TCH-101

B. TECH. (FIRST SEMESTER) MID SEMESTER EXAMINATION, 2018

(All Branches)

ENGINEERING CHEMISTRY

Time: 1:30 Hours

Maximum Marks: 50

Note:(i). This question paper contains two Sections.

(ii) Both Sections are compulsory.

Section—A

- 1. Fill in the blanks/True/False: (1×5=5 Marks)
 - (a) MOT theory was proposed by
 - (b) O₂ and N₂ molecules are diamagnetic.

(True/False)

- (c) AlCl₃ is nucleophile. (True/False)
- (d) The bond order of Li₂ molecule is
- 2. Attempt any five parts: (3×5=15 Marks)
 - (a) Explain about Free radicals.

(2)

TCH-101

- (b) Define Electrophiles with examples.
- (c) Write about Electromeric effect.
- (d) Write a short note on Carbenes.
- (e) Briefly write about homo and heterolytic cleavage of a covalent bond.
- (f) Write a short note on resonance in organic compounds.

Section-B

- 3. Attempt any two parts of choice from (a), (b) and (c). (5×2=10 Marks)
 - (a) Write a short note on Carbocations. Also write about its structure and stability.
 - (b) Explain Metallic bonding with the help of Electron sea theory.
 - (c) Write the main postulates of VSEPR theory with the example of shape of NH₃ molecule.
- 4. Attempt any two parts of choice from (a), (b) and (c). (5×2=10 Marks)
 - (a) Differentiate between bonding and antibonding molecular orbitals.
 - (b) Write a short note on hydrogen bonding with its classification and applications.
 - (c) Explain, why p-nitrophenol and onitrophenol has different solubility in water.

(3)

- 5. Attempt any two parts of choice from (a), (b) and (c). (5×2=10 Marks)
 - (a) Explain Aromatic electrophilic substitution reaction with the mechanism of nitration of benzene.
 - (b) Write the mechanisms of $S_N^1 \& S_N^2$ nucleophilic substitution reactions.
 - (c) Draw the MOT diagram of HF molecule with its bond order and magnetic nature.

TCH-101

500

F. No. : b-50