# **SBML Model Report**

# Model name: "Pathak2013 - MAPK activation in response to various biotic stresses"



May 5, 2016

## 1 General Overview

This is a document in SBML Level 2 Version 4 format. This model was created by the following two authors: Vijayalakshmi Chelliah<sup>1</sup> and Rajesh Kumar Pathak<sup>2</sup> at November 15<sup>th</sup> 2013 at 12:03 a.m. and last time modified at February 25<sup>th</sup> 2015 at 12:11 a.m. Table 1 provides an overview of the quantities of all components of this model.

Table 1: Number of components in this model, which are described in the following sections.

Element	Element Quantity		Quantity
compartment types	0	compartments	3
species types	0	species	52
events	0	constraints	0
reactions	88	function definitions	0
global parameters	176	unit definitions	6
rules	0	initial assignments	0

## **Model Notes**

Pathak2013 - MAPK activation in response to various biotic stresses

MAPK activation mechanism in response to various biotic (fungal and bacterial pathogens) stress conditions in plants

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This model is described in the article:Modeling of the MAPK machinery activation in response to various abiotic and biotic stresses in plants by a system biology approach.Pathak RK, Taj G, Pandey D, Arora S, Kumar A.Bioinformation 2013; 9(9): 443-449

Abstract:

Mitogen-Activated Protein Kinases (MAPKs) cascade plays an important role in regulating plant growth and development, generating cellular responses to the extracellular stimuli. MAPKs cascade mainly consist of three sub-families i.e. mitogen-activated protein kinase kinase (MAPKK), mitogen-activated protein kinase kinase (MAPKK) and mitogen activated protein kinase (MAPK), several cascades of which are activated by various abiotic and biotic stresses. In this work we have modeled the holistic molecular mechanisms essential to MAPKs activation in response to several abiotic and biotic stresses through a system biology approach and performed its simulation studies. As extent of abiotic and biotic stresses goes on increasing, the process of cell division, cell growth and cell differentiation slow down in time dependent manner. The models developed depict the combinatorial and multicomponent signaling triggered in response to several abiotic and biotic factors. These models can be used to predict behavior of cells in event of various stresses depending on their time and exposure through activation of complex signaling cascades.

This model is hosted on BioModels Database and identifiedby: BIOMD0000000492.

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## 2 Unit Definitions

This is an overview of six unit definitions.

#### 2.1 Unit substance

Name substance

**Definition** mmol

# 2.2 Unit volume

Name volume

**Definition** ml

#### 2.3 Unit area

Name area

**Definition** m<sup>2</sup>

# 2.4 Unit length

Name length

**Definition** m

## 2.5 Unit time

Name time

**Definition** s

# 2.6 Unit per\_second

Definition  $s^{-1}$ 

# 3 Compartments

This model contains three compartments.

Table 2: Properties of all compartments.

Id	Name	SBO	Spatial Dimensions	Size	Unit	Constant	Outside
default			3	1	litre	<b></b>	
c1	Cytosol		3	1	litre	<u></u>	default
c2	Nucleus		3	1	litre	$\overline{\mathbb{Z}}$	c1

# 3.1 Compartment default

This is a three dimensional compartment with a constant size of one ml.

# 3.2 Compartment c1

This is a three dimensional compartment with a constant size of one ml, which is surrounded by default.

Name Cytosol

# 3.3 Compartment c2

This is a three dimensional compartment with a constant size of one ml, which is surrounded by c1 (Cytosol).

Name Nucleus

# 4 Species

This model contains 52 species. Section 7 provides further details and the derived rates of change of each species.

Table 3: Properties of each species.

Id	Name	Compartment	Derived Unit	Constant	Boundary Condi- tion
s1	Fungal pathogen	default	mmol		
s2	Bacterial pathogen	default	mmol		
s3	LysM	c1	mmol		
s4	PRRs	c1	mmol		
<b>s</b> 5	FLS2	c1	mmol		
<b>s</b> 6	LRR	c1	mmol		
s7	MAPKKK	c1	mmol		
s8	MAPKKK	c1	mmol		
s9	MAPKKK1	c1	mmol		
s10	MAPKKK18	c1	mmol		
s11	MAPKKK19	c1	mmol		
s12	MAPKKK20	c1	mmol		
s13	EDR1	c1	mmol		
s14	MAPKK	c1	mmol		
s15	MAPKK	c1	mmol		
s16	MAPKK2	c1	mmol		
s17	MAPKK4	c1	mmol		
s18	MAPKK5	c1	mmol		
s19	MAPKK9	c1	mmol		
s20	MAPK	c1	mmol		
s21	MAPK	c1	mmol		
s22	MAPK2	c1	mmol		$\Box$

Id	Name	Compartment	Derived Unit	Constant	Boundary Condi- tion
s23	MAPK3	c1	mmol		
s24	MAPK4	c1	mmol		
s25	MAPK6	c1	mmol		
s28	WRKY1	c2	mmol		
s29	WRKY1	c2	mmol		
s30	MYB2	c2	mmol		
s31	MYB2	c2	mmol		$\Box$
s32	WRKY33	c2	mmol		
s33	WRKY33	c2	mmol		$\Box$
s34	WRKY6	c2	mmol		
s35	WRKY6	c2	mmol		$\Box$
s36	MYB4	c2	mmol		
s37	MYB4	c2	mmol		$\Box$
s38	WRKY25	c2	mmol		
s39	WRKY25	c2	mmol		
s40	WRKY12	c2	mmol		
s41	WRKY12	c2	mmol		
s42	WRKY22	c2	mmol		
s43	WRKY22	c2	mmol		
s44	WRKY28	c2	mmol		
s45	WRKY28	c2	mmol		
s46	WRKY29	c2	mmol		
s47	WRKY29	c2	mmol		
s48	MYB44	c2	mmol		
s49	NAC	c2	mmol		$\Box$
s50	bZIP	c2	mmol		$\Box$
s51	AP2	c2	mmol		$\Box$

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Id	Name	Compartment	Derived Unit	Constant	Boundary Condi- tion
s52	Response	default	mmol		
s26	SIMK	c1	mmol	$\Box$	
s27	SAMK	c1	mmol	$\Box$	$\Box$

# **5 Parameters**

This model contains 176 global parameters.

Table 4: Properties of each parameter.

Id	Name	SBO	Value	Unit	Constant
kdiss_re1	Dissociation constant of reaction re1	0000282	1.0	$s^{-1}$	Ø
kass_re1	Association constant of reaction re1	0000337	1.0	$s^{-1}$	Ø
kdiss_re2	Dissociation constant of reaction re2	0000282	1.0	$s^{-1}$	Ø
kass_re2	Association constant of reaction re2	0000337	1.0	$s^{-1}$	Ø
kdiss_re3	Dissociation constant of reaction re3	0000282	1.0	$s^{-1}$	Ø
kass_re3	Association constant of reaction re3	0000337	1.0	$s^{-1}$	Ø
kdiss_re4	Dissociation constant of reaction re4	0000282	1.0	$s^{-1}$	Ø
kass_re4	Association constant of reaction re4	0000337	1.0	$s^{-1}$	Ø
kdiss_re5	Dissociation constant of reaction re5	0000282	1.0	$s^{-1}$	Ø
kass_re5	Association constant of reaction re5	0000337	1.0	$s^{-1}$	Ø
kdiss_re6	Dissociation constant of reaction re6	0000282	1.0	$s^{-1}$	Ø
kass_re6	Association constant of reaction re6	0000337	1.0	$s^{-1}$	$\square$

Id	Name	SBO	Value	Unit	Constant
kdiss_re7	Dissociation constant of reaction re7	0000282	1.0	s <sup>-1</sup>	Ø
kass_re7	Association constant of reaction re7	0000337	1.0	$s^{-1}$	Ø
kdiss_re8	Dissociation constant of reaction re8	0000282	1.0	$s^{-1}$	Ø
kass_re8	Association constant of reaction re8	0000337	1.0	$s^{-1}$	Ø
kdiss_re9	Dissociation constant of reaction re9	0000282	1.0	$s^{-1}$	$\mathbf{Z}$
kass_re9	Association constant of reaction re9	0000337	1.0	$s^{-1}$	Ø
kdiss_re10	Dissociation constant of reaction re10	0000282	1.0	$s^{-1}$	Ø
kass_re10	Association constant of reaction re10	0000337	1.0	$s^{-1}$	Ø
kdiss_re11	Dissociation constant of reaction	0000282	1.0	$s^{-1}$	Ø
kass_re11	Association constant of reaction re11	0000337	1.0	$s^{-1}$	Ø
kdiss_re12	Dissociation constant of reaction re12	0000282	1.0	$s^{-1}$	Ø
kass_re12	Association constant of reaction re12	0000337	1.0	$s^{-1}$	Ø
kdiss_re13	Dissociation constant of reaction re13	0000282	1.0	$s^{-1}$	Ø
kass_re13	Association constant of reaction re13	0000337	1.0	$s^{-1}$	Ø

Id	Name	SBO	Value	Unit	Constant
kdiss_re14	Dissociation constant of reaction re14	0000282	1.0	$s^{-1}$	Ø
kass_re14	Association constant of reaction re14	0000337	1.0	$s^{-1}$	✓
kdiss_re15	Dissociation constant of reaction re15	0000282	1.0	$s^{-1}$	<b>⊿</b>
kass_re15	Association constant of reaction re15	0000337	1.0	$s^{-1}$	Ø
kdiss_re16	Dissociation constant of reaction re16	0000282	1.0	$s^{-1}$	Ø
kass_re16	Association constant of reaction re16	0000337	1.0	$s^{-1}$	Z
kdiss_re17	Dissociation constant of reaction re17	0000282	1.0	$s^{-1}$	Z
kass_re17	Association constant of reaction re17	0000337	1.0	$s^{-1}$	Ø
kdiss_re18	Dissociation constant of reaction re18	0000282	1.0	$s^{-1}$	Ø
kass_re18	Association constant of reaction re18	0000337	1.0	$s^{-1}$	Ø
kdiss_re19	Dissociation constant of reaction re19	0000282	1.0	$s^{-1}$	Ø
kass_re19	Association constant of reaction re19	0000337	1.0	$s^{-1}$	Ø
kdiss_re20	Dissociation constant of reaction re20	0000282	1.0	$s^{-1}$	Ø
kass_re20	Association constant of reaction re20	0000337	1.0	$s^{-1}$	Ø

