# BPL\_IEC\_operation script with FMPy

The key library FMPy is installed.

After the installation a small application BPL\_IEC\_operation is loaded and run. You can continue with this example if you like.

!lsb\_release -a # Actual VM Ubuntu version used by Google No LSB modules are available. Distributor ID: Ubuntu Description: Ubuntu 22.04.3 LTS Release: 22.04 Codename: jammy %env PYTH0NPATH= → env: PYTH0NPATH= !wget https://repo.anaconda.com/miniconda/Miniconda3-py312\_24.3.0-0-Linux-x86\_64. !chmod +x Miniconda3-py312\_24.3.0-0-Linux-x86\_64.sh !bash ./Miniconda3-py312 24.3.0-0-Linux-x86 64.sh -b -f -p /usr/local import sys sys.path.append('/usr/local/lib/python3.12/site-packages/') → --2024-05-21 06:05:54-- https://repo.anaconda.com/miniconda/Miniconda3-py312 Resolving repo.anaconda.com (repo.anaconda.com)... 104.16.191.158, 104.16.32. Connecting to repo.anaconda.com (repo.anaconda.com)|104.16.191.158|:443... com HTTP request sent, awaiting response... 200 OK Length: 143351488 (137M) [application/octet-stream] Saving to: 'Miniconda3-py312\_24.3.0-0-Linux-x86\_64.sh' Miniconda3-py312\_24 100%[===========] 136.71M in 0.7s 195MB/s 2024-05-21 06:05:54 (195 MB/s) - 'Miniconda3-py312\_24.3.0-0-Linux-x86\_64.sh' : PREFIX=/usr/local Unpacking payload ... Installing base environment... Preparing transaction: ...working... done Executing transaction: ...working... done installation finished. !conda update -n base -c defaults conda --yes → Channels: - defaults Platform: linux-64

Collecting package metadata (repodata.json): done

Solving environment: done

## Package Plan ##

environment location: /usr/local

added / updated specs:

- conda

The following packages will be downloaded:

package	build	
conda-24.5.0 frozendict-2.4.2 openssl-3.0.13	py312h06a4308_0 py312h06a4308_0 py312h06a4308_0 h7f8727e_1	1.2 MB 36 KB 5.2 MB
	 Total:	6.5 MB

The following NEW packages will be INSTALLED:

frozendict pkgs/main/linux-64::frozendict-2.4.2-py312h06a4308\_0

The following packages will be UPDATED:

conda 24.3.0-py312h06a4308\_0 --> 24.5.0-py312h0 3.0.13-h7f8727e\_0 --> 3.0.13-h7f872 openssl

Downloading and Extracting Packages:

openssl-3.0.13 | 5.2 MB 0% 0/1 [00:00<?, ?it/s] conda-24.5.0 | 1.2 MB | : 0% 0/1 [00:00<?, ?it/s]

frozendict-2.4.2 | 36 KB | : 0% 0/1 [00:00<?, ?it/s]

frozendict-2.4.2 | 36 KB 1 : 44% 0.43853215920344746/1 [00:00<00:00,</pre> openssl-3.0.13 | 5.2 MB 0% 0.002997347135570501/1 [00:00<00:58 | :

| 5.2 MB | : 88% 0.8842174049932978/1 [00:00<00:00,</pre> openssl-3.0.13 | : 100% 1.0/1 [00:00<00:00, 1.73it/s] conda-24.5.0 | 1.2 MB

Preparing transaction: done Verifying transaction: done Executing transaction: done

!conda --version !python --version

conda 24.5.0 Python 3.12.2

!conda install -c conda-forge fmpy --yes # Install the key package



Preparing transaction: done Verifying transaction: done Executing transaction: done #!conda install matplotlib --yes

Now specific installation and the run simulations. Start with connecting to Github. Then upload the two files:

- FMU BPL\_IEC\_operation\_linux\_om\_me.fmu
- Setup-file BPL\_IEC\_operation\_fmpy\_explore.py

```
%bash
git clone https://github.com/janpeter19/BPL_IEC_operation

Cloning into 'BPL_IEC_operation'...

%cd BPL_IEC_operation

/content/BPL_IEC_operation
```

### BPL\_IEC\_operation

process\_diagram()

Authors: Karl Johan Brink and Jan Peter Axelsson

In this notebook we show operation of a typical ion-exchange chromatography step. The impact of pH is also illustrated.

