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# Procedure for measuring extraction efficiency

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Works for me

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#### ABSTRACT

Capturing dissolved organic matter from seawater using solid phase extraction is not 100% efficient. We use this protocol to quantify the amount of organic carbon retained during the extraction step. Comparing this value to the concentration of dissolved organic carbon in the initial water sample allows us to quantify the efficiency of the extraction procedure.

#### PROTOCOL CITATION

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# MATERIALS TEXT

Extract from solid phase extraction

40 ml glass vials with Teflon-lined septa

Concentrated hydrochloric acid

pipet and pipet tips

Vacufuge

 $TOC\ analyzer\ (our\ lab\ uses\ a\ Shimadzu\ TOC-V_{CSH}\ total\ organic\ carbon\ analyzer\ with\ a\ TNM-1 module\ for\ total\ nitrogen)$ 

## Procedure for measuring extraction efficiency

- This protocol starts with the eluent from a solid phase extraction resin. Dry down eluents in 38 mL vials (all the way down, no solvent remaining!).
- While drying down prepare **□40 mL** EPA vials with **□25 mL** MQ water and **□25 μl □12 M** HCl.

3

Once the **B** mL vials have dried down fully add **1** mL of MQ water and vortex.

4

Add  $\Box 400~\mu l$  of this vortexed solution to your prepped  $\Box 40~mL$  EPA vials. The final volume should be  $\Box 25.425~mL$ .

5

Load the  $\square 40 \text{ mL}$  vial on to the autosampler for the TOC analyzer ( $\square 150 \mu I$  is injected onto the column).

6 Once these are done, dry down the remaining solution in the **B** mL vials for long term storage.

# **Extraction efficiency calculation:**

Note that this calculation assumes that you have dried down the entire extract for this analysis and that no sample has been removed from the extract for other analyses.

Example for 466  $\mu$ M NPOC in DOC (*not* TOC) sample and 240 ml total volume of extract: (240 ml) (466  $\mu$ M) = (1 ml) (x) x = 111,840  $\mu$ M in extract if 100% efficient

Actual measured carbon in extract example (for sample with 46  $\mu$ M measured carbon): (400  $\mu$ l) (x) = (46  $\mu$ M) (25 ml) (1000  $\mu$ l / 1 ml) x = 2875  $\mu$ M

% efficiency = 2875  $\mu$ M / 111,840  $\mu$ M = < 3%