



DEC 28, 2022

# Measurement of dissolved carbohydrate

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## OPEN ACCESS

**DOI:**  
[dx.doi.org/10.17504/protocols.io.bp2l6168zvqe/v1](https://dx.doi.org/10.17504/protocols.io.bp2l6168zvqe/v1)

**Protocol Citation:** Ying-Yu Hu, Zoe V. Finkel 2022. Measurement of dissolved carbohydrate. **protocols.io** <https://dx.doi.org/10.17504/protocols.io.bp2l6168zvqe/v1>

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**Protocol status:** Working  
We use this protocol and it's working

**Created:** Jul 11, 2022

**Last Modified:** Dec 28, 2022

**PROTOCOL integer ID:**  
66472

**Keywords:** Dissolved total carbohydrate, Dissolved polysaccharide, Dissolved monosaccharide, TPTZ method, Ferricyanide, hydrolysis

## ABSTRACT

Here we describe a protocol to measure the dissolved carbohydrate, including total dissolved monosaccharides and total dissolved polysaccharides.

For total dissolved carbohydrate measurement, freeze-dried dissolved carbohydrate samples are initially vortexed in 9 M H<sub>2</sub>SO<sub>4</sub> for 15 s. The solution is diluted for a final H<sub>2</sub>SO<sub>4</sub> molarity of 1.6 M and hydrolyzed for 3 hours at 90 °C. The hydrolysate is alkalized by adding 12 M NaOH to the hydrolysate, the ratio of [H<sup>+</sup>] from hydrolysate to [OH<sup>-</sup>] from NaOH is 0.82. The alkalized hydrolysate is oxidized by ferricyanide solution. The absorbance of TPTZ-Fe<sup>2+</sup> complex is measured in microtiter plate at 595 nm.

For total dissolved monosaccharide measurement, freeze-dried dissolved carbohydrate samples are alkalized by 12 M NaOH and then oxidized by ferricyanide solution. The absorbance of TPTZ-Fe<sup>2+</sup> complex is measured in microtiter plate at 595 nm.

Our method has shown high reproducibility in aldohexoses, ketohexoses, deoxysugars, aldopentoses, uronic acid and amino sugars. The low limit of detection is 0.024 µg C/mL.

## SAFETY WARNINGS



Ferric waste should be disposed into trace metal waste container.  
Waste acid should be neutralized before disposed into sink.

## Sample collection

1 GFF filter is combusted for 04:00:00 at 450 °C

12h

12h

Glass filter holder is combusted for 02:00:00 at 500 °C  
Glass filter funnel, flask and 10 mL centrifuge tubes are combusted for 06:00:00 at 500 °C

Equipment	
Disposable Glass Screw-Cap Centrifuge Tubes	NAME
10 mL	TYPE
Corning®	BRAND
99502-10	SKU

Tube caps are acid-washed.

Equipment	
Polypropylene Screw Caps	NAME
Linerless, 15-415	TYPE
Kimble Chase	BRAND
73805-15415	SKU

- 2
- Filter microalgae sample and collect the filtrate, using gentle vacuum pressure (130 mm Hg).
- 3
- Transfer 5 mL filtrate into centrifuge tube and flash freeze.

Note
Three tubes for total dissolved monosaccharide and three tubes for total dissolved carbohydrate measurement.

- 4
- Freeze dry samples before measurement.


## Glucose standards


### 5 Primary standard solution

5.1 In a 2 mL microtube, weigh 1 ~ 2 mg D-glucose  D-glucose **Sigma Aldrich Catalog #G8270-100G**

5.2 Add Milli-Q for a final concentration of 1 mg/mL (>600 µL).

### 6 Secondary standard for total dissolved carbohydrate

6.1 Add  45 µL primary solution into a 2 mL microtube

6.2 Add  955 µL Milli-Q and then vortex for a good mix

6.3 In 10 mL centrifuge tubes, prepare the following standard solutions:

SD	Secondary solution (uL)	Milli-Q (uL)
TCHO-SD1	0	100
TCHO-SD2	20	80
TCHO-SD3	40	60
TCHO-SD4	60	40
TCHO-SD5	80	20
TCHO-SD6	100	0

7 Secondary standard for total dissolved monosaccharide

7.1 Add 10 µL primary solution into a 2 mL microtube

7.2 Add 990 µL Milli-Q and then vortex for a good mix


7.3 In 12 mL amber vials, prepare the following standard solutions:



SD	Secondary solution (uL)	Milli-Q (uL)	12 M NaOH (uL)
MCHO-SD1	0	984	16
MCHO-SD2	10	974	16
MCHO-SD3	20	964	16
MCHO-SD4	50	934	16
MCHO-SD5	100	884	16
MCHO-SD6	150	834	16

