

Virginia Ambient Air Monitoring 2009 Data Report



Department of Environmental Quality

Commonwealth of Virginia Department of Environmental Quality



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On The Cover

We would like to thank Elise Spangler for her contributions to the front cover.

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Introduction

The 2009 Virginia Ambient Air Monitoring Data Report is a compilation of air pollutant measurements made by the Virginia Department of Environmental Quality, the City of Alexandria, Fairfax County, the U.S. Department of Agriculture Forest Service, and the National Park Service. This report satisfies the requirements of the U.S. Environmental Protection Agency (EPA) for the reporting of air quality data as specified in the Code of Federal Regulations (<http://www.access.gpo.gov/cgi-bin/cfrassemble.cgi?title=200740>), Title 40, Part 58, Appendix F (http://edocket.access.gpo.gov/cfr_2007/julqtr/pdf/40cfr58.61.pdf).

Ambient air quality was measured at 47 locations within the Commonwealth during 2009. These monitoring sites were established in accordance with EPA's siting criteria contained in 40 CFR Part 58, Appendices D and E (http://edocket.access.gpo.gov/cfr_2007/julqtr/pdf/40cfr58.61.pdf), and monitoring network operations conformed to EPA guidance documents and generally accepted air quality monitoring practices. All data reported for these monitoring sites were quality assured in accordance with requirements contained in 40 CFR Part 58, Appendix A (http://edocket.access.gpo.gov/cfr_2007/julqtr/pdf/40cfr58.61.pdf).

Ambient concentrations of carbon monoxide, nitrogen dioxide, and sulfur dioxide were within the EPA's national ambient air quality standards (NAAQS) in 2009. Virginia experienced far fewer exceedances of the ozone pollution standard in 2009 than in any year in the past. In 2009, Northern Virginia had 4 days when an eight-hour ozone average greater than .075 ppm was recorded at one or more monitoring stations in the area. The remainder of the state had no ozone exceedances recorded at any of the monitors. This can largely be attributed to meteorological conditions during the normal ozone recording season, i. e. April through October. This period was cooler and wetter than average which resulted in less ozone formation in the lower atmosphere. This also caused the three year average at 7 stations to fall below NAAQS standard of 0.075.

The Office of Air Quality Monitoring (AQM) added a lead monitor to the network. The monitor was placed at the Cherry Hill Circle (109-H) site in Roanoke. Also AQM began operating a particulate monitor (PM_{2.5}) at Salem High School (110-C). The Hampton monitoring site was moved to Hogan Drive in Newport News (180-O). This relocation was temporary and the site will be moved permanently to a site at the NASA Langley Research Center. Also in 2009, the Fairfax County Board of Supervisors opted to fund the Fairfax County air monitoring program for only one additional year (7/2009 through 7/2010). The program will be ended permanently effective on 7/1/2010. AQM also operated a particulate monitor at the Central Elementary School in Lexington part of the year in support of air permitting efforts in the Valley Regional Office.

We are responsible for seeing that the Virginia ambient air monitoring network is maintained and operated in accordance with state and federal guidelines. Personnel from DEQ regional offices, the City of Alexandria, Fairfax County Health Department, the National Park Service, and the U.S. Department of Agriculture Forest Service conduct the daily operations at these sites. One of our primary jobs is to support these people in their monitoring efforts. This is done by:

- calibrating air monitoring instrumentation and associated support equipment on a set schedule
- auditing the instrumentation to insure that it is operating within set standards
- troubleshooting instrumentation problems reported by the operators
- supplying field operators with necessary items so they can perform their job properly
- repairing malfunctioning sampling instrumentation and ancillary equipment

Other functions:

- respond to regional and locality requests for special sampling such as emergency response or to answer citizen complaints
- coordinate efforts with the regional offices and localities to determine new air monitoring site locations
- conduct AQM generated special sampling projects to characterize a community's air
- furnish ambient air data to the regional offices, localities, Central Office, EPA and the EPA database
- answer FOIA requests for ambient air sampling data
- work with the regions and the localities to see that area monitoring needs are met
- work with EPA to see that necessary state and federal monitoring needs are met
- support VISTAS (Visibility Improvement State and Tribal Association) and MARAMA (Mid-Atlantic Regional Air Management Association of the Southeast) on routine and special projects

Criteria Pollutant Monitoring:

A portion of the air monitoring network is made up of instruments that sample for the Criteria Pollutants. Sulfur dioxide, nitrogen dioxide, carbon monoxide, ozone and particulate matter (PM₁₀ & PM_{2.5}) can injure health, harm the environment and cause property damage. EPA calls these pollutants criteria air pollutants because they have regulated them by first developing health-based criteria (science-based guidelines) as the basis for setting permissible limits. One set of limits (primary standard) protects health; another set of limits (secondary standard) is intended to prevent environmental and property damage

Special Monitoring:

In addition to overseeing the air sampling network for criteria pollutants, AQM conducts routine and short term sampling for VOCs (volatile organic compounds), Carbonyls, Toxic Metals and NO_y (total reactive nitrogen). Sampled VOCs are made up of 39 HAPs (Hazardous Air Pollutants) and 56 Hydrocarbon Ozone Precursors.

1. What is the Clean Air Act?

The Clean Air Act is a federal law that provides for the protection of human health and the environment. The original Clean Air Act was passed in 1963, and the 1970 version of the law resulted in the creation of the U.S. Environmental Protection Agency (EPA), which was charged with setting and enforcing ambient air quality standards. The law was amended in 1977, and most recently in 1990. Most of the activities of the Virginia Department of Environmental Quality's Air Division come from mandates of the Clean Air Act, and are overseen by the EPA. More information on the 1990 amendments to the Clean Air Act can be found at: <http://www.epa.gov/air/caa/>.

2. What is a criteria air pollutant?

The Clean Air Act names six air pollutants that are commonly found in the air throughout the United States, and that can injure humans by causing respiratory and cardiovascular problems, and harm the environment by impairing visibility, and causing damage to animals, crops, vegetation and buildings. EPA has developed health-based criteria for these pollutants through scientific studies, and has established regulations setting permissible levels of these pollutants in the air. The "criteria" pollutants are: carbon monoxide, sulfur dioxide, nitrogen dioxide, ozone, particulate matter, and lead, and the limits that have been set for them are the National Ambient Air Quality Standards (NAAQS).

3. What is the difference between a primary and secondary National Ambient Air Quality Standard?

The National Ambient Air Quality Standards are divided into two types. The first type, the primary standard, is designed to protect human health, especially those who are most vulnerable such as children and the elderly, and people suffering from asthma, emphysema, chronic bronchitis, and heart ailments. The second type, the secondary standard, is designed to prevent damage to property and the environment. For a list of the primary and secondary National Ambient Air Quality Standards, see <http://www.epa.gov/air/criteria.html> or page 67 of this report.

4. How is the location of an air monitoring station decided?

Generally, the deciding factor in all Virginia air monitoring sampling is to determine where the highest pollutant concentrations will occur, and place the sampler as near as possible to that location. A wind rose is typically used to determine the prevailing wind direction for an area and identify the downwind direction from a probable source. A wind rose is a meteorological map showing the frequency and strength of winds from different directions at a specific location.

For typical criteria pollutant monitoring, the federal guidelines on siting an air monitor for measuring maximum concentrations are followed. These guidelines not only encourage siting in areas with free airflow and a minimum amount of obstructions, but they also give the height requirements for the sample inlet and the desired separation distances from obstructions such as tree lines, localized sources such as oil furnace flues, and other influences that can skew the data.

Other determining factors for placing air monitoring stations include:

- ❖ security of the site
- ❖ safety of the operator
- ❖ availability of electric power and communication service
- ❖ accessibility of the site

For more specific information, consult EPA's Quality Assurance Handbook for Air Pollution Measurement Systems, Volume II, Section 6, <http://www.epa.gov/ttn/amtic/qalist.html>

5. **How large of an area does an air monitoring station represent?**

The sampling area of a monitoring site is dependent on the parameters selected for representation, such as:

- type of pollutants being sampled
- rural vs. urban sampling
- source oriented, population oriented, or background oriented
- sampling for pollution transported from outside the Commonwealth

Many sites are also dependant on topography and meteorology of an area, which play an important role. Federal guidelines spell out the general area of representation. Some examples of varied air sampling sites are:

- A background research site in central Virginia may represent an area with a radius of 50 to 100 kilometers.
- An ozone or fine particulate site in the Shenandoah Valley may represent an elongated area with an axis running with the valley and is a hundred kilometers long but only twenty-five kilometers wide.
- A carbon monoxide sampling site in an urban street canyon setting may represent an area of only a few blocks in radius.
- A source oriented site in south central Virginia may represent an area from 0.5 to 4 kilometers in radius.

For more specific information, consult EPA's Quality Assurance Handbook for Air Pollution Measurement Systems, Volume II, Section 6, <http://www.epa.gov/ttn/amtic/qalist.html>

6. **What is a “nonattainment” area?**

A nonattainment area is a geographic area that has been determined by EPA as not meeting the air quality standards for one or more pollutants. Typically, an area is declared nonattainment based on data collected at one or more ambient air monitoring sites within the area. However, sometimes the nonattainment designation can be made based on the use of air quality models that use monitoring data from other areas. In Virginia, nonattainment areas are designated for two of the criteria pollutants, ozone and fine particulate matter (PM_{2.5}).

7. **How can I find out if I live in a nonattainment area?**

A list of nonattainment areas in Virginia can be found in this report on page 71. On the web, EPA has a comprehensive list of all nonattainment areas in the country at <http://www.epa.gov/air/oagps/greenbk/>.

8. What are the impacts of nonattainment designation?

To demonstrate how they plan to achieve federal air quality standards, states must draft a "State Implementation Plan," or SIP. This plan lists specific actions that the state will undertake to improve and maintain acceptable air quality, and a time frame for accomplishing these goals. The SIP may require new factories to install the newest and most effective air pollution control technologies. Other actions could be requiring older factories to retrofit their smokestacks with better pollution control devices, requiring an area to sell only reformulated gasoline during the summer months, requiring vapor recovery systems on gasoline pumps, and requiring vehicle exhaust emission checks, to name a few. SIP development is a lengthy process, and involves negotiation between the state and the EPA until it is finalized.

9. What is a Maintenance Area?

A maintenance area is an area that has been formally designated nonattainment, but is now recognized by EPA as now meeting the NAAQS. A maintenance area must have an approved "maintenance plan" to meet and maintain air quality standards.

10. What is an Early Action Compact (EAC) area?

In April 2004, EPA published a final rule listing areas in the country designated as not attaining the 8-hour ozone ambient air quality standard. A few areas, including two in Virginia, Roanoke and Frederick County/Winchester, entered into Early Action Compacts (EAC) with EPA before the nonattainment designations were published, because they were facing the possibility of being designated nonattainment for ozone. The compacts allowed the participating areas to come up with their own plan for meeting the 8-hour ozone standard, provided they meet certain milestones and they attain the 8-hour ozone standard no later than December 31, 2007. As part of the EAC, EPA agreed to defer the nonattainment designation. The two EAC areas in Virginia were designated attainment as of April 15, 2008.

www.epa.gov/air/oaqps/greenbk/encs2.html#virginia

11. How can I get current or historical air quality data?

Current ozone data for Virginia, as well as current AQI and air quality forecasts can be obtained at www.deq.virginia.gov/airquality/homepage.html. Summary air quality data for ozone and PM2.5 can be found on the DEQ website at

www.deq.virginia.gov/airquality/homepage.html and

www.deq.virginia.gov/airmon/pm25home.html.

Annual monitoring data reports for DEQ from 2001 to the present can be found at

<http://www.deq.virginia.gov/airmon/publications.html>. EPA provides monitoring and emissions data, as well as maps, on the web at <http://www.epa.gov/air/data/index.html> and

www.epa.gov/airexplorer/index.htm. Detailed data for monitoring sites in Virginia can also be obtained by contacting the VA DEQ Office of Air Quality Monitoring.

12. What do I do if I have a complaint about air quality in my neighborhood?

Contact the DEQ regional office in your area. To see a list of regional offices and phone numbers, see page 58 of this report, or visit www.deq.virginia.gov/prep.

13. Who can I call about an indoor air quality problem, such as mold or radon gas?

Your local health department may be able to assist you with some indoor air quality problems. See www.vdh.state.va.us/lhd for the health department office in your area. Other excellent sources of information on indoor air quality can be found on EPA's website at www.epa.gov/iaq/index.html and through the American Lung Association website at www.lungusa.org.

Criteria Pollutants

PM_{2.5} is particulate matter (PM) that is less than or equal to 2.5 micrometers (a micrometer is one millionth of a meter) in aerodynamic diameter. These particles are often called “fine particles” because of their small size. Fine particles originate from a variety of man-made stationary and mobile sources, such as factory smoke stacks and diesel engines, as well as from natural sources, such as forest fires. These particles may be emitted directly into the air, or they may be formed by chemical reaction in the atmosphere from gaseous emissions of sulfur dioxide (SO₂), nitrogen oxides (NO_x), and volatile organic compounds (VOCs).

Scientific research has linked fine particle pollution to human health problems. The particles are easily inhaled deep into the lungs, and can actually enter the bloodstream. Particle pollution is of particular concern to people with heart or lung disease, such as coronary artery disease, congestive heart failure, asthma, or chronic obstructive pulmonary disease (COPD). Older adults are at risk because they may have underlying, undiagnosed heart or lung problems. Young children are also at risk because their lungs are still developing, they are more likely to have asthma or acute respiratory disease, and they tend to spend longer periods of time at high activity levels, causing them to inhale more particles than someone at rest. Even otherwise healthy people may suffer short-term symptoms such as eye, nose, throat irritation, coughing, and shortness of breath during episodes of high particulate levels.

PM_{2.5} air quality standards were implemented by EPA in 1997 to protect against the health effects of fine particle pollution. In September 2006, EPA announced revisions to the National Ambient Air Quality Standards (NAAQS) for particulate matter. While the long-term PM_{2.5} annual average standard of 15.0 µg/m³ remained the same, the short-term 24-hour average PM_{2.5} standard was significantly reduced from 65 µg/m³ to 35 µg/m³. This was done to better protect public health, based on a large body of scientific evidence which supported the stricter limits. For more information on the revisions to the particulate matter standards, see www.epa.gov/air/particlepollution/pdfs/20060921_factsheet.pdf.

In addition to health problems, fine particle pollution contributes to haze that causes deterioration of visibility in scenic areas, and also deposits harmful compounds on the soil and water. Unlike ozone, which is a seasonal pollutant in most areas of the country, particle pollution can occur year-round, and is monitored throughout the year in Virginia. The Virginia DEQ PM_{2.5} monitoring network uses three different types of samplers to monitor fine particulate in the state:

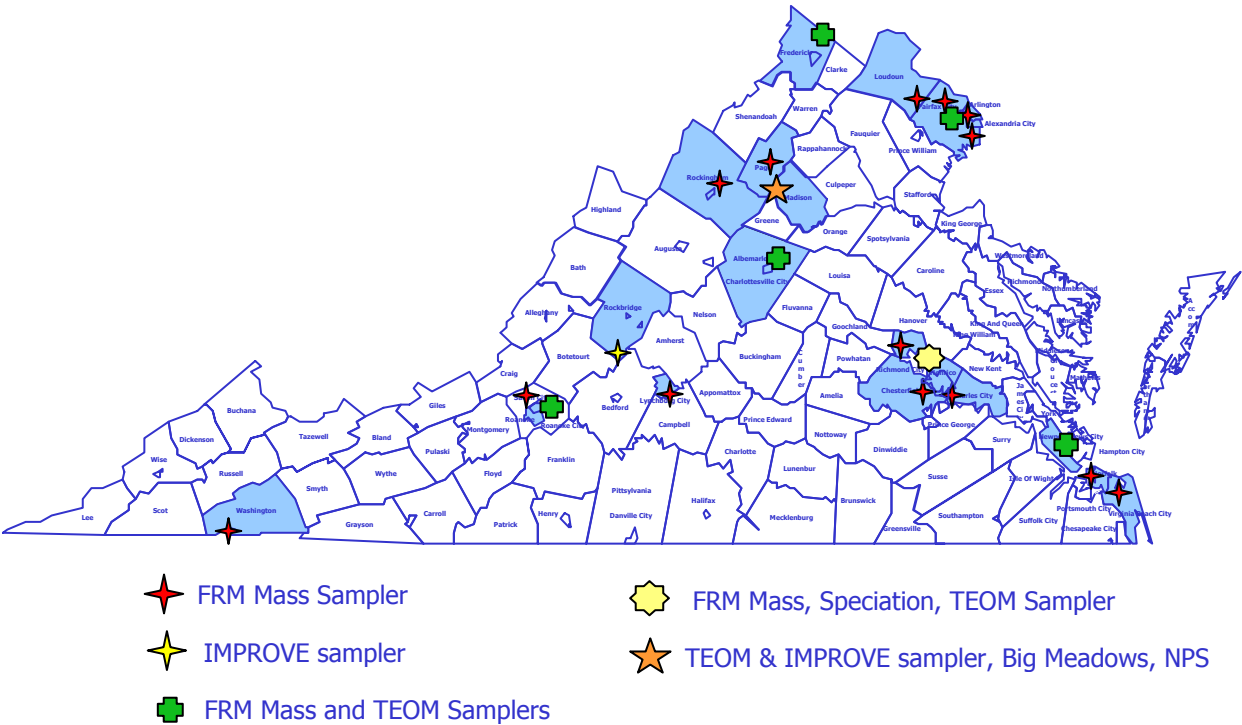
PM_{2.5} 24-hour Mass Sampler: This Federal Reference Method (FRM) sampler collects particulate matter on a stretched Teflon filter media. Four samplers (Henrico Co., Roanoke, Virginia Beach, and Fairfax Co.) collect 24-hour samples every day. The rest of these samplers collect 24-hour samples on a one-in-three day schedule. Filters are retrieved from the field and shipped via courier to the Virginia Division of Consolidated Laboratory Services (DCLS) in Richmond. At the laboratory, the filters are equilibrated for a minimum of 24 hours prior to the final weighing.

PM_{2.5} 24-hour Speciation Sampler: This sampler collects particulate matter on nylon, Teflon, and quartz filters in three sampling modules. These modules are picked up by the operator after the sampling period, and shipped refrigerated to the EPA contract laboratory. The lab analyzes the filters for mass loading, trace elements (such as aluminum, antimony, arsenic, barium, bromine, and zirconium), cations (ammonium, potassium, sodium), anions (nitrate, sulfate), and carbons (carbonate carbon, elemental carbon, and organic carbon). The speciation network in Virginia consists of one monitor, located in Henrico Co., and this sampler operates on a one in three day sampling schedule.

PM_{2.5} Continuous Monitor: This sampler collects particulate samples on a continuous basis, and data are compiled into hourly averages. The sampler utilizes a Tapered Element Oscillating Microbalance (TEOM) in the sampling design. TEOM samplers are operated in Hampton Roads, Henrico Co., Roanoke, Fairfax Co., and Big Meadows in Shenandoah National Park, Frederick Co., and Albemarle Co.

Each type of PM_{2.5} sampler has a unique function. The FRM samplers collect data that are used to determine if the state is complying with the national ambient air quality standards (NAAQS) for particulate matter. The speciation sampler collects data about the composition of particulate matter in Virginia, and is useful for identifying potential sources of air pollution both within and outside the state boundaries. The FRM and speciation monitors are manual, filter-based methods, and the samples they collect must be transported to a laboratory for processing. Consequently, they are not useful for reporting real-time air quality conditions. The TEOM is a continuous particulate monitor that provides hourly data on fine particulate levels. The data are polled each hour by a central computer at DEQ, and then used to compute the current air quality index, which is posted on the agency website at www.deq.virginia.gov/airquality/homepage.html. The data are also simultaneously sent to EPA's national air quality website at www.airnow.gov.

In addition to the PM_{2.5} network operated by the DEQ, the National Park Service and the USDA Forest Service operate PM_{2.5} samplers at Big Meadows in Shenandoah National Park, and in Rockbridge Co. as part of the IMPROVE (Interagency Monitoring of Protected Visual Environments) network. This network employs different sampling methods than those used by the DEQ. Data for the IMPROVE network can be found on the internet at <http://vista.cira.colostate.edu/improve>.



PM2.5 Monitoring Network

NAAQS Standards

Primary Standard for PM_{2.5}:

- Annual Arithmetic Mean – the 3 year average of the weighted annual mean PM_{2.5} concentration must not exceed 15.0 µg/m³.
- 24-Hour concentration – the 3 year average of the 98th percentile of 24-hour concentrations must not exceed 35 µg/m³.

Secondary Standard for PM_{2.5}:

- Same as Primary.

| 2007-2009 PM_{2.5} 24-hour Averages, 98th Percentile Values (µg/m³) | | | | |
|--|-------------|-------------|-------------|--|
| Site | 2007 | 2008 | 2009 | 3-Year Average (NAAQ = 35 µg/m³) |
| (101-E) Bristol | 28.0 | 25.9 | 19.7 | 25 |
| (29-D) Page Co. | 30.0 | 25.0 | 20.8 | 25 |
| (109-M) Roanoke | 31.9 | 27.3** | 20.5 | 27 |
| (26-F) Rockingham Co. | 32.0 | 24.7 | 21.7 | 26 |
| (155-Q) Lynchburg | 30.5 | 25.4 | 18.5 | 25 |
| (71-D) Chesterfield Co. | 30.7 | 22.8 | 19.0 | 24 |
| (72-M) Henrico Co. | 32.0 | 25.3** | 20.8 | 26 |
| (72-N) Henrico Co. | 30.4 | 21.9 | 20.2 | 24 |
| (75-B) Charles City Co. | 30.5 | 22.1 | 18.5 | 24 |
| (181-A1) Norfolk | 27.3 | 30.3 | 17.7 | 25 |
| (184-J) Va. Beach | 28.7 | 38.6** | 20.1 | 29 |
| (38-I) Loudoun Co. | 27.7 | 27.5** | 20.0 | 25 |
| (47-T) Arlington Co. | 29.5 | 23.4 | 23.2 | 27 |
| (46-B9) Franconia, Fairfax Co. | 31.9 | 28.4 | 24.2 | 28 |
| (L-46-A8) McLean, Fairfax Co. | 30.9 | 25.6 | 21.2 | 26 |
| (L-46-C1) Annandale, Fairfax Co. | 29.5 | 22.7* | 20.8 | 24 |

* Annual value did not meet completeness criteria.

** In 2009, VA DEQ submitted documentation to EPA requesting exclusion of high values that resulted from large wildfires in the Dismal Swamp and eastern North Carolina during the summer of 2008, under the Exceptional Events rule, 40 CFR, 50.14. As of August 2010, the EPA has agreed to exclude high values from the monitor in Norfolk (181-A1), but has yet to rule on data affected by the same event from the following sites: Va. Beach (184-J), Hampton (179-C, which was discontinued in April 2009), Henrico Co. (72-M), Roanoke (109-M), and Loudoun Co. (38-I).

NAAQS Standards

Primary Standard for PM_{2.5}:

- Annual Arithmetic Mean – the 3 year average of the weighted annual mean PM_{2.5} concentration must not exceed 15.0 µg/m³.
- 24-Hour concentration – the 3 year average of the 98th percentile of 24-hour concentrations must not exceed 35 µg/m³.

Secondary Standard for PM_{2.5}:

- Same as Primary.

| 2007-2009 PM_{2.5} Weighted Annual Arithmetic Means (µg/m³) | | | | |
|---|-------------|-------------|-------------|--|
| Site | 2007 | 2008 | 2009 | 3-Year Average (NAAQ = 15 µg/m³) |
| (101-E) Bristol | 13.9 | 10.6 | 9.2 | 11.2 |
| (29-D) Page Co. | 12.5 | 10.5 | 8.8 | 10.6 |
| (109-M) Roanoke | 14.2 | 12.0 | 9.4 | 11.5 |
| (26-F) Rockingham Co. | 13.7 | 11.5 | 9.8 | 11.7 |
| (155-Q) Lynchburg | 13.1 | 10.0 | 8.4 | 10.5 |
| (71-D) Chesterfield Co. | 13.0 | 11.3 | 9.2 | 11.2 |
| (72-M) Henrico Co. | 12.4 | 10.7 | 9.3 | 10.8 |
| (72-N) Henrico Co. | 12.4 | 10.5 | 8.8 | 10.6 |
| (75-B) Charles City Co. | 11.9 | 10.5 | 8.6 | 10.3 |
| (181-A1) Norfolk | 11.4 | 11.4 | 9.4 | 11.5 |
| (184-J) Va. Beach | 11.2 | 11.8 | 9.2 | 10.7 |
| (38-I) Loudoun Co. | 12.8 | 11.5 | 9.2 | 11.2 |
| (47-T) Arlington Co. | 13.8 | 12.0 | 10.1 | 11.9 |
| (46-B9) Franconia, Fairfax Co. | 12.5 | 11.1 | 9.8 | 11.1 |
| (L-46-A8) McLean, Fairfax Co. | 13.5 | 11.8 | 9.7 | 11.7 |
| (L-46-C1) Annandale, Fairfax Co. | 13.3 | 11.2* | 9.5 | 11.3 |

* Annual value did not meet completeness criteria.

3-Day Monitoring Schedule for PM2.5 2009

| January | | | | | | |
|---------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| | | | | 1 | 2 | 3 |
| 4 | 5 | 6 | 7 | 8 | 9 | 10 |
| 11 | 12 | 13 | 14 | 15 | 16 | 17 |
| 18 | 19 | 20 | 21 | 22 | 23 | 24 |
| 25 | 26 | 27 | 28 | 29 | 30 | 31 |

| February | | | | | | |
|----------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| 1 | 2 | 3 | 4 | 5 | 6 | 7 |
| 8 | 9 | 10 | 11 | 12 | 13 | 14 |
| 15 | 16 | 17 | 18 | 19 | 20 | 21 |
| 22 | 23 | 24 | 25 | 26 | 27 | 28 |
| | | | | | | |

| March | | | | | | |
|-------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| 1 | 2 | 3 | 4 | 5 | 6 | 7 |
| 8 | 9 | 10 | 11 | 12 | 13 | 14 |
| 15 | 16 | 17 | 18 | 19 | 20 | 21 |
| 22 | 23 | 24 | 25 | 26 | 27 | 28 |
| 29 | 30 | 31 | | | | |

| April | | | | | | |
|-------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| | | | 1 | 2 | 3 | 4 |
| 5 | 6 | 7 | 8 | 9 | 10 | 11 |
| 12 | 13 | 14 | 15 | 16 | 17 | 18 |
| 19 | 20 | 21 | 22 | 23 | 24 | 25 |
| 26 | 27 | 28 | 29 | 30 | | |

| May | | | | | | |
|-----|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| | | | | | 1 | 2 |
| 3 | 4 | 5 | 6 | 7 | 8 | 9 |
| 10 | 11 | 12 | 13 | 14 | 15 | 16 |
| 17 | 18 | 19 | 20 | 21 | 22 | 23 |
| 24 | 25 | 26 | 27 | 28 | 29 | 30 |
| 31 | | | | | | |

| June | | | | | | |
|------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| | 1 | 2 | 3 | 4 | 5 | 6 |
| 7 | 8 | 9 | 10 | 11 | 12 | 13 |
| 14 | 15 | 16 | 17 | 18 | 19 | 20 |
| 21 | 22 | 23 | 24 | 25 | 26 | 27 |
| 28 | 29 | 30 | | | | |

| July | | | | | | |
|------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| | | | 1 | 2 | 3 | 4 |
| 5 | 6 | 7 | 8 | 9 | 10 | 11 |
| 12 | 13 | 14 | 15 | 16 | 17 | 18 |
| 19 | 20 | 21 | 22 | 23 | 24 | 25 |
| 26 | 27 | 28 | 29 | 30 | 31 | |

| August | | | | | | |
|--------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| | | | | | | 1 |
| 2 | 3 | 4 | 5 | 6 | 7 | 8 |
| 9 | 10 | 11 | 12 | 13 | 14 | 15 |
| 16 | 17 | 18 | 19 | 20 | 21 | 22 |
| 23 | 24 | 25 | 26 | 27 | 28 | 29 |
| 30 | 31 | | | | | |

| September | | | | | | |
|-----------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| | | 1 | 2 | 3 | 4 | 5 |
| 6 | 7 | 8 | 9 | 10 | 11 | 12 |
| 13 | 14 | 15 | 16 | 17 | 18 | 19 |
| 20 | 21 | 22 | 23 | 24 | 25 | 26 |
| 27 | 28 | 29 | 30 | | | |

| October | | | | | | |
|---------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| | | | | 1 | 2 | 3 |
| 4 | 5 | 6 | 7 | 8 | 9 | 10 |
| 11 | 12 | 13 | 14 | 15 | 16 | 17 |
| 18 | 19 | 20 | 21 | 22 | 23 | 24 |
| 25 | 26 | 27 | 28 | 29 | 30 | 31 |

| November | | | | | | |
|----------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| 1 | 2 | 3 | 4 | 5 | 6 | 7 |
| 8 | 9 | 10 | 11 | 12 | 13 | 14 |
| 15 | 16 | 17 | 18 | 19 | 20 | 21 |
| 22 | 23 | 24 | 25 | 26 | 27 | 28 |
| 29 | 30 | | | | | |

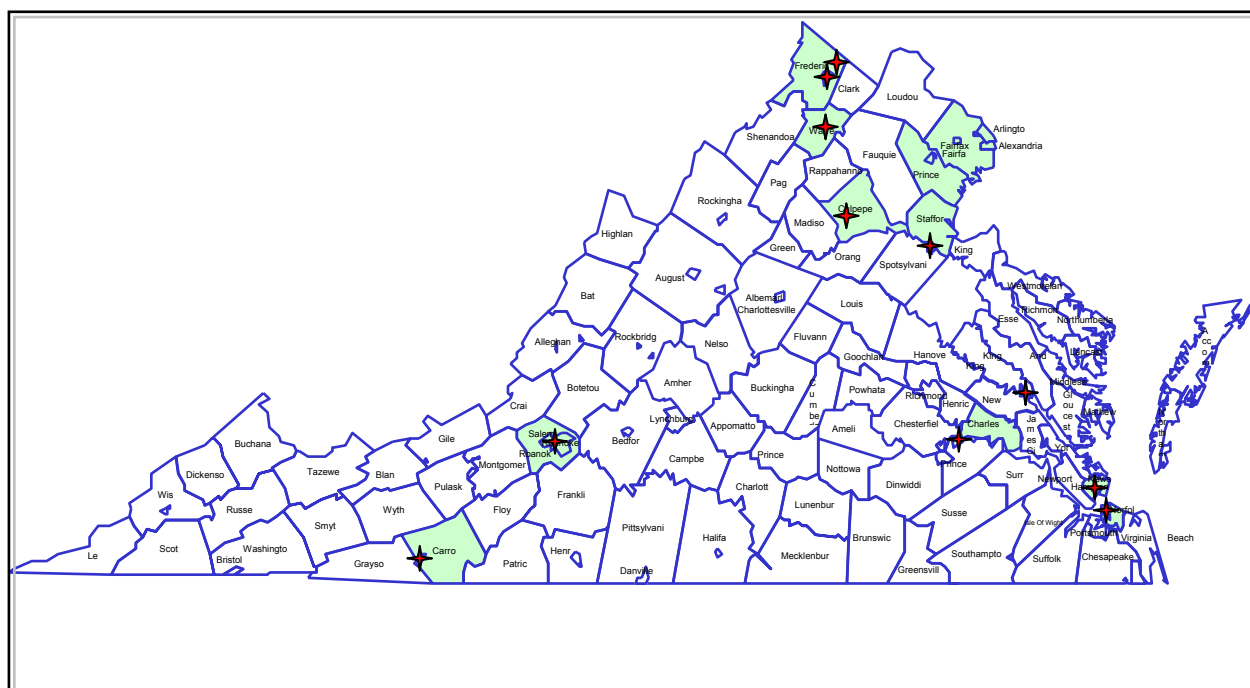
| December | | | | | | |
|----------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| | | 1 | 2 | 3 | 4 | 5 |
| 6 | 7 | 8 | 9 | 10 | 11 | 12 |
| 13 | 14 | 15 | 16 | 17 | 18 | 19 |
| 20 | 21 | 22 | 23 | 24 | 25 | 26 |
| 27 | 28 | 29 | 30 | 31 | | |

PM₁₀ is particulate matter comprised of solid particles or liquid droplets with an aerodynamic diameter of less than or equal to 10 micrometers, and is sometimes referred to as “coarse particles.” PM₁₀ particles are larger than PM_{2.5}, but are still in a size range that can pose health problems because they can be inhaled, and retained in the human respiratory system, causing breathing difficulties, and eye, nose and throat irritation. In addition to the health effects of PM₁₀, these particles can impair visibility, can contribute to climate change, and result in “acidic dry deposition.” Acidic dry deposition occurs when particles containing acidic compounds fall to the ground. The acidic particles can corrode surfaces that they settle on, and can increase the acidity of the soil and water.

