BPL_IEC_operation script with PyFMI

The key library PyFMI is installed.

After the installation a small application BPL_IEC_operation is loaded and run. You can continue with this example if you like.

```
!lsb_release -a # Actual VM Ubuntu version used by Google
    No LSB modules are available.
    Distributor ID: Ubuntu
    Description:
                   Ubuntu 22.04.3 LTS
    Release:
                    22.04
    Codename:
                    jammy
%env PYTHONPATH=
    env: PYTHONPATH=
!wget https://repo.anaconda.com/miniconda/Miniconda3-py310_23.1.0-1-Linux-x86_64.
!chmod +x Miniconda3-py310_23.1.0-1-Linux-x86_64.sh
!bash ./Miniconda3-py310 23.1.0-1-Linux-x86 64.sh -b -f -p /usr/local
import sys
sys.path.append('/usr/local/lib/python3.10/site-packages/')
    --2024-03-09 10:41:46-- https://repo.anaconda.com/miniconda/Miniconda3-py310
    Resolving repo.anaconda.com (repo.anaconda.com)... 104.16.131.3, 104.16.130.3
    Connecting to repo.anaconda.com (repo.anaconda.com)|104.16.131.3|:443... connecting
    HTTP request sent, awaiting response... 200 OK
    Length: 74403966 (71M) [application/x-sh]
    Saving to: 'Miniconda3-py310_23.1.0-1-Linux-x86_64.sh'
    172MB/s
                                                                      in 0.4s
    2024-03-09 10:41:46 (172 MB/s) - 'Miniconda3-py310_23.1.0-1-Linux-x86_64.sh' :
    PREFIX=/usr/local
    Unpacking payload ...
    Installing base environment...
    Downloading and Extracting Packages
    Downloading and Extracting Packages
    Preparing transaction: done
    Executing transaction: done
    installation finished.
```

!conda update -n base -c defaults conda --yes

Preparing transaction: done Verifying transaction: done Executing transaction: done

```
!conda --version
!python --version
```

conda 24.1.2 Python 3.10.13

!conda install -c conda-forge pyfmi --yes # Install the key package

```
Preparing transaction: done
Verifying transaction: done
Executing transaction: done
```

Preparation of BPL_IEC_operation

Now specific installation and the run simulations. Start with connecting to Github. Then upload the two files:

- FMU BPL_IEC_Column_system_linux_om_me
- Setup-file BPL_IEC_explore

```
%%bash
git clone https://github.com/janpeter19/BPL_IEC_operation
    Cloning into 'BPL_IEC_operation'...
%cd BPL_IEC_operation
    /content/BPL_IEC_operation
```

BPL_IEC_operation

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In this notebook we show operation of a typical ion-exchange chromatography step. The impact of pH is also illustrated.

The model is based on the simplified model [1].

Note that both disp() and describe() takes values from the last simulation

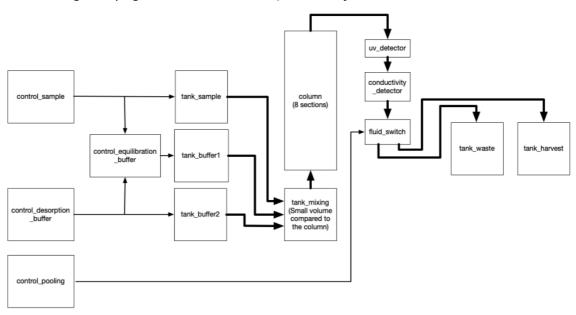
and the command process_diagram() brings up the main configuration

Brief information about a command by help(), eg help(simu)
Key system information is listed with the command system_info()

```
%matplotlib inline
plt.rcParams['figure.figsize'] = [36/2.54, 30/2.54]
```

