SBML Model Report

Model name: "Qi2013 - IL-6 and IFN crosstalk model (non-competitive)"



May 6, 2016

1 General Overview

This is a document in SBML Level 2 Version 4 format. This model was created by the following two authors: Vincent Knight-Schrijver¹ and Yun-Feng Qi² at August 19th 2014 at 1:45 p.m. and last time modified at October sixth 2014 at 11:44 a.m. Table 1 provides an overview of the quantities of all components of this model.

Table 1: Number of components in this model, which are described in the following sections.

Element	Quantity	Element	Quantity
compartment types	0	compartments	4
species types	0	species	111
events	0	constraints	0
reactions	127	function definitions	11
global parameters	204	unit definitions	1
rules	2	initial assignments	0

Model Notes

Qi2013 - IL-6 and IFN crosstalk model(non-competitive)This model [BIOMD0000000543]describes the crosstalk between IFN-gamma and IL-6 induced signalling; it aims to outline mechanisms and factors that may control the interaction between both signalling pathways, discussing a role

¹EMBL-EBI, vknight@ebi.ac.uk

²National Engineering Laboratory for Druggable Gene and Protein Screening, Northeast Normal University, China, qiyf104@nenu.edu.cn

of heterodimer formation in signalling dysfunction.

To account for the possibility of different IFNR and gp130binding sites for STAT1 and STAT3, model 1 [BIOMD000000543] assumes that there is no competition between STAT1 and STAT3 for the receptor complexes (includes two extra reactions).

The reverse of this is true in model 2 [BIOMD000000544] where it generally is assumed that there is competition between STAT1 and STAT3 for the receptor complexes.

This model is described in the article: Elucidating the crosstalk mechanism between IFN-gamma and IL-6 via mathematical modelling. Qi YF, Huang YX, Wang HY, Zhang Y, Bao YL, Sun LG, Wu Y, Yu CL, Song ZB, Zheng LH, Sun Y, Wang GN, Li YX.BMC Bioinformatics 2013; 14: 41

Abstract:

BACKGROUND: Interferon-gamma (IFN-gamma) and interleukin-6 (IL-6) are multifunctional cytokines that regulate immune responses, cell proliferation, and tumour development and progression, which frequently have functionally opposing roles. The cellular responses to both cytokines are activated via the Janus kinase/signal transducer and activator of transcription (JAK/STAT) pathway. During the past 10 years, the crosstalk mechanism between the IFNgamma and IL-6 pathways has been studied widely and several biological hypotheses have been proposed, but the kinetics and detailed crosstalk mechanism remain unclear. RESULTS: Using established mathematical models and new experimental observations of the crosstalk between the IFN-gamma and IL-6 pathways, we constructed a new crosstalk model that considers three possible crosstalk levels: (1) the competition between STAT1 and STAT3 for common receptor docking sites; (2) the mutual negative regulation between SOCS1 and SOCS3; and (3) the negative regulatory effects of the formation of STAT1/3 heterodimers. A number of simulations were tested to explore the consequences of cross-regulation between the two pathways. The simulation results agreed well with the experimental data, thereby demonstrating the effectiveness and correctness of the model. CONCLUSION: In this study, we developed a crosstalk model of the IFN-gamma and IL-6 pathways to theoretically investigate their cross-regulation mechanism. The simulation experiments showed the importance of the three crosstalk levels between the two pathways. In particular, the unbalanced competition between STAT1 and STAT3 for IFNR and gp130 led to preferential activation of IFN-gamma and IL-6, while at the same time the formation of STAT1/3 heterodimers enhanced preferential signal transduction by sequestering a fraction of the activated STATs. The model provided a good explanation of the experimental observations and provided insights that may inform further research to facilitate a better understanding of the cross-regulation mechanism between the two pathways.

This model is hosted on BioModels Database and identified by: BIOMD0000000543.

To cite BioModels Database, please use: BioModels Database: An enhanced, curated and annotated resource for published quantitative kinetic models.

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2 Unit Definitions

This is an overview of five unit definitions of which four are predefined by SBML and not mentioned in the model.

2.1 Unit substance

Name substance

Definition nmol

2.2 Unit volume

Notes Litre is the predefined SBML unit for volume.

Definition 1

2.3 Unit area

Notes Square metre is the predefined SBML unit for area since SBML Level 2 Version 1.

Definition m^2

2.4 Unit length

Notes Metre is the predefined SBML unit for length since SBML Level 2 Version 1.

Definition m

2.5 Unit time

Notes Second is the predefined SBML unit for time.

Definition s

3 Compartments

This model contains four compartments.

Table 2: Properties of all compartments.

Outside

3.1 Compartment default

This is a three dimensional compartment with a constant size of one litre.

Name Extracellular

3.2 Compartment compartment_1

This is a three dimensional compartment with a constant size of one litre.

Name Cell

3.3 Compartment c2

This is a three dimensional compartment with a constant size of one litre.

Name Nucluse1

3.4 Compartment c3

This is a three dimensional compartment with a constant size of one litre.

Name Nucluse2

4 Species

This model contains 111 species. The boundary condition of six of these species is set to true so that these species' amount cannot be changed by any reaction. Section 9 provides further details and the derived rates of change of each species.

Table 3: Properties of each species.

Id	Name	Compartment	Derived Unit	Constant	Boundary Condi- tion
species_1	IL6	default	$nmol \cdot l^{-1}$	\checkmark	
species_79	IFN	default	$nmol \cdot l^{-1}$		
species_17	(IL6-gp80-gp130-JAK)2*-STAT1C- STAT3C	default	$\operatorname{nmol} \cdot 1^{-1}$		
species_67	(IL6-gp80-gp130-JAK)2*-STAT1C	default	$nmol \cdot l^{-1}$	\Box	\Box
s118	(IFN-R-JAK)2*-STAT3C	${\tt compartment_1}$	$\operatorname{nmol} \cdot 1^{-1}$		
s119	(IFN-R-JAK)2*-STAT3C*	${ t compartment}_{ t 1}$	$\operatorname{nmol} \cdot 1^{-1}$		
s120	STAT1C*-STAT3C*	${\tt compartment_1}$	$nmol \cdot l^{-1}$		
s122	STAT1N*-STAT3N*	${\tt compartment_1}$	$nmol \cdot l^{-1}$		
s126	PP2-STAT1N*-STAT3N*	${\tt compartment_1}$	$nmol \cdot l^{-1}$		
s135	PP1-STAT1C*-STAT3C*	${\tt compartment_1}$	$nmol \cdot l^{-1}$		
species_2	gp80	${\tt compartment_1}$	$nmol \cdot l^{-1}$	\square	
species_3	IL6-gp80	${\tt compartment_1}$	$nmol \cdot l^{-1}$		
${\tt species_4}$	gp130	${\tt compartment_1}$	$nmol \cdot l^{-1}$		
species_5	JAK(IFN)	${\tt compartment_1}$	$nmol \cdot l^{-1}$		
species_6	gp130-JAK	${\tt compartment_1}$	$nmol \cdot l^{-1}$		
species_7	IL6-gp80-gp130-JAK	${\tt compartment_1}$	$nmol \cdot l^{-1}$		
species_8	(IL6-gp80-gp130-JAK)2	${\tt compartment_1}$	$nmol \cdot l^{-1}$		
species_9	(IL6-gp80-gp130-JAK)2*	${\tt compartment_1}$	$nmol \cdot l^{-1}$		\Box
species_10	(IL6-gp80-gp130-JAK)2*-STAT3C	${\tt compartment_1}$	$\mathrm{nmol}\cdot\mathrm{l}^{-1}$		\Box
species_11	STAT3C	${\tt compartment_1}$	$nmol \cdot l^{-1}$		

Id	Name	Compartment	Derived Unit	Constant	Boundary Condi- tion
species_12	STAT3C*	compartment_1	$nmol \cdot 1^{-1}$		
species_13	(IL6-gp80-gp130-JAK)2*-STAT3C*	$compartment_1$	$\mathrm{nmol}\cdot\mathrm{l}^{-1}$		
species_14	(STAT3C*)2	$compartment_1$	$\mathrm{nmol}\cdot\mathrm{l}^{-1}$		
species_15	(IL6-gp80-gp130-JAK)2*-SOCS3	${\tt compartment_1}$	$\mathrm{nmol}\cdot\mathrm{l}^{-1}$		\Box
species_16	(IL6-gp80-gp130-JAK)2*-SHP2	${\tt compartment_1}$	$\mathrm{nmol}\cdot\mathrm{l}^{-1}$		\Box
species_18	STAT3C-STAT3C*	$compartment_1$	$\operatorname{nmol} \cdot 1^{-1}$		\Box
species_19	SOCS3	$compartment_1$	$\operatorname{nmol} \cdot 1^{-1}$		\Box
species_20	PP1	$compartment_1$	$\operatorname{nmol} \cdot 1^{-1}$		\Box
species_21	PP1-STAT3C*	$compartment_1$	$\operatorname{nmol} \cdot 1^{-1}$		\Box
species_22	PP1-(STAT3C*)2	$compartment_1$	$\operatorname{nmol} \cdot 1^{-1}$		\Box
species_31	mRNA-SOCS3C	$compartment_1$	$\operatorname{nmol} \cdot 1^{-1}$		
species_32	Grb2	$compartment_1$	$\mathrm{nmol}\cdot\mathrm{l}^{-1}$		
species_33	(IL6-gp80-gp130-JAK)2*-SHP2*	$compartment_1$	$\mathrm{nmol}\cdot\mathrm{l}^{-1}$		\Box
species_34	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2	compartment_1	$nmol \cdot l^{-1}$		
species_35	SOS	compartment_1	$nmol \cdot l^{-1}$		
species_36	SHP2*-Grb2	compartment_1	$nmol \cdot l^{-1}$		
species_37	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2- SOS	compartment_1	$\operatorname{nmol} \cdot 1^{-1}$		
species_38	Ras-GDP	${\tt compartment_1}$	$\mathrm{nmol}\cdot\mathrm{l}^{-1}$		\Box
species_39	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2- SOS-Ras-GDP	${\tt compartment_1}$	$\operatorname{nmol} \cdot 1^{-1}$		
species_40	Ras-GTP	$compartment_1$	$\operatorname{nmol} \cdot 1^{-1}$		\Box
species_41	Raf	${\tt compartment_1}$	$\mathrm{nmol}\cdot\mathrm{l}^{-1}$		
species_42	Raf-Ras-GTP	${ t compartment}_{ t 1}$	$\mathrm{nmol}\cdot\mathrm{l}^{-1}$		
species_43	Ras-GTP*	$compartment_1$	$\mathrm{nmol}\cdot\mathrm{l}^{-1}$		
species_44	Raf*	$compartment_1$	$nmol \cdot l^{-1}$		

Id	Name	Compartment	Derived Unit	Constant	Boundary Condi- tion
species_45	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2- SOS-Ras-GTP	compartment_1	$nmol \cdot l^{-1}$		
species_46	SHP2*-Grb2-SOS	compartment_1	$nmol \cdot l^{-1}$		
species_47	Grb2-SOS	compartment_1	$nmol \cdot l^{-1}$		
species_48	SHP2*	compartment_1	$nmol \cdot l^{-1}$		
species_49	Phosp1	$compartment_1$	$nmol \cdot l^{-1}$		
species_50	Raf*-Phosp1	$compartment_1$	$nmol \cdot l^{-1}$		\Box
species_51	MEK	$compartment_1$	$nmol \cdot l^{-1}$		
species_52	MEK-Raf*	compartment_1	$nmol \cdot l^{-1}$		\Box
species_53	MEK-P	$compartment_1$	$nmol \cdot l^{-1}$		
species_54	MEK-P-Raf*	${\tt compartment_1}$	$nmol \cdot l^{-1}$		
species_55	MEK-PP	${\tt compartment_1}$	$\operatorname{nmol} \cdot 1^{-1}$		\Box
species_56	MEK-PP-Phosp2	${\tt compartment_1}$	$nmol \cdot l^{-1}$		\Box
species_57	Phosp2	${\tt compartment_1}$	$nmol \cdot l^{-1}$		
species_58	MEK-P-Phosp2	$compartment_1$	$nmol \cdot l^{-1}$		
species_59	ERK	$compartment_1$	$nmol \cdot l^{-1}$		
species_60	ERK-MEK-PP	${\tt compartment_1}$	$nmol \cdot l^{-1}$		
species_61	ERK-P	${\tt compartment_1}$	$\operatorname{nmol} \cdot 1^{-1}$		\Box
species_62	ERK-P-MEK-PP	${\tt compartment_1}$	$\operatorname{nmol} \cdot 1^{-1}$		
species_63	ERK-PP	${\tt compartment_1}$	$nmol \cdot l^{-1}$		
species_64	Phosp3	${\tt compartment_1}$	$nmol \cdot l^{-1}$		
species_65	ERK-PP-Phosp3	${\tt compartment_1}$	$nmol \cdot l^{-1}$		
species_66	ERK-P-Phosp3	${\tt compartment_1}$	$\operatorname{nmol} \cdot 1^{-1}$		
species_74	CEBPi	${\tt compartment_1}$	$\operatorname{nmol} \cdot 1^{-1}$		
species_75	CEBP	${\tt compartment_1}$	$\mathrm{nmol}\cdot \mathrm{l}^{-1}$		\Box
species_76	CEBPn	${\tt compartment_1}$	$\operatorname{nmol} \cdot 1^{-1}$		\Box
species_78	R-JAK	$compartment_1$	$n \text{mol} \cdot l^{-1}$		

∞	Id	Name	Compartment	Derived Unit	Constant	Boundary Condi- tion
	species_80	IFN-R-JAK	compartment_1	$nmol \cdot 1^{-1}$		
	species_81	(IFN-R-JAK)2	compartment_1	$\operatorname{nmol} \cdot 1^{-1}$	\Box	\Box
	species_82	(IFN-R-JAK)2*	compartment_1	$\operatorname{nmol} \cdot 1^{-1}$	\Box	\Box
	species_83	(IFN-R-JAK)2*-STAT1C	compartment_1	$\operatorname{nmol} \cdot 1^{-1}$	\Box	
	species_84	STAT1C	compartment_1	$\operatorname{nmol} \cdot 1^{-1}$	\Box	\Box
	species_85	STAT1C*	compartment_1	$\operatorname{nmol} \cdot 1^{-1}$	\Box	
	species_86	(IFN-R-JAK)2*-STAT1C*	compartment_1	$\operatorname{nmol} \cdot 1^{-1}$	\Box	
L .	species_87	(STAT1C*)2	compartment_1	$\operatorname{nmol} \cdot 1^{-1}$	\Box	
Produced by SBML2PTFX	species_88	(IFN-R-JAK)2*-SHP2	compartment_1	$\operatorname{nmol} \cdot 1^{-1}$	\Box	\Box
duc	species_90	PP1-STAT1C*	compartment_1	$\operatorname{nmol} \cdot 1^{-1}$	\Box	
ced	species_91	STAT1C-STAT1C*	$compartment_1$	$\operatorname{nmol} \cdot 1^{-1}$	\Box	\Box
by	species_98	mRNA-SOCS1C	$compartment_1$	$\operatorname{nmol} \cdot 1^{-1}$	\Box	
8	species_99	SOCS1	$compartment_1$	$\mathrm{nmol}\cdot\mathrm{l}^{-1}$	\Box	\Box
\leq	species_100	(IFN-R-JAK)2*-SOCS1	compartment_1	$\mathrm{nmol}\cdot\mathrm{l}^{-1}$	\Box	\Box
ZE ZEI	species_101	PP1-(STAT1C*)2	compartment_1	$\mathrm{nmol}\cdot\mathrm{l}^{-1}$	\Box	\Box
Ţ.	species_104	(IFN-R-JAK)2*-STAT1C-SOCS1	compartment_1	$\mathrm{nmol}\cdot\mathrm{l}^{-1}$	\Box	
	species_105	(IFN-R-JAK)2*-STAT1C-SHP2-SOCS1	compartment_1	$\operatorname{nmol} \cdot 1^{-1}$	\Box	
	species_106	(IFN-R-JAK)2*-STAT1C-SHP2	compartment_1	$\operatorname{nmol} \cdot 1^{-1}$	\Box	
	species_107	R	$compartment_1$	$\operatorname{nmol} \cdot 1^{-1}$	\Box	
	species_108	SHP2	$compartment_1$	$\mathrm{nmol}\cdot\mathrm{l}^{-1}$	\Box	
	s137	STAT3*	compartment_1	$\mathrm{nmol}\cdot\mathrm{l}^{-1}$	\Box	
	s136	STAT1*	compartment_1	$\mathrm{nmol}\cdot\mathrm{l}^{-1}$	\Box	$\overline{\mathbf{Z}}$
	s138	(IL6-gp80-gp130-JAK)2*-STAT1	compartment_1	$\operatorname{nmol} \cdot 1^{-1}$	\Box	
	s139	(IL6-gp80-gp130-JAK)2*-STAT1*	compartment_1	$\operatorname{nmol} \cdot 1^{-1}$	\Box	\Box
	s140	JAK(IL-6)	${\tt compartment_1}$	$\mathrm{nmol}\cdot\mathrm{l}^{-1}$	\Box	\Box
	species_68	(IFN-R-JAK)2*-STAT1C-STAT3C	compartment_1	$\operatorname{nmol} \cdot 1^{-1}$	\Box	\Box
	species_92	(STAT1N*)2	c2	$nmol \cdot l^{-1}$		

Id	Name	Compartment	Derived Unit	Constant	Boundary Condi- tion
species_94	PP2-STAT1N*	c2	$nmol \cdot l^{-1}$		
species_95	STAT1N*	c2	$nmol \cdot l^{-1}$	\Box	\Box
species_96	STAT1N	c2	$nmol \cdot l^{-1}$	\Box	\Box
species_97	mRNA-SOCS1N	c2	$nmol \cdot l^{-1}$	\Box	
species_102	PP2-(STAT1N*)2	c2	$nmol \cdot l^{-1}$	\Box	
species_103	STAT1N-STAT1N*	c2	$nmol \cdot l^{-1}$	\Box	\Box
species_23	(STAT3N*)2	с3	$nmol \cdot l^{-1}$		
species_24	PP2	с3	$nmol \cdot l^{-1}$		
species_25	PP2-(STAT3N*)2	с3	$nmol \cdot l^{-1}$		
species_26	STAT3N*	с3	$nmol \cdot l^{-1}$		
species_27	PP2-STAT3N*	с3	$nmol \cdot l^{-1}$	\Box	
species_28	STAT3N	с3	$nmol \cdot l^{-1}$	\Box	
species_29	STAT3N-STAT3N*	с3	$nmol \cdot l^{-1}$	\Box	\Box
$species_30$	mRNA-SOCS3N	c3	$nmol \cdot l^{-1}$	\Box	\Box

5 Parameters

This model contains 204 global parameters.

Table 4: Properties of each parameter.

Id	Name	SBO	Value	Unit	Constant
parameter_1	k1f		0.100		$ \mathcal{L} $
parameter_2	k1r		0.050		\mathbf{Z}
parameter_3	k2f		0.100		$\overline{\mathbf{Z}}$
$parameter_4$	k2r		0.050		$\overline{\mathbf{Z}}$
parameter_5	k3f		0.020		$\overline{\mathbf{Z}}$
$parameter_6$	k3r		0.020		$ \overline{\checkmark} $
$parameter_11$	k6f		0.040		
$parameter_12$	k6r		0.200		$ \overline{\checkmark} $
$parameter_13$	k7		0.005		
$parameter_14$	k8f		0.008		
$parameter_15$	k8r		0.800		
$parameter_16$	k9		0.400		
$parameter_17$	k10f		0.005		
$parameter_{-}18$	k10r		0.500		
$parameter_27$	k16f		0.020		
$parameter_28$	k16r		0.100		
parameter_29	k17f		0.001		
$parameter_30$	k17r		0.200		
$parameter_31$	k18		0.003		
$parameter_32$	k19f		0.001		
$parameter_33$	k19r		0.200		
$parameter_34$	k20		0.003		
parameter_35	k21f		0.001		
parameter_36	k21r		0.200		
$parameter_37$	k22		0.003		
$parameter_38$	k23f		$2 \cdot 10^{-7}$		
$parameter_39$	k23r		0.200		
$parameter_40$	k24		0.005		
$parameter_41$	k25f		0.100		
$parameter_42$	k25r		0.020		
$parameter_43$	k26f		0.001		
$parameter_44$	k26r		0.200		
$parameter_45$	k27		0.005		
$parameter_46$	k28f		0.001		
$parameter_47$	k28r		0.200		
$parameter_48$	k29		0.005		
parameter_49	k30f		0.200		\square

Id	Name	SBO Value	Unit	Constant
parameter_50	k30r	$2 \cdot 10^{-7}$		
parameter_51	k31	0.050		$\overline{\mathbf{Z}}$
$parameter_52$	k32f	0.010		$ \overline{\mathscr{L}} $
parameter_53	k32r	400.000		$ \overline{\mathbf{Z}} $
$parameter_54$	k33	0.001		$ \overline{\mathbf{Z}} $
parameter_55	k34	0.010		$ \overline{\mathbf{Z}} $
parameter_56	k35f	5.000		$ \overline{\mathbf{Z}} $
parameter_57	k35r	0.100		
parameter_58	k36	$5 \cdot 10^{-4}$		
parameter_59	k37	$6 \cdot 10^{-4}$		$ \overline{\mathbf{Z}} $
$parameter_60$	k38	0.003		$ \overline{\mathbf{Z}} $
$parameter_61$	k39f	6.000		$ \overline{\mathbf{Z}} $
parameter_62	k39r	0.060		$ \overline{\mathbf{Z}} $
parameter_63	k40f	0.010		$ \overline{\mathbf{Z}} $
$parameter_64$	k40r	0.550		$ \overline{\mathbf{Z}} $
$parameter_65$	k41f	0.010		$ \overline{\mathscr{L}} $
parameter_66	k41r	0.021		$ \overline{\mathbf{Z}} $
parameter_67	k42f	0.015		$ \overline{\mathbf{Z}} $
parameter_68	k42r	1.300		$ \overline{\mathbf{Z}} $
parameter_69	k43f	0.500		$ \overline{\mathbf{Z}} $
$parameter_{-}70$	k43r	10^{-4}		
$parameter_71$	k44f	0.001		
$parameter_{-}72$	k44r	0.005		
parameter_73	k45f	1.000		
$parameter_{-}74$	k45r	$7 \cdot 10^{-4}$		$ \overline{\mathbf{Z}} $
parameter_75	k46f	0.008		$ \overline{\mathbf{Z}} $
$parameter_{-}76$	k46r	0.400		\mathbf{Z}
$parameter_{-}77$	k47f	0.023		\mathbf{Z}
$parameter_{-}78$	k47r	$2.2 \cdot 10^{-4}$		
$parameter_{-}79$	k48f	0.470		\mathbf{Z}
parameter_80	k48r	$2.45 \cdot 10^{-4}$		\mathbf{Z}
parameter_81	k49f	0.300		\mathbf{Z}
parameter_82	k49r	0.021		\mathbf{Z}
parameter_83	k50f	0.002		\mathbf{Z}
parameter_84	k50r	0.005		
parameter_85	k51f	1.700		
parameter_86	k51r	340.000		
parameter_87	k52f	0.300		
parameter_88	k52r	$9 \cdot 10^{-4}$		
parameter_89	k53f	0.010		
$parameter_90$	k53r	0.550		
$parameter_91$	k54f	0.300		