The National Ambient Air Quality Standards, or NAAQS, for particulate matter were revised in September 2006. EPA changed the existing standards for PM₁₀ by revoking the annual standard of 50 micrograms per cubic meter, because current scientific evidence did not support a link between long-term exposure to coarse particles and health problems. However, the 24-hour PM₁₀ standard was retained to protect citizens from effects of short-term exposures. For additional information on the revised particulate matter standards, see www.epa.gov/air/particlepollution/pdfs/20060921_factsheet.pdf.

To measure PM₁₀, ambient air is drawn into a sampler that uses a particle size discrimination inlet. The inlet is designed so that particles in the size range of 10 micrometers (also called microns) or below stay suspended in the air stream, while larger particles settle out. The sample air flows across an 8 x 10 inch micro-quartz filter at a rate of 40 cubic feet per minute for a 24-hour period. The particles are captured on the filter, which is weighed before and after sampling, and the PM₁₀ concentration is determined by dividing the change in filter mass by the volume of sampled air. The resulting PM₁₀ concentration is reported as micrograms per cubic meter (µg/m³). The filters are processed at the DEQ Office of Air Quality Monitoring. The normal sampling schedule is once every sixth day from midnight to midnight.

PM₁₀ Monitoring Sites



PM10 Monitoring Sites

NAAQS Standards

Primary Standard for PM₁₀:

- ➡ 24-Hour concentration not to exceed 150 $\mu\text{g}/\text{m}^3$ more than once per year averaged over three years.

Secondary Standard for PM₁₀:

- ➡ Same as Primary.

| 2007-2009 PM ₁₀ 24-Hour Average Concentrations (units in $\mu\text{g}/\text{m}^3$) | | | | | | | |
|--|---------------------|---------------------|---------------------|---------------------|---------------------|---------------------|----------------------------------|
| Site | 2007 | | 2008 | | 2009 | | >150 $\mu\text{g}/\text{m}^3$ |
| | 1 st Max | 2 nd Max | 1 st Max | 2 nd Max | 1 st Max | 2 nd Max | |
| (23-A) Carroll Co. | 39 | 37 | 29 | 28 | 40 | 30 | 0 |
| (30-E) Warren Co. | 46 | 32 | 38 | 36 | 31 | 24 | 0 |
| (134-C) Winchester | 51 | 45 | 36 | 36 | 29 | 28 | 0 |
| (109-H) Roanoke | 75 | 58 | 68 | 61 | 65 | 63 | 0 |
| (154-M) Hopewell | 42 | 35 | 36 | 34 | 35 | 27 | 0 |
| (82-C) King William Co. | 41 | 36 | 36 | 35 | 35 | 27 | 0 |
| (181-A1) Norfolk | 39 | 36 | 88 | 69 | 45 | 35 | 0 |
| (42-B) Culpeper Co. | 50 | 36 | 39 | 33 | 27 | 26 | 0 |
| (130-E) Fredericksburg | 50 | 39 | 40 | 39 | 30 | 28 | 0 |

6-Day Monitoring Schedule for PM10 2009

| January | | | | | | |
|---------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| | | | | 1 | 2 | 3 |
| 4 | 5 | 6 | 7 | 8 | 9 | 10 |
| 11 | 12 | 13 | 14 | 15 | 16 | 17 |
| 18 | 19 | 20 | 21 | 22 | 23 | 24 |
| 25 | 26 | 27 | 28 | 29 | 30 | 31 |

| February | | | | | | |
|----------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| 1 | 2 | 3 | 4 | 5 | 6 | 7 |
| 8 | 9 | 10 | 11 | 12 | 13 | 14 |
| 15 | 16 | 17 | 18 | 19 | 20 | 21 |
| 22 | 23 | 24 | 25 | 26 | 27 | 28 |
| | | | | | | |

| March | | | | | | |
|-------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| 1 | 2 | 3 | 4 | 5 | 6 | 7 |
| 8 | 9 | 10 | 11 | 12 | 13 | 14 |
| 15 | 16 | 17 | 18 | 19 | 20 | 21 |
| 22 | 23 | 24 | 25 | 26 | 27 | 28 |
| 29 | 30 | 31 | | | | |

| April | | | | | | |
|-------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| | | | 1 | 2 | 3 | 4 |
| 5 | 6 | 7 | 8 | 9 | 10 | 11 |
| 12 | 13 | 14 | 15 | 16 | 17 | 18 |
| 19 | 20 | 21 | 22 | 23 | 24 | 25 |
| 26 | 27 | 28 | 29 | 30 | | |

| May | | | | | | |
|-----|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| | | | | | 1 | 2 |
| 3 | 4 | 5 | 6 | 7 | 8 | 9 |
| 10 | 11 | 12 | 13 | 14 | 15 | 16 |
| 17 | 18 | 19 | 20 | 21 | 22 | 23 |
| 24 | 25 | 26 | 27 | 28 | 29 | 30 |
| 31 | | | | | | |

| June | | | | | | |
|------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| | 1 | 2 | 3 | 4 | 5 | 6 |
| 7 | 8 | 9 | 10 | 11 | 12 | 13 |
| 14 | 15 | 16 | 17 | 18 | 19 | 20 |
| 21 | 22 | 23 | 24 | 25 | 26 | 27 |
| 28 | 29 | 30 | | | | |

| July | | | | | | |
|------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| | | | 1 | 2 | 3 | 4 |
| 5 | 6 | 7 | 8 | 9 | 10 | 11 |
| 12 | 13 | 14 | 15 | 16 | 17 | 18 |
| 19 | 20 | 21 | 22 | 23 | 24 | 25 |
| 26 | 27 | 28 | 29 | 30 | 31 | |

| August | | | | | | |
|--------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| | | | | | | 1 |
| 2 | 3 | 4 | 5 | 6 | 7 | 8 |
| 9 | 10 | 11 | 12 | 13 | 14 | 15 |
| 16 | 17 | 18 | 19 | 20 | 21 | 22 |
| 23 | 24 | 25 | 26 | 27 | 28 | 29 |
| 30 | 31 | | | | | |

| September | | | | | | |
|-----------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| | | 1 | 2 | 3 | 4 | 5 |
| 6 | 7 | 8 | 9 | 10 | 11 | 12 |
| 13 | 14 | 15 | 16 | 17 | 18 | 19 |
| 20 | 21 | 22 | 23 | 24 | 25 | 26 |
| 27 | 28 | 29 | 30 | | | |

| October | | | | | | |
|---------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| | | | | 1 | 2 | 3 |
| 4 | 5 | 6 | 7 | 8 | 9 | 10 |
| 11 | 12 | 13 | 14 | 15 | 16 | 17 |
| 18 | 19 | 20 | 21 | 22 | 23 | 24 |
| 25 | 26 | 27 | 28 | 29 | 30 | 31 |

| November | | | | | | |
|----------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| 1 | 2 | 3 | 4 | 5 | 6 | 7 |
| 8 | 9 | 10 | 11 | 12 | 13 | 14 |
| 15 | 16 | 17 | 18 | 19 | 20 | 21 |
| 22 | 23 | 24 | 25 | 26 | 27 | 28 |
| 29 | 30 | | | | | |

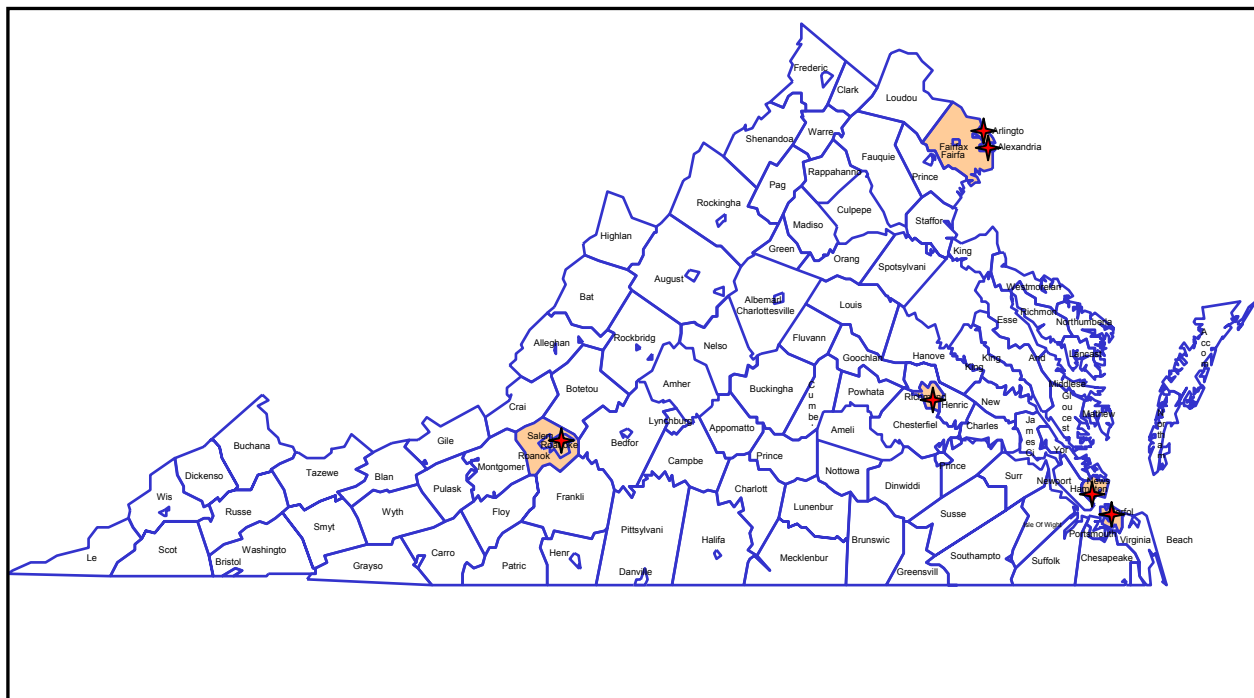
| December | | | | | | |
|----------|----|----|----|----|----|----|
| Su | M | Tu | W | Th | F | Sa |
| | | 1 | 2 | 3 | 4 | 5 |
| 6 | 7 | 8 | 9 | 10 | 11 | 12 |
| 13 | 14 | 15 | 16 | 17 | 18 | 19 |
| 20 | 21 | 22 | 23 | 24 | 25 | 26 |
| 27 | 28 | 29 | 30 | 31 | | |

Carbon monoxide (CO) is a colorless, odorless gas that is produced by incomplete burning of carbon compounds in fossil fuels (gasoline, natural gas, coal, oil, etc.). Over half of the CO emissions in the country come from motor vehicle exhaust. Other sources include construction equipment, boats, lawnmowers, woodstoves, forest fires, and industrial manufacturing processes.

CO concentrations are higher in the vicinity of heavily traveled highways, and drop rapidly the further the distance from the road. Ambient levels of carbon monoxide tend to be higher in the colder months due to "thermal inversions" that trap pollutants close to the ground. A thermal inversion occurs when the temperature of the air next to the ground is colder than air above it. When this happens, the air resists vertical mixing that can help the pollutants to disperse, forming a layer of smog close to the ground.

Carbon monoxide is harmful because it reacts in the bloodstream, reducing the amount of oxygen that is supplied to the heart and brain. CO can be harmful at lower levels to people who suffer from cardiovascular disease, like angina, arteriosclerosis, or congestive heart failure. At high levels, CO can impair brain function, causing vision problems, reduce manual dexterity, and reduce ability to perform complicated tasks. At very high levels, carbon monoxide can be deadly.

Carbon monoxide in the ambient air is measured continuously with an electronic instrument that uses NDIR, "non-dispersive infrared" photometry. The instrument has a pump that continuously draws air through a sample chamber that contains an infrared light source and a detector. Any CO molecules that are present in the sample air absorb some of the infrared light, reducing the intensity of the light reaching the detector. The portion of the infrared light absorbed by the CO molecules is converted into an electrical signal corresponding to the CO concentration, and stored in the instrument computer.



CO Monitoring Network

NAAQS Standards

Primary Standard for CO:

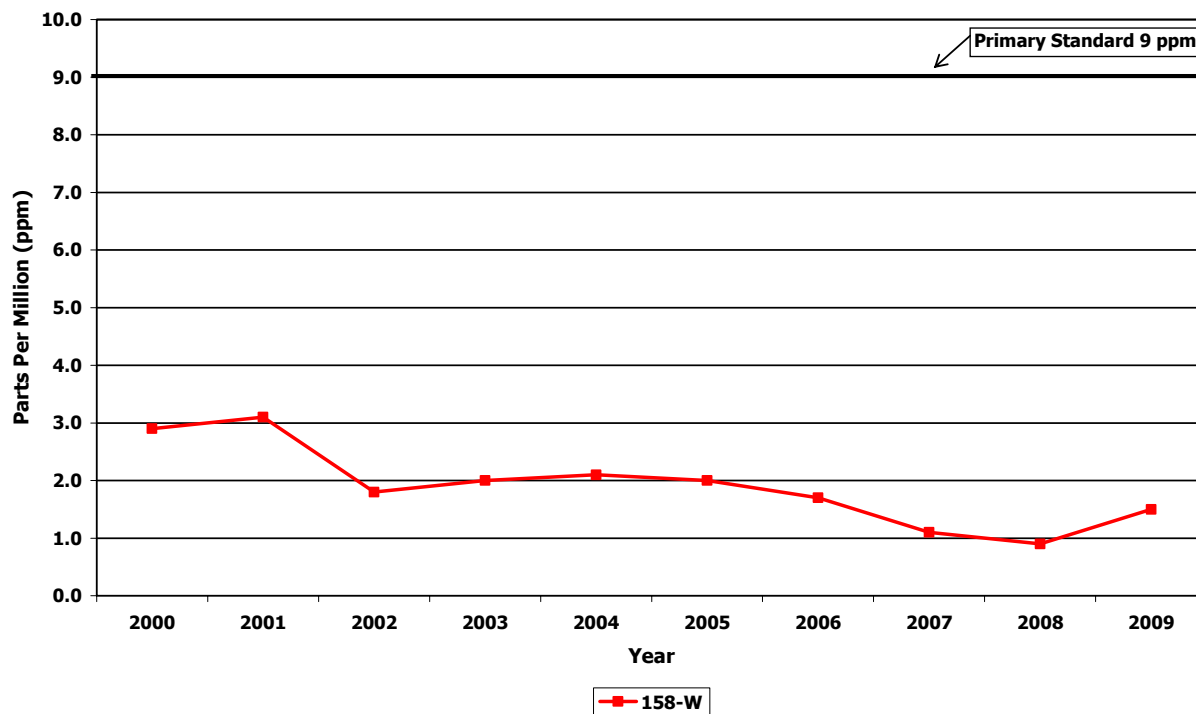
- 👉 8-hour average not to exceed 9 ppm (10 mg/m³) more than once per year.
- 👉 1-hour average not to exceed 35 ppm (40 mg/m³) more than once per year.

There are no Secondary Standards for CO because it does not harm vegetation or buildings.

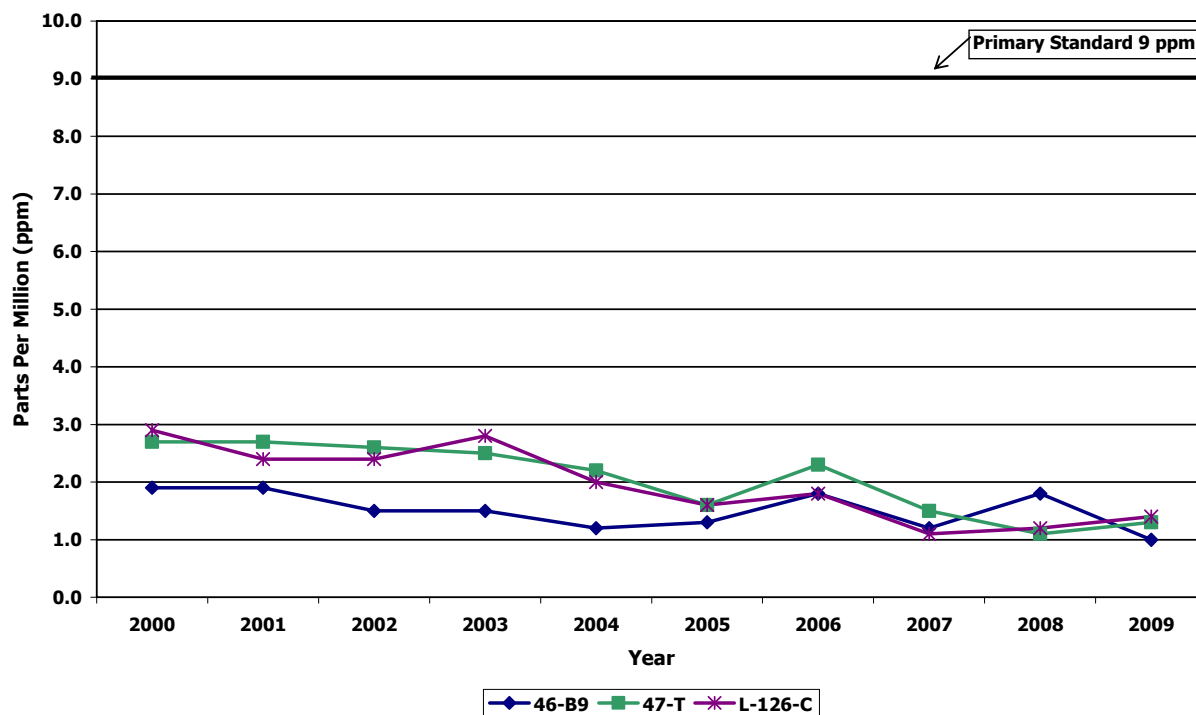
| Site | 2009 | | | |
|-----------------------------|----------------------|----------------------|----------------------|----------------------|
| | 1-Hour Avg. (ppm) | | 8-Hour Avg. (ppm) | |
| | 1 st Max. | 2 nd Max. | 1 st Max. | 2 nd Max. |
| (109-M) Roanoke | 2.5 | 2.4 | 2.0 | 1.7 |
| (158-W) Richmond | 2.3 | 2.1 | 1.7 | 1.5 |
| (181-A1) Norfolk | 1.5 | 1.3 | 0.8 | 0.8 |
| (46-B9) Fairfax Co. | 1.4 | 1.3 | 1.1 | 1.0 |
| (47-T) Arlington Co. | 1.7 | 1.7 | 1.6 | 1.3 |
| (L-126-C) Alexandria | 1.8 | 1.7 | 1.4 | 1.4 |

* Eight Hour Averages stated as Ending Hour

Carbon Monoxide - Piedmont Region Eight Hour 2nd Maximum



Carbon Monoxide - Northern Region Eight Hour 2nd Maximum

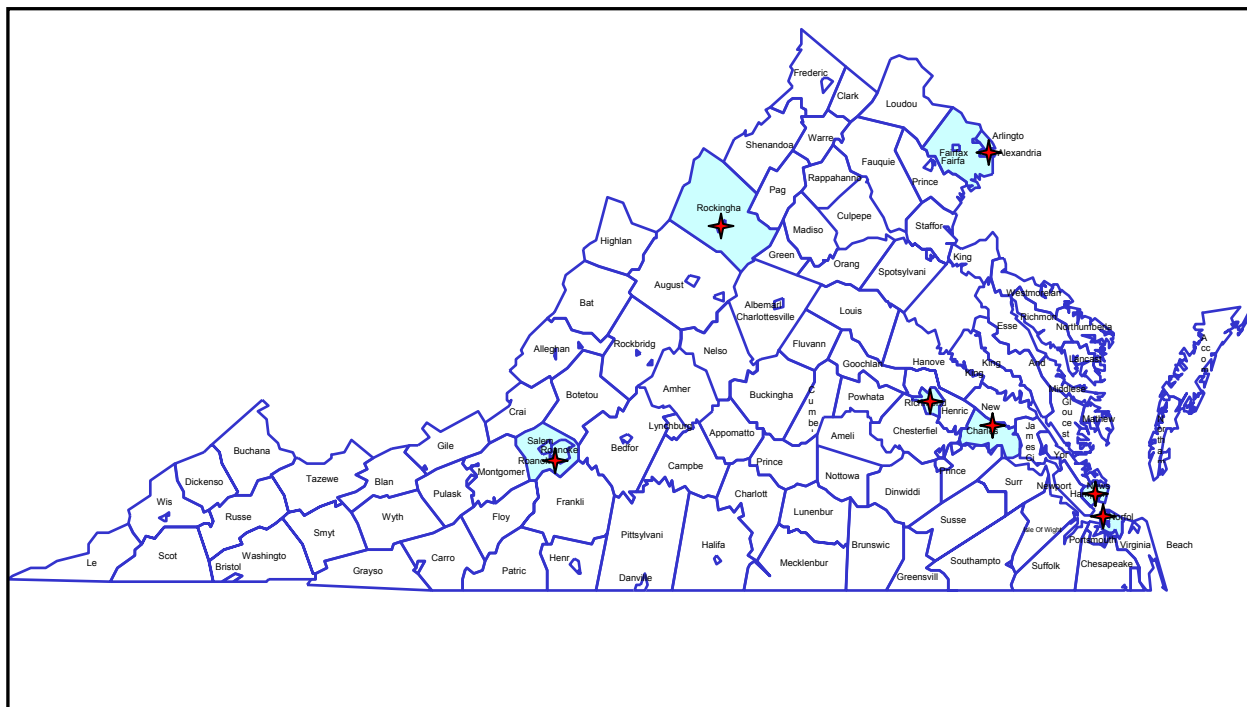


Sulfur Dioxide (SO₂) is a colorless gas that has a strong odor. It results from burning of fuels containing sulfur (such as coal and oil), petroleum refining, and smelting (extracting metals from ore), and it also occurs naturally from volcanic eruptions. SO₂ can dissolve in water vapor to produce sulfuric acid, and it can also interact with other gases and particles in the air to produce sulfate aerosols that are capable of traveling long distances in the atmosphere.

EPA has developed primary and secondary air quality standards for SO₂. The primary standards are designed to protect people from the health effects of sulfur dioxide gas, which include respiratory problems for people with asthma and for those who are active outdoors. Long-term exposure to high concentrations of sulfur dioxide gas can cause respiratory illness and aggravate existing heart conditions. Sulfate particles that are formed from SO₂ gas can be inhaled, and are associated with increased respiratory symptoms and disease.

Secondary standards for sulfur dioxide protect against damage to vegetation and buildings, and against decreased visibility. The acids that can form from SO₂ and water vapor contribute to acid deposition (commonly called "acid rain") which causes damage to the leaves of plants and trees, making them vulnerable to disease, and can increase the acidity of lakes and streams, making them unsuitable for aquatic life. Acid deposition also causes deterioration of materials on buildings, monuments, and sculptures. Finally, small sulfate particles, formed when SO₂ gas reacts with other gases and particles in the air, contribute to haze that causes decreased visibility in many areas of the country.

Sulfur dioxide is monitored continuously with an electronic instrument using ultraviolet fluorescence detection. The instrument has a pump that pulls outside air into a sample chamber containing a high intensity ultraviolet (UV) light. Any SO₂ molecules in the sample air absorb some of the UV light, become excited, and then fluoresce, releasing light characteristic of SO₂. The fluorescence is detected with a photomultiplier tube (a tube that detects very small amounts of light and multiplies the signal many times), and the resulting signal, which corresponds to the amount of SO₂ in the sample, is converted to an SO₂ concentration by the instrument computer.



SO2 Monitoring Network

NAAQS Standards

Primary Standards for SO₂:

- ➡ Annual Arithmetic Mean not to exceed 0.03 ppm (80 µg/m³).
- ➡ 24-Hour concentration not to exceed 0.14 ppm (365 µg/m³) more than once per year.

Secondary Standard for SO₂:

- ➡ 3-Hour concentration not to exceed 0.5 ppm (1300 µg/m³) more than once per year.

| Site | 2009 | | | |
|--------------------------------|----------------------|----------------------|----------------------|----------------------|
| | 24-Hour Avg. (ppm) | | 3-Hour Avg. (ppm) | |
| | 1 st Max. | 2 nd Max. | 1 st Max. | 2 nd Max. |
| (26-F) Rockingham Co. | .007 | .007 | .016 | .012 |
| (19-A6) Roanoke Co. | .006 | .006 | .013 | .013 |
| (75-B) Charles City Co. | .012 | .011 | .067 | .031 |
| (158-W) Richmond | .010 | .010 | .032 | .026 |
| (L-126-C) Alexandria | .025 | .016 | .055 | .046 |

NAAQS Standards

Primary Standards for SO₂:

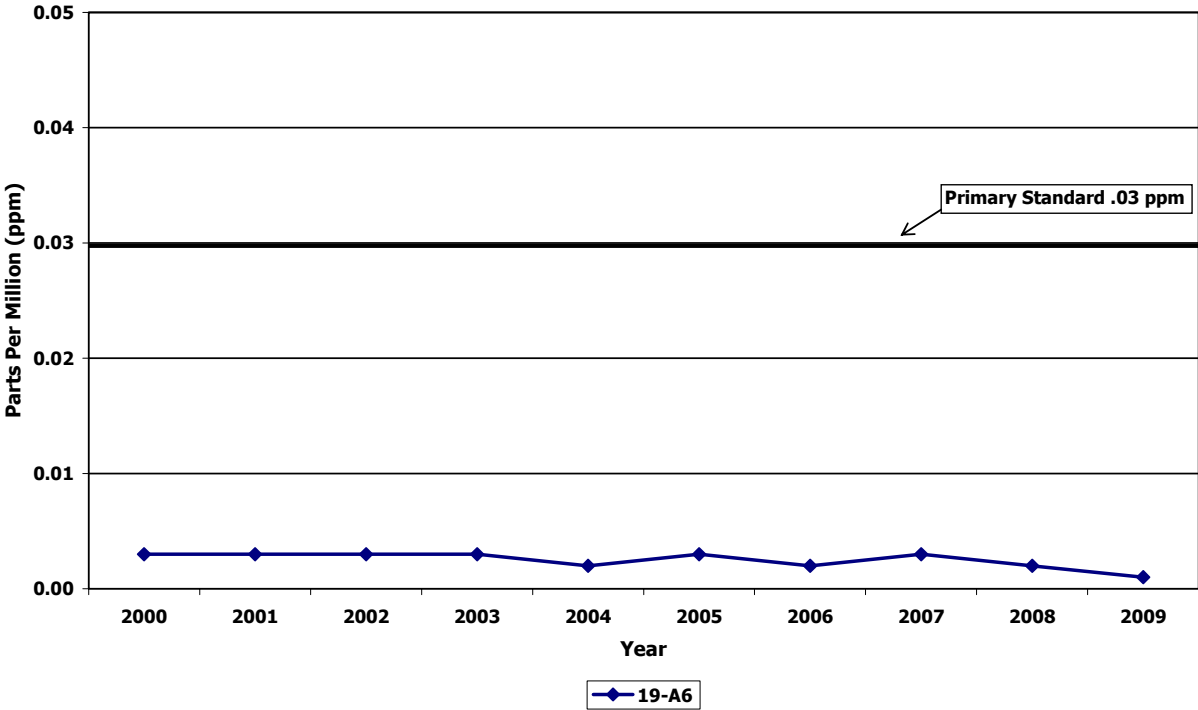
- Annual Arithmetic Mean not to exceed 0.03 ppm (80 µg/m³).
- 24-Hour concentration not to exceed 0.14 ppm (365 µg/m³) more than once per year.

