

### UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS General Certificate of Education Advanced Level

CANDIDATE NAME					
CENTRE NUMBER			CANDIDATE NUMBER		

571913318

PHYSICS 9702/43

Paper 4 A2 Structured Questions

October/November 2011

2 hours

Candidates answer on the Question Paper.

No Additional Materials are required.

#### **READ THESE INSTRUCTIONS FIRST**

Write your Centre number, candidate number and name on all the work you hand in.

Write in dark blue or black pen.

You may use a soft pencil for any diagrams, graphs or rough working.

Do not use staples, paper clips, highlighters, glue or correction fluid.

DO NOT WRITE IN ANY BARCODES.

Answer all questions.

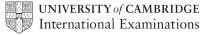
You may lose marks if you do not show your working or if you do not use appropriate units.

At the end of the examination, fasten all your work securely together.

The number of marks is given in brackets [ ] at the end of each question or part question.

For Examiner's Use			
1			
2			
3			
4			
5			
6			
7			
8			
9			
10			
11			
12			
Total			

This document consists of 21 printed pages and 3 blank pages.



 $G = 6.67 \times 10^{-11} \,\mathrm{N}\,\mathrm{m}^2\,\mathrm{kg}^{-2}$ 

## **Data**

speed of light in free space,	$c = 3.00 \times 10^8 \mathrm{ms^{-1}}$
permeability of free space,	$\mu_0 = 4\pi \times 10^{-7} \mathrm{Hm^{-1}}$
permittivity of free space,	$\varepsilon_0 = 8.85 \times 10^{-12}  \mathrm{F  m^{-1}}$
elementary charge,	$e = 1.60 \times 10^{-19} \mathrm{C}$
the Planck constant,	$h = 6.63 \times 10^{-34} \mathrm{Js}$
unified atomic mass constant,	$u = 1.66 \times 10^{-27} \text{ kg}$
rest mass of electron,	$m_{\rm e} = 9.11 \times 10^{-31} \text{ kg}$
rest mass of proton,	$m_{\rm p} = 1.67 \times 10^{-27}  \rm kg$
molar gas constant,	$R = 8.31 \text{ J K}^{-1} \text{ mol}^{-1}$
the Avogadro constant,	$N_{\rm A} = 6.02 \times 10^{23}  {\rm mol}^{-1}$
the Boltzmann constant,	$k = 1.38 \times 10^{-23} \mathrm{JK^{-1}}$

gravitational constant,

#### **Formulae**

uniformly accelerated motion,	$s = ut + \frac{1}{2}at^2$
	$v^2 = u^2 + 2as$

work done on/by a gas, 
$$W = p\Delta V$$

gravitational potential, 
$$\phi = -\frac{Gm}{r}$$

hydrostatic pressure, 
$$p = \rho gh$$

pressure of an ideal gas, 
$$p = \frac{1}{3} \frac{Nm}{V} < c^2 >$$

simple harmonic motion, 
$$a = -\omega^2 x$$

velocity of particle in s.h.m., 
$$v = v_0 \cos \omega t$$
 
$$v = \pm \omega \sqrt{(x_0^2 - x^2)}$$

electric potential, 
$$V = \frac{Q}{4\pi\varepsilon_0 r}$$

capacitors in series, 
$$1/C = 1/C_1 + 1/C_2 + \dots$$

capacitors in parallel, 
$$C = C_1 + C_2 + \dots$$

energy of charged capacitor, 
$$W = \frac{1}{2}QV$$

resistors in series, 
$$R = R_1 + R_2 + \dots$$

resistors in parallel, 
$$1/R = 1/R_1 + 1/R_2 + \dots$$

alternating current/voltage, 
$$X = X_0 \sin \omega t$$

radioactive decay, 
$$X = X_0 \exp(-\lambda t)$$

decay constant, 
$$\lambda \, = \frac{0.693}{t_{\scriptscriptstyle \frac{1}{2}}}$$

# **Section A**

For
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Use

Answer **all** the questions in the spaces provided.

