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# Unified pH measurement

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1 Works for me Share

dx.doi.org/10.17504/protocols.io.bybxpspn

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ABSTRACT

The purpose of this document is to present technical guidance of measuring  $pH_{abs}^{H20}$  difference ( $\Delta pH_{abs}^{H20}$  value) between two solutions by differential potentiometry in a cell with two compartments connected by a salt bridge filled with an ionic liquid.

DOI

dx.doi.org/10.17504/protocols.io.bybxpspn

**EXTERNAL LINK** 

https://analytical.chem.ut.ee/, uniphied.eu

PROTOCOL CITATION

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KEYWORDS

unified acidity, pHabs, unified pH, pH

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MATERIALS TEXT

The ionic liquid triethylpentylammonium bis(trifluoromethylsulfonyl)imide [N2225][NTf2] salt bridge has to be used.

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Turn on the thermostat and electrometer at least half an hour before starting the work.

## Software

Start Quick IV Measurement Software.

Computer cannot go to sleep during measurements or the communication between computer and instruments is lost and data collection stops.

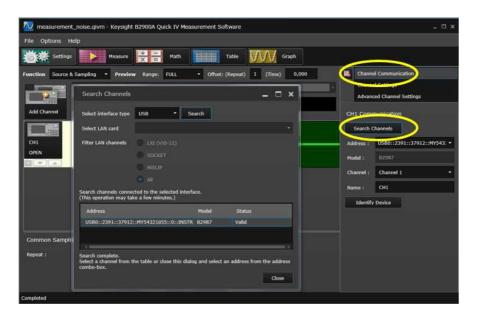
2 On the left hand pick Function "Source & Sampling".



Position of the Function choice "Source & Sampling".

Alternatively, open previously saved QIVM file with measurement settings.

Right hand side click on tab "Channel Communication". Click on "Search Channels", which opens a new window. Select USB interface and search for the channel. After the search click on the channel name to choose the channel and close the window.



Position of the tabs "Channel Communication" and "Search Channels".

This must be done every time the program is started.

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4 Settings can be saved as QIVM file, which can be used for measurements. The saved file can be opened under tab "File". The settings will remain the same between the measurements if the program is not closed.



Program settings.

Filling the cell 8m

5

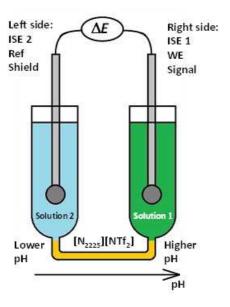


₿ 25.0 °C

Cell from Gebr. Rettberg GmbH.

	VistaShield Faraday cage
	Gamry VistaShield
	UT cell Special water-jacketed glass cell Rettberg Glass cell
	U2c Thermostat MLW U2c
	The side ports must be open.
6	Fill capillary with a syringe or pipette with the ionic liquid, so that the level of ionic liquid is 1 mm to 2 mm below the $^{2m}$ bottoms of the measurement compartments. $\blacksquare$ <b>0.1 mL</b>
	Ionic liquid has high viscosity, therefore flows very slowly, and can leave a film on the wall of the pipette tip or syringe during dispense.
7	The test solutions are added simultaneously to both compartments, raising solution levels in the compartments synchronously. <b>20 mL both test solutions</b>
	Equal amount by weight (not volume!) of solutions must be used. The volume of the lighter solution is calculated from the weight of the heavier solution using the densities of the two solutions in hand. This is especially important for solutions containing acetonitrile.
	Minimum volume that can be used is 13 ml.
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One option how to fill the cell. This configuration gives positive  $\Delta pH$  values. Figure shows connections to a pH meter, a potentiostat and an electrometer.

2m

- After adding the test solutions, the air bubbles are removed from the capillary ends with a syringe or a pipette.
- 9 Insert electrodes one-by-one.

EST-0601 Glass electrode half-cell Izmeritelnaya tekhnika EST-0601

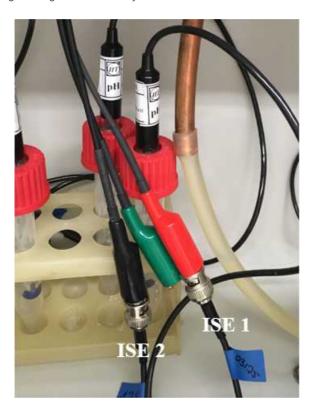
- 9.1 Electrode is rinsed with water, gently dried, and then rinsed with solution to be measured.
- 10 Connect electrodes to the instrument.