Secondary Standard for SO₂:

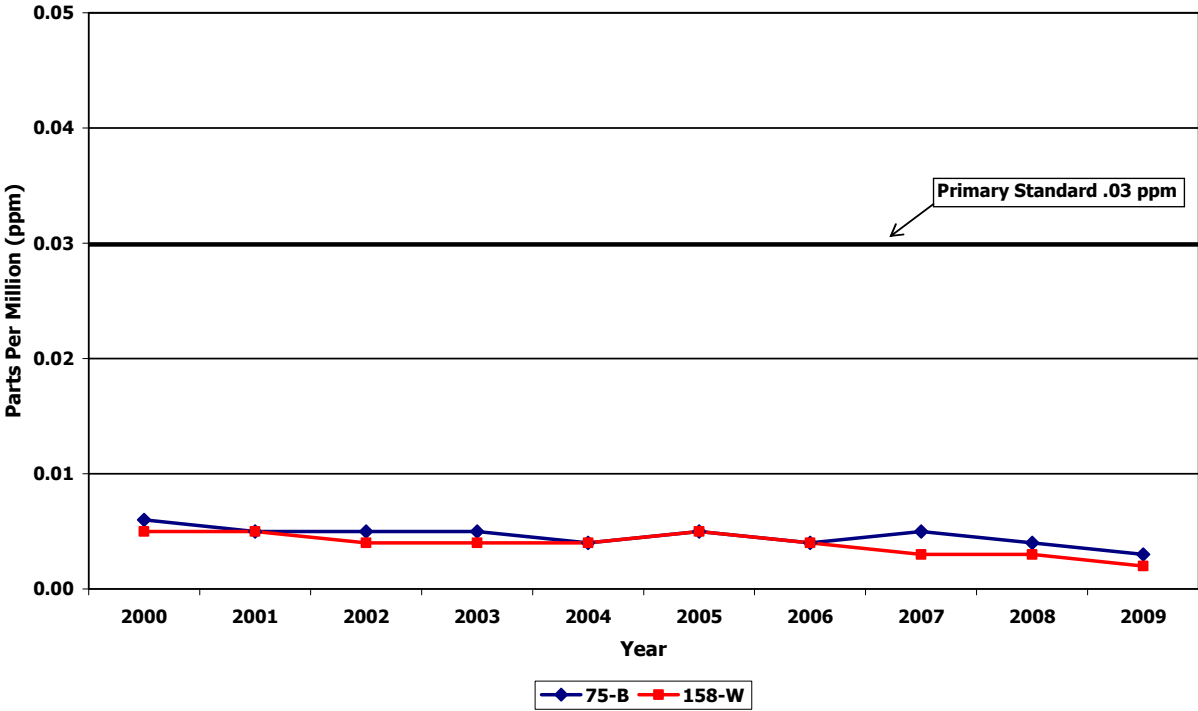
- 3-Hour concentration not to exceed 0.5 ppm (1300 µg/m³) more than once per year.

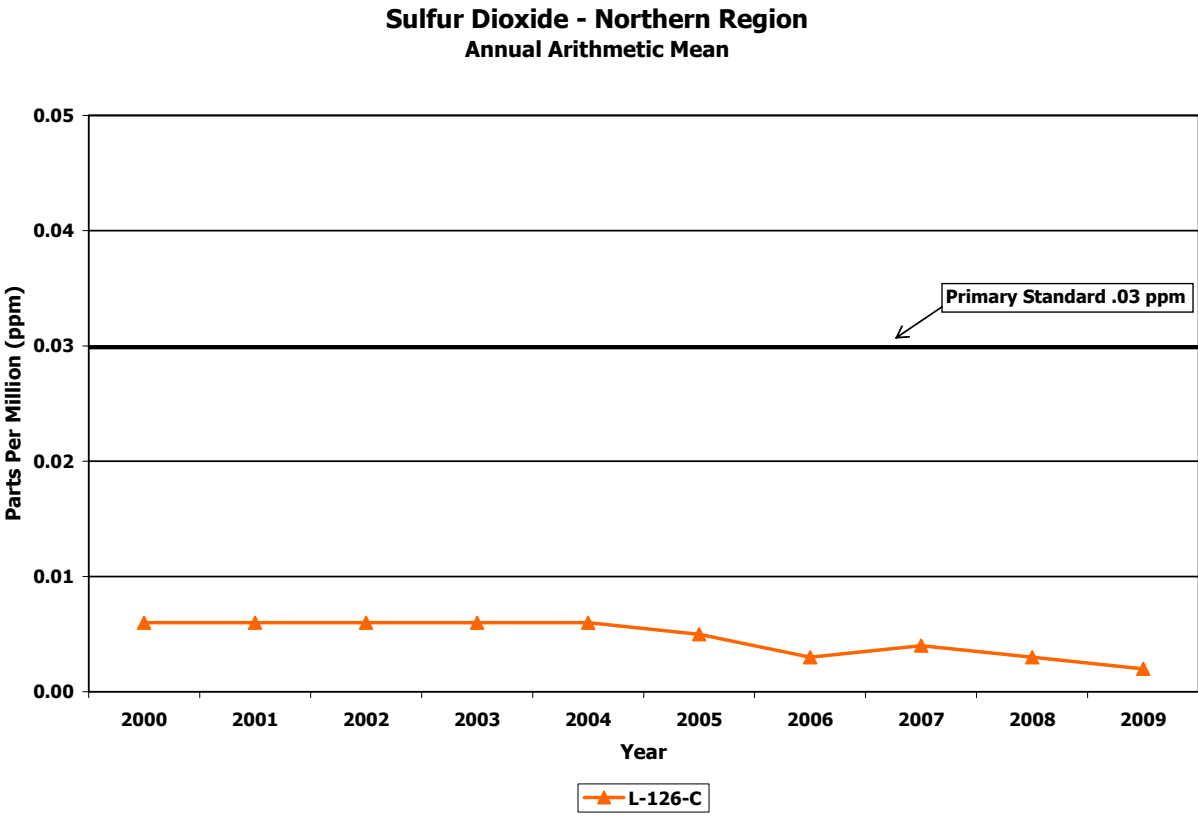
| Site | Annual Arithmetic Mean (ppm) | | | | | | | | | |
|--------------------------------|------------------------------|------|------|------|------|------|------|------|------|------|
| | 2000 | 2001 | 2002 | 2003 | 2004 | 2005 | 2006 | 2007 | 2008 | 2009 |
| (26-F) Rockingham Co. | -- | -- | -- | -- | -- | .002 | .002 | .001 | .001 | .002 |
| (19-A6) Roanoke Co. | .003 | .003 | .003 | .003 | .002 | .003 | .002 | .003 | .002 | .001 |
| (75-B) Charles City Co. | .006 | .005 | .005 | .005 | .004 | .005 | .004 | .005 | .004 | .003 |
| (158-W) Richmond | .005 | .005 | .004 | .004 | .004 | .005 | .004 | .003 | .003 | .002 |
| (L-126-C) Alexandria | .006 | .006 | .006 | .006 | .006 | .005 | .003 | .004 | .003 | .002 |

Sulfur Dioxide - Blue Ridge Region
Annual Arithmetic Mean



Sulfur Dioxide - Piedmont Region
Annual Arithmetic Mean



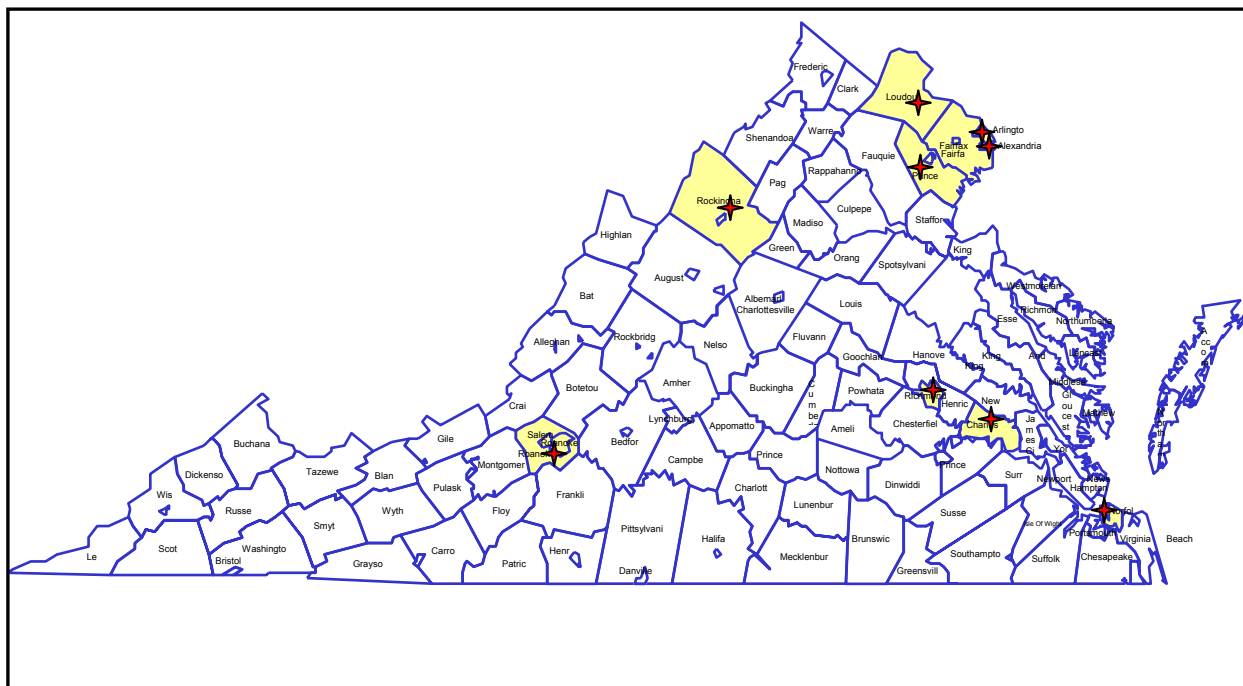


Nitrogen dioxide (NO₂) is one in a group of gases referred to as oxides of nitrogen (NO_x). Nitrogen dioxide, which is characterized by a reddish-brown color and pungent odor, along with the other NO_x gases, results from high-temperature burning of fossil fuels in automobiles, power plants, and other industrial, commercial, and residential sources. NO_x can occur naturally from lightning, forest fires, and bacterial processes that take place in soil.

NO_x pollution contributes to a wide range of problems in the environment. Ground-level ozone, a major component of “smog”, forms when NO_x and volatile organic compounds (VOCs) react in the presence of sunlight. NO_x also reacts with other gases and particles in the air to form acids that contribute to acid deposition, and to form small particles that can be inhaled into the lungs. NO_x contributes to water quality deterioration by depositing nitrogen into water bodies, upsetting the nutrient balance and causing oxygen depletion in the water so that fish and other aquatic life cannot survive. Nitrate particles and nitrogen dioxide also contribute to visibility impairment by blocking light transmission.

EPA has established primary and secondary air quality standards for NO₂ because it can cause lung irritation and respiratory problems in humans. Small particles formed from reaction of NO_x gases with other compounds can be inhaled deep into the lungs and cause or worsen respiratory conditions such as emphysema and bronchitis, and can aggravate existing heart conditions.

Nitrogen oxides are measured continuously with electronic instruments using the “gas phase chemiluminescence” method. The instrument has a pump that draws ambient air into a reaction chamber. Inside the chamber, the air is mixed with a high concentration of ozone (O₃). Any nitric oxide (NO) present in the sample air reacts with O₃ to produce NO₂. The NO₂ molecules created by the reaction are in an excited state, and emit light characteristic of NO₂ – this is called “chemiluminescence.” The light produced in the reaction is detected with a photomultiplier tube, and the resulting signal is converted to a number reflecting the concentration of NO in the ambient air by the instrument computer. The instrument then activates a valve that diverts incoming ambient air into a “converter”, which converts any NO₂ in the ambient air to NO by reduction reaction. After the air passes through the converter, it is sent to the reaction chamber where the NO and O₃ react to produce NO₂. The chemiluminescence produced by the reaction is converted to a signal that reflects the concentration of NO_x in the ambient air. The instrument then calculates the NO₂ concentration using the difference between the measured NO and NO_x concentrations.



NO₂ Monitoring Network

NAAQS Standards

Primary Standard for NO₂:

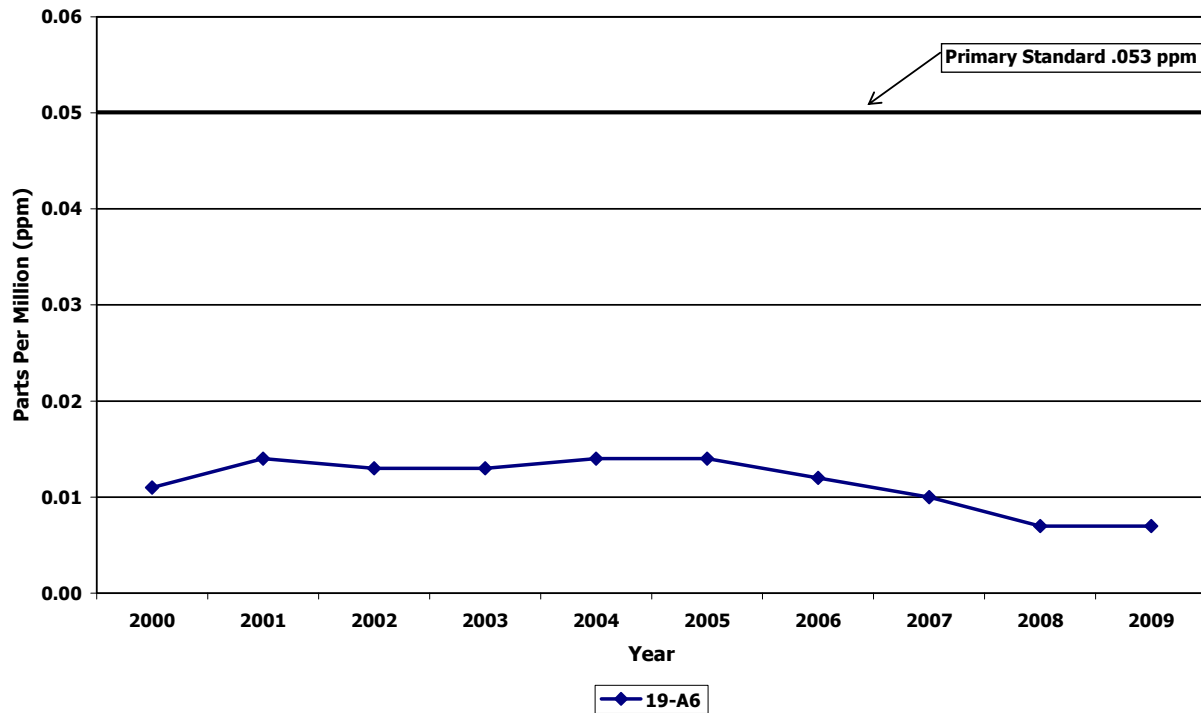
- Annual Arithmetic Mean not to exceed 0.053 ppm (100 µg/m³).

Secondary Standard for NO₂:

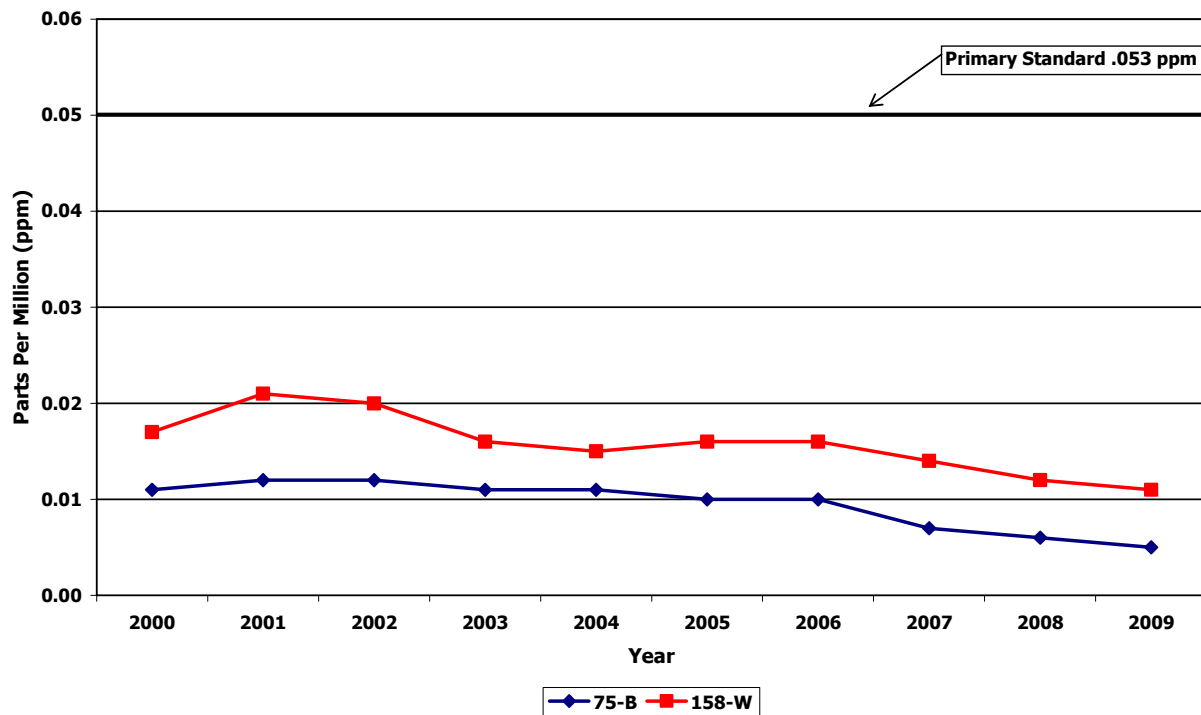
- Same as primary.

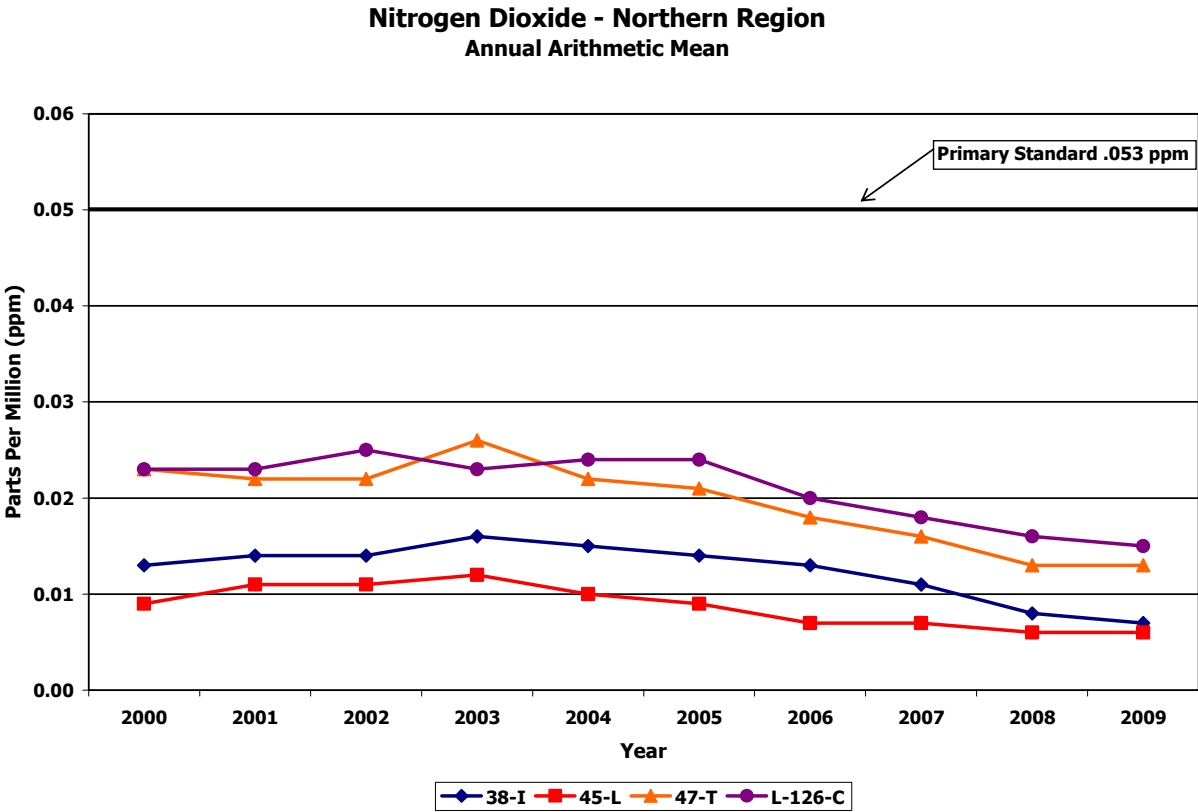
| Site | Annual Arithmetic Mean (ppm) | | | | | | | | | |
|----------------------------------|------------------------------|------|------|------|------|------|------|------|------|------|
| | 2000 | 2001 | 2002 | 2003 | 2004 | 2005 | 2006 | 2007 | 2008 | 2009 |
| (26-F) Rockingham Co. | -- | -- | -- | -- | -- | .014 | .012 | .011 | .011 | .009 |
| (19-A6) Roanoke Co. | .011 | .014 | .013 | .013 | .014 | .014 | .012 | .010 | .007 | .007 |
| (75-B) Charles City Co. | .011 | .012 | .012 | .011 | .011 | .010 | .010 | .007 | .006 | .005 |
| (158-W) Richmond | .017 | .021 | .020 | .016 | .015 | .016 | .016 | .014 | .012 | .011 |
| (38-I) Loudoun Co. | .013 | .014 | .014 | .016 | .015 | .014 | .013 | .011 | .008 | .007 |
| (45-L) Prince William Co. | .009 | .011 | .011 | .012 | .010 | .009 | .007 | .007 | .006 | .006 |
| (47-T) Arlington Co. | .023 | .022 | .022 | .026 | .022 | .021 | .018 | .016 | .013 | .013 |
| (L-126-C) Alexandria | .023 | .023 | .025 | .023 | .024 | .024 | .020 | .018 | .016 | .015 |

Nitrogen Dioxide - Blue Ridge Region Annual Arithmetic Mean



Nitrogen Dioxide - Piedmont Region Annual Arithmetic Mean





Ozone (O₃) is a gas comprised of three oxygen atoms. It is unstable, and a strong oxidizing agent, and will react readily with other compounds to decay to the more stable diatomic oxygen (O₂).

Ozone can be good or bad, depending on its location in the atmosphere. "Good" ozone occurs naturally in the stratosphere, about 10-30 miles above the earth's surface, where it forms a layer that filters the sun's ultraviolet rays before they reach the surface where they can cause harm to animals and plants. "Bad" ozone, or ground-level ozone, occurs when chemicals found in the atmosphere at earth's surface react in the presence of intense sunlight. Ozone at ground level is harmful because it can cause a variety of health problems, as well as damage to plants and materials. Since ground-level ozone is not emitted directly, it is called a "secondary" pollutant. The chemicals needed to form ozone, NO_x and hydrocarbons (also called volatile organic compounds, or VOCs), can come from motor vehicle exhaust, power plants, industrial emissions, gasoline vapors, chemical solvents, as well as natural sources such as lightning, forest fires, and plant decomposition. Ozone, and the chemicals that produce ozone, can travel hundreds of miles from their sources, so that even rural areas with few pollutant emissions can occasionally experience high ozone levels. Efforts to control ground-level ozone involve limiting emissions of NO_x and VOCs, or "ozone precursors," that are necessary for ozone production.

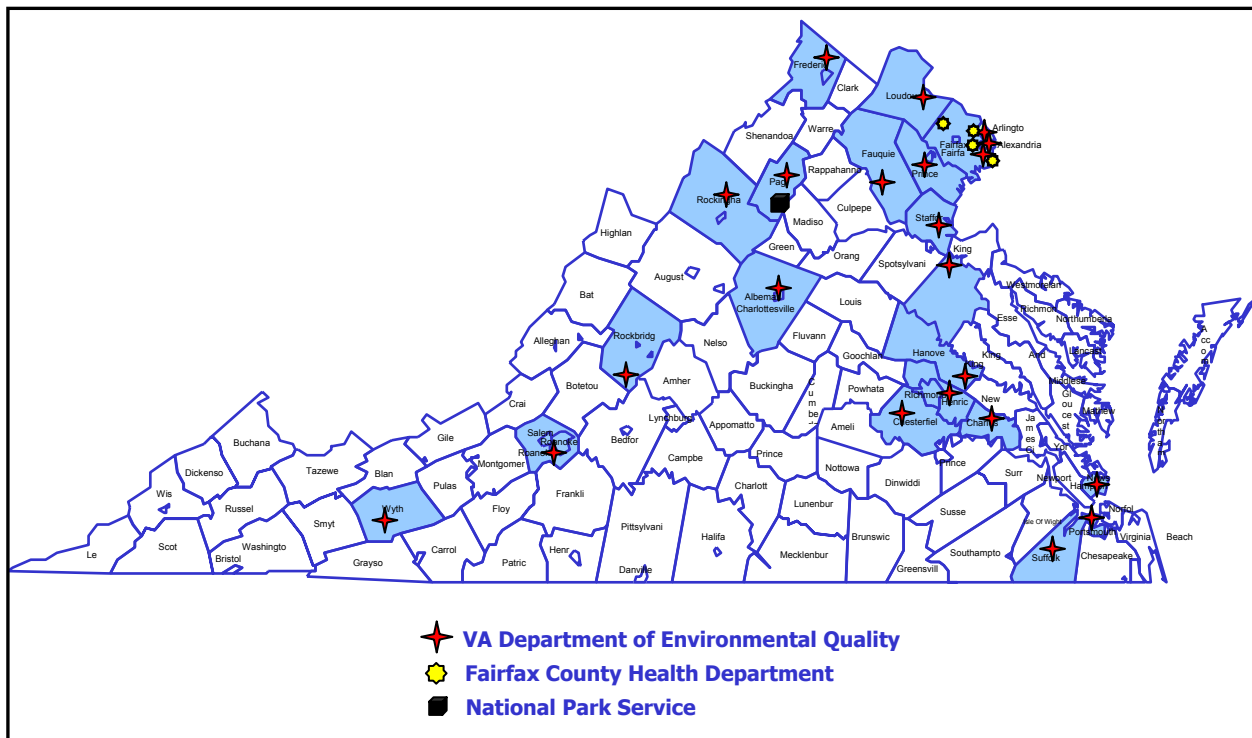
Ground-level ozone is a seasonal pollutant, and the length of the ozone season varies across the country. In some areas, the season may last most of the year, but in Virginia it is usually only a problem during the late spring to summer months when the sunlight is most intense. Virginia is only required to operate its ozone monitors from the months of April to October, although a few sites operate year-round. In addition to the seasonal pattern, ozone also has a strong diurnal (daily) pattern at low altitudes, so that it is usually depressed at night, but begins to build during the day after the sun rises.

EPA has created primary and secondary air quality standards for ground-level ozone because of its adverse affects on public health and welfare. In humans, ozone can irritate lung airways, causing sunburn-like inflammation, and can induce symptoms such as wheezing, coughing, and pain when taking a deep breath. Although people with existing respiratory problems, such as asthma and emphysema, are most vulnerable, young children and otherwise healthy people can also suffer respiratory problems from ozone exposure. Scientific studies have shown that even at low levels, ozone can trigger breathing problems for sensitive individuals. In addition to human health problems, ozone can damage the leaves of plants and trees, making them susceptible to disease, insects, and harsh weather. Ozone can also cause rubber to harden and crack, and some painted surfaces to fade more quickly.

Ozone is measured continuously with electronic instruments using “ultraviolet (UV) absorption photometry.” The method is based on the principle that ozone strongly absorbs UV light at 254 nanometers (a nanometer is equal to a distance of one billionth of a meter). The ozone monitor has a sample pump that draws ambient air into it and splits the air into two gas streams. In one stream, the air passes through an “ozone scrubber”, which cleanses the sample air of any ozone. Then the clean air passes through a sample cell that contains a UV light source and a detector. The detector measures the intensity of the light in the sample cell, providing a zero reference. The second air stream is sent straight into the sample cell, bypassing the scrubber. Any ozone present in the incoming air will absorb some of the UV light in the sample cell, reducing the amount of light reaching the detector. The instrument then calculates the ozone concentration of the ambient air from the difference in the light intensities measured between the scrubbed, or “zero” air, and the unscrubbed air.

Daily ozone forecasts for selected metropolitan areas and hourly ozone values for all Virginia ozone monitoring sites can be viewed for the months of April to October on the DEQ web page at <http://www.deq.virginia.gov/airquality/homepage>. In addition, animated ozone maps for Virginia and other parts of the United States are available at <http://www.airnow.gov/>.

The National Park Service operated one ozone monitor at Big Meadows in Shenandoah National Park in 2009. Daily data from this site are available at the DEQ website, and historical data may be obtained from the National Park Service, or by internet at <http://12.45.109.6/>.