Id	Name	SBO Value	Unit	Constant
parameter_92	k54r	$9 \cdot 10^{-4}$		$ \overline{\checkmark} $
parameter_93	k55f	0.030		$ \overline{\checkmark} $
$parameter_94$	k55r	0.064		\square
$parameter_95$	k56f	0.030		\mathbf{Z}
parameter_96	k56r	0.043		
$parameter_97$	k57f	0.072		
parameter_98	k57r	0.200		
parameter_99	k58	1.000		
$\mathtt{parameter} extsf{-}100$	k59f	0.011		\mathbf{Z}
parameter- _101	k59r	0.002		
parameter- _102	k60	3.500		\mathbf{Z}
parameter- _103	k61f	0.011		Ø
parameter- _104	k61r	0.002		\square
parameter- _105	k62	2.900		\square
parameter- _106	k63f	0.014		Ø
parameter- _107	k63r	0.800		\square
parameter- _108	k64	0.058		\square
parameter- _109	k65f	$2.5\cdot10^{-4}$		Ø
parameter- _110	k65r	0.500		Ø
parameter- _111	k66	0.058		Ø
parameter- _112	k67f	$1.1 \cdot 10^{-4}$		\square
parameter- _113	k67r	0.033		
parameter- _114	k68	16.000		Ø
parameter- _115	k69f	$1.1 \cdot 10^{-4}$		Ø
parameter- _116	k69r	0.033		Ø

Id	Name	SBO Value	Unit C	Constant
parameter- _117	k70	6.700		
parameter- _118	k71f	0.014		\square
parameter- _119	k71r	0.600		\square
parameter- _120	k72	0.270		\square
parameter- _121	k73f	0.005		\square
parameter- _122	k73r	0.500		\square
parameter- _123	k74	0.300		\square
parameter- $_{ extstyle 124}$	k75f	0.234		\square
parameter- _125	k75r	20000.000		\square
parameter- _126	k76	0.039		\square
parameter- _127	k77f	0.985		\square
parameter- _128	k77r	$9 \cdot 10^{-4}$		\square
parameter- _129	kf1	0.100		\square
parameter- _130	kb1	0.050		
parameter- _131	kf2	0.020		
parameter- _132	kb2	0.020		\square
parameter- _133	kf3	0.040		\square
parameter- _134	kb3	0.200		\square
parameter- _135	kf4	0.005		\square
parameter- _136	kf5	0.008		\square
parameter- _137	kb5	0.800		

Id	Name	SBO Value	Unit	Constant
parameter- _138	kf6	0.400	0	Ø
parameter- _139	kf7	0.003	5	\square
parameter- $_{ m _140}$	kb7	0.500		\square
parameter- $_{ m _141}$	kf8	0.020	0	\square
parameter- _142	kb8	0.100	0	\square
parameter- _143	kf9	0.00	1	\square
parameter- _144	kb9	0.200		\square
parameter- _145	kf10	0.003	3	\square
parameter- _146	kf11	0.00	1	\square
parameter- _147	kb11	0.200	0	\square
parameter- _148	kf12	0.003	3	\square
parameter- _149	kf13	2 · 10 -	7	\square
parameter- _150	kb13	0.200	0	\square
parameter- _151	kf14	0.003	5	\square
parameter- _152	kf15	0.00	1	\square
parameter- _153	kb15	0.200	0	\square
parameter- _154	kf17	0.050	0	\square
parameter- _155	kf16	0.003	5	\square
parameter- $_{ extstyle 156}$	k18a	0.010	0	
parameter- _157	k18b	400.000	0	\square
parameter- _158	kf19	0.00	1	

Id	Name	SBO Value	Unit Constant
parameter- _159	kf20	0.010	Ø
parameter- _160	kf21	0.020	Ø
parameter- _161	kb21	0.100	Ø
parameter- _162	kf22	$5 \cdot 10^{-4}$	Ø
parameter- _163	kf23	$5 \cdot 10^{-4}$	Ø
parameter- _164	kf24	0.001	Ø
parameter- _165	kb24	0.200	Ø
parameter- _166	kf25	0.003	Ø
parameter- _167	kf26	0.005	Ø
parameter- _168	kb26	0.500	Ø
parameter- _169	kf27	0.001	Ø
parameter- _170	kb27	0.200	Ø
parameter- _171	kf28	0.005	Ø
parameter- _172	kf29	$2\cdot 10^{-7}$	Ø
parameter- _173	kb29	0.200	Ø
parameter- _174	kf30	0.008	Ø
parameter- _175	kb30	0.800	Ø
parameter- _176	kf31	0.001	Ø
parameter- $_{ m ar{1}77}$	kb31	0.200	Ø
parameter- _178	kf32	0.003	Ø
parameter- _179	kf33	$5 \cdot 10^{-4}$	Ø

Id	Name	SBO Valu	e Unit	Constant
parameter- _221	n_kf1	0.0	002	
parameter- _222	n_kr1	2.0	000	
parameter- _223	n_k2		200	Ø
parameter- _224	n_kf3	0.0	005	Ø
parameter- _225	n_kr3	0.:	500	\square
parameter- _226	n_kf4	0.0	002	\square
parameter- _227	n_kr4	2.0	000	
parameter- _228	n_k5	0.2	200	
parameter- _229	n_kf6	0.0	005	
parameter- _230	n_kr6	0	500	
parameter- _231	n_kf7	0.0	001	
parameter- _232	n_kr7	400.	000	
parameter- _233	n_kf8	0.0	020	
parameter- _234	n_kr8	0.	100	
parameter- _235	n_kf9	0.0	020	\square
parameter- _236	n_kr9	0.	100	
parameter- _237	n_k10	0.0	005	\square
parameter- _238	n_kf11	0.0	001	\square
parameter- _239	n_kr11	0.2	200	
parameter- _240	n_kf12	0.0	001	\square
parameter- _241	n_kr12	0.3	200	

Id	Name	SBO	Value	Unit	Constant
parameter-	n_k13		0.002		Ø
parameter- _243	n_k14		0.002		\square
parameter- _244	n_k15		0.003		Ø
parameter- _245	n_k16		0.003		Ø
$parameter_{-}7$	n_kf17		0.008		\square
parameter_8	n_kr17		0.800		$\overline{\mathbf{Z}}$
$parameter_9$	n_kf18		0.002		$ \overline{\checkmark} $
$parameter_10$	n_kr18		2.000		
$parameter_19$	n_k19		0.400		
$parameter_20$	n_k20		0.200		
$parameter_21$	n_kf21		0.002		
$parameter_22$	n_kr21		2.000		
$parameter_23$	n_kf22		0.008		$ \overline{\checkmark} $
parameter_24	n_kr22		0.800		\overline{Z}
parameter_25	n_k23		0.200		\overline{Z}
parameter_26	n_k24		0.400		$\overline{\checkmark}$

6 Function definitions

This is an overview of eleven function definitions.

6.1 Function definition function_3

Name Henri-Michaelis-Menten (irreversible)

Arguments substrate, Km, V

Mathematical Expression

$$\frac{V \cdot substrate}{Km + substrate} \tag{1}$$

6.2 Function definition function_2

Name function_4_reaction-34

Arguments parameter_55, [species_31]

Mathematical Expression

parameter_
$$55 \cdot [\text{species}_31]$$
 (2)

6.3 Function definition function_1

Name function_4_reaction-32

Arguments parameter_52, parameter_53, [species_23]

Mathematical Expression

$$\frac{\text{parameter_52} \cdot [\text{species_23}]}{\text{parameter_53} + [\text{species_23}]}$$
(3)

6.4 Function definition function_4

Name function_4_reaction-v15

Arguments vol (c2), parameter_152, parameter_153, [species_24], [species_94], [species_95]

Mathematical Expression

6.5 Function definition function_5

Name function_4_reaction-v16

Arguments vol (c2), parameter_155, [species_94]

Mathematical Expression

$$vol(c2) \cdot parameter_155 \cdot [species_94]$$
 (5)

6.6 Function definition function_6

Name function_4_reaction-v18

Arguments parameter_156, parameter_157, [species_92]

Mathematical Expression

$$\frac{\text{parameter}_156 \cdot [\text{species}_92]}{\text{parameter}_157 + [\text{species}_92]}$$
 (6)

6.7 Function definition function_7

Name function_4_reaction-v20

Arguments parameter_159, [species_98]

Mathematical Expression

parameter_
$$159 \cdot [\text{species}_98]$$
 (7)

6.8 Function definition function_8

Name function_4_reaction-v27

Arguments vol(c2), parameter_169, parameter_170, [species_102], [species_24], [species_92]

Mathematical Expression

$$vol(c2) \cdot (parameter_169 \cdot [species_24] \cdot [species_92] - parameter_170 \cdot [species_102])$$

6.9 Function definition function_9

Name function_4_reaction-v28

Arguments vol (c2), parameter_171, [species_102]

Mathematical Expression

$$vol(c2) \cdot parameter_171 \cdot [species_102]$$
 (9)

6.10 Function definition function_4_reaction_n7_1

Name function_4_reaction_n7_1

Arguments parameter_231, parameter_232, [species_92]

Mathematical Expression

$$\frac{\text{parameter}_231 \cdot [\text{species}_92]}{\text{parameter}_232 + [\text{species}_92]}$$
 (10)

6.11 Function definition function_activation

Name function_activation

Arguments v, a, b, k

Mathematical Expression

$$\frac{\mathbf{v} \cdot \mathbf{a} \cdot \mathbf{b}}{\mathbf{b} + \mathbf{k}} \tag{11}$$

7 Rules

This is an overview of two rules.

7.1 Rule s136

Rule \$136 is an assignment rule for species \$136:

$$s136 = [species_86] + 2 \cdot [species_101] + [s135] + [species_90] + 2 \cdot [species_102] + [s126] + [species_94] + 2 \cdot [species_87] + [s120] + [species_85] + [species_91] + 2 \cdot [species_92] + [s122] + [species_95] + [species_103] + [s139]$$

$$(12)$$

7.2 Rule s137

Rule \$137 is an assignment rule for species \$137:

$$s137 = [s119] + [species_13] + [s135] + 2 \cdot [species_22] + [species_21] + [s126] + 2 \cdot [species_25] + [species_27] + [s120] + [s122] + 2 \cdot [species_14] + [species_12] + [species_18] + 2 \cdot [species_23] + [species_26] + [species_29]$$

$$(13)$$

8 Reactions

This model contains 127 reactions. All reactions are listed in the following table and are subsequently described in detail. If a reaction is affected by a modifier, the identifier of this species is written above the reaction arrow.

Table 5: Overview of all reactions

N⁰	Id	Name	Reaction Equation SBO
1	re160	reaciton-2	$s140 + species_4$ $\stackrel{s140}{\longleftarrow} species_6$ $species_6$
2	reaction_1	reaction-1	species_2+species_1 species_3 species_3 species_3
3	reaction_8	reaction-10	species_9 + species_12 \(\frac{\text{species}_9, \text{ species}_12, \text{ species}_13}{\text{species}_13} \)
4	reaction_9	reaction-16	2 species_12 species_14 species_14
5	re138	reaction-17	species_9+species_108 species_9, species_108, species_16 species_16
6	re137	reaction-18	$species_{-16} \xrightarrow{species_{-16}} species_{-8} + species_{-108}$
7	${\tt reaction_12}$	reaction-19	species_12+species_20 species_12, species_20, species_21 species_21
8	${\tt reaction_14}$	reaction-20	$species_21 \xrightarrow{species_21} species_11 + species_20$
9	reaction_13	reaction-21	species_14+species_20 species_14, species_20, species_22 species_22
10	reaction_15	reaction-22	$species_{22} \xrightarrow{species_{22}} species_{18} + species_{20}$
11	reaction_16	reaction-23	species_11+species_12
12	reaction_17	reaction-24	$species_{-}14 \xrightarrow{species_{-}14} species_{-}23$
13	reaction_65	reaction-25	species_23 species_26 2 species_26
14	reaction_19	reaction-26	species_24+species_26 species_24, species_26, species_27 species_27
15	reaction_20	reaction-27	species_27 $\xrightarrow{\text{species}_27}$ species_28 + species_24

3	N⁰	Id	Name	Reaction Equation	SBO
	16	reaction_18	reaction-28	species_24 + species_23 species_24, species_23, species_25	ecies_25 species_25
	17	reaction_21	reaction-29	species 25 $\xrightarrow{\text{species} 25}$ species 24 + species 29	
	18	reaction_3	reaction-3		_7 ⇒ species_7
	19	reaction_66	reaction-30	species_29 species_26, species_28 species_28	eies_26+
	20	reaction_22	reaction-31	species_28 species_11	
D	21	reaction_23	reaction-32	$\emptyset \xrightarrow{\text{species}_23, \text{ species}_23} \text{species}_30$	
i po	22	reaction_24	reaction-33	$species_30 \xrightarrow{species_30} species_31$	
	23	reaction_25	reaction-34	$\emptyset \xrightarrow{\text{species}_31, \text{species}_31} \text{species}_19$	
3	24	reaction_26	reaction-35	species_9+species_19 species_9, species_19, speci	es_15 species_15
2	25	reaction_27	reaction-36	species_31 $\xrightarrow{\text{species}_31} \emptyset$	
ATI 1	26	reaction_28	reaction-37	$species_{-}19 \xrightarrow{species_{-}19} \emptyset$	
	27	reaction_73	reaction-38	$species_15 \xrightarrow{species_15} species_8 + species_19$	
	28	reaction_39	reaction-39	species_16 species_33 species_33	
	29	reaction_29	reaction-40	species_33 + species_32 species_33, species_32, species_32	
	30	reaction_30	reaction-41	species_35 + species_34 species_35, species_34, species_35	
	31	reaction_31	reaction-42	species_38 + species_37 species_38, species_37, species_37	
	32	reaction_32	reaction-43	species_39 species_40, species_37 species_37 species_37	ies_40+

Nº	Id	Name	Reaction Equation	SBO
33	reaction_33	reaction-44	1	species_42
34	reaction_34	reaction-45	species_42 species_43, species_44 species_44 species_44	s_43+
35	reaction_35	reaction-46	species_37 + species_43 = species_37, species_43, species_37 = species_37 = species_38	
36	reaction_36	reaction-47	species_45 species_37, species_38 species_38 species_38	
37	reaction_37	reaction-48	species_37 species_46, species_9 species_9 species_9	
38	reaction_38	reaction-49	species_46 species_47, species_48 species_48 species_48	
39	reaction_40	reaction-50	species_47 species_32, species_35 species species_35	s_32+
40	re136	reaction-51	species_48 $\xrightarrow{\text{species}_48}$ species_108	
41	reaction_42	reaction-52	species_33 species_9, species_48 species_48 species_48	_9+
42	reaction_43	reaction-53	species_32+species_48 species_32, species_48, species_48	
43	reaction_44	reaction-54	species_34 species_9, species_36 species_36 species_36	_9+
44	reaction_45	reaction-55	species_35+species_36 species_35, species_36, species_36	ies_46 ⇒ species_46
45	reaction_46	reaction-56	species_33 + species_47 species_47, species_47, species_47	ies_37 species_37
46	reaction_47	reaction-57	species_49 + species_44	ies_50 species_50

24	No	Id	Name	Reaction Equation	SBO
	47	reaction_48	reaction-58	species_50 $\xrightarrow{\text{species}_50}$ species_41 + species_49	
	48	reaction_49	reaction-59	species_44+species_51 species_51, species_51, species_51	cies_52 species_52
	49	reaction_5	reaction-6	2 species_7 species_8 species_8	
	50	reaction_50	reaction-60	species_52 $\xrightarrow{\text{species}_52}$ species_53 + species_44	
	51	reaction_51	reaction-61	species_44+species_53 species_44, species_53, species_54	species_54
	52	reaction_52	reaction-62	species_54 $\xrightarrow{\text{species}_54}$ species_55 + species_44	
Pro	53	reaction_53	reaction-63	species_55 + species_57, species_57, species_57, species_57	$\xrightarrow{\text{cies}_56} \text{species}_56$
duce	54	reaction_54	reaction-64	species_56 $\xrightarrow{\text{species}_56}$ species_57 + species_53	
d by	55	reaction_55	reaction-65	species_53 + species_57 species_57, species_57, species_50	species_58
Produced by SBML2laTEX	56	reaction_56	reaction-66	species_58 $\xrightarrow{\text{species}_58}$ species_57 + species_51	
	57	reaction_57	reaction-67	species_55 + species_59 species_59, species_59, species_59	$\xrightarrow{\text{cies}_60} \text{species}_60$
×	58	reaction_58	reaction-68	species_60 $\xrightarrow{\text{species}_60}$ species_61 + species_55	
	59	reaction_59	reaction-69	species_55 + species_61 = species_55, species_61, species_61	$\xrightarrow{\text{cies}_62} \text{species}_62$
	60	reaction_6	reaction-7	species_8 species_9 species_9	
	61	reaction_60	reaction-70	$species_62 \xrightarrow{species_62} species_63 + species_55$	
	62	reaction_64	reaction-71	species_63 + species_64, species_64, species_64, species_64	species_65
	63	reaction_61	reaction-72	species_65 $\xrightarrow{\text{species}_65}$ species_64 + species_61	
	64	reaction_62	reaction-73	species_61 + species_64, species_64, species_64, species_64	species_66
	65	reaction_63	reaction-74	$species_66 \xrightarrow{species_66} species_64 + species_59$	

N₀	Id	Name	Reaction Equation SBO	_
66	reaction_74	reaction-75	species_74 species_63, species_63, species_74 species_75	
67	reaction_75	reaction-76	$species_{-75} \xrightarrow{species_{-75}} species_{-74}$	
68	reaction_76	reaction-77	$2 \text{ species}_75 \xrightarrow{\text{species}_75, \text{ species}_76} \text{ species}_76$	
69	reaction_4	reaction-8	species_9+species_11 species_11, species_10 species_	s_10
70	reaction_7	reaction-9	$species_10 \xrightarrow{species_10} species_9 + species_12$	
71	re135	reaction-v1	species_5 + species_107 \(\frac{\text{species_107, species_78}}{\text{species_2}} \) species	ies_78
72	reaction_87	reaction-v10	$\frac{\text{species}_{88}}{\text{species}_{88}} + \frac{\text{species}_{81}}{\text{species}_{108}}$	
73	re140	reaction-v11	species_85 + species_20 species_90 species_90 species_90 species_90	ies_90
74	re141	reaction-v12	$\frac{\text{species}_90}{\text{species}_90} \xrightarrow{\text{species}_84 + \text{species}_20}$	
75	reaction_90	reaction-v13	species_84+species_85 species_91 species_85 species_91	ies_91
76	reaction_91	reaction-v14	species_87 $\xrightarrow{\text{species}_87}$ species_92	
77	re144	reaction-v15	species_95+species_24 species_94, species_95 species_95	ies_94
78	re145	reaction-v16	species_94 $\xrightarrow{\text{species}_94}$ species_96 + species_24	
79	reaction_94	reaction-v17	species_96 $\xrightarrow{\text{species}_96}$ species_84	
80	reaction_95	reaction-v18	$\emptyset \xrightarrow{\text{species}_92, \text{species}_92} \text{species}_97$	
81	reaction_96	reaction-v19	species_97 $\xrightarrow{\text{species}_97}$ species_98	
82	reaction_80	reaction-v2	species_79+species_78 species_79, species_78, species_80 species_79+species_78	ies_80
83	reaction_97	reaction-v20	$\emptyset \xrightarrow{\text{species}_98, \text{species}_98} \text{species}_99$	
84	reaction_98	reaction-v21	species_99+species_82 species_99, species_82, species_100 species_99+species_82	cies 100
	· · · · · · · · · · · · · · · · · · ·	· · · · · · · · · · · · · · · · · · ·	Special Control of the Control of th	

26	N⁰	Id	Name	Reaction Equation	SBO
	85	reaction_99	reaction-v22	species_98 $\xrightarrow{\text{species}_98} \emptyset$	
	86	reaction_100	reaction-v23	species_99 $\xrightarrow{\text{species}_99} \emptyset$	
	87	re142	reaction-v24	species_87 + species_20 species_87, species_20, spe	$\xrightarrow{\text{cies}_101} \text{species}_101$
	88	re143	reaction-v25	species_101 $\xrightarrow{\text{species}_101}$ species_91 + species_20	
	89	reaction_103	reaction-v26	2 species_95 species_92 species_92	
	90	re147	reaction-v27	species_92+species_24 species_102, species_24, sp	ecies_92 species_102
Pro	91	re148	reaction-v28	$species_102 \xrightarrow{species_102} species_103 + species_24$	
Produced by SBMLਐਨੀEX	92	reaction_106	reaction-v29	species_96+species_95 species_96, species_95, species_96	cies_103 species_103
d by	93	reaction_79	reaction-v3	2 species_80 species_81 species_81	
SBM	94	reaction_107	reaction-v30	species_84 + species_100 species_84, species_100, species_84	
	95	reaction_110	reaction-v31	species_108 + species_104	
\ <u>\</u>	96	reaction_108	reaction-v32	$species_105 \xrightarrow{species_105} species_99 + species_81 +$	
				species_84 + species_108	
	97	reaction_109	reaction-v33	$species_105 \xrightarrow{species_105} species_99 + species_106$	
	98	reaction_81	reaction-v4	species_81 species_82 species_82	
	99	reaction_82	reaction-v5	species_82 + species_84	cies_83 species_83
	100	reaction_83	reaction-v6	species_83 species_82 + species_85	
	101	reaction_84	reaction-v7	species_82 + species_85 species_82, species_85, spe	cies_86species_86
	102	reaction_85	reaction-v8	2 species_85 species_87 species_87	

N⁰	Id	Name	Reaction Equation SBO
103	reaction_86	reaction-v9	species_82+species_108
104	re112	reaction_n1	species_82+species_11 \(\frac{\text{species}_82, \text{ species}_11, \text{ s}_{118}}{\text{s}_{118}}\)
105	re117	reaction_n10	$s120 \xrightarrow{s120} s122$
106	re153	reaction_n11	species_20 + s120 \rightleftharpoons species_20, s120, s135 s135
107	re126	reaction_n12	$s122 + species_24 \xrightarrow{s122, species_24, s126} s126$
108	re113	reaction_n2	$s118 \xrightarrow{s118} species_{12} + species_{82}$
109	re114	reaction_n3	species_12+species_82 \(\frac{\text{species}_12, \text{ species}_82, \text{ s}_{119}}{\text{s}_{119}} \)
110	re157	reaction_n4	species_9 + species_84 species_9, species_84, s138 s138
111	re158	reaction_n5	$s138 \xrightarrow{s138} species_9 + species_85$
112	re159	reaction_n6	species_85 + species_9 species_9, s139 s139
113	re116	reaction_n7	$\emptyset \xrightarrow{\text{species}_92, \text{species}_92} \text{species}_30$
114	re115	reaction_n8	species_12+species_85 \(\square \) species_85, s120
115	re121	reaction_n9	species_26+species_95 \(\square \) species_95, s122 \(\square \) s122
116	re161	reaction_n13	$s135 \xrightarrow{s135} species_20 + species_12 + species_84$
117	re162	reaction_n14	$s135 \xrightarrow{s135}$ species_25 + species_12 + species_54 $s135 \xrightarrow{s135}$ species_85 + species_11 + species_20
118	re163	reaction_n15	$s126 \xrightarrow{s126} species_26 + species_24 + species_96$
119	re164	reaction_n16	$s126 \xrightarrow{s126} species_95 + species_28 + species_24$
120	reaction_2	reaction_n17	s118 + species 84 species 84, species 68
121	reaction_10	reaction_n18	species_83 + species_11

N⁰	Id	Name	Reaction Equation	SBO
122	reaction_11	reaction_n19	$species_68 \xrightarrow{species_68} s118 + species_85$	
123	reaction_41	reaction_n20	$species_68 \xrightarrow{species_68} species_83 + species_12$	
124	reaction_67	reaction_n21	species_10+species_84 species_10, species_84,	
125	reaction_68	reaction_n22	species_67+species_11 species_67, species_11,	species_17 species_17
126	reaction_69	reaction_n23	$species_17 \xrightarrow{species_17} species_10 + species_85$	
127	reaction_70	reaction_n24	$species_17 \xrightarrow{species_17} species_67 + species_12$	

8.1 Reaction re160

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaciton-2

Reaction equation

$$s140 + species_4 \xrightarrow{s140, species_4, species_6} species_6$$
 (14)

Reactants

Table 6: Properties of each reactant.