B2987A Electrometer / High Resistance
Meter
Electrometer
Keysight B2987A

10.1 Crocodile clips are as following: red is signal and connects to ISE1,

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black is shield and is connected to ISE2 and green is guard. Green stays unused.



Connections to electrodes.

Cable goes through the port in Faraday cage. Electrometer is not in the Faraday cage.

N1415A Triax to Alligator Cable Cable

Keysight Technologies N1415A 👄

#### Measurement

1h

- 11 Start the measurement by clicking on the "Measure" button on the upper panel.
- 12 Data collection. Point is taken at 10 s interval.

1h

3 Save data. Files are named as Date\_ISE1\_Solution1\_vs\_ISE2\_solution2.

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13.1 Go to table view, right click on the table, choose "Export as CSV..." and save the results. The data can be automatically imported into excel file while measuring or after (both under "Excel Navigation"), but this causes constant shift in the comma position.

### Washing

14 Remove the electrodes, rinse them and put them in storage solution.

1m

15 Remove the solutions.

3m

16 Rinse the cell with small amount of ethanol **35 mL** 

2m

17 Rinse three times with plenty of water **20 mL each** 

10m

18 Rinse once with acetone to ease the drying. 

5 mL

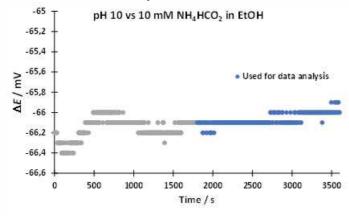
2m

19 Dry cell with compressed air.

2m

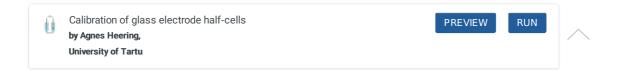
# Data analysis 20m

Points from 30 min to 60 min are used for analysis. Points from 30 min to 60 min are used for analysis. Points from 30 min to 60 min are used for analysis.



Example of a good graph with non-aqueous system.

- 21 Take the average of the chosen data points.
- 22  $\Delta pH = \Delta E / \text{ average slope of the glass electrodes}$



22.1 Start Quick IV Measurement Software.

Computer cannot go to sleep during measurements or the communication between computer and instruments is lost and data collection stops.

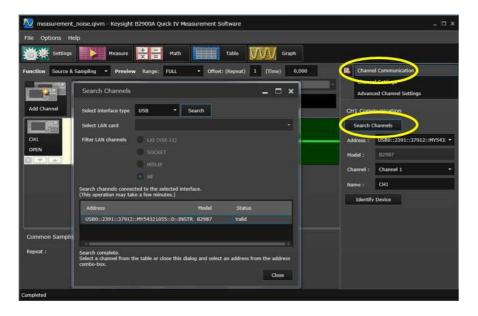
22.2 On the left hand pick Function "Source & Sampling".



Position of the Function choice "Source & Sampling".

Alternatively, open previously saved QIVM file with measurement settings.

22.3 Right hand side click on tab "Channel Communication". Click on "Search Channels", which opens a new window. Select USB interface and search for the channel. After the search, click on the channel name to choose the channel and close the window.



Position of the tabs "Channel Communication" and "Search Channels".

This must be done every time the program is started.

22.4 Settings can be saved as QIVM file, which can be used for measurements. The saved file can be opened under tab "File".

The settings will remain the same between the measurements if the program is not closed.



Program settings.

Range is "Fixed 20V" in case of EST-0601 calibration, otherwise it is "Fixed 2V".

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8 25 °C

Special water-jacketed glass cell from Gebr. Rettberg.