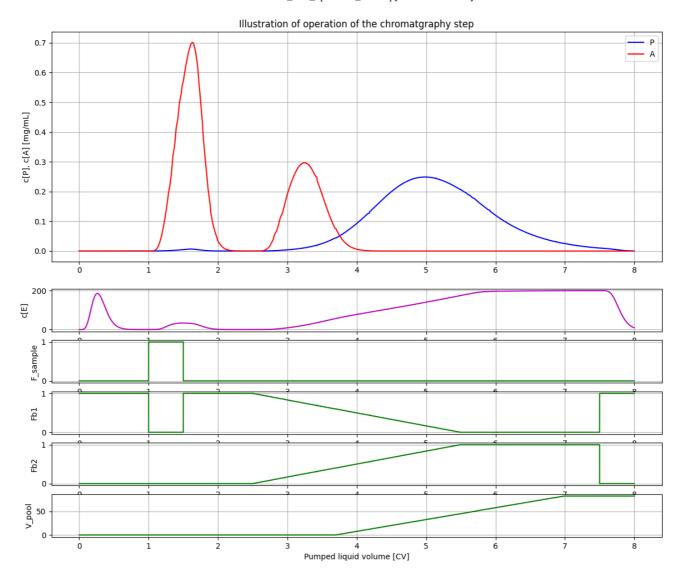
The process diagram is made outside Modelica for illustration of the configurat
process_diagram()

No processDiagram.png file in the FMU, but try the file on disk.



1 Typical parameters for a pilot scale ion exchange chromatography column process setup

```
# From given colunn height (h) diameter (d) and linear flow rate (lfr)
# actual column volume (V) and volume flow rate (VFR) are calculated below.
from numpy import pi
h = 20.0
d = 1.261
a = pi*(d/2)**2
V = h*a
print('V =', np.round(V,1), '[mL]')
lfr = 48
VFR = a*lfr/60
print('VFR =', np.round(VFR,1), '[mL/min]')
                                                                               # P
    V = 25.0 [mL]
    VFR = 1.0 [mL/min]
# Sample concentration product P_in and antagonist A_in
par(P in = 1.0)
par(A_in = 1.0)
par(E_in = 0.0)
# Column properties are described by the size and binding capacity of the resin Q
par(height = h)
par(diameter = d)
par(Q av = 6.0)
# Remaining salt koncentration in the column from prvious batch and eliminated du
init(E start = 50)
# Salt koncentration of the desorption buffer
par(E_in_desorption_buffer = 8.0)
# Flow rate rate through the
par(LFR=lfr)
# Switching points during operation are conveniently described in terms of multip
CV_{ekv} = 1.0
CV ads = 0.5
CV_wash = 1.0
CV_desorb = 3.0
CV_start_pool = 1.2
CV_stop_pool = 4.5
CV_{ekv2} = 2.5
par(scale_volume=True, start_adsorption=CV_ekv*V, stop_adsorption=(CV_ekv+CV_ads)
par(start_desorption=(CV_ekv+CV_ads+CV_wash)*V, stationary_desorption=(CV_ekv+CV_
par(stop_desorption=7.5*V)
par(start_pooling=(CV_ekv+CV_ads+CV_wash+CV_start_pool)*V, stop_pooling=(CV_ekv+C
# Simulation and plot of results
newplot(title='Illustration of operation of the chromatgraphy step', plotType='El
simu((CV_ekv+CV_ads+CV_wash+CV_desorb+CV_ekv2)*V/VFR)
```



Comments of steps of operations:

- 1) Time: 0-1 hours equilibration. Just to illustrate the equilibration process the first part of the column is given an initial value of salt concentration.
- 2) Time: 1-1.5 hours sample is loaded on the column. The product P is adsorbed to the columne and just a small amount passes through and goes to the waste. The antagonist A is much less adsrobed.
- 3) Time: 1.5-2.5 hours washing 1. The column comes to equilibrium and both antagonist and product comes down to low levels.
- 4) Time: 2.5-5.5 hours desorption. A linear gradient of increaseing salt concentration is applied. First the antagonist and later the product comes out.
- 5) Time: 5.5-7.5 hours washing 2 The The column has constant salt concentration and stationary desorption.
- 6) Time: 3.7-7.0 hours pooling of product. The start- and stop of pooling are chosen with tradeoff between maximizing the product pooled and minimize the amount of antagonist in the pooling.
- 7) Time: 7.5-8.0 hours desorption stopped and salt is washed out and preparation of the next batch to come.

