# hydrogen properties v0 = 2.46607e15L = 6.265e8f = 0.4164g0 = 2.0Z = 2.0# physical constants me = 9.11e-28c = 3.00e10e = 4.80e-10Methods for calculations def calculate\_absorption\_coeff(v, v0): first = (e \*\* 2 \* f \* nH) / (4 \* pi \* me \* c) second = ((1 - x) \* g0) / (Z)third = (L) / ((v - v0) \*\* 2 + (L / 4 \* pi) \*\* 2)absorption\_coeff = first \* second \* third return absorption\_coeff def calculate\_intensity(d, wavelength, v0): # using v = c / wavelength formula v = c / wavelength# calculating absorption coefficent absorption\_coeff = calculate\_absorption\_coeff(v, v0) # calculate intensity with absorption coefficient intensity = exp(-absorption\_coeff \* d) return intensity Method for generating and plotting spectrum In [5]: # Reference: https://en.wikipedia.org/wiki/Lyman-alpha\_line # I have read that page and decided to pick wavelength start in 1200 def plot\_spectrum(d, v0, wavelength\_start=1200, wavelength\_end=1240, num=100000): # generate wavelengths with np.linspace wavelengths = np.linspace(wavelength\_start, wavelength\_end, num) # store the intensities that we're going to calculate

Next task is to generate a spectrum of light after it has passed through a slab of gas, and investigate the Lyman α absorption line of hydrogen. The part below shows how you can calculate a spectrum and the necessary

## intensities = [] # calculate intensities for certain wavelength for wavelength in wavelengths: # we're going to use wavelength / 10 \*\* 8 because # we want to represent our wavelength in Angstrom. # plotting fig, ax = plt.subplots(figsize=(8, 8)) ax.plot(wavelengths, intensities) plt.title("Light Intensity vs Wavelength") plt.xlabel("Wavelengths (Angstrom)") plt.ylabel("Intensities") plt.grid() # find minimum intensity of plotting min\_intesity = min(intensities) index\_min\_intesity = intensities.index(min\_intesity) # find wavelength that created minimum intensity # Print-out some information. print("Minimum intensity: {0}".format(min\_intesity)) print("d: {0} \t v0: {1}".format(d, v0)) Subtask 2.1 We need to generate spectrum with: d equals $10^{14}, 10^{18}, 10^{21}$ and v0 equals $2.46607*10^{15}$

from math import pi, exp

import matplotlib.pyplot as plt

information to complete the 3 subtasks.

**Providing constant features** 

# gas slab properties

nH = 0.1

x = 0.1

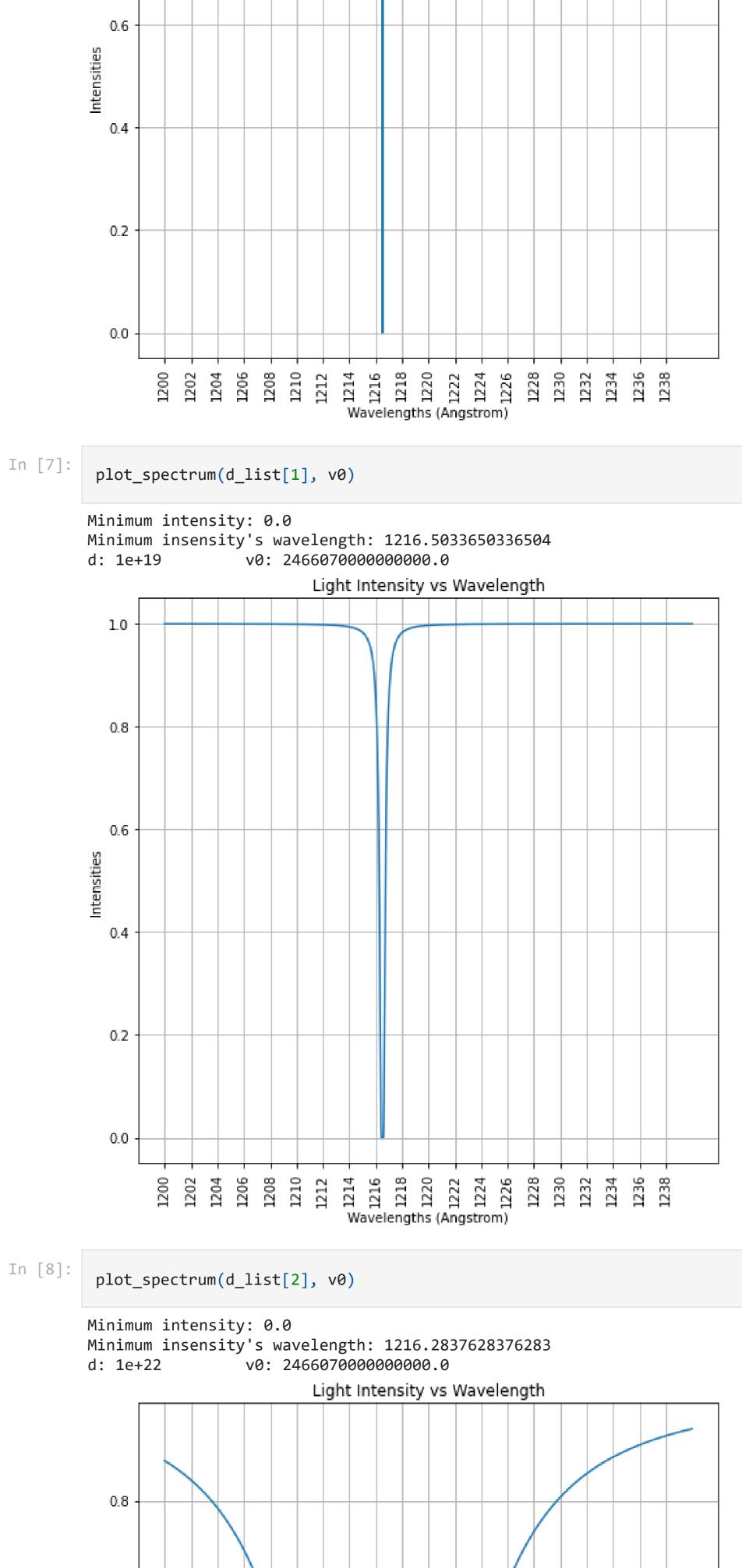
Task 2: Astrophysical Absorption Line Exercise