Id	Name	SBO
s140	JAK(IL-6)	
${\tt species_4}$	gp130	

Modifiers

Table 7: Properties of each modifier.

Id	Name	SBO
s140	JAK(IL-6)	
${ t species_4}$	gp130	
species_6	gp130-JAK	

Product

Table 8: Properties of each product.

Id	Name	SBO
species_6	gp130-JAK	

Kinetic Law

Derived unit contains undeclared units

$$v_1 = \text{vol}(\text{compartment_1}) \cdot (\text{parameter_3} \cdot [\text{s140}] \cdot [\text{species_4}] - \text{parameter_4} \cdot [\text{species_6}])$$
 (15)

8.2 Reaction reaction_1

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-1

Reaction equation

$$species_2 + species_1 \xrightarrow{species_2, species_1, species_3} species_3$$
 (16)

Reactants

Table 9: Properties of each reactant.

Id	Name	SBO
species_2 species_1	gp80 IL6	

Modifiers

Table 10: Properties of each modifier.

Id	Name	SBO
species_2	gp80	
${ t species_1}$	IL6	
species_3	IL6-gp80	

Product

Table 11: Properties of each product.

Id	Name	SBO
species_3	IL6-gp80	

Kinetic Law

Derived unit contains undeclared units

$$v_2 = \text{parameter_1} \cdot [\text{species_2}] \cdot [\text{species_1}] - \text{parameter_2} \cdot [\text{species_3}]$$
 (17)

8.3 Reaction reaction_8

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-10

Reaction equation

Reactants

Table 12: Properties of each reactant.

Id	Name	SBO
species_9 species_12	(IL6-gp80-gp130-JAK)2* STAT3C*	

Modifiers

Table 13: Properties of each modifier.

Id	Name	SBO
species_12	(IL6-gp80-gp130-JAK)2* STAT3C* (IL6-gp80-gp130-JAK)2*-STAT3C*	

Product

Table 14: Properties of each product.

Id	Name	SBO
species_13	(IL6-gp80-gp130-JAK)2*-STAT3C*	

Kinetic Law

Derived unit contains undeclared units

$$v_3 = \text{vol} (\text{compartment}_1) \cdot (\text{parameter}_17 \cdot [\text{species}_9] \cdot [\text{species}_12] - \text{parameter}_18 \cdot [\text{species}_13])$$
(19)

8.4 Reaction reaction_9

This is a reversible reaction of one reactant forming one product influenced by two modifiers.

Name reaction-16

Reaction equation

Reactant

Table 15: Properties of each reactant.

Id	Name	SBO
species_12	STAT3C*	

Modifiers

Table 16: Properties of each modifier.

Id	Name	SBO
species_12	STAT3C*	
species_14	(STAT3C*)2	

Product

Table 17: Properties of each product.

		<u> </u>
Id	Name	SBO
species_14	(STAT3C*)2	

Kinetic Law

Derived unit contains undeclared units

$$v_4 = \text{vol}\left(\text{compartment_1}\right) \cdot \left(\text{parameter_27} \cdot \left[\text{species_12}\right]^2 - \text{parameter_28} \cdot \left[\text{species_14}\right]\right) \ \ (21)$$

8.5 Reaction re138

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-17

Reaction equation

Reactants

Table 18: Properties of each reactant.

Id	Name	SBO
species_9 species_108	(IL6-gp80-gp130-JAK)2* SHP2	

Modifiers

Table 19: Properties of each modifier.

Id	Name	SBO
species_9 species_108 species_16	(IL6-gp80-gp130-JAK)2* SHP2 (IL6-gp80-gp130-JAK)2*-SHP2	

Product

Table 20: Properties of each product.

Id	Name	SBO
species_16	(IL6-gp80-gp130-JAK)2*-SHP2	

Kinetic Law

Derived unit contains undeclared units

$$v_5 = \text{vol} (\text{compartment_1})$$

 $\cdot (\text{parameter_29} \cdot [\text{species_9}] \cdot [\text{species_108}] - \text{parameter_30} \cdot [\text{species_16}])$ (23)

8.6 Reaction re137

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-18

Reaction equation

$$species_16 \xrightarrow{species_16} species_8 + species_108$$
 (24)

Reactant

Table 21: Properties of each reactant.

Id	Name	SBO
species_16	(IL6-gp80-gp130-JAK)2*-SHP2	

Modifier

Table 22: Properties of each modifier.

Id	Name	SBO
species_16	(IL6-gp80-gp130-JAK)2*-SHP2	

Products

Table 23: Properties of each product.

Id	Name	SBO
species_8 species_108	(IL6-gp80-gp130-JAK)2 SHP2	

Kinetic Law

Derived unit contains undeclared units

$$v_6 = \text{vol}(\text{compartment_1}) \cdot \text{parameter_31} \cdot [\text{species_16}]$$
 (25)

8.7 Reaction reaction_12

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-19

Reaction equation

Reactants

Table 24: Properties of each reactant.

Id	Name	SBO
species_12 species_20	STAT3C* PP1	

Modifiers

Table 25: Properties of each modifier.

Id	Name	SBO
species_12	STAT3C*	
species_20	PP1	
${\tt species_21}$	PP1-STAT3C*	

Product

Table 26: Properties of each product.

Id	Name	SBO
species_21	PP1-STAT3C*	

Kinetic Law

Derived unit contains undeclared units

$$v_7 = \text{vol} (\text{compartment_1})$$

 $\cdot (\text{parameter_32} \cdot [\text{species_12}] \cdot [\text{species_20}] - \text{parameter_33} \cdot [\text{species_21}])$ (27)

8.8 Reaction reaction_14

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-20

Reaction equation

$$species_21 \xrightarrow{species_21} species_11 + species_20$$
 (28)

Reactant

Table 27: Properties of each reactant.

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Id	Name	SBO
species_21	PP1-STAT3C*	

Table 28: Properties of each modifier.

Id	Name	SBO
species_21	PP1-STAT3C*	

Products

Table 29: Properties of each product.

Id	Name	SBO
species_11	STAT3C	
species_20	PP1	

Kinetic Law

Derived unit contains undeclared units

$$v_8 = \text{vol}(\text{compartment}_1) \cdot \text{parameter}_34 \cdot [\text{species}_21]$$
 (29)

8.9 Reaction reaction_13

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-21

Reaction equation

Table 30: Properties of each reactant.

Id	Name	SBO
species_14 species_20	(STAT3C*)2 PP1	

Table 31: Properties of each modifier.

Id	Name	SBO
species_14	(STAT3C*)2	
species_20	PP1	
species_22	PP1-(STAT3C*)2	

Product

Table 32: Properties of each product.

Id	Name	SBO
species_22	PP1-(STAT3C*)2	

Kinetic Law

Derived unit contains undeclared units

$$v_9 = \text{vol} (\text{compartment_1})$$

 $\cdot (\text{parameter_35} \cdot [\text{species_14}] \cdot [\text{species_20}] - \text{parameter_36} \cdot [\text{species_22}])$ (31)

8.10 Reaction reaction_15

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-22

Reaction equation

$$species_{22} \xrightarrow{species_{22}} species_{18} + species_{20}$$
 (32)

Table 33: Properties of each reactant.

Id	Name	SBO
species_22	PP1-(STAT3C*)2	

Table 34: Properties of each modifier.

Id	Name	SBO
species_22	PP1-(STAT3C*)2	

Products

Table 35: Properties of each product.

Id	Name	SBO
species_18 species_20	STAT3C-STAT3C* PP1	

Kinetic Law

Derived unit contains undeclared units

$$v_{10} = \text{vol}(\text{compartment_1}) \cdot \text{parameter_37} \cdot [\text{species_22}]$$
 (33)

8.11 Reaction reaction_16

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-23

Reaction equation

$$species_11 + species_12 \xrightarrow{species_11, species_12, species_18} species_18$$
 (34)

Table 36: Properties of each reactant.

Id	Name	SBO
species_11	STAT3C	
species_12	STAT3C*	

Table 37: Properties of each modifier.

Id	Name	SBO
species_11 species_12 species_18		

Product

Table 38: Properties of each product.

Id	Name	SBO
species_18	STAT3C-STAT3C*	

Kinetic Law

Derived unit contains undeclared units

$$v_{11} = \text{vol} (\text{compartment}_1)$$

 $\cdot (\text{parameter}_38 \cdot [\text{species}_11] \cdot [\text{species}_12] - \text{parameter}_39 \cdot [\text{species}_18])$ (35)

8.12 Reaction reaction_17

This is an irreversible reaction of one reactant forming one product influenced by one modifier.

Name reaction-24

Reaction equation

$$species_{14} \xrightarrow{species_{14}} species_{23}$$
 (36)

Table 39: Properties of each reactant	Table 39:	Properties	of each	reactant.
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Tueste est Frep	•	
Id	Name	SBO
species_14	(STAT3C*)2	

Table 40: Properties of each modifier.

Id	Name	SBO
species_14	(STAT3C*)2	

Product

Table 41: Properties of each product.

Id	Name	SBO
species_23	(STAT3N*)2	

Kinetic Law

Derived unit contains undeclared units

$$v_{12} = \text{parameter} \cdot 40 \cdot [\text{species} \cdot 14]$$
 (37)

8.13 Reaction reaction_65

This is a reversible reaction of one reactant forming one product influenced by two modifiers.

Name reaction-25

Reaction equation

$$species_23 \xrightarrow{species_23, species_26} 2 species_26$$
 (38)

Table 42: Properties of each reactant.

Id	Name	SBO
species_23	(STAT3N*)2	

Table 43: Properties of each modifier.

Id	Name	SBO
species_23 species_26	(STAT3N*)2 STAT3N*	

Product

Table 44: Properties of each product.

Id	Name	SBO
species_26	STAT3N*	

Kinetic Law

Derived unit contains undeclared units

$$v_{13} = \text{vol}(c3) \cdot (\text{parameter_41} \cdot [\text{species_23}] - \text{parameter_42} \cdot [\text{species_26}]^2)$$
 (39)

8.14 Reaction reaction_19

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-26

Reaction equation

$$species_24 + species_26 \xrightarrow{species_24, species_26, species_27} species_27 \tag{40}$$

Reactants

Table 45: Properties of each reactant.

Id	Name	SBO
species_24	PP2	
species_26	STAT3N*	

Modifiers

42

Table 46: Properties of each modifier.

Id	Name	SBO
species_24 species_26 species_27		

Table 47: Properties of each product.

Id	Name	SBO
species_27	PP2-STAT3N*	

Kinetic Law

Derived unit contains undeclared units

$$v_{14} = \text{vol}(c3) \cdot (\text{parameter_43} \cdot [\text{species_24}] \cdot [\text{species_26}] - \text{parameter_44} \cdot [\text{species_27}])$$
 (41)

8.15 Reaction reaction_20

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-27

Reaction equation

$$species_27 \xrightarrow{species_27} species_28 + species_24 \tag{42}$$

Reactant

Table 48: Properties of each reactant.

Id	Name	SBO
species_27	PP2-STAT3N*	

Table 49: Properties of each modifier.

	Name	
Id	Name	SBO
species_27	PP2-STAT3N*	

Table 50: Properties of each product.

Id	Name	SBO
species_28 species_24	STAT3N PP2	

Kinetic Law

Derived unit contains undeclared units

$$v_{15} = \text{vol}(c3) \cdot \text{parameter} \cdot 45 \cdot [\text{species} \cdot 27]$$
 (43)

8.16 Reaction reaction_18

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-28

Reaction equation

Reactants

Table 51: Properties of each reactant.

Id	Name	SBO
- F	PP2	
species_23	(STAT3N*)2	

Table 52: Properties of each modifier.

	I	
Id	Name	SBO
species_24 species_23 species_25		

Table 53: Properties of each product.

Id	Name	SBO
species_25	PP2-(STAT3N*)2	

Kinetic Law

Derived unit contains undeclared units

$$v_{16} = \text{vol}(c3) \cdot (\text{parameter_46} \cdot [\text{species_24}] \cdot [\text{species_23}] - \text{parameter_47} \cdot [\text{species_25}])$$
 (45)

8.17 Reaction reaction_21

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-29

Reaction equation

$$species_25 \xrightarrow{species_25} species_24 + species_29 \tag{46}$$

Reactant

Table 54: Properties of each reactant.

Id	Name	SBO
species_25	PP2-(STAT3N*)2	

Table 55: Properties of each modifier.

Id	Name	SBO
species_25	PP2-(STAT3N*)2	

Table 56: Properties of each product.

rue reconstruction of the product.		
Id	Name	SBO
species_24 species_29	PP2 STAT3N-STAT3N*	

Kinetic Law

Derived unit contains undeclared units

$$v_{17} = \text{vol}(c3) \cdot \text{parameter} \cdot 48 \cdot [\text{species} \cdot 25]$$
 (47)

8.18 Reaction reaction_3

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-3

Reaction equation

$$species_3 + species_6 \xrightarrow{species_3, species_6, species_7} species_7$$
 (48)

Reactants

Table 57: Properties of each reactant.

Id	Name	SBO
species_3	IL6-gp80	
species_6	gp130-JAK	

Table 58: Properties of each modifier.

Tuble 50. I roperties of each modifier.		
Id	Name	SBO
species_3 species_6 species_7	gp130-JAK	

Table 59: Properties of each product.

Id	Name	SBO
species_7	IL6-gp80-gp130-JAK	

Kinetic Law

Derived unit contains undeclared units

$$v_{18} = \text{vol} (\text{compartment_1}) \cdot (\text{parameter_5} \cdot [\text{species_3}] \cdot [\text{species_6}] - \text{parameter_6} \cdot [\text{species_7}])$$
(49)

8.19 Reaction reaction_66

This is a reversible reaction of one reactant forming two products influenced by three modifiers.

Name reaction-30

Reaction equation

Reactant

Table 60: Properties of each reactant.

Id	Name	SBO
species_29	STAT3N-STAT3N*	

Table 61: Properties of each modifier.

Id	Name	SBO
species_29 species_26	STAT3N-STAT3N*	
species_28		

Table 62: Properties of each product.

Id	Name	SBO
species_26	STAT3N*	
species_28	STAT3N	

Kinetic Law

Derived unit contains undeclared units

$$v_{19} = \text{vol}(c3) \cdot (\text{parameter_49} \cdot [\text{species_29}] - \text{parameter_50} \cdot [\text{species_26}] \cdot [\text{species_28}])$$
 (51)

8.20 Reaction reaction_22

This is an irreversible reaction of one reactant forming one product influenced by one modifier.

Name reaction-31

Reaction equation

$$species_{28} \xrightarrow{species_{28}} species_{11}$$
 (52)

Reactant

Table 63: Properties of each reactant.

Id	Name	SBO
species_28	STAT3N	

Table 64: Properties of each modifier.

Id	Name	SBO
species_28	STAT3N	

Table 65: Properties of each product.

Id	Name	SBO
species_11	STAT3C	

Kinetic Law

Derived unit contains undeclared units

$$v_{20} = \text{parameter} \cdot 51 \cdot [\text{species} \cdot 28]$$
 (53)

8.21 Reaction reaction_23

This is an irreversible reaction of no reactant forming one product influenced by two modifiers.

Name reaction-32

Reaction equation

$$\emptyset \xrightarrow{\text{species}_23, \text{ species}_23} \text{species}_30$$
 (54)

Modifiers

Table 66: Properties of each modifier.

Id	Name	SBO
-	(STAT3N*)2 (STAT3N*)2	

Product

Table 67: Properties of each product.

Id	Name	SBO
species_30	mRNA-SOCS3N	

Kinetic Law

Derived unit contains undeclared units

$$v_{21} = \text{vol}(c3) \cdot \text{function} \text{1 (parameter} \text{52, parameter} \text{53, [species} \text{23])}$$
 (55)

$$function_1 (parameter_52, parameter_53, [species_23]) = \frac{parameter_52 \cdot [species_23]}{parameter_53 + [species_23]}$$
 (56)

$$function_1 (parameter_52, parameter_53, [species_23]) = \frac{parameter_52 \cdot [species_23]}{parameter_53 + [species_23]}$$
 (57)

8.22 Reaction reaction_24

This is an irreversible reaction of one reactant forming one product influenced by one modifier.

Name reaction-33

Reaction equation

$$species_30 \xrightarrow{species_30} species_31$$
 (58)

Reactant

Table 68: Properties of each reactant.

Id	Name	SBO
species_30	mRNA-SOCS3N	

Table 69: Properties of each modifier.

Id	Name	SBO
species_30	mRNA-SOCS3N	

Table 70: Properties of each product.

Id	Name	SBO
species_31	mRNA-SOCS3C	

Kinetic Law

Derived unit contains undeclared units

$$v_{22} = \text{parameter} \cdot 54 \cdot [\text{species} \cdot 30]$$
 (59)

8.23 Reaction reaction_25

This is an irreversible reaction of no reactant forming one product influenced by two modifiers.

Name reaction-34

Reaction equation

$$\emptyset \xrightarrow{\text{species}_31, \text{ species}_31} \text{species}_19$$
 (60)

Modifiers

Table 71: Properties of each modifier.

Id	Name	SBO
species_31	mRNA-SOCS3C	
species_31	mRNA-SOCS3C	

Product

Table 72: Properties of each product.

Id	Name	SBO
species_19	SOCS3	

Kinetic Law

Derived unit contains undeclared units

$$v_{23} = \text{vol} (\text{compartment_1}) \cdot \text{function_2} (\text{parameter_55}, [\text{species_31}])$$
 (61)

function_2 (parameter_55, [species_31]) = parameter_55
$$\cdot$$
 [species_31] (63)

8.24 Reaction reaction_26

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-35

Reaction equation

$$species_9 + species_19 \xrightarrow{species_9, species_19, species_15} species_15$$
 (64)

Reactants

Table 73: Properties of each reactant.

Id	Name	SBO
species_9 species_19	(IL6-gp80-gp130-JAK)2* SOCS3	

Modifiers

Table 74: Properties of each modifier.

Id	Name	SBO
species_9 species_19	(IL6-gp80-gp130-JAK)2* SOCS3	
•	(IL6-gp80-gp130-JAK)2*-SOCS3	

Product

Table 75: Properties of each product.

Id	Name	SBO
species_15	(IL6-gp80-gp130-JAK)2*-SOCS3	

Kinetic Law

Derived unit contains undeclared units

$$v_{24} = \text{vol} (\text{compartment_1})$$

 $\cdot (\text{parameter_56} \cdot [\text{species_9}] \cdot [\text{species_19}] - \text{parameter_57} \cdot [\text{species_15}])$ (65)

8.25 Reaction reaction_27

This is an irreversible reaction of one reactant forming no product influenced by one modifier.

Name reaction-36

Reaction equation

$$species_31 \xrightarrow{species_31} \emptyset$$
 (66)

Reactant

Table 76: Properties of each reactant.

Id	Name	SBO
species_31	mRNA-SOCS3C	

Modifier

Table 77: Properties of each modifier.

	speries or each mod	***********
Id	Name	SBO
species_31	mRNA-SOCS3C	

Kinetic Law

Derived unit contains undeclared units

$$v_{25} = \text{vol}(\text{compartment_1}) \cdot \text{parameter_58} \cdot [\text{species_31}]$$
 (67)

8.26 Reaction reaction_28

This is an irreversible reaction of one reactant forming no product influenced by one modifier.

Name reaction-37

Reaction equation

$$species_{1}9 \xrightarrow{species_{1}9} \emptyset$$
 (68)

Reactant

Table 78: Properties of each reactant.

Id	Name	SBO
species_19	SOCS3	

Modifier

Table 79: Properties of each modifier.

Id	Name	SBO
species_19	SOCS3	

Kinetic Law

Derived unit contains undeclared units

$$v_{26} = \text{vol}(\text{compartment}_{-1}) \cdot \text{parameter}_{-59} \cdot [\text{species}_{-19}]$$
 (69)

8.27 Reaction reaction_73

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-38

Reaction equation

$$species_{-}15 \xrightarrow{species_{-}15} species_{-}8 + species_{-}19$$
 (70)

Table 80: Properties of each reactant.

Id	Name	SBO
species_15	(IL6-gp80-gp130-JAK)2*-SOCS3	

Table 81: Properties of each modifier.

	· · · · · · · · · · · · · · · · · · ·	
Id	Name	SBO
	1 (01110	
species_15	(IL6-gp80-gp130-JAK)2*-SOCS3	

Products

Table 82: Properties of each product.

Id	Name	SBO
species_8 species_19	(IL6-gp80-gp130-JAK)2 SOCS3	

Kinetic Law

Derived unit contains undeclared units

$$v_{27} = \text{vol}(\text{compartment_1}) \cdot \text{parameter_60} \cdot [\text{species_15}]$$
 (71)

8.28 Reaction reaction_39

This is a reversible reaction of one reactant forming one product influenced by two modifiers.

Name reaction-39

Reaction equation

Reactant

Table 83: Properties of each reactant.

Id	Name	SBO
species_16	(IL6-gp80-gp130-JAK)2*-SHP2	

Table 84: Properties of each modifier.

Id	Name	SBO
_	(IL6-gp80-gp130-JAK)2*-SHP2 (IL6-gp80-gp130-JAK)2*-SHP2*	

Table 85: Properties of each product.

Id	Name	SBO
species_33	(IL6-gp80-gp130-JAK)2*-SHP2*	

Kinetic Law

Derived unit contains undeclared units

$$v_{28} = \text{vol} (\text{compartment_1}) \cdot (\text{parameter_61} \cdot [\text{species_16}] - \text{parameter_62} \cdot [\text{species_33}])$$
 (73)

8.29 Reaction reaction_29

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-40

Reaction equation

Reactants

Table 86: Properties of each reactant.

Id	Name	SBO
species_33 species_32	(IL6-gp80-gp130-JAK)2*-SHP2* Grb2	

Table 87: Properties of each modifier.

Id	Name	SBO
-	(IL6-gp80-gp130-JAK)2*-SHP2*	
species_32	Grb2	
species_34	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2	

Table 88: Properties of each product.