Note that step 4 and 5 is parallel to step 6.

```
# Check mass-balance of P and A
P_mass = model.get('tank_harvest.m[1]') + model.get('tank_waste.m[1]')
A_mass = model.get('tank_harvest.m[2]') + model.get('tank_waste.m[2]')
print('P_mass [mg] =', P_mass)
print('A_mass [mg] = ', A_mass)

P_mass [mg] = [12.42212162]
A_mass [mg] = [12.48878113]
```

These values should be compared with the expected value 12.5 mg, i.e. half a column volume with sample concentration 1 mg/L. The difference is due to numerical errors during simulation.

2 The impact of the slope of the desorption gradient

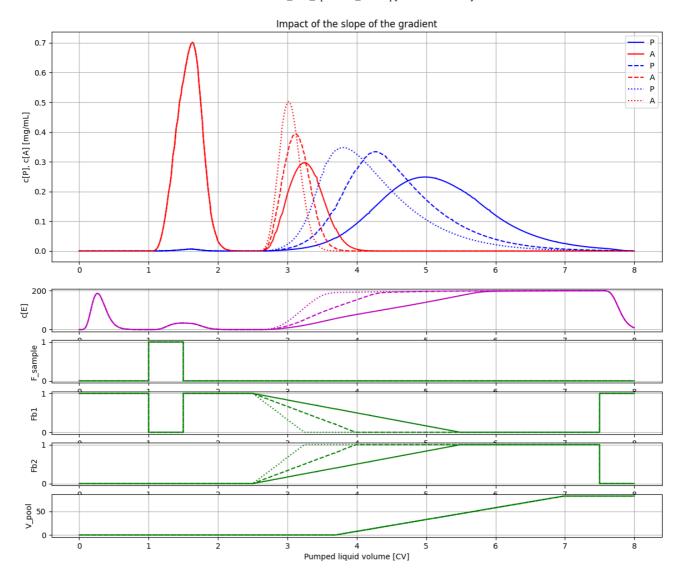
```
# Simulations showing the impact of change of slope of the desorption gradient
newplot(title='Impact of the slope of the gradient', plotType='Elution-conductivi

# Same gradienet as before
par(start_desorption=(CV_ekv+CV_ads+CV_wash)*V, stationary_desorption=(CV_ekv+CV_par(stop_desorption=7.5*V)
simu((CV_ekv+CV_ads+CV_wash+CV_desorb+CV_ekv2)*V/VFR)

# Gradeint finishes after 0.5 of the volume
par(stationary_desorption = (CV_ekv + CV_ads + CV_wash + 0.5*CV_desorb)*V )
simu((CV_ekv+CV_ads+CV_wash+CV_desorb+CV_ekv2)*V/VFR)

# Fradient finishes after 0.25 of the volume
par(stationary_desorption = (CV_ekv + CV_ads + CV_wash + 0.25*CV_desorb)*V )
```

simu((CV_ekv+CV_ads+CV_wash+CV_desorb+CV_ekv2)*V/VFR)



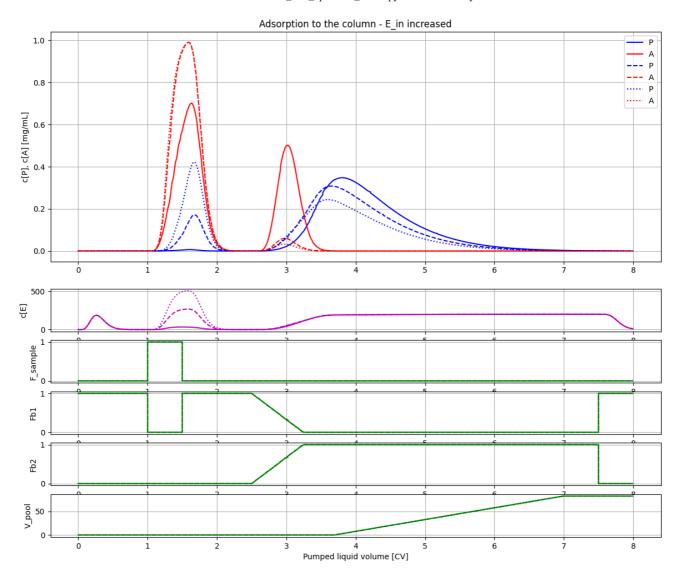
Note the pens shift style for each simulation in the order: solid, dashed, dotted, dash-dotted. The actual simulations done you see in the preceding cell.