NAAQS Standards

Primary Standard for O₃:

- Maximum 8-hour average concentration of 0.075 ppm (157 µg/m³), effective May 27, 2008, based on 3-year average of the annual fourth highest daily maximum 8-hour averages.

Secondary Standard for O₃:

Same as primary

The 8-hour standard is set at 0.075 ppm and is exceeded when an average level of ozone over an 8-hour period is 0.075 ppm. The standard is attained if the fourth highest daily maximum 8-hour average for each of the three most recent years at a monitoring site are averaged, yielding an average less than or equal to 0.075 ppm.

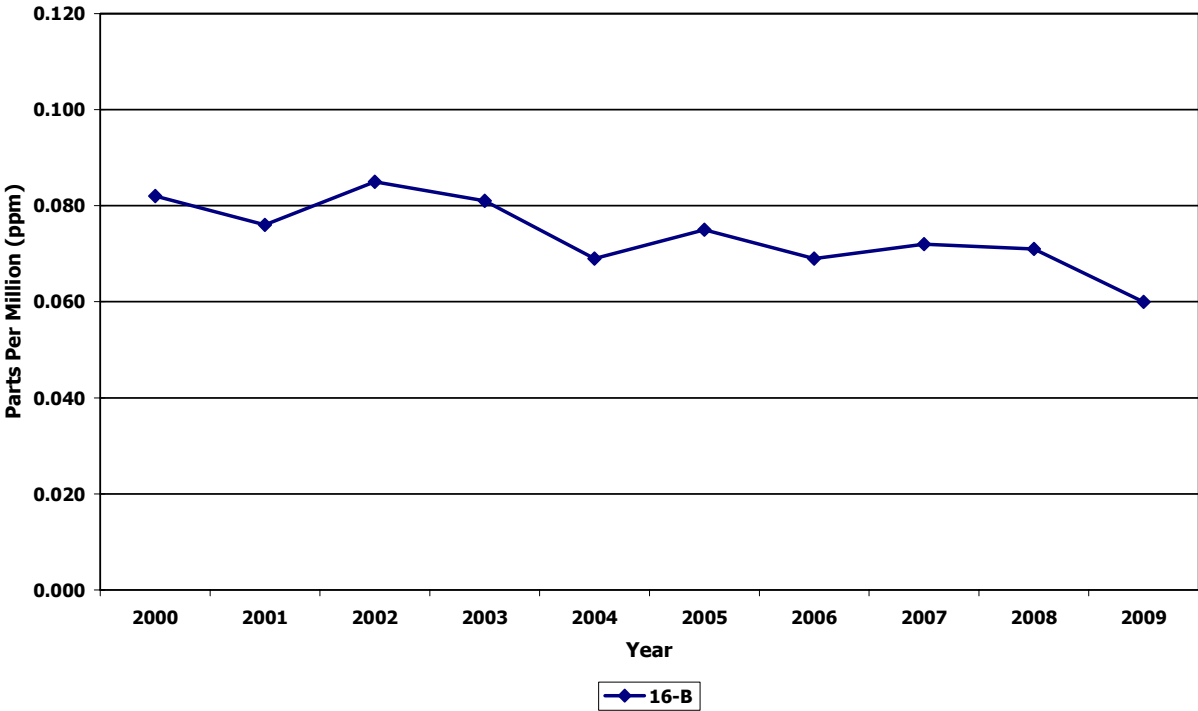
| Site | Days Exceeded 0.075 ppm | 2009 | | | |
|---------------------------|-------------------------|-----------------------------------|----------------------|----------------------|----------------------|
| | | Highest Daily Maximum 8-Hour Avg. | | | |
| | | 1 st Max. | 2 nd Max. | 3 rd Max. | 4 th Max. |
| (16-B) Wythe Co. | 0 | .064 | .064 | .062 | .060 |
| (26-F) Rockingham Co. | 0 | .065 | .064 | .064 | .063 |
| (28-J) Frederick Co. | 0 | .071 | .064 | .064 | .062 |
| (29-D) Page Co. | 0 | .063 | .063 | .063 | .063 |
| (33-A) Albemarle Co. | 0 | .070 | .066 | .066 | .065 |
| (19-A6) Roanoke Co. | 0 | .068 | .065 | .065 | .064 |
| (21-C) Rockbridge Co. | 0 | .064 | .062 | .060 | .060 |
| (71-H) Chesterfield Co. | 0 | .070 | .069 | .065 | .065 |
| (72-M) Henrico Co. | 0 | .074 | .070 | .066 | .065 |
| (73-E) Hanover Co. | 0 | .075 | .069 | .067 | .067 |
| (75-B) Charles City Co. | 0 | .069 | .067 | .064 | .063 |
| (180-O) Newport News * | 0 | .070 | .070 | .067 | .066 |
| (183-E) Suffolk | 0 | .070 | .069 | .068 | .065 |
| (183-F) Suffolk | 0 | .068 | .067 | .067 | .064 |
| (37-B) Fauquier Co. | 0 | .065 | .065 | .063 | .063 |
| (38-I) Loudoun Co. | 0 | .069 | .068 | .068 | .068 |
| (44-A) Stafford Co. | 0 | .069 | .068 | .065 | .064 |
| (45-L) Prince William Co. | 0 | .068 | .065 | .065 | .064 |
| (46-B9) Fairfax Co. | 1 | .080 | .073 | .071 | .070 |
| (47-T) Arlington Co. | 2 | .078 | .077 | .069 | .067 |
| (48-A) Caroline Co. | 0 | .070 | .068 | .067 | .066 |
| (L-46-A8) Fairfax Co. | 0 | .074 | .072 | .068 | .068 |
| (L-46-B3) Fairfax Co. | 0 | .075 | .074 | .071 | .069 |
| (L-46-C1) Fairfax Co. | 1 | .077 | .073 | .072 | .070 |
| (L-46-F) Fairfax Co. | 0 | .068 | .067 | .065 | .065 |
| (L-126-C) Alexandria | 0 | .069 | .067 | .067 | .066 |

* Site added in 2009 as a temporary replacement for Hampton (179-C).

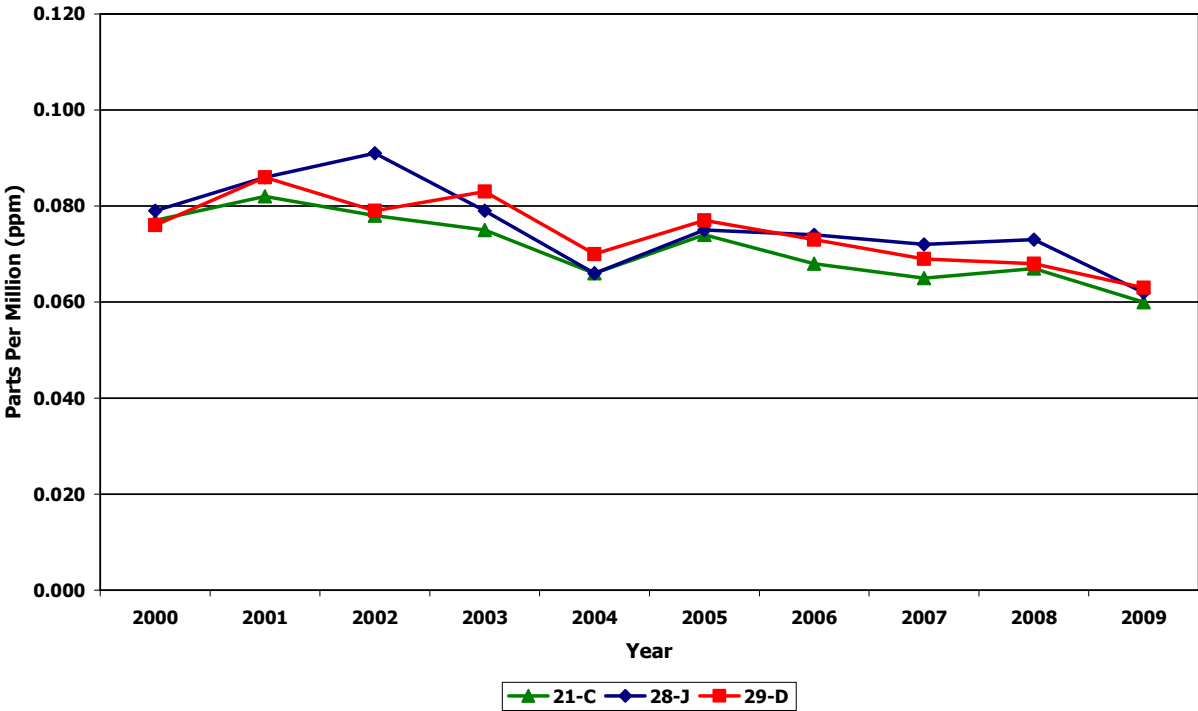
| 2007-2009 Fourth-Highest Daily Maximum 8-Hour Ozone Averages (units parts per million) | | | | | |
|---|---|-------------|-------------|-------------|--|
| | Monitor Location (County/City) | 2007 | 2008 | 2009 | 3-Year Average (NAAQS = .075 ppm) |
| Roanoke EAC Area | Roanoke Co. | .076 | .071 | .064 | .070 |
| Richmond Maintenance Area | Chesterfield Co. | .077 | .080 | .065 | .074 |
| | Henrico Co. | .085 | .086 | .065 | .078 |
| | Hanover Co. | .079 | .080 | .067 | .075 |
| | Charles City Co. | .084 | .084 | .063 | .077 |
| Hampton Roads Maintenance Area | Suffolk City (TCC) | .076 | .077 | .065 | .072 |
| | Suffolk City (Holland) | .078 | .078 | .064 | .073 |
| Winchester EAC Area | Frederick Co. | .072 | .073 | .062 | .069 |
| Fredericksburg Maintenance Area | Stafford Co. | .085 | .069 | .064 | .072 |
| Northern Virginia Nonattainment Area | Loudoun Co. | .086 | .079 | .068 | .077 |
| | Prince William Co. | .076 | .074 | .064 | .071 |
| | Arlington Co. | .088 | .084 | .067 | .079 |
| | Alexandria City | .084 | .075 | .066 | .075 |
| | Fairfax Co. (Lee Park) | .085 | .085 | .070 | .080 |
| | Fairfax Co. (McLean) | .083 | .080 | .068 | .077 |
| | Fairfax Co. (Chantilly) | .078 | .078 | .065 | .073 |
| | Fairfax Co. (Annandale) | .084 | .082 | .070 | .078 |
| | Fairfax Co. (Mt. Vernon) | .088 | .085 | .069 | .080 |
| Shenandoah National Park Maintenance Area | Madison Co. (Big Meadows) | .073 | .078 | .069 | .073 |
| Areas Currently Designated Attainment | Wythe Co. | .072 | .071 | .060 | .067 |
| | Rockbridge Co. | .065 | .067 | .060 | .064 |
| | Rockingham Co. | .069 | .069 | .063 | .067 |
| | Page Co. | .069 | .068 | .063 | .066 |
| | Fauquier Co. | .069 | .068 | .063 | .066 |
| | Caroline Co. | .078 | .080 | .066 | .074 |

A 3-year average greater than .075 ppm exceeds the 8-hour NAAQS for ozone

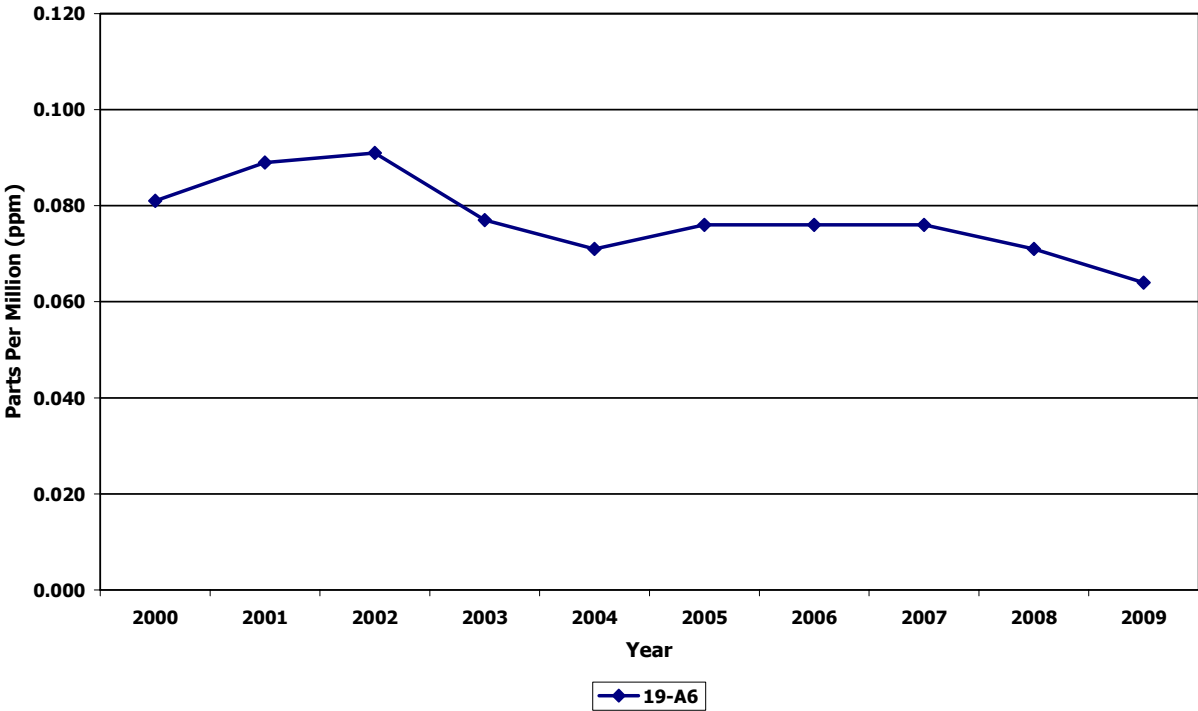
Ozone - Southwest Region
4th Daily Maximum, 8-Hour Value



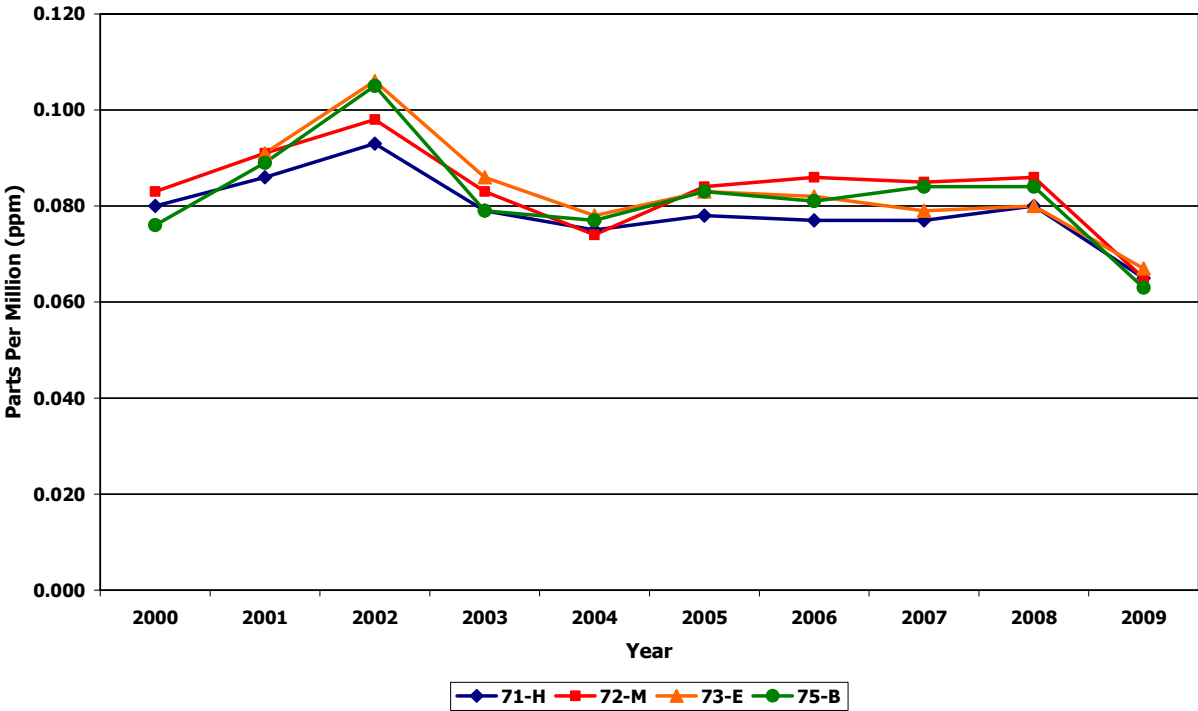
Ozone - Valley Region
4th Daily Maximum, 8-Hour Value



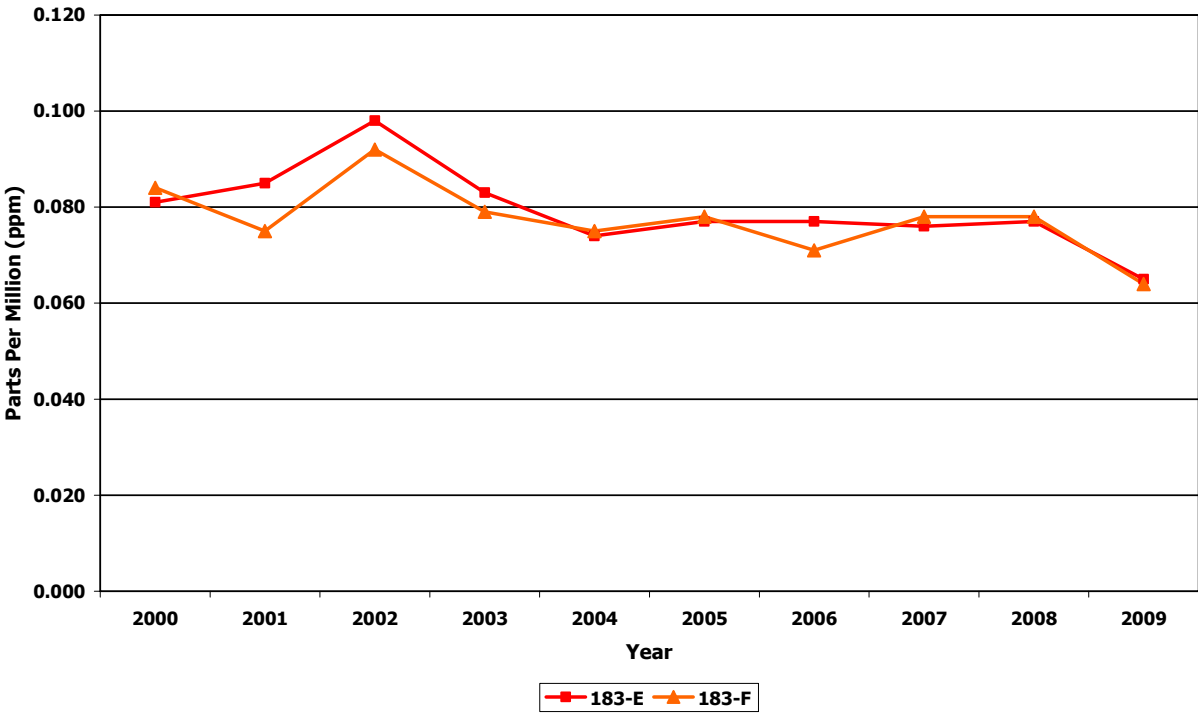
Ozone - Blue Ridge Region
4th Daily Maximum, 8-Hour Value



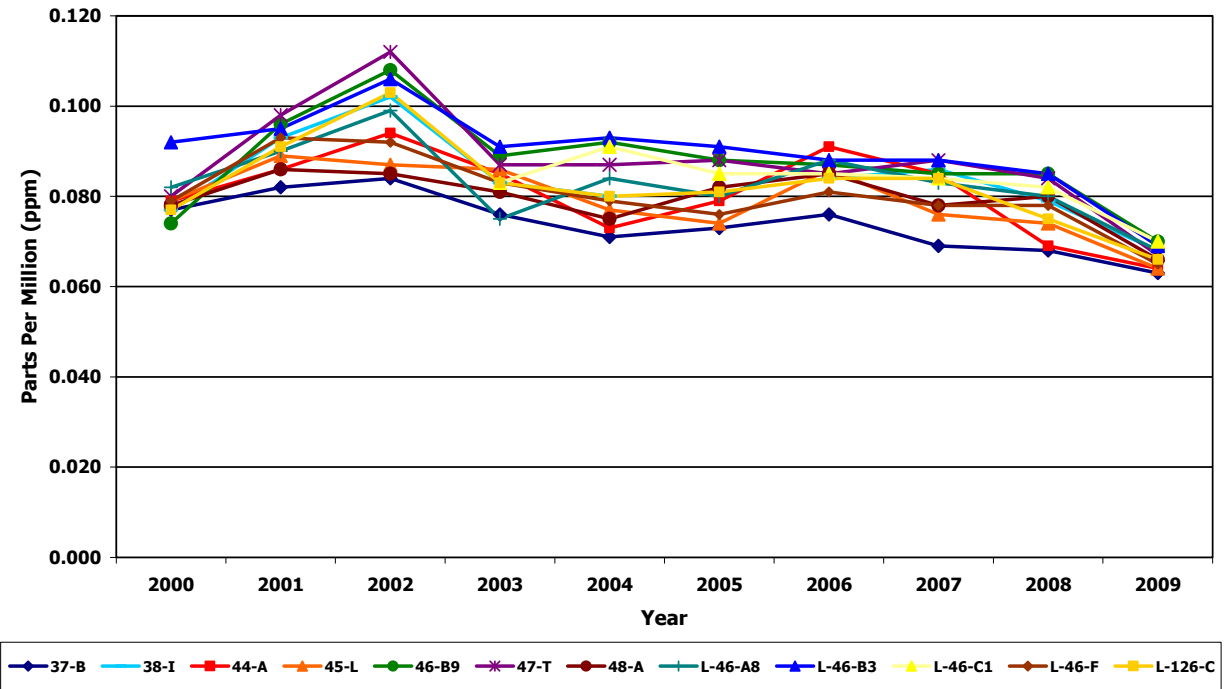
Ozone - Piedmont Region
4th Daily Maximum, 8-Hour Value



Ozone - Tidewater Region
4th Daily Maximum, 8-Hour Value



Ozone - Northern Region
4th Daily Maximum, 8-Hour Value



Acid Deposition Program

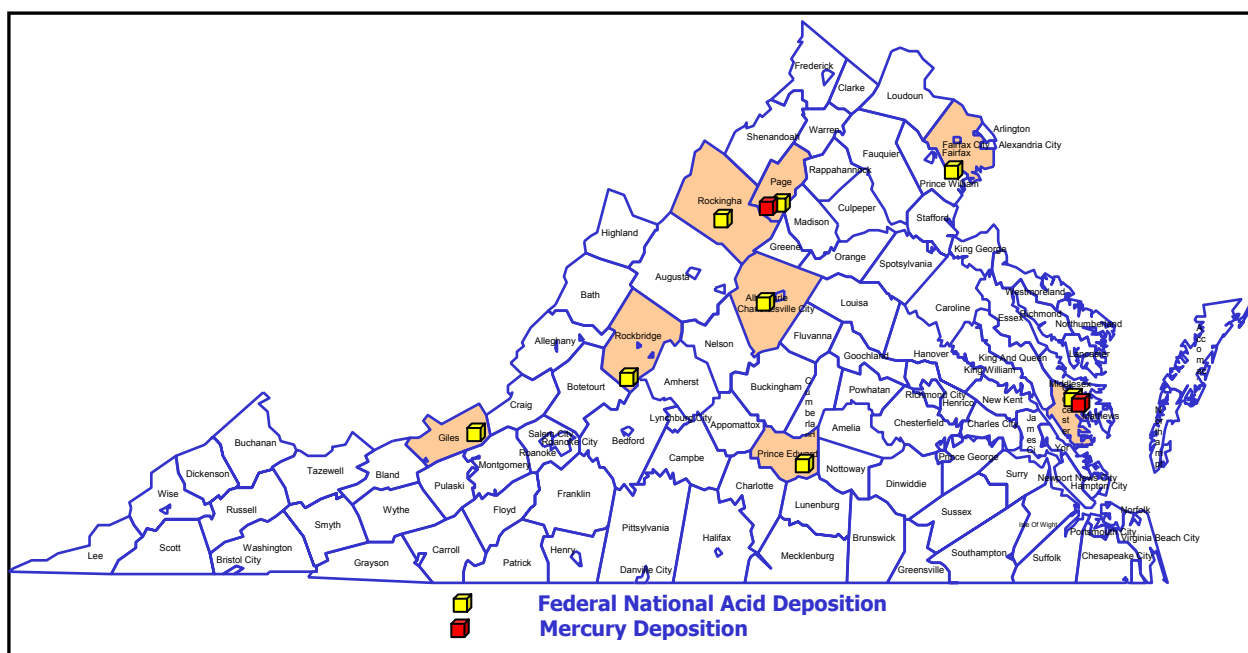
**Photochemical Assessment
Monitoring Stations**

Air Toxics Monitoring Network

The Virginia Department of Environmental Quality sponsored two National Acid Deposition Program (NADP) sites in 2009: Harcum in Gloucester County and Mason Neck in Fairfax County (the later operated through July 14, 2009).

The NADP had eight monitoring sites in Virginia in 2009: Big Meadows (Shenandoah National Park), Hortons Station (Giles County), Charlottesville, Prince Edward County, Harcum (Gloucester County), Natural Bridge Station (Rockbridge County). Harrisonburg (Rockingham County) operated through June 30, 2009 and Mason Neck (Fairfax County) operated through July 14, 2009. NADP site information and data are available on-line at <http://nadp.sws.uiuc.edu>.

In addition to the eight acid deposition monitors, there were two NADP Mercury Deposition Network (MDN) sites in Virginia: Harcum (Gloucester County), and Big Meadows (Shenandoah National Park). MDN site information and data are available on-line at <http://nadp.sws.uiuc.edu/MDN/>.



Acid Precipitation Monitors

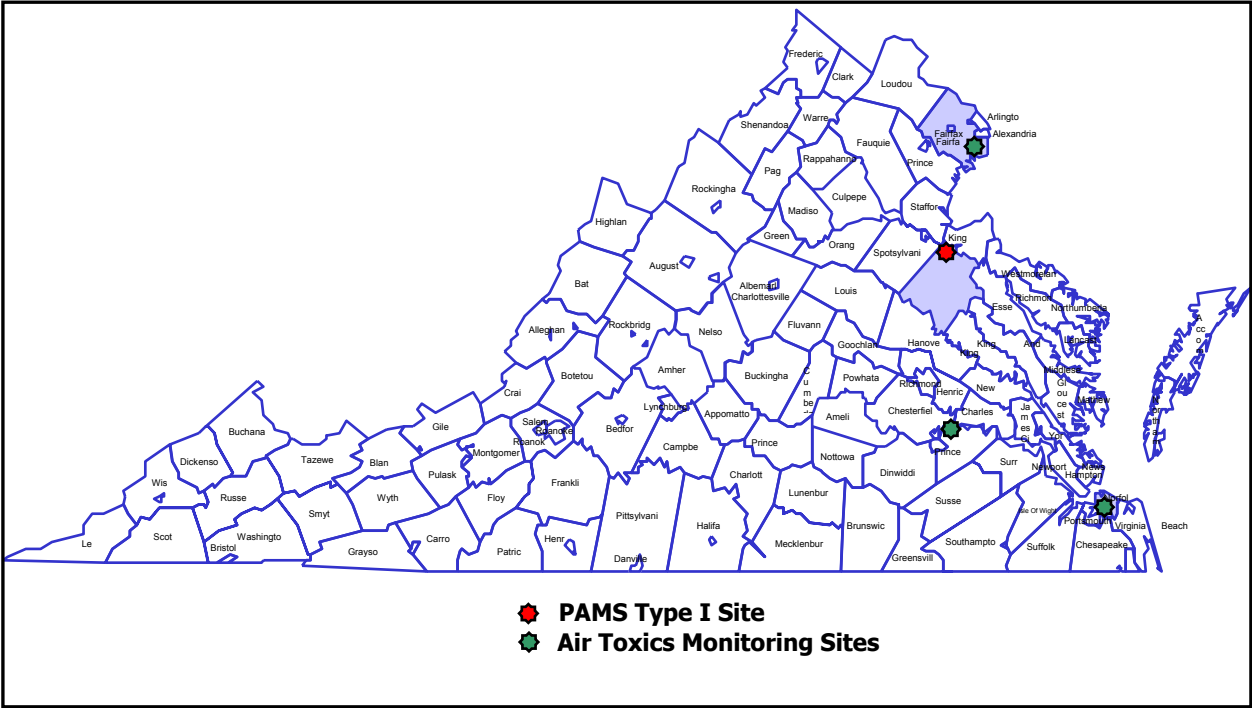
In 2009, the Office of Air Quality Monitoring (AQM) program of the Department of Environmental Quality operated a Photochemical Assessment Monitoring station (PAMS) at Corbin in Caroline County. Additionally, 24-hour PAMS Volatile Organic Compounds (VOC) samples were collected from three core Air Toxics Monitoring Network (ATMN) sites located at Carter G. Woodson Middle School (Woodson) in the City of Hopewell, Lee District Park in Fairfax County and the DEQ Tidewater Regional Office (TRO) in Virginia Beach, using a one in six day sampling schedule.

Corbin was operated as a PAMS Type I site, collecting 24-hour VOC samples every six days. A Type I site measures upwind background ozone precursor concentrations. Hourly samples were collected using an Auto Gas Chromatograph during peak ozone season (months of June, July and August).

AQM used the manual method for collecting ambient air samples. This method involves the collection of integrated, whole samples by using evacuated Summa^T canisters and RMESI (RM Environmental Systems, Inc.) air samplers. Each VOC sample from Corbin was analyzed by the Division of Consolidated Laboratory Services using a Gas Chromatograph/Flame Ionization Detector. Samples from Woodson, Lee District Park, and TRO were analyzed by the Maryland Department of the Environment, Air and Radiation Management Administration, using a Gas Chromatograph/Flame Ionization Detector.