Id	Name	SBO
species_34	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2	

Kinetic Law

Derived unit contains undeclared units

$$v_{29} = \text{vol} (\text{compartment}_1)$$

$$\cdot (\text{parameter}_63 \cdot [\text{species}_33] \cdot [\text{species}_32] - \text{parameter}_64 \cdot [\text{species}_34])$$
(75)

8.30 Reaction reaction_30

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-41

Reaction equation

$$species_35 + species_34 \xrightarrow{species_35, species_34, species_37} species_37$$
 (76)

Reactants

Table 89: Properties of each reactant.

Id	Name	SBO
species_35	SOS	
species_34	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2	

Table 90: Properties of each modifier.

Id	Name	SBO
-	SOS (IL6-gp80-gp130-JAK)2*-SHP2*-Grb2 (IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS	

Table 91: Properties of each product.

Id	Name	SBO
species_37	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS	

Kinetic Law

Derived unit contains undeclared units

$$v_{30} = \text{vol} (\text{compartment_1})$$

 $\cdot (\text{parameter_65} \cdot [\text{species_35}] \cdot [\text{species_34}] - \text{parameter_66} \cdot [\text{species_37}])$ (77)

8.31 Reaction reaction_31

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-42

Reaction equation

$$species_38 + species_37 \xrightarrow{species_38, species_37, species_39} species_39$$
 (78)

Reactants

Table 92: Properties of each reactant.

Id	Name	SBO
species_38 species_37	Ras-GDP (IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS	

Table 93: Properties of each modifier.

Id	Name	SBO
-	Ras-GDP (IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS (IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS-Ras-GDP	

Table 94: Properties of each product.

Id	Name	SBO
species_39	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS-Ras-GDP	

Kinetic Law

Derived unit contains undeclared units

$$v_{31} = \text{vol} (\text{compartment}_1)$$

 $\cdot (\text{parameter}_67 \cdot [\text{species}_38] \cdot [\text{species}_37] - \text{parameter}_68 \cdot [\text{species}_39])$ (79)

8.32 Reaction reaction_32

This is a reversible reaction of one reactant forming two products influenced by three modifiers.

Name reaction-43

Reaction equation

Reactant

Table 95: Properties of each reactant.

Id	Name	SBO
species_39	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS-Ras-GDP	

Table 96: Properties of each modifier.

Id	Name	SBO
species_40	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS-Ras-GDP Ras-GTP (IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS	

Table 97: Properties of each product.

Id	Name	SBO
species_40	Ras-GTP	
species_37	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS	

Kinetic Law

Derived unit contains undeclared units

$$v_{32} = \text{vol}(\text{compartment_1}) \\ \cdot (\text{parameter_69} \cdot [\text{species_39}] - \text{parameter_70} \cdot [\text{species_40}] \cdot [\text{species_37}])$$
(81)

8.33 Reaction reaction_33

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-44

Reaction equation

Table 98: Properties of each reactant.

Id	Name	SBO
species_40		
species_41	Kai	

Table 99: Properties of each modifier.

Id	Name	SBO
1	Ras-GTP Raf Raf-Ras-GTP	

Product

Table 100: Properties of each product.

Id	Name	SBO
species_42	Raf-Ras-GTP	

Kinetic Law

Derived unit contains undeclared units

$$v_{33} = \text{vol} (\text{compartment}_1)$$

 $\cdot (\text{parameter}_71 \cdot [\text{species}_40] \cdot [\text{species}_41] - \text{parameter}_72 \cdot [\text{species}_42])$ (83)

8.34 Reaction reaction_34

This is a reversible reaction of one reactant forming two products influenced by three modifiers.

Name reaction-45

Reaction equation

Table 101: Properties of each reactant.

Id	Name	SBO
species_42	Raf-Ras-GTP	

Table 102: Properties of each modifier.

Id	Name	SBO
species_42 species_43 species_44		

Products

Table 103: Properties of each product.

Id	Name	SBO
species_43 species_44		

Kinetic Law

Derived unit contains undeclared units

$$v_{34} = \text{vol} (\text{compartment_1})$$

$$\cdot (\text{parameter_73} \cdot [\text{species_42}] - \text{parameter_74} \cdot [\text{species_43}] \cdot [\text{species_44}])$$
(85)

8.35 Reaction reaction_35

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-46

Reaction equation

Table 104: Properties of each reactant.

Id	Name	SBO
species_37 species_43	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS Ras-GTP*	

Table 105: Properties of each modifier.

Id	Name	SBO
species_37 species_43	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS Ras-GTP*	
species_45	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS-Ras-GTP	

Product

Table 106: Properties of each product.

Id	Name	SBO
species_45	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS-Ras-GTP	

Kinetic Law

Derived unit contains undeclared units

$$v_{35} = \text{vol} (\text{compartment}_1)$$

$$\cdot (\text{parameter}_75 \cdot [\text{species}_37] \cdot [\text{species}_43] - \text{parameter}_76 \cdot [\text{species}_45])$$
(87)

8.36 Reaction reaction_36

This is a reversible reaction of one reactant forming two products influenced by three modifiers.

Name reaction-47

Reaction equation

Table 107: Properties of each reactant.

Id	Name	SBO
species_45	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS-Ras-GTP	

Table 108: Properties of each modifier.

Id	Name	SBO
-	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS-Ras-GTP (IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS Ras-GDP	

Products

Table 109: Properties of each product.

Id	Name	SBO
species_37 species_38	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS Ras-GDP	

Kinetic Law

Derived unit contains undeclared units

$$v_{36} = \text{vol}(\text{compartment_1})$$

 $\cdot (\text{parameter_77} \cdot [\text{species_45}] - \text{parameter_78} \cdot [\text{species_37}] \cdot [\text{species_38}])$ (89)

8.37 Reaction reaction_37

This is a reversible reaction of one reactant forming two products influenced by three modifiers.

Name reaction-48

Reaction equation

$$species_37 \xrightarrow{species_37, species_46, species_9} species_46 + species_9$$
 (90)

Table 110: Properties of each reactant.

	Tuest Tree tree er euch reuctum.	
Id	Name	SBO
species_37	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS	

Table 111: Properties of each modifier.

Id	Name	SBO
-	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS SHP2*-Grb2-SOS (IL6-gp80-gp130-JAK)2*	

Products

Table 112: Properties of each product.

	1 1	
Id	Name	SBO
species_46 species_9	SHP2*-Grb2-SOS (IL6-gp80-gp130-JAK)2*	

Kinetic Law

Derived unit contains undeclared units

$$v_{37} = \text{vol} (\text{compartment_1})$$

 $\cdot (\text{parameter_79} \cdot [\text{species_37}] - \text{parameter_80} \cdot [\text{species_46}] \cdot [\text{species_9}])$ (91)

8.38 Reaction reaction_38

This is a reversible reaction of one reactant forming two products influenced by three modifiers.

Name reaction-49

Reaction equation

$$species_46 \xrightarrow{species_46, species_47, species_48} species_47 + species_48 \tag{92}$$

Table 113: Properties of each reactant.

Tuesto Tree Portines er euch reuseum.		
Id	Name	SBO
species_46	SHP2*-Grb2-SOS	

Table 114: Properties of each modifier.

Id	Name	SBO
<pre>species_46 species_47 species_48</pre>		

Products

Table 115: Properties of each product.

Id	Name	SBO
species_47 species_48	Grb2-SOS SHP2*	

Kinetic Law

Derived unit contains undeclared units

$$v_{38} = \text{vol} (\text{compartment}_1)$$

 $\cdot (\text{parameter}_81 \cdot [\text{species}_46] - \text{parameter}_82 \cdot [\text{species}_47] \cdot [\text{species}_48])$ (93)

8.39 Reaction reaction_40

This is a reversible reaction of one reactant forming two products influenced by three modifiers.

Name reaction-50

Reaction equation

Table 116: Properties of each reactant.

Id	Name	SBO
species_47	Grb2-SOS	

Table 117: Properties of each modifier.

Id	Name	SBO
species_47	Grb2-SOS	
species_32	Grb2	
species_35	SOS	

Products

Table 118: Properties of each product.

Id	Name	SBO
species_32 species_35		

Kinetic Law

Derived unit contains undeclared units

$$v_{39} = \text{vol} (\text{compartment}_1)$$

 $\cdot (\text{parameter}_83 \cdot [\text{species}_47] - \text{parameter}_84 \cdot [\text{species}_32] \cdot [\text{species}_35])$ (95)

8.40 Reaction re136

This is an irreversible reaction of one reactant forming one product influenced by one modifier.

Name reaction-51

Reaction equation

$$species_48 \xrightarrow{species_48} species_108$$
 (96)

Table 119: Properties of each reactant.

Id	Name	SBO
species_48	SHP2*	

Table 120: Properties of each modifier.

Id	Name	SBO
species_48	SHP2*	

Product

Table 121: Properties of each product.

Id	Name	SBO
species_108	SHP2	

Kinetic Law

Derived unit contains undeclared units

$$v_{40} = \text{vol} (\text{compartment_1}) \cdot \text{function_3} ([\text{species_48}], \text{parameter_86}, \text{parameter_85})$$
 (97)

$$function_3 (substrate, Km, V) = \frac{V \cdot substrate}{Km + substrate}$$
 (98)

$$function_3 (substrate, Km, V) = \frac{V \cdot substrate}{Km + substrate}$$
 (99)

8.41 Reaction reaction_42

This is a reversible reaction of one reactant forming two products influenced by three modifiers.

Name reaction-52

Reaction equation

Table 122: Properties of each reactant.

	• 122. 11 openies of each feature.	
Id	Name	SBO
species_33	(IL6-gp80-gp130-JAK)2*-SHP2*	

Table 123: Properties of each modifier.

Id	Name	SBO
species_9	(IL6-gp80-gp130-JAK)2*-SHP2* (IL6-gp80-gp130-JAK)2* SHP2*	

Products

Table 124: Properties of each product.

Id	Name	SBO
species_9 species_48	(IL6-gp80-gp130-JAK)2* SHP2*	

Kinetic Law

Derived unit contains undeclared units

$$v_{41} = \text{vol} (\text{compartment_1})$$

 $\cdot (\text{parameter_87} \cdot [\text{species_33}] - \text{parameter_88} \cdot [\text{species_9}] \cdot [\text{species_48}])$ (101)

8.42 Reaction reaction_43

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-53

Reaction equation

Table 125: Properties of each reactant.

Id	Name	SBO
species_32	Grb2	
species_48	SHP2*	

Table 126: Properties of each modifier.

Id	Name	SBO
species_32		
species_48	SHP2*	
species_36	SHP2*-Grb2	

Product

Table 127: Properties of each product.

Id	Name	SBO
species_36	SHP2*-Grb2	

Kinetic Law

Derived unit contains undeclared units

$$v_{42} = \text{vol} (\text{compartment_1})$$

 $\cdot (\text{parameter_89} \cdot [\text{species_32}] \cdot [\text{species_48}] - \text{parameter_90} \cdot [\text{species_36}])$ (103)

8.43 Reaction reaction_44

This is a reversible reaction of one reactant forming two products influenced by three modifiers.

Name reaction-54

Reaction equation

$$species_34 \xrightarrow{species_34, species_9, species_36} species_9 + species_36$$
 (104)

Table 128: Properties of each reactant.

Id	Name	SBO
species_34	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2	

Table 129: Properties of each modifier.

Id	Name	SBO
species_9	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2 (IL6-gp80-gp130-JAK)2* SHP2*-Grb2	

Products

Table 130: Properties of each product.

Id	Name	SBO
species_9 species_36	(IL6-gp80-gp130-JAK)2* SHP2*-Grb2	

Kinetic Law

Derived unit contains undeclared units

$$v_{43} = \text{vol}(\text{compartment_1})$$

$$\cdot (\text{parameter_91} \cdot [\text{species_34}] - \text{parameter_92} \cdot [\text{species_9}] \cdot [\text{species_36}])$$
(105)

8.44 Reaction reaction_45

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-55

Reaction equation

$$species_35 + species_36 \xrightarrow{species_35, species_36, species_46} species_46$$
 (106)

Table 131: Properties of each reactant.

Id	Name	SBO
species_35 species_36	SOS SHP2*-Grb2	

Table 132: Properties of each modifier.

Id	Name	SBO
-	SOS SHP2*-Grb2 SHP2*-Grb2-SOS	

Product

Table 133: Properties of each product.

Id	Name	SBO
species_46	SHP2*-Grb2-SOS	

Kinetic Law

Derived unit contains undeclared units

$$v_{44} = \text{vol} (\text{compartment_1})$$

$$\cdot (\text{parameter_93} \cdot [\text{species_35}] \cdot [\text{species_36}] - \text{parameter_94} \cdot [\text{species_46}])$$
(107)

8.45 Reaction reaction_46

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-56

Reaction equation

$$species_33 + species_47 \xrightarrow{species_33, species_47, species_37} species_37$$
 (108)

Table 134: Properties of each reactant.

Id	Name	SBO
species_33 species_47	(IL6-gp80-gp130-JAK)2*-SHP2* Grb2-SOS	

Table 135: Properties of each modifier.

Id	Name	SBO
species_33 species_47	(IL6-gp80-gp130-JAK)2*-SHP2* Grb2-SOS	
-	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS	

Product

Table 136: Properties of each product.

Id	Name	SBO
species_37	(IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS	

Kinetic Law

Derived unit contains undeclared units

$$v_{45} = \text{vol} (\text{compartment_1})$$

 $\cdot (\text{parameter_95} \cdot [\text{species_33}] \cdot [\text{species_47}] - \text{parameter_96} \cdot [\text{species_37}])$ (109)

8.46 Reaction reaction_47

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-57

Reaction equation

Table 137: Properties of each reactant.

Id	Name	SBO
species_49 species_44	Phosp1 Raf*	

Table 138: Properties of each modifier.

Id	Name	SBO
species_49	Phosp1	
species_44	Raf*	
${\tt species_50}$	Raf*-Phosp1	

Product

Table 139: Properties of each product.

Id	Name	SBO
species_50	Raf*-Phosp1	

Kinetic Law

Derived unit contains undeclared units

$$v_{46} = \text{vol}(\text{compartment_1})$$

 $\cdot (\text{parameter_97} \cdot [\text{species_49}] \cdot [\text{species_44}] - \text{parameter_98} \cdot [\text{species_50}])$ (111)

8.47 Reaction reaction_48

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-58

Reaction equation

$$species_50 \xrightarrow{species_50} species_41 + species_49$$
 (112)

Table 140: Properties of each reactant.

Id	Name	SBO
species_50	Raf*-Phosp1	

Table 141: Properties of each modifier.

Id	Name	SBO
species_50	Raf*-Phosp1	

Products

Table 142: Properties of each product.

Id	Name	SBO
species_41	Raf	
species_49	Phosp1	

Kinetic Law

Derived unit contains undeclared units

$$v_{47} = \text{vol}(\text{compartment}_1) \cdot \text{parameter}_99 \cdot [\text{species}_50]$$
 (113)

8.48 Reaction reaction_49

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-59

Reaction equation

$$species_44 + species_51 \xrightarrow{species_44, species_51, species_52} species_52$$
 (114)

Table 143: Properties of each reactant.

Id	Name	SBO
species_44 species_51	Raf* MEK	

Table 144: Properties of each modifier.

Id	Name	SBO
species_44 species_51 species_52	MEK	

Product

Table 145: Properties of each product.

Id	Name	SBO
species_52	MEK-Raf*	

Kinetic Law

Derived unit contains undeclared units

$$v_{48} = \text{vol}(\text{compartment_1}) \\ \cdot (\text{parameter_100} \cdot [\text{species_44}] \cdot [\text{species_51}] - \text{parameter_101} \cdot [\text{species_52}])$$
 (115)

8.49 Reaction reaction_5

This is a reversible reaction of one reactant forming one product influenced by two modifiers.

Name reaction-6

Reaction equation

$$2 \text{ species}_{-7} \xrightarrow{\text{species}_{-8}} \text{ species}_{-8}$$
 (116)

Table 146: Properties of each reactant.

Id	Name	SBO
species_7	IL6-gp80-gp130-JAK	

Table 147: Properties of each modifier.

	1	
Id	Name	SBO
-	IL6-gp80-gp130-JAK (IL6-gp80-gp130-JAK)2	

Product

Table 148: Properties of each product.

	1 1	
Id	Name	SBO
species_8	(IL6-gp80-gp130-JAK)2	

Kinetic Law

Derived unit contains undeclared units

$$v_{49} = \text{vol}(\text{compartment_1}) \cdot (\text{parameter_11} \cdot [\text{species_7}]^2 - \text{parameter_12} \cdot [\text{species_8}])$$
 (117)

8.50 Reaction reaction_50

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-60

Reaction equation

$$species_52 \xrightarrow{species_52} species_53 + species_44$$
 (118)

Table 149: Properties of each reactant.

Id	Name	SBO
species_52	MEK-Raf*	

Table 150: Properties of each modifier.

Id	Name	SBO
species_52	MEK-Raf*	

Products

Table 151: Properties of each product.

Id	Name	SBO
species_53	MEK-P	
${\tt species_44}$	Raf*	

Kinetic Law

Derived unit contains undeclared units

$$v_{50} = \text{vol} (\text{compartment}_{-1}) \cdot \text{parameter}_{-1}02 \cdot [\text{species}_{-5}2]$$
 (119)

8.51 Reaction reaction_51

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-61

Reaction equation

Table 152: Properties of each reactant.

Id	Name	SBO
species_44 species_53	Raf* MEK-P	

Table 153: Properties of each modifier.

Id	Name	SBO
species_44 species_53 species_54		

Product

Table 154: Properties of each product.

Id	Name	SBO
species_54	MEK-P-Raf*	

Kinetic Law

Derived unit contains undeclared units

$$v_{51} = \text{vol}(\text{compartment}_1)$$

 $\cdot (\text{parameter}_103 \cdot [\text{species}_44] \cdot [\text{species}_53] - \text{parameter}_104 \cdot [\text{species}_54])$ (121)

8.52 Reaction reaction_52

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-62

Reaction equation

$$species_54 \xrightarrow{species_54} species_55 + species_44$$
 (122)

Table 155: Properties of each reactant.

THE TO THE TENT TO PET THE ST CHEST TO CHEST		
Id	Name	SBO
species_54	MEK-P-Raf*	

Table 156: Properties of each modifier.

Id	Name	SBO
species_54	MEK-P-Raf*	

Products

Table 157: Properties of each product.

Id	Name	SBO
species_55	MEK-PP	
species_44	Raf*	

Kinetic Law

Derived unit contains undeclared units

$$v_{52} = \text{vol} (\text{compartment}_{-1}) \cdot \text{parameter}_{-1}05 \cdot [\text{species}_{-5}4]$$
 (123)

8.53 Reaction reaction_53

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-63

Reaction equation

$$species_55 + species_57 \xrightarrow{species_55, species_57, species_56} species_56$$
 (124)

Table 158: Properties of each reactant.

Id	Name	SBO
species_55 species_57		

Table 159: Properties of each modifier.

Id	Name	SBO
species_55 species_57 species_56		

Product

Table 160: Properties of each product.

Id	Name	SBO
species_56	MEK-PP-Phosp2	

Kinetic Law

Derived unit contains undeclared units

$$v_{53} = \text{vol}(\text{compartment}_1)$$

 $\cdot (\text{parameter}_106 \cdot [\text{species}_55] \cdot [\text{species}_57] - \text{parameter}_107 \cdot [\text{species}_56])$ (125)

8.54 Reaction reaction_54

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-64

Reaction equation

$$species_56 \xrightarrow{species_56} species_57 + species_53$$
 (126)

Table 161: Properties of each reactant.

Id	Name	SBO
species_56	MEK-PP-Phosp2	

Table 162: Properties of each modifier.

Id	Name	SBO
species_56	MEK-PP-Phosp2	

Products

Table 163: Properties of each product.

Id	Name	SBO
species_57	Phosp2	
species_53	MEK-P	

Kinetic Law

Derived unit contains undeclared units

$$v_{54} = \text{vol} (\text{compartment}_1) \cdot \text{parameter}_108 \cdot [\text{species}_56]$$
 (127)

8.55 Reaction reaction_55

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-65

Reaction equation

$$species_53 + species_57 \xrightarrow{species_53, species_57, species_58} species_58$$
 (128)

Table 164: Properties of each reactant.

Id	Name	SBO
species_53 species_57	MEK-P Phosp2	

Table 165: Properties of each modifier.

Id	Name	SBO
species_53		
species_57 species_58	MEK-P-Phosp2	

Product

Table 166: Properties of each product.

Id	Name	SBO
species_58	MEK-P-Phosp2	

Kinetic Law

Derived unit contains undeclared units

$$v_{55} = \text{vol}(\text{compartment_1})$$

 $\cdot (\text{parameter_109} \cdot [\text{species_53}] \cdot [\text{species_57}] - \text{parameter_110} \cdot [\text{species_58}])$ (129)

8.56 Reaction reaction_56

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-66

Reaction equation

species_58
$$\xrightarrow{\text{species}_58}$$
 species_57 + species_51 (130)

Table 167: Properties of each reactant.

THOIR TOTAL TOPPOTTION OF PHOTO TOPPOTTION		
Id	Name	SBO
species_58	MEK-P-Phosp2	

Table 168: Properties of each modifier.

Tueste Teer Troperties et euen meunien		
Id	Name	SBO
species_58	MEK-P-Phosp2	

Products

Table 169: Properties of each product.

Id	Name	SBO
species_57	Phosp2	
species_51	MEK	

Kinetic Law

Derived unit contains undeclared units

$$v_{56} = \text{vol}(\text{compartment_1}) \cdot \text{parameter_111} \cdot [\text{species_58}]$$
 (131)

8.57 Reaction reaction_57

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-67

Reaction equation

$$species_55 + species_59 \xrightarrow{species_55, species_59, species_60} species_60$$
 (132)

Table 170: Properties of each reactant.

Id	Name	SBO
species_55 species_59		

Table 171: Properties of each modifier.

Id	Name	SBO
species_55	MEK-PP	
species_59	ERK	
${\tt species_60}$	ERK-MEK-PP	

Product

Table 172: Properties of each product.

Id	Name	SBO
species_60	ERK-MEK-PP	

Kinetic Law

Derived unit contains undeclared units

$$v_{57} = \text{vol}(\text{compartment_1})$$

 $\cdot (\text{parameter_112} \cdot [\text{species_55}] \cdot [\text{species_59}] - \text{parameter_113} \cdot [\text{species_60}])$ (133)

8.58 Reaction reaction_58

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-68

Reaction equation

$$species_60 \xrightarrow{species_60} species_61 + species_55$$
 (134)

Table 173: Properties of each reactant.

Table 173. Hoperties of each reactant.		
Id	Name	SBO
species_60	ERK-MEK-PP	

Table 174: Properties of each modifier.

	F	
Id	Name	SBO
species_60	ERK-MEK-PP	

Products

Table 175: Properties of each product.

Id	Name	SBO
species_61	ERK-P	
species_55	MEK-PP	

Kinetic Law

Derived unit contains undeclared units

$$v_{58} = \text{vol} (\text{compartment}_1) \cdot \text{parameter}_114 \cdot [\text{species}_60]$$
 (135)

8.59 Reaction reaction_59

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-69

Reaction equation

$$species_55 + species_61 \xrightarrow{species_65, species_61, species_62} species_62$$
 (136)

Table 176: Properties of each reactant.

Id	Name	SBO
species_55 species_61		

Table 177: Properties of each modifier.

Id	Name	SBO
species_55	MEK-PP	
species_61	ERK-P	
species_62	ERK-P-MEK-PP	

Product

Table 178: Properties of each product.