Id	Name	SBO	Value	Unit	Constant
kdiss_re21	Dissociation constant of reaction re21	0000282	1.0	$s^{-1}$	Ø
kass_re21	Association constant of reaction re21	0000337	1.0	$s^{-1}$	Ø
kdiss_re22	Dissociation constant of reaction re22	0000282	1.0	$s^{-1}$	Ø
kass_re22	Association constant of reaction re22	0000337	1.0	$s^{-1}$	Ø
kdiss_re23	Dissociation constant of reaction re23	0000282	1.0	$s^{-1}$	Ø
kass_re23	Association constant of reaction re23	0000337	1.0	$s^{-1}$	Ø
kdiss_re24	Dissociation constant of reaction re24	0000282	1.0	$s^{-1}$	Ø
kass_re24	Association constant of reaction re24	0000337	1.0	$s^{-1}$	Ø
kdiss_re25	Dissociation constant of reaction re25	0000282	1.0	$s^{-1}$	Ø
kass_re25	Association constant of reaction re25	0000337	1.0	$s^{-1}$	Ø
kdiss_re26	Dissociation constant of reaction re26	0000282	1.0	$s^{-1}$	Ø
kass_re26	Association constant of reaction re26	0000337	1.0	$s^{-1}$	Ø
kdiss_re27	Dissociation constant of reaction re27	0000282	1.0	$s^{-1}$	Ø
kass_re27	Association constant of reaction re27	0000337	1.0	$s^{-1}$	Ø

Id	Name	SBO	Value	Unit	Constant
kdiss_re28	Dissociation constant of reaction re28	0000282	1.0	$s^{-1}$	Ø
kass_re28	Association constant of reaction re28	0000337	1.0	$s^{-1}$	Ø
kdiss_re29	Dissociation constant of reaction re29	0000282	1.0	$s^{-1}$	Ø
kass_re29	Association constant of reaction re29	0000337	1.0	$s^{-1}$	Ø
kdiss_re30	Dissociation constant of reaction re30	0000282	1.0	$s^{-1}$	Ø
kass_re30	Association constant of reaction re30	0000337	1.0	$s^{-1}$	Ø
kdiss_re31	Dissociation constant of reaction re31	0000282	1.0	$s^{-1}$	Ø
kass_re31	Association constant of reaction re31	0000337	1.0	$s^{-1}$	Ø
kdiss_re32	Dissociation constant of reaction re32	0000282	1.0	$s^{-1}$	Ø
kass_re32	Association constant of reaction re32	0000337	1.0	$s^{-1}$	Ø
kdiss_re33	Dissociation constant of reaction re33	0000282	1.0	$s^{-1}$	Ø
kass_re33	Association constant of reaction re33	0000337	1.0	$s^{-1}$	Ø
kdiss_re34	Dissociation constant of reaction re34	0000282	1.0	$s^{-1}$	Ø
kass_re34	Association constant of reaction re34	0000337	1.0	$s^{-1}$	Ø

Id	Name	SBO	Value	Unit	Constant
kdiss_re35	Dissociation constant of reaction re35	0000282	1.0	s <sup>-1</sup>	Ø
kass_re35	Association constant of reaction re35	0000337	1.0	$s^{-1}$	Ø
kdiss_re36	Dissociation constant of reaction re36	0000282	1.0	$s^{-1}$	Ø
kass_re36	Association constant of reaction re36	0000337	1.0	$s^{-1}$	Ø
kdiss_re37	Dissociation constant of reaction re37	0000282	1.0	$s^{-1}$	Ø
kass_re37	Association constant of reaction re37	0000337	1.0	$s^{-1}$	Ø
kdiss_re38	Dissociation constant of reaction re38	0000282	1.0	$s^{-1}$	Ø
kass_re38	Association constant of reaction re38	0000337	1.0	$s^{-1}$	Ø
kdiss_re39	Dissociation constant of reaction re39	0000282	1.0	$s^{-1}$	Ø
kass_re39	Association constant of reaction re39	0000337	1.0	$s^{-1}$	Ø
kdiss_re40	Dissociation constant of reaction re40	0000282	1.0	$s^{-1}$	Ø
kass_re40	Association constant of reaction re40	0000337	1.0	$s^{-1}$	Ø
kdiss_re41	Dissociation constant of reaction re41	0000282	1.0	$s^{-1}$	Ø
kass_re41	Association constant of reaction re41	0000337	1.0	$s^{-1}$	Ø

Id	Name	SBO	Value	Unit	Constant
kdiss_re42	Dissociation constant of reaction re42	0000282	1.0	$s^{-1}$	Ø
kass_re42	Association constant of reaction re42	0000337	1.0	$s^{-1}$	<b>⊿</b>
kdiss_re43	Dissociation constant of reaction re43	0000282	1.0	$s^{-1}$	Ø
kass_re43	Association constant of reaction re43	0000337	1.0	$s^{-1}$	Z
kdiss_re44	Dissociation constant of reaction re44	0000282	1.0	$s^{-1}$	Ø
kass_re44	Association constant of reaction re44	0000337	1.0	$s^{-1}$	Ø
kdiss_re45	Dissociation constant of reaction re45	0000282	1.0	$s^{-1}$	Ø
kass_re45	Association constant of reaction re45	0000337	1.0	$s^{-1}$	Ø
kdiss_re46	Dissociation constant of reaction re46	0000282	1.0	$s^{-1}$	
kass_re46	Association constant of reaction re46	0000337	1.0	$s^{-1}$	
kdiss_re47	Dissociation constant of reaction re47	0000282	1.0	$s^{-1}$	Ø
kass_re47	Association constant of reaction re47	0000337	1.0	$s^{-1}$	Ø
kdiss_re48	Dissociation constant of reaction re48	0000282	1.0	$s^{-1}$	Ø
kass_re48	Association constant of reaction re48	0000337	1.0	$s^{-1}$	Ø

Id	Name	SBO	Value	Unit	Constant
kdiss_re49	Dissociation constant of reaction re49	0000282	1.0	s <sup>-1</sup>	Ø
kass_re49	Association constant of reaction re49	0000337	1.0	$s^{-1}$	Ø
kdiss_re50	Dissociation constant of reaction re50	0000282	1.0	$s^{-1}$	Ø
kass_re50	Association constant of reaction re50	0000337	1.0	$s^{-1}$	Ø
kdiss_re51	Dissociation constant of reaction re51	0000282	1.0	$s^{-1}$	Ø
kass_re51	Association constant of reaction re51	0000337	1.0	$s^{-1}$	Ø
kdiss_re52	Dissociation constant of reaction re52	0000282	1.0	$s^{-1}$	Ø
kass_re52	Association constant of reaction re52	0000337	1.0	$s^{-1}$	Ø
kdiss_re53	Dissociation constant of reaction re53	0000282	1.0	$s^{-1}$	Ø
kass_re53	Association constant of reaction re53	0000337	1.0	$s^{-1}$	Ø
kdiss_re54	Dissociation constant of reaction re54	0000282	1.0	$s^{-1}$	Ø
kass_re54	Association constant of reaction re54	0000337	1.0	$s^{-1}$	Ø
kdiss_re55	Dissociation constant of reaction re55	0000282	1.0	$s^{-1}$	Ø
kass_re55	Association constant of reaction re55	0000337	1.0	$s^{-1}$	Ø

Id	Name	SBO	Value	Unit	Constant
kdiss_re56	Dissociation constant of reaction re56	0000282	1.0	$s^{-1}$	Ø
kass_re56	Association constant of reaction re56	0000337	1.0	$s^{-1}$	<b>⊿</b>
kdiss_re57	Dissociation constant of reaction re57	0000282	1.0	$s^{-1}$	Ø
kass_re57	Association constant of reaction re57	0000337	1.0	$s^{-1}$	Ø
kdiss_re58	Dissociation constant of reaction re58	0000282	1.0	$s^{-1}$	$\mathbf{Z}$
kass_re58	Association constant of reaction re58	0000337	1.0	$s^{-1}$	$\mathbf{Z}$
kdiss_re59	Dissociation constant of reaction re59	0000282	1.0	$s^{-1}$	Ø
kass_re59	Association constant of reaction re59	0000337	1.0	$s^{-1}$	Ø
kdiss_re60	Dissociation constant of reaction re60	0000282	1.0	$s^{-1}$	Ø
kass_re60	Association constant of reaction re60	0000337	1.0	$s^{-1}$	Ø
kdiss_re61	Dissociation constant of reaction re61	0000282	1.0	$s^{-1}$	Ø
kass_re61	Association constant of reaction re61	0000337	1.0	$s^{-1}$	
kdiss_re62	Dissociation constant of reaction re62	0000282	1.0	$s^{-1}$	$\square$
kass_re62	Association constant of reaction re62	0000337	1.0	$s^{-1}$	Ø

Id	Name	SBO	Value	Unit	Constant
kdiss_re63	Dissociation constant of reaction re63	0000282	1.0	s <sup>-1</sup>	Ø
kass_re63	Association constant of reaction re63	0000337	1.0	$s^{-1}$	Ø
kdiss_re64	Dissociation constant of reaction re64	0000282	1.0	$s^{-1}$	Ø
kass_re64	Association constant of reaction re64	0000337	1.0	$s^{-1}$	Ø
kdiss_re65	Dissociation constant of reaction re65	0000282	1.0	$s^{-1}$	Ø
kass_re65	Association constant of reaction re65	0000337	1.0	$s^{-1}$	Ø
kdiss_re66	Dissociation constant of reaction re66	0000282	1.0	$s^{-1}$	Ø
kass_re66	Association constant of reaction re66	0000337	1.0	$s^{-1}$	Ø
kdiss_re67	Dissociation constant of reaction re67	0000282	1.0	$s^{-1}$	Ø
kass_re67	Association constant of reaction re67	0000337	1.0	$s^{-1}$	Ø
kdiss_re68	Dissociation constant of reaction re68	0000282	1.0	$s^{-1}$	Ø
kass_re68	Association constant of reaction re68	0000337	1.0	$s^{-1}$	Ø
kdiss_re69	Dissociation constant of reaction re69	0000282	1.0	$s^{-1}$	Ø
kass_re69	Association constant of reaction re69	0000337	1.0	$s^{-1}$	Ø