All VOC samples were analyzed for the presence of fifty-six target volatile organic precursors, and the measured concentration of Total Nonmethane Organic Compounds (TNMOC).

Detailed PAMS data are available upon written request to the Virginia Department of Environmental Quality, Office of Air Quality Monitoring.



Photochemical Assessment Monitoring Network

2009 Average Concentration of Detectable Volatile Ozone Precursors Photochemical Assessment Monitoring Station (PAMS) Type I - Corbin

Concentrations are in ppbC
(non detects are counted as zeros for statistical purposes)

| Parameter | Compound Name | Num | Minimum | Maximum | Median | Average | StDev |
|--------------|------------------------|-----|---------|---------|--------|---------|--------|
| 43141 | n-dodecane | 0 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 43202 | Ethane | 58 | 1.26 | 9.16 | 4.24 | 4.357 | 1.862 |
| 43203 | Ethylene | 58 | 0.21 | 4.04 | 0.91 | 1.069 | 0.720 |
| 43204 | Propane | 58 | 1.12 | 8.03 | 2.91 | 3.314 | 1.713 |
| 43205 | Propylene | 58 | 0.00 | 0.94 | 0.35 | 0.358 | 0.160 |
| 43206 | Acetylene | 58 | 0.01 | 40.96 | 1.11 | 8.653 | 11.705 |
| 43212 | n-butane | 58 | 0.28 | 3.85 | 1.29 | 1.481 | 0.915 |
| 43214 | Isobutane | 58 | 0.00 | 1.47 | 0.35 | 0.468 | 0.346 |
| 43216 | t-2-butene | 58 | 0.00 | 0.27 | 0.00 | 0.005 | 0.036 |
| 43217 | c-2-butene | 58 | 0.00 | 2.24 | 0.00 | 0.040 | 0.294 |
| 43220 | n-pentane | 58 | 0.20 | 1.69 | 0.65 | 0.717 | 0.351 |
| 43221 | Isopentane | 58 | 0.28 | 2.01 | 0.84 | 0.947 | 0.380 |
| 43224 | 1-pentene | 58 | 0.00 | 0.82 | 0.00 | 0.098 | 0.208 |
| 43226 | t-2-pentene | 58 | 0.00 | 1.95 | 0.00 | 0.071 | 0.288 |
| 43227 | c-2-pentene | 58 | 0.00 | 1.00 | 0.00 | 0.057 | 0.186 |
| 43230 | 3-methylpentane | 58 | 0.00 | 0.42 | 0.18 | 0.182 | 0.099 |
| 43231 | n-hexane | 58 | 0.00 | 0.86 | 0.21 | 0.235 | 0.152 |
| 43232 | n-heptane | 58 | 0.00 | 0.87 | 0.13 | 0.141 | 0.164 |
| 43233 | n-octane | 58 | 0.00 | 0.22 | 0.00 | 0.038 | 0.070 |
| 43235 | n-nonane | 58 | 0.00 | 0.80 | 0.03 | 0.229 | 0.262 |
| 43238 | n-decane | 58 | 0.00 | 1.70 | 0.43 | 0.488 | 0.349 |
| 43242 | Cyclopentane | 58 | 0.00 | 0.18 | 0.00 | 0.013 | 0.035 |
| 43243 | Isoprene | 58 | 0.00 | 21.28 | 0.24 | 3.270 | 5.318 |
| 43244 | 2,2-dimethylbutane | 58 | 0.00 | 0.19 | 0.00 | 0.038 | 0.060 |
| 43245 | 1-hexene | 58 | 0.00 | 1.60 | 0.47 | 0.420 | 0.361 |
| 43247 | 2,4-dimethylpentane | 58 | 0.00 | 0.22 | 0.00 | 0.010 | 0.038 |
| 43248 | Cyclohexane | 58 | 0.00 | 0.33 | 0.00 | 0.022 | 0.064 |
| 43249 | 3-methylhexane | 58 | 0.00 | 0.78 | 0.17 | 0.200 | 0.192 |
| 43250 | 2,2,4-trimethylpentane | 58 | 0.00 | 0.55 | 0.23 | 0.245 | 0.128 |
| 43252 | 2,3,4-trimethylpentane | 58 | 0.00 | 0.08 | 0.00 | 0.001 | 0.011 |
| 43253 | 3-methylheptane | 58 | 0.00 | 0.18 | 0.00 | 0.005 | 0.027 |
| 43261 | Methylcyclohexane | 58 | 0.00 | 0.20 | 0.00 | 0.023 | 0.056 |
| 43262 | Methylcyclopentane | 58 | 0.00 | 0.23 | 0.00 | 0.054 | 0.072 |
| 43263 | 2-methylhexane | 58 | 0.00 | 0.57 | 0.00 | 0.096 | 0.128 |
| 43280 | 1-butene | 58 | 0.00 | 2.04 | 0.26 | 0.479 | 0.441 |
| 43284 | 2,3-dimethylbutane | 58 | 0.00 | 0.26 | 0.00 | 0.051 | 0.063 |
| 43285 | 2-methylpentane | 58 | 0.00 | 0.72 | 0.24 | 0.242 | 0.156 |
| 43291 | 2,3-dimethylpentane | 58 | 0.00 | 0.60 | 0.00 | 0.052 | 0.094 |
| 43954 | n-undecane | 58 | 0.00 | 0.22 | 0.00 | 0.004 | 0.029 |
| 43960 | 2-methylheptane | 58 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 45109 | m/p-xylene | 58 | 0.00 | 0.69 | 0.19 | 0.192 | 0.162 |
| 45201 | Benzene | 58 | 0.19 | 1.44 | 0.52 | 0.563 | 0.247 |
| 45202 | Toluene | 58 | 0.21 | 1.36 | 0.57 | 0.638 | 0.275 |
| 45203 | Ethylbenzene | 58 | 0.00 | 0.27 | 0.00 | 0.024 | 0.064 |
| 45204 | o-xylene | 58 | 0.00 | 0.98 | 0.42 | 0.424 | 0.251 |
| 45207 | 1,3,5-trimethylbenzene | 58 | 0.00 | 0.07 | 0.00 | 0.003 | 0.012 |
| 45208 | 1,2,4-trimethylbenzene | 58 | 0.00 | 0.21 | 0.00 | 0.017 | 0.047 |
| 45209 | n-propylbenzene | 58 | 0.00 | 0.30 | 0.00 | 0.016 | 0.049 |
| 45210 | Isopropylbenzene | 58 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 45211 | o-ethyltoluene | 58 | 0.00 | 0.79 | 0.00 | 0.114 | 0.201 |
| 45212 | m-ethyltoluene | 58 | 0.00 | 1.28 | 0.36 | 0.324 | 0.249 |
| 45213 | p-ethyltoluene | 58 | 0.00 | 0.59 | 0.00 | 0.075 | 0.139 |
| 45218 | m-diethylbenzene | 58 | 0.00 | 0.20 | 0.00 | 0.003 | 0.026 |
| 45219 | p-diethylbenzene | 58 | 0.00 | 0.14 | 0.00 | 0.002 | 0.018 |
| 45220 | Styrene | 58 | 0.00 | 2.08 | 0.06 | 0.130 | 0.282 |
| 45225 | 1,2,3-trimethylbenzene | 58 | 0.00 | 4.21 | 0.63 | 0.892 | 0.926 |
| 43000 | PAMHC | 58 | 10.77 | 63.78 | 28.70 | 31.171 | 12.353 |
| 43102 | TNMOC | 58 | 22.60 | 198.56 | 48.52 | 51.746 | 23.401 |

**2009 Average Concentration of Detectable Volatile Ozone Precursors
Photochemical Assessment Monitoring Station Additional VOC PAMS Sampling –
Carter G. Woodson Middle School, Hopewell**

(Concentrations are in ppbC)
(non detects are counted as zeros for statistical purposes)

| Parameter | Compound Name | Num | Minimum | Maximum | Median | Average | StDev |
|--------------|------------------------|-----|---------|---------|--------|---------|--------|
| 43141 | n-dodecane | 61 | 0.00 | 0.68 | 0.12 | 0.138 | 0.107 |
| 43202 | Ethane | 61 | 1.63 | 20.47 | 5.53 | 6.645 | 4.117 |
| 43203 | Ethylene | 61 | 0.60 | 15.20 | 1.37 | 2.542 | 2.875 |
| 43204 | Propane | 61 | 1.33 | 23.27 | 3.95 | 5.903 | 4.932 |
| 43205 | Propylene | 61 | 0.24 | 4.65 | 0.53 | 1.000 | 1.077 |
| 43206 | Acetylene | 61 | 0.29 | 7.35 | 1.05 | 1.632 | 1.497 |
| 43212 | n-butane | 61 | 0.47 | 19.25 | 2.22 | 3.846 | 4.119 |
| 43214 | Isobutane | 61 | 0.30 | 9.36 | 1.10 | 1.766 | 1.810 |
| 43216 | t-2-butene | 61 | 0.03 | 0.50 | 0.06 | 0.102 | 0.113 |
| 43217 | c-2-butene | 61 | 0.02 | 0.44 | 0.06 | 0.097 | 0.099 |
| 43220 | n-pentane | 61 | 0.51 | 79.21 | 1.51 | 3.365 | 10.032 |
| 43221 | Isopentane | 61 | 0.63 | 42.21 | 2.21 | 3.887 | 5.923 |
| 43224 | 1-pentene | 61 | 0.09 | 1.06 | 0.14 | 0.207 | 0.177 |
| 43226 | t-2-pentene | 61 | 0.05 | 1.66 | 0.19 | 0.289 | 0.294 |
| 43227 | c-2-pentene | 61 | 0.03 | 0.65 | 0.07 | 0.117 | 0.127 |
| 43230 | 3-methylpentane | 61 | 0.16 | 3.99 | 0.49 | 0.765 | 0.786 |
| 43231 | n-hexane | 61 | 0.19 | 4.59 | 0.57 | 0.883 | 0.905 |
| 43232 | n-heptane | 61 | 0.11 | 2.67 | 0.30 | 0.510 | 0.545 |
| 43233 | n-octane | 61 | 0.05 | 0.85 | 0.12 | 0.189 | 0.188 |
| 43235 | n-nonane | 61 | 0.04 | 0.68 | 0.11 | 0.159 | 0.139 |
| 43238 | n-decane | 61 | 0.00 | 0.89 | 0.10 | 0.177 | 0.184 |
| 43242 | Cyclopentane | 61 | 0.05 | 0.91 | 0.14 | 0.194 | 0.176 |
| 43243 | Isoprene | 61 | 0.04 | 11.82 | 0.65 | 2.421 | 3.397 |
| 43244 | 2,2-dimethylbutane | 61 | 0.06 | 0.98 | 0.14 | 0.205 | 0.182 |
| 43245 | 1-Hexene | 61 | 0.00 | 0.32 | 0.05 | 0.072 | 0.063 |
| 43247 | 2,4-dimethylpentane | 61 | 0.04 | 0.98 | 0.12 | 0.192 | 0.198 |
| 43248 | Cyclohexane | 61 | 0.00 | 1.36 | 0.12 | 0.189 | 0.215 |
| 43249 | 3-methylhexane | 61 | 0.13 | 1.59 | 0.33 | 0.428 | 0.306 |
| 43250 | 2,2,4-trimethylpentane | 61 | 0.19 | 5.08 | 0.64 | 0.990 | 1.067 |
| 43252 | 2,3,4-trimethylpentane | 61 | 0.09 | 1.85 | 0.23 | 0.367 | 0.395 |
| 43253 | 3-methylheptane | 61 | 0.04 | 0.78 | 0.09 | 0.146 | 0.160 |
| 43261 | Methylcyclohexane | 61 | 0.06 | 2.09 | 0.17 | 0.269 | 0.325 |
| 43262 | Methylcyclopentane | 61 | 0.14 | 2.44 | 0.36 | 0.529 | 0.474 |
| 43263 | 2-methylhexane | 61 | 0.14 | 2.72 | 0.36 | 0.570 | 0.532 |
| 43280 | 1-butene | 61 | 0.00 | 0.82 | 0.11 | 0.183 | 0.184 |
| 43284 | 2,3-dimethylbutane | 61 | 0.06 | 1.65 | 0.20 | 0.321 | 0.334 |
| 43285 | 2-methylpentane | 61 | 0.24 | 6.68 | 1.16 | 1.579 | 1.294 |
| 43291 | 2,3-dimethylpentane | 61 | 0.07 | 1.14 | 0.17 | 0.252 | 0.224 |
| 43954 | n-undecane | 61 | 0.00 | 0.64 | 0.15 | 0.186 | 0.141 |
| 43960 | 2-methylheptane | 61 | 0.05 | 0.84 | 0.11 | 0.176 | 0.172 |
| 45109 | m/p-xylene | 61 | 0.27 | 6.64 | 0.69 | 1.291 | 1.468 |
| 45201 | Benzene | 61 | 0.28 | 5.08 | 0.89 | 1.319 | 1.087 |
| 45202 | Toluene | 61 | 0.77 | 21.63 | 1.68 | 2.909 | 3.480 |
| 45203 | Ethylbenzene | 61 | 0.10 | 2.25 | 0.24 | 0.445 | 0.503 |
| 45204 | o-xylene | 61 | 0.07 | 2.69 | 0.25 | 0.476 | 0.575 |
| 45207 | 1,3,5-trimethylbenzene | 61 | 0.03 | 1.03 | 0.12 | 0.206 | 0.224 |
| 45208 | 1,2,4-trimethylbenzene | 61 | 0.04 | 2.81 | 0.28 | 0.500 | 0.589 |
| 45209 | n-propylbenzene | 61 | 0.03 | 0.49 | 0.08 | 0.115 | 0.107 |
| 45210 | Isopropylbenzene | 61 | 0.00 | 0.29 | 0.06 | 0.082 | 0.062 |
| 45211 | o-ethyltoluene | 61 | 0.03 | 1.25 | 0.06 | 0.124 | 0.117 |
| 45212 | m-ethyltoluene | 61 | 0.05 | 1.90 | 0.19 | 0.331 | 0.404 |
| 45213 | p-ethyltoluene | 61 | 0.06 | 1.55 | 0.25 | 0.352 | 0.305 |
| 45218 | m-diethylbenzene | 61 | 0.00 | 0.31 | 0.04 | 0.052 | 0.045 |
| 45219 | p-diethylbenzene | 61 | 0.00 | 0.18 | 0.07 | 0.068 | 0.037 |
| 45220 | Styrene | 61 | 0.06 | 0.95 | 0.25 | 0.273 | 0.149 |
| 45225 | 1,2,3-trimethylbenzene | 61 | 0.00 | 0.65 | 0.09 | 0.140 | 0.127 |
| 43000 | PAMHC | 61 | 15.63 | 204.77 | 35.46 | 52.173 | 44.124 |
| 43102 | TNMOC | 61 | 32.93 | 290.09 | 62.82 | 81.012 | 57.245 |

**2009 Average Concentration of Detectable Volatile Ozone Precursors
Photochemical Assessment Monitoring Station Additional VOC PAMS Sampling -
Tidewater Regional Office (TRO)**

Concentrations are in ppbC
(non detects are counted as zeros for statistical purposes)

| Parameter | Compound Name | Num | Minimum | Maximum | Median | Average | StDev |
|--------------|------------------------|-----|---------|---------|--------|---------|--------|
| 43141 | n-dodecane | 61 | 0.00 | 0.50 | 0.10 | 0.134 | 0.101 |
| 43202 | Ethane | 61 | 1.71 | 18.12 | 5.67 | 6.609 | 4.000 |
| 43203 | Ethylene | 61 | 0.58 | 7.31 | 1.41 | 2.063 | 1.629 |
| 43204 | Propane | 61 | 1.14 | 70.69 | 5.27 | 10.605 | 12.443 |
| 43205 | Propylene | 61 | 0.33 | 3.61 | 0.78 | 1.082 | 0.781 |
| 43206 | Acetylene | 61 | 0.37 | 5.97 | 1.11 | 1.460 | 1.094 |
| 43212 | n-butane | 61 | 0.43 | 10.49 | 2.16 | 3.291 | 2.655 |
| 43214 | Isobutane | 61 | 0.30 | 4.74 | 1.20 | 1.511 | 1.009 |
| 43216 | t-2-butene | 61 | 0.03 | 0.34 | 0.08 | 0.119 | 0.087 |
| 43217 | c-2-butene | 61 | 0.03 | 0.28 | 0.07 | 0.106 | 0.070 |
| 43220 | n-pentane | 61 | 0.59 | 6.44 | 1.56 | 1.965 | 1.267 |
| 43221 | Isopentane | 61 | 1.12 | 16.60 | 2.83 | 3.499 | 2.736 |
| 43224 | 1-pentene | 61 | 0.13 | 0.75 | 0.28 | 0.315 | 0.138 |
| 43226 | t-2-pentene | 61 | 0.04 | 1.38 | 0.21 | 0.266 | 0.210 |
| 43227 | c-2-pentene | 61 | 0.00 | 0.63 | 0.08 | 0.110 | 0.095 |
| 43230 | 3-methylpentane | 61 | 0.24 | 2.50 | 0.55 | 0.708 | 0.482 |
| 43231 | n-hexane | 61 | 0.25 | 2.86 | 0.71 | 0.876 | 0.589 |
| 43232 | n-heptane | 61 | 0.15 | 3.37 | 0.42 | 0.539 | 0.480 |
| 43233 | n-octane | 61 | 0.07 | 1.47 | 0.15 | 0.230 | 0.250 |
| 43235 | n-nonane | 61 | 0.07 | 1.03 | 0.15 | 0.199 | 0.167 |
| 43238 | n-decane | 61 | 0.05 | 0.87 | 0.14 | 0.190 | 0.144 |
| 43242 | Cyclopentane | 61 | 0.06 | 0.59 | 0.13 | 0.171 | 0.115 |
| 43243 | Isoprene | 61 | 0.00 | 3.90 | 0.33 | 0.914 | 1.122 |
| 43244 | 2,2-dimethylbutane | 61 | 0.05 | 0.58 | 0.15 | 0.182 | 0.111 |
| 43245 | 1-Hexene | 61 | 0.00 | 0.40 | 0.07 | 0.092 | 0.065 |
| 43247 | 2,4-dimethylpentane | 61 | 0.04 | 0.72 | 0.14 | 0.180 | 0.133 |
| 43248 | Cyclohexane | 61 | 0.04 | 0.63 | 0.14 | 0.185 | 0.121 |
| 43249 | 3-methylhexane | 61 | 0.17 | 1.56 | 0.53 | 0.618 | 0.310 |
| 43250 | 2,2,4-trimethylpentane | 61 | 0.18 | 3.88 | 0.62 | 0.896 | 0.725 |
| 43252 | 2,3,4-trimethylpentane | 61 | 0.06 | 1.07 | 0.23 | 0.298 | 0.207 |
| 43253 | 3-methylheptane | 61 | 0.05 | 0.54 | 0.10 | 0.140 | 0.107 |
| 43261 | Methylcyclohexane | 61 | 0.00 | 3.02 | 0.19 | 0.275 | 0.387 |
| 43262 | Methylcyclopentane | 61 | 0.15 | 1.55 | 0.45 | 0.552 | 0.302 |
| 43263 | 2-methylhexane | 61 | 0.24 | 2.49 | 0.70 | 0.798 | 0.450 |
| 43280 | 1-butene | 61 | 0.00 | 0.72 | 0.17 | 0.230 | 0.164 |
| 43284 | 2,3-dimethylbutane | 61 | 0.09 | 1.50 | 0.23 | 0.297 | 0.244 |
| 43285 | 2-methylpentane | 61 | 0.35 | 4.29 | 1.45 | 1.567 | 0.836 |
| 43291 | 2,3-dimethylpentane | 61 | 0.11 | 1.09 | 0.23 | 0.293 | 0.201 |
| 43954 | n-undecane | 61 | 0.00 | 0.48 | 0.11 | 0.140 | 0.089 |
| 43960 | 2-methylheptane | 61 | 0.07 | 0.81 | 0.15 | 0.193 | 0.129 |
| 45109 | m/p-xylene | 61 | 0.30 | 5.71 | 0.94 | 1.298 | 1.002 |
| 45201 | Benzene | 61 | 0.30 | 3.15 | 0.92 | 1.178 | 0.692 |
| 45202 | Toluene | 61 | 0.34 | 8.46 | 2.17 | 2.692 | 1.812 |
| 45203 | Ethylbenzene | 61 | 0.11 | 1.76 | 0.35 | 0.428 | 0.310 |
| 45204 | o-xylene | 61 | 0.11 | 1.84 | 0.33 | 0.438 | 0.344 |
| 45207 | 1,3,5-trimethylbenzene | 61 | 0.00 | 0.71 | 0.14 | 0.190 | 0.149 |
| 45208 | 1,2,4-trimethylbenzene | 61 | 0.13 | 1.75 | 0.37 | 0.486 | 0.368 |
| 45209 | n-propylbenzene | 61 | 0.00 | 0.37 | 0.09 | 0.119 | 0.076 |
| 45210 | Isopropylbenzene | 61 | 0.00 | 0.19 | 0.07 | 0.074 | 0.040 |
| 45211 | o-ethyltoluene | 61 | 0.00 | 0.25 | 0.06 | 0.069 | 0.050 |
| 45212 | m-ethyltoluene | 61 | 0.07 | 1.19 | 0.24 | 0.315 | 0.240 |
| 45213 | p-ethyltoluene | 61 | 0.05 | 1.02 | 0.20 | 0.258 | 0.188 |
| 45218 | m-diethylbenzene | 61 | 0.00 | 0.28 | 0.05 | 0.059 | 0.040 |
| 45219 | p-diethylbenzene | 61 | 0.03 | 0.38 | 0.07 | 0.085 | 0.056 |
| 45220 | Styrene | 61 | 0.05 | 1.35 | 0.28 | 0.325 | 0.216 |
| 45225 | 1,2,3-trimethylbenzene | 61 | 0.04 | 0.47 | 0.11 | 0.140 | 0.100 |
| 43000 | PAMHC | 61 | 17.31 | 173.14 | 39.32 | 52.051 | 32.970 |
| 43102 | TNMOC | 61 | 27.44 | 207.28 | 70.17 | 80.190 | 39.783 |

**2009 Average Concentration of Detectable Volatile Ozone Precursors
Photochemical Assessment Monitoring Station Additional VOC PAMS Sampling -
Lee District Park**

Concentrations are in ppbC
(non detects are counted as zeros for statistical purposes)

| Parameter | Compound Name | Num | Minimum | Maximum | Median | Average | StDev |
|--------------|------------------------|-----|---------|---------|--------|---------|--------|
| 43141 | n-dodecane | 60 | 0.05 | 0.81 | 0.11 | 0.159 | 0.143 |
| 43202 | Ethane | 60 | 2.50 | 49.13 | 6.00 | 7.531 | 6.600 |
| 43203 | Ethylene | 60 | 0.42 | 7.82 | 1.36 | 1.665 | 1.109 |
| 43204 | Propane | 60 | 1.74 | 17.39 | 3.89 | 4.951 | 3.093 |
| 43205 | Propylene | 60 | 0.22 | 2.83 | 0.62 | 0.708 | 0.407 |
| 43206 | Acetylene | 60 | 0.44 | 6.37 | 1.17 | 1.334 | 0.919 |
| 43212 | n-butane | 60 | 0.77 | 8.90 | 2.07 | 2.713 | 1.940 |
| 43214 | Isobutane | 60 | 0.36 | 3.47 | 1.00 | 1.172 | 0.636 |
| 43216 | t-2-butene | 60 | 0.02 | 0.22 | 0.04 | 0.051 | 0.031 |
| 43217 | c-2-butene | 60 | 0.02 | 0.19 | 0.04 | 0.051 | 0.029 |
| 43220 | n-pentane | 60 | 0.56 | 2.76 | 1.27 | 1.358 | 0.543 |
| 43221 | Isopentane | 60 | 0.61 | 4.33 | 1.69 | 1.954 | 0.894 |
| 43224 | 1-pentene | 60 | 0.07 | 0.33 | 0.14 | 0.154 | 0.048 |
| 43226 | t-2-pentene | 60 | 0.04 | 0.24 | 0.09 | 0.106 | 0.052 |
| 43227 | c-2-pentene | 60 | 0.03 | 0.11 | 0.05 | 0.051 | 0.021 |
| 43230 | 3-methylpentane | 60 | 0.13 | 0.96 | 0.43 | 0.461 | 0.192 |
| 43231 | n-hexane | 60 | 0.15 | 1.21 | 0.51 | 0.561 | 0.234 |
| 43232 | n-heptane | 60 | 0.08 | 1.13 | 0.26 | 0.290 | 0.164 |
| 43233 | n-octane | 60 | 0.04 | 0.32 | 0.14 | 0.144 | 0.063 |
| 43235 | n-nonane | 60 | 0.03 | 0.41 | 0.12 | 0.136 | 0.069 |
| 43238 | n-decane | 60 | 0.04 | 0.55 | 0.12 | 0.145 | 0.085 |
| 43242 | Cyclopentane | 60 | 0.05 | 0.21 | 0.09 | 0.104 | 0.041 |
| 43243 | Isoprene | 60 | 0.04 | 18.93 | 0.21 | 2.136 | 3.654 |
| 43244 | 2,2-dimethylbutane | 60 | 0.06 | 0.28 | 0.13 | 0.132 | 0.046 |
| 43245 | 1-Hexene | 60 | 0.03 | 0.13 | 0.06 | 0.060 | 0.020 |
| 43247 | 2,4-dimethylpentane | 60 | 0.00 | 0.24 | 0.12 | 0.120 | 0.055 |
| 43248 | Cyclohexane | 60 | 0.00 | 0.26 | 0.11 | 0.111 | 0.050 |
| 43249 | 3-methylhexane | 60 | 0.10 | 0.78 | 0.32 | 0.341 | 0.127 |
| 43250 | 2,2,4-trimethylpentane | 60 | 0.10 | 1.46 | 0.53 | 0.606 | 0.314 |
| 43252 | 2,3,4-trimethylpentane | 60 | 0.06 | 0.60 | 0.19 | 0.227 | 0.111 |
| 43253 | 3-methylheptane | 60 | 0.00 | 0.20 | 0.08 | 0.087 | 0.041 |
| 43261 | Methylcyclohexane | 60 | 0.07 | 0.40 | 0.15 | 0.162 | 0.071 |
| 43262 | Methylcyclopentane | 60 | 0.10 | 0.71 | 0.33 | 0.361 | 0.132 |
| 43263 | 2-methylhexane | 60 | 0.05 | 0.94 | 0.38 | 0.402 | 0.175 |
| 43280 | 1-butene | 60 | 0.05 | 0.47 | 0.11 | 0.135 | 0.072 |
| 43284 | 2,3-dimethylbutane | 60 | 0.06 | 0.40 | 0.18 | 0.187 | 0.082 |
| 43285 | 2-methylpentane | 60 | 0.16 | 2.09 | 1.16 | 1.127 | 0.470 |
| 43291 | 2,3-dimethylpentane | 60 | 0.05 | 0.32 | 0.15 | 0.153 | 0.062 |
| 43954 | n-undecane | 60 | 0.00 | 1.01 | 0.23 | 0.235 | 0.135 |
| 43960 | 2-methylheptane | 60 | 0.04 | 0.25 | 0.11 | 0.121 | 0.047 |
| 45109 | m/p-xylene | 60 | 0.19 | 1.81 | 0.72 | 0.774 | 0.381 |
| 45201 | Benzene | 60 | 0.36 | 2.83 | 0.85 | 0.935 | 0.436 |
| 45202 | Toluene | 60 | 0.38 | 3.79 | 1.68 | 1.816 | 0.807 |
| 45203 | Ethylbenzene | 60 | 0.07 | 0.59 | 0.25 | 0.269 | 0.123 |
| 45204 | o-xylene | 60 | 0.07 | 0.66 | 0.23 | 0.261 | 0.128 |
| 45207 | 1,3,5-trimethylbenzene | 60 | 0.04 | 0.32 | 0.11 | 0.130 | 0.067 |
| 45208 | 1,2,4-trimethylbenzene | 60 | 0.07 | 0.71 | 0.28 | 0.311 | 0.156 |
| 45209 | n-propylbenzene | 60 | 0.00 | 0.26 | 0.07 | 0.079 | 0.039 |
| 45210 | Isopropylbenzene | 60 | 0.02 | 0.18 | 0.06 | 0.060 | 0.027 |
| 45211 | o-ethyltoluene | 60 | 0.00 | 0.25 | 0.05 | 0.072 | 0.054 |
| 45212 | m-ethyltoluene | 60 | 0.07 | 0.50 | 0.19 | 0.212 | 0.103 |
| 45213 | p-ethyltoluene | 60 | 0.06 | 0.50 | 0.18 | 0.196 | 0.100 |
| 45218 | m-diethylbenzene | 60 | 0.02 | 0.37 | 0.04 | 0.049 | 0.046 |
| 45219 | p-diethylbenzene | 60 | 0.03 | 0.99 | 0.07 | 0.084 | 0.122 |
| 45220 | Styrene | 60 | 0.03 | 0.47 | 0.22 | 0.229 | 0.106 |
| 45225 | 1,2,3-trimethylbenzene | 60 | 0.00 | 0.53 | 0.08 | 0.102 | 0.074 |
| 43000 | PAMHC | 60 | 17.42 | 103.94 | 33.07 | 38.075 | 16.574 |
| 43102 | TNMOC | 60 | 22.50 | 133.10 | 55.77 | 60.072 | 21.837 |

In 2009, the Office of Air Quality Monitoring (AQM) of the Department of Environmental Quality (DEQ) operated an Air Toxics Monitoring Network (ATMN). The ATMN consists of three separate monitoring programs. The Urban Air Toxics Monitoring Program (UATM), The National Air Toxics Trend Stations Program (NATTS), and The Community Assessment Monitoring Program (CAMP).

