## **SBML Model Report**

# Model name: "Bai2003\_G1phaseRegulation"



May 5, 2016

### 1 General Overview

This is a document in SBML Level 2 Version 4 format. This model was created by the following two authors: Vijayalakshmi Chelliah<sup>1</sup> and Mandri Obeyeseker<sup>2</sup> at March fifth 2010 at 4:22 p. m. and last time modified at February 14<sup>th</sup> 2014 at 1:22 p. m. Table 1 provides an overview of the quantities of all components of this model.

Table 1: Number of components in this model, which are described in the following sections.

Element	Quantity	Element	Quantity
compartment types	0	compartments	1
species types	0	species	6
events	0	constraints	0
reactions	12	function definitions	1
global parameters	26	unit definitions	0
rules	1	initial assignments	0

#### **Model Notes**

This a model from the article:

Theoretical and experimental evidence for hysteresis in cell proliferation.

Bai S, Goodrich D, Thron CD, Tecarro E, Obeyesekere M. Cell Cycle. 2003 Jan-Feb;2(1):46-52. 12695688,

#### **Abstract:**

<sup>&</sup>lt;sup>1</sup>EMBL-EBI, viji@ebi.ac.uk

<sup>&</sup>lt;sup>2</sup>Department of Biomathematics, University of Texas, mandri@biomath.mdacc.tmc.edu

We propose a mathematical model for the regulation of the G1-phase of the mammalian cell cycle taking into account interactions of cyclin D/cdk4, cyclin E/cdk2, Rb and E2F. Mathematical analysis of this model predicts that a change in the proliferative status in response to a change in concentrations of serum growth factors will exhibit the property of hysteresis: the concentration of growth factors required to induce proliferation is higher than the concentration required to maintain proliferation. We experimentally confirmed this prediction in mouse embryonic fibroblasts in vitro. In agreement with the mathematical model, this indicates that changes in proliferative mode caused by small changes in concentrations of growth factors are not easily reversible. Based on this study, we discuss the importance of proliferation hysteresis for cell cycle regulation.

The original model was taken from the Cell Cycle DataBase (CCDB).

Variable added: assignment rule for denoting phosphorylated Rb (Rb\_phosphorylated i.e(RT-RS-R)) created.

To the extent possible under law, all copyright and related or neighbouring rights to this encoded model have been dedicated to the public domain worldwide. Please refer to CCO Public Domain Dedication for more information.

In summary, you are entitled to use this encoded model in absolutely any manner you deem suitable, verbatim, or with modification, alone or embedded it in a larger context, redistribute it, commercially or not, in a restricted way or not.

To cite BioModels Database, please use: Li C, Donizelli M, Rodriguez N, Dharuri H, Endler L, Chelliah V, Li L, He E, Henry A, Stefan MI, Snoep JL, Hucka M, Le Novre N, Laibe C (2010) BioModels Database: An enhanced, curated and annotated resource for published quantitative kinetic models. BMC Syst Biol., 4:92.

#### 2 Unit Definitions

This is an overview of five unit definitions which are all predefined by SBML and not mentioned in the model.

#### 2.1 Unit substance

**Notes** Mole is the predefined SBML unit for substance.

**Definition** mol

#### 2.2 Unit volume

**Notes** Litre is the predefined SBML unit for volume.

**Definition** 1

#### 2.3 Unit area

**Notes** Square metre is the predefined SBML unit for area since SBML Level 2 Version 1.

## $\textbf{Definition}\ m^2$

## 2.4 Unit length

**Notes** Metre is the predefined SBML unit for length since SBML Level 2 Version 1.

**Definition** m

#### 2.5 Unit time

Notes Second is the predefined SBML unit for time.

**Definition** s

## 3 Compartment

This model contains one compartment.

Table 2: Properties of all compartments.

Id	Name	SBO	Spatial Dimensions	Size	Unit	Constant	Outside
cell	cell		3	1	litre	Ø	

## 3.1 Compartment cell

This is a three dimensional compartment with a constant size of one litre.