Id	Name	SBO	Value	Unit	Constant
kdiss_re70	Dissociation constant of reaction re70	0000282	1.0	$s^{-1}$	Ø
kass_re70	Association constant of reaction re70	0000337	1.0	$s^{-1}$	Ø
kdiss_re71	Dissociation constant of reaction re71	0000282	1.0	$s^{-1}$	Ø
kass_re71	Association constant of reaction re71	0000337	1.0	$s^{-1}$	Ø
kdiss_re72	Dissociation constant of reaction re72	0000282	1.0	$s^{-1}$	Ø
kass_re72	Association constant of reaction re72	0000337	1.0	$s^{-1}$	Ø
kdiss_re73	Dissociation constant of reaction re73	0000282	1.0	$s^{-1}$	Ø
kass_re73	Association constant of reaction re73	0000337	1.0	$s^{-1}$	Ø
kdiss_re74	Dissociation constant of reaction re74	0000282	1.0	$s^{-1}$	Ø
kass_re74	Association constant of reaction re74	0000337	1.0	$s^{-1}$	Ø
kdiss_re75	Dissociation constant of reaction re75	0000282	1.0	$s^{-1}$	Ø
kass_re75		0000337	1.0	$s^{-1}$	Ø
kdiss_re76	Dissociation constant of reaction re76	0000282	1.0	$s^{-1}$	Ø
kass_re76	Association constant of reaction re76	0000337	1.0	$s^{-1}$	Ø

Id	Name	SBO	Value	Unit	Constant
kdiss_re77	Dissociation constant of reaction re77	0000282	1.0	$s^{-1}$	Ø
kass_re77	Association constant of reaction re77	0000337	1.0	$s^{-1}$	Ø
kdiss_re78	Dissociation constant of reaction re78	0000282	1.0	$s^{-1}$	Ø
kass_re78	Association constant of reaction re78		1.0	$s^{-1}$	Ø
kdiss_re79	Dissociation constant of reaction re79	0000282	1.0	$s^{-1}$	Ø
kass_re79	Association constant of reaction re79	0000337	1.0	$s^{-1}$	Ø
kdiss_re80	Dissociation constant of reaction re80	0000282	1.0	$s^{-1}$	$\mathbf{Z}$
kass_re80	Association constant of reaction re80	0000337	1.0	$s^{-1}$	Ø
kdiss_re81	Dissociation constant of reaction re81	0000282	1.0	$s^{-1}$	Ø
kass_re81	Association constant of reaction re81	0000337	1.0	$s^{-1}$	Ø
kdiss_re82	Dissociation constant of reaction re82	0000282	1.0	$s^{-1}$	Ø
kass_re82	Association constant of reaction re82	0000337	1.0	$s^{-1}$	Ø
kdiss_re83	Dissociation constant of reaction re83	0000282	1.0	$s^{-1}$	Ø
kass_re83	Association constant of reaction re83	0000337	1.0	$s^{-1}$	Ø

Id	Name	SBO	Value	Unit	Constant
kdiss_re84	Dissociation constant of reaction re84	0000282	1.0	$s^{-1}$	Ø
kass_re84	Association constant of reaction re84	0000337	1.0	$s^{-1}$	Ø
kdiss_re85	Dissociation constant of reaction re85	0000282	1.0	$s^{-1}$	Ø
kass_re85	Association constant of reaction re85	0000337	1.0	$s^{-1}$	Ø
kdiss_re86	Dissociation constant of reaction re86	0000282	1.0	$s^{-1}$	Ø
kass_re86	Association constant of reaction re86	0000337	1.0	$s^{-1}$	Ø
kdiss_re87	Dissociation constant of reaction re87	0000282	1.0	$s^{-1}$	Ø
kass_re87	Association constant of reaction re87	0000337	1.0	$s^{-1}$	Ø
kdiss_re88	Dissociation constant of reaction re88	0000282	1.0	$s^{-1}$	Ø
kass_re88	Association constant of reaction re88	0000337	1.0	$s^{-1}$	<b>⊿</b>

# 6 Reactions

This model contains 88 reactions. All reactions are listed in the following table and are subsequently described in detail. If a reaction is affected by a modifier, the identifier of this species is written above the reaction arrow.

Table 5: Overview of all reactions

N⁰	Id	Name	Reaction Equation	SBO
1	re1		s1 <u></u> ⇒ s3	
2	re2		s1 === s4	
3	re3		s1 <del>←</del> s5	
4	re4		s2 <del>←</del> s4	
5	re5		s2 <del>←</del> s5	
6	re6		s2 <del>←</del> s6	
7	re7		s7 <del>===</del> s8	
8	re8		s3 <del>===</del> s7	
9	re9		s4 <del>===≥</del> s7	
10	re10		s5 <del>←</del> s7	
11	re11		s6 <del>← ``</del> s7	
12	re12		s8 <del>← ``</del> s9	
13	re13		s8 <del>← ``</del> s10	
14	re14		s8 <del>===</del> s11	
15	re15		s8 <u>←</u> s12	
16	re16		s6 <u>←</u> s9	
17	re17		s8 <u>←</u> s13	
18	re18		s5 <u>⇒</u> s13	
19	re19		s5 <del>←</del> s11	
20	re20		s14 <del>←</del> s15	
21	re21		s8 <u>←</u> s14	
22	re22		s15 <del>←</del> s16	
23	re23		s15 <del>←</del> s17	

N₀	Id	Name	Reaction Equation	SBO
24	re24		s15 <del>←</del> s18	
25	re25		s15 <del>←</del> s19	
26	re26		s9 <u>←</u> s16	
27	re27		s9 <u>⇒</u> s17	
28	re28		s9 <del>←                                   </del>	
29	re29		s11 <del>←</del> s19	
30	re30		s20 <del>←</del> s21	
31	re31		s15 <del>←</del> s20	
32	re32		$s21 \Longrightarrow s22$	
33	re33		$s21 \rightleftharpoons s23$	
34	re34		$s21 \Longrightarrow s24$	
35	re35		$s21 \rightleftharpoons s25$	
36	re36		$s21 \Longrightarrow s26$	
37	re37		$s21 \Longrightarrow s27$	
38	re38		$s16 \Longrightarrow s22$	
39	re39		$s17 \Longrightarrow s22$	
40	re40		s17 <u>⇒</u> s23	
41	re41		$s18 \rightleftharpoons s23$	
42	re42		s17 <u>⇒</u> s25	
43	re43		$s16 \Longrightarrow s24$	
44	re44		$s18 \rightleftharpoons s25$	
45	re45		$s28 \rightleftharpoons s29$	
46	re46		s30 <del>←</del> s31	
47	re47		s32 <del>←</del> s33	
48	re48		s34 <u>⇒</u> s35	
49	re49		s36 <u>⇒</u> s37	
50	re50		s38 <u></u> ⇒ s39	
51	re51		s40 <del>←</del> s41	
52	re52		s42 <u>⇒</u> s43	

22	No	Id Name	Reaction Equation	SBO
	53	re53	s44 <u>←</u> s45	
	54	re54	s46 <del>←</del> s47	
	55	re55	s22 <del>← ``</del> s28	
	56	re56	s24 <del>====`</del> s28	
	57	re57	s23 <u>⇒</u> s32	
	58	re58	s24 <u>⇒</u> s32	
	59	re59	s24 <u>⇒</u> s38	
		re60	s23 <u>←</u> s42	
		re61	s25 <u>⇒</u> s42	
F	62	re62	s25 <u></u> ⇒ s46	
Produced by SML2l <sup>AT</sup> EX	63	re63	s25 <del>←</del> s32	
luc		re64	s23 <u>⇒</u> s46	
ed l	65	re65	s23 <del>←</del> s44	
5у (		re66	s25 <u>⇒</u> s44	
99	67	re67	s23 <u>⇒</u> s40	
<u>                                     </u>	68	re68	$s25 \Longrightarrow s36$	
	69 <b>5</b> 0	re69	$s21 \Longrightarrow s30$	
×	70	re70	$s21 \Longrightarrow s48$	
	71	re71	$s21 \Longrightarrow s49$	
		re72	$s21 \Longrightarrow s51$	
	73	re73	$s21 \Longrightarrow s50$	
	74	re74	$s24 \Longrightarrow s34$	
	75 76	re75	$s45 \Longrightarrow s52$	
		re76	$s31 \Longrightarrow s52$	
	77 78	re77	$s49 \Longrightarrow s52$	
		re78	$ \begin{array}{c} s50 \Longleftrightarrow s52 \\ s51 \Longleftrightarrow s52 \end{array} $	
	79 80	re79	$831 \rightleftharpoons 852$ $848 \rightleftharpoons 852$	
	80 81	re80	$\begin{array}{c} 848 \rightleftharpoons 852 \\ 837 \rightleftharpoons 852 \end{array}$	
	01	re81	837 == 832	

N₀	Id	Name	Reaction Equation	SBO
82	re82		s41 <u>⇒</u> s52	
83	re83		s35 <del>←</del> s52	
84	re84		s29 <u>⇒</u> s52	
85	re85		s47 <u>⇒</u> s52	
86	re86		s43 <del>←</del> s52	
87	re87		s39 <u>⇒</u> s52	
88	re88		s33 <del>←</del> s52	

## **6.1 Reaction** re1

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s1 \rightleftharpoons s3$$
 (1)

#### Reactant

Table 6: Properties of each reactant.

Id	Name	SBO
s1	Fungal pathogen	

## **Product**

Table 7: Properties of each product.

Id	Name	SBO
s3	LysM	

# **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_1 = \text{kass\_re1} \cdot \text{s1} - \text{kdiss\_re1} \cdot \text{s3}$$
 (2)

# **6.2 Reaction** re2

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s1 \rightleftharpoons s4$$
 (3)

Table 8: Properties of each reactant.

Id	Name	SBO
s1	Fungal pathogen	

Table 9: Properties of each product.

## **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_2 = \text{kass\_re2} \cdot \text{s1} - \text{kdiss\_re2} \cdot \text{s4} \tag{4}$$

# 6.3 Reaction re3

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s1 \rightleftharpoons s5$$
 (5)

## Reactant

Table 10: Properties of each reactant.

Id	Name	SBO
s1	Fungal pathogen	

# **Product**

Table 11: Properties of each product.

Id	Name	SBO
<b>s</b> 5	FLS2	

#### **Kinetic Law**

$$v_3 = \text{kass\_re3} \cdot \text{s1} - \text{kdiss\_re3} \cdot \text{s5}$$
 (6)

## **6.4 Reaction** re4

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s2 \rightleftharpoons s4$$
 (7)

#### Reactant

Table 12: Properties of each reactant.

Id	Name	SBO
s2	Bacterial pathogen	

## **Product**

Table 13: Properties of each product.

Id	Name	SBO
s4	PRRs	

# **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_4 = \text{kass\_re4} \cdot \text{s2} - \text{kdiss\_re4} \cdot \text{s4}$$
 (8)

# 6.5 Reaction re5

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s2 \rightleftharpoons s5$$
 (9)

Table 14: Properties of each reactant.

