CHE 260 - MIDTERM 2016

For const pressure process We + Wsh = AH = m(h2-h1) $P_1 = 175 \text{ kPa}$ $h_1 = h_2 = 487-01 \text{ hJ/hy}$ Let liq $J_1 = J_2 = 0.001057 \text{ m3/hy}$ $P_2 = 175 \text{ kPa} \text{ } 2 \text{ } \text{ hg} = 487.01 \text{ } \text{ kJ/H}$ 502 = 0.5 hg = 2700-2 kJ/hgh2 = hf+ x2 (hg-hf) = A87.01 + 0.5 (2700.2 - 487.01) = 1593-6 kJ/by. $m = \frac{V_1}{U_1} = \frac{5 \times 10^{-3} \, \text{m}^3}{0.001057 \, \text{m}^3/\text{hy}} = 4.730 \, \text{kg}$ m (h2-h1) Ve TAT + Wsh = Ve (8A)(45×60 B) + 400 RS = 4.730 (1593.6 - 487.01) 1000 J/pg

Ve = 223.8 V.

Argon P, = 900 kPa T = 450°C V, = 80 m/s P2=150 kPa A1 = 60 Cm2 At met 0, = RT1 = 0,2081 kJ × 723.15 K $m = A_1 V_1 = 80 \text{ m/s} \times 60 \times 10^{-4} \text{ m}^2 = 2.871 \text{ g/s}$ Energy Balance Energy Balance $w + 9 = m(h_2 - h_1) + v_2 - v_1^2$ $=250 \, \text{kJ} = 2.871 \, \text{kg} \left(0.5203 \left(T_2 - 450C\right)\right)$ + (150 m/s)2 - (80 m/s)2 2 × 1000 T/RJ-=> $T_2 = 267-2$ °C.

$$\frac{A}{200 \text{ kpa}} \frac{13}{2 \text{ Mpa}}$$

$$x = 0.8 \text{ } 400^{\circ}\text{C}$$

$$V_{A} = V_{B} = 1 \text{ m}^{3}$$

 $P_{2} = 1 \text{ MPR}.$

At 200 12 Pa

$$U_{f} = 0.001061 \text{ m}^{3}/p_{f} \quad U_{g} = 0.88578 \text{ m}^{3}/h_{f}$$

$$U_{f} = 504.50 \quad p_{J}/h_{f} \quad U_{g} = 2024.6 \quad p_{J}/h_{f}.$$

$$U_{A,1} = 504.50 + 0.8 \times 2024.6 = 2124.18 \quad p_{J}/p_{g}$$

$$U_{A,1} = 0.001061 + 0.8 (0.88578 - 0.001061)$$

$$= 0.70884 \quad m^{3}/p_{g}$$

At 2 MPa; 400° C.

$$G_{31} = 0.15122 \, \text{m}^{3}/\text{hg}$$
 $U_{B_1} = 2945.9 \, \text{kg}$

$$M_{A} = \frac{1}{V_{A_{31}}} = \frac{1}{0.70884} \, \text{m}^{3}/\text{hg}$$

$$= 1.411 \, \text{kg}$$

$$m_{B} = \frac{1}{U_{B,1}} = \frac{1}{0.15122} \, m^{3}/hy = 6.613 \, kg$$

$$m_{bt} = m_A + m_B = 1.411 + 6.613 = 8.024 \text{ bg}$$

$$U_2 = \frac{2m^3}{8.024} = 0.24925 \frac{m^3}{hy}, P_2 = 1HPq$$

$$\frac{T_2 - 250}{300 - 250} = \frac{0.24925 - 0.23275}{0.25799 - 0.23275} \Rightarrow T_2 = 289.20$$

$$\frac{U_2 - 2710.4}{2793.7 - 2710.4} = \frac{0.24295 - 0.23275}{0.25799 - 0.23275} \Rightarrow U_2 = \frac{2764.9}{18}$$

$$6 = -U_2 - U_1 \pm -m_{bot} U_2 - (M_4 U_{A_1} + m_B U_{B_1})$$

$$= 8.024 \times 2764 - 9 - (1.411 \times 2124 - 18 \pm 6.613 \times 2945 - 9)$$

$$= -292 - 9 \text{ kJ}$$

(5)
$$400^{\circ}C$$
 (1) $2HR$

(6) $10RR$
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