

Test Prep Day 2

- Notes:
- For an atom/functional group to be oxidized is for it to lose electrons ~~charge increases~~, oxidation number increases
 - Likewise, it is reduced if it gains electrons, and then the oxidation number is reduced.

Problems: (A)

1: A
2: B
3: D
4: C

(B)

1: T T CE
2: F T

(C)

1: D
2: A
3: C