The model is based on the simplified model [1].

```
run -i BPL_IEC_fmpy_explore.py
Linux - run FMU pre-comiled OpenModelica 1.23.0-dev
    Model for bioreactor has been setup. Key commands:
     - par()

    change of parameters and initial values

     - init()

    change initial values only

     - simu()

    simulate and plot

     - newplot() - make a new plot

    show plot from previous simulation

     - show()

    display parameters and initial values from the last simulation

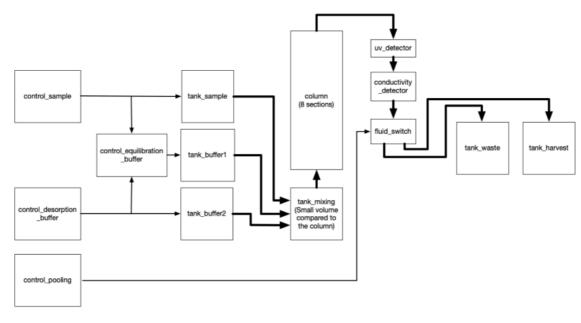
     - disp()

    describe() - describe culture, broth, parameters, variables with values/ur

    Note that both disp() and describe() takes values from the last simulation
    and the command process_diagram() brings up the main configuration
    Brief information about a command by help(), eg help(simu)
    Key system information is listed with the command system_info()
plt.rcParams['figure.figsize'] = [30/2.54, 24/2.54]
```

# The process diagram is made outside Modelica to illustrate the configuration

No processDiagram.png file in the FMU, but try the file on disk.



# 1 Typical parameters an ion exchange chromatography column step

```
# From given column height (h) diameter (d) and linear flow rate (lfr)
# actual column volume (V) and volume flow rate (VFR) are calculated below.

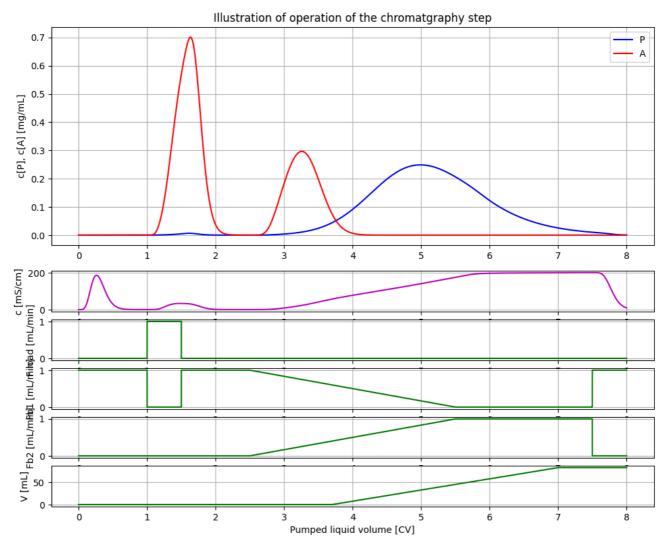
from numpy import pi
h = 20.0
d = 1.261
a = pi*(d/2)**2
V = h*a
print('V =', np.round(V,1), '[mL]')

lfr = 48
VFR = a*lfr/60
print('VFR =', np.round(VFR,1), '[mL/min]')

V = 25.0 [mL]
VFR = 1.0 [mL/min]
```

```
# Sample concentration product P_in and antagonist A_in
par(P_in = 1.0)
par(A in = 1.0)
par(E_in = 0.0)
# Column properties are described by the size and binding capacity of the resin Q
par(height = h)
par(diameter = d)
par(Q av = 6.0)
# Resin parameters - default values used
# Remaining salt koncentration in the column from prvious batch and eliminated du
init(E_start = 50)
# Salt concentration of the desorption buffer
par(E_in_desorption_buffer = 8.0)
# Flow rate rate through the
par(LFR=lfr)
# Switching points during operation are conveniently described in terms of multip
CV ekv = 1.0
CV_ads = 0.5
CV_wash = 1.0
CV_desorb = 3.0
CV_start_pool = 1.2
CV_stop_pool = 4.5
CV wash2 = 2.5
par(scale_volume=True, start_adsorption=CV_ekv*V, stop_adsorption=(CV_ekv+CV_ads)
par(start_desorption=(CV_ekv+CV_ads+CV_wash)*V, stationary_desorption=(CV_ekv+CV_
par(stop_desorption=7.5*V)
par(start_pooling=(CV_ekv+CV_ads+CV_wash+CV_start_pool)*V, stop_pooling=(CV_ekv+C
# Simulation and plot of results
newplot(title='Illustration of operation of the chromatgraphy step', plotType='El
```





