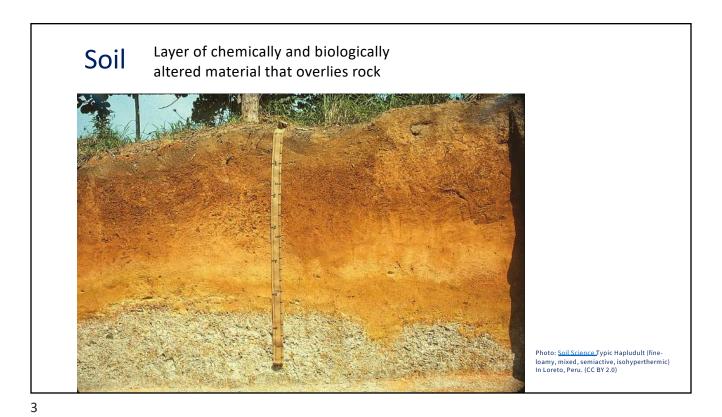
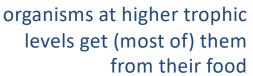


It all comes down to SOIL 12.02.2007



Soil Chemistry **TABLE 21.2 Plant Nutrients and Their Principal Functions Nutrients Principal functions** Carbon, hydrogen, Components of organic molecules oxygen Macronutrients: Nitrogen Component of amino acids, proteins, chlorophyll, nucleic acids Needed in large Phosphorus Component of ATP, NADP, nucleic acids, phospholipids amounts. Potassium Ionic/osmotic balance, pH regulation, regulation of guard cell turgor Calcium Cell wall strengthening and functioning, ionic balance, membrane permeability Magnesium Component of chlorophyll, enzyme activation Sulfur Component of amino acids, proteins Iron Component of proteins (e.g., heme groups), oxidation-reduction reactions Copper Component of enzymes Micronutrients: Manganese Component of enzymes, activation of enzymes Needed in trace Component of enzymes, activation of enzymes, component of ribosomes, Zinc amounts maintenance of membrane integrity Nickel Component of enzymes Molybdenum Component of enzymes Cell wall synthesis, membrane function Boron Chlorine Photosynthesis (water splitting), ionic and electrochemical balance Sources: Salisbury and Ross 1992; Marschner 1995. ECOLOGY, Table 21.2 © 2008 Sinauer Associates, Inc.







Plants get nutrients from their environment

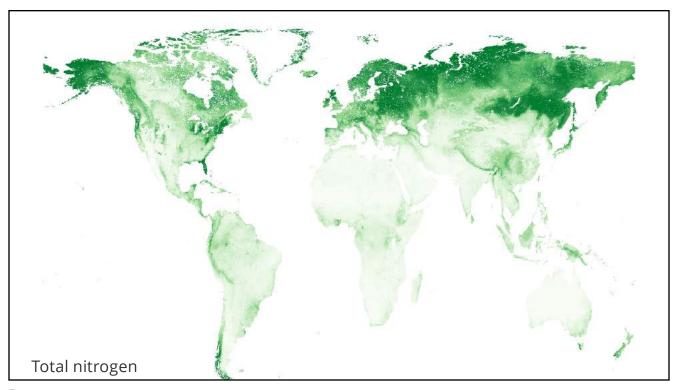
5

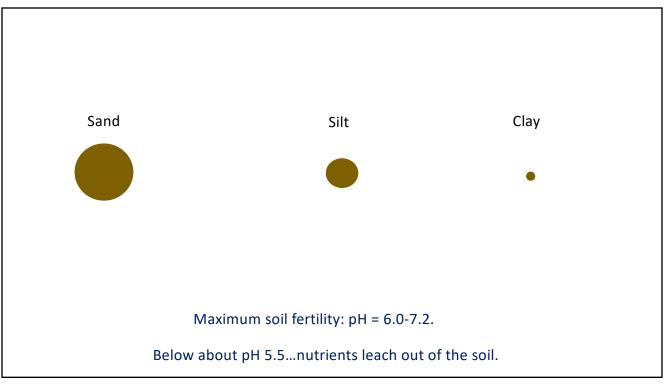
Nutrients must be present in simple, water soluble forms for plant roots to take them up

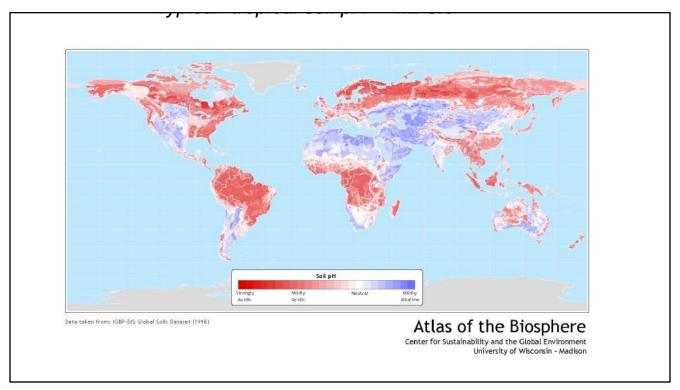


 $\begin{array}{c} \text{Ca}^{2+} \\ \text{K}^+ \\ \text{Mg}^{2+} \end{array}$

Nitrogen is the limiting nutrient in most tropical forests









Most of the nutrients in Trop. Rain Forests are in the living (tree) biomass.







