CHE 260 MIDTERM SOLUTION - 2022

$$\frac{m_{1} + m_{2} = m_{3}}{m_{1}h_{1} + m_{2}h_{2} = m_{3}h_{3}}$$

$$= \frac{m_{1}h_{1} + m_{2}h_{2} = m_{3}h_{3}}{m_{1}h_{1} + m_{2}h_{2} = (m_{1} + m_{2})h_{3}}$$

$$\Rightarrow \frac{m_{1}h_{1} + m_{2}h_{2} = (m_{1} + m_{2})h_{3}}{m_{1}(h_{1} - h_{3}) = m_{2}(h_{3} - h_{2})}$$

$$\Rightarrow \frac{m_{1}c_{p}(T_{1} - T_{3}) = m_{2}c_{p}(T_{3} - T_{2})}{m_{2}c_{p}(T_{1} - T_{3})}$$

$$\Rightarrow \frac{m_{1}c_{p}(T_{1} - T_{3}) = m_{2}c_{p}(T_{3} - T_{2})}{m_{2}c_{p}(T_{1} + m_{2}T_{2})}$$

$$\Rightarrow T_{3} = \frac{m_{1}T_{1} + m_{2}T_{2}}{m_{2}c_{p}(T_{3} + m_{1})}$$

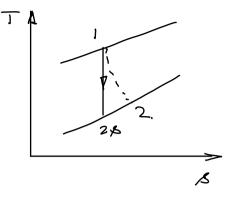
$$T_{3} = \frac{m_{1}T_{1} + m_{2}T_{2}}{m_{2}c_{p}(T_{3} + m_{1})}$$

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T3= 326.7 K.

 $m_1 k_1 + m_2 k_2 + Sgn = m_3 k_3 = m_1 k_3 + m_2 k_3$ => Sgn = $m_1 (k_3 - k_1) + m_2 (k_3 - k_2)$ => Sgn = $m_1 c_1 k_2 + m_2 c_2 k_3 + m_2 k_3$ => Sgn = $m_1 c_2 k_3 + m_2 k_3$ $c_3 k_4 + m_2 k_3 + m_2 k_3$ $c_4 k_5 + k_5 k_4 k_5$

Sgen = 1/8/5 × 1.004 kJ lu 326.7 K + 2/2 × 1.004 kJ × 326.7 K Sgen = 0.0 3 6 kW/K.



$$\frac{T_{2R}}{T_{1}} = \left(\frac{\rho_{2}}{\rho_{1}}\right)^{\frac{\gamma-1}{\gamma}}$$

$$= 725 = 573 \times \left(\frac{200 \text{ kPa}}{2200 \text{ kPa}}\right) \frac{0.395}{1.395} = 290.6 \times 10^{-10}$$

$$\omega_{t,g} = C_P (T_{26} - T_1)$$

$$\omega_{t} = \int_{t}^{t} \omega_{t,8} = 0.85 (-286.1)^{k} \frac{J}{k}$$
, $\omega_{t} = -243.2 \frac{kJ}{k}$

$$T_{2} = T_{1} + \omega_{t}/C_{p}$$

$$= 573 \times + (-243.2 \text{ kJ/kg})$$

$$\frac{1.013 \text{ kJ/kg}}{}$$

(3)
$$1200 \text{ kfa}$$
 300°C
 0°C
 $0^{\circ}\text{$

$$\hat{w} = 0.4 \, \text{kg/s} \times 1.018 \, \text{kJ} (300-20)^2 + 15 \, \text{kW}$$
 $\hat{w} = 129.0 \, \text{kW}$

b) Rate of entropy generation Sgen = a Sair + & Sour.

Sgan =
$$m(Cp ln \frac{T_2}{T_1} - R ln \frac{P_2}{P_1}) + \frac{g_1}{Tswr}$$

IMPa
$$2 h_1 = 3051.6 kJ/hg$$

 $300^{\circ}C$ $8_1 = 7.1246 kJ/hgK$
at 15 kAa
 $h_1 = 225.94 kJ$ $h_2 = 2598.3 kJ$
 $h_3 = 225.94 kJ$ $h_4 = 2598.3 kJ$
 $h_4 = 225.94 kJ$ $h_4 = 2598.3 kJ$
 $h_4 = 0.7549 kJ$ $h_4 = 8.0071 kJ$

For ventrapic expansion &2= &,

$$\frac{\mathcal{L}_{2k} = \frac{k_{2k} - k_f}{k_g - k_f} = \frac{7.1246 - 0.7549}{8.0071 - 0.7549} = 0.8783$$

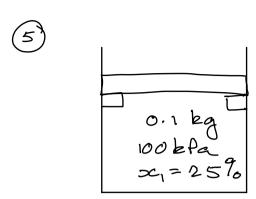
$$h_{28} = h_f + \infty_{28} (h_g - h_f)$$

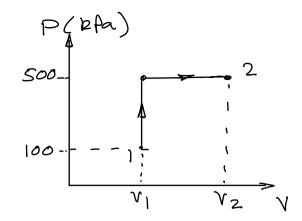
$$= 225.94 + 0.8783 (2598.3 - 225.94)$$
 $h_{28} = 2309.6 = \frac{1}{28} h_f$

$$\omega_{\mathcal{B}} = h_{28} - h_{1} = 2309.6 - 3051.6 = -742.0 \frac{\text{kJ}}{\text{hg}}$$

$$M_{\mathcal{E}} = \frac{\omega_{\alpha}}{\omega_{\delta}} = \frac{-600 \frac{\text{kJ}}{\text{hg}}}{-742.0 \frac{\text{kJ}}{\text{hg}}} = 80.9 \frac{9}{0}$$

$$x_2 = 0.9382$$





at 100 kPa $v_f = 0.001043 \text{ m}^3/\text{kg}$, $v_g = 1.6941 \text{ m}^3/\text{kg}$ $v_f = 417.40 \text{ kJ/kg}$ $v_g = 2505.6 \text{ kJ/kg}$

 $U_1 = U_f + \infty, (U_q - U_f)$ $= 0.001043 \quad m^3/b_f + 0.25(1.6941 - 0.001043) \frac{3}{b_f}$ $U_1 = 0.42431 \quad m^3/b_f$

At $500 \, \text{kp}$, $300^{\circ} \, \text{c} \Rightarrow \sigma_2 = 0.52261 \, \text{m}^3/\text{hg}$ $\text{km} \, \text{ce} \quad \sigma_2 \Rightarrow \sigma_1 \Rightarrow \text{Piston is lifted}.$

Constant pressure process $= -P (V_2 - V_1)$ $= -P m (v_2 - v_1)$ $= -W_{12} = -500 \text{ kfax } 0-1 \text{ kg } (0.52261 - 0.4243) \text{ m}/\text{hg}$ $W_{12} = -4.92 \text{ kJ}$