

Mar 26,  
2020

# 10X TBS Buffer

Bryon Drown<sup>1</sup><sup>1</sup>Northwestern University**1** Works for me dx.doi.org/10.17504/protocols.io.bd85i9y6Kelleher Research Group  
Tech. support email: [kelleher-ofc@northwestern.edu](mailto:kelleher-ofc@northwestern.edu)Bryon Drown  
Northwestern University

## ABSTRACT

Tris buffered saline (TBS) is a commonly used buffer with a wide range of applications in biochemistry. It is often more convenient to prepare a 10X solution so that additional components can be added when preparing the buffer on demand.

## MATERIALS

NAME	CATALOG #	VENDOR
Sodium Chloride	S271	Fisher Scientific
Tris Base	BP152	Fisher Scientific
Tris Hydrochloride	BP153	Fisher Scientific
Hydrochloric Acid	A144S	Fisher Scientific

## SAFETY WARNINGS

Concentrated hydrochloric acid is corrosive by all exposure routes. Always work with concentrated HCl in a fume hood while wearing appropriate PPE.

- 1 Weigh out Tris base **3.13 g**
- 2 Weigh out Tris HCl **27.45 g**
- 3 Weigh out sodium chloride **88 g**
- 4 Dissolve in MilliQ water **900 ml**
- 5 Measure and adjust to **pH7.5** with concentrated HCl and/or NaOH
- 6 Add MilliQ water to a final volume of 1 L



This is an open access protocol distributed under the terms of the [Creative Commons Attribution License](https://creativecommons.org/licenses/by/4.0/), which permits unrestricted use, distribution, and reproduction in any medium, provided the original author and source are credited