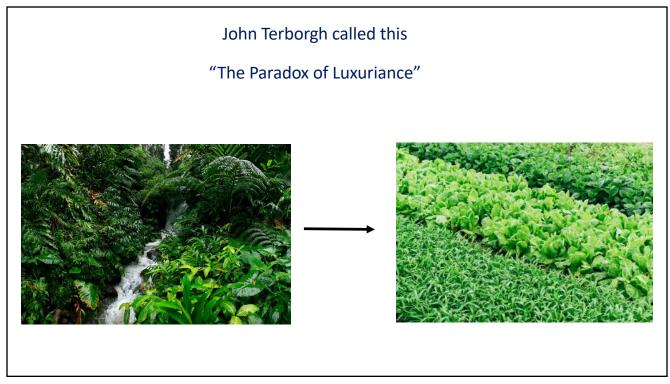
11

When tropical forests are cleared and burned, most nutrients are lost in smoke and ash and soil erosion.



If severe enough, these ecosystems can take centuries to return to their previous state.





John Terborgh called this

"The Paradox of Luxuriance"





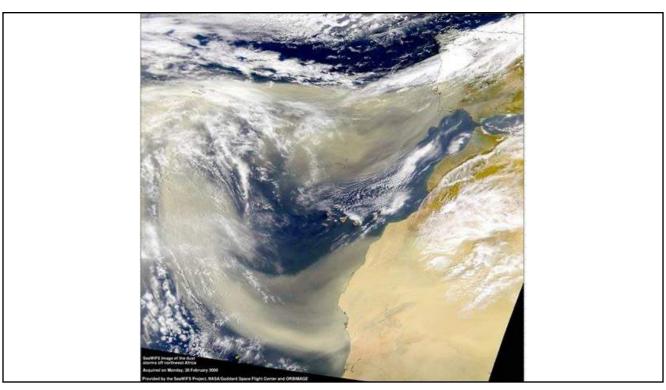
15

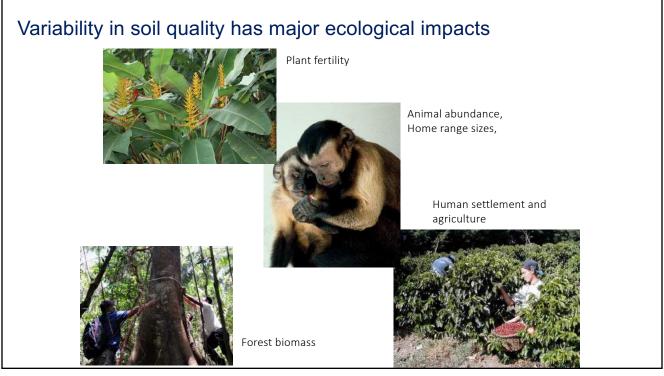
How do nutrients get back into the soils?

- Rain
- •Exposure of new parent material
- •River deposits
- Volcanoes
- •Outer space
- •Nutrient Cycling / Decomposition











Three key properties of soil:

(1) TEXTURE

(2) WATER RETENTION CAPACITY

(3) CHEMISTRY

1) Texture: size distribution of mineral particles

Sand

Silt

Clay

0.002 - 0.5 mm

<0.002 mm

2 - 0.5 mm

1) Texture: size distribution of mineral particles Sand Silt Clay 2 - 0.5 mm 0.002 - 0.5 mm < 0.002 mm 1. Largest Ratio of SA:V 2. Weak negative charge attracts positively charged lons* 3. Absorb water

2) Water Retention Capacity

Determined by type of particles in soil & their distribution throughout the soil

Clay Soils:

Hold large volumes of water

Sandy Soils:

Drain Well



What does this mean for nutrient availability?

3) Soil Chemistry: soil pH strongly influences nutrient availability pH: large concentration of H⁺ (acidic) pH: low concentration of H⁺ (Basic or alkaline) Clay H⁺ A little acidity promotes nutrient availability. How? Below about pH 5.5...nutrients leach out of the soil. Maximum soil fertility: pH = 6.0-7.2