Id	Name	SBO
species_62	ERK-P-MEK-PP	

Kinetic Law

Derived unit contains undeclared units

$$v_{59} = \text{vol}(\text{compartment_1})$$

 $\cdot (\text{parameter_115} \cdot [\text{species_55}] \cdot [\text{species_61}] - \text{parameter_116} \cdot [\text{species_62}])$ (137)

8.60 Reaction reaction_6

This is an irreversible reaction of one reactant forming one product influenced by one modifier.

Name reaction-7

Reaction equation

$$species_8 \xrightarrow{species_8} species_9$$
 (138)

Table 179: Properties of each reactant.

Id	Name	SBO
species_8	(IL6-gp80-gp130-JAK)2	

Table 180: Properties of each modifier.

Id	Name	SBO
species_8	(IL6-gp80-gp130-JAK)2	

Product

Table 181: Properties of each product.

Id	Name	SBO
species_9	(IL6-gp80-gp130-JAK)2*	

Kinetic Law

Derived unit contains undeclared units

$$v_{60} = \text{vol}(\text{compartment_1}) \cdot \text{parameter_13} \cdot [\text{species_8}]$$
 (139)

8.61 Reaction reaction_60

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-70

Reaction equation

species_62
$$\xrightarrow{\text{species}_62}$$
 species_63 + species_55 (140)

Table 182: Properties of each reactant.

Id	Name	SBO
species_62	ERK-P-MEK-PP	

Table 183: Properties of each modifier.

_ racio recorrieperates or calculation.		
Id	Name	SBO
species_62	ERK-P-MEK-PP	

Products

Table 184: Properties of each product.

Id	Name	SBO
species_63 species_55		

Kinetic Law

Derived unit contains undeclared units

$$v_{61} = \text{vol} (\text{compartment}_{-1}) \cdot \text{parameter}_{-117} \cdot [\text{species}_{-62}]$$
 (141)

8.62 Reaction reaction_64

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-71

Reaction equation

Reactants

Table 185: Properties of each reactant.

Id	Name	SBO
Special se	ERK-PP	
species_64	rnosps	

Table 186: Properties of each modifier.

Id	Name	SBO
species_63 species_64 species_65		

Table 187: Properties of each product.

Id	Name	SBO
species_65	ERK-PP-Phosp3	

Kinetic Law

Derived unit contains undeclared units

$$v_{62} = \text{vol}(\text{compartment_1})$$

 $\cdot (\text{parameter_118} \cdot [\text{species_63}] \cdot [\text{species_64}] - \text{parameter_119} \cdot [\text{species_65}])$ (143)

8.63 Reaction reaction_61

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-72

Reaction equation

$$species_65 \xrightarrow{species_65} species_64 + species_61$$
 (144)

Reactant

Table 188: Properties of each reactant.

Id	Name	SBO
species_65	ERK-PP-Phosp3	

Table 189: Properties of each modifier.

Id	Name	SBO
species_65	ERK-PP-Phosp3	

Table 190: Properties of each product.

Id	Name	SBO
species_64		
species_61	ERK-P	

Kinetic Law

Derived unit contains undeclared units

$$v_{63} = \text{vol} (\text{compartment_1}) \cdot \text{parameter_120} \cdot [\text{species_65}]$$
 (145)

8.64 Reaction reaction_62

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-73

Reaction equation

Reactants

Table 191: Properties of each reactant.

Id	Name	SBO
species_61	ERK-P	
species_64	Phosp3	

Table 192: Properties of each modifier.

Id	Name	SBO
species_61 species_64		
-	ERK-P-Phosp3	

Table 193: Properties of each product.

Id	Name	SBO
species_66	ERK-P-Phosp3	

Kinetic Law

Derived unit contains undeclared units

$$v_{64} = \text{vol}(\text{compartment_1})$$

$$\cdot (\text{parameter_121} \cdot [\text{species_61}] \cdot [\text{species_64}] - \text{parameter_122} \cdot [\text{species_66}])$$
(147)

8.65 Reaction reaction_63

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-74

Reaction equation

$$species_66 \xrightarrow{species_66} species_64 + species_59$$
 (148)

Reactant

Table 194: Properties of each reactant.

Id	Name	SBO
species_66	ERK-P-Phosp3	

Table 195: Properties of each modifier.

rue to 1981 troperties of euen mounter.		
Id	Name	SBO
species_66	ERK-P-Phosp3	

Table 196: Properties of each product.

Id	Name	SBO
species_64	•	
species_59	ERK	

Kinetic Law

Derived unit contains undeclared units

$$v_{65} = \text{vol} (\text{compartment_1}) \cdot \text{parameter_123} \cdot [\text{species_66}]$$
 (149)

8.66 Reaction reaction_74

This is an irreversible reaction of one reactant forming one product influenced by three modifiers.

Name reaction-75

Reaction equation

species_74
$$\xrightarrow{\text{species}_63, \text{species}_74}$$
 species_75 $\xrightarrow{\text{species}_75}$ (150)

Reactant

Table 197: Properties of each reactant.

Id	Name	SBO
species_74	CEBPi	

Table 198: Properties of each modifier.

Id	Name	SBO
species_63	ERK-PP	
species_63	ERK-PP	
${\tt species_74}$	CEBPi	

Table 199: Properties of each product.

Id	Name	SBO
species_75	CEBP	

Kinetic Law

Derived unit contains undeclared units

$$v_{66} = \text{vol}(\text{compartment_1})$$
· function_activation(parameter_124,[species_63],[species_74], parameter_125)
(151)

function_activation
$$(v, a, b, k) = \frac{v \cdot a \cdot b}{b + k}$$
 (152)

function_activation
$$(v, a, b, k) = \frac{v \cdot a \cdot b}{b + k}$$
 (153)

8.67 Reaction reaction_75

This is an irreversible reaction of one reactant forming one product influenced by one modifier.

Name reaction-76

Reaction equation

species_75
$$\xrightarrow{\text{species}_75}$$
 species_74 (154)

Table 200: Properties of each reactant.

Id	Name	SBO
species_75	CEBP	

Table 201: Properties of each modifier.

Id	Name	SBO
species_75	CEBP	

Product

Table 202: Properties of each product.

Id	Name	SBO
species_74	CEBPi	

Kinetic Law

Derived unit contains undeclared units

$$v_{67} = \text{vol}(\text{compartment_1}) \cdot \text{parameter_126} \cdot [\text{species_75}]$$
 (155)

8.68 Reaction reaction_76

This is a reversible reaction of one reactant forming one product influenced by two modifiers.

Name reaction-77

Reaction equation

$$2 \text{ species_75}, \frac{\text{species_76}}{\text{species_76}} \text{ species_76}$$
 (156)

Table 203: Properties of each reactant.

Id	Name	SBO
species_75	CEBP	

Table 204: Properties of each modifier.

Id	Name	SBO
species_75		
species_76	CEBPn	

Product

Table 205: Properties of each product.

Id	Name	SBO
species_76	CEBPn	

Kinetic Law

Derived unit contains undeclared units

$$v_{68} = \text{vol} \left(\text{compartment_1} \right) \cdot \left(\text{parameter_127} \cdot [\text{species_75}]^2 - \text{parameter_128} \cdot [\text{species_76}] \right)$$
(157)

8.69 Reaction reaction_4

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-8

Reaction equation

$$species_9 + species_11 \xrightarrow{species_9, species_11, species_10} species_10$$
 (158)

Reactants

Table 206: Properties of each reactant.

Id	Name	SBO
species_9 species_11	(IL6-gp80-gp130-JAK)2* STAT3C	

Table 207: Properties of each modifier.

Id	Name	SBO
species_9 species_11 species_10	(IL6-gp80-gp130-JAK)2* STAT3C (IL6-gp80-gp130-JAK)2*-STAT3C	

Table 208: Properties of each product.

Id	Name	SBO
species_10	(IL6-gp80-gp130-JAK)2*-STAT3C	

Kinetic Law

Derived unit contains undeclared units

$$v_{69} = \text{vol} (\text{compartment_1})$$

 $\cdot (\text{parameter_14} \cdot [\text{species_9}] \cdot [\text{species_11}] - \text{parameter_15} \cdot [\text{species_10}])$ (159)

8.70 Reaction reaction_7

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-9

Reaction equation

$$species_{-}10 \xrightarrow{species_{-}10} species_{-}9 + species_{-}12$$
 (160)

Reactant

Table 209: Properties of each reactant.

Id	Name	SBO
species_10	(IL6-gp80-gp130-JAK)2*-STAT3C	

Table 210: Properties of each modifier.

Id	Name	SBO
species_10	(IL6-gp80-gp130-JAK)2*-STAT3C	

Table 211: Properties of each product.

	1 1	
Id	Name	SBO
species_9 species_12	(IL6-gp80-gp130-JAK)2* STAT3C*	

Kinetic Law

Derived unit contains undeclared units

$$v_{70} = \text{vol}(\text{compartment_1}) \cdot \text{parameter_16} \cdot [\text{species_10}]$$
 (161)

8.71 Reaction re135

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-v1

Reaction equation

$$species_5 + species_107 \xrightarrow{species_5}, species_107, species_78$$
 species_78 (162)

Reactants

Table 212: Properties of each reactant.

Id	Name	SBO
species_5	JAK(IFN)	
species_107	R	

Table 213: Properties of each modifier.

Id	Name	SBO
species_5 species_107 species_78	JAK(IFN) R R-JAK	

Table 214: Properties of each product.

Id	Name	SBO
species_78	R-JAK	

Kinetic Law

Derived unit contains undeclared units

$$v_{71} = \text{vol}(\text{compartment}_1)$$

 $\cdot (\text{parameter}_129 \cdot [\text{species}_5] \cdot [\text{species}_107] - \text{parameter}_130 \cdot [\text{species}_78])$ (163)

8.72 Reaction reaction_87

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-v10

Reaction equation

$$species_88 \xrightarrow{species_88} species_81 + species_108$$
 (164)

Reactant

Table 215: Properties of each reactant.

Id	Name	SBO
species_88	(IFN-R-JAK)2*-SHP2	

Table 216: Properties of each modifier.

Id	Name	SBO
species_88	(IFN-R-JAK)2*-SHP2	

Table 217: Properties of each product.

Id	Name	SBO
species_81 species_108	(IFN-R-JAK)2 SHP2	

Kinetic Law

Derived unit contains undeclared units

$$v_{72} = \text{vol} (\text{compartment_1}) \cdot \text{parameter_145} \cdot [\text{species_88}]$$
 (165)

8.73 Reaction re140

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-v11

Reaction equation

$$species_85 + species_20 \xrightarrow{species_85, species_20, species_90} species_90$$
 (166)

Reactants

Table 218: Properties of each reactant.

Id	Name	SBO
species_85	STAT1C*	
species_20	PP1	

Table 219: Properties of each modifier.

	F	
Id	Name	SBO
species_85 species_20 species_90	STAT1C* PP1 PP1-STAT1C*	

Table 220: Properties of each product.

Id	Name	SBO
species_90	PP1-STAT1C*	

Kinetic Law

Derived unit contains undeclared units

$$v_{73} = \text{vol}(\text{compartment_1})$$

 $\cdot (\text{parameter_146} \cdot [\text{species_85}] \cdot [\text{species_20}] - \text{parameter_147} \cdot [\text{species_90}])$ (167)

8.74 Reaction re141

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-v12

Reaction equation

$$species_90 \xrightarrow{species_90} species_84 + species_20$$
 (168)

Reactant

Table 221: Properties of each reactant.

	F	
Id	Name	SBO
species_90	PP1-STAT1C*	

Table 222: Properties of each modifier.

Tuble 222: I roperties of each inodifier.		
Id	Name	SBO
species_90	PP1-STAT1C*	

Table 223: Properties of each product.

Id	Name	SBO
species_84 species_20	STAT1C PP1	

Kinetic Law

Derived unit contains undeclared units

$$v_{74} = \text{vol}(\text{compartment}_1) \cdot \text{parameter}_148 \cdot [\text{species}_90]$$
 (169)

8.75 Reaction reaction_90

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-v13

Reaction equation

$$species_84 + species_85 \xrightarrow{species_84, species_85, species_91} species_91$$
 (170)

Reactants

Table 224: Properties of each reactant.

Id	Name	SBO
species_84	STAT1C	
species_85	STAT1C*	

Table 225: Properties of each modifier.

14616 228.1	racio 220. Properties of each incumer.		
Id	Name	SBO	
species_84 species_85 species_91			

Table 226: Properties of each product.

Id	Name	SBO
species_91	STAT1C-STAT1C*	

Kinetic Law

Derived unit contains undeclared units

$$v_{75} = \text{vol}(\text{compartment_1})$$

 $\cdot (\text{parameter_149} \cdot [\text{species_84}] \cdot [\text{species_85}] - \text{parameter_150} \cdot [\text{species_91}])$ (171)

8.76 Reaction reaction_91

This is an irreversible reaction of one reactant forming one product influenced by one modifier.

Name reaction-v14

Reaction equation

$$species_87 \xrightarrow{species_87} species_92$$
 (172)

Reactant

Table 227: Properties of each reactant.

Id	Name	SBO
species_87	(STAT1C*)2	

Table 228: Properties of each modifier.

Id	Name	SBO
species_87	(STAT1C*)2	

Table 229: Properties of each product.

Id	Name	SBO
species_92	(STAT1N*)2	

Kinetic Law

Derived unit contains undeclared units

$$v_{76} = \text{parameter} \cdot 151 \cdot [\text{species} \cdot 87]$$
 (173)

8.77 Reaction re144

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-v15

Reaction equation

Reactants

Table 230: Properties of each reactant.

Id	Name	SBO
species_95	STAT1N*	
species_24	PP2	

Table 231: Properties of each modifier.

Id	Name	SBO
species_24 species_94 species_95	PP2-STAT1N*	

Table 232: Properties of each product.

Id	Name	SBO
species_94	PP2-STAT1N*	

Kinetic Law

Derived unit contains undeclared units

$$v_{77} = \text{function_4} (\text{vol} (\text{c2}), \text{parameter_152}, \text{parameter_153}, [\text{species_24}], [\text{species_94}], [\text{species_95}])$$
 (175)

8.78 Reaction re145

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-v16

Reaction equation

$$species_94 \xrightarrow{species_94} species_96 + species_24$$
 (177)

Table 233: Properties of each reactant.

Tuble 255. Troperties of each reactant.		
Id	Name	SBO
species_94	PP2-STAT1N*	

Table 234: Properties of each modifier.

Tuble 254. I Toperties of each inodifier.		
Id	Name	SBO
species_94	PP2-STAT1N*	

Products

Table 235: Properties of each product.

Id	Name	SBO
species_96 species_24	STAT1N PP2	

Kinetic Law

Derived unit contains undeclared units

$$v_{78} = \text{function_5}(\text{vol}(\text{c2}), \text{parameter_155}, [\text{species_94}])$$
 (178)

$$function_5\left(vol\left(c2\right),parameter_155,[species_94]\right) = vol\left(c2\right) \cdot parameter_155 \cdot [species_94] \tag{179}$$

8.79 Reaction reaction_94

This is an irreversible reaction of one reactant forming one product influenced by one modifier.

Name reaction-v17

Reaction equation

$$species_{96} \xrightarrow{species_{96}} species_{84}$$
 (180)

Table 236: Properties of each reactant.

Id	Name	SBO
species_96	STAT1N	

Table 237: Properties of each modifier.

Id	Name	SBO
species_96	STAT1N	

Product

Table 238: Properties of each product.

Id	Name	SBO
species_84	STAT1C	

Kinetic Law

Derived unit contains undeclared units

$$v_{79} = \text{parameter}_154 \cdot [\text{species}_96]$$
 (181)

8.80 Reaction reaction_95

This is an irreversible reaction of no reactant forming one product influenced by two modifiers.

Name reaction-v18

Reaction equation

$$\emptyset \xrightarrow{\text{species_92, species_92}} \text{species_97}$$
 (182)

Modifiers

Table 239: Properties of each modifier.

Id	Name	SBO
-	(STAT1N*)2 (STAT1N*)2	

Product

Table 240: Properties of each product.

Id	Name	SBO
species_97	mRNA-SOCS1N	

Kinetic Law

Derived unit contains undeclared units

$$v_{80} = \text{vol}(c2) \cdot \text{function}_6(\text{parameter}_156, \text{parameter}_157, [\text{species}_92])$$
 (183)

$$function_6 (parameter_156, parameter_157, [species_92]) = \frac{parameter_156 \cdot [species_92]}{parameter_157 + [species_92]}$$

$$(184)$$

$$function_6 (parameter_156, parameter_157, [species_92]) = \frac{parameter_156 \cdot [species_92]}{parameter_157 + [species_92]}$$

$$(185)$$

8.81 Reaction reaction_96

This is an irreversible reaction of one reactant forming one product influenced by one modifier.

Name reaction-v19

Reaction equation

$$species_97 \xrightarrow{species_97} species_98$$
 (186)

Reactant

Table 241: Properties of each reactant.

Id	Name	SBO
species_97	mRNA-SOCS1N	

Table 242: Properties of each modifier.

Id	Name	SBO
species_97	mRNA-SOCS1N	

Table 243: Properties of each product.

Id	Name	SBO
species_98	mRNA-SOCS1C	

Kinetic Law

Derived unit contains undeclared units

$$v_{81} = \text{parameter} \cdot 158 \cdot [\text{species} \cdot 97]$$
 (187)

8.82 Reaction reaction_80

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-v2

Reaction equation

$$species_79 + species_78 \xrightarrow{species_79, species_78, species_80} species_80$$
 (188)

Reactants

Table 244: Properties of each reactant.

Id	Name	SBO
species_79	IFN	
species_78	R-JAK	

Table 245: Properties of each modifier.

Id	Name	SBO
species_79	IFN	
species_78	R-JAK	
species_80	IFN-R-JAK	

Table 246: Properties of each product.

Id	Name	SBO
species_80	IFN-R-JAK	

Kinetic Law

Derived unit contains undeclared units

$$v_{82} = \text{parameter}_131 \cdot [\text{species}_79] \cdot [\text{species}_78] - \text{parameter}_132 \cdot [\text{species}_80]$$
 (189)

8.83 Reaction reaction_97

This is an irreversible reaction of no reactant forming one product influenced by two modifiers.

Name reaction-v20

Reaction equation

$$\emptyset \xrightarrow{\text{species_98, species_98}} \text{species_99}$$
 (190)

Modifiers

Table 247: Properties of each modifier.

Id	Name	SBO
-	mRNA-SOCS1C mRNA-SOCS1C	

Product

Table 248: Properties of each product.

Id	Name	SBO
species_99	SOCS1	

Kinetic Law

Derived unit contains undeclared units

$$v_{83} = \text{vol}(\text{compartment_1}) \cdot \text{function_7}(\text{parameter_159}, [\text{species_98}])$$
 (191)

function_7 (parameter_159, [species_98]) = parameter_159
$$\cdot$$
 [species_98] (192)

function_7 (parameter_159, [species_98]) = parameter_159
$$\cdot$$
 [species_98] (193)

8.84 Reaction reaction_98

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-v21

Reaction equation

Reactants

Table 249: Properties of each reactant.

~- ~
X)2*

Table 250: Properties of each modifier.

Id	Name	SBO
species_99 species_82 species_100	SOCS1 (IFN-R-JAK)2* (IFN-R-JAK)2*-SOCS1	

Table 251: Properties of each product.

Id	Name	SBO
species_100	(IFN-R-JAK)2*-SOCS1	

Kinetic Law

Derived unit contains undeclared units

$$v_{84} = \text{vol}(\text{compartment_1}) \\ \cdot (\text{parameter_160} \cdot [\text{species_99}] \cdot [\text{species_82}] - \text{parameter_161} \cdot [\text{species_100}])$$
(195)

8.85 Reaction reaction_99

This is an irreversible reaction of one reactant forming no product influenced by one modifier.

Name reaction-v22

Reaction equation

$$species_98 \xrightarrow{species_98} \emptyset$$
 (196)

Reactant

Table 252: Properties of each reactant.

Id	Name	SBO
species_98	mRNA-SOCS1C	

Modifier

Table 253: Properties of each modifier.

Id	Name	SBO
species_98	mRNA-SOCS1C	

Kinetic Law

Derived unit contains undeclared units

$$v_{85} = \text{vol}(\text{compartment_1}) \cdot \text{parameter_162} \cdot [\text{species_98}]$$
 (197)

8.86 Reaction reaction_100

This is an irreversible reaction of one reactant forming no product influenced by one modifier.

Name reaction-v23

Reaction equation

$$species_{-99} \xrightarrow{species_{-99}} \emptyset$$
 (198)

Reactant

Table 254: Properties of each reactant.

Id	Name	SBO
species_99	SOCS1	

Modifier

Table 255: Properties of each modifier.

Id	Name	SBO
species_99	SOCS1	

Kinetic Law

Derived unit contains undeclared units

$$v_{86} = \text{vol}(\text{compartment_1}) \cdot \text{parameter_163} \cdot [\text{species_99}]$$
 (199)

8.87 Reaction re142

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-v24

Reaction equation

$$species_87 + species_20 \xrightarrow{species_87, species_20, species_101} species_101 \qquad (200)$$

Table 256: Properties of each reactant.

Id	Name	SBO
species_87 species_20	(STAT1C*)2 PP1	

Table 257: Properties of each modifier.

Id	Name	SBO
species_87	(STAT1C*)2	
species_20	PP1	
species_101	PP1-(STAT1C*)2	

Product

Table 258: Properties of each product.

Id	Name	SBO
species_101	PP1-(STAT1C*)2	

Kinetic Law

Derived unit contains undeclared units

$$v_{87} = \text{vol}(\text{compartment_1}) \\ \cdot (\text{parameter_164} \cdot [\text{species_87}] \cdot [\text{species_20}] - \text{parameter_165} \cdot [\text{species_101}])$$
 (201)

8.88 Reaction re143

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-v25

Reaction equation

$$species_101 \xrightarrow{species_101} species_91 + species_20$$
 (202)

Table 259: Properties of each reactant.

Id	Name	SBO
species_101	PP1-(STAT1C*)2	

Table 260: Properties of each modifier.

Id	Name	SBO
species_101	PP1-(STAT1C*)2	

Products

Table 261: Properties of each product.

Id	Name	SBO
species_91	STAT1C-STAT1C*	
species_20	PP1	

Kinetic Law

Derived unit contains undeclared units

$$v_{88} = \text{vol} (\text{compartment}_{-1}) \cdot \text{parameter}_{-1} 66 \cdot [\text{species}_{-1} 01]$$
 (203)

8.89 Reaction reaction_103

This is a reversible reaction of one reactant forming one product influenced by two modifiers.

Name reaction-v26

Reaction equation

$$2 \text{ species_95}, \text{ species_92}$$
 species_92 (204)

Table 262: Properties of each reactant.

Id	Name	SBO
species_95	STAT1N*	

Table 263: Properties of each modifier.

Id	Name	SBO
species_95	STAT1N*	
species_92	(STAT1N*)2	

Product

Table 264: Properties of each product.

Id	Name	SBO
species_92	(STAT1N*)2	

Kinetic Law

Derived unit contains undeclared units

$$v_{89} = \text{vol}(\text{c2}) \cdot \left(\text{parameter_167} \cdot [\text{species_95}]^2 - \text{parameter_168} \cdot [\text{species_92}] \right) \quad (205)$$

8.90 Reaction re147

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-v27

Reaction equation

Table 265: Properties of each reactant.

