Examples of isotope equations involving alpha decay, beta decay, positron decay, and gamma radiation

- 1. Alpha decay: $^{237}_{93}\text{Np} \longrightarrow ^{11}_{12}\text{Pa} + ^{4}_{2}\text{He}$
- 2. Alpha decay: $^{238}_{92}U \longrightarrow Th + ^{4}_{2}He$
- 3. Alpha decay: $^{235}_{92}U \longrightarrow ^{11}_{92}He$
- 4. Alpha decay: ${}^{226}_{88}$ Ra \longrightarrow Rn + ${}^{4}_{2}$ He
- 5. Alpha decay: $^{210}_{84}$ Po \longrightarrow Pb + $^{4}_{2}$ He
- 6. Beta decay: ${}^{14}_{6}\text{C} \longrightarrow \text{m} \text{N} + \beta^{-} + \bar{\nu}_{e}$
- 7. Beta decay: ${}^{210}_{82}\text{Pb} \longrightarrow \text{ III} \text{Bi} + \beta^- + \bar{\nu}_e$
- 8. Beta decay: ${}_{1}^{3}H \longrightarrow {}_{1}^{4}H + \bar{\nu}_{e}$
- 9. Beta decay: $^{234}_{90}$ Th \longrightarrow $^{11}_{11}$ Pa + β^- + $\bar{\nu}_e$
- 10. Beta decay: $^{131}_{53}I \longrightarrow \text{Tr} Xe + \beta^- + \bar{\nu}_e$
- 11. Positron decay: ${}^{11}_{6}\text{C} \longrightarrow {}^{11}_{6}\text{B} + \beta^{+} + \nu_{e}$
- 12. Positron decay: ${}^{13}_{7}N \longrightarrow C + \beta^{+} + \nu_{e}$
- 13. Positron decay: ${}^{15}_{8}O \longrightarrow N + \beta^{+} + \nu_{e}$
- 14. Positron decay: ${}^{18}_{9}F \longrightarrow {}^{11}_{11}O + \beta^+ + \nu_e$
- 15. Positron decay: $^{22}_{11}$ Na \longrightarrow $^{11}_{11}$ Ne + β^+ + ν_e
- 16. Gamma decay: $^{60}_{27}\text{Co} \longrightarrow ^{\circ\circ}_{11}\text{Co} + \gamma$
- 17. Gamma decay: $^{137}_{55}$ Cs \longrightarrow [1] Cs + γ
- 18. Gamma decay: $^{131}_{53}I \longrightarrow III + \gamma$
- 19. Gamma decay: $^{226}_{88}$ Ra \longrightarrow Ra + γ
- 20. Gamma decay: $^{234}_{90}$ Th \longrightarrow 777 Th + γ