1	its r	nass	net Mars may be considered to be an isolated sphere of diameter $6.79 \times 10^6$ m with of $6.42 \times 10^{23}$ kg concentrated at its centre. f mass 1.40 kg rests on the surface of Mars.
	For	this	rock,
	(a)	(i)	determine its weight,
		(ii)	$\mbox{weight} = \mbox{N } [3]$ show that its gravitational potential energy is $-1.77 \times 10^7 \mbox{J}.$
	(b)		[2] the information in <b>(a)(ii)</b> to determine the speed at which the rock must leave the ace of Mars so that it will escape the gravitational attraction of the planet.
			speed = ms <sup>-1</sup> [3]

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(c)		mean translational kinetic energy $<\!E_{\rm K}\!>$ of a molecule of an ideal gas is given by the ression
		$\langle E_{K} \rangle = \frac{3}{2}kT$
	whe	ere $T$ is the thermodynamic temperature of the gas and $k$ is the Boltzmann constant.
	(i)	Determine the temperature at which the root-mean-square (r.m.s.) speed of hydrogen molecules is equal to the speed calculated in <b>(b)</b> . Hydrogen may be assumed to be an ideal gas. A molecule of hydrogen has a mass of 2 u.
	(ii)	temperature = K [2] State and explain one reason why hydrogen molecules may escape from Mars at temperatures below that calculated in (i).

	esistance thermometer and a thermocouple thermometer are both used at the same e to measure the temperature of a water bath.	For Examiner's Use
	plain why, although both thermometers have been calibrated correctly and are at illibrium, they may record different temperatures.	036
( <b>b)</b> Stat	[2] te	
(i)	in what way the absolute scale of temperature differs from other temperature scales,	
	[1]	
(ii)	what is meant by the absolute zero of temperature.	
	[1]	
	e temperature of a water bath increases from 50.00°C to 80.00°C. ermine, in kelvin and to an appropriate number of significant figures,	
(i)	the temperature 50.00 °C,	
	tomporatura V [1]	
(ii)	the change in temperature of the water bath.	
	temperature change = K [1]	
(ii)		

(a)	Define simple harmonic motion.	For Examin
		Use
	[2]	
(b)	A horizontal plate is vibrating vertically, as shown in Fig. 3.1.	
	cube, mass 5.8 g	
	vertical oscillations frequency 4.5 Hz	
	Fig. 3.1	
	The plate undergoes simple harmonic motion with a frequency of 4.5 Hz and amplitude 3.0 mm. A metal cube of mass 5.8 g rests on the plate.	
	Calculate, for the cube, the energy of oscillation.	
	energy = J [3]	
(c)	The amplitude of oscillation of the plate in <b>(b)</b> is gradually increased. The frequency remains constant.  At one particular amplitude, the cube just loses contact momentarily with the plate.	
	(i) State the position of the plate in its oscillation at the point when the cube loses contact.	
	[2]	

(ii)	Calculate this amplitude of oscillation.	For Examiner's Use
	amplitude = m [2]	

For Examiner's Use

4	(a)	Stat	e two functions of capacitors in electrical circuits.
		1	
		2	
			[2]
	(b)	Thre	ee uncharged capacitors of capacitance $C_1$ , $C_2$ and $C_3$ are connected in series, as wn in Fig. 4.1.
			plate A $C_1$ $C_2$ $C_3$
			Fig. 4.1
		A cł	narge of $+Q$ is put on plate A of the capacitor of capacitance $C_1$ .
		(i)	State and explain the charges that will be observed on the other plates of the capacitors. You may draw on Fig. 4.1 if you wish.
			[2]
		(ii)	Use your answer in (i) to derive an expression for the combined capacitance of the capacitors.
			[2]

(c) A capacitor of capacitance  $12\,\mu\text{F}$  is charged using a battery of e.m.f. 9.0V, as shown in Fig. 4.2.

