

材料科学基础（双语）考试（二）试卷答案

2016~2017 学年第二学期

A 卷选择和判断答案

一、

1-10 dbaab cdacb 11-20 cbacb ddcad 21-30 cbdbd bcaad

二、

1-10 FTFFT FTTT 11-20 FFFFF FTFFF

B 卷选择和判断答案

一、

1-10 daaab cdacb 11-20 cbabb ddcad 21-30 cbdbd bcaad

二、

1-10 FTFFT FTTT 11-20 TFFFF FTFFT

填空和大题答案

三、

1. Isotropic

2. $2\sqrt{5}R$

3. Crystalline

4. Growth

5. Cations Anions

6. Graphite diamond

7. Polymorphism allotropy

8. Smallest Translational

四、
1.

A

	X	Y	Z
Productions	$-\frac{2}{3}a$	$\frac{1}{2}b$	$0c$
in terms of $\frac{1}{2}$	$-\frac{2}{3}$	$\frac{1}{2}$	0
Reductions	-4	3	0
Enclosure	$[\bar{4} \ 3 \ 0]$		

D

	X	Y	Z
	$\frac{a}{6}$	$\frac{b}{2}$	$-c$
	$\frac{1}{6}$	$\frac{1}{2}$	-1
	1	3	-6
	$[1 \ 3 \ \bar{6}]$		

B

	X	Y	Z
	$\frac{2}{3}a$	$-b$	$\frac{2}{3}c$
	$\frac{2}{3}$	-1	$\frac{2}{3}$
	2	-3	2
	$[2 \ \bar{3} \ 2]$		

A $[\bar{4} \ 3 \ 0]$

B $[2 \ \bar{3} \ 2]$

C $[1 \ \bar{3} \ \bar{3}]$

D $[1 \ 3 \ \bar{6}]$

C

	X	Y	Z
	$\frac{1}{3}a$	$-b$	$-c$
	$\frac{1}{3}$	-1	-1
	1	-3	-3
	$[1 \ \bar{3} \ \bar{3}]$		

2.

Plane A

	x	y	z
Intercepts	$\frac{1}{3}a$	$\frac{1}{2}b$	$-\frac{1}{2}c$
in term of $\frac{1}{x}$	$\frac{1}{3}$	$\frac{1}{2}$	$-\frac{1}{2}$
Reciprocals	3	2	-2
Reductions	3	2	-2
Enclosure	(3 2 $\bar{2}$)		

Plane B

	x	y	z
Intercepts	$-\frac{1}{2}a$	∞b	$\frac{1}{2}c$
in term of $\frac{1}{x}$	$-\frac{1}{2}$	∞	$\frac{1}{2}$
Reciprocals	-2	0	2
Reductions	-2	0	2
Enclosure	($\bar{2}$ 0 2)		

Plane A (3 2 $\bar{2}$)

Plane B ($\bar{2}$ 0 2)

3.

3. Please determine the indices for the directions shown in the following hexagonal unit cell. Please show your steps. (2 points)

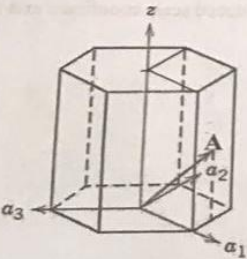
Solution

$[u'v'w'] = [211]$

$$\begin{cases} u = \frac{1}{3}(2u' - v') = \frac{1}{3}(4 - 1) = 1 \\ v = \frac{1}{3}(2v' - u') = \frac{1}{3}(2 - 2) = 0 \\ w = -(u + v) = -(1 + 0) = -1 \end{cases}$$

Thus, $[uvw] = [10\bar{1}]$

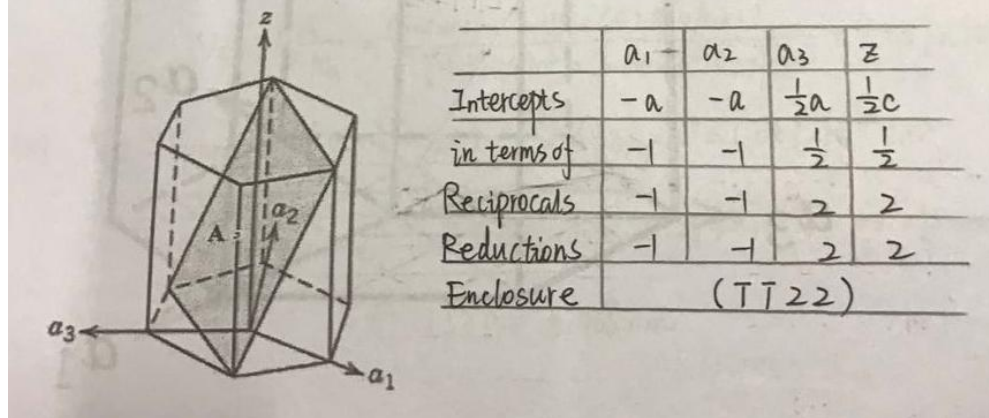
The indices for the direction A is $[10\bar{1}]$.