Comments of steps of operations: 1) Time: 0-1 hours - equilibration. Just to illustrate the equilibration process the first part of the column is given an initial value of salt concentration. 2) Time: 1-1.5 hours - sample is loaded on the column. The product P is adsorbed to the columne and just a small amount passes through and goes to the waste. The antagonist A is much less adsrobed. 3) Time: 1.5-2.5 hours - washing 1. The column comes to equilibrium and both antagonist and product comes down to low levels. 4) Time: 2.5-5.5 hours - desorption. A linear gradient of increaseing salt concentration is applied. First the antagonist and later the product

comes out. 5) Time: 5.5-7.5 hours - washing 2 The The column has constant salt concentration and stationary desorption. 6) Time: 3.7-7.0 hours - pooling of product. The start- and stop of pooling are chosen with trade-off between maximizing the product pooled and minimize the amount of antagonist in the pooling. 7) Time: 7.5-8.0 hours - desorption stopped and salt is washed out and preparation of the next batch to come.

Note that step 4 and 5 is parallel to step 6.

```
# Check mass-balance of P and A
P_mass = model_get('tank_harvest.m[1]') + model_get('tank_waste.m[1]')
A_mass = model_get('tank_harvest.m[2]') + model_get('tank_waste.m[2]')
print('P_mass [mg] =', P_mass)
print('A_mass [mg] =', A_mass)
P_mass [mg] = 12.422130344957237
A_mass [mg] = 12.488781164503445
```

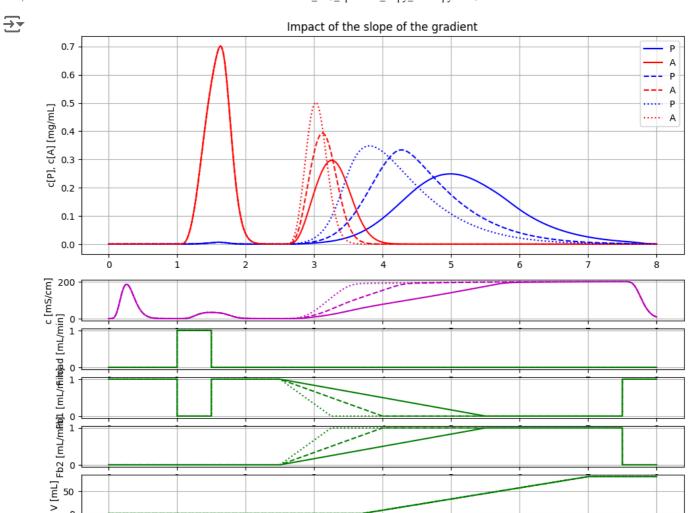
#### 2 The impact of the slope of the desorption gradient

```
# Simulations showing the impact of change of slope of the desorption gradient
newplot(title='Impact of the slope of the gradient', plotType='Elution-conductivi

# Same gradienet as before
par(start_desorption=(CV_ekv+CV_ads+CV_wash)*V, stationary_desorption=(CV_ekv+CV_
par(stop_desorption=7.5*V)
simu((CV_ekv+CV_ads+CV_wash+CV_desorb+CV_wash2)*V/VFR)

# Gradeint finishes after 0.5 of the volume
par(stationary_desorption = (CV_ekv + CV_ads + CV_wash + 0.5*CV_desorb)*V )
simu((CV_ekv+CV_ads+CV_wash+CV_desorb+CV_wash2)*V/VFR)

# Fradient finishes after 0.25 of the volume
par(stationary_desorption = (CV_ekv + CV_ads + CV_wash + 0.25*CV_desorb)*V )
simu((CV_ekv+CV_ads+CV_wash+CV_desorb+CV_wash2)*V/VFR)
```



