

# Overview: Basic concepts on Quantum Mechanics

Classical physics  
Part I

Javier Orduz-Ducuara

<sup>1</sup>Baylor University

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# Objectives

In this course [General objective] participants will. . .

## General Objective (chap. 1)

Appraise historical facts and experiments that started the revolutionary change to Quantum Mechanics such as postulates, entanglement, superposition and others, which are used in Quantum Computing

# How do we do?

In this chapter, participants will. . .

## Particular Objectives

- ▶ Recognize differences between Classical Mechanics and Quantum Mechanics
- ▶ Describe key basic concepts inside Modern Physics
- ▶ Analyze concepts on Physics and Mathematics
- ▶ Study some relevant experiments for the Quantum Mechanics.
- ▶ Do exercise with definitions (Kahoot, forms, and some other exercises)

# What is Quantum Mechanics?



# Quantum Mechanics

## Concept

is a fundamental theory in physics that provides a description of the physical properties of nature at the scale of atoms and subatomic particles.



# The scope of QM



Now it should not be thought that “quantum physics” is something which does not concern the macroscopic world. Actually *all* of physics is quantum physics; the laws of quantum physics as we know them today are our most general laws of nature.

Figure: Source [3]



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# Phenomenological theories

## Phenomenological theories or Classical theories ...

attempt to describe and summarize experimental facts within some limited domain of physics.

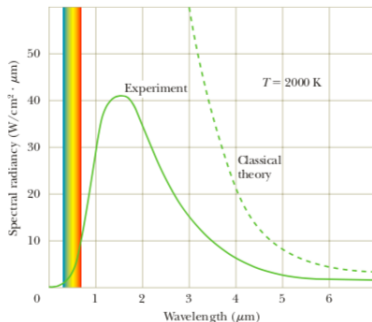
Today, every physical theory is "phenomenological"



## Why do we need QM?

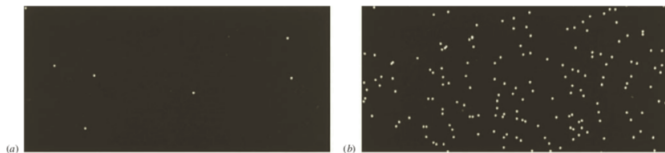
The solid curve shows the experimental spectral radiance for a cavity at 2000 K. Note the failure of the classical theory, which is shown as a dashed curve. The range of visible wavelengths is indicated. (Source [1]).

$$B_{\lambda}(\lambda, T) = \frac{2hc^2}{\lambda^5} \frac{1}{e^{hc/(\lambda k_B T)} - 1} \quad (1)$$

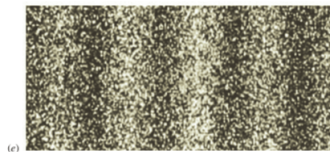


Eq.(1) represents energy per unit volume per unit wavelength

# Double-slit

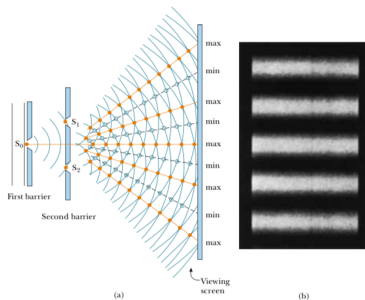


An interference pattern by a beam of electrons in a two-slit interference experiment. Check the ref. [1, pag.1167]



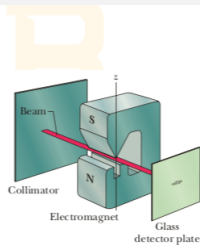
# Double-slit

(a) Schematic diagram of Young's double-slit experiment. Slits  $S_1$  and  $S_2$  behave as coherent sources of light waves that produce an interference pattern on the viewing screen. (b) An enlargement of the center of a fringe pattern formed on the viewing screen with many slits could look like this (Check the ref. [1, pag.1054]).



# Stern-Gerlach

When the electromagnet is off, the silver deposit is a narrow spot. However, when the electromagnet is turned on, the silver deposit should be spread vertically. See ref. [1, pag. 1226]



The reason is that silver atoms are magnetic dipoles, and so vertical magnetic forces act on them as they pass through the vertical magnetic field of the electromagnet; these forces deflect them slightly up or down.



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# Stern-Gerlach

We don't use Lorentz force, since  $q = 0$ . We use

$$U = -\vec{\mu} \cdot \vec{B}$$
$$\frac{dU}{dz} = -\mu_z \frac{dB}{dz}$$

and  $F_z = -\frac{\partial U}{\partial z}$ . Where  $\vec{\mu}$  is (dipole) magnetic moment vector in z-axis and Field vector component (z-axis). Therefore,

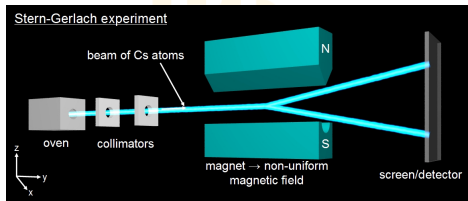
$$F_z = \mu_z \frac{dB}{dz}. \quad (2)$$

Eq. (2) represents the force experimented by an atom with a magnetic dipole moment experiences a force in a non-uniform magnetic field.