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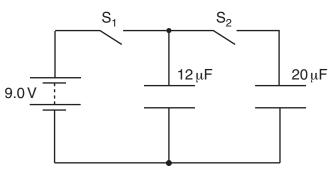


Fig. 4.2

Switch  $\mathbf{S}_1$  is closed and switch  $\mathbf{S}_2$  is open.

(i) The capacitor is now disconnected from the battery by opening S<sub>1</sub>. Calculate the energy stored in the capacitor.

energy =		J	[2]
----------	--	---	-----

(ii) The  $12\,\mu\text{F}$  capacitor is now connected to an uncharged capacitor of capacitance  $20\,\mu\text{F}$  by closing  $S_2$ . Switch  $S_1$  remains open.

The total energy now stored in the two capacitors is  $1.82 \times 10^{-4} \, \text{J}$ .

Suggest why this value is different from your answer in (i).

		[1]

5 The components for a bridge rectifier are shown in Fig. 5.1.



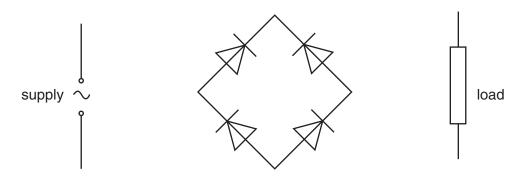


Fig. 5.1

(a)		nplete the circuit of Fig. 5.1 by showing the connections of the supply and of the load ne diodes.
(b)	_	gest one advantage of the use of a bridge rectifier, rather than a single diode, for the ification of alternating current.
		[1]
(c)	Stat	re
	(i)	what is meant by <i>smoothing</i> ,
		[1]
	(ii)	the effect of the value of the capacitance of the smoothing capacitor in relation to smoothing.

6 (a)		Define the <i>tesla</i> .		
			Examiner's Use	
		[0]		

**(b)** A charged particle of mass m and charge +q is travelling with velocity v in a vacuum. It enters a region of uniform magnetic field of flux density B as shown in Fig. 6.1.

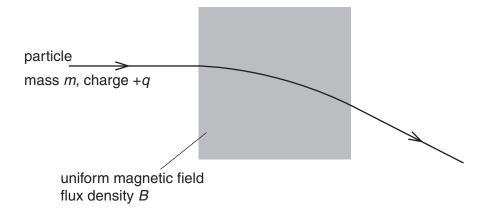


Fig. 6.1

The magnetic field is normal to the direction of motion of the particle. The path of the particle in the field is the arc of a circle of radius r.

(i)	Explain why the path of the particle in the field is the arc of a circle.
	01

(ii) Show that the radius r is given by the expression

$$r = \frac{mv}{Bq}$$
.

[1]

(c) A uniform magnetic field is produced in the region PQRS, as shown in Fig. 6.2.



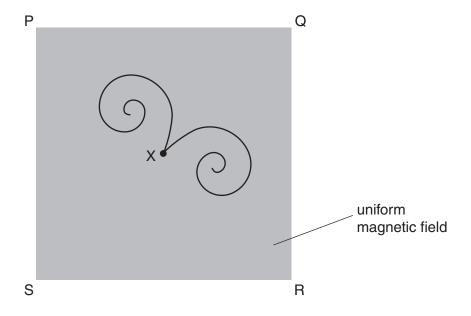


Fig. 6.2

The magnetic field is normal to the page.

At point X, a gamma-ray photon interaction causes two particles to be formed. The paths of these particles are shown in Fig. 6.2.

(i)	Suggest, with a reason, why each of the paths is a spiral, rather than the arc of a circle.
	[2]
(ii)	State and explain what can be deduced from the paths about
	1. the charges on the two particles,
	[2]
	2. the initial speeds of the two particles.
	[2]

7	An explanation of the photoelectric effect includes the terms photon energy and work function
	energy.

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(a) Explain what is meant by

(i)	a photon,
	[2]
(ii)	work function energy.