Pumped liquid volume [CV]

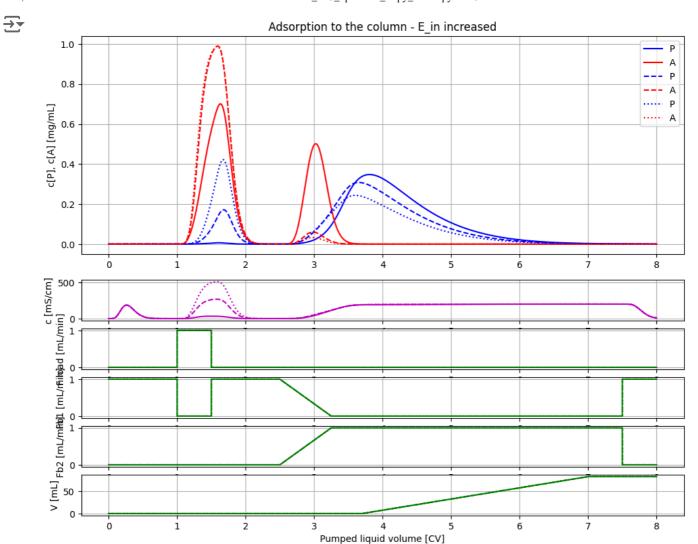
# 3 The impact of salt concentration in the sample

These values should be compared with the expected value 12.5 mg, i.e. half a column volume with sample concentration 1 mg/L. The difference is due to numerical errors during simulation.

```
# Let us investigate the impact of increasing salt concetration in the sample E_i
# Simulate and plot the results
newplot(title='Adsorption to the column - E_in increased', plotType='Elution-cond

for value in [0, 10, 20]:
    par(E_in=value)
    simu((CV_ekv+CV_ads+CV_wash+CV_desorb+CV_wash2)*V/VFR)

# Restore default values
par(k2=0.05, k4=0.3, E_in=0)
```



Note, that increased salt concentration in the sample affect binding of both proteins. During adsorption less is bound. During desoprtion less product P can be harvested but the fraction of antagonist A may be lowered. Thus, some product is lost but the quality in terms of purity is improved.

# 4 The impact of change of binding strength due to pH

There are many factors that contribute to the binding strength. A most important factor is the pH-value of the resin and the characteristic iso-electric point of the protein. The binding strength can be seen as proportional to the difference.

The binding strength of the resin is described by the quotient KP=k1/k2 for the protein P and similarly KA=k3/k4 for the protein A.

Below a few help-functions that describe this idea of the pH difference and its impact on binding strength in terms of the parameters k1, k2, k3, and k4 of the protein-resin interaction.

```
# Define function that describe the proportionality of binding strength ot
# the pH difference of the iso-electric point and the resin
def KP_pH_sensitivity(pI_P=8.0, pH_resin=7.0):
   K_P_nom = 0.0
    coeff pH = 6.0
    return K_P_nom + coeff_pH*(pI_P-pH_resin)
def KA pH sensitivity(pI A=7.1667, pH resin=7.0):
   K A nom = 0.0
    coeff_pH = 1.0
    return K_A_nom + coeff_pH*(pI_A-pH_resin)
def par_pH(pI_P=8.0, pI_A=7.1667, pH_resin=7.0, TP=3.33, TA=20.0):
    if (pI P > pH resin) \& (pI A > pH resin):
        par(k2 = 1/(TP*KP_pH_sensitivity(pI_P=pI_P, pH_resin=pH_resin)))
        par(k4 = 1/(TA*KA_pH_sensitivity(pI_A=pI_A, pH_resin=pH_resin)))
    else:
        print('Both pI P > pH resin and pI A > pH resin must hold - no parameter
# The default parameters of the column
disp('column')
→ diameter : 1.261
    height: 20.0
    x_m : 0.3
    k1: 0.3
    k2: 0.05
    k3: 0.05
    k4: 0.3
    Q_av : 6.0
    E_start: 50
```

