



Feb 15, 2022

Gallyas-silver stain

Christiana.bjorkli 1, María José Lagartos Donate²

¹The Norwegian University of Science and Technology; ²University of Oslo

1

dx.doi.org/10.17504/protocols.io.b44cqysw

Sandvig lab

Christiana.bjorkli

Simplified Gallyas-silver staining protocol that works for pathological staining in rodent and human tissue

DOI

dx.doi.org/10.17504/protocols.io.b44cqysw

Christiana.bjorkli , María José Lagartos Donate 2022. Gallyas-silver stain.

protocols.io

https://dx.doi.org/10.17504/protocols.io.b44cqysw

_____ protocol,

Feb 15, 2022

Feb 15, 2022

58212

Solutions that can be made in advance and stored in a fume hood

1 Make 4% of **NaOH** (sodium hydroxide) - use gloves, store in fume hood

1.1 1g *NaOH* + 25ml *H20*

2 Make 2 x 600ml dH20 – use gloves, store in fume hood

protocols.io

1

Citation: Christiana.bjorkli , MarÃÂa José Lagartos Donate Gallyas-silver stain https://dx.doi.org/10.17504/protocols.io.b44cqysw

2.1 1.5g *ammonium nitrate* + 1.2g *silver nitrate* + 3.6g of 4%*NaOH*



Ammonium nitrate can cause an explosion with metals Reaction between **silver nitrate** and ethanol is explosive Reaction between **silver nitrate** and **NaOH** can form inflammable gases/vapors

- 3 Make acetic acid use gloves, store in fume hood
 - 3.1 0.5%: 3ml acetic acid in 597ml H20 0.1%: 0.6ml acetic acid in 600ml H20 0.05%: 50ml 0.5% acetic acid + 450ml H20
- 4 Make 0.2% **potassium ferricyanide** can be stored up to a week, use gloves, store in fume hood
 - **4.1** 0.6g **potassium ferricyanide** in 300ml H20
- 5 Make 0.5% **sodium thiosulfate** use gloves, store in fume hood
 - 5.1 1.5g sodium thiosulfate in 300ml H20
- 6 Make 25%, 50%, 70%, 95% and 100% **ethanol**

Solutions that need to be prepared on the same day of staining

7 Make **pyridine + acetic acid** - use gloves, store in fume hood

protocols.io

2

7.1 100ml pyridine + 50ml acetic acid



Pyridine is incompatible with rubber, plastics, and metals! Very important to use butyl-rubber gloves with this chemical

- 8 Make silver nitrate (add in this order) use gloves, store in fume hood
 - 8.1 100ml dH20 0.192g ammonium nitrate 0.2g silver nitrate 0.6ml of 4% NaOH pH must be 7.5



Ammonium nitrate can cause an explosion with metals Reaction between **silver nitrate** and **ethanol** is explosive Reaction between **silver nitrate** and **NaOH** can form inflammable gases/vapors

- 9 Make A + B + C (1L) use gloves, store in fume hood
 - 9.1 Solution A

5g anhydrous sodium carbonate + 100ml dH20

Solution B (add in this order)

100ml dH20 + 0.19g ammonium nitrate + 0.2g silver nitrate + 1g silicongulistic (tungolistic) acid

Solution C (add in this order)

100mldH20 + 0.19g ammonium nitrate + 0.2g silver nitrate + 1g silicongulistic (tungolistic) acid + 0.66ml 37% formaldehyde

A + B + C - use gloves, store in fume hood 50ml A + 37.5ml B + 37.5ml C (25±2°C)



Ammonium nitrate can cause an explosion with metals Reaction between **silver nitrate** and **ethanol** is explosive Reaction between **silver nitrate** and **NaOH** can form inflammable gases/vapors

Formaldehyde may cause cancer

Consumables for staining

10 17 x glass petri dishes
20 x glass pipettes
1 x rubber pump for glass pipettes
Butyl rubber gloves
Lab coat with long arms
Protective eyewear
Face mask

Need to use glass petri dishes and glass pipettes for each wash

Gallyas-silver staining

11

Step	Reagent	Minutes	Comments
1	dH20	3 or more	Put in dH20 while preparing other solutions
2	Pyridine + acetic acid	60	On shaker
3	50% ethanol	3	
4	25% ethanol	3	
5	0.05% acetic acid	2.5	
6	0.1% acetic acid	2.5	
7	0.05% acetic acid	10	Leave longer if necessary, to prepare silver nitrate
8	Silver nitrate	60	On shaker – make A+ B + C
9	0.5% acetic acid	10	



4

10	4 . 5 . 6	c /a = ·	B
10	A + B + C	6 (15min	Do not let the tissue become dark,
		human)	time is dependent on sample
11	0.5% acetic acid	1	New
12	0.2% potassium ferricyanide	5	On shaker
13	dH20	1	
14	0.5% acetic acid	1	New, can be reused
15	0.5% sodium thiosulfate	2	On shaker
16	dH20	4	
17	dH20	4	
18	0.5% acetic acid	1	Use the one from the previous step
19	A + B + C	3-4	On shaker
20	0.5% acetic		Use the one from the previous step
21	0.2% potassium ferricyanide	3	On shaker
22	dH20	1	
23	0.5% sodium	2	On shaker
24	dH20	4	
25	dH20	4	
26	0.5% acetic	1	New, can be reused
27	A + B + C	3-4	On shaker
28	0.5% acetic	1	Use the one from the previous step
29	0.2% potassium ferricyanide	10	On shaker
30	dH20	1	
31	0.5% sodium thiosulfate	2	On shaker
32	dH20	Fast wash	
33	dH20	5	
34	Dehydrate	3 in each	50% - 70% -95% - 100% - 100% ethanol
35	Xylene x 2	3 in each	
36	Coverslip		
	1		· ·