NPRE446: Radiation Interactions With Matter I

University of Illinois, Urbana-Champaign

Fall 2019

Instructor: Prof. Kathryn Huff Time: MWF 9:00–10:00

Email: kdhuff@illinois.edu Place: 305 Materials Science

Teaching Assistant Natalie Gaughan

Email: ncg5@illinois.edu

Course Pages:

1. https://compass2g.illinois.edu

2. https://github.com/katyhuff/npre446

3. https://katyhuff.youcanbook.me

TA Office Hours: The teaching assistant for the course, Natalie Gaughan, will hold office hours TBD from TBD to TBD in TBD Talbot Laboratory.

Office Hours: Prof. Huff will hold office hours by appointment only, in her office, 118 Talbot Laboratory, 104 S. Wright St.. Please make use of the teaching assistant and your colleagues before booking an appointment with Prof. Huff. You can make an appointment at katyhuff. youcanbook.me.

Main References: A few essential references for this course will be assigned as readings. The required texts for this course are [1] and [2]. The more recent [3] is not required, but can be used instead of the second edition. Additional, a recommended texts include [4, 5, 6].

- [1] David J. Griffiths. *Introduction to Quantum Mechanics*. Pearson Prentice Hall, Upper Saddle River, NJ, 2nd edition edition, April 2004.
- [2] Sidney Yip. Nuclear Radiation Interactions. World Scientific Publishing Company, October 2014.
- [3] David J. Griffiths and Darrell F. Schroeter. *Introduction to Quantum Mechanics*. Cambridge University Press, Cambridge; New York, NY, 3 edition edition, August 2018.

- [4] Kenneth S. Krane. *Introductory Nuclear Physics*. Wiley, New York, 3 edition edition, October 1987.
- [5] Robert Eisberg and Robert Resnick. Quantum Physics of Atoms, Molecules, Solids, Nuclei, and Particles. John Wiley & Sons, New York, 2nd edition edition, 1985.
- [6] Robley Dunglison Evans. Atomic Nucleus. TATA McGraw-Hill, Bombay, New Delhi, 1955.

Description: The classical and quantum theories of the interaction of radiation (neutrons, photons, and charged particles) with matter are the core components of nuclear and materials science and engineering. At UIUC, we offer a sequence of four courses (446, 447, 521, and 529) at progressively deepening levels on this subject. The sequence, in the aggregate, aims to provide the students with solid training in essential physical principles, mathematical competence, and computational skills. In this course, we provide a quantitative introduction to introductory quantum mechanics, fundamentals of atomic and nuclear physics, and interactions of radiation with matter.

Objectives:

The quantum physics elements of this course will equip students to

- Recognize the limitations of classical theory (stable atomic model, black-body radiation, the photoelectric effect, and Compton scattering).
- Apply classical theory to fundamental problems.
- Explain the implications of wave-particle duality.
- Apply operators in physical equations.
- Solve the Schrödinger equation for canonical problems.
- Understand and apply fundamental concepts (eigenstates, observables, statistical interpretation, probability conservation, bound/unbound states).
- Identify the bound and unbound states in key potentials: square potential, Harmonic oscillator, ladder operators, free particle, δ potential.
- Define, derive, and apply the uncertainty principle.

The atomic and nuclear physics elements of this course will equip students to:

- Fundamentally describe atomic structure and nuclear properties.
- Understand and use nuclear physics terminology.
- Differentiate nuclei and interactions via their size, distributions of charge and mass, abundance, and binding energy.
- Identify and calculate nuclear phenomena via Q-value, separation energy, semi-empirical mass formula, liquid drop model, mass parabola, spin, parity, electromagnetic moments), nuclear force and nuclear structure.

- Define and apply the properties of nuclear forces, deuteron structure, neutron-proton scattering, phase shift, partial wave approximation, scattering length, differential cross section, exchange force model, nuclear shell model, nuclear magic numbers.
- Model and solve equations related to radioactive decay and binary nuclear reactions.

The interaction of radiation with matter portion of this course will equip students to:

- Categorize and characterize neutron interations with matter.
- Identify and quantify neutron-proton scattering, energy dependence of cross sections, and moderation.
- Categorize and characterize gamma interactions with matter.
- Identify and quantify gamma attenuation, the photoelectric effect, Compton scattering, pair production.
- Categorize and characterize heavy and light charged particle interactions with matter.
- Identify and calculate the features of charged particle stopping power, Bragg curve, range, ionization loss, and radiation loss.

Prerequisites:

- Junior standing is recommended.
- MATH 285
- PHYS 211 214, or equivalent.
- Linear Algebra (MATH 125) is not required, but highly recommended.

Grading Policy: Grades will be assigned as a weighted sum of the following work.

Work	Weight
Homework	(35%)
Quizzes	(5%)
Midterm 1	(15%)
Midterm 2	(15%)
Final	(30%)
Total	(100%)

Important Dates: The following dates are subject to change.

