Fundamental Physical Constants — Frequently used constants

	=			Relative std.
Quantity	Symbol	Value	Unit	uncert. $u_{\rm r}$
speed of light in vacuum	c, c_0	299 792 458	${ m m~s^{-1}}$	(exact)
magnetic constant	μ_0	$4\pi \times 10^{-7}$	${\sf N}{\sf A}^{-2}$	(*,
	, -	$= 12.566370614 \times 10^{-7}$	${ m N~A^{-2}}$	(exact)
electric constant $1/\mu_0 c^2$	ϵ_0	$8.854187817\times10^{-12}$	${ m F}{ m m}^{-1}$	(exact)
Newtonian constant				
of gravitation	G	$6.6742(10) \times 10^{-11}$	${ m m}^{3}~{ m kg}^{-1}~{ m s}^{-2}$	1.5×10^{-4}
T	,	0.000.000(11) 10.24		4 - 40 7
Planck constant	h	$6.6260693(11) \times 10^{-34}$	J s	1.7×10^{-7}
$h/2\pi$	\hbar	$1.05457168(18) \times 10^{-34}$	J s	1.7×10^{-7}
elementary charge	e	$1.60217653(14)\times10^{-19}$	C	8.5×10^{-8}
magnetic flux quantum $h/2e$	Φ_0	$2.06783372(18)\times10^{-15}$	Wb	8.5×10^{-8}
conductance quantum $2e^2/h$	G_0	$7.748091733(26) \times 10^{-5}$	S	3.3×10^{-9}
electron mass	$m_{ m e}$	$9.1093826(16) \times 10^{-31}$	kg	1.7×10^{-7}
proton mass	$m_{ m p}$	$1.67262171(29) \times 10^{-27}$	kg	1.7×10^{-7}
proton-electron mass ratio	$m_{ m p}/m_{ m e}$	1836.15267261(85)	8	4.6×10^{-10}
fine-structure constant $e^2/4\pi\epsilon_0\hbar c$	lpha	$7.297352568(24) \times 10^{-3}$		3.3×10^{-9}
inverse fine-structure constant	α^{-1}	137.035 999 11(46)		3.3×10^{-9}
		()		
Rydberg constant $\alpha^2 m_{\rm e} c/2h$	R_{∞}	10973731.568525(73)	m^{-1}	6.6×10^{-12}
Avogadro constant	$N_{ m A}, L$	$6.0221415(10) \times 10^{23}$	mol^{-1}	1.7×10^{-7}
Faraday constant $N_{\rm A}e$	F	96485.3383(83)	$C \text{ mol}^{-1}$	8.6×10^{-8}
molar gas constant	R	8.314472(15)	$\mathrm{J}\ \mathrm{mol^{-1}}\ \mathrm{K^{-1}}$	1.7×10^{-6}
Boltzmann constant R/N_A	k	$1.3806505(24) \times 10^{-23}$	$ m J~K^{-1}$	1.8×10^{-6}
Stefan-Boltzmann constant				
$(\pi^2/60)k^4/\hbar^3c^2$	σ	$5.670400(40) \times 10^{-8}$	$\mathrm{W}~\mathrm{m}^{-2}~\mathrm{K}^{-4}$	7.0×10^{-6}
Non-SI units accepted for use with the SI				
electron volt: (e/C) J	eV	$1.60217653(14)\times10^{-19}$	J	8.5×10^{-8}
(unified) atomic mass unit	٠,	1.00211000(11) // 10	•	0.0 /\ 10
$1 \text{ u} = m_{\text{u}} = \frac{1}{12} m(^{12}\text{C})$	u	$1.66053886(28)\times 10^{-27}$	kg	1.7×10^{-7}
$=10^{-3} \text{ kg mol}^{-1}/N_{\rm A}$				