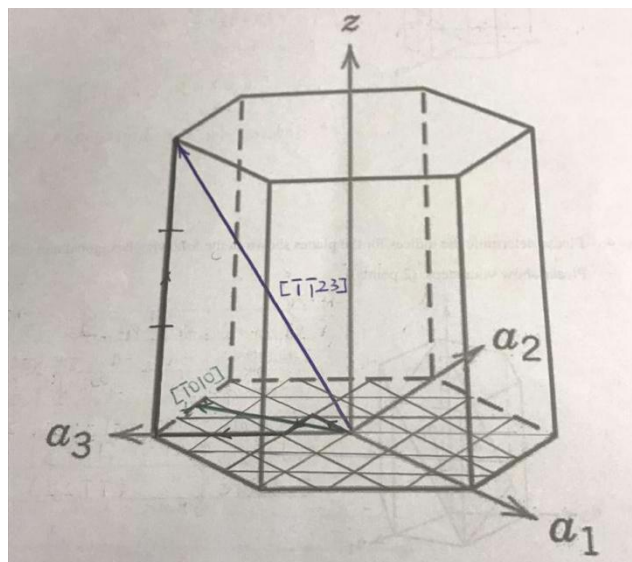
4.

4. Please determine the indices for the planes shown in the following hexagonal unit cell.

Please show your steps. (2 points)

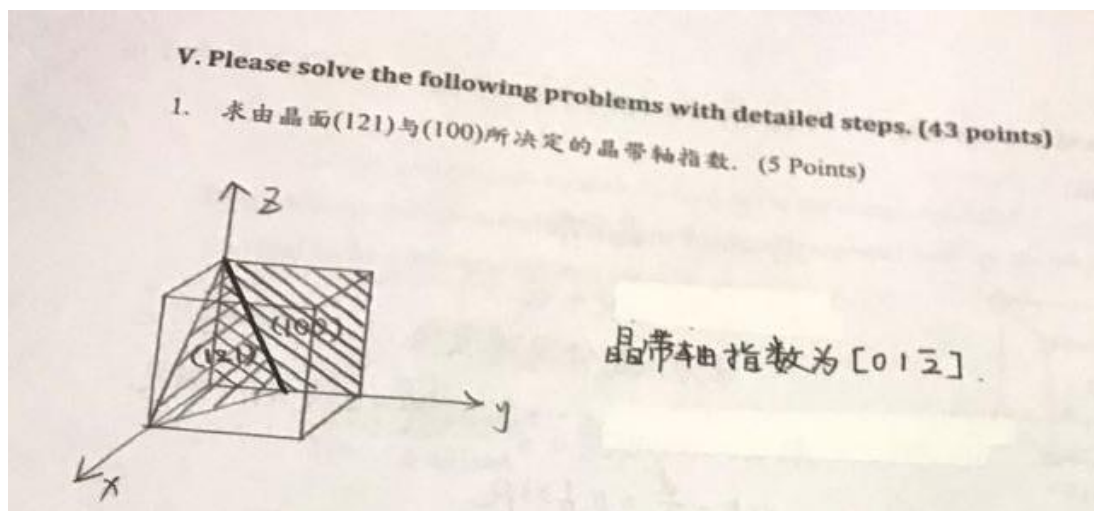


5.



五、

1.



2.

2. Titanium has an HCP crystal structure and a density of 4.51 g/cm^3 .

(1) What is the volume of its unit cell in cubic meters?