Midterm #1 September 25, 2019, 9:00am-10:00am Midterm #2 November 6, 2019, 9:00am-10:00am Final Exam December 18, 2019, 1:30pm-4:00pm

Class Policies:

Integrity: This is an institution of higher learning. You will be swiftly ejected from the course if you are caught undermining its integrity. Note the Student's Quick Reference Guide to Academic Integrity and the Academic Integrity Policy and Procedure.

Attendance: Regular attendance is mandatory. Request approval for absence for extenuating circumstances prior to absence.

Electronics: Active participation is essential and expected. Accordingly, students must turn off all electronic devices (laptop, tablets, cellphones, etc.) during class. Exceptions may be granted for laptops if engaging in computational exercises or taking notes.

Collaboration: Collaboratively reviewing course materials and studying for exams with fellow students can be enriching. This is recommended. However, unless otherwise instructed, homework assignments are to be completed independently and materials submitted as homework should be the result of one's own independent work.

Late Work: Late work has a halflife of 1 hour. That is, adjusted for lateness, your grade G(t) is a decaying percentage of the raw grade G_0 . An assignment turned in t hours late will receive a grade according to the following relation:

$$G(t) = G_0 e^{-\lambda t}$$

where

$$G(t)=$$
 grade adjusted for lateness $G_0=$ raw grade
$$\lambda=\frac{ln(2)}{t_{\frac{1}{2}}}=$$
 decay constant $t=$ time elapsed since due [hours] $t_{1/2}=1=$ half-life [hours]

Make-up Work: There will be no negotiation about late work except in the case of absence documented by an absence letter from the Dean of Students. The university policy for requesting such a letter is in the Student Code. Please note that such a letter is appropriate for many types of conflicts, but that religious conflicts require special early handling. In accordance with university policy, students seeking an excused absence for religious reasons should complete the Request for Accommodation for Religious Observances Form, which can be found on the Office of the Dean of Students website. The student should submit this form to the instructor and the Office of the Dean of Students by the end of the second week of the course to which it applies.

Accessibility: I hope that this course will be inclusive and accommodating for all learners. As such, I am committed upholding the vision and values of Inclusive Illinois in my classroom. With regard to accommodating all learners, please note that many resources are provided through the Division of Disability Resources and Educational Services. To request particular accommodations, please contact me as soon as possible so that we can work out any necessary arrangements.

Other Resources: University students typically experience a wide range of stressors during their time on campus. Accordingly, campus resources exist to help students manage stress levels, mental health, physical health, and emergencies while navigating this environment. I hope you will take advantage of these campus resources as soon as they can be of help.

- The Campus Recreational Centers
- The Counselling Center
- The McKinley Health Clinic
- The McKinley Mental Health Clinic
- The Emergency Dean

Run. Hide. Fight. It is important that we take time to prepare for a situation in which our safety could depend on our ability to react quickly. Please review the university guidance on responding to emergency situations https://police.illinois.edu/emergency-preparedness/run-hide-fight/.

Course Schedule: This schedule is subject to change

Date	Week	Day	\mathbf{Unit}	Chap.	Quiz Due	HW Given	HW Due
08-26	1	M	Intro	G1			
08-28	1	W	Wave Function	G1			
08-30	1	\mathbf{F}	Wave Function	G1		HW1	
09-02	2	M	• No Class •	Q1			
09-04	2	W	Wave Function	G1			
09-06	2	\mathbf{F}	Wave Function	G1			
09-09	3	\mathbf{M}	Wave Function	G1	Q2		
09-11	3	W	Time-Independent Schrödinger	G2			
09-13	3	\mathbf{F}	Time-Independent Schrödinger	G2		HW2	HW1
09-16	4	\mathbf{M}	Time-Independent Schrödinger	G2	Q3		
09-18	4	W	Time-Independent Schrödinger	G2			
09-20	4	\mathbf{F}	Time-Independent Schrödinger	G2			
09-23	5	\mathbf{M}	Review	G3	Q4		
09-25	5	W	ullet Midterm $ullet$		•		
09-27	5	\mathbf{F}	Formalism	Y3		HW3	HW2
09-30	6	M	Nuclear Properties	Y3	Q5		
10-02	6	W	Nuclear Properties	Y3	•		
10-04	6	\mathbf{F}	Nuclear Properties	Y4		HW4	HW3
10-07	7	W	Nuclear Properties	Y4	Q6		
10-09	7	W	Stability of Nuclei	Y4	·		
10-11	7	\mathbf{F}	Stability of Nuclei	Y5		HW5	HW4
10-14	8	\mathbf{M}	Energy Level Models	Y5	Q7		
10-16	8	W	Energy Level Models	Y5	-0		
10-18	8	\mathbf{F}	Nuclear Decay	Y6		HW8	HW7
10-21	9	${ m M}$	Collision Cross Sections	Y7	Q8		
10-23	9	W	Collision Cross Sections	7	400		
10-25	9	\mathbf{F}	Collision Cross Sections	7		HW9	HW8
10-28	10	M	Collision Cross Sections	·	Q9	11,10	11,10
10-30	10	W	Collision Cross Sections	7	Q 0		
11-01	10	\mathbf{F}	Neutron Scattering	7		HW10	HW9
11-04	11	M	Neutron Scattering	9	Q10	,, _ ,	
11-06	11	W	Review	9	Q-0		
11-08	11	F	• Midterm •	9		HW11	HW10
11-11	12	M	Neutron Scattering	9	Q11	11 11 11	11,1,10
11-13	12	W	Neutron Scattering	9	Q11		
11-15	12	F	γ Scattering & Absorption	9		HW12	HW11
11-18	13	M	γ Scattering & Absorption	Ü		11 11 12	11,,11
11-20	13	W	γ Scattering & Absorption				
11-22	13	F	γ Scattering & Absorption				
11-25	14	M	• No Class •	10	Q12		
11-27	14	W	• No Class •	10	W12		
11-27	14	F	• No Class •	11		HW13	HW12
12-02	15	M	Charged Particle Stopping	12	Q13	11 11 10	11 11 14
12-02	15 15	W	Charged Particle Stopping Charged Particle Stopping	12	Ø10		
12-04	15 15	F	Charged Particle Stopping Charged Particle Stopping	12			HW13
12-00	16	M	Charged Particle Stopping Charged Particle Stopping	12			
12-09	16	W	Review	14			page 6 of 7
12-11	10 17	W	• Final Exam •				
14-10	т (v v	▼ Filiai Exaili ♥				

