FePVCdon,i}\;(28)\;-\;\{^{Total}E_{whole,FePTFEacc,i}\;(-4.87154\;\text{a.}\\ \text{u.})/\;^{Total}N_{whole,FePTFEacc,i}\;(28)\}\;\times\;^{Total}N_{whole,FePTFEacc,i}\;(28). \end{array}$

$$= E_{Av.whole,FePVCdon,i}(-0.15582a.u.) \times^{Total} N_{whole,FePVCdon,i}(28)$$

$$- E_{Av.whole,FePTFEacc,i}(-0.17398a.u.) \times^{Total} N_{whole,FePTFEacc,i}(28)$$

$$> 0$$

$$(10)$$

Therefore, $\triangle C^{\text{Net.norm}} TRfE(-0.12500)$ is obtained in settings where $E_{\text{Av.whole,FePVCdon,i}}$ (-0.15582 a.u.) > $E_{\text{Av.whole,FePVFEacc,i}}$ (-0.17398 a.u.).

We concluded that the PVC surface comprising whole,FePVC_{don,i} donated *f-electrons* of $\Delta C^{\text{Net.norm}}$ TRFE(-0.12500) to the whole,FePTFE_{acc.i} via the feasible *f-electron* transfer pathways in the settings where E_{AV.whole,FePVCdon,i} (-0.15582 a.u.) > E_{AV.whole,FePTFEacc.i} (-0.17398 a.u.). Consequently, the PVC surface results in *positive*

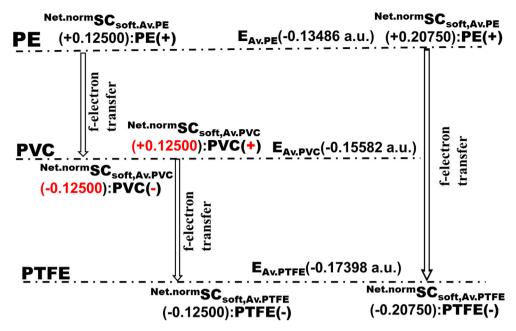


Fig. 4. Correlations of the net surface charges, average energy levels, and triboelectric series. Regarding the soft contact between PE and PVC, PE exhibited PE(+) with Net. norm $SC_{soft.Av.PVC}(-0.12500)$, while PVC exhibited PVC(-) with Net.norm $SC_{soft.Av.PVC}(-0.12500)$ because $EAv.PE(-0.13486 \text{ a.u.}) > E_{Av.PVC}(-0.15582 \text{ a.u.})$. Conversely, regarding the soft contact between PVC and PTFE, PVC exhibited PVC(+) with Net.norm $SC_{soft.Av.PVC}(+0.12500)$ since $E_{Av.PVC}(-0.15582 \text{ a.u.}) > E_{Av.PTEacc.i}(-0.17398 \text{ a.u.})$. These results indicated that the net surface charge depended on the average energy level of the *naked-activated-charge-sources*. The alignment of EAv.PE(-0.13486 a.u.)- $E_{Av.PVC}(-0.15582 \text{ a.u.})$ - $E_{Av.PVC}(-0.15582 \text{ a.u.})$ is identical to the alignment from positive to negative charge sign on the triboelectric series for PE-PVC-PTFE.

net surface charge of PVC (PVC(+)) with Net.normSC $_{soft,Av.}$ PVC(+0.12500). Although the PTFE surface comprising Whole,Fe_PTFE accepts f-electrons of $\Delta C^{Net.norm}TRfE(-0.12500)$, and results in negative net surface charge of PTFE (PTFE(-)) with Net.normSC $_{soft,Av.}$ PTFE(-0.12500).

3.2.4. Correlations among the charge signs, average energy levels and triboelectric series

We focused on the surface charge of PVC that was induced via contact with PE or PTFE.

Firstly, regarding CE between PE and PVC in a vacuum, we concluded that the PE surface comprising $^{whole,Fe}PE_{don,i}$ donated the f-electrons of $\Delta C^{\text{Net.norm}}TRfE(-0.12500)$ to $^{whole,Fe}PVC_{acc,i}$ via the feasible f-electron transfer path ways in settings where $E_{Av.whole,FePEdon,i}$ (-0.13486 a.u.) > $E_{Av.whole,FePVCacc,i}$ (-0.15582 a.u.). Thus, the PE surface yielded PE(+) with $^{\text{Net.norm}}SC_{soft,Av,PE}$ (+0.12500), while the PVC surface comprising $^{whole,Fe}PVC_{acc,i}$ accepted the f-electrons of $\Delta C^{\text{Net.norm}}TRfE$ (-0.12500), yielding PVC(-) with $^{\text{Net.norm}}SC_{soft,Av,PVC}$ (-0.12500).

