

**JUNE 2002** 

### **GCE Advanced Level**

# **MARK SCHEME**

**MAXIMUM MARK: 60** 

SYLLABUS/COMPONENT:9702/4

**PHYSICS** (STRUCTURED QUESTIONS (A2 CORE))

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## Categorisation of marks

The marking scheme categorises marks on the MACB scheme.

**B** marks: These are awarded as <u>independent</u> marks, which do not depend on other marks. For a B-mark to be scored, the point to which it refers must be seen specifically in the candidate's answer.

M marks: These are <u>method</u> marks upon which A-marks (accuracy marks) later depend. For an M-mark to be scored, the point to which it refers must be seen in the candidate's answer. If a candidate fails to score a particular M-mark, then none of the dependent A-marks can be scored.

C marks: These are <u>compensatory</u> method marks which can be scored even if the points to which they refer are not written down by the candidate, providing subsequent working gives evidence that they must have known it. For example, if an equation carries a C-mark and the candidate does not write down the actual equation but does correct working which shows he/she knew the equation, then the C-mark is awarded.

A marks: These are accuracy or <u>answer</u> marks which either depend on an M-mark, or allow a C-mark to be scored.

# Conventions within the marking scheme

### BRACKETS

Where brackets are shown in the marking scheme, the candidate is not required to give the brackets information in order to earn the available marks.

### UNDERLINING

In the marking scheme, underlining indicates information that is essential for marks to be awarded.

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1	(a)	$g = GM/R^2$ $M = 9.81 \times (6.38 \times 10^6)^2 / 6.67 \times 10^{-11}$ $= 5.99 \times 10^{24} \text{ kg}$ (allow 2 marks if $g = 9.8 \text{ N kg}^{-1} \text{ used}$ , I mark if $g = 10 \text{ N kg}^{-1} \text{ used}$ )	C1 M1 A0	[2]
	(b)	(i) $T = 24 \text{ hours}$ $\omega = 2\pi/(24 \times 3600) \text{ or } 2\pi/T$ $= 7.27 \times 10^{-5} \text{ rad s}^{-1}$ (ii) $mr\omega^2 = GMm/r^2$ $r^3 = 7.55 \times 10^{22}$ $r = 4.23 \times 10^7 \text{ m}$	C1 C1 A1 C1 C1 A1	[3]
2	(a)	<ul> <li>(i) volume increases on evaporation         so work done pushing back the atmosphere         (ii) E<sub>k</sub> of atoms constant (as no temperature change)         E<sub>p</sub> changes because separation of atoms changes         so internal energy changes because U = E<sub>k</sub> + E<sub>p</sub></li> </ul>	BI BI B1	[5]
	(b)	$\Delta U = \Delta W + \Delta Q$ argument leading to $\Delta Q$ being positive	Mi Al	[2]
3	(a)	(i) mean kinetic energy of the atoms / molecules / particles (ii) at absolute zero, atoms have no kinetic energy	Αl	[3]
	(b)	(i) $pV = nRT$ $n = (1.2 \times 10^5 \times 2.0 \times 10^{-2})/(8.31 \times 310)$ = 0.93  mol (ii) total amount = $(1.20 + 0.93)$ $(1.20 + 0.93) = (4.0 \times 10^{-2} \times p)/(8.31 \times 310)$ $p = 1.37 \times 10^5 \text{ Pa}$	C1 C1 A1 C1 C1 A1	[6]

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4	(a)	• •	acceleration proportional to distance (from fixed point) / displacement Meand directed towards fixed point	1 11	<b>i</b> ]
	(b)		correct period	81 //1 //1 [3	3]
	(c)		1	31 31 [2	2]
5	(a)			M1 A1 [2	2]
	(b)	(ii)	C = Q/V	31 31 31 [3	3]
	(c)	`	energy = $\frac{1}{2}CV^2$ or $\frac{1}{2}QV$ potential = $(2.0 \times 10^{-6}) / (1.67 \times 10^{-11}) = 1.2 \times 10^{5} \text{ V}$	B1 C1 C1	4]
6	(a)	i	no e.m.f. when current constant	B1 B1 B1 [	[3]
	(b)			Cl	[1] [2]
				Bl Bl	[3]

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7	(a)	photoelectric effect	Βl	[1]
	. , .,	reasonable line extrapolated $6.8 \times 10^{14}$ Hz (allow $\pm 0.4 \times 10^{14}$ Hz) attempt at finding gradient working shown to give $6.6 \times 10^{-34}$ J s Hz (allow $\pm 0.4 \times 10^{14}$ Hz)	Al Ml	[4]
	(c)	line: same gradient		[2]
	(d)	maximum corresponds to electron emitted from surface		[2]

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