3 The impact of salt concentration in the sample

```
# Let us investigate the impact of increasing salt concetration in the sample E_i
# Simulate and plot the results
newplot(title='Adsorption to the column - E_in increased', plotType='Elution-cond

for value in [0, 10, 20]:
    par(E_in=value)
    simu((CV_ekv+CV_ads+CV_wash+CV_desorb+CV_ekv2)*V/VFR)

# Restore default values
par(k2=0.05, k4=0.3, E_in=0)
```



Note, that increased salt concentration in the sample affect binding of both proteins. During adsorption less is bound. During desoprtion less product P can be harvested but the fraction of antagonist A may be lowered. Thus, some product is lost but the quality in terms of purity is improved.

4 The impact of change of binding strength due to pH

There are many factors that contribute to the binding strength. A most important factor is the pH-value of the resin and the characteristic iso-electric point of the protein. The binding strength can be seen as proportional to the difference.

The binding strength of the resin is described by the quotient KP=k1/k2 for the protein P and similarly KA=k3/k4 for the protein A.

Below a few help-functions that describe this idea of the pH difference and its impact on binding strength in terms of the parameters k1, k2, k3, and k4 of the protein-resin interaction.

```
# Define function that describe the proportionality of binding strength ot
# the pH difference of the iso-electric point and the resin
def KP_pH_sensitivity(pI_P=8.0, pH_resin=7.0):
   coeff pH = 6.0
    return coeff pH*(pI P-pH resin)
def KA pH sensitivity(pI A=7.1667, pH resin=7.0):
    coeff_pH = 1.0
    return coeff_pH*(pI_A-pH_resin)
def par_pH(pI_P=8.0, pI_A=7.1667, pH_resin=7.0, TP=3.33, TA=20.0):
    if (pI_P > pH_resin) \& (pI_A > pH_resin):
        par(k2 = 1/(TP*KP_pH_sensitivity(pI_P=pI_P, pH_resin=pH_resin)))
        par(k4 = 1/(TA*KA_pH_sensitivity(pI_A=pI_A, pH_resin=pH_resin)))
   else:
        print('Both pI_P > pH_resin and pI_A > pH_resin must hold - no parameter
# The default parameters of the column
disp('column')
    diameter: 1.261
    height : 20.0
    x_m : 0.3
    k1: 0.3
    k2: 0.05
    k3: 0.05
    k4: 0.3
```

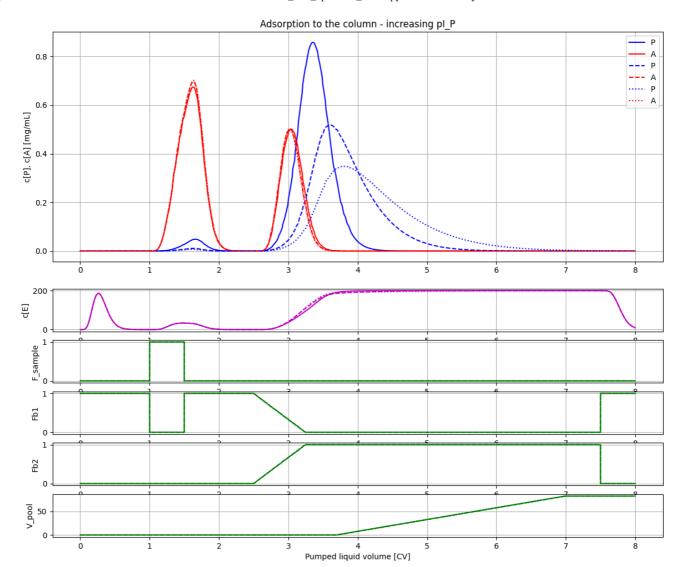
Q_av : 6.0
E_start : 50.0

```
# Let us investigate the impact of change of the iso-electric pH for protein P

# Simulate and plot the results
newplot(title='Adsorption to the column - increasing pI_P', plotType='Elution-con

for value in [7.2, 7.6, 8.0]:
    par_pH(pI_P=value, pI_A=7.1667, pH_resin=7.0)
    simu((CV_ekv+CV_ads+CV_wash+CV_desorb+CV_ekv2)*V/VFR)

# Restore default values
par(k2 = 0.05, k4 = 0.3)
```