Id	Name	SBO
s2	Bacterial pathogen	

Table 15: Properties of each product.

Id	Name	SBO
s5	FLS2	

## **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_5 = \text{kass\_re5} \cdot \text{s2} - \text{kdiss\_re5} \cdot \text{s5}$$
 (10)

# 6.6 Reaction re6

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s2 \rightleftharpoons s6$$
 (11)

## Reactant

Table 16: Properties of each reactant.

Id	Name	SBO
s2	Bacterial pathogen	

# **Product**

Table 17: Properties of each product.

Id	Name	SBO
s6	LRR	

#### **Kinetic Law**

$$v_6 = \text{kass\_re6} \cdot \text{s2} - \text{kdiss\_re6} \cdot \text{s6}$$
 (12)

## **6.7 Reaction** re7

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s7 \rightleftharpoons s8$$
 (13)

#### Reactant

Table 18: Properties of each reactant.

Id	Name	SBO
s7	MAPKKK	

## **Product**

Table 19: Properties of each product.

Id	Name	SBO
s8	MAPKKK	

## **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_7 = \text{kass\_re7} \cdot \text{s7} - \text{kdiss\_re7} \cdot \text{s8}$$
 (14)

# 6.8 Reaction re8

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s3 \rightleftharpoons s7$$
 (15)

Table 20: Properties of each reactant.

Id	Name	SBO
s3	LysM	

Table 21: Properties of each product.

Id	Name	SBO
ຣ7	MAPKKK	

## **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_8 = \text{kass\_re8} \cdot \text{s3} - \text{kdiss\_re8} \cdot \text{s7}$$
 (16)

## 6.9 Reaction re9

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s4 \rightleftharpoons s7$$
 (17)

## Reactant

Table 22: Properties of each reactant.

Id	Name	SBO
s4	PRRs	

# **Product**

Table 23: Properties of each product.

Id	Name	SBO
s7	MAPKKK	

#### **Kinetic Law**

$$v_9 = \text{kass\_re}9 \cdot \text{s4} - \text{kdiss\_re}9 \cdot \text{s7} \tag{18}$$

# 6.10 Reaction re10

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s5 \rightleftharpoons s7$$
 (19)

#### Reactant

Table 24: Properties of each reactant.

Id	Name	SBO
s5	FLS2	

## **Product**

Table 25: Properties of each product.

Id	Name	SBO
s7	MAPKKK	

# **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{10} = \text{kass\_re10} \cdot \text{s5} - \text{kdiss\_re10} \cdot \text{s7}$$
 (20)

# 6.11 Reaction re11

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s6 \rightleftharpoons s7$$
 (21)

Table 26: Properties of each reactant.

Id	Name	SBO
s6	LRR	

Table 27: Properties of each product.

Id	Name	SBO
ຣ7	MAPKKK	

## **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{11} = \text{kass\_re11} \cdot \text{s6} - \text{kdiss\_re11} \cdot \text{s7}$$
 (22)

## 6.12 Reaction re12

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s8 \rightleftharpoons s9$$
 (23)

## Reactant

Table 28: Properties of each reactant.

Id	Name	SBO
s8	MAPKKK	

# **Product**

Table 29: Properties of each product.

Id	Name	SBO
s9	MAPKKK1	

#### **Kinetic Law**

$$v_{12} = \text{kass\_re12} \cdot \text{s8} - \text{kdiss\_re12} \cdot \text{s9}$$
 (24)

# 6.13 Reaction re13

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s8 \rightleftharpoons s10$$
 (25)

#### Reactant

Table 30: Properties of each reactant.

Id	Name	SBO
<b>s</b> 8	MAPKKK	

## **Product**

Table 31: Properties of each product.

Id	Name	SBO
s10	MAPKKK18	

# **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{13} = \text{kass\_re13} \cdot \text{s8} - \text{kdiss\_re13} \cdot \text{s10}$$
 (26)

# 6.14 Reaction re14

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s8 \rightleftharpoons s11$$
 (27)

Table 32: Properties of each reactant.

Id	Name	SBO
s8	MAPKKK	

Table 33: Properties of each product.

Id	Name	SBO
s11	MAPKKK19	

## **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{14} = \text{kass\_re14} \cdot \text{s8} - \text{kdiss\_re14} \cdot \text{s11}$$
 (28)

## 6.15 Reaction re15

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s8 \rightleftharpoons s12$$
 (29)

## Reactant

Table 34: Properties of each reactant.

Id	Name	SBO
88	MAPKKK	

# **Product**

Table 35: Properties of each product.

Id	Name	SBO
s12	MAPKKK20	

#### **Kinetic Law**

$$v_{15} = \text{kass\_re15} \cdot \text{s8} - \text{kdiss\_re15} \cdot \text{s12}$$
 (30)

# 6.16 Reaction re16

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s6 \rightleftharpoons s9$$
 (31)

#### Reactant

Table 36: Properties of each reactant.

Id	Name	SBO
s6	LRR	

## **Product**

Table 37: Properties of each product.

Id	Name	SBO
s9	MAPKKK1	

# **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{16} = \text{kass\_re16} \cdot \text{s6} - \text{kdiss\_re16} \cdot \text{s9}$$
 (32)

# 6.17 Reaction re17

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s8 \rightleftharpoons s13$$
 (33)

Table 38: Properties of each reactant.

Id	Name	SBO
<b>s</b> 8	MAPKKK	

Table 39: Properties of each product.

Id	Name	SBO
s13	EDR1	

## **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{17} = \text{kass\_re17} \cdot \text{s8} - \text{kdiss\_re17} \cdot \text{s13}$$
 (34)

## 6.18 Reaction re18

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s5 \rightleftharpoons s13$$
 (35)

## Reactant

Table 40: Properties of each reactant.

Id	Name	SBO
<b>s</b> 5	FLS2	

# **Product**

Table 41: Properties of each product.

Id	Name	SBO
s13	EDR1	

#### **Kinetic Law**

$$v_{18} = \text{kass\_re18} \cdot \text{s5} - \text{kdiss\_re18} \cdot \text{s13}$$
 (36)

## 6.19 Reaction re19

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s5 \rightleftharpoons s11$$
 (37)

#### Reactant

Table 42: Properties of each reactant.

Id	Name	SBO
s5	FLS2	

## **Product**

Table 43: Properties of each product.

Id	Name	SBO
s11	MAPKKK19	

# **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{19} = \text{kass\_re19} \cdot \text{s5} - \text{kdiss\_re19} \cdot \text{s11}$$
 (38)

# 6.20 Reaction re20

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s14 \rightleftharpoons s15$$
 (39)

Table 44: Properties of each reactant.

Id	Name	SBO
s14	MAPKK	

Table 45: Properties of each product.

Id	Name	SBO
s15	MAPKK	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{20} = \text{kass\_re20} \cdot \text{s14} - \text{kdiss\_re20} \cdot \text{s15}$$
 (40)

### **6.21 Reaction** re21

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s8 \rightleftharpoons s14$$
 (41)

### Reactant

Table 46: Properties of each reactant.

Id	Name	SBO
88	MAPKKK	

## **Product**

Table 47: Properties of each product.

Id	Name	SBO
s14	MAPKK	

#### **Kinetic Law**

$$v_{21} = \text{kass\_re21} \cdot \text{s8} - \text{kdiss\_re21} \cdot \text{s14}$$
 (42)

## 6.22 Reaction re22

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s15 \rightleftharpoons s16$$
 (43)

#### Reactant

Table 48: Properties of each reactant.

Id	Name	SBO
s15	MAPKK	

### **Product**

Table 49: Properties of each product.

Id	Name	SBO
s16	MAPKK2	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{22} = \text{kass\_re22} \cdot \text{s15} - \text{kdiss\_re22} \cdot \text{s16}$$
 (44)

## 6.23 Reaction re23

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s15 \rightleftharpoons s17$$
 (45)

Table 50: Properties of each reactant.

Id	Name	SBO
s15	MAPKK	

Table 51: Properties of each product.

Id	Name	SBO
s17	MAPKK4	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{23} = \text{kass\_re23} \cdot \text{s15} - \text{kdiss\_re23} \cdot \text{s17} \tag{46}$$

## 6.24 Reaction re24

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s15 \rightleftharpoons s18$$
 (47)

### Reactant

Table 52: Properties of each reactant.

Id	Name	SBO
s15	MAPKK	

## **Product**

Table 53: Properties of each product.

Id	Name	SBO
s18	MAPKK5	

#### **Kinetic Law**

$$v_{24} = \text{kass\_re24} \cdot \text{s15} - \text{kdiss\_re24} \cdot \text{s18}$$
 (48)

### 6.25 Reaction re25

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s15 \rightleftharpoons s19$$
 (49)

#### Reactant

Table 54: Properties of each reactant.

Id	Name	SBO
s15	MAPKK	

### **Product**

Table 55: Properties of each product.

Id	Name	SBO
s19	MAPKK9	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{25} = \text{kass\_re25} \cdot \text{s15} - \text{kdiss\_re25} \cdot \text{s19}$$
 (50)

## 6.26 Reaction re26

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s9 \rightleftharpoons s16$$
 (51)

Table 56: Properties of each reactant.

Id	Name	SBO
s9	MAPKKK1	

Table 57: Properties of each product.

Id	Name	SBO
s16	MAPKK2	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{26} = \text{kass\_re26} \cdot \text{s9} - \text{kdiss\_re26} \cdot \text{s16}$$
 (52)

## **6.27 Reaction** re27

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s9 \rightleftharpoons s17$$
 (53)

### Reactant

Table 58: Properties of each reactant.

Id	Name	SBO
<b>s</b> 9	MAPKKK1	

## **Product**

Table 59: Properties of each product.

Id	Name	SBO
s17	MAPKK4	

#### **Kinetic Law**

$$v_{27} = \text{kass\_re27} \cdot \text{s9} - \text{kdiss\_re27} \cdot \text{s17}$$
 (54)

## 6.28 Reaction re28

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s9 \rightleftharpoons s18$$
 (55)

#### Reactant

Table 60: Properties of each reactant.

Id	Name	SBO
<b>s</b> 9	MAPKKK1	

### **Product**

Table 61: Properties of each product.

Id	Name	SBO
s18	MAPKK5	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{28} = \text{kass\_re28} \cdot \text{s9} - \text{kdiss\_re28} \cdot \text{s18} \tag{56}$$

## 6.29 Reaction re29

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s11 \rightleftharpoons s19$$
 (57)

Table 62: Properties of each reactant.