The UATM program consisted of three sites that were located at the Carter G. Woodson Middle School in Hopewell, DEQ Tidewater Regional Office (TRO) in Virginia Beach, and Lee District Park in Fairfax County. Sampling at these sites consisted of Volatile Organic Compounds (VOC), Carbonyls, and Total Suspended Particulate (TSP) Metals.

The UATM sites had a sampling schedule consisting of 24-hour samples collected every 6th day. Data from these sites will be used to characterize air toxics concentrations in the respective urban areas.

AQM used the manual method for collecting ambient air samples for VOC analysis. Whole air samples were collected using evacuated Silco^T canisters and RMESI (RM Environmental Systems, Inc.) air samplers. Each sample was analyzed by the Maryland Department of the Environment, Air and Radiation Management Administration, using a Gas Chromatograph equipped with a Mass Selective Detector, using method TO15.

Carbonyls were collected on DNPH (2,4-Dinitrophenylhydrazine) treated sorbent tubes using ATEC 8000 cartridge samplers. Samples were analyzed by the Philadelphia Health Department using a Liquid Chromatographic procedure, using method TO11A.

The Metals were collected using a high volume Total Suspended Particulate (TSP) sampler. Samples were analyzed by the Division of Consolidated Laboratory Services (DCLS). Analysis utilized inductively coupled plasma mass spectrometry (ICP-MS) using method IO-3.1 and IO-3.5.

Detailed data collected at these sites in 2009 are available upon written request to the Virginia Department of Environmental Quality, Office of Air Quality Monitoring.

The NATTS program operated one station located at the MathScience Innovation Center (MSIC) in Henrico County.

The NATTS site had a sampling schedule consisting of 24-hour samples collected every 6th day. Data from these sites will be entered into the EPA's data system and used along with the rest of the data from all of the NATTS sites nationally.

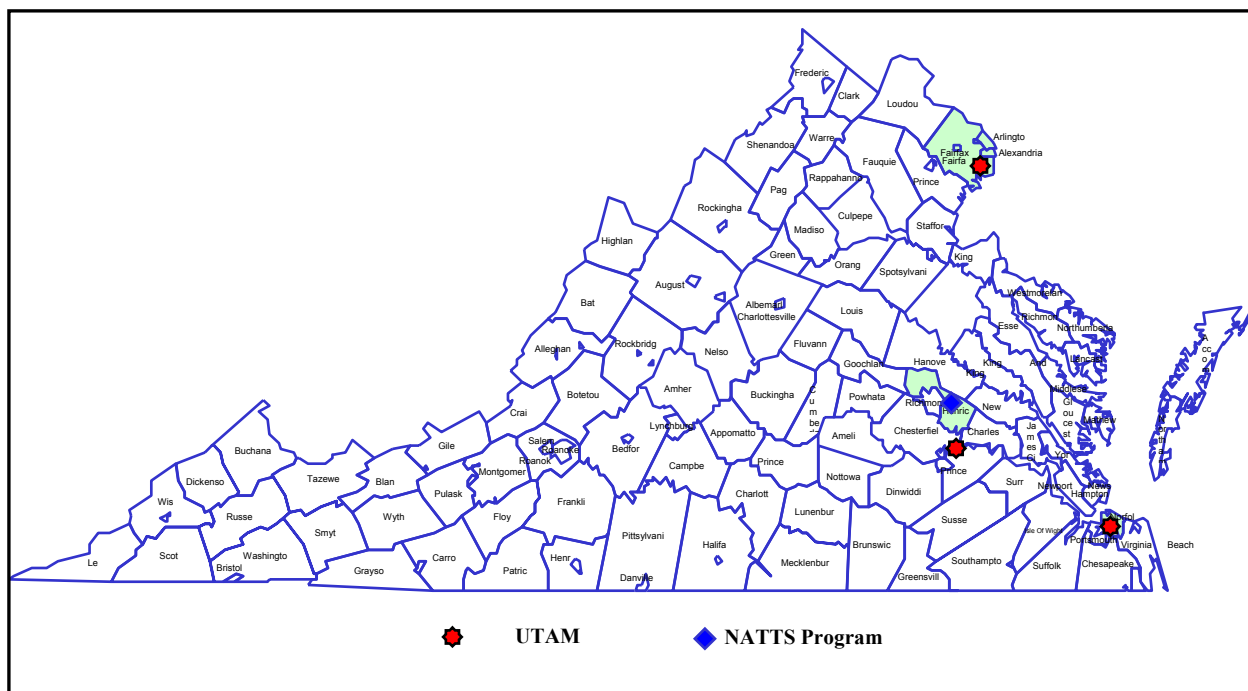
AQM used the manual method for collecting ambient air samples for VOC analysis. Whole air samples were collected using evacuated Silco^T canisters and RMESI (RM Environmental Systems, Inc.) air samplers. Each sample was analyzed by the Division of Consolidated Laboratory Services (DCLS), using a Gas Chromatograph equipped with a Mass Selective Detector, utilizing method TO15.

Carbonyls were collected on DNPH (2,4-Dinitrophenylhydrazine) treated sorbent tubes using ATEC 8000 cartridge samplers. Samples were analyzed by DCLS using a Liquid Chromatographic procedure, and the TO11A method.

The Metals were collected using a high volume 10 micron Particulate Matter (PM10) sampler. Samples were analyzed by the DCLS. Analysis utilized Inductively Coupled Plasma Mass Spectrometry (ICP-MS) using method IO-3.1 and IO-3.5.

Detailed data collected at this site in 2009 are available upon written request to the Virginia Department of Environmental Quality, Office of Air Quality Monitoring.

The Community Air Monitoring Program (CAMP) consists of special studies undertaken when conditions warrant. The locations and target compounds monitored are based on specific conditions and needs. The reports from these studies are published independently of this annual report.



Air Toxics Monitoring Network

Detectable VOC in 24-Hour Canister Samples (UATM)
GC/MSD - Carter G. Woodson Middle School, Hopewell, VA
January 1 to December 31, 2009 - Concentrations are in ppbV
(non detects are counted as zeros for statistical purposes)

| Parameter | Compound Name | Num | Minimum | Maximum | Median | Average | StDev |
|-----------|----------------------------------|-----|---------|---------|--------|---------|-------|
| 42153 | Carbon Disulfide | 61 | 0.01 | 0.13 | 0.02 | 0.022 | 0.019 |
| 43207 | Freon 113 | 61 | 0.08 | 0.27 | 0.10 | 0.111 | 0.031 |
| 43208 | Freon 114 | 61 | 0.02 | 0.04 | 0.02 | 0.022 | 0.005 |
| 43209 | Ethyl Acetate | 61 | 0.00 | 0.56 | 0.01 | 0.037 | 0.081 |
| 43218 | 1,3-Butadiene | 61 | 0.01 | 0.29 | 0.02 | 0.054 | 0.067 |
| 43231 | Hexane | 61 | 0.03 | 0.95 | 0.10 | 0.161 | 0.179 |
| 43232 | Heptane | 61 | 0.01 | 0.36 | 0.04 | 0.062 | 0.065 |
| 43248 | Cyclohexane | 61 | 0.00 | 0.17 | 0.02 | 0.052 | 0.047 |
| 43312 | Isopropyl Alcohol | 61 | 0.13 | 2.78 | 0.48 | 0.681 | 0.533 |
| 43372 | 2-Methoxy-2-Methyl-Propane | 61 | 0.00 | 0.23 | 0.00 | 0.019 | 0.048 |
| 43505 | Acrolein | 61 | 0.00 | 2.94 | 0.15 | 0.252 | 0.443 |
| 43551 | Acetone | 61 | 1.29 | 8.01 | 2.91 | 3.353 | 1.535 |
| 43552 | Methyl ethyl Ketone (2-butanone) | 61 | 0.02 | 1.18 | 0.24 | 0.254 | 0.167 |
| 43559 | Methyl butyl Ketone (2-hexanone) | 61 | 0.00 | 0.08 | 0.00 | 0.006 | 0.014 |
| 43560 | Methyl isobutyl Ketone | 61 | 0.00 | 0.07 | 0.00 | 0.005 | 0.014 |
| 43702 | Acetonitrile | 30 | 0.00 | 3.82 | 0.36 | 0.592 | 0.740 |
| 43704 | Acrylonitrile | 61 | 0.00 | 0.13 | 0.04 | 0.049 | 0.031 |
| 43801 | Chloromethane | 61 | 0.54 | 2.06 | 0.65 | 0.671 | 0.187 |
| 43802 | Dichloromethane | 61 | 0.13 | 0.66 | 0.24 | 0.249 | 0.074 |
| 43803 | Chloroform | 61 | 0.01 | 0.05 | 0.02 | 0.025 | 0.010 |
| 43804 | Carbon Tetrachloride | 61 | 0.08 | 0.13 | 0.11 | 0.109 | 0.011 |
| 43806 | Bromoform (Tribromomethane) | 61 | 0.00 | 0.03 | 0.00 | 0.003 | 0.006 |
| 43811 | Trichlorofluoromethane | 61 | 0.25 | 0.32 | 0.29 | 0.288 | 0.013 |
| 43812 | Chloroethane | 61 | 0.00 | 1.20 | 0.01 | 0.043 | 0.156 |
| 43813 | 1,1-Dichloroethane | 61 | 0.00 | 0.02 | 0.00 | 0.002 | 0.005 |
| 43814 | Methyl chloroform | 61 | 0.01 | 0.06 | 0.01 | 0.015 | 0.008 |
| 43815 | Ethylene dichloride | 61 | 0.01 | 0.12 | 0.02 | 0.022 | 0.015 |
| 43817 | Tetrachloroethylene | 61 | 0.00 | 0.21 | 0.01 | 0.024 | 0.029 |
| 43818 | 1,1,2,2-Tetrachloroethane | 61 | 0.00 | 0.02 | 0.00 | 0.003 | 0.006 |
| 43819 | Bromomethane | 61 | 0.00 | 0.03 | 0.01 | 0.012 | 0.006 |
| 43820 | 1,1,2-Trichloroethane | 61 | 0.00 | 0.02 | 0.00 | 0.003 | 0.006 |
| 43823 | Dichlorodifluoromethane | 61 | 0.56 | 0.69 | 0.62 | 0.630 | 0.027 |
| 43824 | Trichloroethylene | 61 | 0.00 | 0.03 | 0.01 | 0.008 | 0.009 |
| 43826 | 1,1-Dichloroethylene | 61 | 0.00 | 0.03 | 0.00 | 0.005 | 0.007 |
| 43828 | Bromodichloromethane | 61 | 0.00 | 0.03 | 0.00 | 0.004 | 0.007 |
| 43829 | 1,2-Dichloropropane | 61 | 0.00 | 0.04 | 0.00 | 0.005 | 0.009 |
| 43830 | trans-1,3-Dichloropropylene | 61 | 0.00 | 0.07 | 0.00 | 0.003 | 0.010 |
| 43831 | cis-1,3-Dichloropropylene | 61 | 0.00 | 0.11 | 0.00 | 0.004 | 0.015 |
| 43832 | Dibromochloromethane | 61 | 0.00 | 0.02 | 0.00 | 0.002 | 0.005 |
| 43838 | Trans-1,2-Dichloroethene | 61 | 0.00 | 0.01 | 0.00 | 0.001 | 0.003 |
| 43839 | cis-1,2-Dichloroethene | 61 | 0.00 | 0.02 | 0.00 | 0.002 | 0.005 |
| 43843 | Ethylene Dibromide | 61 | 0.00 | 0.05 | 0.00 | 0.004 | 0.009 |
| 43844 | Hexachlorobutadiene | 61 | 0.01 | 0.07 | 0.01 | 0.013 | 0.008 |
| 43860 | Vinyl Chloride | 61 | 0.00 | 0.02 | 0.00 | 0.002 | 0.005 |

Detectable VOC in 24-Hour Canister Samples (UATM)
GC/MSD - Carter G. Woodson Middle School, Hopewell, VA, cont.
January 1 to December 31, 2009 - Concentrations are in ppbV
 (non detects are counted as zeros for statistical purposes)

| Parameter | Compound Name | Num | Minimum | Maximum | Median | Average | StDev |
|-----------|------------------------|-----|---------|---------|--------|---------|-------|
| 45109 | m/p-Xylene | 61 | 0.01 | 0.79 | 0.05 | 0.106 | 0.150 |
| 45201 | Benzene | 61 | 0.05 | 0.95 | 0.18 | 0.238 | 0.201 |
| 45202 | Toluene | 61 | 0.07 | 3.52 | 0.18 | 0.338 | 0.515 |
| 45203 | Ethylbenzene | 61 | 0.01 | 0.28 | 0.02 | 0.044 | 0.052 |
| 45204 | o-Xylene | 61 | 0.00 | 0.32 | 0.02 | 0.042 | 0.058 |
| 45207 | 1,3,5-Trimethylbenzene | 61 | 0.00 | 0.08 | 0.01 | 0.016 | 0.016 |
| 45208 | 1,2,4-Trimethylbenzene | 61 | 0.01 | 0.34 | 0.03 | 0.059 | 0.070 |
| 45213 | p-Ethyltoluene | 61 | 0.00 | 0.31 | 0.01 | 0.027 | 0.045 |
| 45220 | Styrene | 61 | 0.00 | 0.19 | 0.02 | 0.034 | 0.032 |
| 45801 | Chlorobenzene | 61 | 0.00 | 0.11 | 0.00 | 0.009 | 0.018 |
| 45805 | 1,2-Dichlorobenzene | 61 | 0.00 | 0.09 | 0.00 | 0.008 | 0.015 |
| 45806 | 1,3-Dichlorobenzene | 61 | 0.00 | 0.10 | 0.00 | 0.007 | 0.017 |
| 45807 | 1,4-Dichlorobenzene | 61 | 0.00 | 0.11 | 0.01 | 0.020 | 0.021 |
| 45809 | Benzyl Chloride | 61 | 0.00 | 0.07 | 0.00 | 0.007 | 0.014 |
| 45810 | 1,2,4-Trichlorobenzene | 61 | 0.00 | 0.06 | 0.01 | 0.015 | 0.012 |
| 46401 | Tetrahydrofuran | 61 | 0.00 | 0.07 | 0.00 | 0.010 | 0.017 |

Detectable VOC in 24-Hour Canister Samples (UATM)
GC/MSD - Tidewater Regional Office (TRO) – Va. Beach, VA
January 1 to December 31, 2009— Concentrations are in ppbV
(non detects are counted as zeros for statistical purposes)

| Parameter | Compound Name | Num | Minimum | Maximum | Median | Average | StDev |
|-----------|----------------------------------|-----|---------|---------|--------|---------|-------|
| 42153 | Carbon Disulfide | 61 | 0.01 | 0.26 | 0.05 | 0.070 | 0.063 |
| 43207 | Freon 113 | 61 | 0.09 | 0.12 | 0.10 | 0.097 | 0.008 |
| 43208 | Freon 114 | 61 | 0.02 | 0.04 | 0.02 | 0.022 | 0.005 |
| 43209 | Ethyl Acetate | 61 | 0.00 | 0.08 | 0.01 | 0.014 | 0.019 |
| 43218 | 1,3-Butadiene | 61 | 0.00 | 0.21 | 0.03 | 0.050 | 0.046 |
| 43231 | Hexane | 61 | 0.05 | 0.58 | 0.12 | 0.159 | 0.116 |
| 43232 | Heptane | 61 | 0.00 | 0.56 | 0.05 | 0.067 | 0.074 |
| 43248 | Cyclohexane | 61 | 0.00 | 0.13 | 0.03 | 0.051 | 0.043 |
| 43312 | Isopropyl Alcohol | 61 | 0.13 | 2.89 | 0.25 | 0.394 | 0.447 |
| 43372 | 2-Methoxy-2-Methyl-Propane | 61 | 0.00 | 0.02 | 0.00 | 0.003 | 0.006 |
| 43505 | Acrolein | 61 | 0.00 | 1.26 | 0.16 | 0.241 | 0.257 |
| 43551 | Acetone | 61 | 1.47 | 10.47 | 4.16 | 4.393 | 1.856 |
| 43552 | Methyl ethyl Ketone (2-butanone) | 61 | 0.11 | 0.93 | 0.29 | 0.322 | 0.162 |
| 43559 | Methyl butyl Ketone (2-hexanone) | 61 | 0.00 | 0.26 | 0.01 | 0.018 | 0.035 |
| 43560 | Methyl isobutyl Ketone | 61 | 0.00 | 0.22 | 0.01 | 0.013 | 0.030 |
| 43702 | Acetonitrile | 30 | 0.32 | 6.87 | 0.95 | 1.341 | 1.385 |
| 43704 | Acrylonitrile | 61 | 0.00 | 0.16 | 0.04 | 0.046 | 0.025 |
| 43801 | Chloromethane | 61 | 0.53 | 2.08 | 0.69 | 0.727 | 0.202 |
| 43802 | Dichloromethane | 61 | 0.16 | 0.39 | 0.25 | 0.254 | 0.050 |
| 43803 | Chloroform | 61 | 0.01 | 0.05 | 0.02 | 0.024 | 0.010 |
| 43804 | Carbon Tetrachloride | 61 | 0.09 | 0.14 | 0.11 | 0.110 | 0.011 |
| 43806 | Bromoform (Tribromomethane) | 61 | 0.00 | 0.02 | 0.00 | 0.001 | 0.004 |
| 43811 | Trichlorofluoromethane | 61 | 0.26 | 0.35 | 0.29 | 0.292 | 0.016 |
| 43812 | Chloroethane | 61 | 0.00 | 0.08 | 0.01 | 0.013 | 0.016 |
| 43813 | 1,1-Dichloroethane | 61 | 0.00 | 0.02 | 0.00 | 0.002 | 0.005 |
| 43814 | Methyl chloroform | 61 | 0.01 | 0.03 | 0.01 | 0.012 | 0.005 |
| 43815 | Ethylene dichloride | 61 | 0.01 | 0.04 | 0.02 | 0.020 | 0.007 |
| 43817 | Tetrachloroethylene | 61 | 0.01 | 1.02 | 0.07 | 0.119 | 0.162 |
| 43818 | 1,1,2,2-Tetrachloroethane | 61 | 0.00 | 0.02 | 0.00 | 0.003 | 0.006 |
| 43819 | Bromomethane | 61 | 0.00 | 0.04 | 0.01 | 0.012 | 0.007 |
| 43820 | 1,1,2-Trichloroethane | 61 | 0.00 | 0.02 | 0.00 | 0.002 | 0.005 |
| 43823 | Dichlorodifluoromethane | 61 | 0.60 | 0.71 | 0.64 | 0.644 | 0.025 |
| 43824 | Trichloroethylene | 61 | 0.00 | 0.04 | 0.00 | 0.007 | 0.010 |
| 43826 | 1,1-Dichloroethylene | 61 | 0.00 | 0.02 | 0.00 | 0.005 | 0.006 |
| 43828 | Bromodichloromethane | 61 | 0.00 | 0.03 | 0.00 | 0.004 | 0.007 |
| 43829 | 1,2-Dichloropropane | 61 | 0.00 | 0.04 | 0.00 | 0.004 | 0.009 |
| 43830 | trans-1,3-Dichloropropylene | 61 | 0.00 | 0.01 | 0.00 | 0.001 | 0.003 |
| 43831 | cis-1,3-Dichloropropylene | 61 | 0.00 | 0.02 | 0.00 | 0.001 | 0.004 |
| 43832 | Dibromochloromethane | 61 | 0.00 | 0.02 | 0.00 | 0.002 | 0.005 |
| 43838 | Trans-1,2-Dichloroethene | 61 | 0.00 | 0.02 | 0.00 | 0.002 | 0.006 |
| 43839 | cis-1,2-Dichloroethene | 61 | 0.00 | 0.02 | 0.00 | 0.002 | 0.006 |
| 43843 | Ethylene Dibromide | 61 | 0.00 | 0.02 | 0.00 | 0.001 | 0.004 |
| 43844 | Hexachlorobutadiene | 61 | 0.01 | 0.02 | 0.01 | 0.011 | 0.003 |
| 43860 | Vinyl Chloride | 61 | 0.00 | 0.03 | 0.00 | 0.003 | 0.007 |

Detectable VOC in 24-Hour Canister Samples (UATM)
GC/MSD - Tidewater Regional Office (TRO) – Va. Beach, VA, cont.
January 1 to December 31, 2009– Concentrations are in ppbV
(non detects are counted as zeros for statistical purposes)

| Parameter | Compound Name | Num | Minimum | Maximum | Median | Average | StDev |
|-----------|------------------------|-----|---------|---------|--------|---------|-------|
| 45109 | m/p-Xylene | 61 | 0.01 | 0.61 | 0.07 | 0.104 | 0.109 |
| 45201 | Benzene | 61 | 0.07 | 0.67 | 0.18 | 0.226 | 0.135 |
| 45202 | Toluene | 61 | 0.03 | 1.43 | 0.23 | 0.310 | 0.263 |
| 45203 | Ethylbenzene | 61 | 0.01 | 0.19 | 0.03 | 0.041 | 0.035 |
| 45204 | o-Xylene | 61 | 0.01 | 0.21 | 0.03 | 0.041 | 0.040 |
| 45207 | 1,3,5-Trimethylbenzene | 61 | 0.00 | 0.06 | 0.01 | 0.013 | 0.011 |
| 45208 | 1,2,4-Trimethylbenzene | 61 | 0.01 | 0.24 | 0.04 | 0.051 | 0.043 |
| 45213 | p-Ethyltoluene | 61 | 0.00 | 0.21 | 0.01 | 0.020 | 0.030 |
| 45220 | Styrene | 61 | 0.01 | 0.07 | 0.03 | 0.027 | 0.014 |
| 45801 | Chlorobenzene | 61 | 0.00 | 0.02 | 0.00 | 0.004 | 0.006 |
| 45805 | 1,2-Dichlorobenzene | 61 | 0.00 | 0.02 | 0.00 | 0.004 | 0.006 |
| 45806 | 1,3-Dichlorobenzene | 61 | 0.00 | 0.01 | 0.00 | 0.002 | 0.004 |
| 45807 | 1,4-Dichlorobenzene | 61 | 0.00 | 0.02 | 0.01 | 0.009 | 0.006 |
| 45809 | Benzyl Chloride | 61 | 0.00 | 0.02 | 0.00 | 0.003 | 0.006 |
| 45810 | 1,2,4-Trichlorobenzene | 61 | 0.00 | 0.02 | 0.01 | 0.011 | 0.005 |
| 46401 | Tetrahydrofuran | 61 | 0.00 | 0.74 | 0.00 | 0.022 | 0.095 |

Detectable VOC in 24-Hour Canister Samples (UATM)
GC/MSD - Lee District Park - Fairfax County, VA
January 1 to December 31, 2009 - Concentrations are in ppbV
(non detects are counted as zeros for statistical purposes)

| Parameter | Compound Name | Num | Minimum | Maximum | Median | Average | StDev |
|-----------|----------------------------------|-----|---------|---------|--------|---------|-------|
| 42153 | Carbon Disulfide | 61 | 0.01 | 0.18 | 0.01 | 0.023 | 0.028 |
| 43207 | Freon 113 | 61 | 0.09 | 0.12 | 0.09 | 0.094 | 0.007 |
| 43208 | Freon 114 | 61 | 0.02 | 0.04 | 0.02 | 0.021 | 0.004 |
| 43209 | Ethyl Acetate | 61 | 0.00 | 0.11 | 0.00 | 0.014 | 0.026 |
| 43218 | 1,3-Butadiene | 61 | 0.00 | 0.14 | 0.02 | 0.030 | 0.025 |
| 43231 | Hexane | 61 | 0.03 | 0.25 | 0.09 | 0.097 | 0.045 |
| 43232 | Heptane | 61 | 0.01 | 0.11 | 0.03 | 0.034 | 0.020 |
| 43248 | Cyclohexane | 61 | 0.00 | 0.13 | 0.02 | 0.046 | 0.046 |
| 43312 | Isopropyl Alcohol | 61 | 0.10 | 28.47 | 0.25 | 1.014 | 3.806 |
| 43372 | 2-Methoxy-2-Methyl-Propane | 61 | 0.00 | 0.01 | 0.00 | 0.001 | 0.004 |
| 43505 | Acrolein | 61 | 0.00 | 0.88 | 0.14 | 0.168 | 0.150 |
| 43551 | Acetone | 61 | 1.23 | 6.14 | 2.98 | 3.132 | 1.252 |
| 43552 | Methyl ethyl Ketone (2-butanone) | 61 | 0.04 | 0.52 | 0.20 | 0.239 | 0.128 |
| 43559 | Methyl butyl Ketone (2-hexanone) | 61 | 0.00 | 0.04 | 0.00 | 0.007 | 0.010 |
| 43560 | Methyl isobutyl Ketone | 61 | 0.00 | 0.03 | 0.00 | 0.004 | 0.007 |
| 43702 | Acetonitrile | 30 | 0.04 | 3.62 | 0.19 | 0.356 | 0.697 |
| 43704 | Acrylonitrile | 61 | 0.00 | 0.09 | 0.03 | 0.034 | 0.017 |
| 43801 | Chloromethane | 61 | 0.52 | 0.73 | 0.63 | 0.636 | 0.050 |
| 43802 | Dichloromethane | 61 | 0.09 | 0.36 | 0.23 | 0.239 | 0.057 |
| 43803 | Chloroform | 61 | 0.01 | 0.05 | 0.02 | 0.025 | 0.008 |
| 43804 | Carbon Tetrachloride | 61 | 0.04 | 0.13 | 0.11 | 0.104 | 0.015 |
| 43806 | Bromoform (Tribromomethane) | 61 | 0.00 | 0.01 | 0.00 | 0.001 | 0.003 |
| 43811 | Trichlorofluoromethane | 61 | 0.27 | 0.34 | 0.29 | 0.290 | 0.015 |
| 43812 | Chloroethane | 61 | 0.00 | 0.03 | 0.01 | 0.007 | 0.008 |
| 43813 | 1,1-Dichloroethane | 61 | 0.00 | 0.02 | 0.00 | 0.001 | 0.004 |
| 43814 | Methyl chloroform | 61 | 0.01 | 0.03 | 0.01 | 0.011 | 0.004 |
| 43815 | Ethylene dichloride | 61 | 0.00 | 0.03 | 0.02 | 0.018 | 0.006 |
| 43817 | Tetrachloroethylene | 61 | 0.01 | 0.09 | 0.03 | 0.028 | 0.016 |
| 43818 | 1,1,2,2-Tetrachloroethane | 61 | 0.00 | 0.01 | 0.00 | 0.001 | 0.003 |
| 43819 | Bromomethane | 61 | 0.00 | 0.02 | 0.01 | 0.011 | 0.005 |
| 43820 | 1,1,2-Trichloroethane | 61 | 0.00 | 0.01 | 0.00 | 0.001 | 0.003 |
| 43823 | Dichlorodifluoromethane | 61 | 0.59 | 0.71 | 0.63 | 0.636 | 0.028 |
| 43824 | Trichloroethylene | 61 | 0.00 | 0.02 | 0.01 | 0.010 | 0.006 |
| 43826 | 1,1-Dichloroethylene | 61 | 0.00 | 0.02 | 0.00 | 0.002 | 0.005 |
| 43828 | Bromodichloromethane | 61 | 0.00 | 0.02 | 0.00 | 0.002 | 0.005 |
| 43829 | 1,2-Dichloropropane | 61 | 0.00 | 0.02 | 0.00 | 0.002 | 0.005 |
| 43830 | trans-1,3-Dichlopropylene | 61 | 0.00 | 0.01 | 0.00 | 0.000 | 0.002 |
| 43831 | cis-1,3-Dichlopropylene | 61 | 0.00 | 0.01 | 0.00 | 0.001 | 0.003 |
| 43832 | Dibromochloromethane | 61 | 0.00 | 0.01 | 0.00 | 0.001 | 0.003 |
| 43838 | Trans-1,2-Dichloroethene | 61 | 0.00 | 0.01 | 0.00 | 0.001 | 0.002 |
| 43839 | cis-1,2-Dichloroethene | 61 | 0.00 | 0.01 | 0.00 | 0.001 | 0.003 |
| 43843 | Ethylene Dibromide | 61 | 0.00 | 0.01 | 0.00 | 0.001 | 0.003 |
| 43844 | Hexachlorobutadiene | 61 | 0.01 | 0.02 | 0.01 | 0.011 | 0.003 |
| 43860 | Vinyl Chloride | 61 | 0.00 | 0.02 | 0.00 | 0.001 | 0.004 |

