

4. Relation between the Einstein coefficients. (Lasers)

To find out the relation, we assume that:

- i. The atoms and the radiation are thermal equilibrium.
- ii. The radiation is identical with black body radiation and consistent with Planck's radiation law for any value of  $T$ .
- iii. The population densities  $N_1$  and  $N_2$  at the lower and upper energy levels respectively, are constant in time and are distributed according to Boltzmann law in the energy levels.

The above conditions indicate that rate of change of atoms at the level  $E_2$  must be equal to zero. the number of transitions from  $E_2$  to  $E_1$  must be equal to the transitions from  $E_1$  to  $E_2$ .

so, we have,  $N_{12} = N_{21}$

$$N_1 P_{12} = N_2 P_{21}$$

$$N_1 B_{12} Q = N_2 [A_{21} + B_{21} Q]$$

$$N_1 B_{12} Q = N_2 A_{21} + N_2 B_{21} Q$$

$$N_2 A_{21} = Q(N_2 B_{21} - N_1 B_{12})$$

$$Q = N_2 A_{21}$$

$$N_2 B_{21} - N_1 B_{12}$$

$$Q = \frac{A_{21}/B_{21}}{\left[ \frac{N_1 B_{12}}{N_2 B_{21}} - 1 \right]}$$

According to Boltzmann distribution law, no. of items atoms  $N_1$  and  $N_2$  in energy states  $E_1$  and  $E_2$  in thermal equilibrium at temperature  $T$  are given,

$$N_1 = N_0 e^{-(E_1/k_B T)} \quad \text{and} \quad N_2 = N_0 e^{-(E_2/k_B T)}$$

where,  $N_0$  is the total no. of atoms and  $k_B$   $k_B$  is boltzmann's constant.

$$N_1 = e^{(E_2/k_B T)} - (e^{E_1/k_B T}) = e^{(E_2 - E_1/k_B T)}$$

$N_2$

$$\therefore N_1 = e^{(h\nu/k_B T)} \quad \dots E_2 - E_1 = h\nu$$

— energy of each photon of radiation

from (1),

$$Q = \frac{(A_{21}/B_{21})}{\left[ \frac{B_{12}}{B_{21}} e^{(h\nu/k_B T)} - 1 \right]} \quad — (2)$$

according to planck's radiation formula,

$$Q = \frac{8\pi h\nu^3}{c^3 [e^{(h\nu/k_B T)} - 1]} \quad — (3)$$

comparing equation (2) and (3),

$$\frac{B_{12}}{B_{21}} = 1 \Rightarrow B_{12} = B_{21} \quad — \text{answer}$$

$$\text{also, } A_{21} = \frac{8\pi h\nu^3}{c^3} \quad — \text{answer}$$

These equations present the relation between Einstein's coefficients.