Second, we concluded that the PVC surface comprising <code>whole,FePVCdon,i</code> donated the *f-electrons* of $\Delta C^{Net,norm}TRfE$ (-0.12500) to the <code>whole,FePTFEacc,i</code>, via the feasible *f-electron* transfer pathways in settings where $E_{AV,whole,FePVCdon,i}$ (-0.15582~a.u.) $> E_{AV,whole,FePTFEacc,i}$ (-0.17398~a.u.). Consequently, the PVC surface comprising <code>whole,FePVCdon,i</code> resulted in PVC(+) with <code>Net,normSC_soft,Av,PVC(+0.12500)</code>. Conversely, the PTFE surface comprising <code>whole,FePTFEacc,i</code> accepted the <code>f-electrons</code> of $\Delta C^{Net,norm}TRfE(-0.12500)$ to yield PTFE(-) with <code>Net,normSC_soft,Av,PTFE(-0.12500)</code>.

Regarding the soft contact between PE and PVC, PVC exhibited PVC(-) with $^{\rm Net.norm}SC_{\rm soft,Av.PVC}$ (-0.12500) because $E_{\rm Av.whole,FePE-don,i}$ (-0.13486~a.u.) > $E_{\rm Av.whole,FePVCacc,i}$ (-0.15582~a.u.). Conversely, in the soft contact between PVC and PTFE, PVC exhibited PVC(+) with $^{\rm Net.norm}SC_{\rm soft,Av.PVC}(+0.12500)$ because $E_{\rm Av.whole,FePVCdon,i}$ (-0.15582~a.u.) > $E_{\rm Av.whole,FePTFEacc,i}$ (-0.17398~a.u.), indicating the net surface charge depended on the average energy level of the <code>naked-activated-charge-sources</code> (Fig. 4).

Conversely, by aligning PE, PVC, and PTFE to their average energy levels, the alignment of PE(EAv.PE; -0.13486 a.u.)-PVC ($E_{\text{Av.PVC}}$; -0.15582 a.u.)-PTFE($E_{\text{Av.PTFE}}$; -0.17398 a.u.) was identical to the alignment from the positive to negative charge signs on the triboelectric series [28], indicating that the triboelectric series might be aligned to the polymer exhibiting high to low average energy levels.

4. Conclusions

Since CE is accompanied by polymer destruction, we assumed that it is also accompanied by covalent bond scission, which is induced by the "hard contacting" of polymers in a vacuum. Naked-activated-mechano-anions and naked-activated-mechanocations, which are produced by the heterogeneous covalent bond scission of the polymer main chain, and the naked-activatedmechano-radicals, which are produced by the homogeneous covalent bond scission of the polymer main chain, accounted for the charge sources. Moreover, these charge sources were anchored to the polymer surface and isolated on the surface in a vacuum. Thus, such a polymer surface exhibited a non-uniform charge pattern comprising a nano-scale charge region that was obtained from the charge sources. Although the surface exhibited a non-uniform charge pattern, the net charge was neutral. Sequentially, owing to "soft contacting," which does not induce covalent bond scission, the transfer of an *f-electron* from the charge sources as the donor to another charging source as the acceptor via feasible electron transfer pathways was accomplished via settings where the energy level of the charge source (donor) was higher than that the acceptor one, wherein the activation energy of the transfer of the f-electron was expected to be zero. A net surface charge was obtained employing a setting in which the average energy level of the charge sources as the donor is higher than that as acceptors. Consequently, the surface comprising the charge sources with a high average energy level exhibited positive net surface charge. However, another

surface comprising charge sources with a low average energy level exhibited a negative net surface charge.

We concluded thus: 1. The charge sources were nakedactivated-mechano-anions, naked-activated-mechano-cations and naked-activated-mechano-radicals, which were anchored and isolated on the surface in a vacuum. 2. An f-electron acted as the charge carrier. 3. The net surface charge was obtained via felectron transfer from the donor charge sources exhibiting a high average energy level to the acceptor charge sources exhibiting a low average energy level. Consequently, the surfaces comprising the donors and acceptors exhibited positive and negative net surface charges, respectively. 4. The net surface charge depended on the average energy level of the charge sources. 5. The alignment of PE(EAv.PE; -0.13486 a.u.)-PVC($E_{Av.PVC}$; -0.15582 a.u.)-PTFE (EAV.PTFE; -0.17398 a.u.) was identical to the alignment from the positive to negative charge signs on the triboelectric series [28]. The triboelectric series might be aligned to a polymer exhibiting high to low average energy levels.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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