(2) If the c/a ratio is 1.58, compute the values of c and a . (6 points)

Ti: Atomic weight: 47.87 g/mol

Solution

(1) According to the density equation

$$V_c = \frac{nA}{\rho \cdot N_A} = \frac{(6 \text{ atoms/unit cell}) \times (47.87 \text{ g/mol})}{(4.51 \text{ g/cm}^3) \times (6.02 \times 10^{23} \text{ atom/mol})} = 1.057 \times 10^{-22} \text{ cm}^3/\text{unit cell}$$

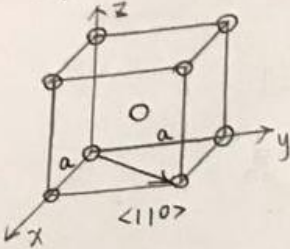
(2) $V = 3V_{\text{para}} = 3 \times (a \times \frac{\sqrt{3}}{2} a \times c) = \frac{3\sqrt{3}}{2} \times 1.58 \times a^3 = 1.057 \times 10^{-22} \text{ cm}^3$

Thus, $a = \sqrt[3]{\frac{2}{3\sqrt{3}} \times 1.057 \times 10^{-22} \text{ cm}^3} = 0.295 \times 10^{-7} \text{ cm} = 0.295 \text{ nm}$

$c = 1.58a = 0.466 \text{ nm}$

3.

解:



沿 $\langle 110 \rangle$ 晶向有:

$$\sqrt{2}a = 2R + d$$

d 为八面体间隙直径.

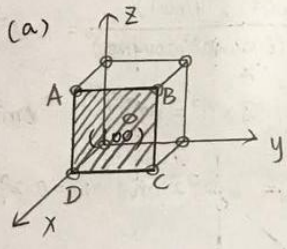
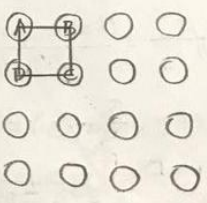
$$\Rightarrow d = \sqrt{2}a - 2R = \left(\frac{4\sqrt{2}}{\sqrt{3}} - 2\right)R = 1.266R$$

$$\therefore r = \frac{d}{2} = 0.633R.$$

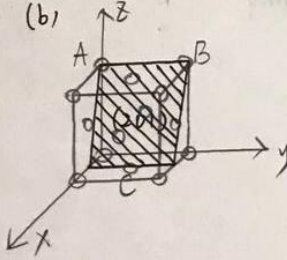
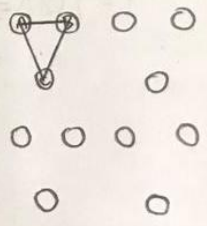
4.

Solution

(a)

(b)

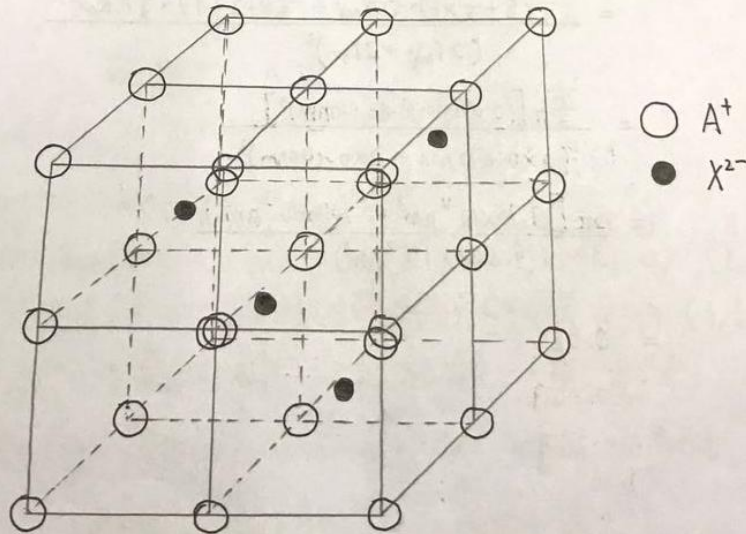
5.

Solution

$$(a) \quad \frac{r_{A^+}}{r_{X^{2-}}} = \frac{0.125 \text{ nm}}{0.145 \text{ nm}} = 0.862 \in (0.732, 1)$$

Thus, the coordination number for each A^+ is 8.

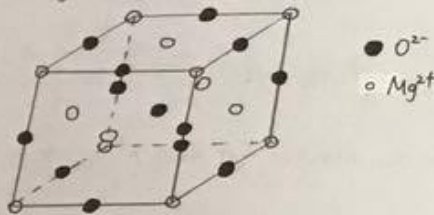
(b). The crystal structure for A_2X should be similar to Fluorite.



6.

Solution

MgO has the same crystal structure of NaCl.