Id	Name	SBO
species_92 species_24	(STAT1N*)2 PP2	

Table 266: Properties of each modifier.

Id	Name	SBO
species_102 species_24 species_92	PP2-(STAT1N*)2 PP2 (STAT1N*)2	

Product

Table 267: Properties of each product.

Id	Name	SBO
species_102	PP2-(STAT1N*)2	

Kinetic Law

Derived unit contains undeclared units

 $v_{90} = \text{function_8} (\text{vol}(c2), \text{parameter_169}, \text{parameter_170}, [\text{species_102}], [\text{species_24}], [\text{species_24}])$

$$\begin{aligned} &\text{function_8} \left(\text{vol} \left(\text{c2} \right), \text{parameter_169}, \text{parameter_170}, [\text{species_102}], [\text{species_24}], [\text{species_92}] \right) \\ &= &\text{vol} \left(\text{c2} \right) \cdot \left(\text{parameter_169} \cdot [\text{species_24}] \cdot [\text{species_92}] - \text{parameter_170} \cdot [\text{species_102}] \right) \end{aligned}$$

8.91 Reaction re148

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-v28

Reaction equation

$$species_102 \xrightarrow{species_102} species_103 + species_24$$
 (209)

Table 268: Properties of each reactant.

Id	Name	SBO
species_102	PP2-(STAT1N*)2	

Table 269: Properties of each modifier.

Id	Name	SBO
species_102	PP2-(STAT1N*)2	

Products

Table 270: Properties of each product.

Id	Name	SBO
species_103	STAT1N-STAT1N*	
${\tt species_24}$	PP2	

Kinetic Law

Derived unit contains undeclared units

$$v_{91} = \text{function}_{9}(\text{vol}(\text{c2}), \text{parameter}_{171}, [\text{species}_{102}])$$
 (210)

$$function_9 \left(vol\left(c2\right), parameter_171, [species_102] \right) = vol\left(c2\right) \cdot parameter_171 \cdot [species_102] \tag{211}$$

8.92 Reaction reaction_106

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-v29

Reaction equation

$$species_96 + species_95 \xrightarrow{species_96, species_95, species_103} species_103$$
 (212)

Table 271: Properties of each reactant.

Id	Name	SBO
species_96	STAT1N STAT1N*	

Table 272: Properties of each modifier.

Id	Name	SBO
species_96 species_95 species_103	STAT1N STAT1N* STAT1N-STAT1N*	

Product

Table 273: Properties of each product.

Id	Name	SBO
species_103	STAT1N-STAT1N*	

Kinetic Law

Derived unit contains undeclared units

$$v_{92} = \text{vol}\left(\text{c2}\right) \cdot \left(\text{parameter_172} \cdot \left[\text{species_96}\right] \cdot \left[\text{species_95}\right] - \text{parameter_173} \cdot \left[\text{species_103}\right]\right)$$
 (213)

8.93 Reaction reaction_79

This is a reversible reaction of one reactant forming one product influenced by two modifiers.

Name reaction-v3

Reaction equation

$$2 \text{ species_80} \xrightarrow{\text{species_80, species_81}} \text{ species_81}$$
 (214)

Table 274: Properties of each reactant.

Id	Name	SBO
species_80	IFN-R-JAK	

Table 275: Properties of each modifier.

Id	Name	SBO
species_80	IFN-R-JAK	
species_81	(IFN-R-JAK)2	

Product

Table 276: Properties of each product.

Id	Name	SBO
species_81	(IFN-R-JAK)2	

Kinetic Law

Derived unit contains undeclared units

$$v_{93} = \text{vol} \left(\text{compartment}_1 \right) \cdot \left(\text{parameter}_133 \cdot [\text{species}_80]^2 - \text{parameter}_134 \cdot [\text{species}_81] \right)$$
(215)

8.94 Reaction reaction_107

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-v30

Reaction equation

$$species_84 + species_100 \xrightarrow{species_84, species_100, species_104} species_104 \tag{216}$$

Table 277: Properties of each reactant.

Id	Name	SBO
species_84 species_100	STAT1C (IFN-R-JAK)2*-SOCS1	

Table 278: Properties of each modifier.

Id	Name	SBO
_	STAT1C (IFN-R-JAK)2*-SOCS1 (IFN-R-JAK)2*-STAT1C-SOCS1	

Product

Table 279: Properties of each product.

Id	Name	SBO
species_104	(IFN-R-JAK)2*-STAT1C-SOCS1	

Kinetic Law

Derived unit contains undeclared units

$$v_{94} = vol(compartment_1) \\ \cdot (parameter_174 \cdot [species_84] \cdot [species_100] - parameter_175 \cdot [species_104])$$
 (217)

8.95 Reaction reaction_110

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-v31

Reaction equation

$$species_108 + species_104 \xrightarrow{species_108, species_104, species_105} species_105$$
 (218)

Table 280: Properties of each reactant.

Id	Name	SBO
species_108 species_104	SHP2 (IFN-R-JAK)2*-STAT1C-SOCS1	

Table 281: Properties of each modifier.

Id	Name	SBO
_	SHP2 (IFN-R-JAK)2*-STAT1C-SOCS1 (IFN-R-JAK)2*-STAT1C-SHP2-SOCS1	

Product

Table 282: Properties of each product.

Id	Name	SBO
species_105	(IFN-R-JAK)2*-STAT1C-SHP2-SOCS1	

Kinetic Law

Derived unit contains undeclared units

$$v_{95} = \text{vol} \left(\text{compartment_1} \right)$$

$$\cdot \left(\text{parameter_176} \cdot \left[\text{species_108} \right] \cdot \left[\text{species_104} \right] - \text{parameter_177} \cdot \left[\text{species_105} \right] \right)$$
(219)

8.96 Reaction reaction_108

This is an irreversible reaction of one reactant forming four products influenced by one modifier.

Name reaction-v32

Reaction equation

$$species_105 \xrightarrow{species_105} species_99 + species_81 + species_84 + species_108$$
 (220)

Table 283: Properties of each reactant.

Id	Name	SBO
species_105	(IFN-R-JAK)2*-STAT1C-SHP2-SOCS1	

Table 284: Properties of each modifier.

Id	Name	SBO
species_105	(IFN-R-JAK)2*-STAT1C-SHP2-SOCS1	

Products

Table 285: Properties of each product.

Name	SBO
SOCS1	_
(IFN-R-JAK)2	
STAT1C	
SHP2	
	SOCS1 (IFN-R-JAK)2 STAT1C

Kinetic Law

Derived unit contains undeclared units

$$v_{96} = \text{vol} (\text{compartment_1}) \cdot \text{parameter_178} \cdot [\text{species_105}]$$
 (221)

8.97 Reaction reaction_109

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-v33

Reaction equation

$$species_{-}105 \xrightarrow{species_{-}105} species_{-}99 + species_{-}106$$
 (222)

Table 286: Properties of each reactant.

Id	Name	SBO
species_105	(IFN-R-JAK)2*-STAT1C-SHP2-SOCS1	

Table 287: Properties of each modifier.

Id	Name	SBO
species_105	(IFN-R-JAK)2*-STAT1C-SHP2-SOCS1	

Products

Table 288: Properties of each product.

Id	Name	SBO
species_99 species_106	SOCS1 (IFN-R-JAK)2*-STAT1C-SHP2	

Kinetic Law

Derived unit contains undeclared units

$$v_{97} = \text{vol} (\text{compartment}_{-1}) \cdot \text{parameter}_{-179} \cdot [\text{species}_{-105}]$$
 (223)

8.98 Reaction reaction_81

This is an irreversible reaction of one reactant forming one product influenced by one modifier.

Name reaction-v4

Reaction equation

$$species_81 \xrightarrow{species_81} species_82$$
 (224)

Table 289: Pr	operties of	each	reactant.
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Table 207. Hoperties of each reactant.		
Id	Name	SBO
species_81	(IFN-R-JAK)2	

Table 290: Properties of each modifier.

Id	Name	SBO
species_81	(IFN-R-JAK)2	

Product

Table 291: Properties of each product.

Id	Name	SBO
species_82	(IFN-R-JAK)2*	

Kinetic Law

Derived unit contains undeclared units

$$v_{98} = \text{vol}(\text{compartment}_{-1}) \cdot \text{parameter}_{-135} \cdot [\text{species}_{-81}]$$
 (225)

8.99 Reaction reaction_82

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-v5

Reaction equation

Table 292: Properties of each reactant.

Id	Name	SBO
species_82 species_84	(IFN-R-JAK)2* STAT1C	

Table 293: Properties of each modifier.

Id	Name	SBO
species_84	(IFN-R-JAK)2* STAT1C (IFN-R-JAK)2*-STAT1C	

Product

Table 294: Properties of each product.

Id	Name	SBO
species_83	(IFN-R-JAK)2*-STAT1C	

Kinetic Law

Derived unit contains undeclared units

$$v_{99} = \text{vol}(\text{compartment}_1)$$

 $\cdot (\text{parameter}_136 \cdot [\text{species}_82] \cdot [\text{species}_84] - \text{parameter}_137 \cdot [\text{species}_83])$ (227)

8.100 Reaction reaction_83

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction-v6

Reaction equation

$$species_83 \xrightarrow{species_83} species_82 + species_85$$
 (228)

Table 295: Properties of each reactant.

Id	Name	SBO
species_83	(IFN-R-JAK)2*-STAT1C	

Table 296: Properties of each modifier.

Id	Name	SBO
species_83	(IFN-R-JAK)2*-STAT1C	

Products

Table 297: Properties of each product.

Id	Name	SBO
species_82 species_85	(IFN-R-JAK)2* STAT1C*	

Kinetic Law

Derived unit contains undeclared units

$$v_{100} = \text{vol}(\text{compartment}_1) \cdot \text{parameter}_138 \cdot [\text{species}_83]$$
 (229)

8.101 Reaction reaction_84

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-v7

Reaction equation

Reactants

Table 298: Properties of each reactant.

Id	Name	SBO
species_82 species_85	(IFN-R-JAK)2* STAT1C*	

Table 299: Properties of each modifier.

Id	Name	SBO
species_85	(IFN-R-JAK)2* STAT1C* (IFN-R-JAK)2*-STAT1C*	

Table 300: Properties of each product.

Id	Name	SBO
species_86	(IFN-R-JAK)2*-STAT1C*	

Kinetic Law

Derived unit contains undeclared units

$$v_{101} = \text{vol} (\text{compartment_1})$$

$$\cdot (\text{parameter_139} \cdot [\text{species_82}] \cdot [\text{species_85}] - \text{parameter_140} \cdot [\text{species_86}])$$
(231)

8.102 Reaction reaction_85

This is a reversible reaction of one reactant forming one product influenced by two modifiers.

Name reaction-v8

Reaction equation

$$2 \text{ species_85, species_87} \text{ species_87}$$
 species_87 (232)

Reactant

Table 301: Properties of each reactant.

Id	Name	SBO
species_85	STAT1C*	

Table 302: Properties of each modifier.

Id	Name	SBO
species_85		
species_87	(STAT1C*)2	

Table 303: Properties of each product.

Id	Name	SBO
species_87	(STAT1C*)2	

Kinetic Law

Derived unit contains undeclared units

$$v_{102} = \text{vol} \left(\text{compartment_1} \right) \cdot \left(\text{parameter_141} \cdot [\text{species_85}]^2 - \text{parameter_142} \cdot [\text{species_87}] \right)$$
(233)

8.103 Reaction reaction_86

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction-v9

Reaction equation

Reactants

Table 304: Properties of each reactant.

Id	Name	SBO
species_82 species_108	(IFN-R-JAK)2* SHP2	

Table 305: Properties of each modifier.

Id	Name	SBO
species_82 species_108 species_88	(IFN-R-JAK)2* SHP2 (IFN-R-JAK)2*-SHP2	

Table 306: Properties of each product.

Id	Name	SBO
species_88	(IFN-R-JAK)2*-SHP2	

Kinetic Law

Derived unit contains undeclared units

$$v_{103} = \text{vol} (\text{compartment}_1)$$

 $\cdot (\text{parameter}_143 \cdot [\text{species}_82] \cdot [\text{species}_108] - \text{parameter}_144 \cdot [\text{species}_88])$ (235)

8.104 Reaction re112

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction_n1

Reaction equation

$$species_82 + species_11 \xrightarrow{species_82, species_11, s118} s118$$
 (236)

Reactants

Table 307: Properties of each reactant.

Id	Name	SBO
species_82 species_11	(IFN-R-JAK)2* STAT3C	

Table 308: Properties of each modifier.

Id	Name	SBO
species_82 species_11 s118	(IFN-R-JAK)2* STAT3C (IFN-R-JAK)2*-STAT3C	

Table 309: Properties of each product.

Id	Name	SBO
s118	(IFN-R-JAK)2*-STAT3C	

Kinetic Law

Derived unit contains undeclared units

$$v_{104} = vol\left(compartment_1\right) \cdot \left(parameter_221 \cdot [species_82] \cdot [species_11] - parameter_222 \cdot [s118]\right) \tag{237}$$

8.105 Reaction re117

This is an irreversible reaction of one reactant forming one product influenced by one modifier.

Name reaction_n10

Reaction equation

$$s120 \xrightarrow{s120} s122$$
 (238)

Reactant

Table 310: Properties of each reactant.

Id	Name	SBO
s120	STAT1C*-STAT3C*	

Table 311: Properties of each modifier.

Id	Name	SBO
s120	STAT1C*-STAT3C*	

Table 312: Properties of each product.

Id	Name	SBO
s122	STAT1N*-STAT3N*	

Kinetic Law

Derived unit contains undeclared units

$$v_{105} = \text{vol} (\text{compartment}_1) \cdot \text{parameter}_237 \cdot [\text{s}120]$$
 (239)

8.106 Reaction re153

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction_n11

Reaction equation

$$species_20 + s120 \xrightarrow{species_20, s120, s135} s135$$
 (240)

Reactants

Table 313: Properties of each reactant.

Id	Name	SBO
species_20 s120	PP1 STAT1C*-STAT3C*	

Table 314: Properties of each modifier.

Id	Name	SBO
species_20 s120 s135	PP1 STAT1C*-STAT3C* PP1-STAT1C*-STAT3C*	

Table 315: Properties of each product.

Id	Name	SBO
s135	PP1-STAT1C*-STAT3C*	

Kinetic Law

Derived unit contains undeclared units

$$v_{106} = \text{vol} (\text{compartment_1}) \cdot (\text{parameter_238} \cdot [\text{species_20}] \cdot [\text{s120}] - \text{parameter_239} \cdot [\text{s135}])$$
(241)

8.107 Reaction re126

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction_n12

Reaction equation

$$s122 + species_24 \xrightarrow{s122, species_24, s126} s126$$
 (242)

Reactants

Table 316: Properties of each reactant.

Id	Name	SBO
s122 species_24	STAT1N*-STAT3N* PP2	

Table 317: Properties of each modifier.

Id	Name	SBO
s122 species_24 s126	STAT1N*-STAT3N* PP2 PP2-STAT1N*-STAT3N*	

Table 318: Properties of each product.

Id	Name	SBO
s126	PP2-STAT1N*-STAT3N*	

Kinetic Law

Derived unit contains undeclared units

$$v_{107} = \text{parameter} 240 \cdot [\text{s}122] \cdot [\text{species} 24] - \text{parameter} 241 \cdot [\text{s}126]$$
 (243)

8.108 Reaction re113

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction_n2

Reaction equation

$$s118 \xrightarrow{s118} species_12 + species_82$$
 (244)

Reactant

Table 319: Properties of each reactant.

Id	Name	SBO
s118	(IFN-R-JAK)2*-STAT3C	

Table 320: Properties of each modifier.

Id	Name	SBO
s118	(IFN-R-JAK)2*-STAT3C	

Table 321: Properties of each product.

-more v=-v-representation products		
Id	Name	SBO
species_12 species_82	STAT3C* (IFN-R-JAK)2*	

Kinetic Law

Derived unit contains undeclared units

$$v_{108} = \text{vol}(\text{compartment}_1) \cdot \text{parameter}_223 \cdot [\text{s}118]$$
 (245)

8.109 Reaction re114

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction_n3

Reaction equation

$$species_12 + species_82 \xrightarrow{species_12, species_82, s119} s119$$
 (246)

Reactants

Table 322: Properties of each reactant.

Id	Name	SBO
species_12	STAT3C*	
species_82	(IFN-R-JAK)2*	

Table 323: Properties of each modifier.

Id	Name	SBO
species_12 species_82 s119	STAT3C* (IFN-R-JAK)2* (IFN-R-JAK)2*-STAT3C*	

Table 324: Properties of each product.

Id	Name	SBO
s119	(IFN-R-JAK)2*-STAT3C*	

Kinetic Law

Derived unit contains undeclared units

$$v_{109} = \text{vol} \left(\text{compartment_1} \right) \cdot \left(\text{parameter_224} \cdot [\text{species_12}] \cdot [\text{species_82}] - \text{parameter_225} \cdot [\text{s119}] \right)$$
(247)

8.110 Reaction re157

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction_n4

Reaction equation

species_9 + species_84
$$\rightleftharpoons$$
 species_84, s138 s138 (248)

Reactants

Table 325: Properties of each reactant.

Id	Name	SBO
species_9 species_84	(IL6-gp80-gp130-JAK)2* STAT1C	

Table 326: Properties of each modifier.

Id	Name	SBO
species_9 species_84 s138	(IL6-gp80-gp130-JAK)2* STAT1C (IL6-gp80-gp130-JAK)2*-STAT1	

Table 327: Properties of each product.

Id	Name	SBO
s138	(IL6-gp80-gp130-JAK)2*-STAT1	

Kinetic Law

Derived unit contains undeclared units

$$v_{110} = \text{vol} \left(\text{compartment_1} \right) \cdot \left(\text{parameter_226} \cdot [\text{species_9}] \cdot [\text{species_84}] - \text{parameter_227} \cdot [\text{s138}] \right)$$
 (249)

8.111 Reaction re158

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction_n5

Reaction equation

$$s138 \xrightarrow{s138} species_9 + species_85$$
 (250)

Reactant

Table 328: Properties of each reactant.

Id	Name	SBO
s138	(IL6-gp80-gp130-JAK)2*-STAT1	

Table 329: Properties of each modifier.

Id	Name	SBO
s138	(IL6-gp80-gp130-JAK)2*-STAT1	

Table 330: Properties of each product.

Id	Name	SBO
species_9 species_85	(IL6-gp80-gp130-JAK)2* STAT1C*	

Kinetic Law

Derived unit contains undeclared units

$$v_{111} = \text{vol}(\text{compartment_1}) \cdot \text{parameter_228} \cdot [\text{s}138]$$
 (251)

8.112 Reaction re159

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction_n6

Reaction equation

$$species_85 + species_9 \xrightarrow{species_85, species_9, s139} s139$$
 (252)

Reactants

Table 331: Properties of each reactant.

Id	Name	SBO
species_85		
species_9	(IL6-gp80-gp130-JAK)2*	

Table 332: Properties of each modifier.

Id	Name	SBO
species_85 species_9 s139	STAT1C* (IL6-gp80-gp130-JAK)2* (IL6-gp80-gp130-JAK)2*-STAT1*	

Table 333: Properties of each product.

Id	Name	SBO
s139	(IL6-gp80-gp130-JAK)2*-STAT1*	

Kinetic Law

Derived unit contains undeclared units

$$v_{112} = \text{vol} \left(\text{compartment_1} \right) \cdot \left(\text{parameter_229} \cdot \left[\text{species_85} \right] \cdot \left[\text{species_9} \right] - \text{parameter_230} \cdot \left[\text{s139} \right] \right)$$
 (253)

8.113 Reaction re116

This is an irreversible reaction of no reactant forming one product influenced by two modifiers.

Name reaction_n7

Reaction equation

$$\emptyset \xrightarrow{\text{species_92, species_92}} \text{species_30}$$
 (254)

Modifiers

Table 334: Properties of each modifier.

Id	Name	SBO
-	(STAT1N*)2 (STAT1N*)2	

Product

Table 335: Properties of each product.

Id	Name	SBO
species_30	mRNA-SOCS3N	

Kinetic Law

Derived unit contains undeclared units

$$v_{113} = \text{vol}(c3) \cdot \text{function_4_reaction_n7_1} \text{ (parameter_231, parameter_232, [species_92])}$$
 (255)

function_4_reaction_n7_1 (parameter_231, parameter_232, [species_92])
$$= \frac{\text{parameter}_231 \cdot [\text{species}_92]}{\text{parameter}_232 + [\text{species}_92]}$$
(257)

8.114 Reaction re115

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction_n8

Reaction equation

$$species_12 + species_85 \xrightarrow{species_12, species_85, s120} s120$$
 (258)

Reactants

Table 336: Properties of each reactant.

Id	Name	SBO
species_12	STAT3C*	
species_85	STAT1C*	

Table 337: Properties of each modifier.

Id	Name	SBO
species_12 species_85 s120		

Table 338: Properties of each product.

Id	Name	SBO
s120	STAT1C*-STAT3C*	

Kinetic Law

Derived unit contains undeclared units

$$v_{114} = \text{vol} \left(\text{compartment_1} \right) \cdot \left(\text{parameter_233} \cdot [\text{species_12}] \cdot [\text{species_85}] - \text{parameter_234} \cdot [\text{s120}] \right)$$
 (259)

8.115 Reaction re121

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction_n9

Reaction equation

$$species_26 + species_95 \xrightarrow{species_26, species_95, s122} s122$$
 (260)

Reactants

Table 339: Properties of each reactant.

Id	Name	SBO
species_26 species_95	STAT3N* STAT1N*	

Table 340: Properties of each modifier.

Id	Name	SBO
species_26 species_95 s122		

Table 341: Properties of each product.

Id	Name	SBO
s122	STAT1N*-STAT3N*	

Kinetic Law

Derived unit contains undeclared units

$$v_{115} = \text{parameter.} 235 \cdot [\text{species.} 26] \cdot [\text{species.} 95] - \text{parameter.} 236 \cdot [\text{s}122]$$
 (261)

8.116 Reaction re161

This is an irreversible reaction of one reactant forming three products influenced by one modifier.

Name reaction_n13

Reaction equation

$$s135 \xrightarrow{s135} species_20 + species_12 + species_84$$
 (262)

Reactant

Table 342: Properties of each reactant.

Id	Name	SBO
s135	PP1-STAT1C*-STAT3C*	

Table 343: Properties of each modifier.

Id	Name	SBO
s135	PP1-STAT1C*-STAT3C*	

Table 344: Properties of each product.

Id	Name	SBO
species_20 species_12 species_84	PP1 STAT3C* STAT1C	

Kinetic Law

Derived unit contains undeclared units

$$v_{116} = \text{vol} (\text{compartment}_1) \cdot \text{parameter}_242 \cdot [\text{s}135]$$
 (263)

8.117 Reaction re162

This is an irreversible reaction of one reactant forming three products influenced by one modifier.

Name reaction_n14

Reaction equation

$$s135 \xrightarrow{s135} species_85 + species_11 + species_20$$
 (264)

Reactant

Table 345: Properties of each reactant.

Id	Name	SBO
s135	PP1-STAT1C*-STAT3C*	

Table 346: Properties of each modifier.

Id	Name	SBO
s135	PP1-STAT1C*-STAT3C*	

Table 347: Properties of each product.