UT cell

Special water-jacketed glass cell

Rettberg Glass cell

VistaShield Faraday cage

Gamry VistaShield

U2c

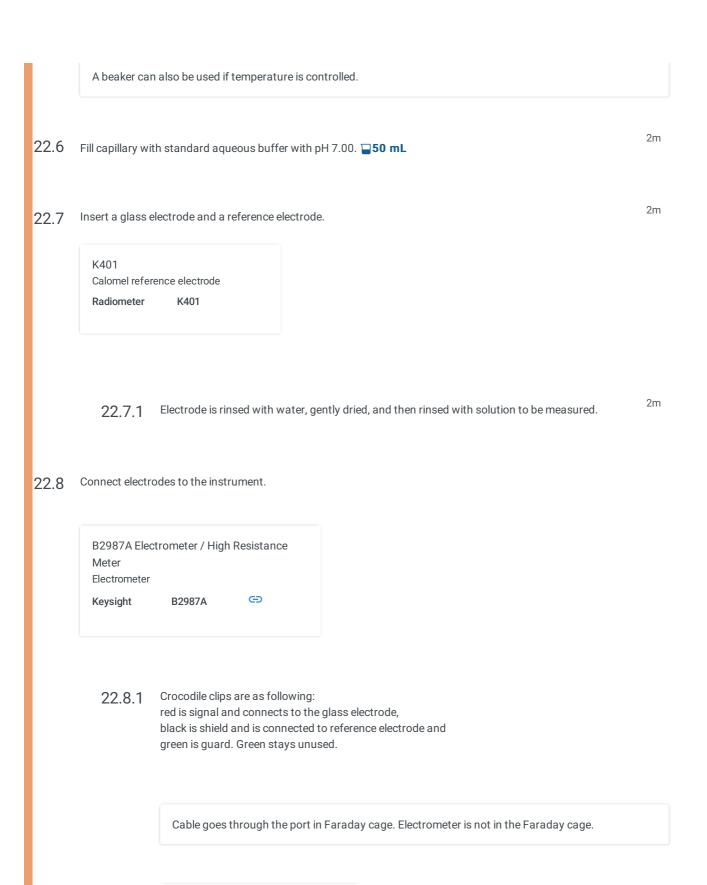
Thermostat

MLW U2c

The side ports must be open.

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N1415A Triax to Alligator Cable

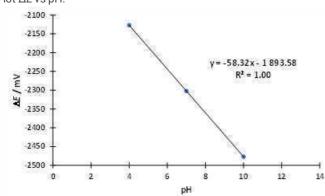
Keysight Technologies N1415A 👄

22.9	Start the measurement by clicking on the "Measure" button on the upper panel.	
2.10	Data collection. Point is taken at 10 s interval.	1h
	Calibration duration must be the same as later used in pH measurements.	
2.11	Save data. Files are named as Date_GE_vs_ref_pHx.	
	SAVE THE FILE BEFORE NEW MEASUREMENT! Otherwise, data is lost. Data is given in volts and seconds.	
	Go to table view, right click on the table, choose "Export as CSV" and save the results. The data cautomatically imported into excel file while measuring or after (both under "Excel Navigation"), but causes constant shift in the comma position.	
2.12	Remove the electrodes.	
2.13	Remove the solution.	
2.14	Rinse three times with plenty of water <b>20 mL each</b>	
2.15	Rinse once with acetone to ease the drying.   5 mL	
2.16	Dry cell with compressed air.	
2.17	Repeat measurements with standard buffers pH 4 and pH 10.	
2.18	Points from 30 min to 60 min are used for analysis.	

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2.19 Take the average of the chosen data points.

2.20 Plot  $\Delta E$  vs pH.



Calibration graph of a glass electrode half-cell.

2.21 Obtain the slope and intercept of the glass electrode-half cell.

23 Insert the  $\Delta pH$  into the ladder.

Symmetry 2020, 12 (7), 1150; https://doi.org/10.3390/sym12071150

Mesurement

24 Repeat with a new pair of solutions.

Data analysis 20m

25 Repeat the data analysis with all the measured potentials.

26 Insert the data into the ladder.

27 Do the minimization process.

Symmetry 2020, 12 (7), 1150; https://doi.org/10.3390/sym12071150

- $28 \quad \text{Obtain the pH values for all the measured solutions from the minimization process.} \\$
- 29 Calculate uncertainty.

Analytica Chimica Acta, **2021**, 1182, 338923, <u>doi.org/10.1016/j.aca.2021.338923</u>