Ionic radius of Mg^{2+} : 0.072 nm
 O^{2-} : 0.140 nm

$$IPF = \frac{n_{Mg^{2+}} V_{Mg^{2+}} + n_{O^{2-}} V_{O^{2-}}}{a^3}$$

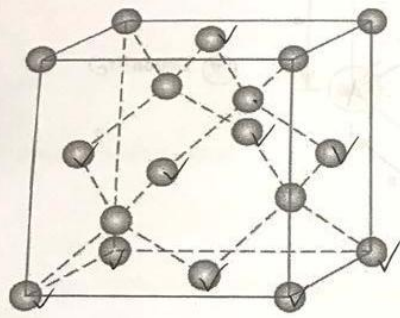
$$= \frac{(\frac{1}{8} \times 8 + \frac{1}{2} \times 6) \times \frac{4}{3} \pi R_{Mg^{2+}}^3 + (\frac{1}{4} \times 12 + 1) \times \frac{4}{3} \pi R_{O^{2-}}^3}{(2r_{Mg^{2+}} + 2r_{O^{2-}})^3}$$

$$= \frac{\frac{16}{3} \pi [(0.072 \text{ nm})^3 + (0.140 \text{ nm})^3]}{(2 \times 0.072 \text{ nm} + 2 \times 0.140 \text{ nm})^3}$$

$$= \frac{2\pi [3.73 \times 10^{-4} \text{ nm}^3 + 2.744 \times 10^{-3} \text{ nm}^3]}{3 \times (9.528 \times 10^{-3} \text{ nm}^3)}$$

$$= 0.69$$

7.



$$\frac{\sqrt{3}}{4} a = 2R$$

Atomic radius of Si: 0.118 nm

● Si

Solution

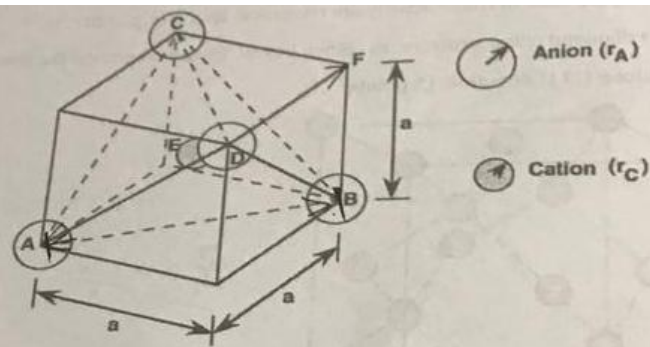
The crystal structure of Si is similar to ZnS .

The coordinations of Si: $(0,0,0)$, $(1,0,1)$, $(1,0,0)$, $(0,1,1)$, $(1,1,1)$, $(0,0,1)$, $(0,1,0)$, $(1,1,0)$, $(\frac{1}{2}, 0, \frac{1}{2})$, $(\frac{1}{2}, \frac{1}{2}, 0)$, $(0, \frac{1}{2}, \frac{1}{2})$, $(\frac{1}{2}, 1, \frac{1}{2})$, $(\frac{1}{2}, \frac{1}{2}, 1)$, $(1, \frac{1}{2}, \frac{1}{2})$, $(\frac{1}{4}, \frac{1}{4}, \frac{3}{4})$, $(\frac{1}{4}, \frac{3}{4}, \frac{1}{4})$, $(\frac{3}{4}, \frac{1}{4}, \frac{1}{4})$, $(\frac{3}{4}, \frac{3}{4}, \frac{3}{4})$.

Thus, along the $[111]$, there are two atoms in total.

$$LD_{[111]} = \frac{2 \text{ atoms}}{\sqrt{3} a} = \frac{2 \text{ atoms}}{\sqrt{3} \times \frac{2R}{\frac{\sqrt{3}}{4}}} = \frac{1}{4R} = \frac{1}{4 \times 0.118 \text{ nm}} = 2.119 \text{ nm}^{-1}$$

8.



Solution

① $AB = 2r_A = \sqrt{2}a$

$$\Rightarrow a = \sqrt{2} r_A$$

There will be an anion located at F position.

$$\textcircled{2} AEF = 2r_A + 2r_C = \sqrt{AB^2 + BF^2} \\ = \sqrt{4r_A^2 + 2r_A^2} = \sqrt{6} r_A$$

$$\Rightarrow 2 + 2 \frac{r_c}{r_a} = \sqrt{6}$$

Thus, $\frac{r_c}{r_a} = \frac{\sqrt{6}-2}{2} = 0.225$