Id	Name	SBO
species_85 species_11 species_20	STAT1C* STAT3C PP1	

Kinetic Law

Derived unit contains undeclared units

$$v_{117} = \text{vol} \left(\text{compartment}_{-1} \right) \cdot \text{parameter}_{-243} \cdot [\text{s}135]$$
 (265)

8.118 Reaction re163

This is an irreversible reaction of one reactant forming three products influenced by one modifier.

Name reaction_n15

Reaction equation

$$s126 \xrightarrow{s126} species_26 + species_24 + species_96$$
 (266)

Reactant

Table 348: Properties of each reactant.

Id	Name	SBO
s126	PP2-STAT1N*-STAT3N*	

Modifier

Table 349: Properties of each modifier.

Id	Name	SBO
s126	PP2-STAT1N*-STAT3N*	

Table 350: Properties of each product.

Id	Name	SBO
species_26	STAT3N*	
species_24	PP2	
species_96	STAT1N	

Kinetic Law

Derived unit contains undeclared units

$$v_{118} = parameter_244 \cdot [s126] \tag{267}$$

8.119 Reaction re164

This is an irreversible reaction of one reactant forming three products influenced by one modifier.

Name reaction_n16

Reaction equation

$$s126 \xrightarrow{s126} species_95 + species_28 + species_24$$
 (268)

Reactant

Table 351: Properties of each reactant.

Id	Name	SBO
s126	PP2-STAT1N*-STAT3N*	

Modifier

Table 352: Properties of each modifier.

Id	Name	SBO
s126	PP2-STAT1N*-STAT3N*	

Table 353: Properties of each product.

Id	Name	SBO
species_95 species_28 species_24		

Kinetic Law

Derived unit contains undeclared units

$$v_{119} = parameter_245 \cdot [s126] \tag{269}$$

8.120 Reaction reaction_2

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction_n17

Reaction equation

$$s118 + species_84 \xrightarrow{s118, species_84, species_68} species_68$$
 (270)

Reactants

Table 354: Properties of each reactant.

Id	Name	SBO
s118 species_84	(IFN-R-JAK)2*-STAT3C STAT1C	

Modifiers

Table 355: Properties of each modifier.

Id	Name	SBO
s118 species_84 species_68	(IFN-R-JAK)2*-STAT3C STAT1C (IFN-R-JAK)2*-STAT1C-STAT3C	

Table 356: Properties of each product.

Id	Name	SBO
species_68	(IFN-R-JAK)2*-STAT1C-STAT3C	

Kinetic Law

Derived unit contains undeclared units

$$v_{120} = vol (compartment_1) \cdot (parameter_7 \cdot [s118] \cdot [species_84] - parameter_8 \cdot [species_68])$$
 (271)

8.121 Reaction reaction_10

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction_n18

Reaction equation

Reactants

Table 357: Properties of each reactant.

Id	Name	SBO
species_83 species_11	(IFN-R-JAK)2*-STAT1C STAT3C	

Modifiers

Table 358: Properties of each modifier.

Id	Name	SBO
species_11	(IFN-R-JAK)2*-STAT1C STAT3C (IFN-R-JAK)2*-STAT1C-STAT3C	

Table 359: Properties of each product.

Id	Name	SBO
species_68	(IFN-R-JAK)2*-STAT1C-STAT3C	

Kinetic Law

Derived unit contains undeclared units

$$v_{121} = \text{vol} (\text{compartment}_1)$$

 $\cdot (\text{parameter}_9 \cdot [\text{species}_83] \cdot [\text{species}_11] - \text{parameter}_10 \cdot [\text{species}_68])$ (273)

8.122 Reaction reaction_11

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction_n19

Reaction equation

species_68
$$\xrightarrow{\text{species}_68}$$
 s118 + species_85 (274)

Reactant

Table 360: Properties of each reactant.

Id	Name	SBO
species_68	(IFN-R-JAK)2*-STAT1C-STAT3C	

Modifier

Table 361: Properties of each modifier.

Id	Name	SBO
species_68	(IFN-R-JAK)2*-STAT1C-STAT3C	

Table 362: Properties of each product.

Id	Name	SBO
s118 species_85	(IFN-R-JAK)2*-STAT3C STAT1C*	

Kinetic Law

Derived unit contains undeclared units

$$v_{122} = \text{vol} (\text{compartment}_{-1}) \cdot \text{parameter}_{-19} \cdot [\text{species}_{-68}]$$
 (275)

8.123 Reaction reaction_41

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction_n20

Reaction equation

$$species_68 \xrightarrow{species_68} species_83 + species_12$$
 (276)

Reactant

Table 363: Properties of each reactant.

Id	Name	SBO
species_68	(IFN-R-JAK)2*-STAT1C-STAT3C	

Modifier

Table 364: Properties of each modifier.

Id	Name	SBO
species_68	(IFN-R-JAK)2*-STAT1C-STAT3C	

Table 365: Properties of each product.

Tuest tot. Treperiors of tuest products		
Id	Name	SBO
species_83 species_12	(IFN-R-JAK)2*-STAT1C STAT3C*	

Kinetic Law

Derived unit contains undeclared units

$$v_{123} = \text{vol} (\text{compartment_1}) \cdot \text{parameter_20} \cdot [\text{species_68}]$$
 (277)

8.124 Reaction reaction_67

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction_n21

Reaction equation

Reactants

Table 366: Properties of each reactant.

Id	Name	SBO
species_10	(IL6-gp80-gp130-JAK)2*-STAT3C	
species_84	STAT1C	

Modifiers

Table 367: Properties of each modifier.

Id	Name	SBO
species_84	(IL6-gp80-gp130-JAK)2*-STAT3C STAT1C (IL6-gp80-gp130-JAK)2*-STAT1C-STAT3C	

Table 368: Properties of each product.

Id	Name	SBO
species_17	(IL6-gp80-gp130-JAK)2*-STAT1C-STAT3C	

Kinetic Law

Derived unit contains undeclared units

$$v_{124} = \text{parameter} \cdot 21 \cdot [\text{species} \cdot 10] \cdot [\text{species} \cdot 84] - \text{parameter} \cdot 22 \cdot [\text{species} \cdot 17]$$
 (279)

8.125 Reaction reaction_68

This is a reversible reaction of two reactants forming one product influenced by three modifiers.

Name reaction_n22

Reaction equation

$$species_67 + species_11 \xrightarrow{species_67, species_11, species_17} species_17 \tag{280}$$

Reactants

Table 369: Properties of each reactant.

Id	Name	SBO
species_67 species_11	(IL6-gp80-gp130-JAK)2*-STAT1C STAT3C	

Modifiers

Table 370: Properties of each modifier.

Id	Name	SBO
species_11	(IL6-gp80-gp130-JAK)2*-STAT1C STAT3C (IL6-gp80-gp130-JAK)2*-STAT1C-STAT3C	

Table 371: Properties of each product.

Id	Name	SBO
species_17	(IL6-gp80-gp130-JAK)2*-STAT1C-STAT3C	

Kinetic Law

Derived unit contains undeclared units

$$v_{125} = \text{parameter} \cdot 23 \cdot [\text{species} \cdot 67] \cdot [\text{species} \cdot 11] - \text{parameter} \cdot 24 \cdot [\text{species} \cdot 17]$$
 (281)

8.126 Reaction reaction_69

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction_n23

Reaction equation

species_17
$$\xrightarrow{\text{species}_17}$$
 species_10 + species_85 (282)

Reactant

Table 372: Properties of each reactant.

Id	Name	SBO
species_17	(IL6-gp80-gp130-JAK)2*-STAT1C-STAT3C	

Modifier

Table 373: Properties of each modifier.

Id	Name	SBO
species_17	(IL6-gp80-gp130-JAK)2*-STAT1C-STAT3C	

Table 374: Properties of each product.

Id	Name	SBO
species_10 species_85	(IL6-gp80-gp130-JAK)2*-STAT3C STAT1C*	

Kinetic Law

Derived unit contains undeclared units

$$v_{126} = \text{parameter} \cdot 25 \cdot [\text{species} \cdot 17]$$
 (283)

8.127 Reaction reaction_70

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name reaction_n24

Reaction equation

$$species_{17} \xrightarrow{species_{17}} species_{67} + species_{12}$$
 (284)

Reactant

Table 375: Properties of each reactant.

Id	Name	SBO
species_17	(IL6-gp80-gp130-JAK)2*-STAT1C-STAT3C	

Modifier

Table 376: Properties of each modifier.

Id	Name	SBO
species_17	(IL6-gp80-gp130-JAK)2*-STAT1C-STAT3C	

Table 377: Properties of each product.

Id	Name	SBO
species_67 species_12	(IL6-gp80-gp130-JAK)2*-STAT1C STAT3C*	

Kinetic Law

Derived unit contains undeclared units

$$v_{127} = \text{parameter} \cdot 26 \cdot [\text{species} \cdot 17]$$
 (285)

9 Derived Rate Equations

When interpreted as an ordinary differential equation framework, this model implies the following set of equations for the rates of change of each species.

Identifiers for kinetic laws highlighted in gray cannot be verified to evaluate to units of SBML substance per time. As a result, some SBML interpreters may not be able to verify the consistency of the units on quantities in the model. Please check if

- parameters without an unit definition are involved or
- volume correction is necessary because the hasOnlySubstanceUnits flag may be set to false and spacialDimensions> 0 for certain species.

9.1 Species species_1

Name IL6

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in two reactions (as a reactant in reaction_1 and as a modifier in reaction_1), which do not influence its rate of change because this constant species is on the boundary of the reaction system:

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{species}_{-1} = 0 \tag{286}$$

9.2 Species species_79

Name IFN

Initial concentration $0.1 \text{ nmol} \cdot l^{-1}$

This species takes part in two reactions (as a reactant in reaction_80 and as a modifier in reaction_80), which do not influence its rate of change because this constant species is on the boundary of the reaction system:

$$\frac{d}{dt} \text{species}_{.79} = 0 \tag{287}$$

9.3 Species species_17

Name (IL6-gp80-gp130-JAK)2*-STAT1C-STAT3C

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in eight reactions (as a reactant in reaction_69, reaction_70 and as a product in reaction_67, reaction_68 and as a modifier in reaction_67, reaction_68, reaction_69, reaction_70).

$$\frac{d}{dt} \text{species}_{17} = v_{124} + v_{125} - v_{126} - v_{127}$$
 (288)

9.4 Species species_67

Name (IL6-gp80-gp130-JAK)2*-STAT1C

Initial concentration 0 nmol ⋅ 1⁻¹

This species takes part in three reactions (as a reactant in reaction_68 and as a product in reaction_70 and as a modifier in reaction_68).

$$\frac{d}{dt} \text{species}_{.67} = |v_{127}| - |v_{125}| \tag{289}$$

9.5 Species s118

Name (IFN-R-JAK)2*-STAT3C

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in seven reactions (as a reactant in re113, reaction_2 and as a product in re112, reaction_11 and as a modifier in re112, re113, reaction_2).

$$\frac{\mathrm{d}}{\mathrm{d}t}s118 = |v_{104}| + |v_{122}| - |v_{108}| - |v_{120}| \tag{290}$$

9.6 Species s119

Name (IFN-R-JAK)2*-STAT3C*

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in two reactions (as a product in re114 and as a modifier in re114).

$$\frac{d}{dt}s119 = v_{109} \tag{291}$$

9.7 Species s120

Name STAT1C*-STAT3C*

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in six reactions (as a reactant in re117, re153 and as a product in re115 and as a modifier in re117, re153, re115).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{s}120 = |v_{114}| - |v_{105}| - |v_{106}| \tag{292}$$

9.8 Species s122

Name STAT1N*-STAT3N*

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in five reactions (as a reactant in re126 and as a product in re117, re121 and as a modifier in re126, re121).

$$\frac{\mathrm{d}}{\mathrm{d}t}s122 = v_{105} + v_{115} - v_{107} \tag{293}$$

9.9 Species s126

Name PP2-STAT1N*-STAT3N*

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in six reactions (as a reactant in re163, re164 and as a product in re126 and as a modifier in re126, re163, re164).

$$\frac{\mathrm{d}}{\mathrm{d}t}s126 = |v_{107}| - |v_{118}| - |v_{119}| \tag{294}$$

9.10 Species s135

Name PP1-STAT1C*-STAT3C*

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in six reactions (as a reactant in re161, re162 and as a product in re153 and as a modifier in re153, re161, re162).

$$\frac{\mathrm{d}}{\mathrm{d}t}s135 = v_{106} - v_{116} - v_{117} \tag{295}$$

9.11 Species species_2

Name gp80

Initial concentration $8 \text{ nmol} \cdot 1^{-1}$

This species takes part in two reactions (as a reactant in reaction_1 and as a modifier in reaction_1), which do not influence its rate of change because this constant species is on the boundary of the reaction system:

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{species}.2 = 0 \tag{296}$$

9.12 Species species_3

Name IL6-gp80

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in four reactions (as a reactant in reaction_3 and as a product in reaction_1 and as a modifier in reaction_1, reaction_3).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{species}_{3} = v_{2} - v_{18} \tag{297}$$

9.13 Species species_4

Name gp130

Initial concentration $0.8 \text{ nmol} \cdot l^{-1}$

This species takes part in two reactions (as a reactant in re160 and as a modifier in re160), which do not influence its rate of change because this constant species is on the boundary of the reaction system:

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{species}_{-}4 = 0 \tag{298}$$

9.14 Species species_5

Name JAK(IFN)

Initial concentration 12 nmol·1⁻¹

This species takes part in two reactions (as a reactant in re135 and as a modifier in re135).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{species}_5 = -\nu_{71} \tag{299}$$

9.15 Species species_6

Name gp130-JAK

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in four reactions (as a reactant in reaction_3 and as a product in re160 and as a modifier in re160, reaction_3).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{species}_{.6} = |v_1| - |v_{18}| \tag{300}$$

9.16 Species species_7

Name IL6-gp80-gp130-JAK

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in four reactions (as a reactant in reaction_5 and as a product in reaction_3 and as a modifier in reaction_3, reaction_5).

$$\frac{d}{dt} \text{species}_{-7} = |v_{18}| - 2 v_{49} \tag{301}$$

9.17 Species species_8

Name (IL6-gp80-gp130-JAK)2

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in six reactions (as a reactant in reaction_6 and as a product in re137, reaction_73, reaction_5 and as a modifier in reaction_5, reaction_6).

$$\frac{d}{dt} \text{species}_{8} = |v_{6}| + |v_{27}| + |v_{49}| - |v_{60}|$$
(302)

9.18 Species species_9

Name (IL6-gp80-gp130-JAK)2*

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in 21 reactions (as a reactant in reaction_8, re138, reaction_26, reaction_4, re157, re159 and as a product in reaction_37, reaction_42, reaction_44, reaction_6, reaction_7, re158 and as a modifier in reaction_8, re138, reaction_26, reaction_37, reaction_42, reaction_44, reaction_4, re157, re159).

$$\frac{d}{dt} \text{species}_{9} = v_{37} + v_{41} + v_{43} + v_{60} + v_{70} + v_{111} - v_{3} - v_{5} - v_{24} - v_{69} - v_{110} - v_{112}$$
(303)

9.19 Species species_10

Name (IL6-gp80-gp130-JAK)2*-STAT3C

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in seven reactions (as a reactant in reaction_7, reaction_67 and as a product in reaction_4, reaction_69 and as a modifier in reaction_4, reaction_7, reaction_67).

$$\frac{d}{dt} \text{species}_{10} = |v_{69}| + |v_{126}| - |v_{70}| - |v_{124}| \tag{304}$$

9.20 Species species_11

Name STAT3C

Initial concentration 1000 nmol·l⁻¹

This species takes part in 13 reactions (as a reactant in reaction_16, reaction_4, re112, reaction_10, reaction_68 and as a product in reaction_14, reaction_22, re162 and as a modifier in reaction_16, reaction_4, re112, reaction_10, reaction_68).

$$\frac{d}{dt} \text{species}_{11} = v_8 + v_{20} + v_{117} - v_{11} - v_{69} - v_{104} - v_{121} - v_{125}$$
 (305)

9.21 Species species_12

Name STAT3C*

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in 17 reactions (as a reactant in reaction_8, reaction_9, reaction_12, reaction_16, re114, re115 and as a product in reaction_7, re113, re161, reaction_41, reaction_70 and as a modifier in reaction_8, reaction_9, reaction_12, reaction_16, re114, re115).

$$\frac{d}{dt} \text{species}_{12} = v_{70} + v_{108} + v_{116} + v_{123} + v_{127} - v_{3} - 2v_{4} - v_{7} - v_{11} - v_{109} - v_{114}$$
(306)

9.22 Species species_13

Name (IL6-gp80-gp130-JAK)2*-STAT3C*

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in two reactions (as a product in reaction_8 and as a modifier in reaction_8).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{species}_{-13} = v_3 \tag{307}$$

9.23 Species species_14

Name (STAT3C*)2

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in six reactions (as a reactant in reaction_13, reaction_17 and as a product in reaction_9 and as a modifier in reaction_9, reaction_13, reaction_17).

$$\frac{d}{dt} \text{species}_{14} = |v_4| - |v_9| - |v_{12}| \tag{308}$$

9.24 Species species_15

Name (IL6-gp80-gp130-JAK)2*-SOCS3

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in four reactions (as a reactant in reaction_73 and as a product in reaction_26 and as a modifier in reaction_26, reaction_73).

$$\frac{d}{dt} \text{species}_{-15} = |v_{24}| - |v_{27}| \tag{309}$$

9.25 Species species_16

Name (IL6-gp80-gp130-JAK)2*-SHP2

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in six reactions (as a reactant in re137, reaction_39 and as a product in re138 and as a modifier in re138, re137, reaction_39).

$$\frac{d}{dt} \text{species}_{-16} = |v_5| - |v_6| - |v_{28}| \tag{310}$$

9.26 Species species_18

Name STAT3C-STAT3C*

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in three reactions (as a product in reaction_15, reaction_16 and as a modifier in reaction_16).

$$\frac{d}{dt} \text{species}_{-}18 = |v_{10}| + |v_{11}| \tag{311}$$

9.27 Species species_19

Name SOCS3

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in six reactions (as a reactant in reaction_26, reaction_28 and as a product in reaction_25, reaction_73 and as a modifier in reaction_26, reaction_28).

$$\frac{d}{dt} \text{species}_{19} = |v_{23}| + |v_{27}| - |v_{24}| - |v_{26}|$$
(312)

9.28 Species species_20

Name PP1

Initial concentration 50 nmol·1⁻¹

This species takes part in 16 reactions (as a reactant in reaction_12, reaction_13, re140, re142, re153 and as a product in reaction_14, reaction_15, re141, re143, re161, re162 and as a modifier in reaction_12, reaction_13, re140, re142, re153).

$$\frac{d}{dt} \text{species} 20 = v_8 + v_{10} + v_{74} + v_{88} + v_{116} + v_{117} - v_7 - v_9 - v_{73} - v_{87} - v_{106}$$
(313)

9.29 Species species_21

Name PP1-STAT3C*

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in four reactions (as a reactant in reaction_14 and as a product in reaction_12 and as a modifier in reaction_12, reaction_14).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{species.}21 = |v_7| - v_8 \tag{314}$$

9.30 Species species_22

Name PP1-(STAT3C*)2

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in four reactions (as a reactant in reaction_15 and as a product in reaction_13 and as a modifier in reaction_13, reaction_15).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{species}.22 = v_9 - v_{10} \tag{315}$$

9.31 Species species_31

Name mRNA-SOCS3C

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in five reactions (as a reactant in reaction_27 and as a product in reaction_24 and as a modifier in reaction_25, reaction_25, reaction_27).

$$\frac{d}{dt}$$
 species_31 = $v_{22} - v_{25}$ (316)

9.32 Species species_32

Name Grb2

Initial concentration 85 nmol·1⁻¹

This species takes part in six reactions (as a reactant in reaction_29, reaction_43 and as a product in reaction_40 and as a modifier in reaction_29, reaction_40, reaction_43).

$$\frac{d}{dt} \text{species}_{32} = |v_{39}| - |v_{29}| - |v_{42}| \tag{317}$$

9.33 Species species_33

Name (IL6-gp80-gp130-JAK)2*-SHP2*

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in eight reactions (as a reactant in reaction_29, reaction_42, reaction_46 and as a product in reaction_39 and as a modifier in reaction_39, reaction_29, reaction_42, reaction_46).

$$\frac{d}{dt} \text{species}_{33} = |v_{28}| - |v_{29}| - |v_{41}| - |v_{45}| \tag{318}$$

9.34 Species species_34

Name (IL6-gp80-gp130-JAK)2*-SHP2*-Grb2

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in six reactions (as a reactant in reaction_30, reaction_44 and as a product in reaction_29 and as a modifier in reaction_29, reaction_30, reaction_44).

$$\frac{d}{dt} \text{species}_{34} = |v_{29}| - |v_{30}| - |v_{43}| \tag{319}$$

9.35 Species species_35

Name SOS

Initial concentration 34 nmol·l⁻¹

This species takes part in six reactions (as a reactant in reaction_30, reaction_45 and as a product in reaction_40 and as a modifier in reaction_30, reaction_40, reaction_45).

$$\frac{d}{dt} \text{species}_{35} = |v_{39}| - |v_{30}| - |v_{44}| \tag{320}$$

9.36 Species species_36

Name SHP2*-Grb2

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in six reactions (as a reactant in reaction_45 and as a product in reaction_43, reaction_44 and as a modifier in reaction_43, reaction_44, reaction_45).

$$\frac{d}{dt} \text{species}_{36} = |v_{42}| + |v_{43}| - v_{44}$$
 (321)

9.37 Species species_37

Name (IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in 14 reactions (as a reactant in reaction_31, reaction_35, reaction_37 and as a product in reaction_30, reaction_32, reaction_36, reaction_46 and as a modifier in reaction_30, reaction_31, reaction_32, reaction_35, reaction_36, reaction_37, reaction_46).

$$\frac{d}{dt} \text{species}_{37} = v_{30} + v_{32} + v_{36} + v_{45} - v_{31} - v_{35} - v_{37}$$
(322)

9.38 Species species_38

Name Ras-GDP

Initial concentration 19000 nmol·1⁻¹

This species takes part in four reactions (as a reactant in reaction_31 and as a product in reaction_36 and as a modifier in reaction_31, reaction_36).

$$\frac{d}{dt}$$
 species_38 = $|v_{36}| - |v_{31}|$ (323)

9.39 Species species_39

Name (IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS-Ras-GDP

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in four reactions (as a reactant in reaction_32 and as a product in reaction_31 and as a modifier in reaction_31, reaction_32).

$$\frac{d}{dt} \text{species}_{39} = |v_{31}| - |v_{32}| \tag{324}$$

9.40 Species species_40

Name Ras-GTP

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in four reactions (as a reactant in reaction_33 and as a product in reaction_32 and as a modifier in reaction_32, reaction_33).

$$\frac{d}{dt} \text{species}_{40} = |v_{32}| - |v_{33}| \tag{325}$$

9.41 Species species_41

Name Raf

Initial concentration 67 nmol·1⁻¹

This species takes part in three reactions (as a reactant in reaction_33 and as a product in reaction_48 and as a modifier in reaction_33).