Id	Name	SBO
s11	MAPKKK19	

Table 63: Properties of each product.

Id	Name	SBO
s19	MAPKK9	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{29} = \text{kass\_re29} \cdot \text{s11} - \text{kdiss\_re29} \cdot \text{s19}$$
 (58)

### 6.30 Reaction re30

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s20 \rightleftharpoons s21$$
 (59)

### Reactant

Table 64: Properties of each reactant.

Id	Name	SBO
s20	MAPK	

## **Product**

Table 65: Properties of each product.

Id	Name	SBO
s21	MAPK	

#### **Kinetic Law**

$$v_{30} = \text{kass\_re30} \cdot \text{s20} - \text{kdiss\_re30} \cdot \text{s21}$$
 (60)

## **6.31 Reaction** re31

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s15 \rightleftharpoons s20$$
 (61)

#### Reactant

Table 66: Properties of each reactant.

Id	Name	SBO
s15	MAPKK	

### **Product**

Table 67: Properties of each product.

Id	Name	SBO
s20	MAPK	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{31} = \text{kass\_re31} \cdot \text{s15} - \text{kdiss\_re31} \cdot \text{s20}$$
 (62)

## **6.32 Reaction** re32

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s21 \rightleftharpoons s22$$
 (63)

Table 68: Properties of each reactant.

Id	Name	SBO
s21	MAPK	

Table 69: Properties of each product.

Id	Name	SBO
s22	MAPK2	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{32} = \text{kass\_re32} \cdot \text{s21} - \text{kdiss\_re32} \cdot \text{s22}$$
 (64)

### 6.33 Reaction re33

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s21 \rightleftharpoons s23$$
 (65)

### Reactant

Table 70: Properties of each reactant.

Id	Name	SBO
s21	MAPK	

## **Product**

Table 71: Properties of each product.

Id	Name	SBO
s23	MAPK3	

#### **Kinetic Law**

$$v_{33} = \text{kass\_re33} \cdot \text{s21} - \text{kdiss\_re33} \cdot \text{s23}$$
 (66)

### 6.34 Reaction re34

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s21 \rightleftharpoons s24$$
 (67)

#### Reactant

Table 72: Properties of each reactant.

Id	Name	SBO
s21	MAPK	

### **Product**

Table 73: Properties of each product.

Id	Name	SBO
s24	MAPK4	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{34} = \text{kass\_re34} \cdot \text{s21} - \text{kdiss\_re34} \cdot \text{s24}$$
 (68)

## 6.35 Reaction re35

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s21 \rightleftharpoons s25$$
 (69)

Table 74: Properties of each reactant.

Id	Name	SBO
s21	MAPK	

Table 75: Properties of each product.

Id	Name	SBO
s25	MAPK6	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{35} = \text{kass\_re35} \cdot \text{s21} - \text{kdiss\_re35} \cdot \text{s25} \tag{70}$$

### 6.36 Reaction re36

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s21 \rightleftharpoons s26$$
 (71)

### Reactant

Table 76: Properties of each reactant.

Id	Name	SBO
s21	MAPK	

## **Product**

Table 77: Properties of each product.

Id	Name	SBO
s26	SIMK	

#### **Kinetic Law**

$$v_{36} = \text{kass\_re36} \cdot \text{s21} - \text{kdiss\_re36} \cdot \text{s26}$$
 (72)

### **6.37 Reaction** re37

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s21 \rightleftharpoons s27$$
 (73)

#### Reactant

Table 78: Properties of each reactant.

Id	Name	SBO
s21	MAPK	

### **Product**

Table 79: Properties of each product.

Id	Name	SBO
s27	SAMK	

# **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{37} = \text{kass\_re37} \cdot \text{s21} - \text{kdiss\_re37} \cdot \text{s27} \tag{74}$$

## 6.38 Reaction re38

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s16 \rightleftharpoons s22$$
 (75)

Table 80: Properties of each reactant.

Id	Name	SBO
s16	MAPKK2	

Table 81: Properties of each product.

Id	Name	SBO
s22	MAPK2	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{38} = \text{kass\_re38} \cdot \text{s16} - \text{kdiss\_re38} \cdot \text{s22} \tag{76}$$

### 6.39 Reaction re39

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s17 \rightleftharpoons s22$$
 (77)

### Reactant

Table 82: Properties of each reactant.

Id	Name	SBO
s17	MAPKK4	

## **Product**

Table 83: Properties of each product.

Id	Name	SBO
s22	MAPK2	

#### **Kinetic Law**

$$v_{39} = \text{kass\_re39} \cdot \text{s17} - \text{kdiss\_re39} \cdot \text{s22}$$
 (78)

### 6.40 Reaction re40

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s17 \rightleftharpoons s23$$
 (79)

#### Reactant

Table 84: Properties of each reactant.

Id	Name	SBO
s17	MAPKK4	

### **Product**

Table 85: Properties of each product.

Id	Name	SBO
s23	MAPK3	

## **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{40} = \text{kass\_re40} \cdot \text{s17} - \text{kdiss\_re40} \cdot \text{s23}$$
 (80)

## **6.41 Reaction** re41

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s18 \rightleftharpoons s23$$
 (81)

Table 86: Properties of each reactant.

Id	Name	SBO
s18	MAPKK5	

Table 87: Properties of each product.

Id	Name	SBO
s23	MAPK3	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{41} = \text{kass\_re41} \cdot \text{s18} - \text{kdiss\_re41} \cdot \text{s23}$$
 (82)

### 6.42 Reaction re42

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s17 \rightleftharpoons s25$$
 (83)

### Reactant

Table 88: Properties of each reactant.

Id	Name	SBO
s17	MAPKK4	

## **Product**

Table 89: Properties of each product.

Id	Name	SBO
s25	MAPK6	

#### **Kinetic Law**

$$v_{42} = \text{kass\_re42} \cdot \text{s17} - \text{kdiss\_re42} \cdot \text{s25}$$
 (84)

### 6.43 Reaction re43

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s16 \rightleftharpoons s24$$
 (85)

#### Reactant

Table 90: Properties of each reactant.

Id	Name	SBO
s16	MAPKK2	

### **Product**

Table 91: Properties of each product.

Id	Name	SBO
s24	MAPK4	

# **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{43} = \text{kass\_re43} \cdot \text{s16} - \text{kdiss\_re43} \cdot \text{s24} \tag{86}$$

## 6.44 Reaction re44

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s18 \rightleftharpoons s25$$
 (87)

Table 92: Properties of each reactant.

Id	Name	SBO
s18	MAPKK5	

Table 93: Properties of each product.

Id	Name	SBO
s25	MAPK6	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{44} = \text{kass\_re44} \cdot \text{s18} - \text{kdiss\_re44} \cdot \text{s25}$$
 (88)

### 6.45 Reaction re45

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s28 \rightleftharpoons s29$$
 (89)

### Reactant

Table 94: Properties of each reactant.

Id	Name	SBO
s28	WRKY1	

## **Product**

Table 95: Properties of each product.

Id	Name	SBO
s29	WRKY1	

#### **Kinetic Law**

$$v_{45} = \text{kass\_re45} \cdot \text{s28} - \text{kdiss\_re45} \cdot \text{s29}$$
 (90)

## 6.46 Reaction re46

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s30 \rightleftharpoons s31$$
 (91)

#### Reactant

Table 96: Properties of each reactant.

Id	Name	SBO
s30	MYB2	

### **Product**

Table 97: Properties of each product.

Id	Name	SBO
s31	MYB2	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{46} = \text{kass\_re46} \cdot \text{s30} - \text{kdiss\_re46} \cdot \text{s31}$$
 (92)

## 6.47 Reaction re47

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s32 \rightleftharpoons s33$$
 (93)

Table 98: Properties of each reactant.

Id	Name	SBO
s32	WRKY33	

Table 99: Properties of each product.

Id	Name	SBO
s33	WRKY33	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{47} = \text{kass\_re47} \cdot \text{s32} - \text{kdiss\_re47} \cdot \text{s33} \tag{94}$$

### 6.48 Reaction re48

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s34 \rightleftharpoons s35$$
 (95)

### Reactant

Table 100: Properties of each reactant.

Id	Name	SBO
s34	WRKY6	·

## **Product**

Table 101: Properties of each product.

Id	Name	SBO
s35	WRKY6	

#### **Kinetic Law**

$$v_{48} = \text{kass\_re48} \cdot \text{s34} - \text{kdiss\_re48} \cdot \text{s35}$$
 (96)

### 6.49 Reaction re49

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s36 \rightleftharpoons s37$$
 (97)

#### Reactant

Table 102: Properties of each reactant.

Id	Name	SBO
s36	MYB4	

### **Product**

Table 103: Properties of each product.

Id	Name	SBO
s37	MYB4	

# **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{49} = \text{kass\_re49} \cdot \text{s36} - \text{kdiss\_re49} \cdot \text{s37}$$
 (98)

## 6.50 Reaction re50

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s38 \rightleftharpoons s39$$
 (99)

Table 104: Properties of each reactant.

Id	Name	SBO
s38	WRKY25	

Table 105: Properties of each product.

Id	Name	SBO
s39	WRKY25	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{50} = \text{kass\_re50} \cdot \text{s38} - \text{kdiss\_re50} \cdot \text{s39}$$
 (100)

### **6.51 Reaction** re51

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s40 \rightleftharpoons s41$$
 (101)

### Reactant

Table 106: Properties of each reactant.

Id	Name	SBO
s40	WRKY12	

## **Product**

Table 107: Properties of each product.

Id	Name	SBO
s41	WRKY12	

#### **Kinetic Law**

$$v_{51} = \text{kass\_re51} \cdot \text{s40} - \text{kdiss\_re51} \cdot \text{s41}$$
 (102)

## **6.52 Reaction** re52

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s42 \rightleftharpoons s43$$
 (103)

#### Reactant

Table 108: Properties of each reactant.

Id	Name	SBO
s42	WRKY22	

### **Product**

Table 109: Properties of each product.

Id	Name	SBO
s43	WRKY22	

# **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{52} = \text{kass\_re52} \cdot \text{s42} - \text{kdiss\_re52} \cdot \text{s43} \tag{104}$$

## 6.53 Reaction re53

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s44 \rightleftharpoons s45$$
 (105)

Table 110: Properties of each reactant.

Id	Name	SBO
s44	WRKY28	

Table 111: Properties of each product.

Id	Name	SBO
s45	WRKY28	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{53} = \text{kass\_re53} \cdot \text{s44} - \text{kdiss\_re53} \cdot \text{s45} \tag{106}$$

### 6.54 Reaction re54

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s46 \rightleftharpoons s47$$
 (107)

### Reactant

Table 112: Properties of each reactant.