Detectable VOC in 24-Hour Canister Samples (UATM)
GC/MSD - Lee District Park - Fairfax County, VA, cont.
January 1 to December 31, 2009 - Concentrations are in ppbV
(non detects are counted as zeros for statistical purposes)

| Parameter | Compound Name | Num | Minimum | Maximum | Median | Average | StDev |
|-----------|------------------------|-----|---------|---------|--------|---------|-------|
| 45109 | m/p-Xylene | 61 | 0.01 | 0.22 | 0.05 | 0.058 | 0.038 |
| 45201 | Benzene | 61 | 0.07 | 0.60 | 0.16 | 0.183 | 0.089 |
| 45202 | Toluene | 61 | 0.03 | 0.58 | 0.17 | 0.189 | 0.101 |
| 45203 | Ethylbenzene | 61 | 0.01 | 0.09 | 0.03 | 0.028 | 0.015 |
| 45204 | o-Xylene | 61 | 0.00 | 0.09 | 0.02 | 0.023 | 0.016 |
| 45207 | 1,3,5-Trimethylbenzene | 61 | 0.00 | 0.03 | 0.01 | 0.009 | 0.006 |
| 45208 | 1,2,4-Trimethylbenzene | 61 | 0.01 | 0.09 | 0.03 | 0.032 | 0.016 |
| 45213 | p-Ethyltoluene | 61 | 0.01 | 0.03 | 0.01 | 0.013 | 0.006 |
| 45220 | Styrene | 61 | 0.01 | 0.04 | 0.02 | 0.021 | 0.007 |
| 45801 | Chlorobenzene | 61 | 0.00 | 0.02 | 0.00 | 0.003 | 0.005 |
| 45805 | 1,2-Dichlorobenzene | 61 | 0.00 | 0.01 | 0.00 | 0.003 | 0.005 |
| 45806 | 1,3-Dichlorobenzene | 61 | 0.00 | 0.01 | 0.00 | 0.002 | 0.004 |
| 45807 | 1,4-Dichlorobenzene | 61 | 0.00 | 0.03 | 0.01 | 0.010 | 0.005 |
| 45809 | Benzyl Chloride | 61 | 0.00 | 0.01 | 0.00 | 0.002 | 0.004 |
| 45810 | 1,2,4-Trichlorobenzene | 61 | 0.00 | 0.03 | 0.01 | 0.011 | 0.006 |
| 46401 | Tetrahydrofuran | 61 | 0.00 | 0.02 | 0.00 | 0.001 | 0.004 |

UATM Sampling 2009 Summary Statistical Analysis 24 Hour Carbonyl

Concentrations are in ppbV
(non detects are counted as zeros for statistical purposes)

| Site | Parameter | Compound Name | Num | Minimum | Maximum | Median | Average | StDev |
|--|-----------|------------------------|-----|---------|---------|--------|---------|-------|
| Carter G. Woodson Middle School | 43502 | Formaldehyde | 61 | 0.08 | 7.40 | 2.67 | 3.170 | 1.697 |
| | 43503 | Acetaldehyde | 61 | 0.12 | 2.93 | 1.33 | 1.440 | 0.560 |
| | 43504 | Propionaldehyde | 61 | 0.06 | 0.86 | 0.37 | 0.379 | 0.147 |
| | 43551 | Acetone | 61 | 0.42 | 13.17 | 3.99 | 4.401 | 2.415 |
| | 43552 | Methyl Ethyl Ketone | 61 | 0.10 | 2.00 | 0.68 | 0.669 | 0.305 |
| | 43560 | Methyl Isobutyl Ketone | 61 | 0.00 | 0.19 | 0.00 | 0.018 | 0.038 |

| Site | Parameter | Compound Name | Num | Minimum | Maximum | Median | Average | StDev |
|----------------------------------|-----------|------------------------|-----|---------|---------|--------|---------|-------|
| Tidewater Regional Office | 43502 | Formaldehyde | 60 | 0.07 | 6.04 | 2.53 | 2.718 | 1.116 |
| | 43503 | Acetaldehyde | 60 | 0.25 | 2.75 | 1.19 | 1.324 | 0.584 |
| | 43504 | Propionaldehyde | 60 | 0.22 | 1.13 | 0.45 | 0.539 | 0.249 |
| | 43551 | Acetone | 60 | 0.00 | 7.69 | 3.36 | 3.307 | 2.004 |
| | 43552 | Methyl Ethyl Ketone | 60 | 0.13 | 1.31 | 0.63 | 0.574 | 0.276 |
| | 43560 | Methyl Isobutyl Ketone | 60 | 0.00 | 0.46 | 0.00 | 0.053 | 0.083 |

| Site | Parameter | Compound Name | Num | Minimum | Maximum | Median | Average | StDev |
|-----------------|-----------|------------------------|-----|---------|---------|--------|---------|-------|
| Lee Park | 43502 | Formaldehyde | 61 | 0.19 | 4.93 | 2.12 | 2.432 | 1.250 |
| | 43503 | Acetaldehyde | 61 | 0.42 | 4.67 | 1.68 | 1.866 | 0.857 |
| | 43504 | Propionaldehyde | 61 | 0.11 | 3.15 | 0.62 | 0.702 | 0.540 |
| | 43551 | Acetone | 61 | 1.25 | 9.07 | 3.94 | 4.118 | 1.731 |
| | 43552 | Methyl Ethyl Ketone | 61 | 0.18 | 1.46 | 0.69 | 0.771 | 0.303 |
| | 43560 | Methyl Isobutyl Ketone | 61 | 0.00 | 0.46 | 0.00 | 0.023 | 0.070 |

TSP Metals Sampling 2009 Summary Statistical Analysis (UATM)

Concentrations are in ng/m³
(non detects are counted as zeros for statistical purposes)

| Site | Parameter | Compound Name | Num | Minimum | Maximum | Median | Average | StDev |
|--|-----------|---------------|-----|---------|---------|--------|---------|-------|
| Carter G. Woodson Middle School | 12103 | Arsenic | 61 | 0.12 | 4.26 | 0.70 | 0.928 | 0.693 |
| | 12105 | Beryllium | 61 | 0.00 | 0.02 | 0.00 | 0.003 | 0.005 |
| | 12110 | Cadmium | 61 | 0.00 | 0.33 | 0.07 | 0.088 | 0.072 |
| | 12112 | Chromium | 61 | 1.53 | 5.63 | 2.09 | 2.286 | 0.629 |
| | 12128 | Lead | 61 | 0.73 | 19.43 | 2.67 | 3.349 | 2.813 |
| | 12132 | Manganese | 61 | 1.08 | 32.76 | 6.70 | 8.578 | 6.137 |
| | 12136 | Nickel | 61 | 0.55 | 6.18 | 1.14 | 1.356 | 0.903 |

| Site | Parameter | Compound Name | Num | Minimum | Maximum | Median | Average | StDev |
|----------------------------------|-----------|---------------|-----|---------|---------|--------|---------|--------|
| Tidewater Regional Office | 12103 | Arsenic | 58 | 0.65 | 35.84 | 3.77 | 5.354 | 5.532 |
| | 12105 | Beryllium | 58 | 0.00 | 0.23 | 0.00 | 0.013 | 0.037 |
| | 12110 | Cadmium | 58 | 0.00 | 0.99 | 0.35 | 0.405 | 0.260 |
| | 12112 | Chromium | 58 | 6.03 | 16.60 | 8.73 | 8.844 | 1.819 |
| | 12128 | Lead | 58 | 3.38 | 37.84 | 10.59 | 11.584 | 6.142 |
| | 12132 | Manganese | 58 | 5.07 | 166.10 | 19.83 | 25.910 | 23.103 |
| | 12136 | Nickel | 58 | 2.91 | 16.07 | 6.09 | 6.801 | 2.854 |

| Site | Parameter | Compound Name | Num | Minimum | Maximum | Median | Average | StDev |
|-----------------|-----------|---------------|-----|---------|---------|--------|---------|-------|
| Lee Park | 12103 | Arsenic | 60 | 0.15 | 2.38 | 0.85 | 0.884 | 0.444 |
| | 12105 | Beryllium | 60 | 0.00 | 0.09 | 0.00 | 0.001 | 0.002 |
| | 12110 | Cadmium | 60 | 0.00 | 0.56 | 0.12 | 0.129 | 0.103 |
| | 12112 | Chromium | 60 | 1.52 | 2.57 | 2.03 | 2.030 | 0.257 |
| | 12128 | Lead | 60 | 0.44 | 5.58 | 2.23 | 2.441 | 1.199 |
| | 12132 | Manganese | 60 | 0.98 | 39.53 | 5.57 | 6.918 | 5.894 |
| | 12136 | Nickel | 60 | 0.58 | 5.77 | 1.05 | 1.302 | 0.847 |

NATTS Sampling 2009 Summary Statistical Analysis

Concentrations are in $\mu\text{g}/\text{m}^3$

(non detects and negative values are counted as zeros for statistical purposes)

| Site | Parameter | Compound Name | Num | Minimum | Maximum | Median | Average | StDev |
|-------------------------------------|-----------|---------------|-----|---------|---------|--------|---------|-------|
| MathScience Innovation Center | 82103 | Arsenic | 60 | 0.11 | 2.34 | 0.61 | 0.730 | 0.477 |
| | 82105 | Beryllium | 60 | 0.00 | 0.05 | 0.00 | 0.003 | 0.008 |
| | 82110 | Cadmium | 60 | 0.02 | 0.74 | 0.16 | 0.200 | 0.150 |
| | 82112 | Chromium | 60 | 1.26 | 3.83 | 1.99 | 2.036 | 0.409 |
| | 82128 | Lead | 60 | 0.51 | 29.99 | 2.02 | 2.775 | 3.842 |
| | 82132 | Manganese | 60 | 0.43 | 10.38 | 2.50 | 2.736 | 1.635 |
| | 82136 | Nickel | 60 | 0.38 | 2.15 | 0.84 | 0.914 | 0.356 |

24 Hour Carbonyl

Concentrations are in $\mu\text{g}/\text{m}^3$

(non detects are counted as zeros for statistical purposes)

| Site | Parameter | Compound Name | Num | Minimum | Maximum | Median | Average | StDev |
|-------------------------------------|-----------|------------------------|-----|---------|---------|--------|---------|-------|
| MathScience Innovation Center | 43502 | Formaldehyde | 61 | 0.51 | 6.41 | 2.28 | 2.82 | 1.429 |
| | 43503 | Acetaldehyde | 61 | 0.55 | 2.72 | 1.25 | 1.34 | 0.441 |
| | 43504 | Propionaldehyde | 61 | 0.00 | 0.63 | 0.22 | 0.20 | 0.201 |
| | 43551 | Acetone | 61 | 1.78 | 11.10 | 3.54 | 4.04 | 1.727 |
| | 43552 | Methyl Ethyl Ketone | 61 | 0.00 | 1.02 | 0.48 | 0.49 | 0.207 |
| | 43560 | Methyl Isobutyl Ketone | 61 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |

Detectable VOC in 24-Hour Canister Samples (NATTS)
GC/MSD - MathScience Innovation Center - Henrico County, VA
January 1 to June 30, 2009- Concentrations are in ppbV
(non detects are counted as zeros for statistical purposes)

| Parameter | Compound Name | Num | Minimum | Maximum | Median | Average | StDev |
|-----------|-----------------------------|-----|---------|---------|--------|---------|-------|
| 43205 | Propylene | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 43207 | Freon 113 | 60 | 0.00 | 0.17 | 0.09 | 0.073 | 0.044 |
| 43208 | Freon 114 | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 43209 | Ethyl Acetate | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 43218 | 1,3-Butadiene | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 43231 | Hexane | 60 | 0.00 | 0.66 | 0.00 | 0.078 | 0.138 |
| 43232 | Heptane | 60 | 0.00 | 0.28 | 0.00 | 0.015 | 0.047 |
| 43248 | Cyclohexane | 60 | 0.00 | 0.12 | 0.00 | 0.004 | 0.019 |
| 43372 | Methyl Tert-Butyl Ether | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 43505 | Acrolein | 60 | 0.00 | 0.88 | 0.18 | 0.199 | 0.173 |
| 43702 | Acetonitrile | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 43704 | Acrylonitrile | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 43801 | Chloromethane | 60 | 0.25 | 0.85 | 0.51 | 0.509 | 0.118 |
| 43802 | Dichloromethane | 60 | 0.00 | 0.38 | 0.13 | 0.142 | 0.071 |
| 43803 | Chloroform | 60 | 0.00 | 0.09 | 0.00 | 0.002 | 0.012 |
| 43804 | Carbon Tetrachloride | 60 | 0.00 | 0.26 | 0.08 | 0.076 | 0.044 |
| 43806 | Bromoform (Tribromomethane) | 60 | 0.00 | 0.08 | 0.00 | 0.001 | 0.010 |
| 43811 | Trichlorofluoromethane | 60 | 0.10 | 1.10 | 0.25 | 0.279 | 0.158 |
| 43812 | Chloroethane | 60 | 0.00 | 0.08 | 0.00 | 0.001 | 0.010 |
| 43813 | 1,1-Dichloroethane | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 43814 | Methyl chloroform | 60 | 0.00 | 0.08 | 0.00 | 0.001 | 0.010 |
| 43815 | Ethylene dichloride | 60 | 0.00 | 0.10 | 0.00 | 0.003 | 0.017 |
| 43817 | Tetrachloroethylene | 60 | 0.00 | 0.11 | 0.00 | 0.008 | 0.025 |
| 43818 | 1,1,2,2-Tetrachloroethane | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 43819 | Bromomethane | 60 | 0.00 | 0.08 | 0.00 | 0.001 | 0.010 |
| 43820 | 1,1,2-Trichloroethane | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 43823 | Dichlorodifluoromethane | 60 | 0.22 | 0.75 | 0.50 | 0.481 | 0.094 |
| 43824 | Trichloroethylene | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 43826 | 1,1-Dichloroethylene | 60 | 0.00 | 0.08 | 0.00 | 0.001 | 0.010 |
| 43828 | Bromodichloromethane | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 43829 | 1,2-Dichloropropane | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 43830 | trans-1,3-Dichlopropylene | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 43831 | cis-1,3-Dichlopropylene | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 43832 | Dibromochloromethane | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 43838 | Trans-1,2-Dichloroethene | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 43839 | cis-1,2-Dichloroethene | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 43843 | Ethylene Dibromide | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 43844 | Hexachlorobutadiene | 60 | 0.00 | 0.20 | 0.00 | 0.003 | 0.026 |
| 43860 | Vinyl Chloride | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 45109 | M/P Xylene | 60 | 0.00 | 0.44 | 0.04 | 0.086 | 0.109 |
| 45201 | Benzene | 60 | 0.00 | 0.65 | 0.15 | 0.188 | 0.145 |
| 45202 | Toluene | 60 | 0.00 | 8.60 | 0.24 | 0.510 | 1.136 |
| 45203 | Ethylbenzene | 60 | 0.00 | 0.26 | 0.00 | 0.029 | 0.055 |
| 45204 | O-Xylene | 60 | 0.00 | 0.17 | 0.00 | 0.021 | 0.045 |
| 45207 | 1,3,5-Trimethylbenzene | 60 | 0.00 | 0.11 | 0.00 | 0.003 | 0.017 |
| 45208 | 1,2,4-Trimethylbenzene | 60 | 0.00 | 0.53 | 0.00 | 0.050 | 0.099 |
| 45213 | P-Ethyltoluene | 60 | 0.00 | 0.11 | 0.00 | 0.006 | 0.024 |
| 45220 | Styrene | 60 | 0.00 | 0.14 | 0.00 | 0.008 | 0.028 |
| 45801 | Chlorobenzene | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |
| 45805 | 1,2-Dichlorobenzene | 60 | 0.00 | 0.10 | 0.00 | 0.002 | 0.013 |
| 45806 | 1,3-Dichlorobenzene | 60 | 0.00 | 0.09 | 0.00 | 0.002 | 0.012 |
| 45807 | 1,4-Dichlorobenzene | 60 | 0.00 | 0.15 | 0.00 | 0.019 | 0.040 |
| 45810 | 1,2,4-Trichlorobenzene | 60 | 0.00 | 0.13 | 0.00 | 0.004 | 0.022 |
| 46401 | Tetrahydrofuran | 60 | 0.00 | 0.00 | 0.00 | 0.000 | 0.000 |

AQI (Air Quality Index)



What is the AQI?

The air quality index (AQI) is a measurement designed to indicate how clean or polluted the air is in an area, and it also provides information about health effects associated with air pollution. The index is reported daily, or in some cases continuously, and calculated from measured concentrations of five major pollutants regulated by the Clean Air Act: ground-level ozone, particulate matter, carbon monoxide, sulfur dioxide, and nitrogen dioxide. EPA has established national ambient air quality standards (NAAQS) for each of these pollutants to protect public health, and the index is derived from the NAAQS. State and local agencies are required to report the AQI in areas where the population is 350,000 or more, although it is often reported in additional areas as a public service.

How does the AQI work?

The AQI range is from 0 to 500, with the low numbers representing good air quality and the high numbers indicating unhealthy, or even hazardous air quality. The index is divided into six categories with coordinating color codes. In addition, each category has a health-related message associated with it, to inform the public of possible health effects that may arise as a result of breathing polluted air.

Generally, an index of 100 corresponds to the national air quality standard for the pollutant, which is the level that EPA has established to protect public health. Levels below 100 are considered satisfactory, while numbers above 100 are considered unhealthy, first for sensitive groups, and then for the general public as the index value increases.

How is the AQI calculated?

The AQI is calculated from air pollution measurements collected at monitoring sites across the country. The reporting agency must calculate an index for each pollutant from the measured concentrations at all monitoring sites in an area using a standard formula developed by EPA. The pollutant with the highest index is reported as the "primary pollutant", and the highest index is reported as the AQI for the area. If the AQI is above 100, then the agency must report which groups may be sensitive to the primary pollutant. If two or more pollutants have indexes above 100, then the agency must report all groups that may be affected by those pollutants.

In Virginia, as well as most of the nation, the pollutants of greatest concern are ground-level ozone, and airborne particulate matter. Currently, the AQI is only reported for those two pollutants in Virginia.

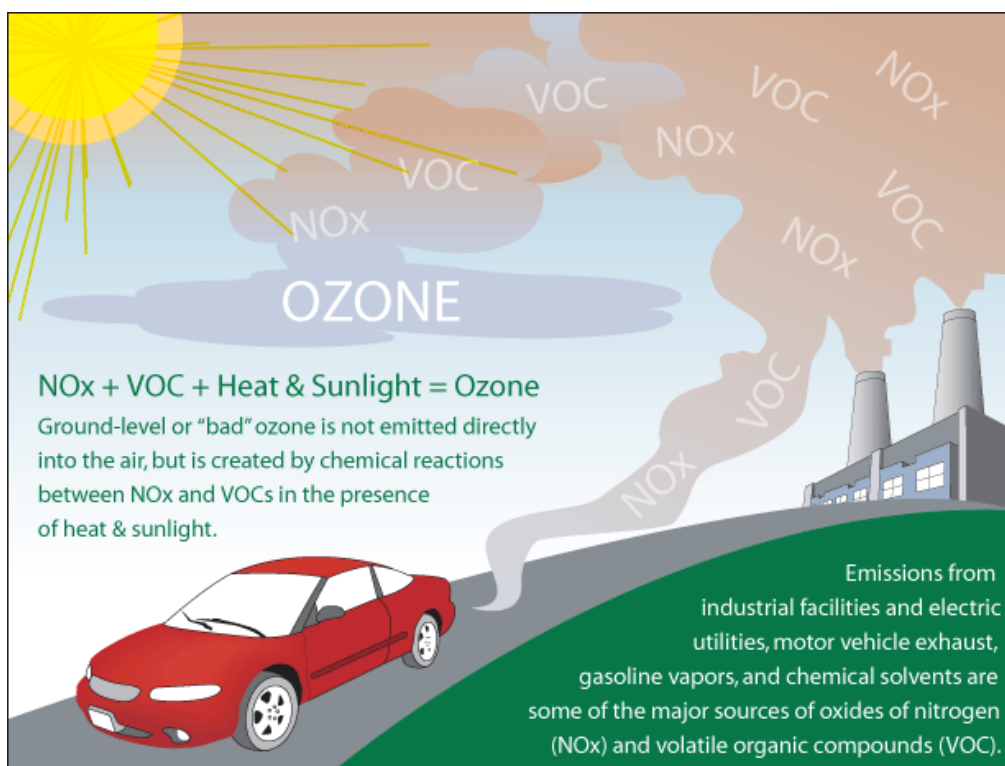
How do I find the AQI for my area?

DEQ reports the air quality index for Roanoke, Winchester, Richmond, Hampton Roads, and Northern Virginia for ozone and particulate matter on the internet at www.deq.virginia.gov/airquality/homepage.html. The AQI for particulate matter is reported year-round and the AQI for ozone is reported during the months of April to October, which is ozone season in Virginia. Air quality forecasts and current ozone data can be obtained at the DEQ site, as well as links to other air quality websites. EPA also reports air quality conditions for the United States at www.airnow.gov.

In addition to the internet, current and forecasted AQI levels are broadcast on local television and radio weather reports in many areas, as well as printed in newspapers. By reaching out to the public using these different media, individuals can plan their activities to reduce exposure during episodes of poor air quality, and they can also take steps to reduce pollution.

For detailed information about the AQI, and on health effects of the pollutants that are included in the AQI, visit www.airnow.gov.

| Air Quality Index Levels of Health Concern | Numerical Value | Meaning |
|--|--------------------|--|
| Good | 0-50 | Air quality is considered satisfactory, and air pollution poses little or no risk. |
| Moderate | 51-100 | Air quality is acceptable; however, for some pollutants there may be a moderate health concern for a very small number of people who are unusually sensitive to air pollution. |
| Unhealthy for Sensitive Groups | 101-150 | Members of sensitive groups may experience health effects. The general public is not likely to be affected. |
| Unhealthy | 151-200 | Everyone may begin to experience health effects; members of sensitive groups may experience more serious health effects. |
| Very Unhealthy | 201-300 | Health alert: everyone may experience more serious health effects. |
| Hazardous | > 300 | Health warnings of emergency conditions. The entire population is more likely to be affected. |



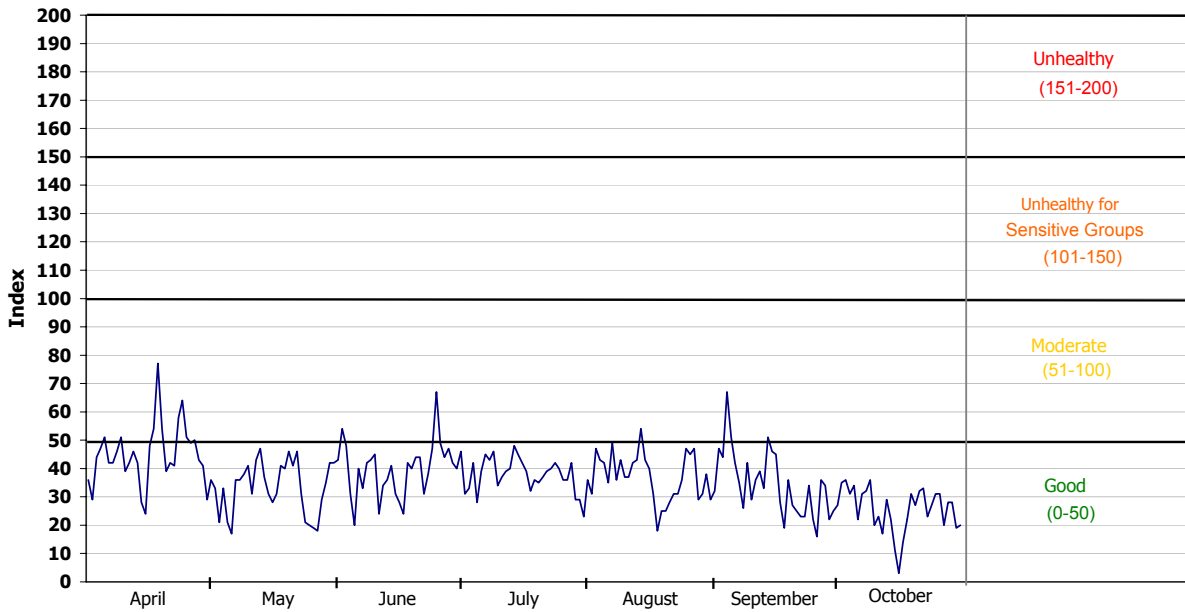
| Every day tips: | Ozone Action Day tips: |
|---|---|
| <ul style="list-style-type: none"> ✚ Conserve energy—at home, at work, everywhere. ✚ Defer use of gasoline-powered lawn and garden equipment. Follow gasoline refueling instructions for efficient vapor recovery. Be careful not to spill fuel and always tighten your gas cap securely. ✚ Keep car, boat, and other engines tuned up according to manufacturers' specification. ✚ Be sure your tires are properly inflated. ✚ Carpool, use public transportation, bike, or walk whenever possible. ✚ Use environmentally safe paints and cleaning products whenever possible. ✚ Some products that you use at your home or office are made with smog-forming chemicals that can evaporate into the air when you use them. Follow manufacturers' recommendations for use and properly seal cleaners, paints, and other chemicals to prevent evaporation into the air. | <ul style="list-style-type: none"> ✚ Conserve electricity and set your air conditioner at a higher temperature. ✚ Choose a cleaner commute—share a ride to work or use public transportation. Bicycle or walk to errands when possible. ✚ Defer use of gasoline-powered lawn and garden equipment. ✚ Refuel cars and trucks after dusk. ✚ Combine errands and reduce trips. ✚ Limit engine idling. ✚ Use household, workshop, and garden chemicals in ways that keep evaporation to a minimum, or try to delay using them when poor air quality is forecast. |

For more information, please visit these sites:

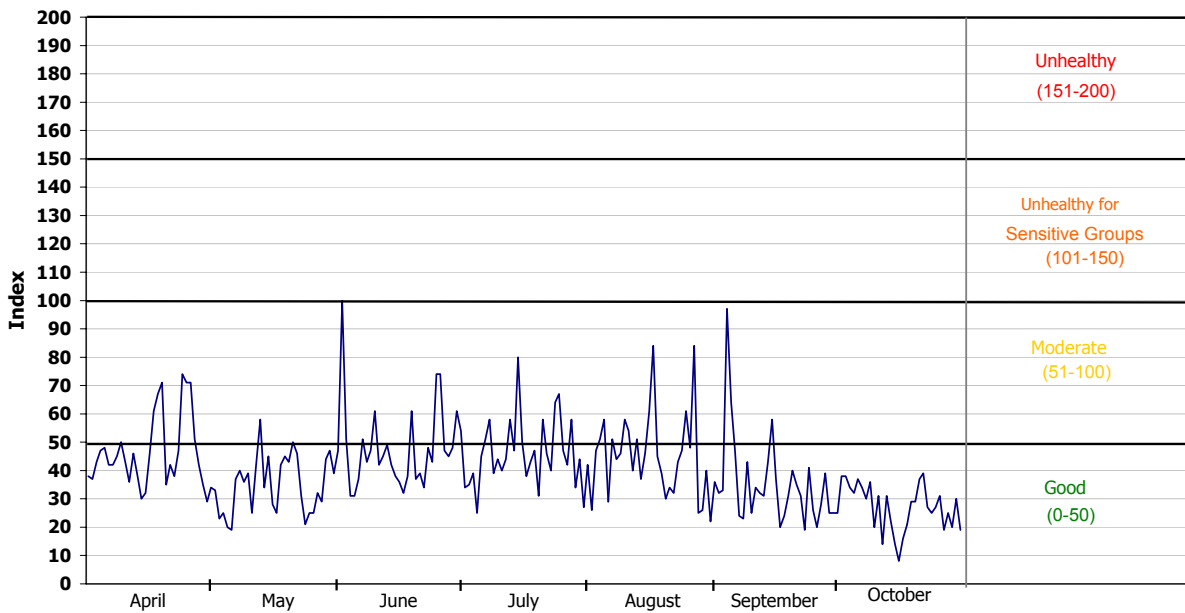
<http://www.epa.gov/otaq/consumer/18-youdo.pdf>

www.airnow.gov/index.cfm?action=resources.whatyoucando

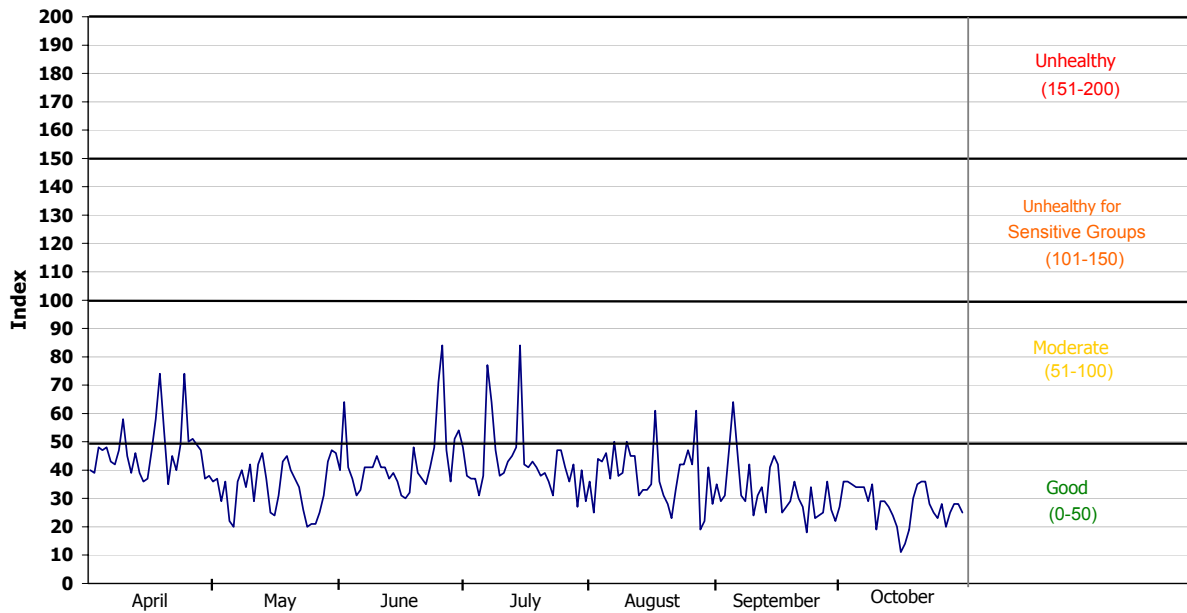
Ozone Air Quality Index Roanoke Area 2009