$$\frac{d}{dt} \text{species}_{-}41 = v_{47} - v_{33} \tag{326}$$

9.42 Species species 42

Name Raf-Ras-GTP

Initial concentration 0 nmol·1⁻¹

This species takes part in four reactions (as a reactant in reaction_34 and as a product in reaction_33 and as a modifier in reaction_33, reaction_34).

$$\frac{d}{dt} \text{species}_{42} = |v_{33}| - |v_{34}| \tag{327}$$

9.43 Species species_43

Name Ras-GTP*

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in four reactions (as a reactant in reaction_35 and as a product in reaction_34 and as a modifier in reaction_34, reaction_35).

$$\frac{d}{dt} \text{species}_{43} = |v_{34}| - |v_{35}| \tag{328}$$

9.44 Species species_44

Name Raf*

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in ten reactions (as a reactant in reaction_47, reaction_49, reaction_51 and as a product in reaction_34, reaction_50, reaction_52 and as a modifier in reaction_34, reaction_47, reaction_49, reaction_51).

$$\frac{d}{dt} \text{species}_{44} = |v_{34}| + |v_{50}| + |v_{52}| - |v_{46}| - |v_{48}| - |v_{51}|$$
(329)

9.45 Species species_45

Name (IL6-gp80-gp130-JAK)2*-SHP2*-Grb2-SOS-Ras-GTP

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in four reactions (as a reactant in reaction_36 and as a product in reaction_35 and as a modifier in reaction_35, reaction_36).

$$\frac{d}{dt} \text{species}_{-}45 = |v_{35}| - |v_{36}| \tag{330}$$

9.46 Species species_46

Name SHP2*-Grb2-SOS

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in six reactions (as a reactant in reaction_38 and as a product in reaction_37, reaction_45 and as a modifier in reaction_37, reaction_38, reaction_45).

$$\frac{d}{dt} \text{species}_{46} = |v_{37}| + |v_{44}| - |v_{38}| \tag{331}$$

9.47 Species species_47

Name Grb2-SOS

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in six reactions (as a reactant in reaction_40, reaction_46 and as a product in reaction_38 and as a modifier in reaction_38, reaction_40, reaction_46).

$$\frac{d}{dt} \text{species}_{47} = |v_{38}| - |v_{39}| - |v_{45}| \tag{332}$$

9.48 Species species 48

Name SHP2*

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in eight reactions (as a reactant in re136, reaction_43 and as a product in reaction_38, reaction_42 and as a modifier in reaction_38, re136, reaction_42, reaction_43).

$$\frac{d}{dt} \text{species}_{48} = |v_{38}| + |v_{41}| - |v_{40}| - |v_{42}| \tag{333}$$

9.49 Species species_49

Name Phosp1

Initial concentration 67 nmol·1⁻¹

This species takes part in three reactions (as a reactant in reaction_47 and as a product in reaction_48 and as a modifier in reaction_47).

$$\frac{d}{dt} \text{species}_{49} = |v_{47}| - |v_{46}| \tag{334}$$

9.50 Species species_50

Name Raf*-Phosp1

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in four reactions (as a reactant in reaction_48 and as a product in reaction_47 and as a modifier in reaction_47, reaction_48).

$$\frac{d}{dt} \text{species}_{.50} = |v_{46}| - |v_{47}| \tag{335}$$

9.51 Species species_51

Name MEK

Initial concentration 41667 nmol·l⁻¹

This species takes part in three reactions (as a reactant in reaction_49 and as a product in reaction_56 and as a modifier in reaction_49).

$$\frac{d}{dt} \text{species}_51 = |v_{56}| - |v_{48}| \tag{336}$$

9.52 Species species_52

Name MEK-Raf*

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in four reactions (as a reactant in reaction_50 and as a product in reaction_49 and as a modifier in reaction_49, reaction_50).

$$\frac{d}{dt} \text{species}_{.52} = |v_{48}| - |v_{50}| \tag{337}$$

9.53 Species species_53

Name MEK-P

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in six reactions (as a reactant in reaction_51, reaction_55 and as a product in reaction_50, reaction_54 and as a modifier in reaction_51, reaction_55).

$$\frac{d}{dt} \text{species}_53 = |v_{50}| + |v_{54}| - |v_{51}| - |v_{55}| \tag{338}$$

9.54 Species species_54

Name MEK-P-Raf*

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in four reactions (as a reactant in reaction_52 and as a product in reaction_51 and as a modifier in reaction_51, reaction_52).

$$\frac{d}{dt} \text{species}_54 = |v_{51}| - |v_{52}| \tag{339}$$

9.55 Species species_55

Name MEK-PP

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in nine reactions (as a reactant in reaction_53, reaction_57, reaction_59 and as a product in reaction_52, reaction_58, reaction_60 and as a modifier in reaction_53, reaction_57, reaction_59).

$$\frac{d}{dt} \text{species}_55 = |v_{52}| + |v_{58}| + |v_{61}| - |v_{53}| - |v_{57}| - |v_{59}|$$
(340)

9.56 Species species_56

Name MEK-PP-Phosp2

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in four reactions (as a reactant in reaction_54 and as a product in reaction_53 and as a modifier in reaction_53, reaction_54).

$$\frac{d}{dt} \text{species}_56 = |v_{53}| - |v_{54}| \tag{341}$$

9.57 Species species_57

Name Phosp2

Initial concentration $67 \text{ nmol} \cdot l^{-1}$

This species takes part in six reactions (as a reactant in reaction_53, reaction_55 and as a product in reaction_54, reaction_56 and as a modifier in reaction_53, reaction_55).

$$\frac{d}{dt} \text{species}_57 = |v_{54}| + |v_{56}| - |v_{53}| - |v_{55}| \tag{342}$$

9.58 Species species_58

Name MEK-P-Phosp2

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in four reactions (as a reactant in reaction_56 and as a product in reaction_55 and as a modifier in reaction_55, reaction_56).

$$\frac{d}{dt} \text{species}_{.58} = |v_{55}| - |v_{56}| \tag{343}$$

9.59 Species species_59

Name ERK

Initial concentration 35000 nmol·1⁻¹

This species takes part in three reactions (as a reactant in reaction_57 and as a product in reaction_63 and as a modifier in reaction_57).

$$\frac{d}{dt} \text{species}_{.59} = |v_{65}| - |v_{57}| \tag{344}$$

9.60 Species species_60

Name ERK-MEK-PP

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in four reactions (as a reactant in reaction_58 and as a product in reaction_57 and as a modifier in reaction_57, reaction_58).

$$\frac{d}{dt} \text{species}_{.60} = |v_{57}| - |v_{58}| \tag{345}$$

9.61 Species species_61

Name ERK-P

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in six reactions (as a reactant in reaction_59, reaction_62 and as a product in reaction_58, reaction_61 and as a modifier in reaction_59, reaction_62).

$$\frac{d}{dt} \text{species}_{61} = |v_{58}| + |v_{63}| - |v_{59}| - |v_{64}| \tag{346}$$

9.62 Species species_62

Name ERK-P-MEK-PP

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in four reactions (as a reactant in reaction_60 and as a product in reaction_59 and as a modifier in reaction_59, reaction_60).

$$\frac{d}{dt} \text{species}_{.62} = |v_{59}| - |v_{61}| \tag{347}$$

9.63 Species species_63

Name ERK-PP

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in five reactions (as a reactant in reaction_64 and as a product in reaction_60 and as a modifier in reaction_64, reaction_74, reaction_74).

$$\frac{d}{dt}$$
 species_63 = $v_{61} - v_{62}$ (348)

9.64 Species species_64

Name Phosp3

Initial concentration $16667 \text{ nmol} \cdot l^{-1}$

This species takes part in six reactions (as a reactant in reaction_64, reaction_62 and as a product in reaction_61, reaction_63 and as a modifier in reaction_64, reaction_62).

$$\frac{d}{dt} \text{species}_{64} = |v_{63}| + |v_{65}| - |v_{62}| - |v_{64}| \tag{349}$$

9.65 Species species_65

Name ERK-PP-Phosp3

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in four reactions (as a reactant in reaction_61 and as a product in reaction_64 and as a modifier in reaction_64, reaction_61).

$$\frac{d}{dt} \text{species}_{-65} = |v_{62}| - |v_{63}| \tag{350}$$

9.66 Species species_66

Name ERK-P-Phosp3

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in four reactions (as a reactant in reaction_63 and as a product in reaction_62 and as a modifier in reaction_62, reaction_63).

$$\frac{d}{dt} \text{species}_{.66} = |v_{64}| - |v_{65}| \tag{351}$$

9.67 Species species_74

Name CEBPi

Initial concentration 40493 nmol·1⁻¹

This species takes part in three reactions (as a reactant in reaction_74 and as a product in reaction_75 and as a modifier in reaction_74).

$$\frac{d}{dt} \text{species}_{.74} = |v_{67}| - |v_{66}| \tag{352}$$

9.68 Species species_75

Name CEBP

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in five reactions (as a reactant in reaction_75, reaction_76 and as a product in reaction_74 and as a modifier in reaction_75, reaction_76).

$$\frac{d}{dt} \text{species}_{.75} = v_{66} - v_{67} - 2 v_{68} \tag{353}$$

9.69 Species species_76

Name CEBPn

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in two reactions (as a product in reaction_76 and as a modifier in reaction_76).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{species}_{.76} = v_{68} \tag{354}$$

9.70 Species species_78

Name R-JAK

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in four reactions (as a reactant in reaction_80 and as a product in re135 and as a modifier in re135, reaction_80).

$$\frac{d}{dt} \text{species}_{.78} = |v_{71}| - |v_{82}| \tag{355}$$

9.71 Species species_80

Name IFN-R-JAK

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in four reactions (as a reactant in reaction_79 and as a product in reaction_80 and as a modifier in reaction_80, reaction_79).

$$\frac{d}{dt} \text{species}_{-80} = |v_{82}| - 2 v_{93} \tag{356}$$

9.72 Species species_81

Name (IFN-R-JAK)2

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in six reactions (as a reactant in reaction_81 and as a product in reaction_87, reaction_79, reaction_108 and as a modifier in reaction_79, reaction_81).

$$\frac{d}{dt} \text{species}_{81} = |v_{72}| + |v_{93}| + |v_{96}| - |v_{98}| \tag{357}$$

9.73 Species species_82

Name (IFN-R-JAK)2*

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in 15 reactions (as a reactant in reaction_98, reaction_82, reaction_84, reaction_86, re112, re114 and as a product in reaction_81, reaction_83, re113 and as a modifier in reaction_98, reaction_82, reaction_84, reaction_86, re112, re114).

$$\frac{d}{dt} species_{-}82 = v_{98} + v_{100} + v_{108} - v_{84} - v_{99} - v_{101} - v_{103} - v_{104} - v_{109}$$
 (358)

9.74 Species species_83

Name (IFN-R-JAK)2*-STAT1C

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in seven reactions (as a reactant in reaction_83, reaction_10 and as a product in reaction_82, reaction_41 and as a modifier in reaction_82, reaction_83, reaction_10).

$$\frac{d}{dt} \text{species}_{-83} = |v_{99}| + |v_{123}| - |v_{100}| - |v_{121}|$$
(359)

9.75 Species species_84

Name STAT1C

Initial concentration $1000 \text{ nmol} \cdot 1^{-1}$

This species takes part in 16 reactions (as a reactant in reaction_90, reaction_107, reaction_82, re157, reaction_2, reaction_67 and as a product in re141, reaction_94, reaction_108, re161 and as a modifier in reaction_90, reaction_107, reaction_82, re157, reaction_2, reaction_67).

$$\frac{d}{dt} species_84 = v_{74} + v_{79} + v_{96} + v_{116} - v_{75} - v_{94} - v_{99} - v_{110} - v_{120} - v_{124}$$
 (360)

9.76 Species species_85

Name STAT1C*

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in 17 reactions (as a reactant in re140, reaction_90, reaction_84, reaction_85, re159, re115 and as a product in reaction_83, re158, re162, reaction_11, reaction_69 and as a modifier in re140, reaction_90, reaction_84, reaction_85, re159, re115).

$$\frac{d}{dt} \text{species_85} = v_{100} + v_{111} + v_{117} + v_{122} + v_{126} - v_{73} - v_{75} - v_{101} - 2 v_{102} - v_{112} - v_{114}$$
(361)

9.77 Species species_86

Name (IFN-R-JAK)2*-STAT1C*

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in two reactions (as a product in reaction_84 and as a modifier in reaction_84).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{species}_{.86} = v_{101} \tag{362}$$

9.78 Species species_87

Name (STAT1C*)2

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in six reactions (as a reactant in reaction_91, re142 and as a product in reaction_85 and as a modifier in reaction_91, re142, reaction_85).

$$\frac{d}{dt} \text{species_87} = |v_{102}| - |v_{76}| - |v_{87}| \tag{363}$$

9.79 Species species_88

Name (IFN-R-JAK)2*-SHP2

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in four reactions (as a reactant in reaction_87 and as a product in reaction_86 and as a modifier in reaction_87, reaction_86).

$$\frac{d}{dt} \text{species}_{88} = |v_{103}| - |v_{72}| \tag{364}$$

9.80 Species species_90

Name PP1-STAT1C*

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in four reactions (as a reactant in re141 and as a product in re140 and as a modifier in re140, re141).

$$\frac{d}{dt} \text{species}_{.90} = |v_{73}| - |v_{74}| \tag{365}$$

9.81 Species species_91

Name STAT1C-STAT1C*

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in three reactions (as a product in reaction_90, re143 and as a modifier in reaction_90).

$$\frac{d}{dt} \text{species}_{.91} = |v_{75}| + |v_{88}| \tag{366}$$

9.82 Species species_98

Name mRNA-SOCS1C

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in five reactions (as a reactant in reaction_99 and as a product in reaction_96 and as a modifier in reaction_97, reaction_97, reaction_99).

$$\frac{d}{dt} \text{species}_{.98} = |v_{81}| - |v_{85}| \tag{367}$$

9.83 Species species_99

Name SOCS1

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in seven reactions (as a reactant in reaction_98, reaction_100 and as a product in reaction_97, reaction_108, reaction_109 and as a modifier in reaction_98, reaction_100).

$$\frac{d}{dt} \text{species_99} = |v_{83}| + |v_{96}| + |v_{97}| - |v_{84}| - |v_{86}|$$
(368)

9.84 Species species_100

Name (IFN-R-JAK)2*-SOCS1

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in four reactions (as a reactant in reaction_107 and as a product in reaction_98 and as a modifier in reaction_98, reaction_107).

$$\frac{d}{dt} \text{species}_{-100} = |v_{84}| - |v_{94}| \tag{369}$$

9.85 Species species_101

Name PP1-(STAT1C*)2

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in four reactions (as a reactant in re143 and as a product in re142 and as a modifier in re142, re143).

$$\frac{d}{dt} \text{species}_{-101} = |v_{87}| - |v_{88}| \tag{370}$$

9.86 Species species_104

Name (IFN-R-JAK)2*-STAT1C-SOCS1

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in four reactions (as a reactant in reaction_110 and as a product in reaction_107 and as a modifier in reaction_107, reaction_110).

$$\frac{d}{dt} \text{species}_{-104} = |v_{94}| - |v_{95}| \tag{371}$$

9.87 Species species_105

Name (IFN-R-JAK)2*-STAT1C-SHP2-SOCS1

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in six reactions (as a reactant in reaction_108, reaction_109 and as a product in reaction_110 and as a modifier in reaction_110, reaction_108, reaction_109).

$$\frac{d}{dt} \text{species}_{105} = |v_{95}| - |v_{96}| - |v_{97}| \tag{372}$$

9.88 Species species_106

Name (IFN-R-JAK)2*-STAT1C-SHP2

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in one reaction (as a product in reaction_109).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{species}_{106} = v_{97} \tag{373}$$

9.89 Species species_107

Name R

Initial concentration $12 \text{ nmol} \cdot l^{-1}$

This species takes part in two reactions (as a reactant in re135 and as a modifier in re135).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{species}_{-107} = -v_{71} \tag{374}$$

9.90 Species species_108

Name SHP2

Initial concentration 100 nmol·l⁻¹

This species takes part in ten reactions (as a reactant in re138, reaction_110, reaction_86 and as a product in re137, re136, reaction_87, reaction_108 and as a modifier in re138, reaction_110, reaction_86).

$$\frac{d}{dt} \text{species} \cdot 108 = v_6 + v_{40} + v_{72} + v_{96} - v_5 - v_{95} - v_{103}$$
 (375)

9.91 Species s137

Name STAT3*

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

Involved in rule s137

One rule determines the species' quantity.

9.92 Species s136

Name STAT1*

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

Involved in rule s136

One rule determines the species' quantity.

9.93 Species s138

Name (IL6-gp80-gp130-JAK)2*-STAT1

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in four reactions (as a reactant in re158 and as a product in re157 and as a modifier in re157, re158).

$$\frac{\mathrm{d}}{\mathrm{d}t}s138 = |v_{110}| - |v_{111}| \tag{376}$$

9.94 Species s139

Name (IL6-gp80-gp130-JAK)2*-STAT1*

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in two reactions (as a product in re159 and as a modifier in re159).

$$\frac{d}{dt}s139 = v_{112} \tag{377}$$

9.95 Species s140

Name JAK(IL-6)

Initial concentration $12 \text{ nmol} \cdot l^{-1}$

This species takes part in two reactions (as a reactant in re160 and as a modifier in re160).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{s}140 = -v_1\tag{378}$$

9.96 Species species_68

Name (IFN-R-JAK)2*-STAT1C-STAT3C

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in eight reactions (as a reactant in reaction_11, reaction_41 and as a product in reaction_2, reaction_10 and as a modifier in reaction_2, reaction_10, reaction_11, reaction_41).

$$\frac{d}{dt} \text{species_68} = |v_{120}| + |v_{121}| - |v_{122}| - |v_{123}|$$
(379)

9.97 Species species_92

Name (STAT1N*)2

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in nine reactions (as a reactant in re147 and as a product in reaction_91, reaction_103 and as a modifier in reaction_95, reaction_95, reaction_103, re147, re116, re116).

$$\frac{d}{dt} \text{species}_{92} = |v_{76}| + |v_{89}| - |v_{90}| \tag{380}$$

9.98 Species species_94

Name PP2-STAT1N*

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in four reactions (as a reactant in re145 and as a product in re144 and as a modifier in re144, re145).

$$\frac{d}{dt} \text{species}_{.94} = |v_{77}| - |v_{78}| \tag{381}$$

9.99 Species species_95

Name STAT1N*

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in nine reactions (as a reactant in re144, reaction_103, reaction_106, re121 and as a product in re164 and as a modifier in re144, reaction_103, reaction_106, re121).

$$\frac{d}{dt} \text{species_95} = |v_{119} - v_{77}| - 2 v_{89} - |v_{92}| - |v_{115}|$$
(382)

9.100 Species species_96

Name STAT1N

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in six reactions (as a reactant in reaction_94, reaction_106 and as a product in re145, re163 and as a modifier in reaction_94, reaction_106).

$$\frac{d}{dt} \text{species_96} = |v_{78}| + |v_{118}| - |v_{79}| - |v_{92}|$$
(383)

9.101 Species species_97

Name mRNA-SOCS1N

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in three reactions (as a reactant in reaction_96 and as a product in reaction_95 and as a modifier in reaction_96).

$$\frac{d}{dt} \text{species}_{.97} = |v_{80}| - |v_{81}| \tag{384}$$

9.102 Species species_102

Name PP2-(STAT1N*)2

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in four reactions (as a reactant in re148 and as a product in re147 and as a modifier in re147, re148).

$$\frac{d}{dt} \text{species}_{-102} = |v_{90}| - |v_{91}| \tag{385}$$

9.103 Species species_103

Name STAT1N-STAT1N*

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in three reactions (as a product in re148, reaction_106 and as a modifier in reaction_106).

$$\frac{d}{dt} \text{species}_{-103} = |v_{91}| + |v_{92}| \tag{386}$$

9.104 Species species_23

Name (STAT3N*)2

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in seven reactions (as a reactant in reaction_65, reaction_18 and as a product in reaction_17 and as a modifier in reaction_65, reaction_18, reaction_23, reaction_23).

$$\frac{d}{dt} \text{species} 23 = |v_{12}| - |v_{13}| - |v_{16}|$$
(387)

9.105 Species species_24

Name PP2

Initial concentration 60 nmol·l⁻¹

This species takes part in 16 reactions (as a reactant in reaction_19, reaction_18, re144, re147, re126 and as a product in reaction_20, reaction_21, re145, re148, re163, re164 and as a modifier in reaction_19, reaction_18, re144, re147, re126).

$$\frac{d}{dt} species_2 24 = v_{15} + v_{17} + v_{78} + v_{91} + v_{118} + v_{119} - v_{14} - v_{16} - v_{77} - v_{90} - v_{107}$$
(388)

9.106 Species species_25

Name PP2-(STAT3N*)2

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in four reactions (as a reactant in reaction_21 and as a product in reaction_18 and as a modifier in reaction_18, reaction_21).

$$\frac{d}{dt} \text{species.} 25 = |v_{16}| - |v_{17}| \tag{389}$$

9.107 Species species_26

Name STAT3N*

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in nine reactions (as a reactant in reaction_19, re121 and as a product in reaction_65, reaction_66, re163 and as a modifier in reaction_65, reaction_19, reaction_66, re121).

$$\frac{d}{dt} \text{species}_{26} = 2 v_{13} + v_{19} + v_{118} - v_{14} - v_{115}$$
(390)

9.108 Species species_27

Name PP2-STAT3N*

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in four reactions (as a reactant in reaction_20 and as a product in reaction_19 and as a modifier in reaction_19, reaction_20).

$$\frac{d}{dt}$$
 species_27 = $v_{14} - v_{15}$ (391)

9.109 Species species_28

Name STAT3N

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in six reactions (as a reactant in reaction_22 and as a product in reaction_20, reaction_66, re164 and as a modifier in reaction_66, reaction_22).

$$\frac{d}{dt} \text{species}_2 = |v_{15}| + |v_{19}| + |v_{119}| - |v_{20}| \tag{392}$$

9.110 Species species_29

Name STAT3N-STAT3N*

Initial concentration $0 \text{ nmol} \cdot 1^{-1}$

This species takes part in three reactions (as a reactant in reaction_66 and as a product in reaction_21 and as a modifier in reaction_66).

$$\frac{d}{dt} \text{species}_{29} = |v_{17}| - |v_{19}| \tag{393}$$

9.111 Species species_30

Name mRNA-SOCS3N

Initial concentration $0 \text{ nmol} \cdot l^{-1}$

This species takes part in four reactions (as a reactant in reaction_24 and as a product in reaction_23, re116 and as a modifier in reaction_24).

$$\frac{d}{dt} \text{species}_{30} = v_{21} + v_{113} - v_{22} \tag{394}$$

SML2ATEX was developed by Andreas Dräger^a, Hannes Planatscher^a, Dieudonné M Wouamba^a, Adrian Schröder^a, Michael Hucka^b, Lukas Endler^c, Martin Golebiewski^d and Andreas Zell^a. Please see http://www.ra.cs.uni-tuebingen.de/software/SBML2LaTeX for more information.

^aCenter for Bioinformatics Tübingen (ZBIT), Germany

^bCalifornia Institute of Technology, Beckman Institute BNMC, Pasadena, United States

^cEuropean Bioinformatics Institute, Wellcome Trust Genome Campus, Hinxton, United Kingdom

^dEML Research gGmbH, Heidelberg, Germany