(b) In an experiment to investigate the photoelectric effect, a student measures the wavelength  $\lambda$  of the light incident on a metal surface and the maximum kinetic energy  $E_{\text{max}}$  of the emitted electrons. The variation with  $E_{\text{max}}$  of  $\frac{1}{\lambda}$  is shown in Fig. 7.1.

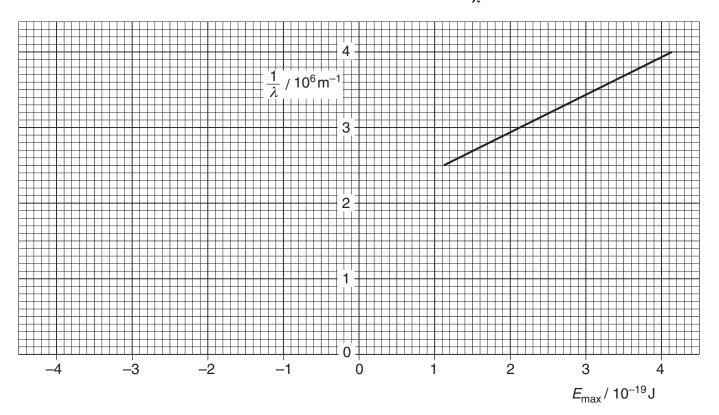


Fig. 7.1

(i)	The work function energy of the metal surface is $\Phi$ . State an equation, in terms of $\lambda$ , $\Phi$ and $E_{\rm max}$ , to represent conservation of energy for the photoelectric effect. Explain any other symbols you use.
	[2]

(ii)	Use	e your answer in (i) and Fig. 7.1 to determine	For
	1.	the work function energy $arPhi$ of the metal surface,	Examiner's Use
		Φ = J [2]	
	2.	a value for the Planck constant.	
		Planck constant =	

8

Radon-222, when found in atmospheric air, can present a health hazard. Safety measurs should be taken when the activity of radon-222 exceeds 200 Bq per cubic metre of air.  (a) (i) Define radioactive <i>decay constant</i> .  (ii) Show that the decay constant of radon-222 is 2.1 × 10 <sup>-6</sup> s <sup>-1</sup> .  (b) A volume of 1.0 m <sup>3</sup> of atmospheric air contains 2.5 × 10 <sup>25</sup> molecules.  Calculate the ratio  number of air molecules in 1.0 m <sup>3</sup> of atmospheric air number of radon-222 atoms in 1.0 m <sup>3</sup> of atmospheric air for the minimum activity of radon-222 at which safety measures should be taken.	E
(ii) Show that the decay constant of radon-222 is 2.1 × 10 <sup>-6</sup> s <sup>-1</sup> .  b) A volume of 1.0 m³ of atmospheric air contains 2.5 × 10 <sup>25</sup> molecules.  Calculate the ratio  number of air molecules in 1.0 m³ of atmospheric air number of radon-222 atoms in 1.0 m³ of atmospheric air for the minimum activity of radon-222 at which safety measures should be taken.	<b>?</b> \$
(ii) Show that the decay constant of radon-222 is 2.1 × 10 <sup>-6</sup> s <sup>-1</sup> .  (ii) Show that the decay constant of radon-222 is 2.1 × 10 <sup>-6</sup> s <sup>-1</sup> .  (iii) Show that the decay constant of radon-222 is 2.1 × 10 <sup>-6</sup> s <sup>-1</sup> .	
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(ii) Show that the decay constant of radon-222 is $2.1 \times 10^{-6}  \mathrm{s}^{-1}$ .  2) A volume of $1.0  \mathrm{m}^3$ of atmospheric air contains $2.5 \times 10^{25}$ molecules.  Calculate the ratio $\frac{\mathrm{number\ of\ air\ molecules\ in\ 1.0 m}^3  \mathrm{of\ atmospheric\ air}}{\mathrm{number\ of\ radon-222\ atoms\ in\ 1.0 m}^3  \mathrm{of\ atmospheric\ air}}$ for the minimum activity of radon-222 at which safety measures should be taken.	
A volume of 1.0 m <sup>3</sup> of atmospheric air contains 2.5 × 10 <sup>25</sup> molecules.  Calculate the ratio number of air molecules in 1.0 m <sup>3</sup> of atmospheric airnumber of radon-222 atoms in 1.0 m <sup>3</sup> of atmospheric air  for the minimum activity of radon-222 at which safety measures should be taken.	2]
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number of radon-222 atoms in 1.0 m <sup>3</sup> of atmospheric air for the minimum activity of radon-222 at which safety measures should be taken.	
for the minimum activity of radon-222 at which safety measures should be taken.	
ratio =	
	3]
	3]