# Stern-Gerlach

But the experiment showed



A silver atom consists of many electrons, each with a spin magnetic moment and an orbital magnetic moment. All those moments vectorially cancel out except for a single electron, and the orbital dipole moment of that electron is zero.



# Stern-Gerlach

Combining the dipole moment  $\vec{\mu}$  of a silver atom is the spin magnetic dipole moment of that single electron.

Therefore, we find

$$F_z^{\uparrow} = +\mu_z \frac{dB}{dz}. \quad (3)$$

$$F_z^{\downarrow} = -\mu_z \frac{dB}{dz}. \quad (4)$$

Reminder:

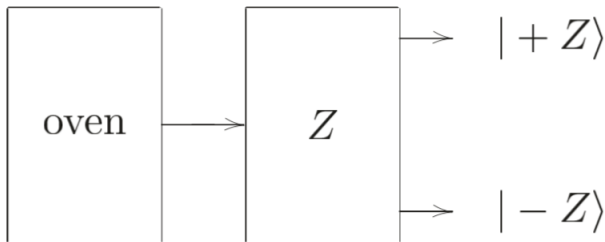
$$\vec{\mu}_s = -\frac{e}{m} \vec{S} \quad (5)$$

is the magnetic dipole moment,  $\vec{S}$  is the spin angular momentum (or spin),  $e$  and  $m$  are the electric charge and the mass of electron.



## Extension of Stern-Gerlach apparatus (STA)

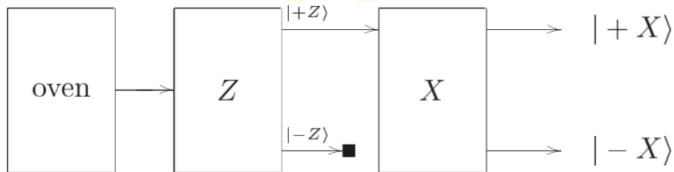
We use one STA [2, pag. 43]





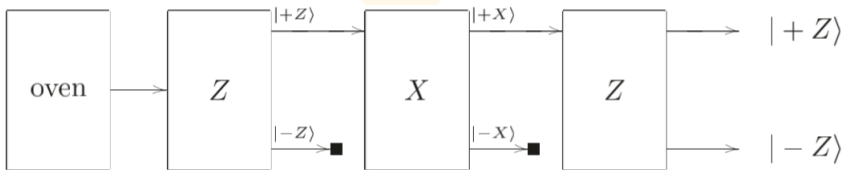
## Extension of Stern-Gerlach apparatus (STA)

We use two STA



# Extension of Stern-Gerlach apparatus (STA)

We use three STA



# Analysis

We use next assignments

$$|+Z\rangle \leftarrow |0\rangle$$

$$|-Z\rangle \leftarrow |1\rangle$$

$$|+X\rangle \leftarrow \frac{|0\rangle + |1\rangle}{\sqrt{2}}$$

$$|-X\rangle \leftarrow \frac{|0\rangle - |1\rangle}{\sqrt{2}}$$

# Analysis

The results of the cascaded SG experiment can be explained by assuming that the  $\hat{z}$  SGA measures the spin (qubit) in the computational basis  $|0\rangle, |1\rangle$ , and the  $\hat{x}$  SGA measures the spin with respect to the basis

$$\frac{|0\rangle + |1\rangle}{\sqrt{2}}, \frac{|0\rangle - |1\rangle}{\sqrt{2}}. \quad (6)$$

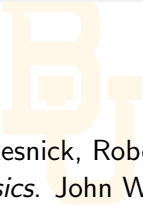
In the cascaded  $\hat{z} \hat{x} \hat{z}$  experiment, if we assume that the spins are in the state  $|+Z\rangle = |0\rangle = (|+X\rangle + |-X\rangle)/\sqrt{2}$  after exiting the first SG experiment, then the probability for obtaining  $|+X\rangle$  out of the second apparatus is  $\frac{1}{2}$ , and the probability for  $|-X\rangle$  is  $\frac{1}{2}$ . Similarly, the probability for obtaining  $|+Z\rangle$  out of the third apparatus is  $\frac{1}{2}$ .

## Conclusions

- ▶ We recognized differences between Classical Mechanics and Quantum Mechanics
- ▶ We described key basic concepts inside Modern Physics
- ▶ We analyze concepts on Physics and Mathematics.
- ▶ We studied some relevant experiments for the Quantum Mechanics.

Go to next lectures: **Part II**

# References

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- [1] Halliday, David and Resnick, Robert and Walker, Jearl. *Fundamentals of physics*. John Wiley and Sons, 2013.
  - [2] Michael A Nielsen and Isaac Chuang. Quantum computation and quantum information, 2002.
  - [3] Wichmann, Eyvind Hugo. *Quantum Physics: Berkeley Physics Course Vol. 4*, volume 4. Tata McGraw-Hill Education, 2010.