Name cell

# 4 Species

This model contains six species. Section 9 provides further details and the derived rates of change of each species.

Table 3: Properties of each species.

Id	Name	Compartment	Derived Unit	Constant	Boundary Condi- tion
D_1	D	cell	$\text{mol} \cdot 1^{-1}$		
E_1	E	cell	$\text{mol} \cdot l^{-1}$	$\Box$	
$R_{-}1$	R	cell	$\text{mol} \cdot l^{-1}$	$\Box$	
$RS_{-1}$	RS	cell	$\text{mol} \cdot l^{-1}$	$\Box$	
${\tt theta\_1}$	theta	cell	$\text{mol} \cdot l^{-1}$	$\Box$	
X_1	X	cell	$\text{mol} \cdot 1^{-1}$	$\Box$	$\Box$

## **5 Parameters**

This model contains 26 global parameters.

Table 4: Properties of each parameter.

Id	Name	SBO Value	Unit	Constant
GF_1	GF	6.300	0	
$k1_{-}1$	k1	0.050		$\mathbf{Z}$
$dD_{-}1$	dD	0.400	0	$\overline{\mathbf{Z}}$
$aD_{-}1$	aD	0.400	0	$\overline{\mathbf{Z}}$
aE_1	aE	0.160	0	$\overline{\mathbf{Z}}$
k2_1	k2	1000.000	0	$\overline{\mathbf{Z}}$
aF_1	aF	0.900	0	$\overline{\mathbf{Z}}$
$pX_{-1}$	pX	0.480	0	$\overline{\mathbb{Z}}$
RT_1	RT	2.500	0	$\overline{\mathbf{Z}}$
$qX_{-1}$	qX	0.800	0	$\overline{\mathbf{Z}}$
pS_1	pS	0.600	0	$\overline{Z}$
pD_1	pD	0.480	0	$\overline{Z}$
$qD_{-}1$	qD	0.600	0	$\overline{\mathbf{Z}}$
$pE_{-}1$	pЕ	0.096	6	$\overline{Z}$
$qE_{-}1$	qE	0.600	0	$   \overline{\mathbf{Z}} $
$atheta_1$	atheta	0.050	0	$   \overline{\mathbf{Z}} $
k3_1	k3	1.500	0	$   \overline{\mathbf{Z}} $
$\mathtt{dtheta}_{\mathtt{-}}\mathtt{1}$	dtheta	0.120	0	
${ t qtheta}_{ t 1}$	qtheta	0.300	0	
$aX_{-}1$	aX	0.080	0	
$f_{-}1$	f	0.350	0	
$g_{-}1$	g	0.528	8	$   \overline{\mathbf{Z}} $
$dX_{-}1$	dX	1.040	0	$   \overline{\mathbf{Z}} $
$dE_1$	dE	0.200	0	$   \overline{\mathbf{Z}} $
fC11	fe	0.003	3	$   \overline{\mathbf{Z}} $
${ t Rb\_phos}$	Rb_phosphorylated	0.000	0	

## **6 Function definition**

This is an overview of one function definition.

## **6.1 Function definition Mass\_Action\_2\_1**

Name Mass\_Action\_2

Arguments k1, S1, S2

## **Mathematical Expression**

$$k1 \cdot S1 \cdot S2 \tag{1}$$

## 7 Rule

This is an overview of one rule.

## 7.1 Rule Rb\_phos

Rule  $Bb\_phos$  is an assignment rule for parameter  $Bb\_phos$ :

$$Rb_{-}phos = RT_{-}1 - [RS_{-}1] - [R_{-}1]$$
 (2)

## 8 Reactions

This model contains twelve reactions. All reactions are listed in the following table and are subsequently described in detail. If a reaction is affected by a modifier, the identifier of this species is written above the reaction arrow.

