Entropy of the ideal gas (P3.8; LN1.4)

$$S(k) = k_B \ln \Omega(k)$$
;  $\Omega = ur$  of status accessible at  $2$ 

Volume  $V = L \times L \times L$ 
 $\Psi = e^{i \frac{L}{k} \cdot L}$  Periodic boundary conditions.

 $k = \frac{2\pi}{L}(n_x, n_y, n_z)$ 

Grid in  $3D$ .  $\{p_i\}$ : grid in  $3N$  dim.  $\{p_i\} = P$  ( $3N$ )

Grid const:  $\frac{2\pi h}{L} = hk$ .  $(p = hk)$ 

States have energy  $P = \sum_{i} \frac{p_i^2}{2m} = \frac{p^2}{2m}$ 

States with  $P = \sum_{i} \frac{p_i^2}{2m} = \frac{p^2}{2m}$ 
 $P(P, 3N) = \frac{dV(P, 3N)}{dP} dP = \frac{dV($ 

So 
$$S = k_B \int N \ln \left( \frac{V}{h^3} (2\pi M R)^{3/2} \right) + \ln 3N - \left( \frac{3N+1}{2} \right) \left( \ln \left( \frac{3N+1}{2} \right) - 1 \right) + k_B \ln 3p$$

$$= k_B \int N \ln \left( \frac{V}{h^3} \left( \frac{4\pi m R}{3N} \right)^{3/2} + \frac{3}{2} \right) + O(\ln N) + \frac{3N+1}{2} \ln N$$

$$\frac{1}{T} = \frac{\partial S}{\partial \mathcal{R}}\Big|_{\mathcal{H}, \mathcal{V}^-} = \frac{3}{2} \frac{\mathcal{N} k_{\mathcal{B}}}{\mathcal{R}}, \quad \sigma: \quad \mathcal{R} = \frac{3}{2} \mathcal{N} k_{\mathcal{B}} \mathcal{T}$$

$$C_V = \frac{3}{2} N k_B$$

Furthermore: 
$$P = \frac{\partial S}{\partial V} = \frac{Nk_B}{V} \rightarrow \frac{PV = Nk_BT}{V}$$

$$S = k_B N \left[ ln \left( \frac{V}{h^3} \left( \frac{4\pi m R}{3N} \right)^{3/2} + \frac{3}{2} \right] =$$

$$k_{B}N \left[ \ln V/h^{3} + \frac{3}{2} \ln \frac{4\pi M}{3} + \frac{3}{2} \ln \frac{2}{N} + \frac{3}{2} \right]$$

P2

Restra factor 
$$\frac{1}{N!}$$

$$S_{QM} = S_{CL} - k_B N(\ln N - 1)$$

$$= k_B N \left[ \ln \frac{V}{N} + \frac{3}{2} \ln \frac{4\pi m}{3 h^2} + \frac{3}{2} \ln \frac{R}{N} + \frac{5}{2} \right]$$

This is extensive.