import numpy as np

In [2]:

## intensities.append(calculate\_intensity(d, wavelength / 10 \*\* 8, v0)) plt.xticks(np.arange(wavelength\_start, wavelength\_end, 2), rotation=90) min\_intensity\_wavelength = wavelengths[index\_min\_intesity] d\_list = [10e14, 10e18, 10e21] plot\_spectrum(d\_list[0], v0) Minimum intensity: 6.673493654441234e-26 Minimum insensity's wavelength: 1216.510565105651 Light Intensity vs Wavelength 1.0

## print("Minimum insensity's wavelength: {0}".format(min\_intensity\_wavelength)) In [6]: 0.8 0.6 0.2 0.0



0.6

0.2

0.0

0.6

0.2

0.0

Task 2.3 Commenting

• All graphs has same shape.

• When d increased more light gets absorbed.

1200 1202 1204 1206 1206 1210 1210 1216 1220 1224 1226 1228 1228 1230 1231 1231 1231 1232 1233

Wavelengths (Angstrom)

• All graphs minimum intensity in specific wavelength and that that wavelength is 1216 Angstrom.

• When v0 = 2.46632e15 graphics tend to be get 0 intensity faster than v0 = v0 = 2.46607e15 value.

• Actually all graphics has 0 intensity. There is some very very little values in some plottings but we can assume that they are 0.

• Lastly, understanding and proofing that Lyman alpha lines was interesting for me. I was scared at the beginning but now I can even comment about it.

Wavelengths (Angstrom) Subtask 2.2 We need to generate spectrum with: d equals  $10^{14}, 10^{18}, 10^{21}$  and v0 equals  $2.46632*10^{15}$ In [9]:  $v0_{new} = 2.46632e15$ plot\_spectrum(d\_list[0], v0\_new) Minimum intensity: 7.887566410202593e-18 Minimum insensity's wavelength: 1216.3873638736388 v0: 24663200000000000.0 Light Intensity vs Wavelength 1.0 0.8 0.6 0.2 0.0 1200 1202 1204 1206 1206 1210 1210 1216 1220 1224 1226 1228 1228 1230 1231 1231 1232 1233 In [10]: plot\_spectrum(d\_list[1], v0\_new) Minimum intensity: 0.0 Minimum insensity's wavelength: 1216.380163801638 v0: 24663200000000000.0 d: 1e+19 Light Intensity vs Wavelength 1.0 0.8 0.6 0.4 0.2 0.0 1200 1202 1204 1206 1208 1210 1210 1216 1226 1226 1228 1228 1228 1230 1231 1231 1232 1233 Wavelengths (Angstrom) In [11]: plot\_spectrum(d\_list[2], v0\_new) Minimum intensity: 0.0 Minimum insensity's wavelength: 1216.160561605616 d: 1e+22 v0: 24663200000000000.0 Light Intensity vs Wavelength

1214 1216 1218 1220 1224 1226 1230 1231 1234 1233