Table 5: Overview of all reactions

N⁰	Id	Name	Reaction Equation	SBO
1	_1	cyclinD synthesis	$\emptyset \longrightarrow D_{-}1$	
2	${\tt cyclinD}_{-}1$	cyclinD degradation	$D_{-}1 \xrightarrow{D_{-}1, E_{-}1} \emptyset$	
3	cyclinEsynthesis	s-cyclinE synthesis	$\emptyset \xrightarrow{\text{theta\_1}} E_{-1}$	
4	cyclinEdegradati	io <b>ny</b> clinE degradation	$E_{-1} \xrightarrow{X_{-1}, E_{-1}} \emptyset$	
5	pRBsynthesis_1	pRB synthesis	$\emptyset \xrightarrow{RS_{-1}, R_{-1}, X_{-1}} R_{-1}$	
6	pRBdeplation_1	pRB/E2F complex association	$R\_1 + theta\_1 \longrightarrow RS\_1$	
7	_2	pRB/E2F complex dissociation via cyclin D	$RS_{-1} \xrightarrow{RS_{-1}, D_{-1}} theta_{-1}$	
8	null2_1	pRB/E2F complex dissociation via cyclin E	$RS_{-1} \xrightarrow{RS_{-1}, E_{-1}} theta_{-1}$	
9	null3_1	E2F synthesis	$\emptyset \xrightarrow{\text{theta}\_1} \text{theta}\_1$	
10	10 E2FdegradationviaE2Fldegradationgrasselbnycle progression _1		theta_1 $\xrightarrow{\text{theta}_1, X_1} \emptyset$	
11	11 cellcycleprogressichheycle progression go		$\emptyset \xrightarrow{E1, \text{ theta}1, X1} X1$	
12	cellcycleprogres	ssáchlsyok-progression slow	$X_{-}1 \xrightarrow{X_{-}1} \emptyset$	

#### 8.1 Reaction \_1

This is an irreversible reaction of no reactant forming one product.

Name cyclinD synthesis

## **Reaction equation**

$$\emptyset \longrightarrow D_{-1}$$
 (3)

#### **Product**

Table 6: Properties of each product.

Id	Name	SBO
$D_{-}1$	D	

#### **Kinetic Law**

**Derived unit** contains undeclared units

$$v_1 = \text{vol} (\text{cell}) \cdot \text{aD}_{-1} \cdot \frac{\text{GF}_{-1}}{\text{k1}_{-1}^1 + \text{GF}_{-1}}$$
 (4)

## 8.2 Reaction cyclinD\_1

This is an irreversible reaction of one reactant forming no product influenced by two modifiers.

Name cyclinD degradation

#### **Reaction equation**

$$D_{-1} \xrightarrow{D_{-1}, E_{-1}} \emptyset \tag{5}$$

## Reactant

Table 7: Properties of each reactant.

Id	Name	SBO
D_1	D	

Table 8: Properties of each modifier.

Id	Name	SBO
D_1	D	
E_1	Е	

**Derived unit** contains undeclared units

$$v_2 = \text{vol}(\text{cell}) \cdot \text{dD}_{-1} \cdot [\text{E}_{-1}] \cdot [\text{D}_{-1}]$$
(6)

## **8.3 Reaction** cyclinEsynthesis\_1

This is an irreversible reaction of no reactant forming one product influenced by one modifier.

Name cyclinE synthesis

### **Reaction equation**

$$\emptyset \xrightarrow{\text{theta}\_1} E_\_1 \tag{7}$$

#### **Modifier**

Table 9: Properties of each modifier.

Id	Name	SBO
theta_1	theta	

#### **Product**

Table 10: Properties of each product.

Id	Name	SBO
E_1	Е	

## **Kinetic Law**

Derived unit contains undeclared units

$$v_3 = \text{vol}\,(\text{cell}) \cdot aE_{-1} \cdot \left(\frac{GF_{-1}}{k2_{-1}^1 + GF_{-1}} + aF_{-1} \cdot [\text{theta}_{-1}]\right)$$
 (8)

## **8.4 Reaction** cyclinEdegradation\_1

This is an irreversible reaction of one reactant forming no product influenced by two modifiers.