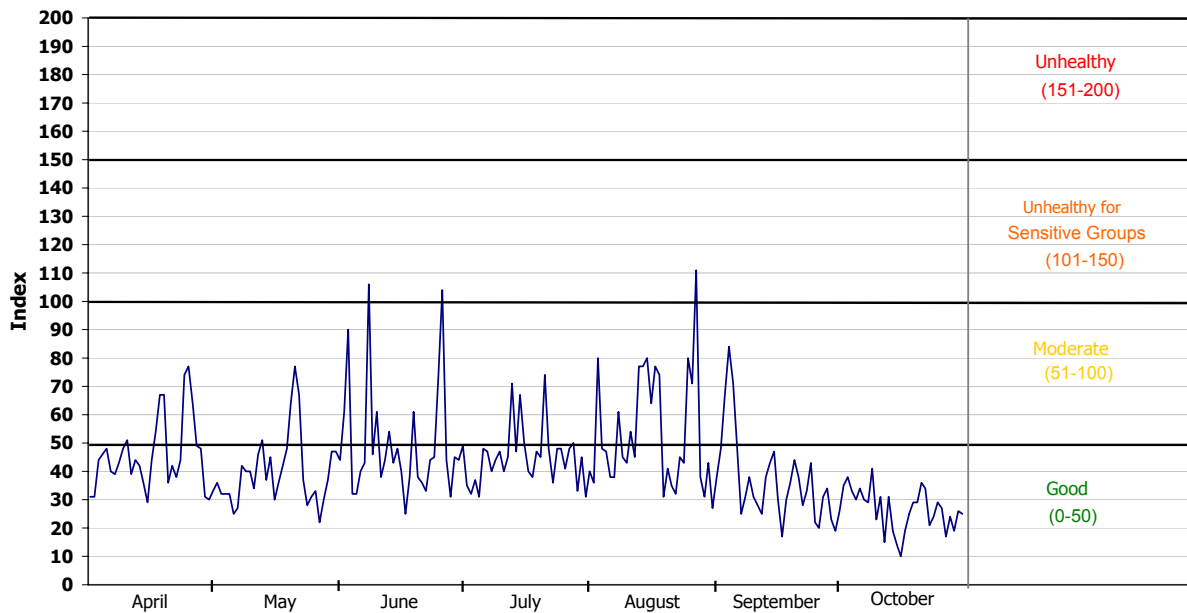
Ozone Air Quality Index Richmond - Petersburg Areas 2009



**Ozone Air Quality Index
Norfolk - Virginia Beach - Newport News Areas
2009**



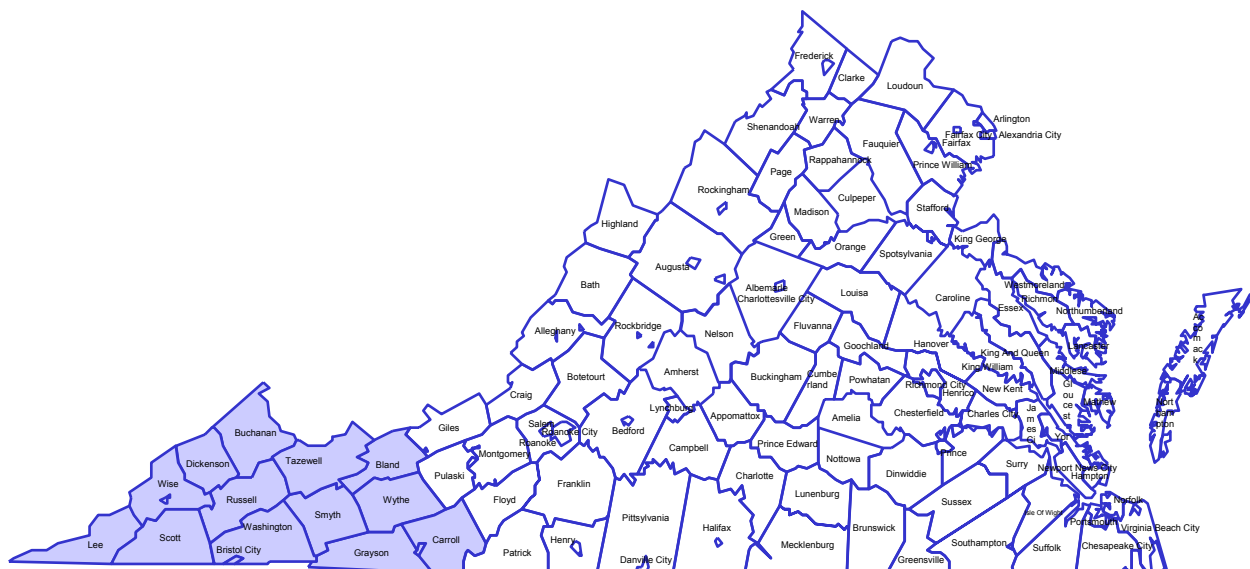
**Ozone Air Quality Index
Washington, DC Area
2009**



Appendix A

| | |
|-------------------|---|
| AQM | Air Quality Monitoring |
| AQCR | Air Quality Control Region |
| ATMN | Air Toxics Monitoring Network |
| Avg. | Average |
| CAMP | Community Assessment Monitoring |
| CO | Carbon Monoxide |
| DEQ | Department of Environmental Quality |
| EAC | Early Action Compacts |
| EPA | Environmental Protection Agency |
| IMPROVE | Interagency Monitoring of Protected Visual Environments |
| LAT | Latitude |
| LONG | Longitude |
| MARAMA | Mid-Atlantic Regional Air Management Association |
| MET. | Meteorological Instrumentation |
| MSA | Metropolitan Statistical Area |
| NA | Not Available |
| NAMS | National Air Monitoring Stations |
| NATTS | National Air Toxics Trend Stations |
| NMOC | Non-Methane Organic Compounds |
| NO ₂ | Nitrogen Dioxide |
| NUM | Number of Samples |
| O ₃ | Ozone |
| PAMHC | Total PAMS Hydrocarbon |
| PAMS | Photochemical Assessment Monitoring Station |
| PM ₁₀ | Particulate Matter with an aerodynamic diameter less than or equal to 10 microns |
| PM _{2.5} | Particulate Matter with an aerodynamic diameter less than or equal to 2.5 microns |
| POLLUT. | Pollutant |
| ppbC | Part Per Billion of Carbon |
| ppbv | Part Per Billion by volume |
| ppm | Part Per Million |
| SLAMS | State and Local Air Monitoring Station |
| SO ₂ | Sulfur Dioxide |
| STD | Standard |
| STDEV | Standard Deviation |
| TEOM | Tapered Element Oscillating Microbalance (a method for continuously measuring PM _{2.5} in ambient air) |
| TNMOC | Total Nonmethane Organic Compound |
| UATM | Urban Air Toxics Monitoring Program |
| ug/m ³ | Micrograms per cubic meter |
| VISTAS | Visibility Improvement State and Tribal Association of the Southeast |
| VOC | Volatile Organic Compounds |

Abbreviation Table



Southwest Monitoring Network 2009

| STATION NUMBER | POLLUT. | LOCATION | EPA ID | CITY/COUNTY | LAT/LONG |
|----------------|-------------------|---------------------------------|-------------|----------------------------|-----------------------|
| 16-B | O ₃ | Sewage Disposal Plant | 51-197-0002 | Rural Retreat Wythe Co. | 36.89117 -81.25423 |
| 23-A | PM ₁₀ | Gladeville Elementary School | 51-035-0001 | Galax Carroll Co. | 36.70067 -80.87978 |
| 101-E | PM _{2.5} | Highland View Elementary School | 51-520-0006 | Bristol | 36.60800 -82.16410 |

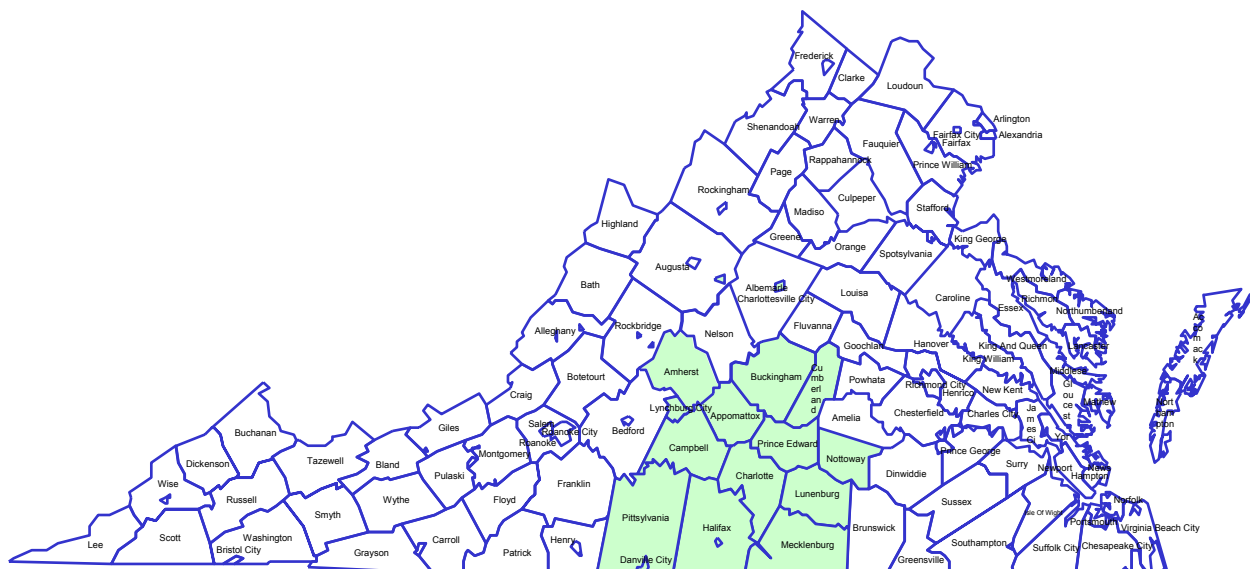
Contact Information for this Region:
 Southwest Regional Office
 Dallas Sizemore, Director
 P.O. Box 1688
 355 Deadmore Street
 Abingdon, VA 24212
 (276) 676-4800



Contact information for this Region:
Valley Regional Office
Amy T. Owens, Director
P.O. Box 3000
4411 Early Road
Harrisonburg, VA 22801
(540) 574-7800

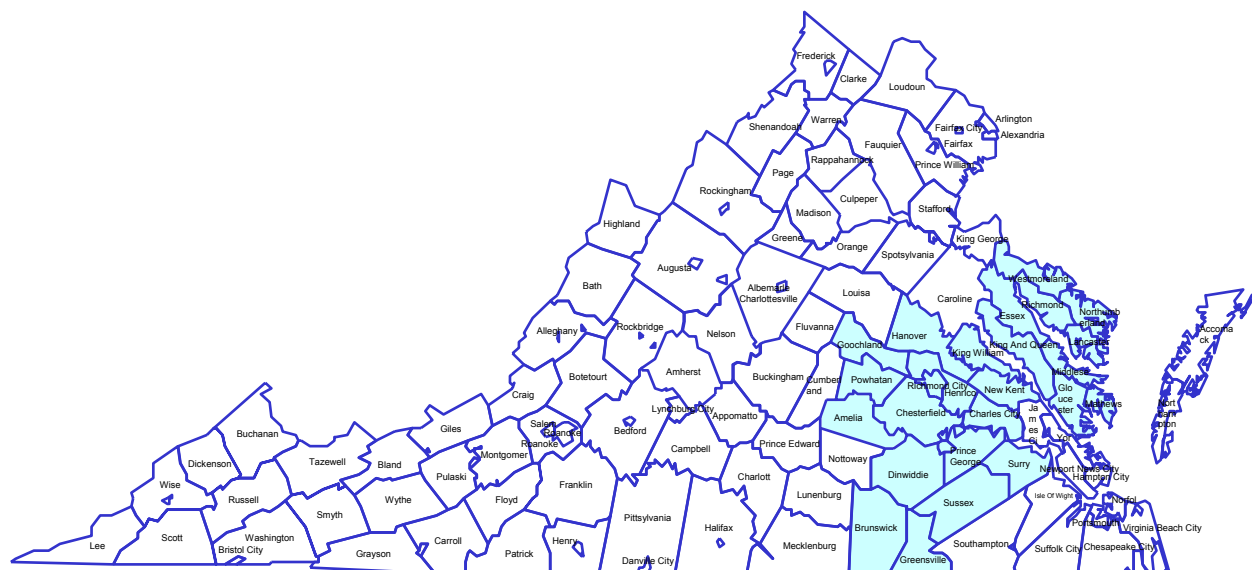


Contact information for this Region:
Blueridge Regional Office
Robert Weld, Director
3019 Peters Creek Road
Roanoke, VA 24019
(540) 562-6700



| STATION NUMBER | POLLUT. | LOCATION | EPA ID | CITY/COUNTY | LAT/LONG |
|----------------|-------------------|--------------------------------|-------------|-------------|-----------------------|
| 155-Q | PM _{2.5} | Leesville Hwy. & Greystone Dr. | 51-680-0015 | Lynchburg | 37.33175 -79.21478 |

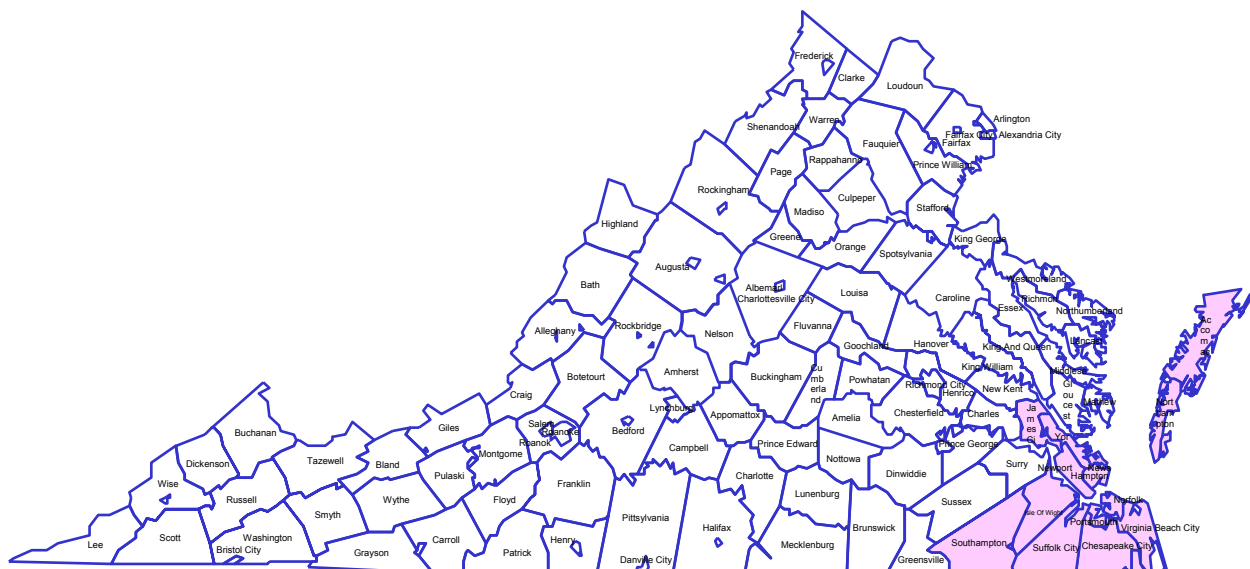
-69-



Piedmont Monitoring Network 2009

| STATION NUMBER | POLLUT. | LOCATION | EPA ID | CITY/COUNTY | LAT/LONG |
|----------------|--|---|-------------|--------------------------------|-----------------------|
| 71-D | PM _{2.5} | Bensley Armory | 51-041-003 | Chesterfield Co. | 37.43467 -77.45118 |
| 71-H | O ₃ | Beach Road Highway Shop | 51-041-0004 | Chesterfield Co. | 37.35748 -77.59355 |
| 72-M | O ₃ , VOC, PM _{2.5} , TEOM, Speciation Toxics | MathScience Innovation Center 2401 Hartman Street | 51-087-0014 | Henrico Co. | 37.55652 -77.40027 |
| 72-N | PM _{2.5} | DEQ-Piedmont Regional Office 4949-A Cox Road | 51-087-0015 | Henrico Co. | 37.67132 -77.56640 |
| 73-E | O ₃ | McClellan Road | 51-085-0003 | Hanover Co. | 37.60613 -77.21880 |
| 75-B | O ₃ , NO ₂ , SO ₂ , PM _{2.5} | Charles City County Route 608 | 51-036-0002 | Charles City Co. | 37.34438 -77.25925 |
| 82-C | PM ₁₀ | West Point Elementary School Thompson Ave. & Chelsea Rd. | 51-101-0003 | West Point King William Co. | 37.55793 -76.79540 |
| 154-M | PM ₁₀ | Carter G. Woodson Middle School 1000 Winston Churchill Dr. | 51-670-0010 | Hopewell | 37.28962 -77.29182 |
| 158-W | CO, SO ₂ , NO ₂ | Science Museum of Virginia DMV and Leigh Street | 51-760-0024 | Richmond | 37.56260 -77.46500 |

Contact Information for this Region:
Piedmont Regional Office
Michael Murphy, Director
4949-A Cox Road
Glen Allen, VA 23060
(804) 527-5020

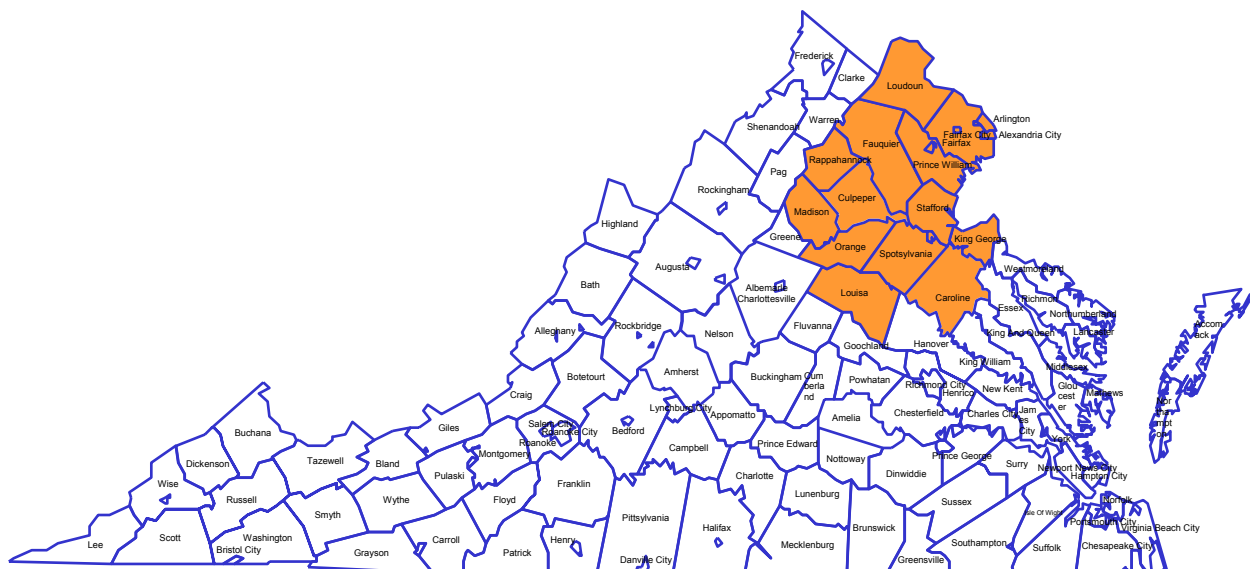


Tidewater Monitoring Network 2009

| STATION NUMBER | POLLUT. | LOCATION | EPA ID | CITY/COUNTY | LAT/LONG |
|----------------|--|--|-------------|--------------|-----------------------|
| 180-O | CO, SO ₂ , NO ₂ , O ₃ , PM _{2.5} , PM ₁₀ , TEOM | Newport News Public Schools Hogan Dr. | 51-700-0013 | Newport News | 37.09980 -76.48105 |
| 181-A1 | CO, SO ₂ , NO ₂ , PM ₁₀ , PM _{2.5} | NOAA Property 2 nd and Woodis Avenue | 51-710-0024 | Norfolk | 36.85555 -76.30135 |
| 183-E | O ₃ | Tidewater Community College Frederick Campus | 51-800-0004 | Suffolk | 36.90118 -76.43808 |
| 183-F | O ₃ | Tidewater Research Station | 51-800-0005 | Suffolk | 36.66525 -76.73078 |
| 184-J | PM _{2.5} , Toxics | DEQ – Tidewater Regional Office 5636 Southern Blvd. | 51-810-0008 | Va. Beach | 36.84188 -76.18123 |

Contact information for this Region:

Francis L. Daniel, Director
5636 Southern Blvd.
Virginia Beach, VA 23462
(757) 518-2000



Northern Monitoring Network 2009

| STATION NUMBER | POLLUT. | LOCATION | EPA ID | CITY/COUNTY | LAT/LONG |
|----------------|---|--|-------------|---------------------------|------------------------|
| 37-B | O ₃ | Phelps Wildlife Area Route 651 | 51-061-0002 | Sumerduck Fauquier Co. | 38.47367 -77.76772 |
| 38-I | O ₃ , NO ₂ , PM _{2.5} | Broad Run High School Route 641 | 51-107-1005 | Ashburn Loudoun Co. | 39.02473 -77.48925 |
| 42-B | PM ₁₀ | Farmington Elementary School Sunset Lane | 51-047-0002 | Culpeper Culpeper Co. | 38.45552 -78.01120 |
| 44-A | O ₃ | Widewater Elementary School Den Rich Road | 51-179-0001 | Widewater Stafford Co. | 38.48123 -77.37040" |
| 45-L | O ₃ , NO ₂ | Long Park Route 15 | 51-153-0009 | Prince William Co. | 38.85287 -77.63462 |
| 46-B9 | O ₃ , CO, PM _{2.5} | Lee District Park Telegraph Road | 51-059-0030 | Franconia Fairfax Co. | 38.77335 -77.10468 |
| 47-T | CO, NO ₂ , O ₃ , PM _{2.5} | Aurora Hills Visitors Center 18 th and Hayes Streets | 51-013-0020 | Arlington Co. | 38.85770 -77.05922 |
| 48-A | O ₃ , NO _y , VOC, PAMS | U.S.G.S. Geomagnetic Center | 51-033-0001 | Corbin Caroline Co. | 38.20087 -77.37742 |
| 130-E | PM ₁₀ | Hugh Mercer Elementary School 2100 Cowan Boulevard | 51-630-0004 | Fredericksburg | 38.30225 -77.48712 |

| STATION NUMBER | POLLUT. | LOCATION | EPA ID | CITY/COUNTY | LAT/LONG |
|----------------|--|--|-------------|-----------------------------|-----------------------|
| L-46-A8 | CO, SO ₂ , O ₃ , NO ₂ , PM _{2.5} | McLean Governmental Center 1437 Balls Hill Road | 51-059-5001 | McLean Fairfax Co. | 38.93260 -77.19822 |
| L-46-B3 | O ₃ , PM ₁₀ | Mt. Vernon Fire Station 2675 Sherwood Hall Lane | 51-059-0018 | Mount Vernon Fairfax Co. | 38.74232 -77.07743 |
| L-46-F | CO, SO ₂ , O ₃ , NO ₂ , PM ₁₀ | Upper Cub Run Drive | 51-059-0005 | Chantilly Fairfax Co. | 38.89410 -77.46520 |
| L-46-C1 | CO, SO ₂ , O ₃ , NO ₂ , PM _{2.5} , TEOM | Mason Governmental Center 6507 Columbia Pike | 51-059-1005 | Annandale Fairfax Co. | 38.83738 -77.16338 |
| L-126-C | CO, SO ₂ , O ₃ , NO ₂ , PM ₁₀ | Alexandria Health Department 517 North Saint Asaph Street | 51-510-0009 | Alexandria | 38.81040 -77.04435 |
| L-126-H | PM ₁₀ | 435 Ferdinand Day Drive | 51-510-0020 | Alexandria | 38.80493 -77.12687 |
| N-35-A | O ₃ , TEOM, IMPROVE, | Big Meadows, National Park Service | 51-113-0003 | Madison Co. | 38.52280 -78.43487 |

Contact Information for this Region:
Northern Regional Office
Thomas Faha, Director
13901 Crown Court
Woodbridge, VA 22193
(703) 583-3800

| Minimum Number of Observations | |
|--|------------------------------------|
| 3-Hour Average | 3 Consecutive Hourly Observations |
| 8-Hour | 6 Hourly Observations |
| 24-Hour | 18 Hourly Observations |
| Quarterly Averages (PM _{2.5} , PM ₁₀) | 75% of Scheduled Samples |
| Yearly Averages (Continuous Instruments) | 75% of Total Possible Observations |
| Yearly Averages (PM _{2.5} , PM ₁₀) | Four Complete Quarterly Averages |

Data Capture Criteria

National Ambient Air Quality Standards

| POLLUTANT | PRIMARY STANDARD | | SECONDARY STANDARD | |
|--|--|-----------------------------------|--------------------|------------------|
| | µg/m ³ | ppm | µg/m ³ | ppm |
| CARBON MONOXIDE 8-hour concentration 1-hour concentration | 10,000 ^a 40,000 ^a | 9 ^a 35 ^a | None | None |
| SULFUR DIOXIDE Annual arithmetic mean 24-hour concentration 3-hour concentration | 80 365 ^a | 0.03 0.14 ^a | 1300 ^a | 0.5 ^a |
| NITROGEN DIOXIDE Annual arithmetic mean | 100 | 0.053 | Same as primary | |
| OZONE 8-hour concentration 1-hour concentration** | 157 ^b | 0.075 ^b | Same as primary | |
| LEAD Quarterly arithmetic mean (October 2008) 3-month rolling average | 1.5 0.15 | | Same as primary | |
| PARTICULATE MATTER PM_{2.5} Annual arithmetic mean 24-hour concentration PM₁₀ 24-hour concentration | 15.0 ^e 35 ^d 150 ^e | | Same as primary | |

^a Not to be exceeded more than once a year

^b 3-year average of the 4th highest 8-hour concentration may not exceed 0.075 ppm

^c Based on a 3-year average of annual arithmetic mean PM2.5 concentrations

^d Based on a 3-year average of 98th percentile of 24-hour PM2.5 concentrations

^e Not to be exceeded more than once per year on average over 3 years

Please see www.epa.gov/air/criteria.html for additional information concerning NAAQS.

NAMS/SLAMS 2009

| REGION | PM _{2.5} | PM ₁₀ | CO | SO ₂ | NO ₂ | O ₃ | TOTAL |
|-------------------|-------------------|------------------|----------|-----------------|-----------------|----------------|-----------|
| Southwest | 1 | 1 | --- | --- | --- | 1 | 3 |
| Valley | 4 | 2 | --- | 1 | 1 | 5 | 13 |
| Blue Ridge | 3 | 1 | 1 | 1 | 1 | 1 | 8 |
| Piedmont | 4 | 3 | 1 | 2 | 3 | 4 | 17 |
| Tidewater | 3 | 2 | 2 | 2 | 1 | 3 | 13 |
| *Northern | 5 | 6 | 5 | 4 | 8 | 13 | 41 |
| TOTAL | 19 | 15 | 9 | 10 | 14 | 27 | 94 |

* This region's sites are operated by DEQ, Fairfax Co., and Alexandria

Number of Criteria Pollutant Monitoring Sites

**8-Hour Ozone Nonattainment Area
(1997 Std.)**

Northern Virginia

Arlington County
Fairfax County
Loudoun County
Prince William County
City of Alexandria
City of Fairfax
City of Falls Church
City of Manassas
City of Manassas Park

**PM_{2.5} Nonattainment Area
Designations (1997 Std.)**

Northern Virginia

Arlington County
Fairfax County
Loudoun County
Prince William County
City of Alexandria
City of Fairfax
City of Falls Church
City of Manassas
City of Manassas Park

**The following are Maintenance Areas
(Previously Nonattainment Areas)**

Fredericksburg

Spotsylvania County
Stafford County
City of Fredericksburg

Richmond

Charles City County
Chesterfield County
Hanover County
Henrico County
Prince George County
City of Colonial Heights
City of Hopewell
City of Petersburg
City of Richmond

**The following are Maintenance Areas
(cont.)**

Norfolk-Va. Beach-Newport News

Gloucester County
Isle of Wright County
James City County
York County
City of Chesapeake
City of Hampton
City of Newport News
City of Norfolk
City of Poquoson
City of Portsmouth
City of Suffolk
City of Virginia Beach
City of Williamsburg

Shenandoah National Park

Page County
Madison County*
(* the portions in SNP)

Ozone & PM_{2.5} Nonattainment Area Designations

Appendix B

AIRSDData – Access to national and state air pollution concentrations and emissions data
<http://www.epa.gov/air/data>

Air Explorer – Collection of user-friendly visualization tools for air quality monitoring
<http://www.epa.gov/airexplorer>

Air Now – Ozone mapping, AQI, and real time data
<http://www.airnow.gov>

Air Now – Air Quality Index Information
<http://www.airnow.gov/index.cfm?action=aqibasics.aqi>

American Lung Association:
<http://www.lungusa.org/>

Department of Environmental Quality link:
<http://www.deq.virginia.gov/>

Education for teachers and children:
<http://www.epa.gov/kids>

IMPROVE
<http://vista.cira.colostate.edu/improve>

MARAMA
<http://www.marama.org/index.html>

Nonattainment area descriptions:
<http://epa.gov/oar/oaqps/greenbk>

U.S. EPA:
<http://www.epa.gov>

VISTAS:
<http://www.vistas-sesarm.org>

2009 3-Day Monitoring Schedule for PM2.5 and 6-Day Monitoring Schedule for PM10:
<http://www.epa.gov/ttn/amtic/calendar.html>

EPA's Technology Transfer Network (TTN) – Ambient Monitoring Technology Information Center (AMTC)
<http://www.epa.gov/ttn/amtic>

Code of Federal Regulations – 40 CFR 50 & 58

Virginia Ambient Air Monitoring Data Reports

DEQ Monthly/Quarterly Reports 2000 – 2009

Air Quality System (AQS)

References