Id	Name	SBO
s46	WRKY29	

## **Product**

Table 113: Properties of each product.

Id	Name	SBO
s47	WRKY29	

#### **Kinetic Law**

$$v_{54} = \text{kass\_re54} \cdot \text{s46} - \text{kdiss\_re54} \cdot \text{s47}$$
 (108)

## 6.55 Reaction re55

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s22 \rightleftharpoons s28$$
 (109)

#### Reactant

Table 114: Properties of each reactant.

Id	Name	e SBO
s2	2 MAP	K2

### **Product**

Table 115: Properties of each product.

Id	Name	SBO
s28	WRKY1	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{55} = \text{kass\_re55} \cdot \text{s22} - \text{kdiss\_re55} \cdot \text{s28}$$
 (110)

## 6.56 Reaction re56

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s24 \rightleftharpoons s28$$
 (111)

Table 116: Properties of each reactant.

Id	Name	SBO
s24	MAPK4	

Table 117: Properties of each product.

Id	Name	SBO
s28	WRKY1	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{56} = \text{kass\_re56} \cdot \text{s24} - \text{kdiss\_re56} \cdot \text{s28}$$
 (112)

## 6.57 Reaction re57

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s23 \rightleftharpoons s32$$
 (113)

### Reactant

Table 118: Properties of each reactant.

Id	Name	SBO
s23	MAPK3	

## **Product**

Table 119: Properties of each product.

Id	Name	SBO
s32	WRKY33	

#### **Kinetic Law**

$$v_{57} = \text{kass\_re57} \cdot \text{s23} - \text{kdiss\_re57} \cdot \text{s32}$$
 (114)

## 6.58 Reaction re58

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s24 \rightleftharpoons s32$$
 (115)

#### Reactant

Table 120: Properties of each reactant.

Id	Name	SBO
s24	MAPK4	

### **Product**

Table 121: Properties of each product.

Id	Name	SBO
s32	WRKY33	

## **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{58} = \text{kass\_re58} \cdot \text{s24} - \text{kdiss\_re58} \cdot \text{s32} \tag{116}$$

## 6.59 Reaction re59

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s24 \rightleftharpoons s38$$
 (117)

Table 122: Properties of each reactant.

Id	Name	SBO
s24	MAPK4	

Table 123: Properties of each product.

Id	Name	SBO
s38	WRKY25	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{59} = \text{kass\_re59} \cdot \text{s24} - \text{kdiss\_re59} \cdot \text{s38}$$
 (118)

## 6.60 Reaction re60

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s23 \rightleftharpoons s42$$
 (119)

### Reactant

Table 124: Properties of each reactant.

Id	Name	SBO
s23	MAPK3	

## **Product**

Table 125: Properties of each product.

Id	Name	SBO
s42	WRKY22	

#### **Kinetic Law**

$$v_{60} = \text{kass\_re60} \cdot \text{s23} - \text{kdiss\_re60} \cdot \text{s42}$$
 (120)

## 6.61 Reaction re61

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s25 \rightleftharpoons s42$$
 (121)

#### Reactant

Table 126: Properties of each reactant.

Id	Name	SBO
s25	MAPK6	

### **Product**

Table 127: Properties of each product.

Id	Name	SBO
s42	WRKY22	

## **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{61} = \text{kass\_re61} \cdot \text{s25} - \text{kdiss\_re61} \cdot \text{s42}$$
 (122)

## 6.62 Reaction re62

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s25 \Longrightarrow s46$$
 (123)

Table 128: Properties of each reactant.

Id	Name	SBO
s25	MAPK6	

Table 129: Properties of each product.

Id	Name	SBO
s46	WRKY29	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{62} = \text{kass\_re62} \cdot \text{s25} - \text{kdiss\_re62} \cdot \text{s46}$$
 (124)

## 6.63 Reaction re63

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s25 \rightleftharpoons s32$$
 (125)

### Reactant

Table 130: Properties of each reactant.

Id	Name	SBO
s25	MAPK6	

## **Product**

Table 131: Properties of each product.

Id	Name	SBO
s32	WRKY33	

#### **Kinetic Law**

$$v_{63} = \text{kass\_re63} \cdot \text{s25} - \text{kdiss\_re63} \cdot \text{s32}$$
 (126)

## **6.64 Reaction** re64

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s23 \rightleftharpoons s46$$
 (127)

#### Reactant

Table 132: Properties of each reactant.

Id	Name	SBO
s23	MAPK3	

### **Product**

Table 133: Properties of each product.

Id	Name	SBO
s46	WRKY29	

# **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{64} = \text{kass\_re64} \cdot \text{s23} - \text{kdiss\_re64} \cdot \text{s46} \tag{128}$$

## 6.65 Reaction re65

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s23 \Longrightarrow s44$$
 (129)

Table 134: Properties of each reactant.

Id	Name	SBO
s23	MAPK3	

Table 135: Properties of each product.

Id	Name	SBO
s44	WRKY28	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{65} = \text{kass\_re65} \cdot \text{s23} - \text{kdiss\_re65} \cdot \text{s44} \tag{130}$$

## 6.66 Reaction re66

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s25 \rightleftharpoons s44$$
 (131)

### Reactant

Table 136: Properties of each reactant.

Id	Name	SBO
s25	MAPK6	

## **Product**

Table 137: Properties of each product.

Id	Name	SBO
s44	WRKY28	

#### **Kinetic Law**

$$v_{66} = \text{kass\_re66} \cdot \text{s25} - \text{kdiss\_re66} \cdot \text{s44} \tag{132}$$

## **6.67 Reaction** re67

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s23 \rightleftharpoons s40$$
 (133)

#### Reactant

Table 138: Properties of each reactant.

Id	Name	SBO
s23	MAPK3	

### **Product**

Table 139: Properties of each product.

Id	Name	SBO
s40	WRKY12	

## **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{67} = \text{kass\_re67} \cdot \text{s23} - \text{kdiss\_re67} \cdot \text{s40} \tag{134}$$

## 6.68 Reaction re68

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s25 \rightleftharpoons s36$$
 (135)

Table 140: Properties of each reactant.

Id	Name	SBO
s25	MAPK6	

Table 141: Properties of each product.

Id	Name	SBO
s36	MYB4	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{68} = \text{kass\_re68} \cdot \text{s25} - \text{kdiss\_re68} \cdot \text{s36}$$
 (136)

## 6.69 Reaction re69

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s21 \rightleftharpoons s30$$
 (137)

### Reactant

Table 142: Properties of each reactant.

Id	Name	SBO
s21	MAPK	

## **Product**

Table 143: Properties of each product.

Id	Name	SBO
s30	MYB2	

#### **Kinetic Law**

$$v_{69} = \text{kass\_re69} \cdot \text{s21} - \text{kdiss\_re69} \cdot \text{s30}$$
 (138)

## **6.70 Reaction** re70

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s21 \rightleftharpoons s48$$
 (139)

#### Reactant

Table 144: Properties of each reactant.

Id	Name	SBO
s21	MAPK	

### **Product**

Table 145: Properties of each product.

Id	Name	SBO
s48	MYB44	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{70} = \text{kass\_re70} \cdot \text{s21} - \text{kdiss\_re70} \cdot \text{s48} \tag{140}$$

## **6.71 Reaction** re71

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s21 \rightleftharpoons s49$$
 (141)

Table 146: Properties of each reactant.

Id	Name	SBO
s21	MAPK	

Table 147: Properties of each product.

Id	Name	SBO
s49	NAC	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{71} = \text{kass\_re71} \cdot \text{s21} - \text{kdiss\_re71} \cdot \text{s49} \tag{142}$$

## **6.72 Reaction** re72

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s21 \rightleftharpoons s51$$
 (143)

### Reactant

Table 148: Properties of each reactant.

Id	Name	SBO
s21	MAPK	

## **Product**

Table 149: Properties of each product.

Id	Name	SBO
s51	AP2	

#### **Kinetic Law**

$$v_{72} = \text{kass\_re72} \cdot \text{s21} - \text{kdiss\_re72} \cdot \text{s51}$$
 (144)

## 6.73 Reaction re73

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s21 \rightleftharpoons s50$$
 (145)

#### Reactant

Table 150: Properties of each reactant.

Id	Name	SBO
s21	MAPK	

### **Product**

Table 151: Properties of each product.

Id	Name	SBO
s50	bZIP	

## **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{73} = \text{kass\_re73} \cdot \text{s21} - \text{kdiss\_re73} \cdot \text{s50}$$
 (146)

## 6.74 Reaction re74

This is a reversible reaction of one reactant forming one product.

## **Reaction equation**

$$s24 \rightleftharpoons s34$$
 (147)

Table 152: Properties of each reactant.

Id	Name	SBO
s24	MAPK4	

Table 153: Properties of each product.

Id	Name	SBO
s34	WRKY6	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{74} = \text{kass\_re74} \cdot \text{s24} - \text{kdiss\_re74} \cdot \text{s34}$$
 (148)

### 6.75 Reaction re75

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s45 \rightleftharpoons s52$$
 (149)

### Reactant

Table 154: Properties of each reactant.

Id	Name	SBO
s45	WRKY28	

### **Product**

Table 155: Properties of each product.

Id	Name	SBO
s52	Response	

#### **Kinetic Law**

$$v_{75} = \text{kass\_re75} \cdot \text{s45} - \text{kdiss\_re75} \cdot \text{s52}$$
 (150)

# 6.76 Reaction re76

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s31 \rightleftharpoons s52$$
 (151)

#### Reactant

Table 156: Properties of each reactant.

Id	Name	SBO
s31	MYB2	

### **Product**

Table 157: Properties of each product.

Id	Name	SBO
s52	Response	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{76} = \text{kass\_re76} \cdot \text{s31} - \text{kdiss\_re76} \cdot \text{s52}$$
 (152)

# **6.77 Reaction** re77

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s49 \Longrightarrow s52$$
 (153)

#### Reactant

Table 158: Properties of each reactant.

Id	Name	SBO
s49	NAC	

Table 159: Properties of each product.

Id	Name	SBO
s52	Response	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{77} = \text{kass\_re77} \cdot \text{s49} - \text{kdiss\_re77} \cdot \text{s52}$$
 (154)

### 6.78 Reaction re78

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s50 \rightleftharpoons s52$$
 (155)

### Reactant

Table 160: Properties of each reactant.

Id	Name	SBO
s50	bZIP	

### **Product**

Table 161: Properties of each product.