Equipment	
Storage Vials and Closures	NAME
12 mL amber	TYPE
Thermo Scientific	BRAND
B7800-12A	SKU
VWR 66030-686	SPECIFICATIONS

Hydrolysis of total dissolved carbohydrate

8 Prepare water bath 95 °C



9 Add  100  $\mu\text{L}$  Milli-Q to each tube with freeze-dried sample.

10 Use reverse pipetting technique, add  100  $\mu\text{L}$  18 M  $\text{H}_2\text{SO}_4$  to standard solution/sample, immediately vortex for  00:00:15 (monitored by timer or stopwatch)

15s

Note

Do not cap the centrifuge tube!

11 Add  900  $\mu\text{L}$  Milli-Q, tightly cap the centrifuge tube, and vortex for  00:00:05 .

5s

12 Place tube into water bath, log the time.

Note


Hydrolysis duration for each sample/blank/standard should be accurately monitored.

13 After all samples are in the water bath, reduce temperature to  90  $^{\circ}\text{C}$  .

14 Label amber vials for TPTZ measurement with white oil based sharpie.

# of vials = # of samples + # of blanks + # of standards



15 As soon as hydrolysis duration reaches 3 hours, remove the tube from water bath, let it sit in the tap water bath with ice to quickly stop hydrolysis.

16 Keep all hydrolysate in a dark cabinet at  Room temperature .

## Prepare TPTZ reagents



17 12 M NaOH

17.1 Add 15 mL Milli-Q water into a 50 mL Falcon tube.

17.2 Add  12 g NaOH pellet into the water, swirl and have the pellets completely dissolved, let it cool down to  Room temperature .

17.3 Transfer the solution into a 25 mL PP volumetric flask, rinse the tube three times by small amount of Milli-Q and combine the rinsed water into flask, top with Milli-Q water to 25 mL.




18 Alkaline solution for potassium ferricyanide

Dissolve  400 mg NaOH and  20 g  $\text{Na}_2\text{CO}_3$  in volumetric flask and top to 1 L by Milli-Q. Store at room temperature.

 NaOH Fisher Scientific Catalog #BP359-500

  $\text{Na}_2\text{CO}_3$  VWR international Ltd Catalog #97061-972

19 Sodium acetate solution

19.1 Dissolve  164 g sodium acetate,  42 g citric acid and  300 g acetic acid in a 1 L volumetric flask and top to 1 L with Milli-Q water.

### Note

In this solution, sodium acetate, citric acid and acetic acid is 2 M, 0.2 M and 5 M respectively.


☒ Sodium acetate anhydrous **Fisher Scientific Catalog #BP333-500**

☒ Citric acid **Sigma Aldrich Catalog # 251275-500G**

☒ Acetic acid **Fisher Scientific Catalog #M1000632500**

19.2 Store at room temperature.

19.3 Dispense solution by serological pipet to avoid having salt precipitated around sealing surface of the bottle.



20 3 M acetic acid  
Weigh  180 g acetic acid in fumehood, transfer the acid into volumetric flask, top to 1 L with Milli-Q water.  
Store at room temperature.

40m

## TPTZ method

21 Prepare boiling bath

22 TPTZ reagents

22.1 Potassium ferricyanide (Reagent A)  
Weigh  23 mg potassium ferricyanide and transfer into a 100 mL amber reagent bottle. Add  
 100 mL alkaline solution, vortex until powder is completely dissolved. It is stable for two weeks at room temperature.

☒  $\text{K}_3[\text{Fe}(\text{CN})_6]$  **Fisher Scientific Catalog #AC424120050**

Equipment	
<b>Reagent bottle</b>	NAME
100 mL, amber	TYPE
VWR	BRAND
14216-240	SKU

## 22.2

### Ferric chloride (Reagent B)

Ferric chloride hexahydrate is in spherical shape. It is hard to weigh exact 54 mg for a 100 mL solution. Pick a very small ferric chloride ball and log the weight. Transfer the ball into a 100 mL amber reagent bottle.

Calculate the acetate solution required.

Add acetate solution into the amber bottle, vortex until the ball is completely dissolved.

$$V_{\text{acetate}} = 100 \times W_{\text{actual}} / 54$$

#### Note

This reagent needs to be prepared right prior to analysis. It can only be stable for no more than two days.