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Please turn over for Section B.

#### **Section B**

For Examiner's Use

Answer all the questions in the spaces provided.

9	(a)	The resistance of a light-dependent resistor (LDR) is approximately $500 \Omega$ in daylight.
		Suggest an approximate value for the resistance of the LDR in darkness.

resistance = .....  $\Omega$  [1]

**(b)** An electronic light-meter is used to warn when light intensity becomes low. A light-dependent resistor is connected into the circuit of Fig. 9.1.

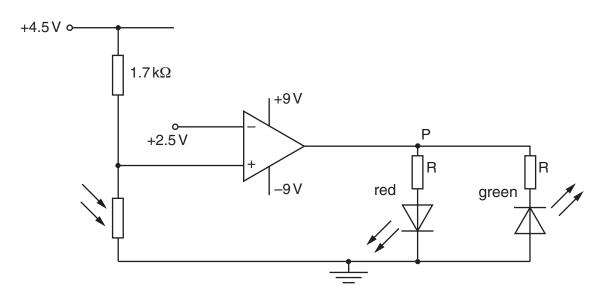


Fig. 9.1

The operational amplifier (op-amp) is ideal.

The resistors R are to ensure that the light-emitting diodes (LEDs) do not over-heat.

(i	) On Fig. 9.1.	, mark the polari	ty of the point	P for the red LED	to be emitting light.	[1]
١-	,	,	.,			L - 1

(ii)	The LDR is in daylight and has a resistance of $500\Omega$ . State and explain which diode, red or green, will be emitting light.	
(iii)	The intensity of the light decreases and the LDR is in darkness. State and explain the effect on the LEDs of this change in intensity.	[O]

10	Explain the principles of the generation and detection of ultrasound waves.					
		Use				
	ro1					

For Examiner's Use

		20			
11	(a)	Distinguish between sharpness and contrast in X-ray imaging.			
		sharpness:			
		contrast:			
		[2]			
	(b)	A student investigates the absorption of X-ray radiation in a model arm. A cross-section of the model arm is shown in Fig. 11.1.			
		4.0 cm muscle bone			
		ВВ			
		M M			
		8.0 cm			
		Fig. 11.1			
		Parallel X-ray beams are directed along the line MM and along the line BB. The linear absorption coefficients of the muscle and of the bone are $0.20\mathrm{cm}^{-1}$ and $12\mathrm{cm}^{-1}$ respectively.			
		Calculate the ratio			
		intensity of emergent X-ray beam from model intensity of incident X-ray beam on model			
		for a parallel X-ray beam directed along the line			
		(i) MM,			
		ratio =[2]			

	(ii)	BB.	For Examiner's Use
		ratio =[3]	
(c)	Sta	te whether your answers in (b) would indicate that the X-ray image	
	(i)	is sharp,	
		[1]	
	(ii)	has good contrast.	

For Examiner's Use

12	In a stat	cellular phone network, a region is divided into a number of cells, each with its own base ion.
	(a)	Suggest and explain two reasons why a region is divided into a number of cells.
		1
		2
		[4]
(b) A passenger in a car is using a mobile phone as the car moves across several of Outline how it is ensured that the phone call is continuous.		
		[4]

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