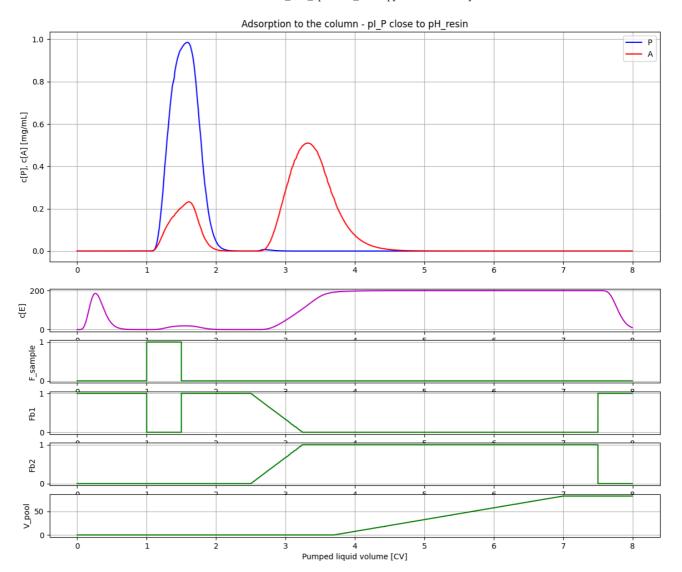
Note, with increasing pI_P the binding of P increase which leads less loss of product during adsorption. During desorption the peak height is lower with increasing binding strenght, but the total amount of product P that can be harvested is higher, due to the smaller loss during adsorption.

```
# Let us investigate the impact of pI_P close to pH_resin

# Simulate and plot the results
newplot(title='Adsorption to the column - pI_P close to pH_resin', plotType='Elut

for value in [7.0001]:
    par_pH(pI_P=value, pI_A=8)
    simu((CV_ekv+CV_ads+CV_wash+CV_desorb+CV_ekv2)*V/VFR)

# Restore default values
par(k2=0.05, k4=0.3)
```

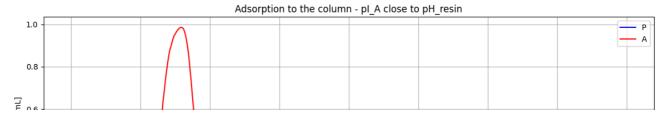


```
# Let us investigate the impact of pI_A close to pH_resin

# Simulate and plot the results
newplot(title='Adsorption to the column - pI_A close to pH_resin', plotType='Elut

for value in [7.001]:
    par_pH(pI_P=8.0, pI_A=value)
    simu((CV_ekv+CV_ads+CV_wash+CV_desorb+CV_ekv2)*V/VFR)

# Restore default values
par(k2=0.05, k4=0.3)
```



Let us also investigate the impact of salt concentration of the desorptions buf

Simulate and plot the results
newplot(title='Adsorption to the column - desorption buffer salt conc varied', pl

```
for value in [8.0, 16.0]:
    par(E_in_desorption_buffer=value)
    par_pH(pI_P=8.0, pI_A=7.001, pH_resin=7.0)
    simu((CV_ekv+CV_ads+CV_wash+CV_desorb+CV_ekv2)*V/VFR)
```

Restore default values
par(E_in_desorption_buffer=8.0)
par(k2=0.05, k4=0.3)