## 22.3

### TPTZ (Reagent C)

Estimate the total volume required for the assay: 2 mL X (standard # + blank # + sample #)

For each 100 mL TPTZ reagent, weigh and transfer 78 mg TPTZ into an amber reagent bottle, add 100 mL acetic acid solution, vortex until the powder is completely dissolved.

 **TPTZ Sigma Aldrich Catalog #T253-5G**

#### Note

This solution is stored at room temperature and stable for one week.

## 23


### Total dissolved carbohydrate samples




23.1 Use reverse pipetting technique, transfer  750 µL hydrolysate of standard/sample to amber vial.

23.2 Add  250 µL 12 M NaOH and vortex.

24 Total dissolved monosaccharide samples


24.1 Add  1200 µL Milli-Q into the tube with freeze-dried sample

24.2 Use reverse pipetting technique, transfer  984 µL solution to amber vial.

24.3 Add  16 µL 12 M NaOH and vortex.

25 In a room with dim light, add  1 mL Reagent A into each amber vial.



26 Tightly cap the vial and vortex.

27 Keep in a boiling water bath for  00:10:00

10m

28 Remove boiling bath from the heat, keep all vials in the hot water and move them into the room with dim light.

29 Add  1 mL Reagent B and  2 mL Reagent C into the vial and vortex.

30 Shake at  Room temperature for  00:30:00 .

30m

31 Under dim light, using reverse pipetting, load 250 uL of blanks, standards, and samples into the microplate (duplicate).  
Load column by column. After one column has been loaded, immediately cover the column with a lid, which has a black membrane on the top to protect sample from light.

	1	2	3	4	5	6	7	8	9	10	11	12
A	MCHO-SD1	MCHO-SD1										
B	MCHO-SD2	MCHO-SD2										
C	MCHO-SD3	MCHO-SD3										
D	MCHO-SD4	MCHO-SD4										
E	MCHO-SD5	MCHO-SD5										
F	MCHO-SD6	MCHO-SD6										
G												
H												

Microplate layout for dissolved monosaccharide samples

	1	2	3	4	5	6	7	8	9	10	11	12
A	TCHO-SD1	TCHO-SD1										
B	TCHO-SD2	TCHO-SD2										
C	TCHO-SD3	TCHO-SD3										
D	TCHO-SD4	TCHO-SD4										
E	TCHO-SD5	TCHO-SD5										
F	TCHO-SD6	TCHO-SD6										
G												

	1	2	3	4	5	6	7	8	9	10	11	12
H												

Microplate layout for total dissolved carbohydrate samples

32 Read in microplate reader:

Shake for 5 s at 600 rpm in a continuous and high force mode

Read endpoint 595 nm with a measurement time 100 ms

## UV/VIS spectra (optional)

33 Hydrolysate

33.1 Load  200 µL hydrolysate into microplate.

33.2 Blank:  
Milli-Q : H<sub>2</sub>SO<sub>4</sub> = 10:1

34 Monosaccharide solutions

34.1 Load  200 µL solution into microplate.

34.2 Blank: Milli-Q

35 Scan UV/VIS spectra from 200 to 400 nm at a step of 1 nm.

## Calculation

### 36 Total dissolved carbohydrate

36.1 Subtract the average absorbance of blank (0 ug glucose) from the absorbance of each standard for total dissolved carbohydrate.

36.2 Obtain standard curve by plotting blank subtracted absorbance (Abs') versus carbon (uM C)

$$Abs' = a * C_{(uM)} + b$$

36.3 Subtract the average absorbance of blank (0 ug glucose) from the absorbance of each sample

36.4

$$C_{(uM)} = (Abs' - b) / a$$
$$TCHO_{(uMC)} = C * (1.1/5) / 0.75$$

### 37 Total dissolved monosaccharide

37.1 Subtract the average absorbance of blank (0 ug glucose) from the absorbance of each standard for total dissolved monosaccharide.

37.2 Obtain standard curve by plotting blank subtracted absorbance (Abs') versus carbon (uM C)

$$Abs' = a * C_{(uM)} + b$$

37.3 Subtract the average absorbance of blank (0 ug glucose) from the absorbance of each sample

37.4

$$C_{(uM)} = (Abs' - b)/a$$

$$MCHO_{(uMC)} = C * (1.2/5)/0.984$$

38 Total dissolved polysaccharide

$$PCHO_{(uMC)} = TCHO_{(uMC)} - MCHO_{(uMC)}$$

## Waste disposal

39 All hydrolysate and TPTZ reagent C need to be neutralized by soda before disposed into the sink.

40 TPTZ reagent B is collected in trace metal waste container.