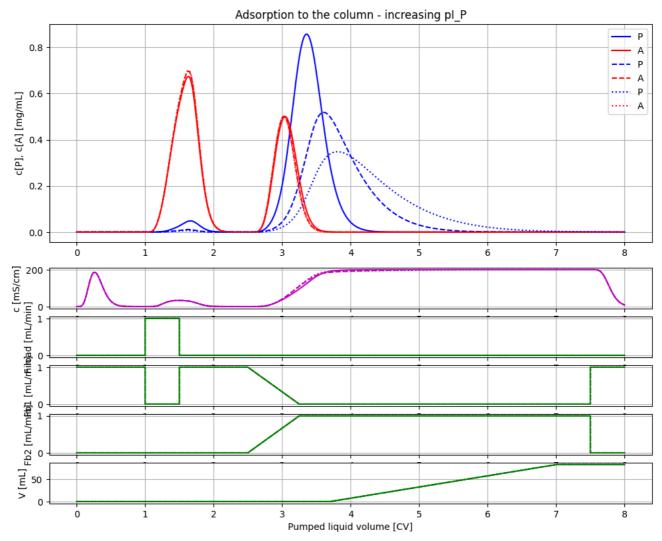
```
# Let us investigate the impact of change of the iso-electric pH for protein P

# Simulate and plot the results
newplot(title='Adsorption to the column - increasing pI_P', plotType='Elution-con

for value in [7.2, 7.6, 8.0]:
    par_pH(pI_P=value, pI_A=7.1667, pH_resin=7.0)
    simu((CV_ekv+CV_ads+CV_wash+CV_desorb+CV_wash2)*V/VFR)

# Restore default values
par(k2 = 0.05, k4 = 0.3)
```





Note, with increasing pI\_P the binding of P increase which leads less loss of product during adsorption. During desorption the peak height is lower with increasing binding strenght, but the total amount of product P that can be harvested is higher, due to the smaller loss during adsorption.

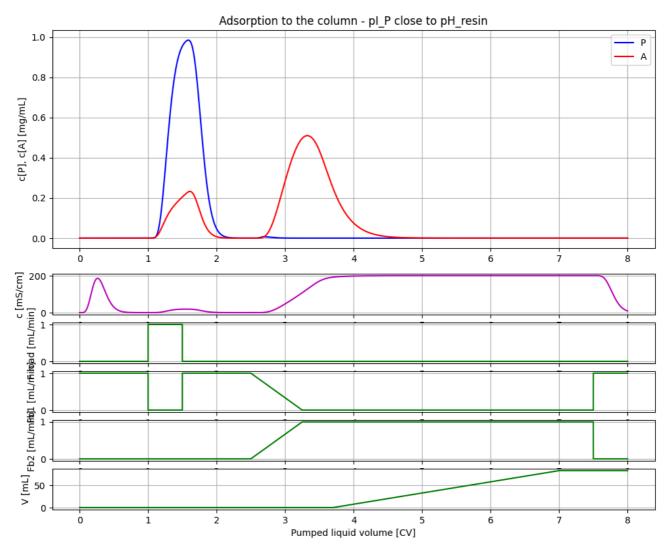
```
# Let us investigate the impact of pI_P close to pH_resin

# Simulate and plot the results
newplot(title='Adsorption to the column - pI_P close to pH_resin', plotType='Elut

for value in [7.0001]:
    par_pH(pI_P=value, pI_A=8)
    simu((CV_ekv+CV_ads+CV_wash+CV_desorb+CV_wash2)*V/VFR)

# Restore default values
par(k2=0.05, k4=0.3)
```





```
# Let us investigate the impact of pI_A close to pH_resin

# Simulate and plot the results
newplot(title='Adsorption to the column - pI_A close to pH_resin', plotType='Elut

for value in [7.001]:
    par_pH(pI_P=8.0, pI_A=value)
    simu((CV_ekv+CV_ads+CV_wash+CV_desorb+CV_wash2)*V/VFR)

# Restore default values
par(k2=0.05, k4=0.3)
```

# Let us also investigate the impact of salt concentration of the desorptions buf