Name cyclinE degradation

## **Reaction equation**

$$E_{-1} \xrightarrow{X_{-1}, E_{-1}} \emptyset \tag{9}$$

#### Reactant

Table 11: Properties of each reactant.

Id	Name	SBO
E_1	Е	

#### **Modifiers**

Table 12: Properties of each modifier.

Id	Name	SBO
X_1	X	
$E_{-}1$	E	

#### **Kinetic Law**

**Derived unit** contains undeclared units

$$v_4 = \text{vol}\left(\text{cell}\right) \cdot dE_{-1} \cdot [X_{-1}] \cdot [E_{-1}] \tag{10}$$

## **8.5 Reaction** pRBsynthesis\_1

This is an irreversible reaction of no reactant forming one product influenced by three modifiers.

Name pRB synthesis

#### **Reaction equation**

$$\emptyset \xrightarrow{RS_{-1}, R_{-1}, X_{-1}} R_{-1}$$
 (11)

Table 13: Properties of each modifier.

Id	Name	SBO
RS_1	RS	
$R_{-}1$	R	
$X_{-}1$	X	

#### **Product**

Table 14: Properties of each product.

Id	Name	SBO
R_1	R	

#### **Kinetic Law**

**Derived unit** contains undeclared units

$$v_{5} = vol\left(cell\right) \cdot \frac{pX_{-}1 \cdot (RT_{-}1 - [RS_{-}1] - [R_{-}1]) \cdot [X_{-}1]}{qX_{-}1 + RT_{-}1 - [RS_{-}1] - [R_{-}1] + [X_{-}1]}$$
(12)

## **8.6 Reaction pRBdeplation\_1**

This is an irreversible reaction of two reactants forming one product.

Name pRB/E2F complex association

### **Reaction equation**

$$R_1 + theta_1 \longrightarrow RS_1$$
 (13)

### Reactants

Table 15: Properties of each reactant.

Id	Name	SBO
R_1	R	
theta_1	theta	

#### **Product**

Table 16: Properties of each product.

Id	Name	SBO
RS_1	RS	

**Derived unit** contains undeclared units

$$v_6 = \text{vol}(\text{cell}) \cdot \text{Mass\_Action\_2\_1}(\text{pS\_1}, [\text{R\_1}], [\text{theta\_1}])$$

$$\tag{14}$$

Mass\_Action\_2\_1 (k1, S1, S2) = 
$$k1 \cdot S1 \cdot S2$$
 (15)

Mass\_Action\_2\_1 (k1, S1, S2) = 
$$k1 \cdot S1 \cdot S2$$
 (16)

#### 8.7 Reaction \_2

This is an irreversible reaction of one reactant forming one product influenced by two modifiers.

Name pRB/E2F complex dissociation via cyclin D

## **Reaction equation**

$$RS_{-}1 \xrightarrow{RS_{-}1, D_{-}1} theta_{-}1$$
 (17)

### Reactant

Table 17: Properties of each reactant.

Id	Name	SBO
RS_1	RS	

Table 18: Properties of each modifier.

Id	Name	SBO
RS_1	RS	
$D_{-}1$	D	

#### **Product**

Table 19: Properties of each product.

Id	Name	SBO
theta_1	theta	

#### **Kinetic Law**

**Derived unit** contains undeclared units

$$v_7 = \text{vol}(\text{cell}) \cdot \frac{\text{pD}_{-}1 \cdot [\text{RS}_{-}1] \cdot [\text{D}_{-}1]}{\text{qD}_{-}1 + [\text{RS}_{-}1] + [\text{D}_{-}1]}$$
 (18)

#### 8.8 Reaction null2\_1

This is an irreversible reaction of one reactant forming one product influenced by two modifiers.

Name pRB/E2F complex dissociation via cyclin E

## **Reaction equation**

$$RS_{-1} \xrightarrow{RS_{-1}, E_{-1}} theta_{-1}$$
 (19)

## Reactant

Table 20: Properties of each reactant.

