

ODE and Diffusion Models of Tumor Growth

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1 Introduction

Researchers have been studying cancer and the patterns behind tumor growth for thousands of years. Hippocrates was accredited with possibly being the first to describe a tumor, and tumors were mainly described by their color and their response to touch. However, ancient physicians did not separate benign and malignant tumors, but they acknowledged the tumor's ability to infect nearby tissue. The application of mathematical models to tumor growth did not start to gain traction until much later, however. In the early 20th century, one of the most famous uses of the diffusion model was first developed: Hill released a paper discussing how the movement of lactic acid and oxygen throughout the body could be shaped into a diffusion problem (Hill, 1928). This paper eventually led to Burton's paper which modeled the growth of tumors using diffusion, applying many of the mathematical tools employed by Hill some years before (Burton, 1966). This paper will present a brief overview of the mathematics used in these papers and will recreate some of the graphics produced as well.

2 Hill's Paper

Hill's paper, titled 'Diffusion of Oxygen and Lactic Acid through Tissues', set the scene for most of the diffusion models developed in the late 20th century to present. In the introduction of this paper, Hill states that the 'diffusion of dissolved substances through cells and tissues is a determining factor in many vital processes'. The diffusion constant, k , is introduced, along with its usages. It is representative of the number of unit quantities of a substance that diffuse throughout an area of 1 cm^2 in a minute. The diffusion constant is usually a small value when related to systems that are outside of the body; for example, the diffusion constant would be very small when studying the diffusion of a gas within a large room. However, when studying systems within the body, the diffusion constant can be quite large. Hill gives the following example when explaining this:

Object	Diameter	Time to 90% Saturation
Cylinder	1 cm	185 minutes
Actual Nerve	0.7 mm	54 seconds
Single Nerve Fiber	7 μ (0.0007 cm)	0.0054 seconds

As seen by this example, many processes of diffusion throughout the body are extremely rapid. However, at the time of Hill's paper, another problem was evident: not many problems in diffusion could be completely solved using mathematical methods. Even when evaluating these equations, the existence of a unique solution was only guaranteed with certain initial conditions. The steady state also needed to have specific parameters in order for the result to be of any interest.

2.1 Diffusion Within a Steady State

The first problem that Hill dealt with was the case where oxygen was being diffused from a gaseous or liquid phase. It was assumed that the oxygen was being used up by metabolic processes at some rate a and that the concentration of oxygen, y_0 , is maintained constant at $x = 0$.

Let x be the distance from the constant source of oxygen to any point within the tissue. Given the assumptions above, the rate of diffusion across any unit area will be $J = -k \frac{dy}{dx}$, where J is the rate of diffusion, k is the diffusion rate and the (-) represents that the diffusion is occurring in the opposite direction from the source. This is more commonly known as Fick's first law of diffusion. To find the measure of this rate of accumulation, Fick's second law can be applied:

$$\frac{dJ}{dx} = k \frac{d^2y}{dx^2} \quad (1)$$

Hill then combined these equations along with a , the rate of usage by the metabolic process, to come up with the first diffusion equation for this model:

$$\frac{dy}{dt} + a = k \frac{d^2y}{dx^2} \quad (2)$$