

Instructions

- 1) This exam requires the Respondus Lockdown Browser with Monitoring; you will not be able to take the exam with any other browser nor if you don't have a webcam.
- 2) You are allowed to use this coversheet, a calculator, paper and pen/pencil during this exam.
- 3) Other references or aids are not allowed. Headphones are not allowed.
- 4) You are responsible to locate an appropriate testing environment where you can work alone while having reliable internet access.
- 5) Your "Environmental Check" must include the entire room and items on the desk.
- 6) You should remain in view of the webcam throughout the entire exam.

PERIODIC TABLE OF THE ELEMENTS

1 IA	2 IIA	3 IIIB	4 IVB	5 VB	6 VIB	7 VIIB	8 -----	9 VIII	10 -----	11 IB	12 IIB	13 IIIA	14 IVA	15 VA	16 VIA	17 VIIA	18 VIII A
1 H 1.008																	2 He 4.003
3 Li 6.941	4 Be 9.012											5 B 10.81	6 C 12.01	7 N 14.01	8 O 16.00	9 F 19.00	10 Ne 20.18
11 Na 22.99	12 Mg 24.31											13 Al 26.98	14 Si 28.09	15 P 30.97	16 S 32.07	17 Cl 35.45	18 Ar 39.95
19 K 39.10	20 Ca 40.08	21 Sc 44.96	22 Ti 47.88	23 V 50.94	24 Cr 52.00	25 Mn 54.94	26 Fe 55.85	27 Co 58.93	28 Ni 58.69	29 Cu 63.55	30 Zn 65.39	31 Ga 69.72	32 Ge 72.61	33 As 74.92	34 Se 78.96	35 Br 79.90	36 Kr 83.80
37 Rb 85.47	38 Sr 87.62	39 Y 88.91	40 Zr 91.22	41 Nb 92.91	42 Mo 95.94	43 Tc (98)	44 Ru 101.1	45 Rh 102.9	46 Pd 106.4	47 Ag 107.9	48 Cd 112.4	49 In 114.8	50 Sn 118.7	51 Sb 121.8	52 Te 127.6	53 I 126.9	54 Xe 131.3
55 Cs 132.9	56 Ba 137.3	57 La 138.9	72 Hf 178.5	73 Ta 180.9	74 W 183.8	75 Re 186.2	76 Os 190.2	77 Ir 192.2	78 Pt 195.1	79 Au 197.0	80 Hg 200.6	81 Tl 204.4	82 Pb 207.2	83 Bi 209.0	84 Po (209)	85 At (210)	86 Rn (222)
87 Fr (223)	88 Ra 226.0	89 Ac 227.0	104 Rf (267)	105 Db (268)	106 Sg (269)	107 Bh (270)	108 Hs (277)	109 Mt (278)	110 Ds (281)	111 Rg (282)	112 Cn (285)	113 Nh (286)	114 Fl (289)	115 Mc (289)	116 Lv (293)	117 Ts (294)	118 Og (294)

58 Ce 140.1	59 Pr 140.9	60 Nd 144.2	61 Pm (145)	62 Sm 150.4	63 Eu 152.0	64 Gd 157.3	65 Tb 158.9	66 Dy 162.5	67 Ho 164.9	68 Er 167.3	69 Tm 168.9	70 Yb 173.0	71 Lu 175.0
90 Th 232.0	91 Pa 231.0	92 U 238.0	93 Np (237)	94 Pu (244)	95 Am (243)	96 Cm (247)	97 Bk (247)	98 Cf (251)	99 Es (252)	100 Fm (257)	101 Md (258)	102 No (259)	103 Lr (262)

Constants and Factors:

$$R = 8.314 \frac{J}{mol \cdot K} = 0.08206 \frac{L \cdot atm}{mol \cdot K}$$

$$1 atm = 760 torr$$

$$N_A = 6.022 \times 10^{23}$$

$$h = 6.63 \times 10^{-34} J \cdot s$$

$$c = 3.00 \times 10^8 m/s$$

$$R_h = 1.097 \times 10^7 m^{-1}$$

$$B = -2.18 \times 10^{-18} J$$

$$m_e = 9.11 \times 10^{-31} kg$$

$$T_K = T_C + 273.15$$

Equations:

$$q = mC_S\Delta T$$

$$\Delta H_{RXN}^\circ = \sum_{Prod} n\Delta H_f^\circ - \sum_{React} n\Delta H_f^\circ$$

$$PV = nRT$$

$$u_{rms} = \sqrt{\frac{3RT}{M}}$$

$$P_A = \chi_A \cdot P_T$$

$$c = \lambda\nu$$

$$E = h\nu$$

$$p = mv$$

$$\lambda = \frac{h}{p}$$

$$\frac{1}{\lambda} = R_h \left(\frac{1}{n_a^2} - \frac{1}{n_b^2} \right) E = \frac{-Z^2 B}{n^2}$$

$$E = \frac{kQ_1Q_2}{r}$$

$$b.o. = 1/2(\#e^- - \#e^{-*})$$

$$\Delta H \approx \sum_{bonds\ broken} D - \sum_{bonds\ formed} D$$