Course Schedule: Note that this schedule is subject to change

Date	Week	Day	\mathbf{Unit}	Chap.	\mathbf{Quiz}	$\mathbf{H}\mathbf{W}$	HW
						Given	Due
1++?	1++?	j++į	1++?	j++j	j++į	j++;	1++?
j++;	j++i	j++į	j++į	j++į	j++;	j++į	j++į
j++;	j++i	j++į	j++į	j++į	j++;	j++į	j++į
j++;	j++i	j++į	j++į	j++į	j++į	j++į	1++?
j++;	j++i	j++į	j++į	j++į	j++;	j++į	j++į
j++;	j++į	j++į	j++ <i>į</i>	j++į	j++;	j++į	j++;
j++;	j++į	j++į	j++ <i>į</i>	j++į	j++;	j++į	j++;
j++;	j++i	j++į	j++į	j++į	j++;	j++į	j++į
j++;	j++i	j++į	j++į	j++į	j++;	j++į	j++į
j++j	1++?	j++į	1++?	j++j	j++į	1++?	1++?
j++j	1++?	j++į	j++;	j++;	1++5	j++;	j++j
j++j	1++?	j++į	j++;	j++;	1++5	j++;	j++j
j++;	j++i	j++į	j++į	j++į	j++;	j++į	j++į
j++;	j++i	j++į	j++į	j++į	j++;	j++į	j++į
j++;	j++i	j++į	j++į	j++į	j++;	j++į	j++į
j++;	j++i	j++į	j++į	j++į	j++;	j++į	j++į
j++;	j++i	j++į	j++į	j++į	j++;	j++į	j++į
j++;	j++i	j++į	j++į	j++į	j++;	j++į	j++į
j++;	j++i	j++į	j++į	j++į	j++;	j++į	j++į
j++j	1++?	j++į	j++;	j++;	1++5	j++;	j++j
j++;	j++į	j++į	j++ <i>į</i>	j++į	j++;	j++į	j++j
j++;	j++į	j++į	j++ <i>į</i>	j++į	j++;	j++į	j++j
j++;	j++i	j++į	j++į	j++į	j++;	j++į	j++į
j++;	j++i	j++į	j++į	j++į	j++;	j++į	j++į
j++;	j++i	j++į	j++į	j++į	j++;	j++į	j++į
j++;	j++i	j++į	j++į	j++į	j++;	j++į	j++į
j++j	1++?	j++į	1++?	j++j	j++į	1++?	1++?
j++;	j++i	j++į	j++į	j++į	j++į	j++į	1++?
j++;	j++i	j++į	j++į	j++į	j++;	j++į	j++į
j++;	j++i	j++į	j++į	j++į	j++;	j++į	j++į
j++;	j++i	j++į	j++į	j++į	j++;	j++į	j++į
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j++;	j++i	j++į	j++į	j++į	j++;	j++į	j++į
j++;	j++i	j++į	j++į	j++į	j++į	j++į	1++?
j++j	j++j	j++į	j++ <i>į</i>	j++j	j++į		j++j
j++j	j++j	j++į	j++ <i>į</i>	j++j	j++į	j++j	j++j
j++j	j++j	j++į	j++ <i>į</i>	j++j	j++į	j++j	j++j
j++j	j++j	j++į	j++ <i>į</i>	j++j	j++į	j++j	j++j
1++?	1++?	j++;	1++?	j++;	j++i	1++?	1++?