# ChemTools

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### Table of contents

- Introduction
  - ChemTools
  - Features
- Chemical reactors
  - Concepts
  - Physical and mathematical model
- 3 Description
  - Project description
  - Input
  - Output
  - Structure
- Demo
  - Closed reactor demo



### What is ChemTools?

- ChemTools is a collection of thermochemical tools.
- Available on GitHub under the MIT license
- Language: C++
- Build system: CMake
- Catch2 Unit tests

- Closed homogeneous chemical reactors with one or two properties held constant:
  - pressure,
  - volume,
  - temperature and pressure, or
  - temperature and volume.
- Closed homogeneous chemical reactors with user-defined Python scripts providing functions for:
  - the volume and its time derivative as a function of time, or
  - the temperature and pressure as a function of time.
- Flamelet library generator base:
  - the  $\beta$ -PDF approach, or
  - the Presumed Mapping Function approach.



## What is a closed homogeneous reactor?

- An enclosed volume in which a chemical reaction takes place, i.e. a chemical process vessel
- No inlets or outlets
- An initial gas mixture exists at an given initial state:
  - Species composition
  - Pressure
  - Temperature
  - Volume
- The gas mixture reacts from this initial state over a specified period of time
- Purpose: used to understand chemical processes through reaction kinetics

# Governing equations of closed homogeneous reactors

- Zero dimensional reactors: transient and space-independent
- Mass and energy conservation in a mixture containing K species:

IVP ODE system 
$$\begin{cases} \rho \frac{dY_k}{dt} = W_k \dot{\omega}_k, & k = 1, 2, ..., K \\ \rho \frac{dT}{dt} = \sum_{k=1}^K h_k W_k \dot{\omega}_k \end{cases}$$

 $Y_k$  = mass fraction of species k (dependent variables)

*T* = temperature (dependent variables)

t = time (independent variable)

$$\dot{\omega}_k = \dot{\omega}_k(Y_1, Y_2, \dots, Y_K, T) = \text{net chemical production rate (non-linear)}$$

 $\rho = density$ 

 $W_k$  = molecular weight of species k

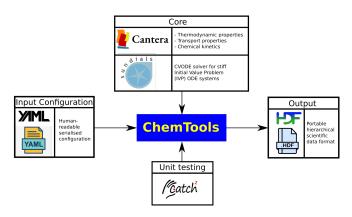
 $h_k$  = specific enthalpy of species k

 $c_p$  = specific heat at constant pressure

Fast and slow time scales lead to a stiff ODE system



## Description and dependencies



#### Other libraries:

- LLNL units for the handling of physical units
- fmt for formatting
- Python C API
- Various Boost libraries: filesystem, optional, algorithm, etc.
- spdlog for logging (coming soon)
- Eigen for numerical linear algebra

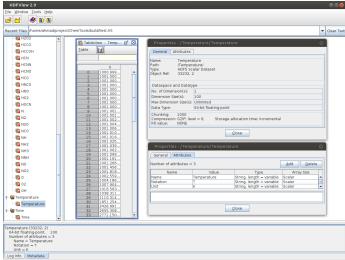


# YAML input configuration



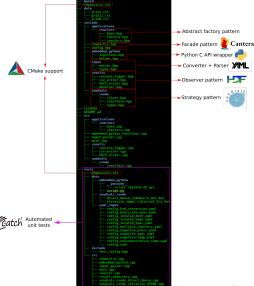
```
CONSTANT PRESSURE
- Reactor:
  Mechanism:
    Description: Constant pressure reactor 1
                  ../data/gri30.cti
    Path:
 Description:
                 case set 1
 Cases:
   Case:
     Description: useful case 1.1
     Pressure:
       Unit: Pa
       Value: 101325.0
     Temperature:
       Unit: Kelvin
       Value: 1300.0
     Composition:
       Type: mole fraction
       Value: [[CH4, 0.1], [02, 0.5], [N2, 0.4]]
     Time:
        Unit: ms
       Value: 1.0
    Case:
     Description: useful case 1.2
     Pressure:
       Unit: atm
       Value: 2.0
     Temperature:
       Unit: degC
       Value: 25.0
     Composition:
       Type: concentration
       Unit: kmol/cm^3
       Value: [[CH4, 0.1], [02, 0.5], [N2, 0.4]]
     Time:
        Unit: min
        Value: 1.0
```

# HDF5 output





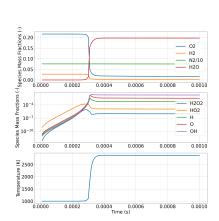
# Project structure and general implementation details



### Constant volume reactor: hydrogen oxidation

- Hydrogen oxidation at constant volume
- Global reaction:  $2H_2 + O_2 \rightarrow 2H_2O + \text{ Heat (not realistic)}$
- Many more intermediate species: OH, O, H, H<sub>2</sub>O<sub>2</sub>, HO<sub>2</sub>, etc.

```
- Reactor: CONSTANT VOLUME
  Mechanism:
    Description: GRI-Mech 3.0
   Path: ../data/gri30.cti
  Description: case set 1
  Cases.
      Description: Combustion of hydrogen
      Pressure:
        Unit: atm
        Value: 1.0
      Temperature:
        Unit: Kelvin
Value: 1001.0
      Composition:
        Type: mole fraction
        Value: [[CH4.2.0], [02.1.0], [N2.4.0]]
        Unit: ms
        Value: 1.0
```



# Thank you