Id	Name	SBO
s52	Response	

#### **Kinetic Law**

$$v_{78} = \text{kass\_re78} \cdot \text{s50} - \text{kdiss\_re78} \cdot \text{s52}$$
 (156)

### 6.79 Reaction re79

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s51 \rightleftharpoons s52$$
 (157)

#### Reactant

Table 162: Properties of each reactant.

Id	Name	SBO
s51	AP2	

### **Product**

Table 163: Properties of each product.

Id	Name	SBO
s52	Response	

# **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{79} = \text{kass\_re79} \cdot \text{s51} - \text{kdiss\_re79} \cdot \text{s52}$$
 (158)

# 6.80 Reaction re80

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s48 \Longrightarrow s52$$
 (159)

#### Reactant

Table 164: Properties of each reactant.

Id	Name	SBO
s48	MYB44	

Table 165: Properties of each product.

Id	Name	SBO
s52	Response	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{80} = \text{kass\_re80} \cdot \text{s48} - \text{kdiss\_re80} \cdot \text{s52}$$
 (160)

### 6.81 Reaction re81

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s37 \rightleftharpoons s52$$
 (161)

### Reactant

Table 166: Properties of each reactant.

s37 MYB	34

### **Product**

Table 167: Properties of each product.

Id	Name	SBO
s52	Response	

#### **Kinetic Law**

$$v_{81} = \text{kass\_re81} \cdot \text{s37} - \text{kdiss\_re81} \cdot \text{s52}$$
 (162)

# 6.82 Reaction re82

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s41 \rightleftharpoons s52$$
 (163)

#### Reactant

Table 168: Properties of each reactant.

Id	Name	SBO
s41	WRKY12	

### **Product**

Table 169: Properties of each product.

Id	Name	SBO
s52	Response	

# Kinetic Law

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{82} = \text{kass\_re82} \cdot \text{s41} - \text{kdiss\_re82} \cdot \text{s52} \tag{164}$$

# 6.83 Reaction re83

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s35 \rightleftharpoons s52$$
 (165)

#### Reactant

Table 170: Properties of each reactant.

Id	Name	SBO
s35	WRKY6	

Table 171: Properties of each product.

Id	Name	SBO
s52	Response	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{83} = \text{kass\_re83} \cdot \text{s35} - \text{kdiss\_re83} \cdot \text{s52}$$
 (166)

### 6.84 Reaction re84

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s29 \Longrightarrow s52$$
 (167)

### Reactant

Table 172: Properties of each reactant.

Id	Name	SBO
s29	WRKY1	

### **Product**

Table 173: Properties of each product.

Id	Name	SBO
s52	Response	

#### **Kinetic Law**

$$v_{84} = \text{kass\_re84} \cdot \text{s29} - \text{kdiss\_re84} \cdot \text{s52}$$
 (168)

# 6.85 Reaction re85

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s47 \rightleftharpoons s52$$
 (169)

#### Reactant

Table 174: Properties of each reactant.

Id	Name	SBO
s47	WRKY29	

### **Product**

Table 175: Properties of each product.

Id	Name	SBO
s52	Response	

# **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{85} = \text{kass\_re85} \cdot \text{s47} - \text{kdiss\_re85} \cdot \text{s52} \tag{170}$$

# 6.86 Reaction re86

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s43 \rightleftharpoons s52$$
 (171)

#### Reactant

Table 176: Properties of each reactant.

Id	Name	SBO
s43	WRKY22	

Table 177: Properties of each product.

Id	Name	SBO
s52	Response	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{86} = \text{kass\_re86} \cdot \text{s43} - \text{kdiss\_re86} \cdot \text{s52}$$
 (172)

### 6.87 Reaction re87

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s39 \Longrightarrow s52$$
 (173)

### Reactant

Table 178: Properties of each reactant.

Id	Name	SBO
s39	WRKY25	

### **Product**

Table 179: Properties of each product.

Id	Name	SBO
s52	Response	

#### **Kinetic Law**

$$v_{87} = \text{kass\_re87} \cdot \text{s39} - \text{kdiss\_re87} \cdot \text{s52}$$
 (174)

### 6.88 Reaction re88

This is a reversible reaction of one reactant forming one product.

# **Reaction equation**

$$s33 \rightleftharpoons s52$$
 (175)

#### Reactant

Table 180: Properties of each reactant.

Id	Name	SBO
s33	WRKY33	

#### **Product**

Table 181: Properties of each product.

Id	Name	SBO
s52	Response	

### **Kinetic Law**

**Derived unit**  $s^{-1} \cdot mmol$ 

$$v_{88} = \text{kass\_re88} \cdot \text{s33} - \text{kdiss\_re88} \cdot \text{s52} \tag{176}$$

# 7 Derived Rate Equations

When interpreted as an ordinary differential equation framework, this model implies the following set of equations for the rates of change of each species.

# 7.1 Species s1

Name Fungal pathogen

Initial amount 0.5

### Charge 0

This species takes part in three reactions (as a reactant in re1, re2, re3).

$$\frac{d}{dt}s1 = -v_1 - v_2 - v_3 \tag{177}$$

# **7.2 Species** s2

Name Bacterial pathogen

Initial amount 0.5

#### Charge 0

This species takes part in three reactions (as a reactant in re4, re5, re6).

$$\frac{d}{dt}s2 = -v_4 - v_5 - v_6 \tag{178}$$

# 7.3 Species s3

Name LysM

Initial amount 0.8

### Charge 0

This species takes part in two reactions (as a reactant in re8 and as a product in re1).

$$\frac{\mathrm{d}}{\mathrm{d}t}s3 = v_1 - v_8 \tag{179}$$

### 7.4 Species s4

Name PRRs

Initial amount 0.8

#### Charge 0

This species takes part in three reactions (as a reactant in re9 and as a product in re2, re4).

$$\frac{d}{dt}s4 = v_2 + v_4 - v_9 \tag{180}$$

# 7.5 Species s5

Name FLS2

Initial amount 0.8

#### Charge 0

This species takes part in five reactions (as a reactant in re10, re18, re19 and as a product in re3, re5).

$$\frac{\mathrm{d}}{\mathrm{d}t}s5 = v_3 + v_5 - v_{10} - v_{18} - v_{19} \tag{181}$$

### 7.6 Species s6

Name LRR

Initial amount 0.8

#### Charge 0

This species takes part in three reactions (as a reactant in re11, re16 and as a product in re6).

$$\frac{\mathrm{d}}{\mathrm{d}t}s6 = v_6 - v_{11} - v_{16} \tag{182}$$

# 7.7 Species s7

Name MAPKKK

Initial amount 1

### Charge 0

This species takes part in five reactions (as a reactant in re7 and as a product in re8, re9, re10, re11).

$$\frac{\mathrm{d}}{\mathrm{d}t}s7 = v_8 + v_9 + v_{10} + v_{11} - v_7 \tag{183}$$

### 7.8 Species s8

Name MAPKKK

Initial amount 1

# $\textbf{Charge} \ \ 0$

This species takes part in seven reactions (as a reactant in re12, re13, re14, re15, re17, re21 and as a product in re7).

$$\frac{\mathrm{d}}{\mathrm{d}t}s8 = v_7 - v_{12} - v_{13} - v_{14} - v_{15} - v_{17} - v_{21} \tag{184}$$

### 7.9 Species s9

Name MAPKKK1

Initial amount 1

# $\textbf{Charge} \ \ 0$

This species takes part in five reactions (as a reactant in re26, re27, re28 and as a product in re12, re16).

$$\frac{\mathrm{d}}{\mathrm{d}t}s9 = v_{12} + v_{16} - v_{26} - v_{27} - v_{28} \tag{185}$$

# **7.10 Species** s10

Name MAPKKK18

Initial amount 1

Charge 0

This species takes part in one reaction (as a product in re13).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{s}10 = v_{13} \tag{186}$$

# **7.11 Species** s11

Name MAPKKK19

Initial amount 1

Charge 0

This species takes part in three reactions (as a reactant in re29 and as a product in re14, re19).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathbf{s}11 = v_{14} + v_{19} - v_{29} \tag{187}$$

# **7.12 Species** s12

Name MAPKKK20

Initial amount 1

Charge 0

This species takes part in one reaction (as a product in re15).

$$\frac{d}{dt}s12 = v_{15} {(188)}$$

# **7.13 Species** s13

Name EDR1

Initial amount 1

Charge 0

This species takes part in two reactions (as a product in re17, re18).

$$\frac{\mathrm{d}}{\mathrm{d}t}s13 = v_{17} + v_{18} \tag{189}$$

### **7.14 Species** s14

Name MAPKK

Initial amount 1

#### Charge 0

This species takes part in two reactions (as a reactant in re20 and as a product in re21).

$$\frac{\mathrm{d}}{\mathrm{d}t}s14 = v_{21} - v_{20} \tag{190}$$

# **7.15 Species** s15

Name MAPKK

Initial amount 1

### Charge 0

This species takes part in six reactions (as a reactant in re22, re23, re24, re25, re31 and as a product in re20).

$$\frac{\mathrm{d}}{\mathrm{d}t}s15 = v_{20} - v_{22} - v_{23} - v_{24} - v_{25} - v_{31} \tag{191}$$

# **7.16 Species** s16

Name MAPKK2

Initial amount 1

# $\textbf{Charge} \ \ 0$

This species takes part in four reactions (as a reactant in re38, re43 and as a product in re22, re26).

$$\frac{\mathrm{d}}{\mathrm{d}t}s16 = v_{22} + v_{26} - v_{38} - v_{43} \tag{192}$$

### **7.17 Species** s17

Name MAPKK4

Initial amount 1

# $\textbf{Charge} \ \ 0$

This species takes part in five reactions (as a reactant in re39, re40, re42 and as a product in re23, re27).

$$\frac{\mathrm{d}}{\mathrm{d}t}s17 = v_{23} + v_{27} - v_{39} - v_{40} - v_{42} \tag{193}$$

### **7.18 Species** s18

Name MAPKK5

Initial amount 1

### Charge 0

This species takes part in four reactions (as a reactant in re41, re44 and as a product in re24, re28).

$$\frac{\mathrm{d}}{\mathrm{d}t}s18 = v_{24} + v_{28} - v_{41} - v_{44} \tag{194}$$

# **7.19 Species** s19

Name MAPKK9

Initial amount 1

### Charge 0

This species takes part in two reactions (as a product in re25, re29).

$$\frac{\mathrm{d}}{\mathrm{d}t}s19 = v_{25} + v_{29} \tag{195}$$

#### **7.20 Species** s20

Name MAPK

Initial amount 1

#### Charge 0

This species takes part in two reactions (as a reactant in re30 and as a product in re31).