Id	Name	SBO
$RS_{-}1$	RS	

#### **Modifiers**

Table 21: Properties of each modifier.

Id	Name	SBO
$RS_{-}1$	RS	
$E_{-}1$	E	

### **Product**

Table 22: Properties of each product.

Id	Name	SBO
theta_1	theta	

**Derived unit** contains undeclared units

$$v_8 = \text{vol} (\text{cell}) \cdot \frac{pE_{-}1 \cdot [RS_{-}1] \cdot [E_{-}1]}{qE_{-}1 + [RS_{-}1] + [E_{-}1]}$$
(20)

## 8.9 Reaction null3\_1

This is an irreversible reaction of no reactant forming one product influenced by one modifier.

Name E2F synthesis

#### **Reaction equation**

$$\emptyset \xrightarrow{\text{theta}\_1} \text{theta}\_1 \tag{21}$$

#### **Modifier**

Table 23: Properties of each modifier.

Id	Name	SBO
theta_1	theta	

#### **Product**

Table 24: Properties of each product.

Id	Name	SBO
theta_1	theta	

## **Kinetic Law**

**Derived unit** contains undeclared units

$$v_9 = \text{vol}\,(\text{cell}) \cdot \text{atheta}_{-1} \cdot \left(\frac{\text{GF}_{-1}}{\text{k3}_{-1}^1 + \text{GF}_{-1}} + \text{fC}_{-1}_{-1} \cdot [\text{theta}_{-1}]\right)$$
 (22)

## 8.10 Reaction E2Fdegradationviacellcycleprogression\_1

This is an irreversible reaction of one reactant forming no product influenced by two modifiers.

Name E2F degradation via cell cycle progression

### **Reaction equation**

theta\_1 
$$\xrightarrow{\text{theta}_-1, X_-1} \emptyset$$
 (23)

#### Reactant

Table 25: Properties of each reactant.

Id	Name	SBO
theta_1	theta	

#### **Modifiers**

Table 26: Properties of each modifier.

Id	Name	SBO
theta_1 X_1	theta X	

#### **Kinetic Law**

**Derived unit** contains undeclared units

$$\nu_{10} = vol\left(cell\right) \cdot dtheta\_1 \cdot \frac{[X\_1]}{qtheta\_1 + [X\_1]} \cdot [theta\_1] \tag{24}$$

## **8.11 Reaction** cellcycleprogression\_1

This is an irreversible reaction of no reactant forming one product influenced by three modifiers.

Name cell cycle progression go

### **Reaction equation**

$$\emptyset \xrightarrow{\text{E\_1, theta\_1, X\_1}} \text{X\_1}$$
 (25)

Table 27: Properties of each modifier.

Id	Name	SBO
E_1	E	
${\tt theta\_1}$	theta	
X_1	X	

#### **Product**

Table 28: Properties of each product.

Id	Name	SBO
$X_{-}1$	X	

#### **Kinetic Law**

**Derived unit** contains undeclared units

$$v_{11} = vol(cell) \cdot (aX_{-1} \cdot [E_{-1}] + f_{-1} \cdot [theta_{-1}] + g_{-1} \cdot [X_{-1}]^2 \cdot [E_{-1}])$$
 (26)

## **8.12 Reaction** cellcycleprogressionslow\_1

This is an irreversible reaction of one reactant forming no product influenced by one modifier.

Name cell cycle progression slow

## **Reaction equation**

$$X_{-1} \xrightarrow{X_{-1}} \emptyset \tag{27}$$

#### Reactant

Table 29: Properties of each reactant.

Id	Name	SBO
X_1	X	

Table 30: Properties of each modifier.

Id	Name	SBO
X_1	X	

Derived unit contains undeclared units

$$v_{12} = \text{vol}(\text{cell}) \cdot dX_{-1} \cdot [X_{-1}]$$

$$(28)$$

## 9 Derived Rate Equations

When interpreted as an ordinary differential equation framework, this model implies the following set of equations for the rates of change of each species.