$$\frac{\mathrm{d}}{\mathrm{d}t}s20 = v_{31} - v_{30} \tag{196}$$

# **7.21 Species** s21

Name MAPK

Initial amount 1

#### Charge 0

This species takes part in twelve reactions (as a reactant in re32, re33, re34, re35, re36, re37, re69, re70, re71, re72, re73 and as a product in re30).

$$\frac{\mathrm{d}}{\mathrm{d}t}s21 = v_{30} - v_{32} - v_{33} - v_{34} - v_{35} - v_{36} - v_{37} - v_{69} - v_{70} - v_{71} - v_{72} - v_{73}$$
 (197)

### **7.22 Species** s22

Name MAPK2

Initial amount 1

#### Charge 0

This species takes part in four reactions (as a reactant in re55 and as a product in re32, re38, re39).

$$\frac{\mathrm{d}}{\mathrm{d}t}s22 = v_{32} + v_{38} + v_{39} - v_{55} \tag{198}$$

## **7.23 Species** s23

Name MAPK3

Initial amount 1

#### Charge 0

This species takes part in eight reactions (as a reactant in re57, re60, re64, re65, re67 and as a product in re33, re40, re41).

$$\frac{\mathrm{d}}{\mathrm{d}t}s23 = v_{33} + v_{40} + v_{41} - v_{57} - v_{60} - v_{64} - v_{65} - v_{67}$$
(199)

#### **7.24 Species** s24

Name MAPK4

Initial amount 1

### Charge 0

This species takes part in six reactions (as a reactant in re56, re58, re59, re74 and as a product in re34, re43).

$$\frac{\mathrm{d}}{\mathrm{d}t}s24 = v_{34} + v_{43} - v_{56} - v_{58} - v_{59} - v_{74} \tag{200}$$

### **7.25 Species** s25

Name MAPK6

Initial amount 1

#### Charge 0

This species takes part in eight reactions (as a reactant in re61, re62, re63, re66, re68 and as a product in re35, re42, re44).

$$\frac{\mathrm{d}}{\mathrm{d}t}s25 = v_{35} + v_{42} + v_{44} - v_{61} - v_{62} - v_{63} - v_{66} - v_{68}$$
(201)

# **7.26 Species** s28

Name WRKY1

Initial amount 1.2

### Charge 0

This species takes part in three reactions (as a reactant in re45 and as a product in re55, re56).

$$\frac{\mathrm{d}}{\mathrm{d}t}s28 = v_{55} + v_{56} - v_{45} \tag{202}$$

# **7.27 Species** s29

Name WRKY1

Initial amount 1.2

### Charge 0

This species takes part in two reactions (as a reactant in re84 and as a product in re45).

$$\frac{d}{dt}s29 = v_{45} - v_{84} \tag{203}$$

### **7.28 Species** s30

Name MYB2

Initial amount 1.5

### Charge 0

This species takes part in two reactions (as a reactant in re46 and as a product in re69).

$$\frac{d}{dt}s30 = v_{69} - v_{46} \tag{204}$$

### **7.29 Species** s31

Name MYB2

Initial amount 1.5

### Charge 0

This species takes part in two reactions (as a reactant in re76 and as a product in re46).

$$\frac{\mathrm{d}}{\mathrm{d}t}s31 = v_{46} - v_{76} \tag{205}$$

### **7.30 Species** s32

Name WRKY33

Initial amount 1.2

#### Charge 0

This species takes part in four reactions (as a reactant in re47 and as a product in re57, re58, re63).

$$\frac{\mathrm{d}}{\mathrm{d}t}s32 = v_{57} + v_{58} + v_{63} - v_{47} \tag{206}$$

# **7.31 Species** s33

Name WRKY33

Initial amount 1.2

# Charge 0

This species takes part in two reactions (as a reactant in re88 and as a product in re47).

$$\frac{d}{dt}s33 = v_{47} - v_{88} \tag{207}$$

# **7.32 Species** s34

Name WRKY6

Initial amount 1.2

### Charge 0

This species takes part in two reactions (as a reactant in re48 and as a product in re74).

$$\frac{d}{dt}s34 = v_{74} - v_{48} \tag{208}$$

# **7.33 Species** s35

Name WRKY6

Initial amount 1.2

#### Charge 0

This species takes part in two reactions (as a reactant in re83 and as a product in re48).

$$\frac{d}{dt}s35 = v_{48} - v_{83} \tag{209}$$

# **7.34 Species** s36

Name MYB4

Initial amount 1.5

### Charge 0

This species takes part in two reactions (as a reactant in re49 and as a product in re68).

$$\frac{d}{dt}s36 = v_{68} - v_{49} \tag{210}$$

# **7.35 Species** s37

Name MYB4

Initial amount 1.5

### Charge 0

This species takes part in two reactions (as a reactant in re81 and as a product in re49).

$$\frac{d}{dt}s37 = v_{49} - v_{81} \tag{211}$$

### **7.36 Species** s38

Name WRKY25

Initial amount 1.2

### Charge 0

This species takes part in two reactions (as a reactant in re50 and as a product in re59).

$$\frac{d}{dt}s38 = v_{59} - v_{50} \tag{212}$$

### **7.37 Species** s39

Name WRKY25

Initial amount 1.2

### Charge 0

This species takes part in two reactions (as a reactant in re87 and as a product in re50).

$$\frac{d}{dt}s39 = v_{50} - v_{87} \tag{213}$$

# **7.38 Species** s40

Name WRKY12

Initial amount 1.2

### Charge 0

This species takes part in two reactions (as a reactant in re51 and as a product in re67).

$$\frac{d}{dt}s40 = v_{67} - v_{51} \tag{214}$$

# **7.39 Species** s41

Name WRKY12

Initial amount 1.2

### Charge 0

This species takes part in two reactions (as a reactant in re82 and as a product in re51).

$$\frac{\mathrm{d}}{\mathrm{d}t} s41 = v_{51} - v_{82} \tag{215}$$

### **7.40 Species** s42

Name WRKY22

Initial amount 1.2

### Charge 0

This species takes part in three reactions (as a reactant in re52 and as a product in re60, re61).

$$\frac{\mathrm{d}}{\mathrm{d}t}s42 = v_{60} + v_{61} - v_{52} \tag{216}$$

### **7.41 Species** s43

Name WRKY22

Initial amount 1.2

### Charge 0

This species takes part in two reactions (as a reactant in re86 and as a product in re52).

$$\frac{\mathrm{d}}{\mathrm{d}t}s43 = v_{52} - v_{86} \tag{217}$$

# **7.42 Species** s44

Name WRKY28

Initial amount 1.2

# Charge 0

This species takes part in three reactions (as a reactant in re53 and as a product in re65, re66).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{s}44 = v_{65} + v_{66} - v_{53} \tag{218}$$

# **7.43 Species** s45

Name WRKY28

Initial amount 1.2

### Charge 0

This species takes part in two reactions (as a reactant in re75 and as a product in re53).

$$\frac{d}{dt}s45 = v_{53} - v_{75} \tag{219}$$

# **7.44 Species** s46

Name WRKY29

Initial amount 1.2

### Charge 0

This species takes part in three reactions (as a reactant in re54 and as a product in re62, re64).

$$\frac{\mathrm{d}}{\mathrm{d}t}s46 = v_{62} + v_{64} - v_{54} \tag{220}$$

### **7.45 Species** s47

Name WRKY29

Initial amount 1.2

### Charge 0

This species takes part in two reactions (as a reactant in re85 and as a product in re54).

$$\frac{d}{dt}s47 = v_{54} - v_{85} \tag{221}$$

# **7.46 Species** s48

Name MYB44

Initial amount 1.5

### Charge 0

This species takes part in two reactions (as a reactant in re80 and as a product in re70).

$$\frac{d}{dt}s48 = v_{70} - v_{80} \tag{222}$$

# **7.47 Species** s49

Name NAC

Initial amount 1.8

### Charge 0

This species takes part in two reactions (as a reactant in re77 and as a product in re71).

$$\frac{d}{dt}s49 = v_{71} - v_{77} \tag{223}$$

### **7.48 Species** s50

Name bZIP

Initial amount 2

### Charge 0

This species takes part in two reactions (as a reactant in re78 and as a product in re73).

$$\frac{d}{dt}s50 = v_{73} - v_{78} \tag{224}$$

# **7.49 Species** s51

Name AP2

Initial amount 2.2

### Charge 0

This species takes part in two reactions (as a reactant in re79 and as a product in re72).

$$\frac{\mathrm{d}}{\mathrm{d}t}s51 = v_{72} - v_{79} \tag{225}$$

### **7.50 Species** s52

Name Response

Initial amount 2.5

#### Charge 0

This species takes part in 14 reactions (as a product in re75, re76, re77, re78, re79, re80, re81, re82, re83, re84, re85, re86, re87, re88).

$$\frac{d}{dt}s52 = v_{75} + v_{76} + v_{77} + v_{78} + v_{79} + v_{80} + v_{81} + v_{82} + v_{83} + v_{84} + v_{85} + v_{86} + v_{87} + v_{88}$$
 (226)

#### **7.51 Species** s26

Name SIMK

Initial amount 1

#### Charge 0

This species takes part in one reaction (as a product in re36).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{s}26 = v_{36} \tag{227}$$

#### **7.52 Species** s27

Name SAMK

Initial amount 1

#### Charge 0

This species takes part in one reaction (as a product in re37).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{s}27 = v_{37} \tag{228}$$

# A Glossary of Systems Biology Ontology Terms

**SBO:0000282** dissociation constant: Equilibrium constant that measures the propensity of a larger object to separate (dissociate) reversibly into smaller components, as when a complex falls apart into its component molecules, or when a salt splits up into its component ions. The dissociation constant is usually denoted Kd and is the inverse of the affinity constant.

**SBO:0000337** association constant: Equilibrium constant that measures the propensity of two objects to assemble (associate) reversibly into a larger component. The association constant is usually denoted Ka and is the inverse of the dissociation constant.

SML2ATEX was developed by Andreas Dräger<sup>a</sup>, Hannes Planatscher<sup>a</sup>, Dieudonné M Wouamba<sup>a</sup>, Adrian Schröder<sup>a</sup>, Michael Hucka<sup>b</sup>, Lukas Endler<sup>c</sup>, Martin Golebiewski<sup>d</sup> and Andreas Zell<sup>a</sup>. Please see http://www.ra.cs.uni-tuebingen.de/software/SBML2LaTeX for more information.

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