Identifiers for kinetic laws highlighted in gray cannot be verified to evaluate to units of SBML substance per time. As a result, some SBML interpreters may not be able to verify the consistency of the units on quantities in the model. Please check if

- parameters without an unit definition are involved or
- volume correction is necessary because the hasOnlySubstanceUnits flag may be set to false and spacialDimensions > 0 for certain species.

#### 9.1 Species D\_1

#### Name D

#### Initial amount 0 mol

This species takes part in four reactions (as a reactant in cyclinD\_1 and as a product in \_1 and as a modifier in cyclinD\_1, \_2).

$$\frac{d}{dt}D_{-}1 = v_1 - v_2 \tag{29}$$

## 9.2 Species E\_1

#### Name E

#### Initial amount 0 mol

This species takes part in six reactions (as a reactant in cyclinEdegradation\_1 and as a product in cyclinEsynthesis\_1 and as a modifier in cyclinD\_1, cyclinEdegradation\_1, null2\_1, cellcycleprogression\_1).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathbf{E}_{-1} = |v_3| - |v_4| \tag{30}$$

#### 9.3 Species R\_1

#### Name R

#### Initial amount 2.5 mol

This species takes part in three reactions (as a reactant in pRBdeplation\_1 and as a product in pRBsynthesis\_1 and as a modifier in pRBsynthesis\_1).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathbf{R}_{-}1 = v_5 - v_6 \tag{31}$$

### 9.4 Species RS\_1

#### Name RS

#### Initial amount 0 mol

This species takes part in six reactions (as a reactant in \_2, null2\_1 and as a product in pRBdeplation\_1 and as a modifier in pRBsynthesis\_1, \_2, null2\_1).

$$\frac{d}{dt}RS_{-1} = v_6 - v_7 - v_8 \tag{32}$$

## 9.5 Species theta\_1

#### Name theta

### Initial amount 0 mol

This species takes part in nine reactions (as a reactant in pRBdeplation\_1, E2Fdegradationviacellcycleprogres\_1 and as a product in \_2, null2\_1, null3\_1 and as a modifier in cyclinEsynthesis\_1, null3\_1, E2Fdegradationviacellcycleprogression\_1, cellcycleprogression\_1).

$$\frac{d}{dt} \text{theta}_{-1} = |v_7| + |v_8| + |v_9| - |v_6| - |v_{10}| \tag{33}$$

#### 9.6 Species X\_1

#### Name X

#### Initial amount 0 mol

This species takes part in seven reactions (as a reactant in cellcycleprogressionslow\_1 and as a product in cellcycleprogression\_1 and as a modifier in cyclinEdegradation\_1, pRBsynthesis\_1, E2Fdegradationviacellcycleprogression\_1, cellcycleprogression-\_1, cellcycleprogressionslow\_1).

$$\frac{d}{dt}X_{-1} = v_{11} - v_{12} \tag{34}$$

 $\mathfrak{BML2}^{AT}$ EX was developed by Andreas Dräger<sup>a</sup>, Hannes Planatscher<sup>a</sup>, Dieudonné M Wouamba<sup>a</sup>, Adrian Schröder<sup>a</sup>, Michael Hucka<sup>b</sup>, Lukas Endler<sup>c</sup>, Martin Golebiewski<sup>d</sup> and Andreas Zell<sup>a</sup>. Please see http://www.ra.cs.uni-tuebingen.de/software/SBML2LaTeX for more information.

<sup>&</sup>lt;sup>a</sup>Center for Bioinformatics Tübingen (ZBIT), Germany

<sup>&</sup>lt;sup>b</sup>California Institute of Technology, Beckman Institute BNMC, Pasadena, United States

 $<sup>^</sup>c\mathrm{European}$  Bioinformatics Institute, Wellcome Trust Genome Campus, Hinxton, United Kingdom

<sup>&</sup>lt;sup>d</sup>EML Research gGmbH, Heidelberg, Germany